

Supporting Information

Nitrogen, Phosphorus and Sulfur co-doped Ultrathin Carbon Nanosheet as a Metal-free Catalyst for Selective Oxidation of Aromatic Alkanes and Oxygen Reduction Reaction

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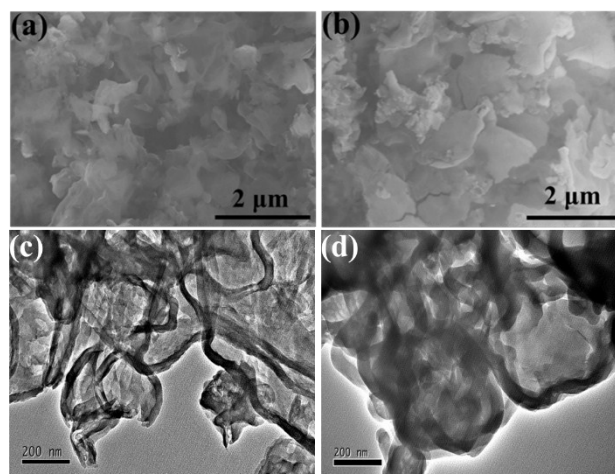


Fig. S1 (a, c) SEM and TEM images of g-C₃N₄, (b, d) SEM and TEM images of g-C₃N₄@PZS.

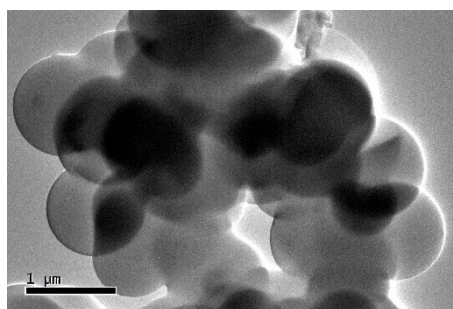


Fig. S2 TEM image of pure PZS

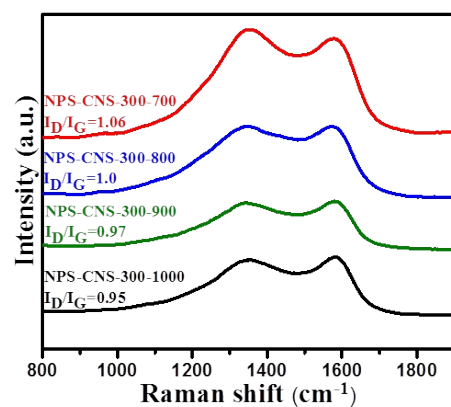


Fig. S3 Raman spectra of NPS-CNS-300-Y

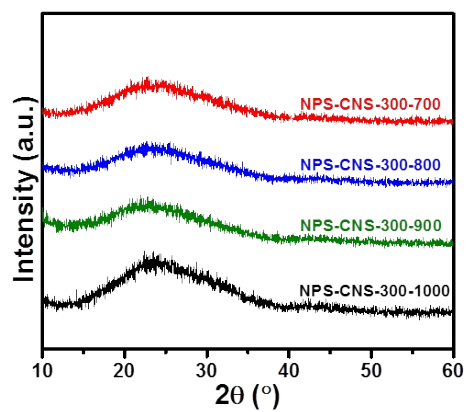


Fig. S4 XRD patterns of NPS-CNS-300-Y

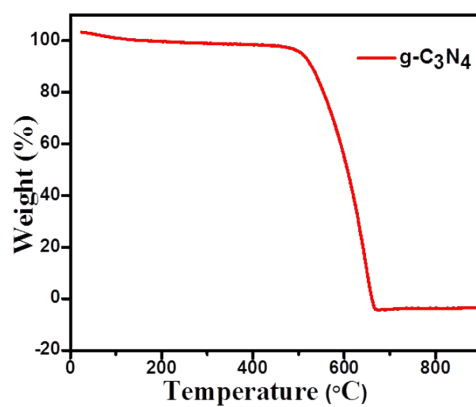


Fig. S5 TGA curve of g- C_3N_4 nanosheets

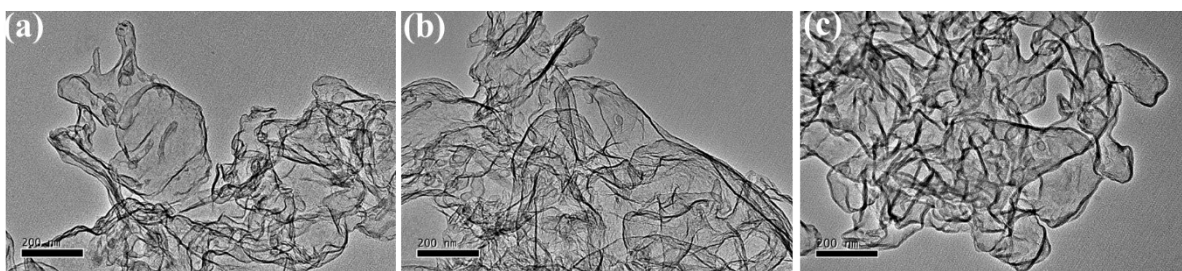


Fig. S6 TEM images of (a) NPS-CNS-300-700, (b) NPS-CNS-300-800 and (c) NPS-CNS-300-900

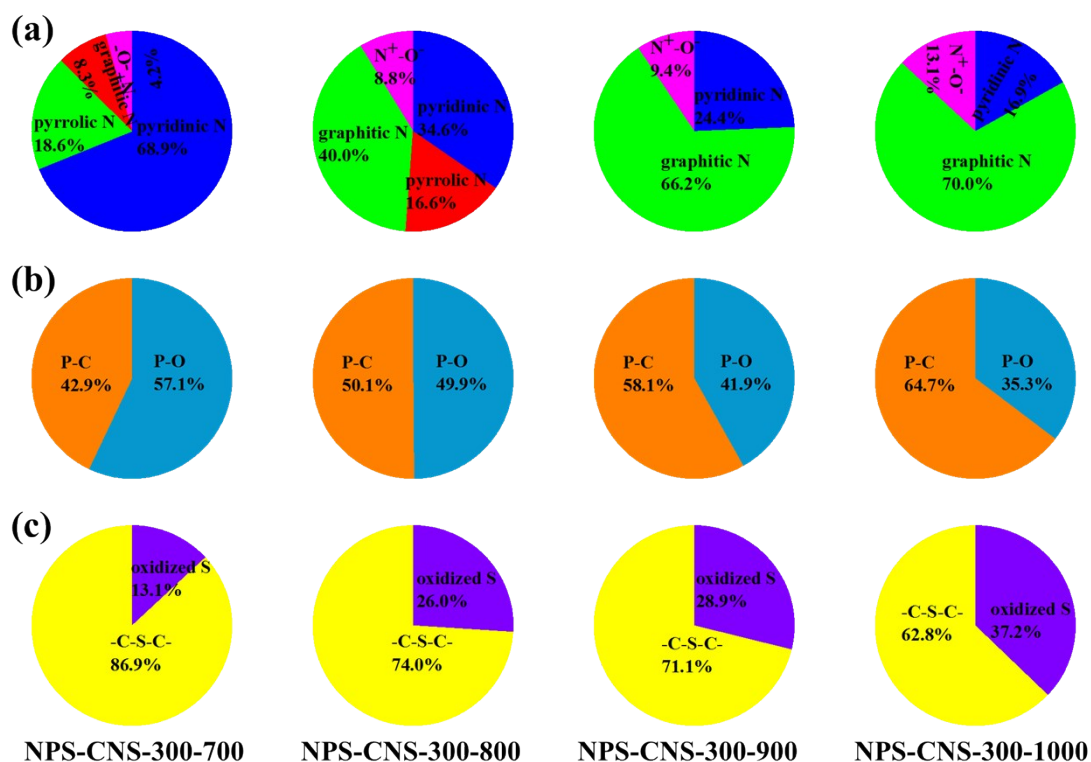


Fig. S7 The relative contents of (a) nitrogen species, (b) phosphorus species and (c) sulphur species in NPS-CNS-300-Y

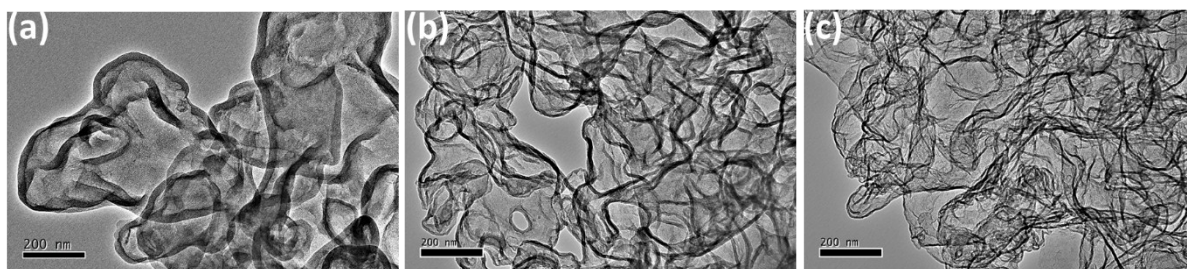


Fig. S8 TEM images of (a) NPS-CNS-100-1000, (b) NPS-CNS-200-1000, (c) NPS-CNS-400-1000

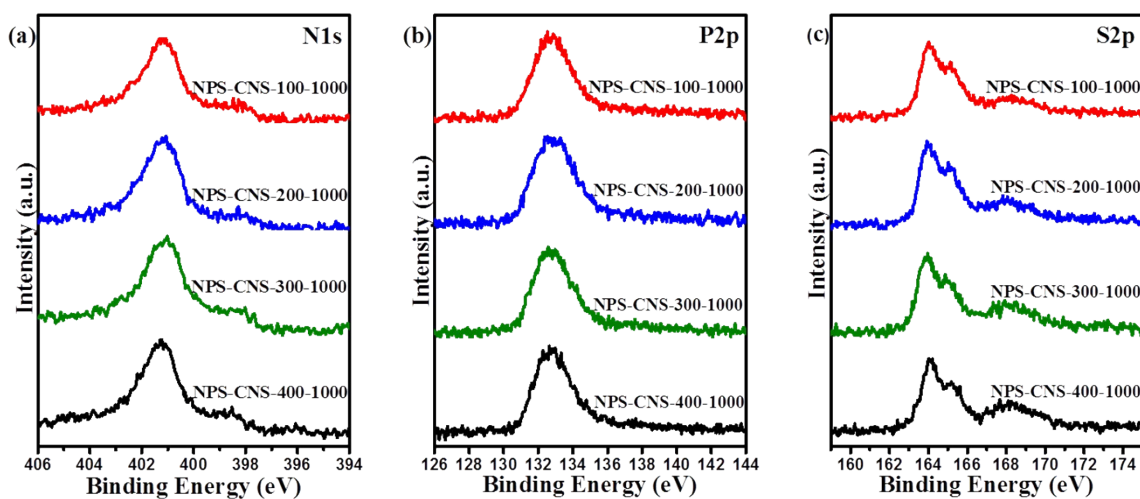


Fig. S9 XPS spectra of NPS-CNS-X-1000: (a) N1s, (b) P2p, and (c) S2p.

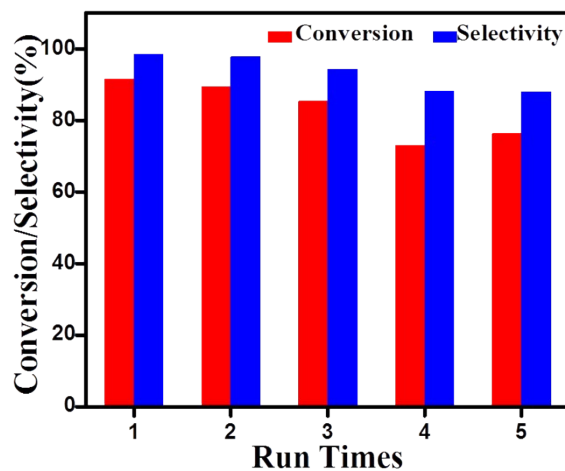


Fig. S10 Reusability of NPS-CNS-300-1000 for ethylbenzene oxidation. Reaction conditions: catalyst (8mg), ethylbenzene (0.5mmol), TBHP (0.5mL, 70wt% in water), H₂O (1mL), 80 °C, 6h.

Table S1 The atomic ratio (calculation from XPS analysis) and BET surface areas of all the samples

Sample	C(%)	N(%)	P(%)	S(%)	O(%)	Surface area (m ² g ⁻¹)
PZS-1000	91.25	1.18	0.74	0.70	6.13	562
NPS-CNS-300-700	53.20	29.12	6.81	0.32	10.56	330
NPS-CNS-300-800	76.48	9.09	2.85	0.53	11.06	755
NPS-CNS-300-900	84.34	4.75	1.60	0.59	8.72	807
NPS-CNS-300-1000	89.10	2.32	0.68	0.49	7.41	1198
NPS-CNS-100-1000	91.30	1.70	0.62	0.59	5.81	873
NPS-CNS-200-1000	89.77	2.12	0.69	0.58	6.85	915
NPS-CNS-400-1000	89.15	2.48	0.69	0.49	7.20	1031