

Supporting information for

Facile synthesis of TiO₂/Mn₃O₄ hierarchical structures for fiber-shaped flexible asymmetric supercapacitor with ultrahigh stable and tailorable performance

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Calculation methods

1. Electrode

Moreover, the linear capacitance (C_l) and volume capacitance (C_v) of the electrode were given and the formula (1, 2) of the CV curve is put as follows:

$$C_l = \frac{\int IdU}{L\Delta U} = \frac{S}{2Lv\Delta U_p} \quad (1)$$

$$C_v = \frac{\int IdU}{V\Delta U} = \frac{S}{2Vv\Delta U_p} \quad (2)$$

Where ΔU_p is potential window, S is the area of the closed CV curve, v is scan rate and L/V is effective linear/volume of the working electrode.

2. Device

The linear capacitance (C_{cell-L}) and volume capacitance (C_{cell-V}) of the device were calculated from the CV curves by the following formulas (3, 4):

$$C_{cell-L} = \frac{\int IdU}{L\Delta U} = \frac{S}{2Lv\Delta U_v} \quad (3)$$

$$C_{cell-V} = \frac{\int IdU}{V\Delta U} = \frac{S}{2Vv\Delta U_v} \quad (4)$$

Where ΔU_v is voltage window, S is the area of the closed CV curve, v is scan rate and L/V is effective length/volume of the device.

Additionally, the volume capacitance (C_{cell-V}) of the device based GCD curves can be calculated from the following equation:

$$C_{cell-V} = \frac{I\Delta t}{\Delta U_v} \quad (5)$$

Where I is the discharge current, Δt is the discharge time, ΔU_v is the voltage window. And the energy and power density could be estimated by the following formulas (6, 7):

$$E = \frac{C_{cell-V}\Delta U_v^2}{2V} \quad (6)$$

$$P = \frac{E}{\Delta t} \quad (7)$$

Where the C_{cell-V} is the volume capacitance of the working device from formulas (5), ΔU_v is the voltage window. V is the effective volume and Δt is the discharge time of GCD curves.

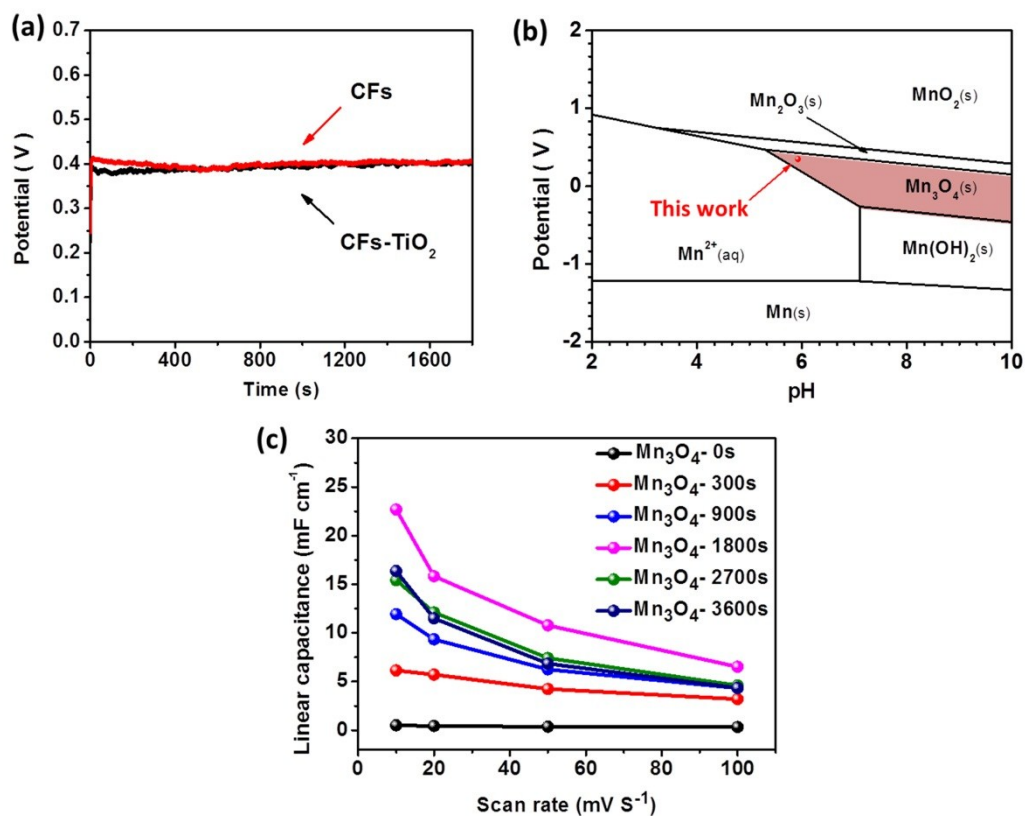


Fig. S1 (a) Voltage-time variation in electrodeposition process, which could reveal the stable voltage when applying slight current at 0.5 mA to the working electrode. (b) Pourbaix (E-pH) diagram of 0.1 M Mn-H₂O system, which shows the thermodynamic stability of possible phases and ionic species in the deposition solution at different pH values and potentials in 70 °C. (c) Various electrodeposition time and corresponding electrochemical performance of the electrodes.

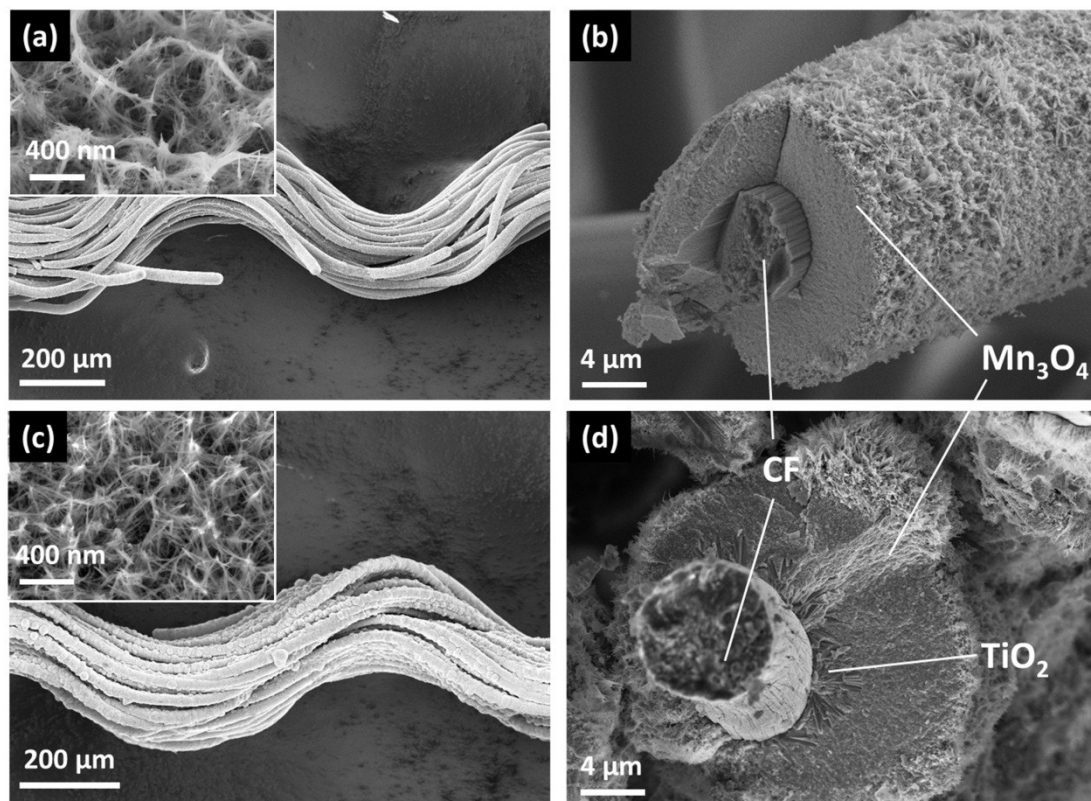


Fig. S2 SEM images of the Mn_3O_4 electrode and $\text{TiO}_2/\text{Mn}_3\text{O}_4$ electrode. (a) Typical image of the fiber-shaped Mn_3O_4 electrode. (b) Cross-sectional image of the Mn_3O_4 nanostructure covering on single carbon fiber. (c) SEM image of the fiber-shaped $\text{TiO}_2/\text{Mn}_3\text{O}_4$ electrode with many typical urchin-like Mn_3O_4 nanospines. (d) Cross-sectional image of the $\text{TiO}_2/\text{Mn}_3\text{O}_4$ nanostructure covering on single carbon fiber.

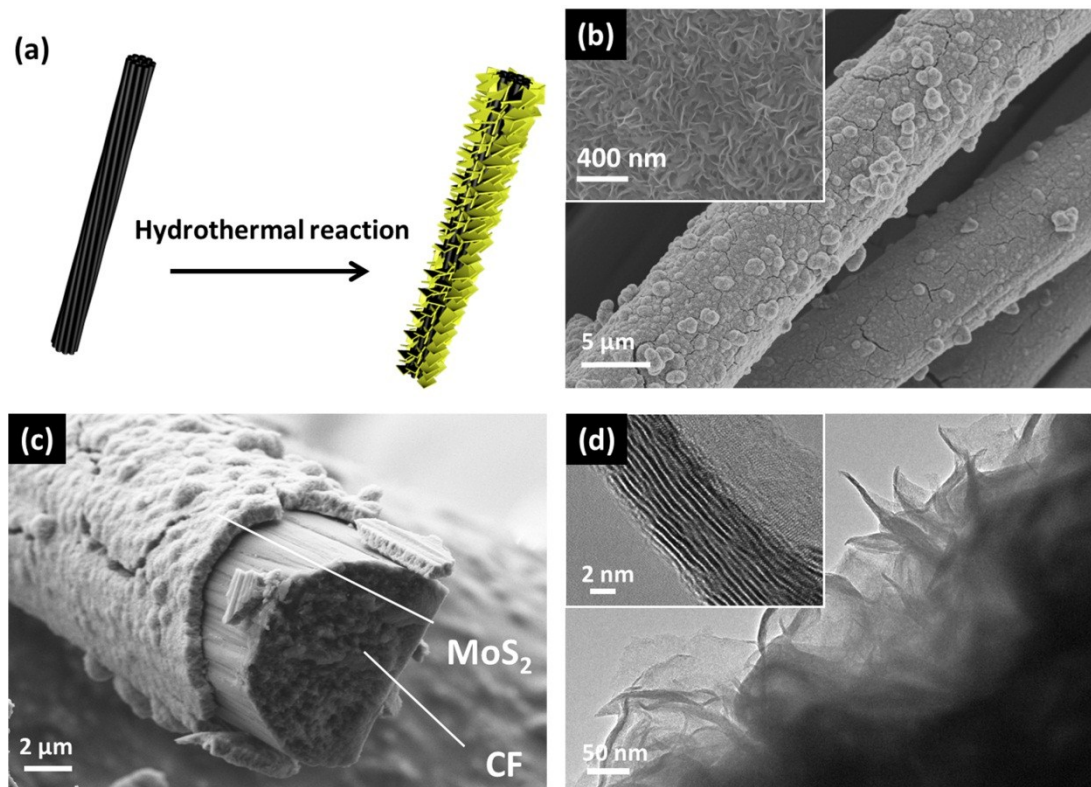


Fig. S3 (a) Schematic illustrating the synthesis procedure of MoS₂ electrode. (b) Typical image of the fiber-shaped MoS₂-coated electrode. (c) Cross-sectional image of the CF. (d) TEM image of the MoS₂ structure, insets show the typical lamellar structure of MoS₂.

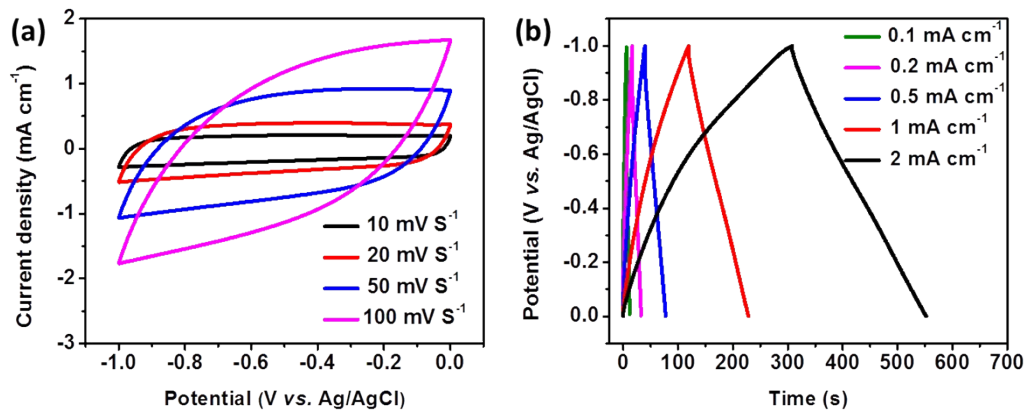


Fig. S4 (a) Different scan rates of CV curves for fiber-shaped MoS₂ electrode. (b) Galvanostatic charge/discharge curves at different current for fiber-shaped MoS₂ electrode. .

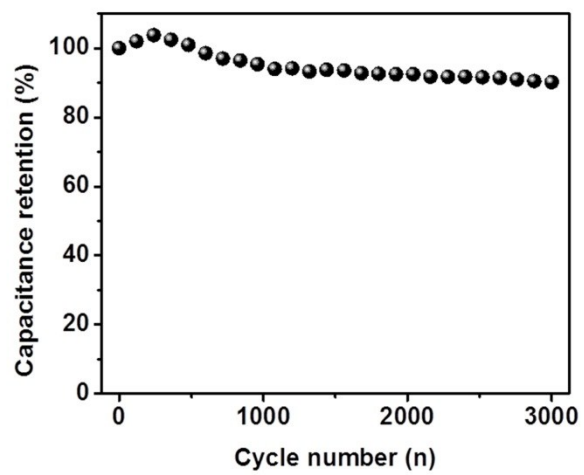


Fig. S5 Cycle stability for fiber-shaped MoS₂//TiO₂/Mn₃O₄ ASC device measured at 100 mV s⁻¹.

Table S1 Electrochemical performances of some previous fiber-shaped supercapacitors

Device configuration	Electrode material	Capacitance	Energy density	Power density	Cycle test	Time
Parallel	MnO ₂	2.5 F cm ⁻³	0.22 mWh cm ⁻³	0.4 W cm ⁻³	84 % at 10000 cycles	2012 [29]
Parallel	CNT/MnO ₂	25.4 F cm ⁻³	3.52 mWh cm ⁻³	127 mW cm ⁻³	n.a.	2014 [10]
Parallel	MnO ₂ @CNT	9.1 mF cm ⁻¹	0.78 mWh cm ⁻³	20 mW cm ⁻³	84.8 % at 1000 cycles	2015 [9]
Parallel	rGO@MnO ₂ @PPy	12.4 F cm ⁻³	0.20 mWh cm ⁻³	150 mW cm ⁻³	92 % at 4950 cycles	2015 [13]
Parallel	MnO ₂ /rGO/CF//GH/CW	2.54 F cm ⁻³	0.9 mWh cm ⁻³	0.2 W cm ⁻³	90 % at 10000 cycles	2015 [28]
Parallel	N-MoO _{3-x} //MnO ₂ @TiN	10.3 mF cm ⁻¹ / 4.1 F cm ⁻³	2.29 mWh cm ⁻³	1.64 W cm ⁻³	80.3 % at 5000 cycles	2016 [30]
Parallel	MoS ₂ //TiO ₂ /Mn ₃ O ₄	15.28 mF cm ⁻¹ / 1/19.84 F cm ⁻³	11.4 mWh cm ⁻³	2.34 W cm ⁻³	90.2 % at 3000 cycles	This work

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