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Supplementary Information - "Thermodynamic Stability and Control of Oxygen Reactivity at Functional Oxide Interfaces: EuO on ITO"

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1 Eu Distillation Condition

Fig. 1 shows HAXPES data of a sample grown under the Eu distillation condition i. e. 4 nm EuO/ITO with the EuO layer grown at $T_S = 400$ °C with $J_{\rm Eu}/J_{\rm O} = 2$. The sample is largely over-oxidized, similar to sample type (i) discussed in the manuscript.

2 Chemical Stability of the YSZ Buffer Layer

In order to investigate the chemical stability of the YSZ buffer layer in sample type (ii), we recorded HAXPES data of the Zr 3d core level. The spectra are depicted in Fig. 2. For all photon energies, we find that the two components of the Zr 3d doublet are located at 182.9 eV and 185.3 eV, which is the expected binding energy for YSZ¹. No other peaks are observed, especially none at 178.8 eV which would be characteristic for metallic Zr⁰. We conclude that the buffer layer is homogeneous and stable against reduction by the EuO layer.

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3 Fitting Procedures

Exemplary fits of Eu 3*d* and In 3*d* spectra are shown in Fig. 3 and Fig. 4. Both core levels show spin-orbit splitting into $3d_{5/2}$ and

 $3d_{3/2}$ components. The Eu 3d spectrum in Fig. 3 is fitted with components for divalent Eu²⁺ (orange, the signature of stoichiometric EuO) and trivalent Eu³⁺ (brown, the signature of over-oxidized phases). Further, plasmonic excitations in the Al capping layer cause satellite peaks (gray) which are shifted by 16 eV with respect to the peak they originate from. The Eu²⁺ component is fitted with a complex multiplet structure². Details of the fitting procedure are describe in³.

The In 3*d* spectrum in Fig. 4 is also spin-orbit split into $3d_{5/2}$ and $3d_{3/2}$. It is fitted with the following components: one located at 444.9 eV, which denotes trivalent In³⁺ (as expected for Sn:In₂O₃)⁴, a second component located at 445.7 eV which can be attributed to screening effects caused by the Sn dopants⁵, and a third component located at 443.9 eV which is characteristic for metallic In⁰.

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Fig. 1 HAXPES spectra of the Eu $3d_{5/2}$ core level of a sample grown under the Eu distillation condition. Spectra normalized to the Eu³⁺ component.



Fig. 3 HAXPES spectrum of the Eu 3d core level of sample type (iii). Spectrum recorded with a photon energy of 5000 eV. Eu²⁺ components are shown in orange, Eu³⁺ in brown, and plasmonic excitation in gray. Black crosses represent the experimental data, gray line is the superposition of all fit components.



Fig. 2 HAXPES spectra of the Zr 3d core level of sample type (ii) discussed in the manuscript. Spectra normalized to the Zr $3d_{5/2}$ components, except for the 2010 eV spectrum which could not be normalized because the intensity is to weak.



Fig. 4 HAXPES spectrum of the ln 3d core level of sample type (iii). Spectrum recorded with a photon energy of 2010 eV. In³⁺ components are shown in dark blue, In⁰ in light blue. Black crosses represent the experimental data, gray line is the superposition of all fit components.