

Supporting Information

Influence of Heteroatoms on the Charge Mobility of Anthracene Derivatives

Lijia Yan, Yang Zhao, Hongtao Yu, Zhao Hu, Yaowu He, Osamu Goto, Chaoyi Yan, Ting Chen, Runfeng Chen, Yueh-Lin Loo, Dmitrii F. Perepichka, Hong Meng* and Wei Huang*

Surface modification of SiO₂ substrates: OTS device substrates we used highly doped n-type Si wafers with a 250 nm dry thermal oxide gate dielectric. These wafers were cleaned in piranha solution (highly corrosive and oxidizing 7:3 mixture of H₂SO₄ and H₂O₂) for 30 min, rinsed with deionized water and dried under a nitrogen stream. For OTS treatment, the wafer was treated with OTS by immersing the cleaned wafer substrate in 0.1 M solution of OTS in toluene at 60 °C for 30 min, and rinsing with toluene and blown dry with nitrogen. The semiconductor layer was deposited over the treated surface through shadow mask.

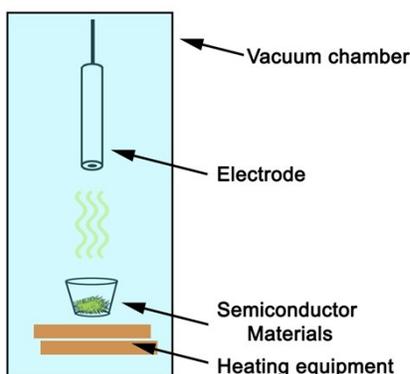


Figure S1. The simplified graph of the evaporating methods for the GCE ($\Phi=3\text{mm}$).

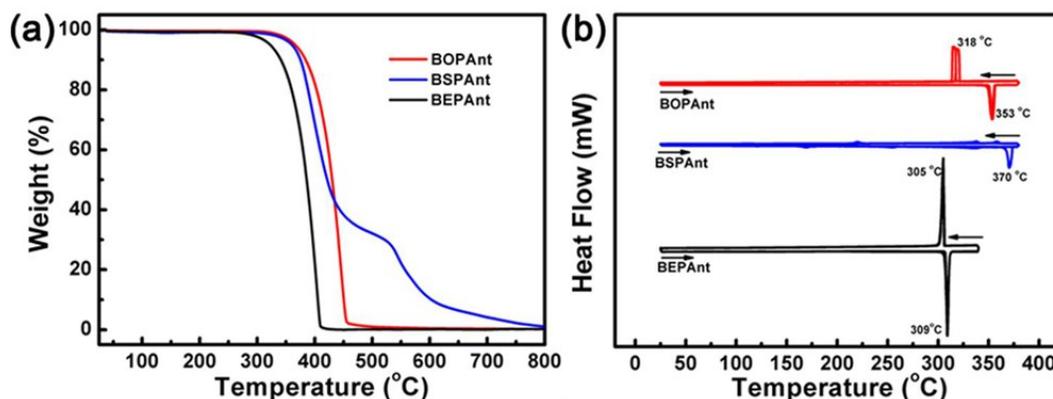
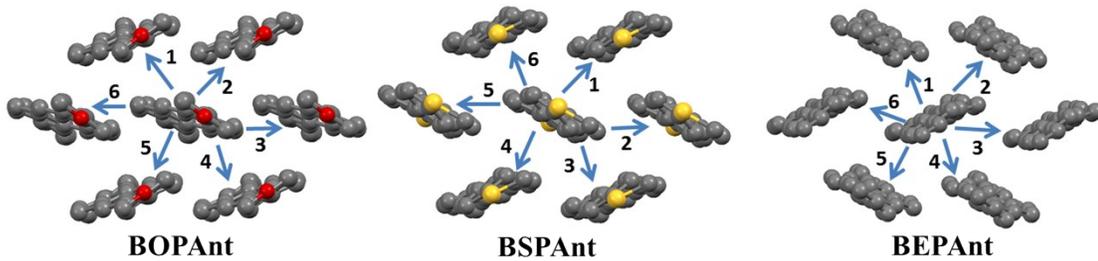


Figure S2. TGA (a) and DSC (b) plots of BOPAnt, BSPAnt and BEPant (scan rate: 10 °C/min).

Table S1. Single Crystal Data of Three Semiconductors

	BOPAnt		BSPAnt		BEPAnt	
Crystal system	Orthorhombic		Triclinic		Monoclinic	
Space group	P b c a		P-1		P 21/c	
Unit cell dimensions	a = 7.44 Å	$\alpha = 90^\circ$	a = 6.03 Å	$\alpha = 92.83^\circ$	a = 22.66 Å	$\alpha = 90^\circ$
	b = 6.16 Å	$\beta = 90^\circ$	b = 7.68 Å	$\beta = 96.55^\circ$	b = 7.72 Å	$\beta = 92.45^\circ$
	c = 42.30 Å	$\gamma = 90^\circ$	c = 22.61 Å	$\gamma = 90.10^\circ$	c = 6.04 Å	$\gamma = 90^\circ$
Z	4		2		2	
Volume	1938.71(15) Å ³		1040.09(11) Å ³		1056.00(7) Å ³	
Packing D	2.06 / nm ³		1.92 / nm ³		1.89 / nm ³	

**Table S2. the hole and electron reorganization energy levels and the respected hole and electron couplings data**

Semi-conductor	pathway	Distance (Å)	Hole coupling (eV)	Electron coupling (eV)	Hole reorganization (eV)	Electron reorganization (eV)	Hole drift mobility (cm ² /Vs)	Electron drift mobility (cm ² /Vs)
BOPAnt	1	4.83	0.07253	0.1692	0.2491	0.2246	0.2635	1.7141
	2	4.83	0.07252	0.1693				
	3	4.83	0.07257	0.1691				
	4	4.83	0.07254	0.1691				
	5	21.37	0.03190	0.0001				
	6	6.16	0.01334	0.0194				
BSPAnt	1	4.881	0.02144	0.03371	0.1829	0.4656	0.1075	0.0074
	2	4.889	0.01202	0.05132				
	3	4.89	0.02181	0.03693				
	4	23.501	0.00110	0.0004				
	5	22.722	0.01265	0.00253				
	6	23.501	0.00176	0.00081				

	1	6.043	0.02328	0.02772				
	2	6.043	0.02328	0.02772				
BEPAnt	3	4.901	0.07939	0.13906	0.1790	0.2190	0.1732 // [a]	0.0699 // [a]
	4	4.901	0.07939	0.13906			0.6583 \perp [b]	1.2594 \perp [b]
	5	4.901	0.07939	0.13906				
	6	4.901	0.07939	0.13906				

[a] Based on BEPAnt crystal parallel aligned ethyl group (Figure S3a); [b]: Based on BEPAnt crystal vertically aligned ethyl group (Figure S3b).

(a)



(b)

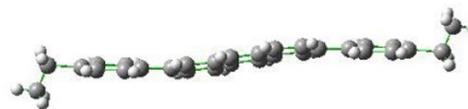


Figure S3. Two possible single crystal structure of BEPAnt: (a) Ethyl parallel to the plane; (b) Ethyl perpendicular to plane.