

Supporting Information

A novel double-perovskite $\text{Gd}_2\text{ZnTiO}_6$: Mn^{4+} red phosphor for UV-based w-LEDs: structure and luminescent properties

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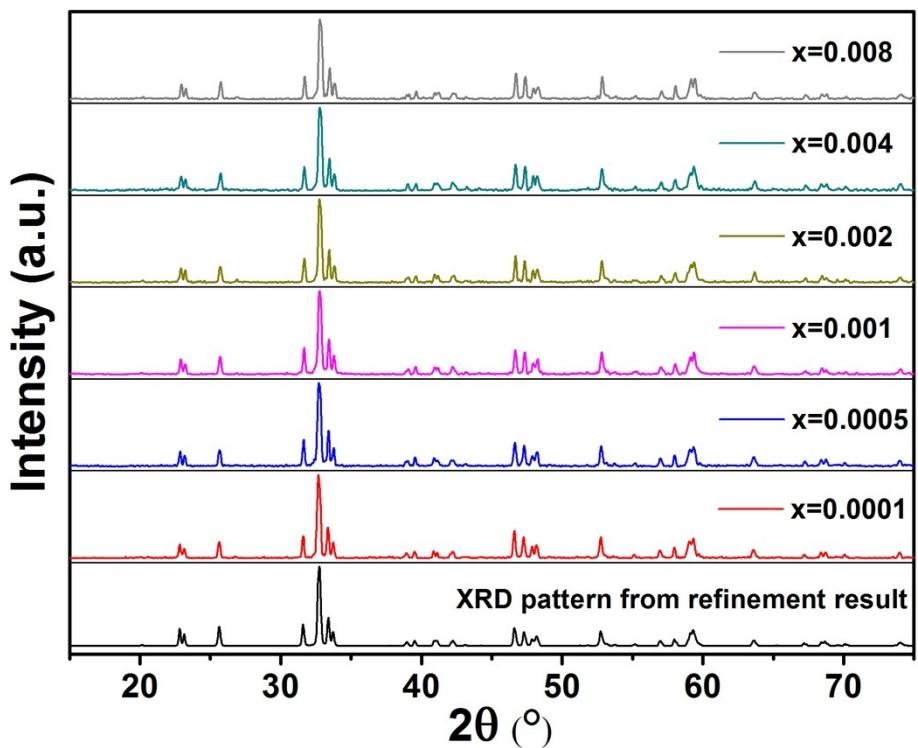


Figure S1. XRD patterns of GZT: $x\text{Mn}^{4+}$ samples and the refinement result for GZT: 0.002Mn $^{4+}$.

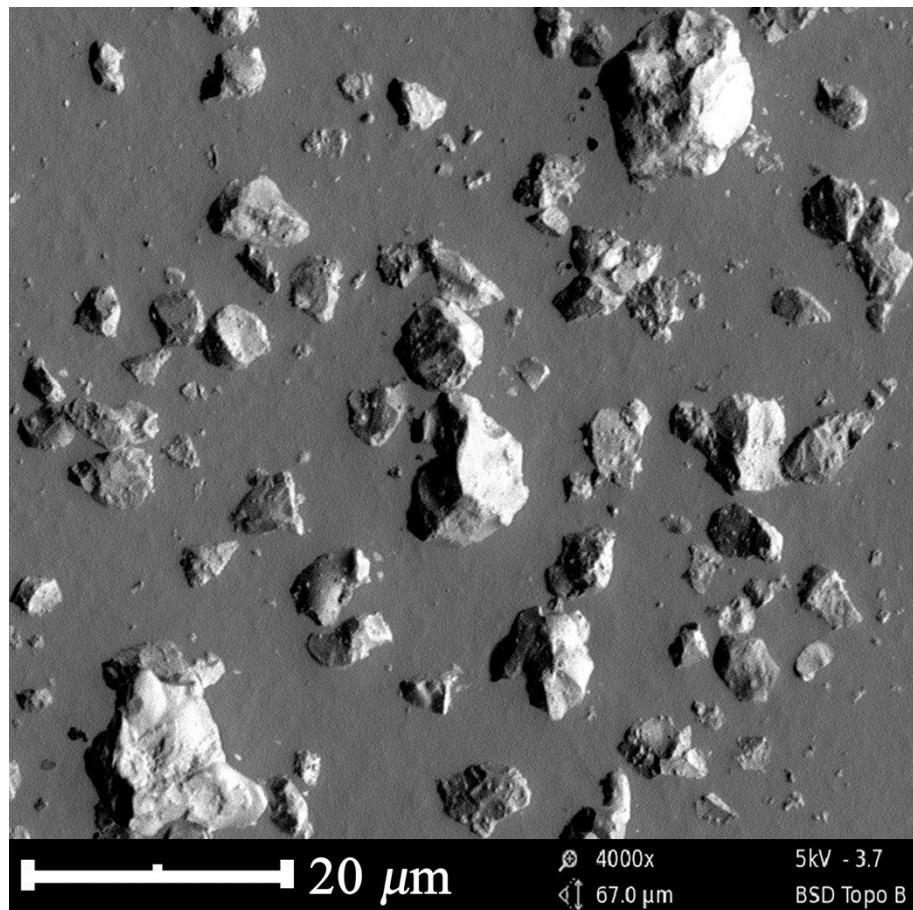


Figure S2. SEM image of GZT: 0.002Mn $^{4+}$ phosphor sample.

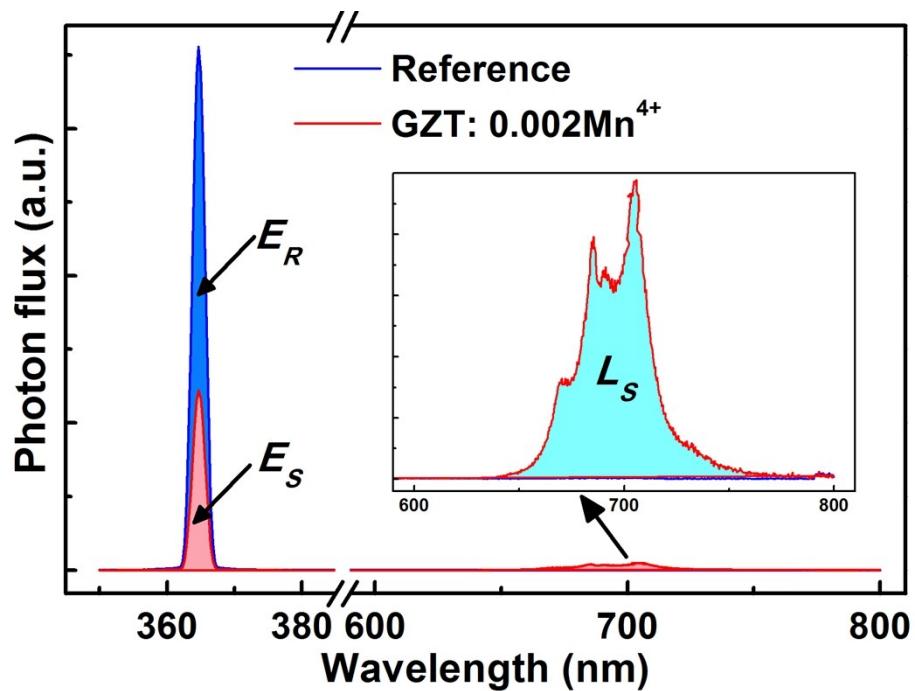


Figure S3. PL spectra of GZT:0.002Mn⁴⁺ and reference sample recorded by a spectrofluorometer equipped with an integrating sphere for quantum efficiency and absorption efficiency calculations; the inset shows the magnified spectral region of 600–800 nm.

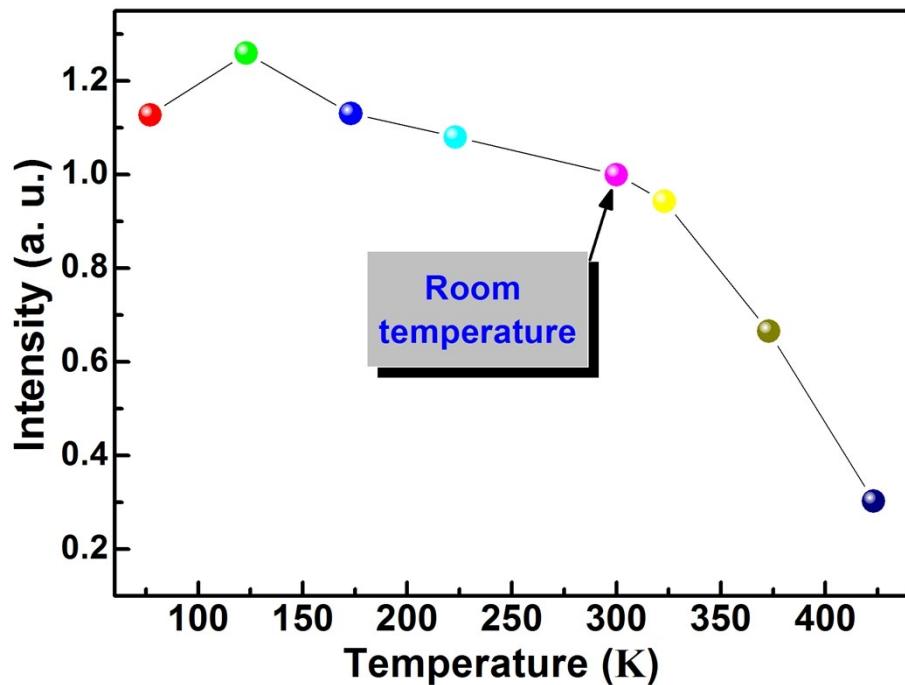


Figure S4. Integrated red PL intensity of GZT:0.002Mn⁴⁺ as a function of temperature

Table S1. The crystal field splitting (D_q), Racah parameters (B , C), Nephelauxetic ratio (β) and ^2E energy level ($E(^2\text{E})$) of Mn^{4+} ion in various hosts

Host	D_q (cm $^{-1}$)	B , (cm $^{-1}$)	C , (cm $^{-1}$)	β	$E(^2\text{E})$, (cm $^{-1}$)	Reference
Na_2SiF_6	2174	775	3475	1.051	16210	1
Na_2SnF_6	2101	589	3873	1.033	16171	2
K_2MnF_6	2183	604	3821	1.029	16129	3
K_2TiF_6	2137	582	3778	1.011	15835	4
Cs_2GeF_6	2063	490	4056	1.033	16032	5
BaSiF_6	2141	568	3879	1.026	16050	6
MMG*	2380	700	3416	0.997	15576	7
$\text{Y}_2\text{Sn}_2\text{O}_7$	2100	700	3515	1.016	15563	8
$\text{CaAl}_{12}\text{O}_{19}$	2132	807	3088	0.999	15244	9-10
$\text{SrMgAl}_{10}\text{O}_{17}$	2237	791	3084	0.989	15152	11
$\text{Sr}_4\text{Al}_{14}\text{O}_{25}$	2222	680	3397	0.983	15361	12-14
$\text{Ba}_2\text{LaNbO}_6$	1780	670	3290	0.958	14679	15
$\text{Y}_3\text{Ga}_5\text{O}_{12}$	1922	699	3197	0.957	14859	16
$\text{Mg}_{14}\text{Ge}_5\text{O}_{24}$	2375	709	3263	0.974	15175	16
CaZrO_3	1850	754	3173	0.983	15054	17
Mg_2TiO_4	2096	700	3348	0.985	15267	18
$\text{Ca}_{14}\text{Zn}_6\text{Al}_{10}\text{O}_{35}$	2165	654	3054	0.906	14451	19-20
LaAlO_3	2123	695	2941	0.907	14034	21
BaTiO_3	1780	738	2820	0.913	13862	22
SrTiO_3	1818	719	2839	0.905	13827	23
$\text{Gd}_2\text{ZnTiO}_6$	1980	639	3132	0.913	14224	this work

MMG*= $3.5\text{MgO}\cdot 0.5\text{MgF}_2\cdot \text{GeO}_2$

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