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Supporting Information

New generation of highly sensitive luminescent thermometers operating in

optical window of biological tissues

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Fig. S1. X-ray diffraction patterns of LiLaP₄O₁₂:Cr,Nd nanocrystals-a; and representative TEM image –b with grain size distribution-c.



Fig. S2. Comparison of absorption and emission spectra of LiLaP₄O₁₂:1%Cr nanocrystals



Fig. S 3. Comparison of absorption and emission spectra of LiLaP₄O₁₂:Cr,Nd nanocrystals with absorption of biological molecules Mel-melanine, NADH – Nicotinamide adenine dinucleotide, coll – collagen, PF – protophorphirin, O2Hb – oxygenated hemoglobin, Hb- deoxygenated hemoglobin, Bil-Billirubine, H2O - water



Fig. S4. Emission spectra of LiLaP₄O₁₂:Cr nanocrystals with different concentration of Cr^{3+} ions –a and integral emission intensity as a function of Cr^{3+} concentration-b

The crossing point between ground state parabola and ${}^{4}T_{2}$ parabola of Cr^{3+} ion in LiLaP₄O₁₂ nanocrystals (activation energy) was determined basing on temperature dependence of emission intensity of LiLaP₄O₁₂:Cr nanocrystals for different concentration of Cr^{3+} ions (Fig. S4) using following equation:

$$I_{em} = \frac{I_0}{1 + \exp\left(\frac{\Delta E}{kT}\right)}$$
(S.1)

where

Iem- emission intensity

I₀- emission intensity at the lowest temperature

 ΔE - activation energy

T-temperature

k- Boltzmann constant



Fig. S5. Temperature dependence of emission intensity of LiLaP₄O₁₂:Cr nanocrystals

Reproducibility of temperature measurement using LiLaP₄O₁₂:1%Cr³⁺,10%Nd³⁺ nanocrystals as a luminescent thermometer was confirmed via 6 cooling-heating cycles measurements at two temperatures 0°C and 50°C. At each temperature emission spectra was recorded after 3 minutes.



Fig. S6 Reproducibility of LIR measurements for $LiLaP_4O_{12}$:1% Cr^{3+} ,10% Nd^{3+} nanocrystals in cooling-heating cycles.