

Electronic Supplementary Information for

Facile synthesis of hierarchical fern leaf-like Sb and its application as additive-free anode for fast reversible Na-ion storage

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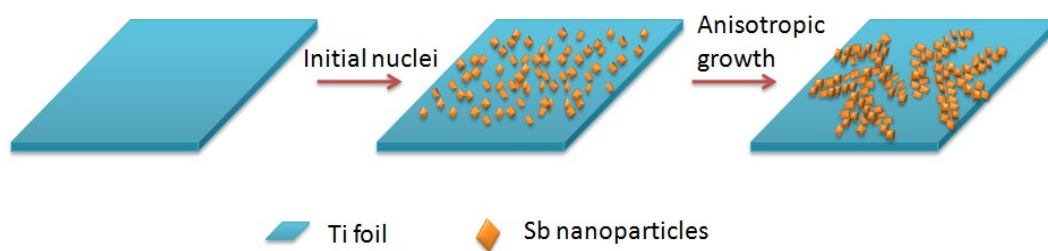


Fig. S1 Scheme illustration of the formation of fern leaf-like Sb on Ti foil.

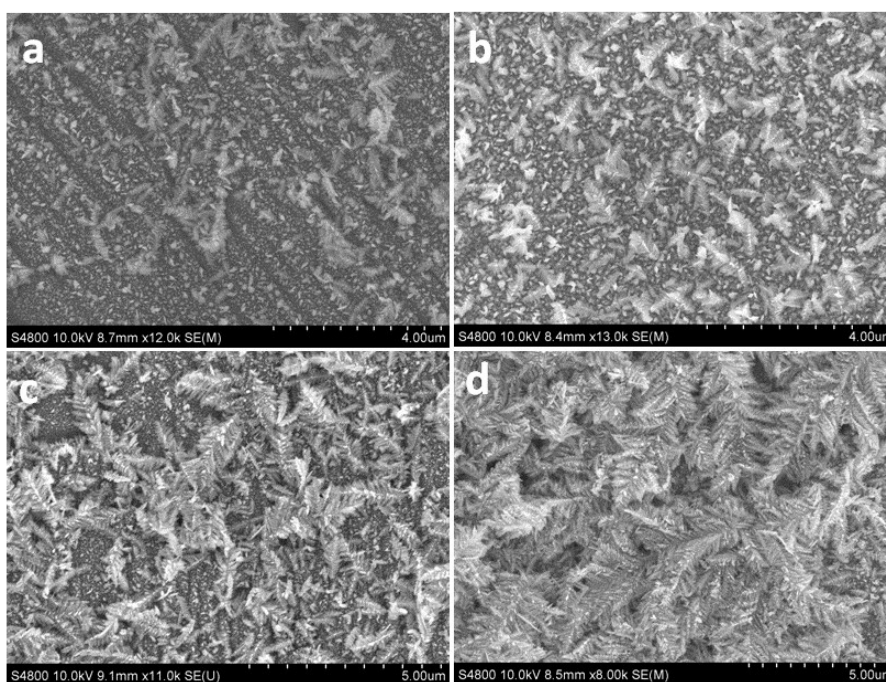


Fig. S2 Fern leaf-like Sb prepared in different electrodeposition time: (a) 3 min, (b) 5 min, (c) 10 min, (d) 15 min.

Table S1 Cycling performance comparison of the as-prepared fern leaf-like Sb with some previously reported Sb-based anodes.

Materials	Current density (A g ⁻¹)	Reversible Capacity (mAh g ⁻¹)		Capacity retention
		2 nd cycle	150 th cycle	
Bi _{0.36} Sb _{0.64} -C ¹	0.2	~494	~320	~64.8%
Sb/C fibers ²	0.1	~422	~380	~90%
rGO/nano Sb composite ³	0.328	~590	~528	~89.5%
Sb/MWCNT nanocomposite ⁴	0.2	~502	~382 (120th)	~76.1%
Porous Sb/Cu ₂ Sb ⁵	0.066	~616	~485 (120th)	~78.7%
Bulk Sb ⁶	0.33	~540	~570	~105.6%
Sb@C coaxial nanotubes ⁷	0.1	~500	~410	~82%
Nanoporous Sb ⁸	0.1	~630	~600	~95.2%
Sb-NiSb-Ni heterostructures ⁹	0.066	~500	~450	~90%
Rod-like Sb-C composite ¹⁰	0.05	~560	~450 (110th)	~80.4%
Sb porous hollow microspheres ¹¹	0.66	~575	~502	~87.4%
Sb/graphene ¹²	0.328	~600	~530	~88.3%
Sb nanocrystals ¹³	0.66	~600	~580 (100th)	~96.7%
Fern leaf-like Sb	0.5	~612	~589	~96.2%

References

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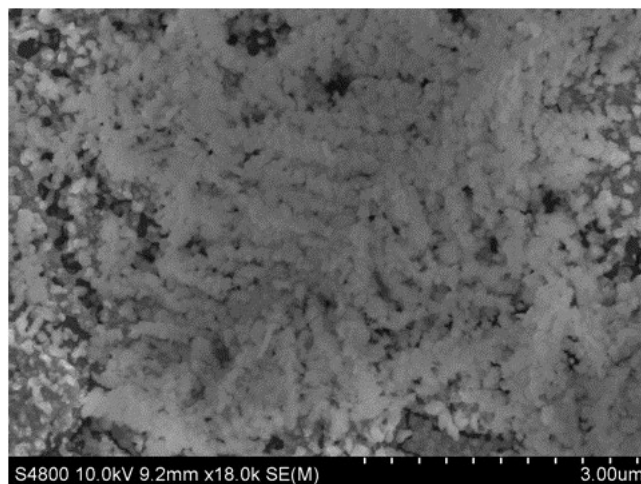


Fig. S3 SEM image of fern leaf-like Sb after 100 cycles at 0.5 A g^{-1} .

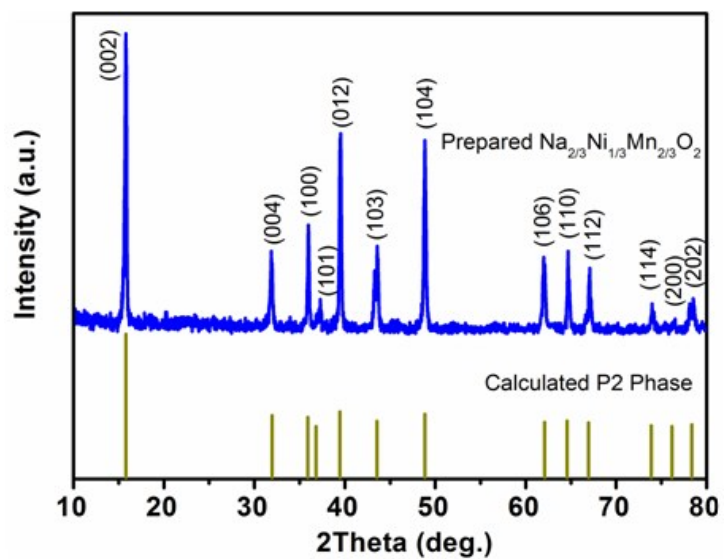


Fig. S4 XRD pattern of layered $\text{P2-Na}_{2/3}\text{Ni}_{1/3}\text{Mn}_{2/3}\text{O}_2$.

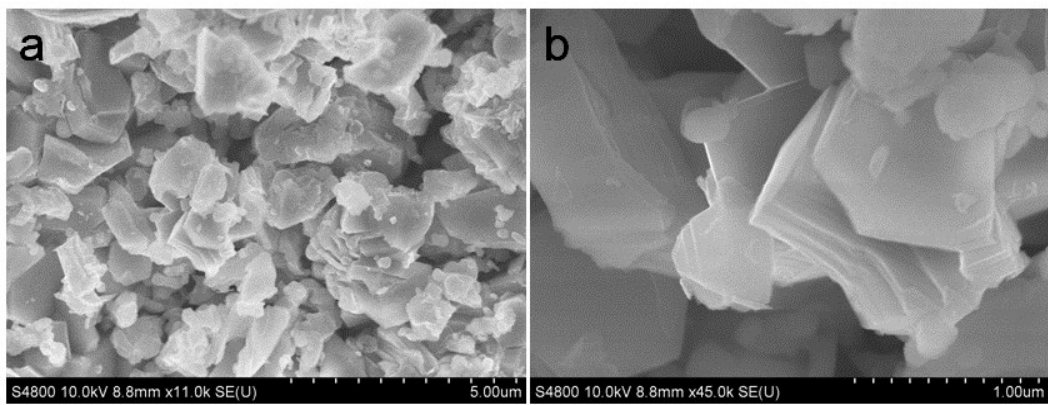


Fig. S5 SEM images of layered P2-Na_{2/3}Ni_{1/3}Mn_{2/3}O₂.

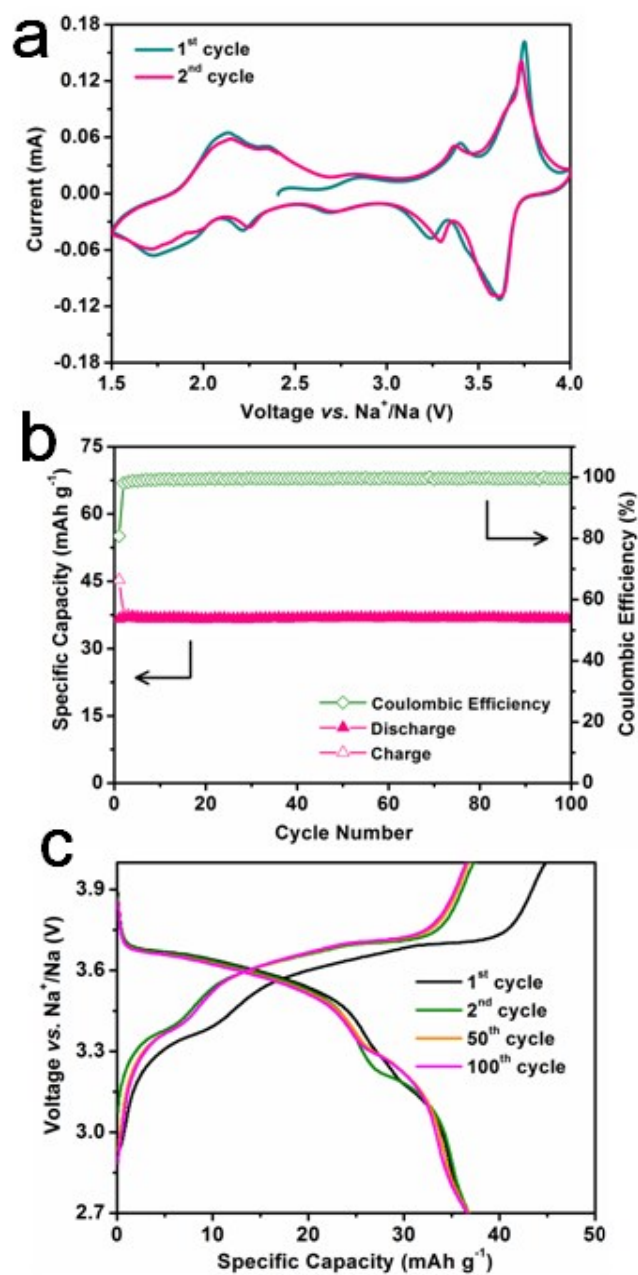


Fig. S6 Electrochemical performance of layered P2-Na_{2/3}Ni_{1/3}Mn_{2/3}O₂ in a Na cell. (a) Cyclic voltammetry at a scan rate of 0.1 mV s⁻¹ between 1.5 to 4.0 V (vs. Na⁺/Na). (b) Cycling performance at a current density of 30 mA g⁻¹, and (c) Galvanostatic voltage profiles in different cycles between 2.7 to 4.0 V (vs. Na⁺/Na).

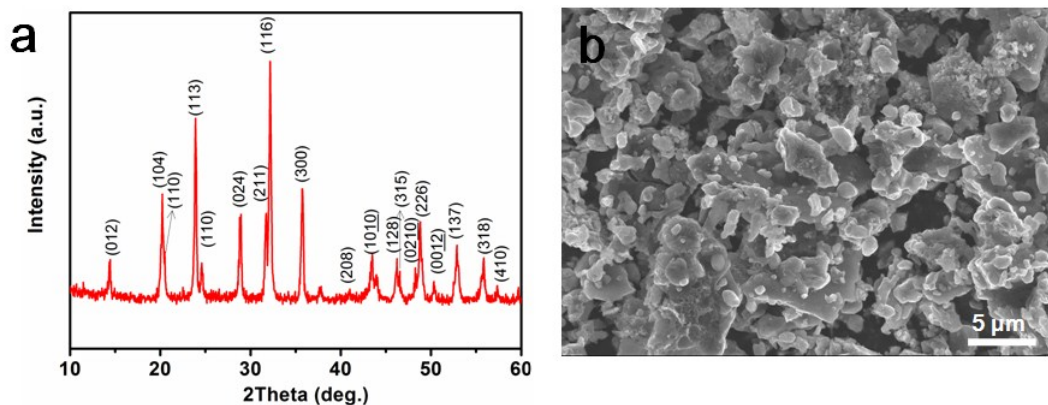


Fig. S7 (a) XRD pattern and (b) SEM image of $\text{Na}_3\text{V}_2(\text{PO}_4)_3/\text{C}$.

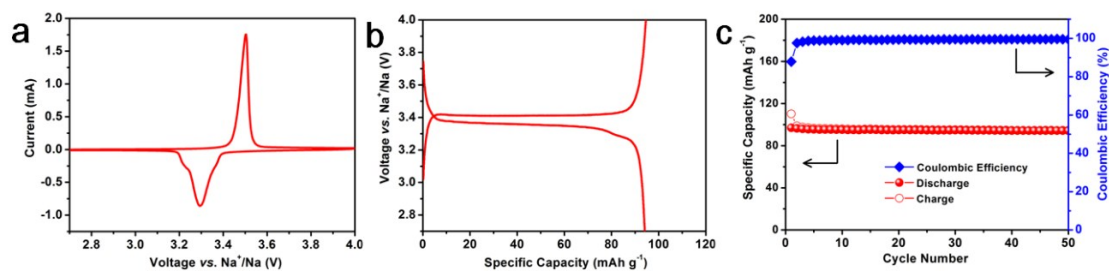


Fig. S8 Electrochemical performance of $\text{Na}_3\text{V}_2(\text{PO}_4)_3/\text{C}$ cathode. (a) Cyclic voltammetry at a scan rate of 0.1 mV s^{-1} between 1.7 to 4.0 V (vs. Na^+/Na). (b) Galvanostatic charge/discharge voltage profile, and (c) cycling performance at a current density of 80 mA g^{-1} between 2.7 to 4.0 V (vs. Na^+/Na).

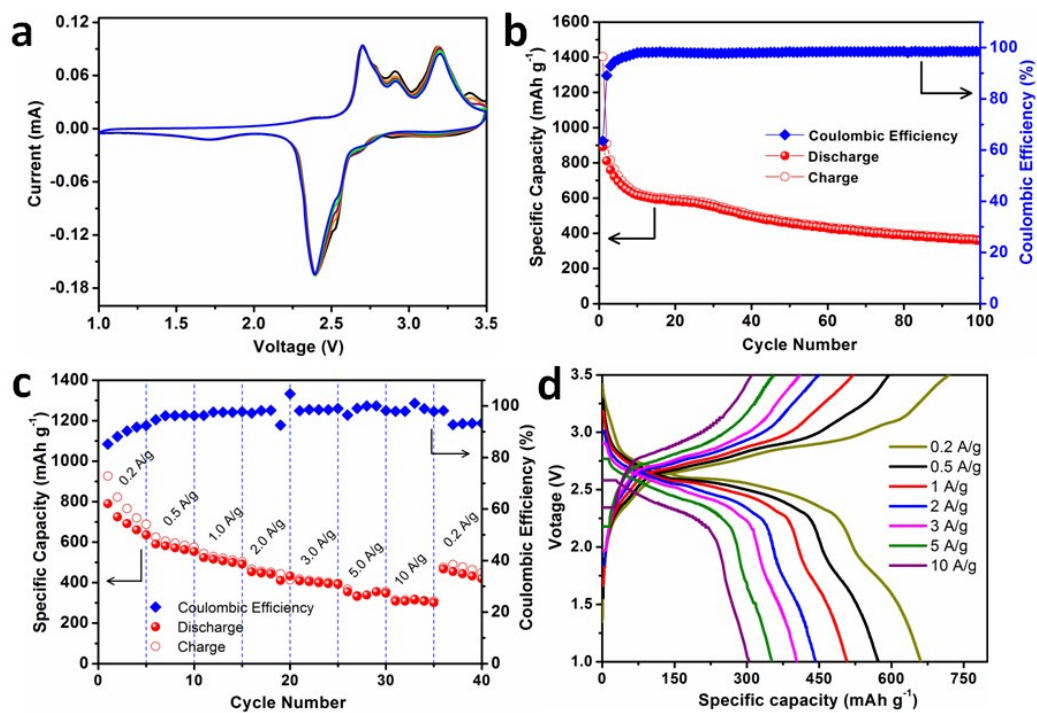


Fig. S9 Electrochemical performance of fern leaf-like Sb//Na₃V₂(PO₄)₃/C full cell. (a) CV curves at a scan rate of 0.3 mV s⁻¹. (b) Cycling performance at a current density of 0.5 A g⁻¹ (with respect to the anode weight). (c) Rate capability (with respect to the anode weight) at various current densities from 0.2 to 10 A g⁻¹. (d) Charge/discharge voltage profiles at various current densities from 0.2 to 10 A g⁻¹.