Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A. This journal is © The Royal Society of Chemistry 2017

Electronic Supplementary Information

Growing a Hydrophilic Nanoporous Shell on a Hydrophobic Catalyst Interface for Aqueous Reactions with High Reaction Efficiency and in situ Catalyst

Recycling

Yajuan Hao,^a Xuan Jiao,^a Houbing Zou,^a Hengquan Yang,^{*a} Jian Liu^{*b}

^aSchool of Chemistry and Chemical Engineering, Shanxi University, Taiyuan 030006, China E-mail: <u>hqyang@sxu.edu.cn</u>
^bDepartment of Chemical Engineering, Curtin University, Perth, WA 6845, Australia E-mail: <u>jian.liu@curtin.edu.au</u>

1. Definition of catalytic efficiency.

- 2. Table S1 Results of hydrogenation of methyl acrylate over Pd-supported catalysts
- 3. Fig. S1 FT-IR spectra of SiO₂, Pd/SF and Pd/SF@NS_{0.3}

Definition of catalytic efficiency. To quantitatively compare the efficiency of different structured catalysts, we defined the catalytic efficiency parameter CE as the number of moles of converted substrate per mole of metal per hour (mol mol⁻¹ h^{-1}).

$$CE = \frac{converted \ substrate(mol)}{Pd(mol) \times h}$$

The conversion was calculated through H_2 consumption monitored by the H_2 pressure decrease and verified by GC analysis. Conversions of less than 20% were used to calculate the efficiency of the catalysts.

catalyst	Pd/SiO ₂	Pd/SF@NS _{0.15}	Pd/SF@NS _{0.3}	Pd/SF@NS _{0.6}	Pd/SF@NS _{0.9}
Conversion (%)	61	94	98	90	84
^a 2 mmol of substrate; 0.1 mol% Pd; 6 mL of water; 25 °C, o.15 MPa H ₂ pressure, 1 h; determined					

by GC.



Figure S1. FT-IR spectra of SiO₂, Pd/SF and Pd/SF@NS_{0.3}.