

Supporting Information

Atomic Layer Deposited Molybdenum Disulfide on Si Photocathodes for Highly Efficient Photoelectrochemical Water Reduction Reaction

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S1. Images of wafer scale ALD MoS₂

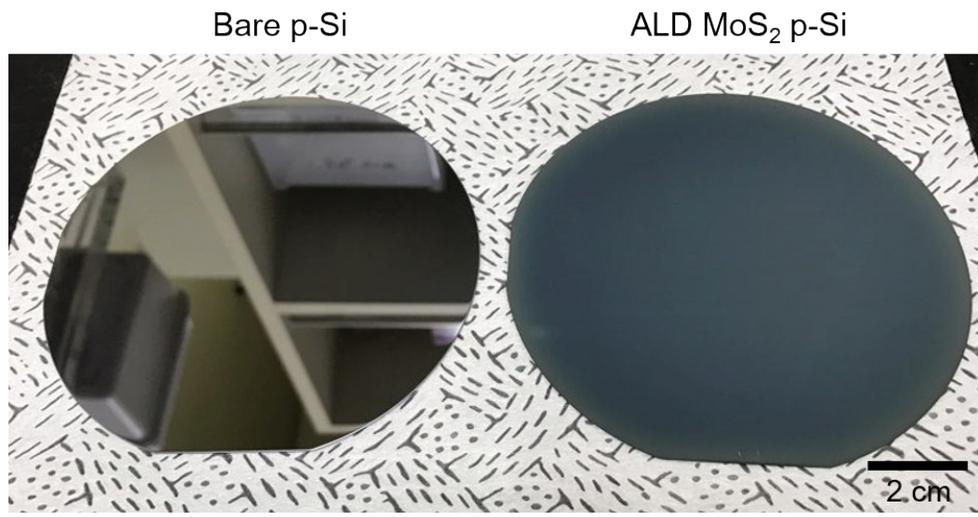


Figure S1. A digital image of a 4 inch p-Si wafer (left) and ALD MoS₂ on 4-inch p-Si wafer with 1180 ALD cycles (right).

S2. SEM images of MoS₂ with various ALD reaction cycles

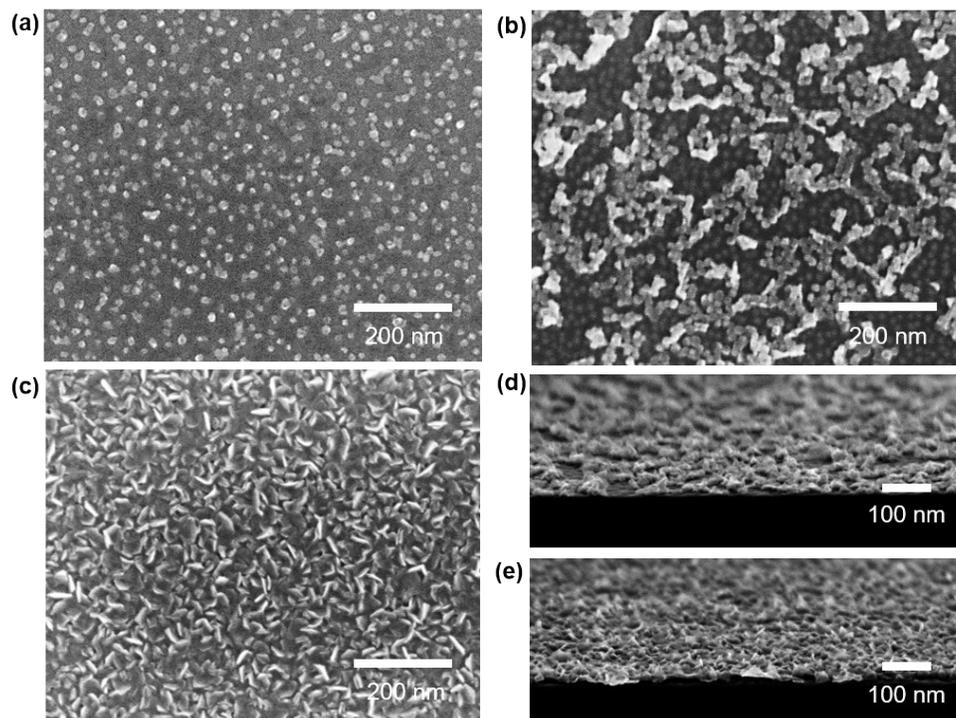


Figure S2. Plan view SEM images of as-grown ALD MoS₂ on p-Si wafers, which were deposited for (a) 60 cycles, (b) 295 cycles, and (c) 1180 cycles. Cross-sectional view SEM image of as-grown ALD MoS₂ film on p-Si wafers, which were deposited for (d) 295 cycles, (e) 1180 cycles.

S3. Growth rates of ALD MoS₂ on Si

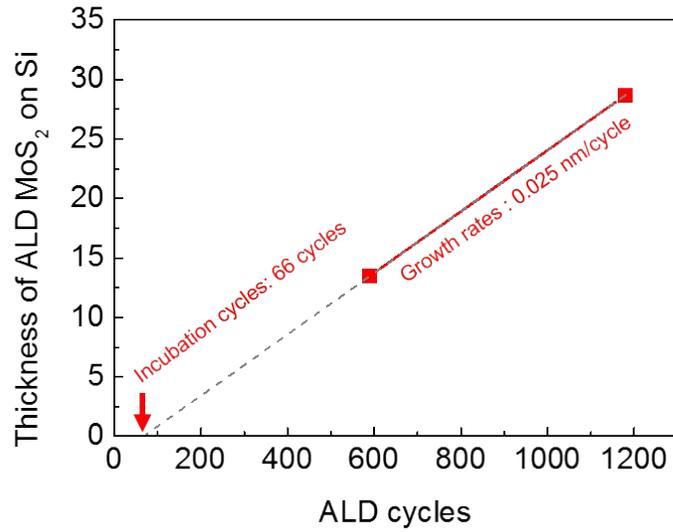


Figure S3. The growth rates of ALD MoS₂ on Si. These data are collected with only continuous ALD MoS₂ film.

S4. XRD and Raman spectra of MoS₂ with various ALD cycles and sulfurization temperatures

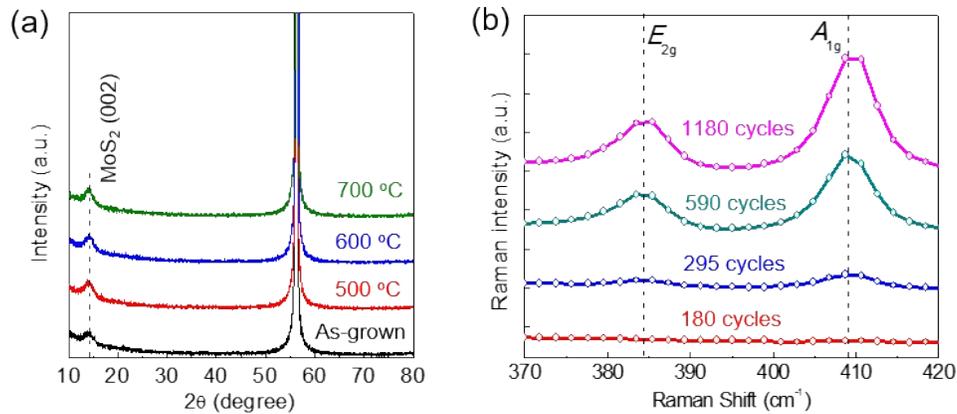


Figure S4. (a) XRD spectra of ALD MoS₂ with various sulfurization temperatures and (b) Raman spectra for sulfurized ALD MoS₂ deposited from 180 to 1180 cycles. Sulfurization temperature is 600 °C.

S5. Electrochemical Impedance Spectroscopy (EIS) of ALD MoS₂ on Si photocathodes

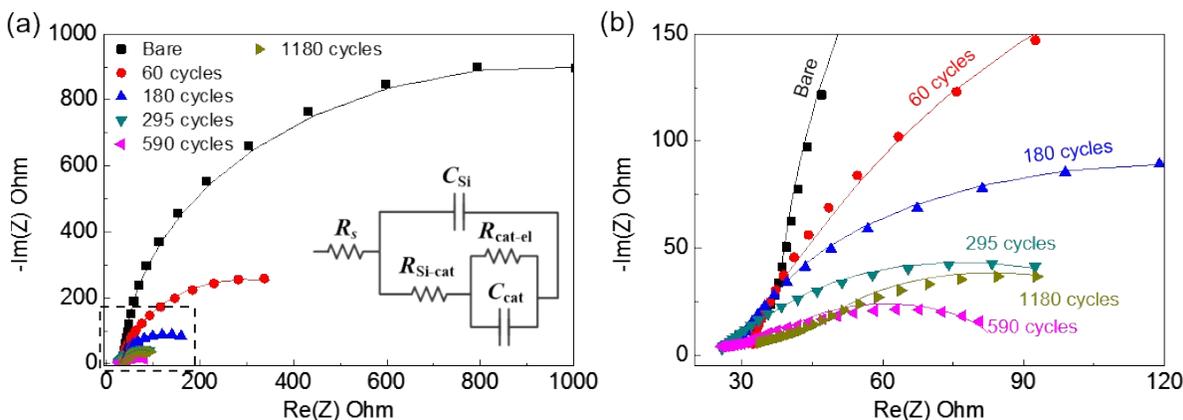


Figure S5. (a) Nyquist impedance plots of sulfurized ALD MoS₂ deposited for reaction cycles from 60 to 1180 cycles. Measurements were carried out at an applied bias of 0 V vs. RHE under 1 sun illumination in 0.5M H₂SO₄. The solid lines correspond to the fitting that uses the equivalent circuit in the inset of (a). (b) The magnified Nyquist plots of the dashed box in (a).

Inset of Fig. S5a shows a simplified equivalent circuit for impedance analysis of ALD MoS₂/p-Si, which was established for photoelectrodes with capacitive co-catalysts, such as MoS₂/Si and Co-Pi/Fe₂O₃.^{1, 2} This circuit is composed of 5 elements of resistances and capacitances; a series resistance of Si, MoS₂, and electrolyte, R_s , a capacitance of the Si substrate, C_{Si} , a charge transfer resistance between Si surface and MoS₂, $R_{\text{Si-cat}}$, a resistance between co-catalyst layer and electrolyte, $R_{\text{cat-el}}$, and a capacitance of MoS₂, C_{cat} . Two resistances generated at interface between two different materials are placed in series along for electrons to flow. In addition, since our ALD MoS₂ is thin, two capacitances of Si and MoS₂, C_{Si} and C_{cat} , are placed in parallel in the equivalent circuit. In order to extract resistance values, we performed 3 sets of EIS measurements for each sample. Note that C_{Si} in ALD MoS₂/p-Si photocathodes has a range

from 0.1 to 0.4 $\mu\text{F}/\text{cm}^2$ which are consistent to well-known capacitance of p-Si obtained by EIS.³

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Table S1. Charge transfer resistances of ALD MoS₂ on Si photocathodes.

Photocathode	ALD cycles	R_s ($\Omega \text{ cm}^2$)	$R_{\text{si-cat}}$ ($\Omega \text{ cm}^2$)	$R_{\text{cat-el}}$ ($\Omega \text{ cm}^2$)
Bare p-Si	-	30.5 \pm 2	-	1622 \pm 140
	60 cycles	25.5 \pm 3	143.93 \pm 20	424.23 \pm 44
	180 cycles	24.99 \pm 2	31.19 \pm 2	144.3 \pm 15
Sulfurized ALD MoS ₂ Si photocathode	295 cycles	27.97 \pm 3	20.02 \pm 5	66.37 \pm 12
	590 cycles	25.57 \pm 3	14.02 \pm 2	38.94 \pm 2
	1180 cycles	36.60 \pm 8	17.14 \pm 2	38.56 \pm 3

Note: R_s is the series resistance of an equivalent circuit; $R_{\text{si-cat}}$ is the resistance at the interface between the catalyst and Si surface; $R_{\text{cat-el}}$ is the resistance at the interface between the electrolyte and catalyst. All resistances are obtained by the average of 3 sets of EIS measurements.

S6. Optical transmittance of MoS₂ on quartz

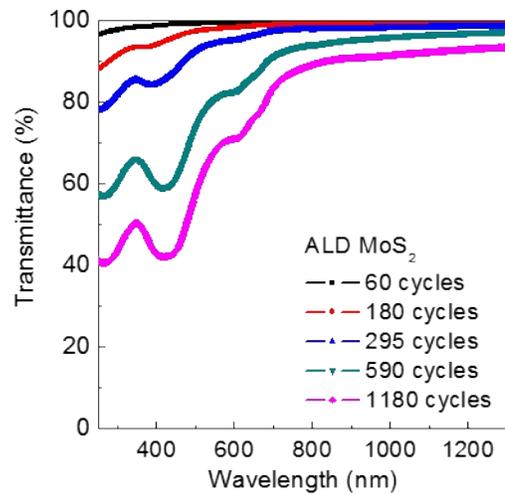


Figure S6. (a) Transmittance of ALD MoS₂ films on quartz that were sulfurized at 600 °C.

S7. PEC performance comparison of our and previously reported MoS₂ on Si photocathodes

Table S2. PEC performance of previously reported MoS₂ Si photocathodes compared with our studies.

#	Photocathode	V_{on} (V vs RHE)	j_{0V} (mA/cm ²)	j_{sat} (mA/cm ²)	methods	refs
1	MoS ₂ /p-Si	0.23	21.7	31	ALD	<i>this work</i>
2	Pt/p-Si	0.21	17	22.5	Electroless-deposition	<i>this work</i>
3	MoS ₂ /p-Si	0.17	24.6	40	Thermolysis followed by a layer transfer	5
4	MoS ₂ /TiO ₂ /p-Si NW	0.25	15	25	Thermolysis of spin-coated precursors	6
5	1T-MoS ₂ /p-Si	0.25	17.5	26.7	CVD	7
6	a-CoMoS _x /p-Si	0.25	17.5	20	Photo-assisted electrodeposition	2
7	MoS ₂ /Mo/n ⁺ p-Si	0.32	17	17.5	Direct sulfurization of Mo layer	8
8	a-MoS _x /Ti/n ⁺ p-Si	0.33	16	16	Electrodeposition	9

Note: V_{on} is the onset potential that is required for producing 1 mA/cm² of current density; j_{0V} is the photocurrent density measured at an applied bias 0 V vs RHE; j_{sat} is the saturation photocurrent density of photocathode; 1T-MoS₂ is the metallic crystalline MoS₂; n⁺p is the phosphorus doped Si emitter layer; Si NW is silicon nanowire; CVD is chemical vapor deposition.

S8. Pt/p-Si Si photocathode

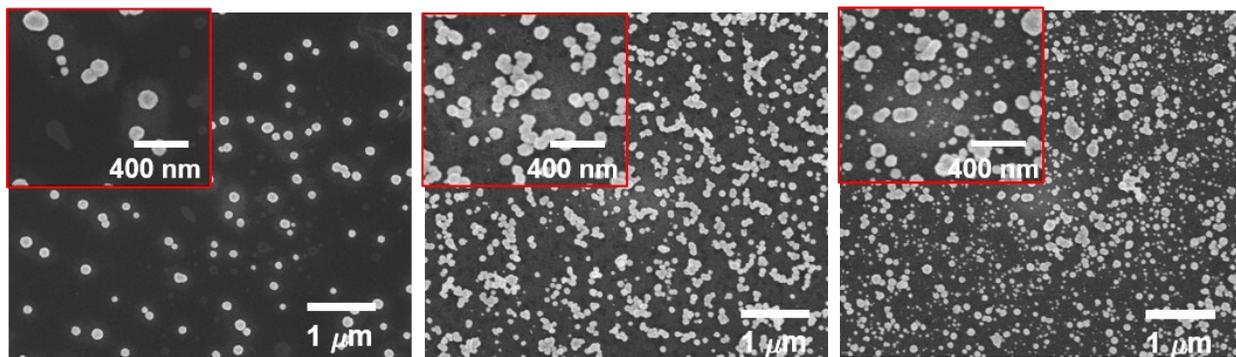


Figure S7. Plan view SEM images of p-Si with Pt nanoparticles deposited for (a) 5, (b) 9, and (c) 12 minutes. Each inset image indicates high resolution plan view SEM images of Pt/p-Si.

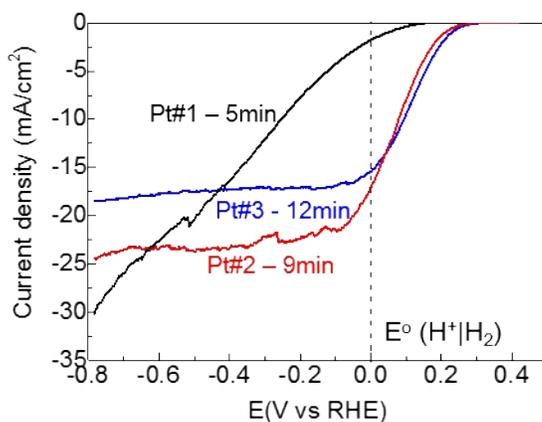


Figure S8. PEC j - V curves of p-Si with Pt nanoparticles deposited for 5 (black solid line), 9 (red solid line), and 12 (blue solid line) minutes in 0.5 M H₂SO₄ under simulated 1 sun illumination.

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