

1 **Electronic Supplementary Information:**

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3 **Vertically Aligned 2D SnS₂ Nanosheets with Strong Photon**
4 **Capturing Capability for Efficient Photoelectrochemical Water**
5 **Splitting**

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20 **Experimental Details**

21 **Synthesis of Vertical 2D SnS₂ Nanosheets.** Vertically aligned 2D SnS₂ nanosheets

1 grown on FTO and CC substrates were synthesized by CVD method in a two-
2 temperature zone tube furnace. In a typical growth procedure, 100 mg $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$
3 powder was loaded in an alumina boat, was placed at the center of the downstream
4 heating zone. A series of growth substrates (FTO and CC) were placed downstream to
5 the $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$ powder. The distance of the $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$ alumina boat and the
6 substrates were 3, 8 and 11 cm, respectively. 200 mg S powder was placed in another
7 alumina boat at the upstream heating zone. Before the start of the growth, the furnace
8 chamber was flushed with 1200 sccm high purity Ar gas for ~20 min to decrease the
9 oxygen content in the chamber. Next, the center of the downstream heating zone was
10 heated to 450°C from room temperature at a ramp rate of 10°C . At the same time, the
11 temperature of the S powder at the upstream heating zone was set to 250°C . The
12 temperature in the downstream heating zone was kept at 450°C for 5 min for the SnS_2
13 growth under a mixture carrier gas of H_2 (15 sccm) and Ar (45 sccm). After the
14 growth, the furnace was naturally cooled down to room temperature and vertically
15 oriented 2D SnS_2 nanosheets on FTO and CC substrates were obtained.

16 **Characterizations.** The crystallographic information for the samples was collected
17 using a Philips Panalytical X'Pert Pro multipurpose diffractometer with Cu-K α X-ray
18 radiation ($\lambda = 1.5418 \text{ \AA}$). The morphologies of the samples were examined using a
19 Hitachi SU8000 field-emission scanning electron microscope (FE-SEM) with a 20 kV
20 accelerating voltage. The detailed microstructure, growth orientations and
21 composition of the as-grown 2D SnS_2 nanosheets were carried out by a Tecnai-G2
22 F30 transmission electron microscop (TEM) attached with an energy dispersive X-ray

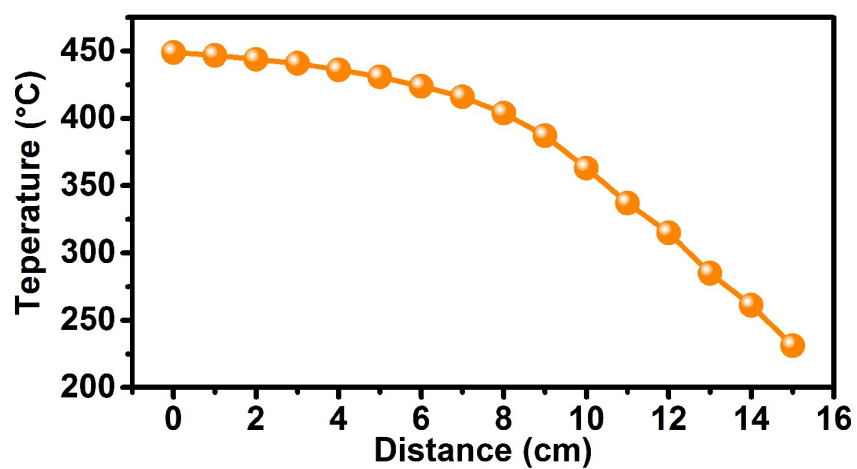
1 spectroscopy (EDS) system, operating at 300 kV. The temperature gradient in the
2 furnace was detected by using a TES-1310 digital thermometer equipped with a
3 WRNK-81530 thermocouple (TES Electrical Electronic Corp., Taipei). The UV-
4 Visible extinction spectra were recorded on a Hitachi U-4100 UV-Vis-NIR
5 spectrophotometer.

6 **Photoelectrochemical Measurements.** Prior to the measurements, the 2D SnS₂
7 nanosheets grown on FTO and CC substrates were first fabricated into photoanodes
8 by soldering a Cu mesh onto the bare part of the substrates. The area of the growth
9 substrates was $\sim 1 \text{ cm} \times 1 \text{ cm}$, and the loading amount of the 2D SnS₂ nanosheets was
10 in the range of 0.25-0.30 mg/cm² ($D_{ss}=11\text{cm}$, $\text{temp}=337\pm 1^\circ\text{C}$). The deposition amount
11 was estimated by weighing the difference in substrate's weight before and after the
12 growth using an analytical balance with 0.01 mg resolution (XS105 DualRange,
13 Mettler Toledo). For the preparation of parallel photoanode counterparts, vertically
14 aligned 2D SnS₂ nanosheets grown on FTO/CC substrates were firstly separated from
15 their growth substrates in ethanol-water solution by using ultrasonic treatment. After
16 that, the 2D SnS₂ nanosheets products were obtained in the ethanol-water solution.
17 Then, the parallel 2D SnS₂ nanosheets photoanodes were obtained as the above-
18 obtained solution was deposited onto a $1 \text{ cm} \times 1 \text{ cm}$ FTO substrate and acquired a
19 loading amount in the range of 0.25-0.30 mg/cm² after drying process.

20 The photoelectrochemical properties were investigated in a conventional three-
21 electrode system on an electrochemical workstation (CHI660D, CH Instruments),
22 with single-compartment quartz cell, a Pt foil counter electrode and an Ag/AgCl

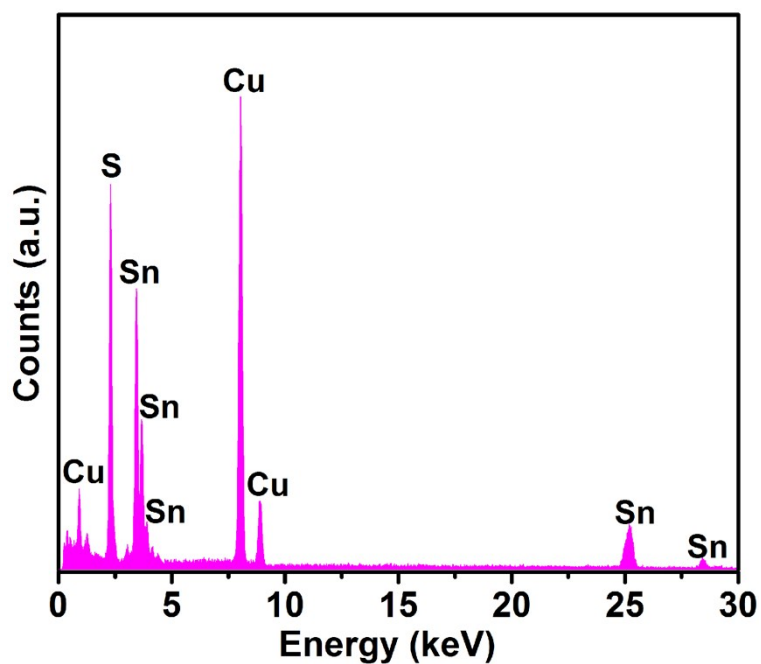
1 reference electrode. The fabricated photoanodes were used as the working electrode.
2 Na₂SO₄ solution (0.5 M, pH=7.58) was used as electrolyte after saturation with N₂ gas
3 for 30 min. The measured Ag/AgCl reference electrode potentials were converted to
4 V vs. reversible hydrogen electrode (RHE) by using the formula $E_{\text{RHE}} = E_{\text{Ag/AgCl}} +$
5 $0.0591\text{pH} + 0.1976 \text{ V}$. The photocurrent was measured by linear sweep voltammetry
6 in the range of 0.2-1.4 V vs. RHE at a scan rate of 10 mV/s. Visible-light irradiation
7 was achieved by a 500 W Xe lamp (CHF-XM500, Trusttech Co., Ltd.) fitted with an
8 AM 1.5 G filter. The light intensity was tuned to one Sun (100 mW/cm²). The time-
9 dependent current change curves were measured at the bias potential of 1.2 V vs.
10 RHE under the same 500 W Xe lamp irradiation. Electrochemical impedance
11 spectroscopy (EIS) was carried out at a bias potential of 1.2 V vs. RHE with a
12 frequency range of 100 kHz-100 mHz using the same lamp. The measured spectra
13 were fitted to the corresponding equivalent circuit model by using the program ZView
14 (Scribner Associates Inc.). The IPCE value was measured under monochromatic light
15 illumination by the 500 W Xe lamp through a monochromator (Omni λ -3007, Zolix
16 Instruments Co., Ltd.). The monochromator was incremented through the spectral
17 range (360-660 nm) to generate a photocurrent action spectrum with a sampling
18 interval of 20 nm. The light intensity was measured using a thermal power sensor
19 (S310C, Thorlabs) equipped with a readout power meter (PM100D, Thorlabs).
20 Conductivity characterizations of bare FTO and different SnS₂/FTO photoelectrodes
21 were performed by using a semiconductor characterization system (Keithley 4200
22 SCS) with a Lakeshore probe station.

1 Supplementary Figures



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3 **Figure S1.** The temperature distribution in the furnace.



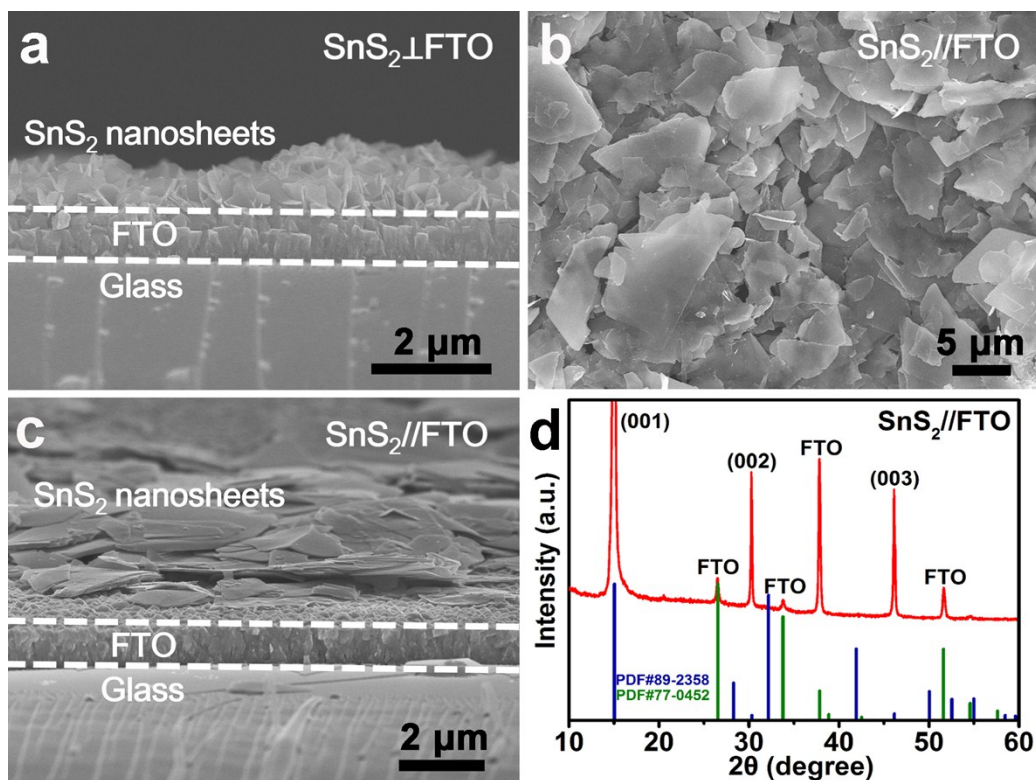
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5 **Figure S2.** EDS spectrum of vertically aligned SnS₂ nanosheets. EDS spectrum of the

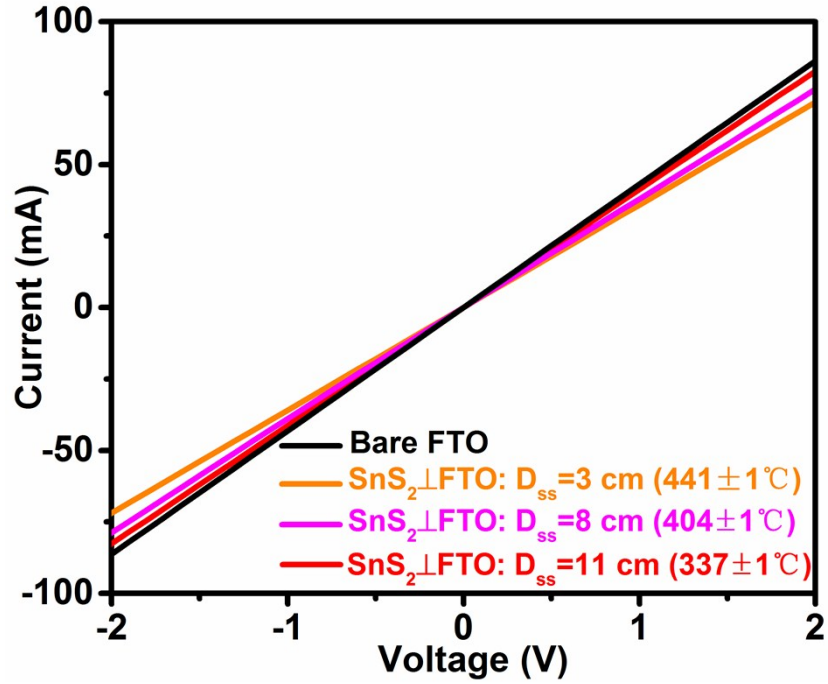
6 as-grown vertically aligned SnS₂ nanosheets confirming the elemental ratio of Sn and

7 S is ~1:2.

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2 **Figure S3.** SEM characterizations of SnS₂⊥FTO and SnS₂//FTO photoelectrodes. (a)
3 Cross-sectional SEM image of SnS₂⊥FTO photoelectrode. (b) SEM image of
4 SnS₂//FTO photoelectrode. (c) Cross-sectional SEM image of SnS₂//FTO
5 photoelectrode. (d) XRD pattern of SnS₂//FTO photoelectrode. The XRD pattern of
6 SnS₂//FTO photoelectrode reveals three diffraction peaks assigned to the (001), (002)
7 and (003) facets for the hexagonal SnS₂, indicating the [001] basal orientation is
8 highly preferred on FTO substrate, which is notably different from that of the
9 SnS₂⊥FTO and SnS₂⊥CC photoelectrodes (main text Fig. 3a).



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2 **Figure S4.** Comparison test of the conductivity of bare FTO and different
 3 SnS₂/FTO photoelectrodes. The current-voltage curves show little or no difference
 4 between the as-grown SnS₂/FTO photoelectrodes and bare FTO, implying
 5 our growth strategy makes no change to the conductivity of FTO after grown catalytic
 6 material, which is essential for the as-grown SnS₂/FTO photoelectrodes whether or
 7 not can be used in high performance PEC water splitting.

**1 Supplementary Table S1. Comparison of PEC performance of the vertically
2 aligned 2D SnS₂ nanosheets with other 2D-based photoelectrode materials.**

Photoelectrode materials	Preparation method	Electrolyte	Photocurrent density (mA/cm ²)	IPCE (%)	Reference
Zr-β-In ₂ S ₃ nanoflake	hydrothermal	0.5 M NaOH	1.1 mA/cm ² at 1.3V vs RHE	2.5% at 300 nm	1
β-In ₂ S ₃ nanoflake	hydrothermal	0.5 M NaOH	0.37 mA/cm ² at 0.7 V vs Ag/AgCl	-	2
CdS-WS ₂ nanosheets	liquid exfoliation	0.05 M Na ₂ S	0.35 mA/cm ² at 0.1 V vs SCE	-	3
CdS-MoS ₂ nanosheets	liquid exfoliation	0.05 M Na ₂ S	0.28 mA/cm ² at 0.1 V vs SCE	-	4
WSe ₂ thin films	liquid exfoliation	1 M H ₂ SO ₄	1.0 mA/cm ² at 0 V vs RHE	20% at 400 nm	5
ZnS thin films	pulsed laser deposition	0.24 M Na ₂ SO ₃	1.6 mA/cm ² at 1.0 V vs Ag/AgCl	8% at 200 nm	6
Au-MoS ₂ nanosheets	hydrothermal	0.1 KH ₂ PO ₄	0.79 mA/cm ² at 0.8 V vs Ag/AgCl	-	7
MoS ₂ nanosheets	hydrothermal	0.1 M KH ₂ PO ₄	0.26 mA/cm ² at 1.0 V vs Ag/AgCl	-	8
SnS ₂ single-layers	liquid exfoliation	0.5 M Na ₂ SO ₄	2.75 mA/cm ² at 1.0 V vs Ag/AgCl	38.7% at 420 nm	9
SnS ₂ nanosheets	solvothetmal	0.5 M Na ₂ SO ₄	0.017 mA/cm ² at 0.8 V vs SCE	-	10
parallel SnS ₂ //FTO	chemical vapor deposition	0.5 M Na ₂ SO ₄	0.91 mA/cm ² at 1.4 V vs. RHE	18.68% at 360 nm	This work
vertically aligned SnS ₂ ⊥ CC			1.92 mA/cm ² at 1.4 V vs. RHE	40.57% at 360 nm	
vertically aligned SnS ₂ ⊥ FTO			1.73 mA/cm ² at 1.4 V vs. RHE	36.76% at 360 nm	

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