Electronic Supplementary Information for:

Dual phosphorescent dinuclear transition metal complexes and their application as triplet photosensitizers for TTA upconversion and photodynamic therapy

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1.0 Molecular structure characterization data

Ru-2:



Figure S1. ¹H NMR spectra of Ru-2 (400 MHz, CD₃CN), 20 °C.



Figure S2. ¹³C NMR spectra of Ru-2 (150 MHz, CD₃CN), 20 °C.



Figure S5. ¹H NMR spectra of Ir-2 (400 MHz, CD₃CN), 20 °C.



2.0 Stern–Volmer plots of triplet excited state lifetime quenching



Figure S9. Stern–Volmer plots generated from triplet excited state lifetime (τ_T) quenching of complexes **Ru-2**, **Ir-2** and Ru(bpy)₃ measured with the increasing DPA concentration in CH₃CN. The lifetimes were measured with the nanosecond time-resolved transient absorption. c [Sensitizers] = 1.0×10^{-5} M, 20 °C.

3.0 Delayed fluorescence



Figure S10. The delayed fluorescence decay of DPA observed in the TTA upconversion with **Ru-2** and **Ir-2** complexes as triplet photosensitizer and DPA as the triplet acceptor, the complexes were excited at 589 nm (nanosecond pulsed OPO laser synchronized with spectrofluorometer) and the emission was due to the upconverted emission of DPA monitored at 428 nm. (a) The prompt fluorescence decay of DPA determined in a different experiment (excited with picosecond 405 nm laser, the decay of the emission was monitored at 428 nm). (b) **Ir-2** as triplet photosensitizer; (c) **Ru-2** as triplet photosensitizer. In deaerated CH₃CN, 25 °C. c [Sensitizers] = 1.0×10^{-5} M, c [DPA] = 2×10^{-3} M.

4.0 Intermolecular triplet energy transfer

Bimolecular quenching constant (k_q):

$$k_q = K_{sv} / \tau_0 \qquad (S1)$$

 K_{sv} Stern-Volmer quenching constant; τ_0 is the triplet state lifetime of the triplet energy donor.

Quenching efficiency (*f*₀):

$$F_{Q} = R_{q}/R_{0}$$
(S2)
$$k_{0} = \frac{4\pi N}{1000} (R_{f} + R_{q}) (D_{f} + D_{q})$$
(S3)

where k_0 is the diffusion-controlled bimolecular quenching rate constant; N is Avogadro's number; R is the collision radius, the sum of the molecule radii of the energy donor (R_f) and the energy quencher (R_q); D is the diffusion coefficients, the sum of the energy donor(D_f) and energy quencher (D_q). Diffusion coefficients can be obtained by the Stokes-Einstein eq. (S4).

$$D = kT/6\pi\eta R \qquad (S4)$$

where k is Boltzmann's constant, η is the solvent viscosity, R is the molecule radius. The radius of **Ir-2** or **Ru-2** is 15.6 Å and the radius of **1** is 5.7 Å. According to eq. (S6), the diffusion coefficients of donor and acceptor were 3.8×10^{-6} cm⁻² s⁻¹ and 1.0×10^{-5} cm⁻² s⁻¹.



Figure S11. Nanosecond time-resolved transient absorption spectra and decay traces of the mixture of **Ru-2** and **1**. (a) the molar ratio is 1:0.2 (**Ru-2**:**1**); (b) the decay trace at 575 nm and (c) at 500 nm; (d) triplet lifetime quenching Stern-Volmer plot of **Ru-2** with the increasing concentration of **1**, $\lambda_{ex} = 570$ nm, c[**Ru-2**] = 1.0 × 10⁻⁵ M, in deaerated CH₃CN at 25 °C.

efficiency (Φ_{ET}) for TTET in the mixture of **Ru-2** and **1**.^a **Ru-2:1** k_{ET}/s^{-1} Φ_{ET} (%) **1**.0.2 2.7×10^4

Table S1. The triplet energy transfer rate constant (k_{ET}) and triplet energy transfer

Ru-2:1	k_{ET}/s^{-1}	$\Phi_{ ext{ET}}$ (%)	
1:0.2	3.7×10^{4}	94.7	
1:0.4	11.2×10^{4}	96.6	
1:0.6	37.9×10^4	98.5	

^aIn deaerated CH₃CN at 25 °C.

5.0 TTA upconversion photos



Figure S11. Upconversion with **Ir-2**, **Ru-2** as triplet sensitizers excited with yellow laser, $\lambda_{ex} = 589$ nm, 5.0 mW. In deaerated CH₃CN, 25 °C. c [Sensitizers] = 1.0×10^{-5} M, c [DPA] = 1.25×10^{-3} M for **Ir-2**, c [DPA] = 1.67×10^{-3} M for **Ru-2**.

6.0 Computational details

Table S2. Electronic Excitation Energies (eV) and corresponding Oscillator Strengths (f),
main Configurations and CI coefficients of the Low-lying Electronic Excited States of the
complex Ru-2 calculated by TDDFT/B3LYP/GENECP/LANL2DZ, CH ₃ CN as the solvent
(PCM model) based on the optimized Ground State Geometries.

				TD-SCF/B3LYP/	GEN	
Electronic transition	Energy ^a (eV/nm)	$oldsymbol{f}^{ ext{b}}$	Composition ^c	CI ^d	Character ^e	
Singlet	$S_0 \rightarrow S_1$	2.2832/543	1.9522	H→L	0.6870	ILCT
	$S_0 \rightarrow S_{13}$	2.8414/436	0.1570	H-2→L	0.3727	MLCT
				H→L+2	0.3765	LL'CT
	$S_0 \rightarrow S_{23}$	3.0333/409	0.1767	H-3→L+1	0.2336	L'LCT
				H-3→L+3	0.2442	LL'CT
				H→L+6	0.2810	L'LCT
Triplet	$S_0 \rightarrow T_1$	1.5504/800	0.0000^{f}	H-8→L	0.2329	ILCT
				H→L	0.6212	ILCT
				H→L+2	0.1924	LLCT
	$S_0 \rightarrow T_2$	2.1295/585	0.0000^{f}	H-8→L	0.4356	ILCT
				H→L	0.2738	ILCT
				H-1→L+1	0.1729	MLCT
				H→L+2	0.2715	LLCT
	$S_0 \rightarrow T_3$	2.1570/575	0.0000 ^f	H-7→L	0.3821	MLCT
				H-1→L	0.2574	MLCT
				H→L+1	0.3444	LLCT

^a Only the selected low-lying excited states are presented. ^b Oscillator strengths. ^c Only the main configurations are presented. ^d The CI coefficients are in absolute values. ^e L stands for BODIPY localized ligand and L' stands for bipyridine localized ligand. ^f No spin-orbital coupling effect was considered, thus the *f* values are zero.

Table S3. Electronic Excitation Energies (eV) and corresponding Oscillator Strengths (*f*), main Configurations and CI coefficients of the Low-lying Electronic Excited States of the complex **Ir-2** calculated by TDDFT/B3LYP/GENECP/LANL2DZ, CH₃CN as the solvent (PCM model) based on the optimized Ground State Geometries.

		TD-SCF/B3LYP/GEN				
Electronic ⁻ transition	Energy ^a (eV/nm)	$f^{ m b}$	Composition ^c	CI ^d	Character ^e	
Singlet	$S_0 \rightarrow S_1$	2.2837/543	1.9723	H-2 –L	0.11933	MLCT
				H-L	0.67951	ILCT
	$S_0 \rightarrow S_{11}$	2.8767/431	0.2980	H-8-L	0.23264	MLCT
				H-L+2	0.58867	ILCT
	$S_0 \rightarrow S_{19}$	3.1523/393	0.2372	H-10-L	0.33818	L'LCT
				H-8-L	0.23258	MLCT
				H-5-L+1	0.24395	ILCT
Triplet	$S_0 \rightarrow T_1$	1.5467/802	$0.0000 \ ^{\rm f}$	H-8-L	0.21184	MLCT
				H-2-L	0.19874	MLCT
				H-L	0.59488	ILCT
	$S_0 \rightarrow T_2$	2.1266/583	$0.0000\mathrm{f}$	H-12-L	0.15212	L'LCT
				H-8-L	0.39479	MLCT
				H-5-L+2	0.23322	ILCT
				H-L	0.29387	ILCT
				H-L+2	0.28617	ILCT
	$S_0 \rightarrow T_3$	2.1648/573	$0.0000{\rm f}$	H-8-L	0.22045	MLCT
				H-2-L	0.50778	MLCT
				H-1-L+1	0.29611	L'LCT; MLCT

^a Only the selected low-lying excited states are presented. ^b Oscillator strengths. ^c Only the main configurations are presented. ^d The CI coefficients are in absolute values. ^e L stands for BODIPY localized ligand and L' stands for bipyridine localized ligand. ^f No spin-orbital coupling effect was considered, thus the *f* values are zero.





Figure S12. Electron density maps of the frontier molecular orbital of the complex **Ir-2**, based on ground state optimized geometry by the TDDFT calculations at the TDDFT/B3LYP/GENECP/LANL2DZ level with Gaussian 09W.



Figure S13. Electron density maps of the frontier molecular orbital of the complex **Ru-2**, based on ground state optimized geometry by the TDDFT calculations at the TDDFT/B3LYP/GENECP/LANL2DZ level with Gaussian 09W.

7.0 The x-y-z coordinates of the triplet optimized geometries of complexes

Complex Ru-2 (DFT/B3LYP/GENECP/LANL2DZ) Charge = 4 Multiplicity = 1

С	3.34019 3.39785 -0.08063
Ċ	1.30692 2.3319 -0.07522
C	1.93809 3.57467 -0.11244
C	2.37678 1.3765 -0.01983
C	2 36611 -0 02944 0 02229
C	3 5661 -0 76303 0 04608
C	3 81159 -2 17608 0 07896
C	5.01137 2.17000 0.07070
C	5 79203 -1 01602 0 04391
N	3 59013 2 07428 -0.02585
N	4 79994 -0 10291 0.02852
R	5.00822 + 4.3793 + 0.02032
D C	5.00022 1.43793 $0.02125.74345$ 1.91000 1.10120
г Г	5 67251 1 82522 1 18627
Г С	3.07331 1.03322 1.10037
	-0.1/0/2 2.08506 -0.09098
П	-0.46747 1.44683 -0.92859
Н	-0.50/4/ 1.58821 0.82453
H	-0./0819 3.0318 -0.1/844
L	7.23472 -0.63438 0.01627
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H	7.50405 -0.22487 -0.96416
Н	7.43656 0.14273 0.75858
С	2.82099 -3.29903 0.11213
Н	2.21455 -3.33009 -0.79899
Н	3.3419 -4.25455 0.20366
Н	2.1292 -3.20539 0.95439
С	1.0591 -0.75364 0.03048
С	0.40843 - 1.02711 1.24287
С	0.47154 -1.16267 -1.17277
С	-0.81225 -1.6983 1.2471
Н	0.86131 -0.71797 2.1802
С	-0.75607 -1.82639 -1.1645
Н	0.97049 -0.95729 -2.11518
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С	-0.11582 8.05633 0.87042
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Ν	-0.71531 9.25837 0.80912
C	-1.13746 9.75431 -0.34627
Ĥ	-1.34213 9.44921 -2.48339
C	-1.78608 11 12263 -0 27467
Ň	-1.88108 11.65492 0.93834

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С	-2.433 12.85657 1.12248
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Č	-1.13705 11.99379 4.43423
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Ч	-4.00047 10.40974 4.41808
Н	-4.47489 11.95966 6.29718
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C	-2 04556 7 40836 4 14966
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C C	-1 8649 6 36531 5 07386
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Н	7 18702 -6 15743 2 24011
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Ν	9.98384 -9.79583 -1.30534
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Н	9.66839 -9.97703 2.0318
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Н	10.98238 -11.13204 -2.53495
Н	11.00068 -12.02566 1.66287
Н	11.67965 -12.62603 -0.67985
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Н	10.30921 -6.29608 -1.44705
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Н	13.11657 -8.09176 -5.02704
Н	12.58571 -5.3593 -1.7626
С	10.5615 -10.86863 -6.59245
Н	12.21428 -9.5626 -6.18861
С	8.69116 -10.57011 -5.07949
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Н	7.71403 -10.8342 -4.75796
Н	8.72595 -12.00752 -6.69645
С	5.70135 -8.78353 -3.58897
С	6.23627 -10.44862 -2.06666
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С	4.36929 -9.21977 -3.68525
C	4.91558 -10.95607 -2.10375
Н	6.96392 -10.91 -1.44537
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с u	3.97113 - 3.70332 - 3.91011
п С	4.50500 -7.1150 -5.41122 9.0579 6.24290 4.79164
с н	2 96181 -10 68804 -2 98213
C C	7 31429 -5 45034 -5 6878
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H	9.07901 -6.0199 -4.59546
Н	7.78467 -4.61732 -6.1824
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Ν	9.35226	-9.608 -	4.43434
Ν	-3.3606	8.61417	2.53139
Ν	-2.08792	11.1338	4.08643

Complex Ir-2 (DFT/B3LYP/GENECP/LANL2DZ)

Charge =	2 Multiplicity = 1
С	3.34019 3.39785 -0.08063
С	1.30692 2.3319 -0.07522
С	1.93809 3.57467 -0.11244
С	2.37678 1.3765 -0.01983
С	2.36611 -0.02944 0.02229
С	3.5661 -0.76303 0.04608
C	3.81159 -2.17608 0.07896
C	5,2004 -2,2994 0,07412
C	5 79203 -1 01602 0 04391
N	3 59013 2 07428 -0 02585
N	4 79994 -0 10291 0 02852
R	$5.00922 \ 1.42702 \ 0.02032$
D E	5.00022 1.43793 $0.02125.74245$ 1.91000 1.10120
г Г	5.74545 1.01909 -1.10139
Г С	5.6/351 1.83522 1.1863/
C U	-0.1/0/2 2.08506 -0.09098
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H	-0.50747 1.58821 0.82453
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Н	7.43656 0.14273 0.75858
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С	0.40843 -1.02711 1.24287
С	0.47154 -1.16267 -1.17277
С	-0.81225 -1.6983 1.2471
Н	0.86131 -0.71797 2.1802
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C	4427 44202 -011583
н	5 18416 4 1978 0 64073
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Н	-2.69691 13.59297 6.98613
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Ν	-1.10148 8.31458 3.92297
С	-5.54539 6.90939 2.69822
Н	-4.36705 5.84518 4.14025
Н	-6.42434 8.21909 1.21856
С	-0.64527 6.2846 5.76088
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С	0.0697 8.26413 4.55877
H	-6.38599 6.24462 2.76893
С	0.33777 7.24611 5.50504
Н	-0.47681 5.49496 6.46921
Н	0.8041 9.00299 4.35277

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C	7 58000 6 54472 1 31047
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L U	7.8775 -0.39177 -1.09014
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C	8.30/07 -7.72262 1.29112
Н	7.72255 -5.89787 -2.01348
Ν	8.578 -7.53946 -1.07819
С	8.80295 -8.19201 0.05436
Н	8.49046 -8.26192 2.19906
С	9.61471 -9.46597 -0.07299
Ν	9.98384 -9.79583 -1.30534
С	9.96357 -10.24551 1.04043
С	10.70302 -10.8963 -1.53819
Ċ	10.72023 -11.40912 0.83012
H	966839 -997703 20318
C	11 09907 -11 74148 -0 47471
с u	10.02222 11.12204 2.52405
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