

## Supporting Information

# Large-scale Synthesize and Screen Printing of Upconversion Hexagonal-phase NaYF<sub>4</sub>: Yb<sup>3+</sup>, Tm<sup>3+</sup>/Er<sup>3+</sup>/Eu<sup>3+</sup> Plates for Security Application

Weijing Yao<sup>1, #</sup>, Qingyong Tian<sup>1, 2#</sup>, Jun Liu<sup>1</sup>, Zhaohui Wu<sup>1</sup>, Shuyuan Cui<sup>1</sup>, Jing Ding<sup>1</sup>, Zhigao Dai<sup>1</sup>,

Wei Wu<sup>1,\*</sup>

<sup>1</sup> Laboratory of Printable Functional Nanomaterials and Printed Electronics, School of Printing and Packaging, Wuhan University, Wuhan 430072, P. R. China

<sup>2</sup> Key Laboratory of Artificial Micro- and Nano-structures of Ministry of Education, School of Physics and Technology, Wuhan University, Wuhan 430072, P. R. China

# These authors contributed equally.

---

\*To whom correspondence should be addressed. Tel: +86-27-68778529. Fax: +86-27-68778433. E-mail:

[weiwu@whu.edu.cn](mailto:weiwu@whu.edu.cn) (W. Wu)

**Table S1** The reaction parameters of the NaYF<sub>4</sub>: 18% Yb<sup>3+</sup>, 2% Tm<sup>3+</sup> (UCMPs) and corresponding morphology and average dimension.

Sample	Process time /h	TSC /mmol	NH <sub>4</sub> F /mmol	NaCl /mmol	OA:EG /ml	Morphology	average dimension (L/T)/ μm	Aspect ratios <sup>a</sup>
S1	6	2	6	2.88	10:5	hexagonal microplates	1.127/0.883	1.27
S2	0.5	2	6	2.88	10:5	nanoparticles	/	/
S3	1	2	6	2.88	10:5	nanoparticles + hexagonal microplates	/	/
S4	2	2	6	2.88	10:5	hexagonal microplates	0.660/0.245	2.69
S5	12	2	6	2.88	10:5	hexagonal microplates	1.125/0.888	1.26
S6	6	1	6	2.88	10:5	irregular microprisms	11.599/4.067	2.85
S7	6	4	6	2.88	10:5	hexagonal microplates	0.774/0.200	3.87
S8	6	2	4	2.88	10:5	nanoparticles	/	/
S9	6	2	12	2.88	10:5	hexagonal microprisms	1.437/1.670	0.86
S10	6	2	6	0	10:5	hexagonal microplates	1.331/1.079	1.23
S11	6	2	6	1	10:5	hexagonal microplates	1.198/0.917	1.31
S12	6	2	6	2.88	7.5:7.5	hexagonal microplates	1.200/0.656	1.83
S13	6	2	6	2.88	5:10	hexagonal microplates	0.921/0.482	1.91

<sup>a</sup> determined that the aspect ratio is equal to the average length divided by the average thickness.

### Ink viscosity and tension

Firstly, the kinematic viscosity of the ink was measured using a 0.4-0.5 mm diameter Capillary Tube Viscometer. We can get the kinematic viscosity through the following formula (eg. (1)). Then, viscosity of UCMPs inks is calculated *via* the conversion formula (eg. (2)). Here, the test temperature of the experiments is 20 °C.

$$\nu = c * t \quad \text{eg.(1)}$$

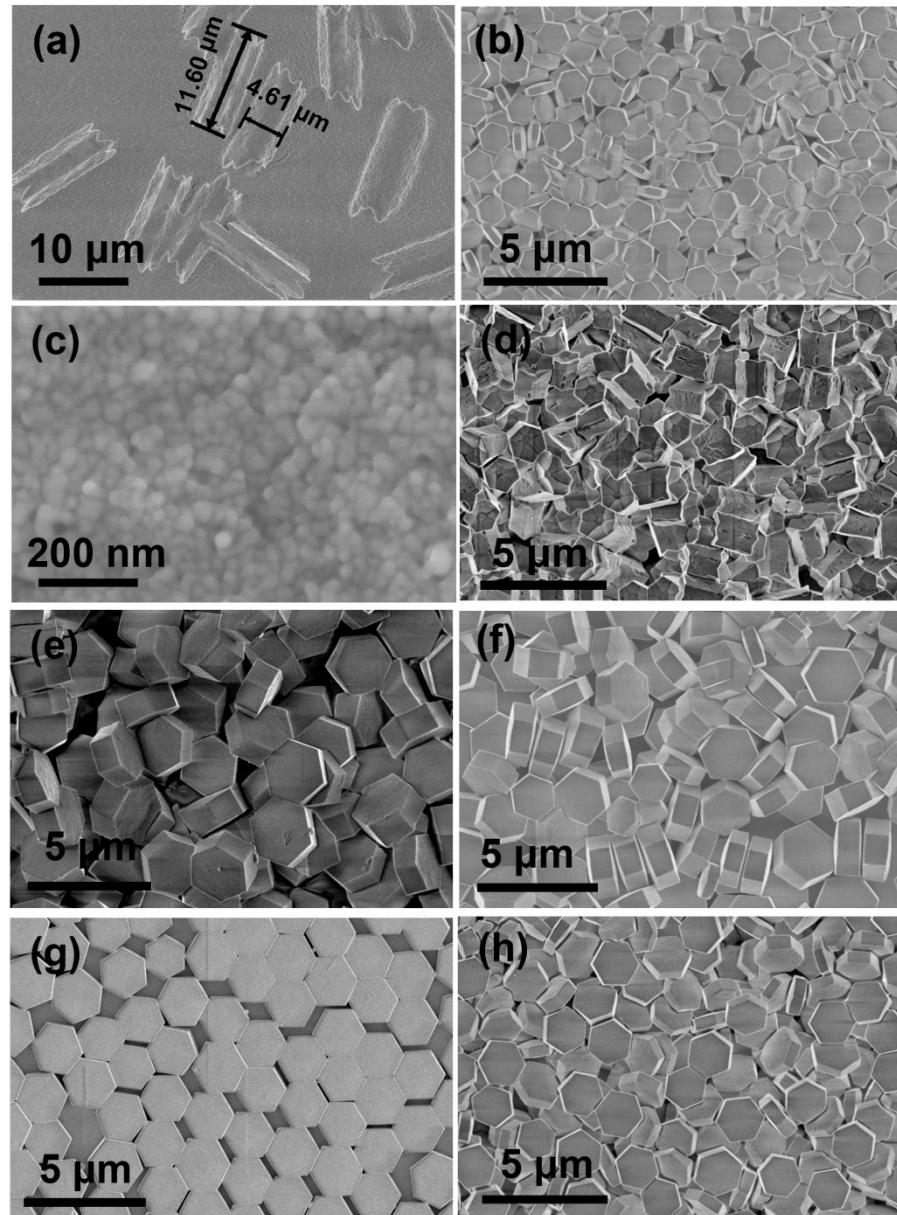
$$\eta = \nu * \rho \quad \text{eg.(2)}$$

where  $\nu$  is the kinematic viscosity of UCMPs inks (mm<sup>2</sup>/s),  $c$  is the viscometer constant (0.004471 mm<sup>2</sup>/s<sup>2</sup>),  $t$  is the time of certain volume liquid through capillary (s),  $\eta$  and  $\rho$  are the viscosity (mPa•s) and density of UCMPs inks (1.13 g/cm<sup>3</sup>), respectively. Then, the experimental parameters of viscosity test are presented in **Table R1**. The calculated viscosity of UCMPs inks is 113.11 mPa•s. The surface tension of PET is 47 mN/m, and the measured surface tension of UCMPs inks is 32

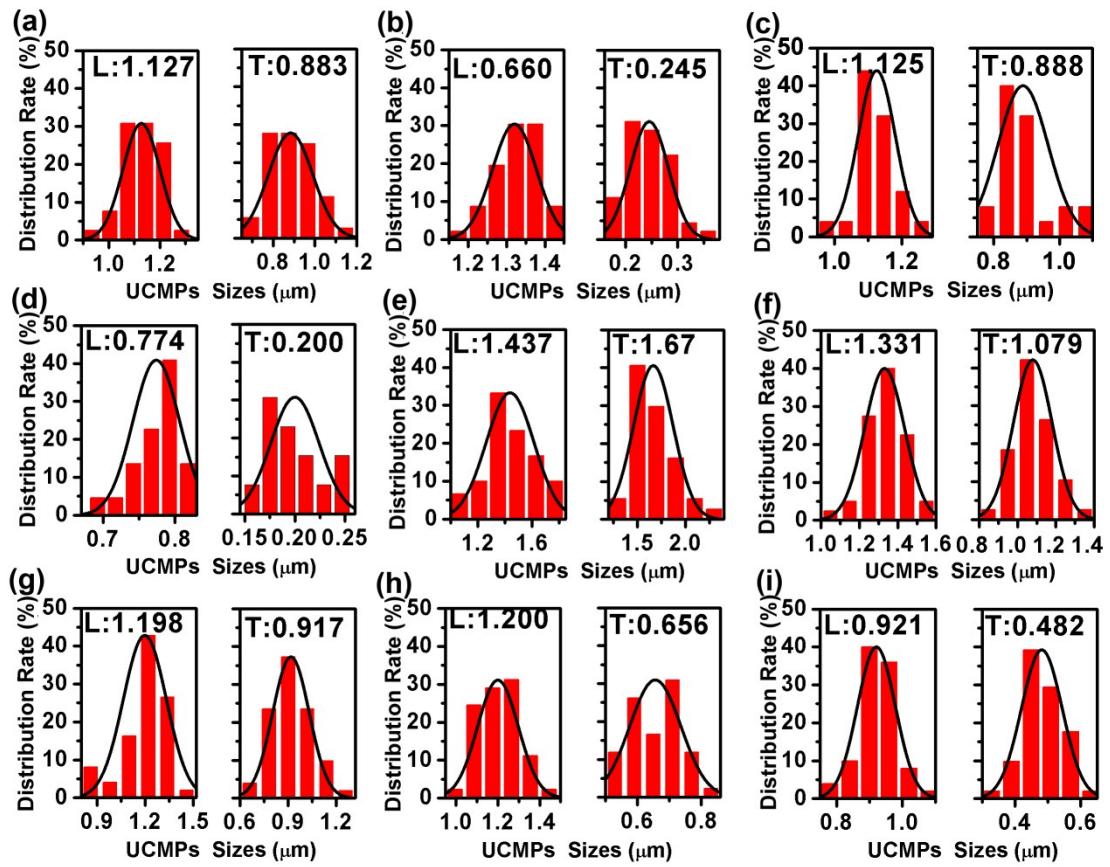
mN/m. The surface tension of substrate is higher than the ink, which is beneficial to the printing.

**Table S2** The experimental parameters of viscosity test.

t (s)	c (mm <sup>2</sup> /s <sup>2</sup> )	v (mm <sup>2</sup> /s)	ρ (g/cm <sup>3</sup> )	η (mPa•s)
22388.00	0.004471	100.10	1.13	113.11

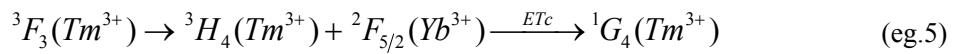
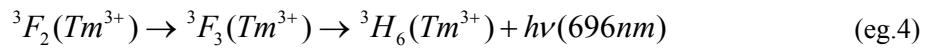
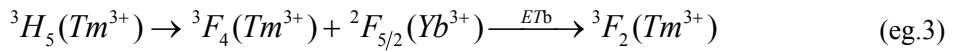
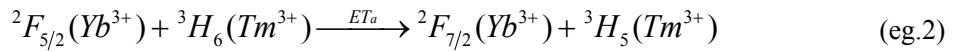
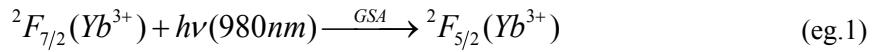


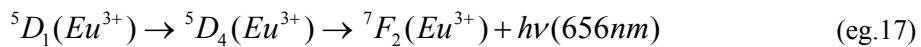
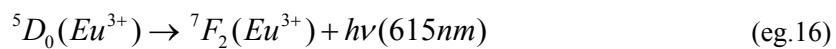
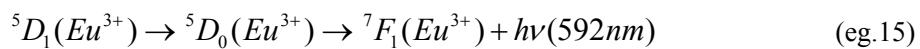
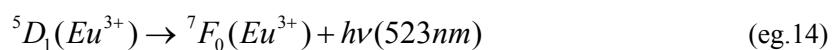
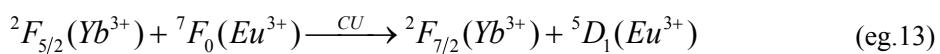
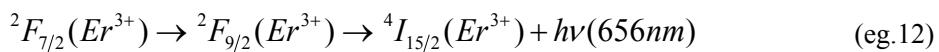
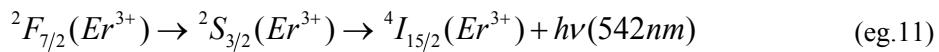
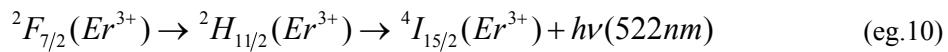
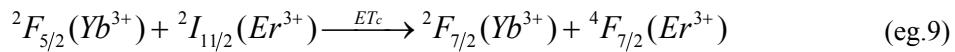
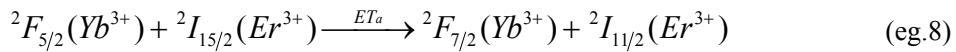
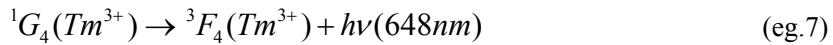
**Figure S1** SEM images of NaYF<sub>4</sub>: Yb<sup>3+</sup>, Tm<sup>3+</sup> UCMPs under different reaction conditions. (a) 1 mmol and (b) 4 mmol TSC, (c) 4 mmol and (d) 12 mmol NH<sub>4</sub>F, (e) 0 mmol and (f) 1 mmol NaCl, (g) 7.5: 7.5 mL and (h) 5: 10 mL OA: EG, respectively.



**Figure S2** Sizes distribution statistics of  $\text{NaYF}_4$ :  $\text{Yb}^{3+}$ ,  $\text{Tm}^{3+}$  UCMPs under different reaction conditions.(a) 6 h, (b) 2 h, (c) 12 h; (d) 4 mmol TSC; (e) 12 mmol  $\text{NH}_4\text{F}$ ; (f) and (g) 0 mmol, 1 mmol  $\text{NaCl}$ ; (h) and (i) 7.5: 7.5 mL, 5: 10 mL OA: EG, respectively.

The energy transition formulas of  $\text{Tm}^{3+}$ ,  $\text{Er}^{3+}$ ,  $\text{Eu}^{3+}$ and  $\text{Yb}^{3+}$  dopant ions.





**Figure S3 The image of screen printing plate.**