

Electronic Supplementary Information

for

**Crystalline Cu-silicide stabilizes the performance of a high
capacity Si-based Li-ion battery anode†**

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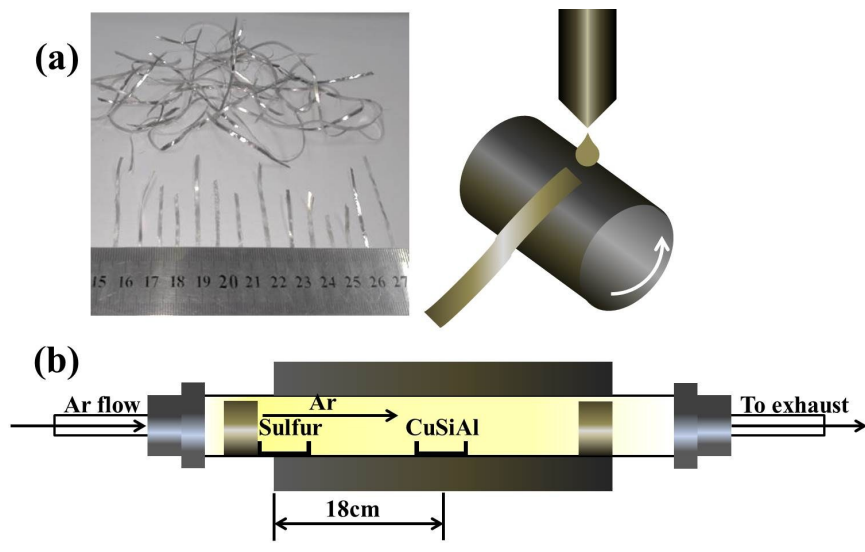


Fig. S1 a) The as-prepared melt spun Cu-Si-Al ribbon (left), and simplified schematics of melt spinner (right). b) Schematic diagram illustrate the Cu-Si-Al-S alloy synthesis setup using a flow tube reactor.

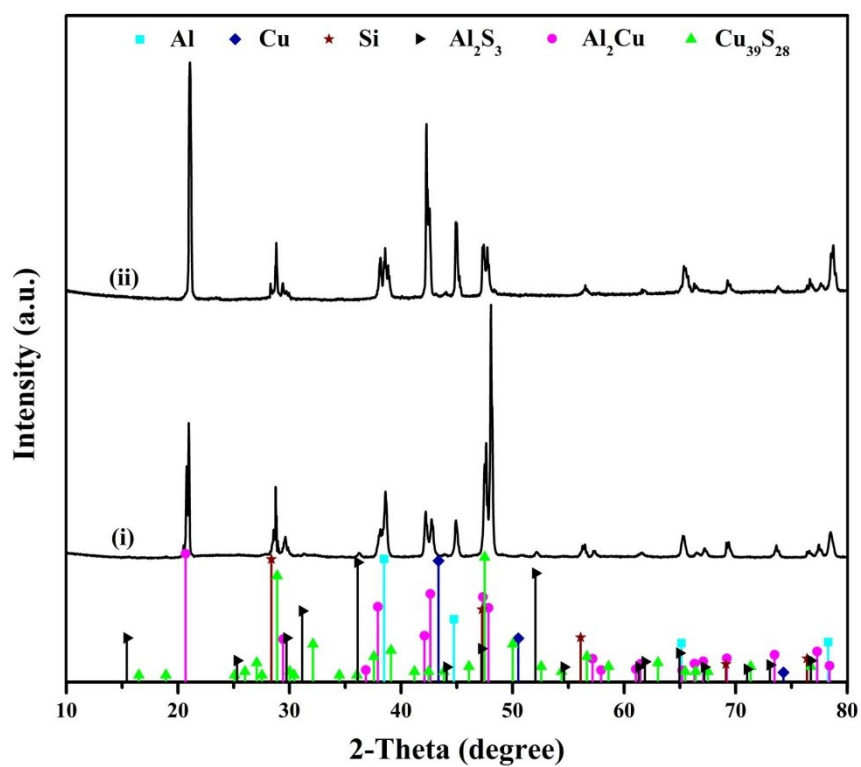


Fig. S2 XRD patterns of i) Cu-Si-Al-S alloy, ii) Cu-Si-Al-S alloy after water etching at 60°C for 36h.

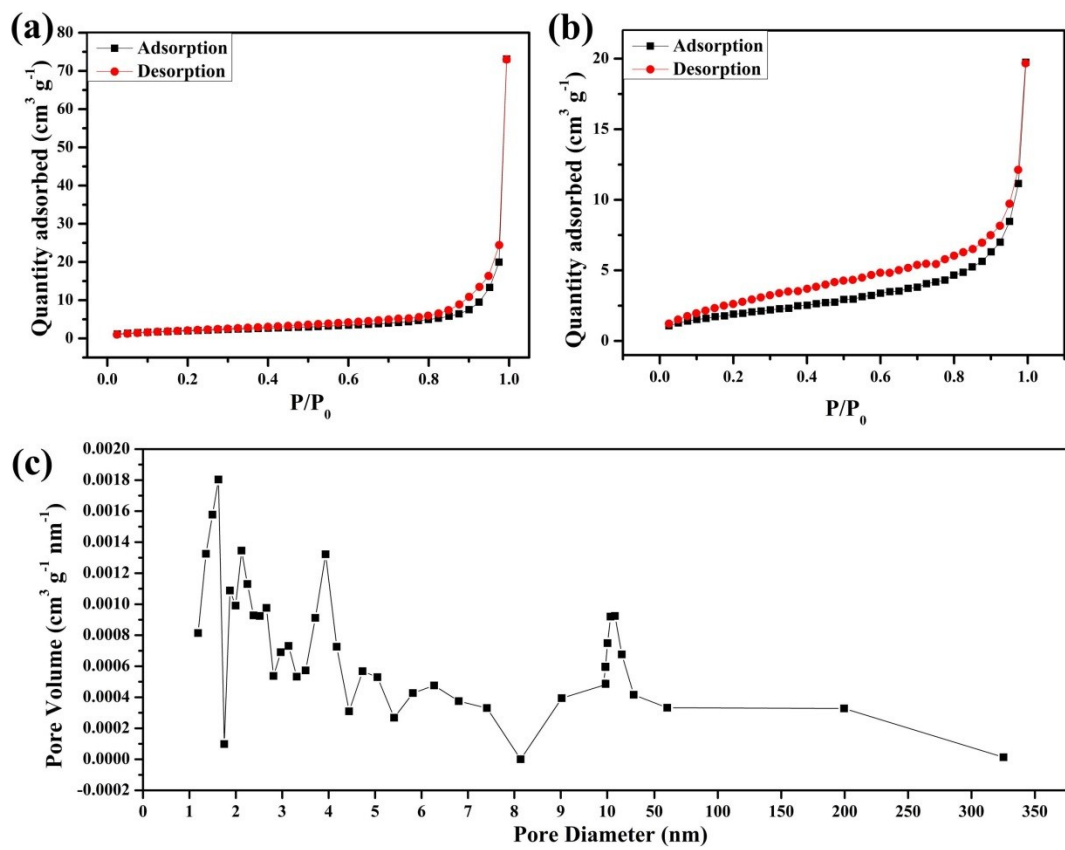


Fig. S3 N_2 adsorption isotherms of a) Si_{SERE} and b) HMSi. c) BJH pore size distribution curve of Si_{SERE} .

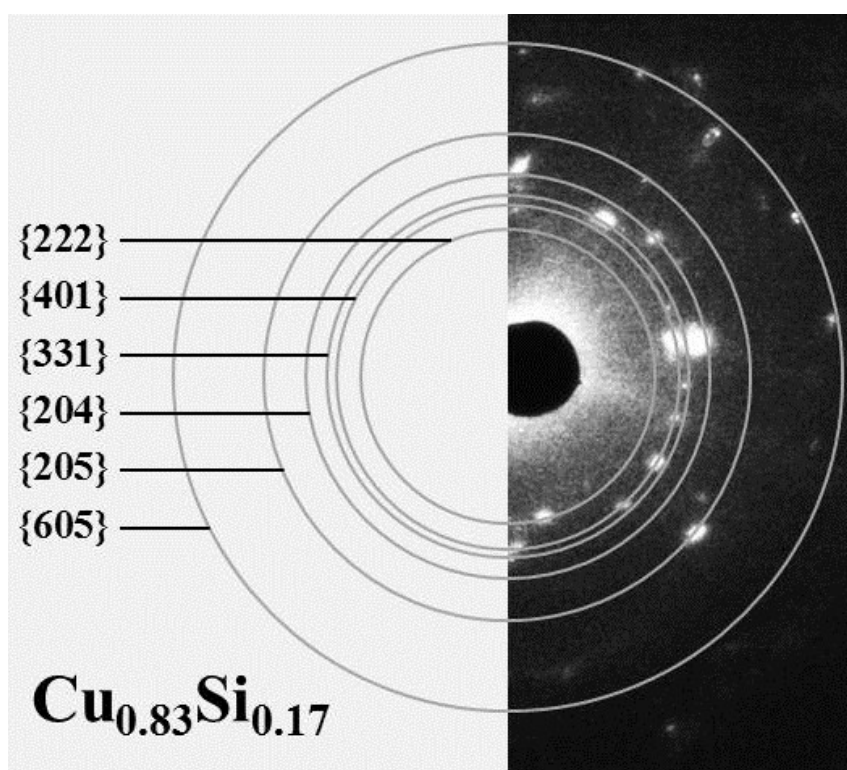


Fig. S4 SAED pattern of pure $\text{Cu}_{0.83}\text{Si}_{0.17}$ particles.

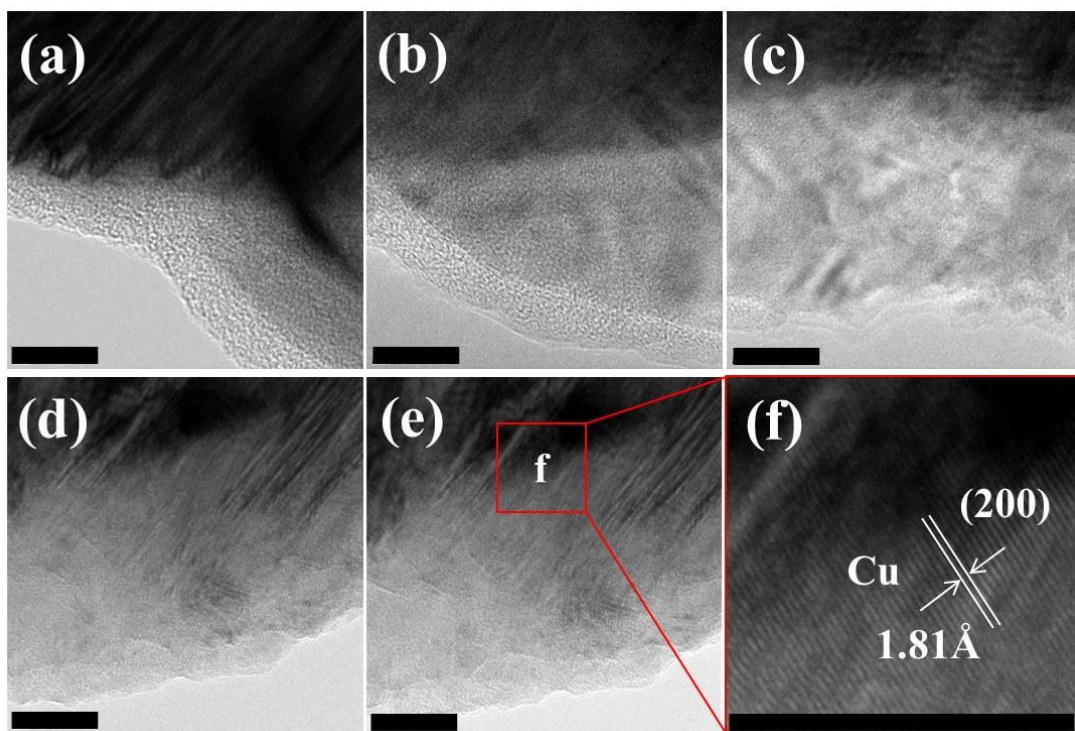


Fig. S5 HRTEM images of Si_{SERE} composite obtained by the In situ TEM experiment during the first lithiation (the scale bar is 10 nm).

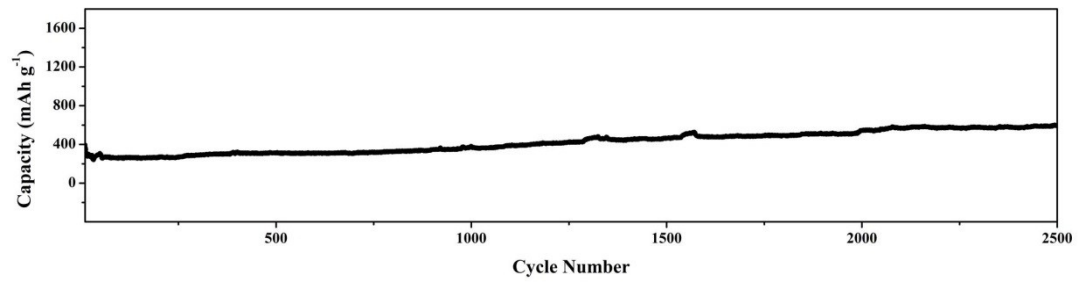


Fig. S6 Long-term cycling stability of Si_{SERE} electrode at a current density of 12 A g⁻¹.

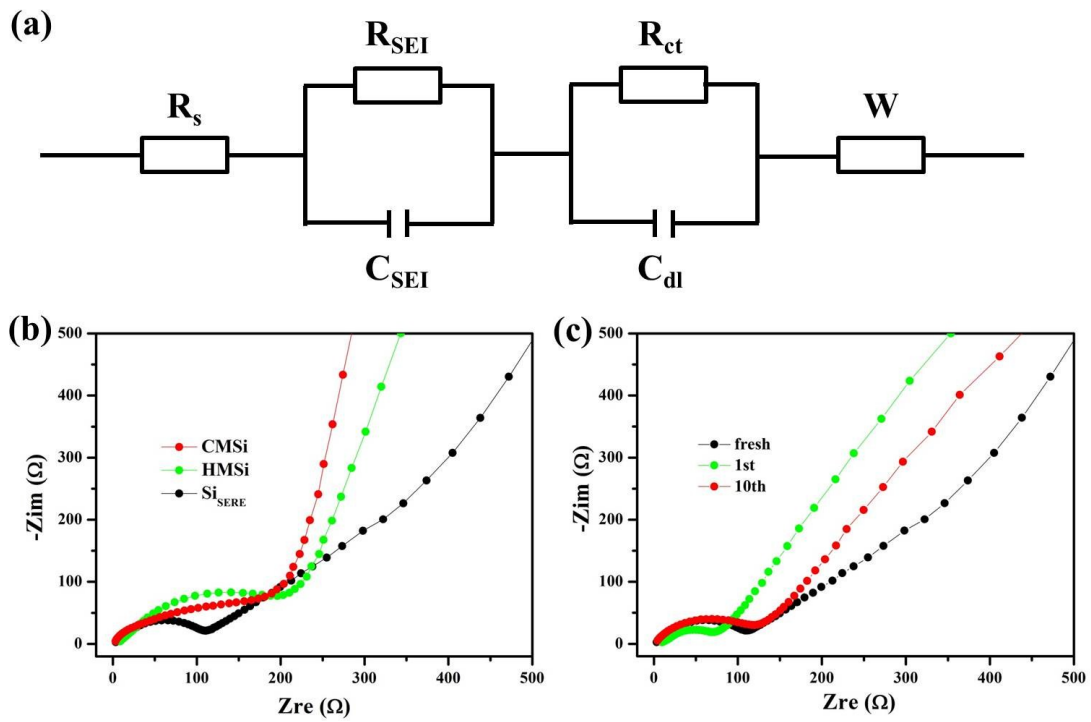


Fig. S7 a) An equivalent circuit model to fit the Nyquist plots. Nyquist plots of b) Si_{SERE} , HMSi, and CMSi at the fresh state; c) Si_{SERE} at different cycling stages by applying a sine wave with an amplitude of 20 mV over a frequency range of 100 kHz to 0.01Hz.

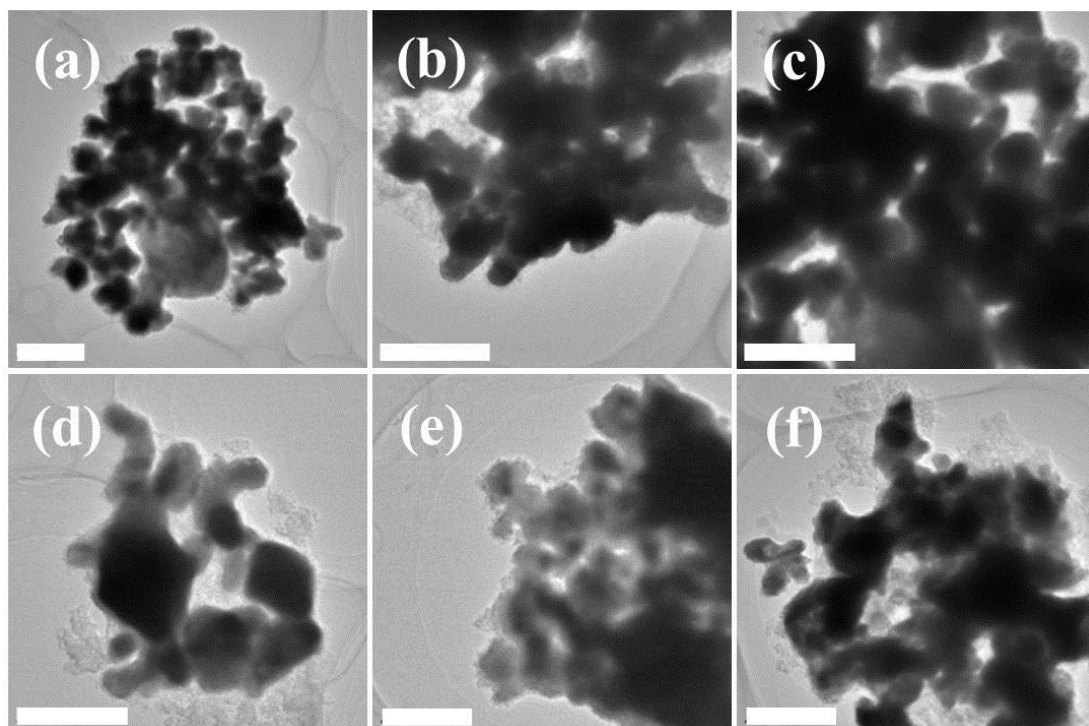


Fig. S8 TEM images of Si_{SERE} electrode after 200 cycles at 1.5 A g⁻¹ (the scale bar is 500nm).

Table S1 The EIS fitting results of different electrode materials at the fresh state.

	R_s (Ω)	R_{ct} (Ω)
Si _{SERE}	14.26	71.7
HMSi	5.272	160.4
CMSi	14.09	199.6