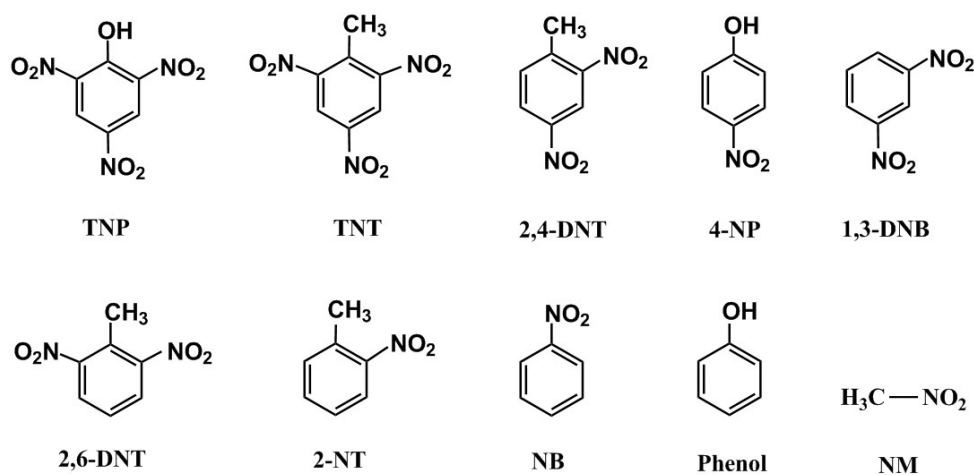


Electronic supplementary information for

Temperature responsive polymer brushes grafted from graphene oxide: an efficient fluorescent sensing platform for 2,4,6-trinitrophenol

Yajiao Song, Jianhua Lü, Bingxin Liu and Changli Lü



Scheme S1 Structures of different nitro-compounds used in the experiments.

Calculation of fluorescence quenching efficiency

The fluorescence quenching efficiency (η) for each analyte was calculated by the following equation:

$$\eta = (I_0 - I) / I_0 \times 100 \%$$

in which I_0 and I are the fluorescence intensities in the absence and presence of analyte, respectively.

Table S1 A comparative study of the K_{sv} , detection limit and medium used for TNP detection of some recent representative reports.

| Publication | Material used | $K_{sv}(M^{-1})$ | Detection limit | Medium Used |
|---|--|-------------------|-----------------|---------------------------------|
| Present work | Alq ₃ -containing Block copolymer brush-GO hybrid | 5.5×10^8 | 2.38 nM | H ₂ O |
| <i>Anal. Chem.</i> , 2012 , 84, 8415 | Alq ₃ nanospheres | 1.4×10^7 | 32.3 nM | Phosphate buffer solution |
| <i>Chem. Eur. J.</i> , 2016 , 22, 2012 | Vinylpyridine Appended anthracene derivatives | 4.3×10^4 | 500 ppb | THF |
| <i>Sensors and Actuators B</i> , 2016 , 230, 746 | Small fluorescent molecule | 7.0×10^4 | 0.47 μ M | THF |
| <i>Cryst. Growth Des.</i> , 2016 , 16, 842 | Metal-organic framework | 4.0×10^4 | 0.18 μ M | EtOH |
| <i>Chem. Eur. J.</i> , 2016 , 22, 4931 | Covalent Triazine Framework | 8.0×10^5 | – | H ₂ O/EtOH (v/v=4:1) |
| <i>Chem. Commun.</i> , 2015 , 51, 7201 | Conjugated polyelectrolyte | 1.0×10^7 | 128 ppb | H ₂ O |
| <i>ACS Appl. Mater. Interfaces</i> , 2014 , 6, 10722 | Graphene oxide | 1.3×10^5 | 125 ppb | Buffer |
| <i>Chem. Mater.</i> , 2014 , 26, 4221 | Graphene derivative | 8.9×10^5 | 300 ppb | H ₂ O/THF (v/v=9:1) |
| <i>Chem. Commun.</i> , 2014 , 50, 15788 | Organic cage | 2.2×10^5 | 6.4 ppb | DCM |
| <i>Chem. Eur. J.</i> , 2014 , 20, 12215 | α -Cyanostilbene derivative | 3.3×10^5 | 0.28 μ M | H ₂ O/THF (v/v=7:3) |

Calculation of Detection Limit

The detection limit plot for TNP was obtained by plotting change in the fluorescence intensity vs the concentration of TNP. The curve demonstrated a linear relationship and the correlation coefficient (R^2) via linear regression analysis was calculated to be 0.99. The limit of detection (LOD) was then calculated using the equation $3\sigma/K$, where σ denotes the standard deviation for the intensity of sensing system in the absence of TNP and was calculated for the peak intensity value using ‘Statistics on Columns’ option in origin software and verified with online calculator. K represents slope of the equation.

$$\begin{aligned}\text{LOD} &= 3\sigma/K \\ &= 3 \times 2427.1 / (3.05 \times 10^{13}) \\ &= 2.38 \times 10^{-9} \text{M}\end{aligned}$$

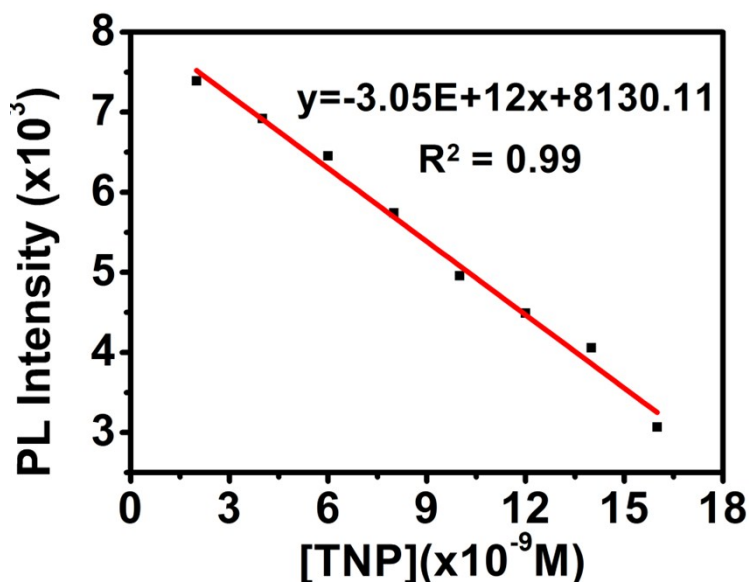


Fig.S1 Relation of PL intensity against the concentration of TNP and linear fit for estimation of detection limit.

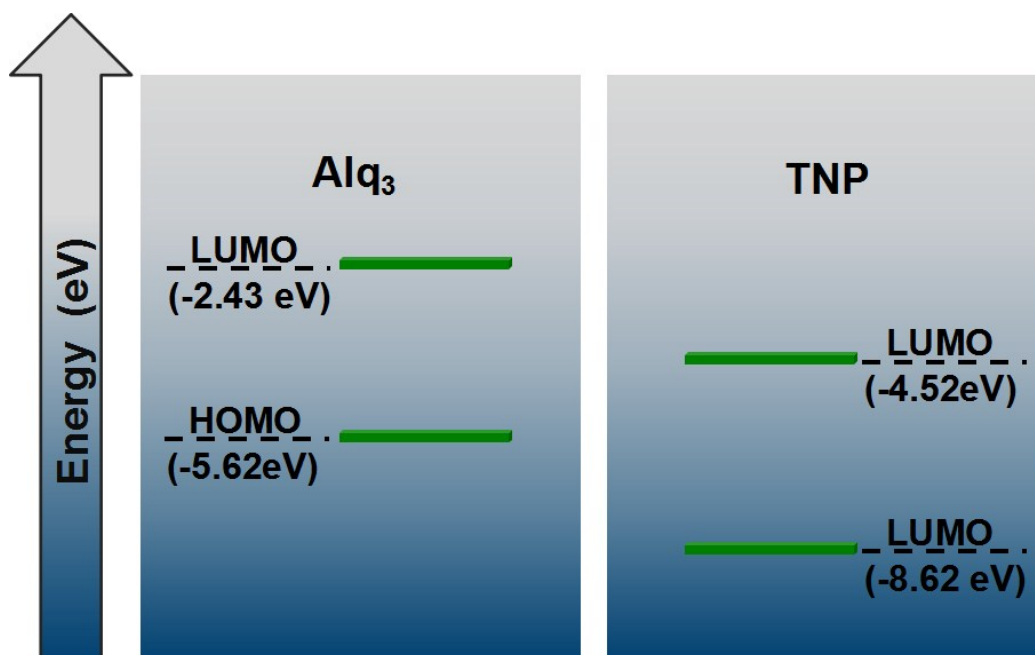


Fig.S2 Calculated energy diagram of Alq₃ and TNP (the calculated energy data come from the reported literatures 43, 48 and 50).

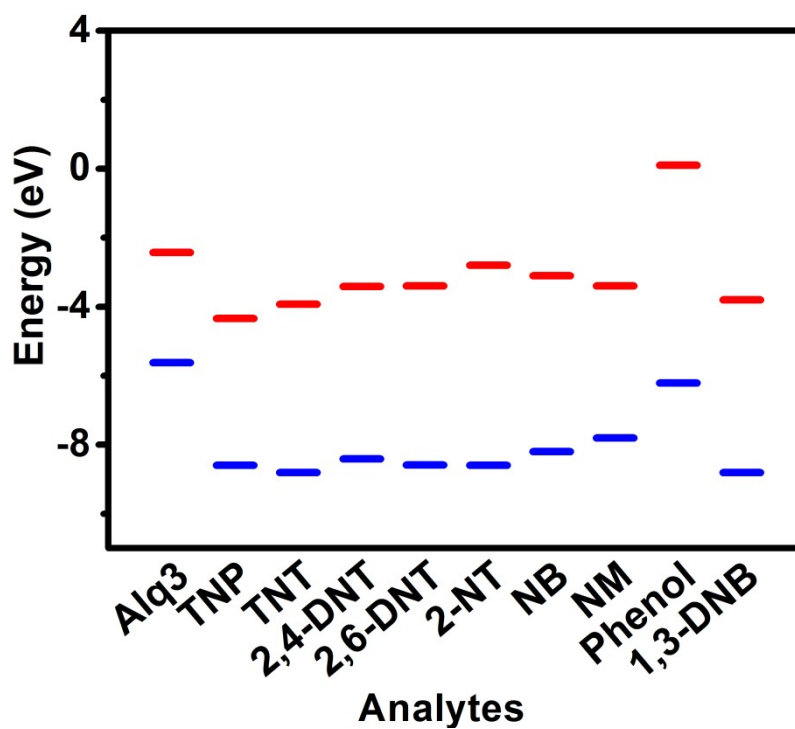


Fig.S3 HOMO and LUMO energies for Alq₃ and selected analytes (the calculated energy data come from the reported literatures 48, 50).