## **Supporting Information**

## One-step synthesis of glycoprotein mimics *in vitro*: improvement of protein activity, stability and application in CPP hydrolysis

Yuecheng Cui,<sup>a</sup> Zhenhua Li,<sup>a</sup> Lei Wang,<sup>a</sup> Feng Liu,<sup>a</sup> Yuqi Yuan,<sup>a</sup> Hongwei Wang,<sup>\*a</sup>

Lulu Xue,<sup>a</sup> Jingjing Pan,<sup>a</sup> Gaojian Chen,<sup>b</sup> Hong Chen<sup>a</sup> and Lin Yuan<sup>\*a</sup>

*a* State and Local Joint Engineering Laboratory for Novel Functional Polymeric

Materials, College of Chemistry, Chemical Engineering and Materials Science,

Soochow University, 199 Ren'ai Road, Suzhou 215123, P. R. China.

*b* Center for Soft Condensed Matter Physics and Interdisciplinary Research, Soochow University, Suzhou 215006, P. R. China

E-mail: wanghw@suda.edu.cn; yuanl@suda.edu.cn



**Fig. S1** <sup>1</sup>H NMR spectrum of MAG. Solvent: D<sub>2</sub>O.



Fig. S2 Aqueous GPC of PMAG with increasing molecular weight.



**Fig. S3** <sup>1</sup>H NMR of PMAG (a) and the corresponding pyridyl-PMAG (b). Solvent:  $D_2O$ . The conversion of pyridyl-PMAG was calculated as below<sup>1, 2</sup>:

Conversion (%) = 
$$\frac{\delta_{8.4ppm}}{\delta_{2.5ppm}/2} \times 100\%$$
 = 95.28%



**Fig. S4** Free thiol content of PPase(N124C), PMAG-PPase conjugates (after purification) and WT (wild type) PPase (Mean  $\pm$  SD, n = 3). The conversion of conjugate was calculated as below:

Conversion (%) = 
$$\frac{A_{N124C}-A_{PMAG-PPase}}{A_{N124C}-A_{WT-PPase}} \times 100\% = \frac{98.26-9.68}{98.26-5.36} \times 100\% = 95.35\%$$



**Fig. S5** SDS-PAGE result. Lane M: molecular weight markers; lane A: PPase; lane B: PMAG-PPase conjugate (PM<sub>2</sub>-PPase); lane C: conjugate reduced by DTT.



**Fig. S6** MALDI-TOF mass of PPase, MAG-PPase and PM<sub>2</sub>-PPase. MAG-PPase conjugate was synthesized via thiol-ene click chemistry according to Cobo *et al.*<sup>3</sup> The process was similar with the synthesis of PMAG-conjugates, which the feed molar ratio of MAG/PPase was 50/1, reacted in 10 mM Tris-HCl buffer (pH 8.0) and measured via MALDI-TOF mass.



**Fig. S7** Specific activities vs. pH value of free PPase, MAG-PPase and PMAG-PPase conjugates of different PMAG molecular weight ( $M_n(PMAG) = 4.5$ , 16.0, and 29.0 kDa for PM<sub>1</sub>, PM<sub>3</sub>, and PM<sub>4</sub> respectively) as a function of pH value. Mean  $\pm$  SD, n = 3.



Fig. S8 Tolerance of  $PM_2$ -PPase conjugate to lysozyme and chitinase digestion (relative to the specific activity of  $PM_2$ -PPase without the incubation of glycosidases, Mean  $\pm$  SD, n = 3).



**Fig. S9** Relative rate of CPP hydrolysis using PPase and PM<sub>2</sub>-PPase, respectively (relative to the hydrolysis rate of unconjugated PPase).

## References

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- 2 C. Boyer, A. Granville, T. P. Davis and V. Bulmus, J. Polym. Sci. Part A: Polym. Chem., 2009, 47, 3773-3794.
- 3 I. Cobo, M. Li, B. S. Sumerlin and S. Perrier, Nat. Mater. 2015, 14, 143-159.