Supplementary Information

## Clustering of Hydroxyapatite on Super-Twisted Collagen Microfibril under Mechanical Tension

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Atom	Charge <sup>1</sup>	Non-bonded van der Waals parameters <sup>2, 3</sup>
Ca	+2.0	$\varepsilon = 0.496 \text{ kJ/mol}, \sigma = 0.294 \text{ nm}$
Р	+2.6	$\varepsilon = 2.447 \text{ kJ/mol}, \sigma = 0.349 \text{ nm}$
O(P)	-1.4	$\varepsilon = 1.054 \text{ kJ/mol}, \sigma = 0.303 \text{ nm}$
O(H)	-1.6	$\varepsilon = 0.490 \text{ kJ/mol}, \sigma = 0.309 \text{ nm}$
Н	+0.6	$\varepsilon = 0.192 \text{ kJ/mol}, \sigma = 0.140 \text{ nm}$
Bond type		Bond parameters <sup>2, 4</sup>
P-O stretch		$r_0 = 0.157 \text{ nm}, k_b = 359824 \text{ kJ/mol/nm}^2$
O-P-O bend		$\theta_0 = 109.5^\circ, k_\theta = 1046 \text{ kJ/mol/rad}^2$
O-H stretch		$r_0 = 0.096 \text{ nm}, k_b = 438767 \text{ kJ/mol/nm}^2$

Supplementary Table S1. Force field parameters of HAP.

In Table S1,  $\varepsilon$  is the depth of the potential well of Lennard-Jones potential, and  $\sigma$  is the distance at which the van der Waals potential is zero.

Lennard-Jones potential:  $V_{LJ} = 4\varepsilon \left[ \left( \frac{\sigma}{r} \right)^{12} - \left( \frac{\sigma}{r} \right)^{6} \right]$ 

 $r_0$  is the equilibrium bond length for covalent bond stretching, and  $k_b$  is the force constant.

 $\theta_0$  is the equilibrium bond angle for covalent bond angle bending, and  $k_0$  is the force constant.

Refer to GROMACS<sup>5</sup> and its manual for the formulations of the potential functions.



**Supplementary Figure S1.** Discrete HAP ion systems for building collagen microfibril-HAP composite. (a) Equivalent to 12 unit cells of HAP. (b) Equivalent to 96 unit cells of HAP, in the final model of collagen microfibril-HAP composite.



**Supplementary Figure S2.** Flow chart of simulation stages, with atomic configuration at the end of each stage.



**Supplementary Figure S3.** Comparison of number of hydrogen bonds within collagen and between collagen and water, for two systems at the same simulation time (one un-stretched configuration during 10 ns extra equilibration, and one configuration stretched for 3.5 ns to 10% followed by 6.5 ns extra equilibration).



*Supplementary Figure S4. Histograms of the HAP cluster size distribution of the two configurations (after 10 ns extra NPT vs after 3.5 ns stretching to 10% plus 6.5 ns extra NPT). The abscissa is the cluster size intervals, and the ordinate is the number of clusters in each interval.* 



**Supplementary Figure S5.** Distribution of actual tensile strain of backbone C-C bonds, at nominal engineering strain of 10% in collagen microfibril. The majority (over 80%) of the C-C bonds experienced a strain less than 4%, and only 3 out of 3013 C-C bonds had a strain larger than 10%.

Supplementary Movies. 2 movies are available in the SI. These movies show the following:

**Movie S1.** Uniaxial tension of a D-period collagen microfibril and HAP cluster formation, followed by extra equilibration. For collagen microfibril, the final strain is 10% at 3.5 ns, and the pulling rate is 0.002 nm/ps. During 6.5 ns extra equilibration, collagen microfibril is kept at 10% strain by only fixing the N- and C- termini. Collagen chains are in cyan. In HAP minerals, Ca is in Green, P in Gold, O in Red, and H in White.

**Movie S2.** 10 ns extra equilibration of a D-period collagen microfibril and HAP model, without mechanical stretching. Collagen chains are in cyan. In HAP minerals, Ca is in Green, P in Gold, O in Red, and H in White.

## **Supplementary References**

References

- 1. S. Hauptmann, H. Dufner, J. Brickmann, S. M. Kast and R. S. Berry, *Physical Chemistry Chemical Physics*, 2003, **5**, 635-639.
- 2. R. Bhowmik, K. S. Katti and D. Katti, *Polymer*, 2007, **48**, 664-674.
- 3. R. B. Best, X. Zhu, J. Shim, P. E. M. Lopes, J. Mittal, M. Feig and A. D. MacKerell, *Journal of Chemical Theory and Computation*, 2012, **8**, 3257-3273.
- 4. T.-J. Lin, *PhD dissertation, Department of Polymer Engineering, The University of Akron,* 2013.
- 5. H. J. C. Berendsen, D. van der Spoel and R. van Drunen, *Computer Physics Communications*, 1995, **91**, 43-56.