

# Atomically Thin Ferromagnetic Half-Metallic Pyrazine-Fused Mn-Porphyrin Sheet: A Slow Spin Relaxation System

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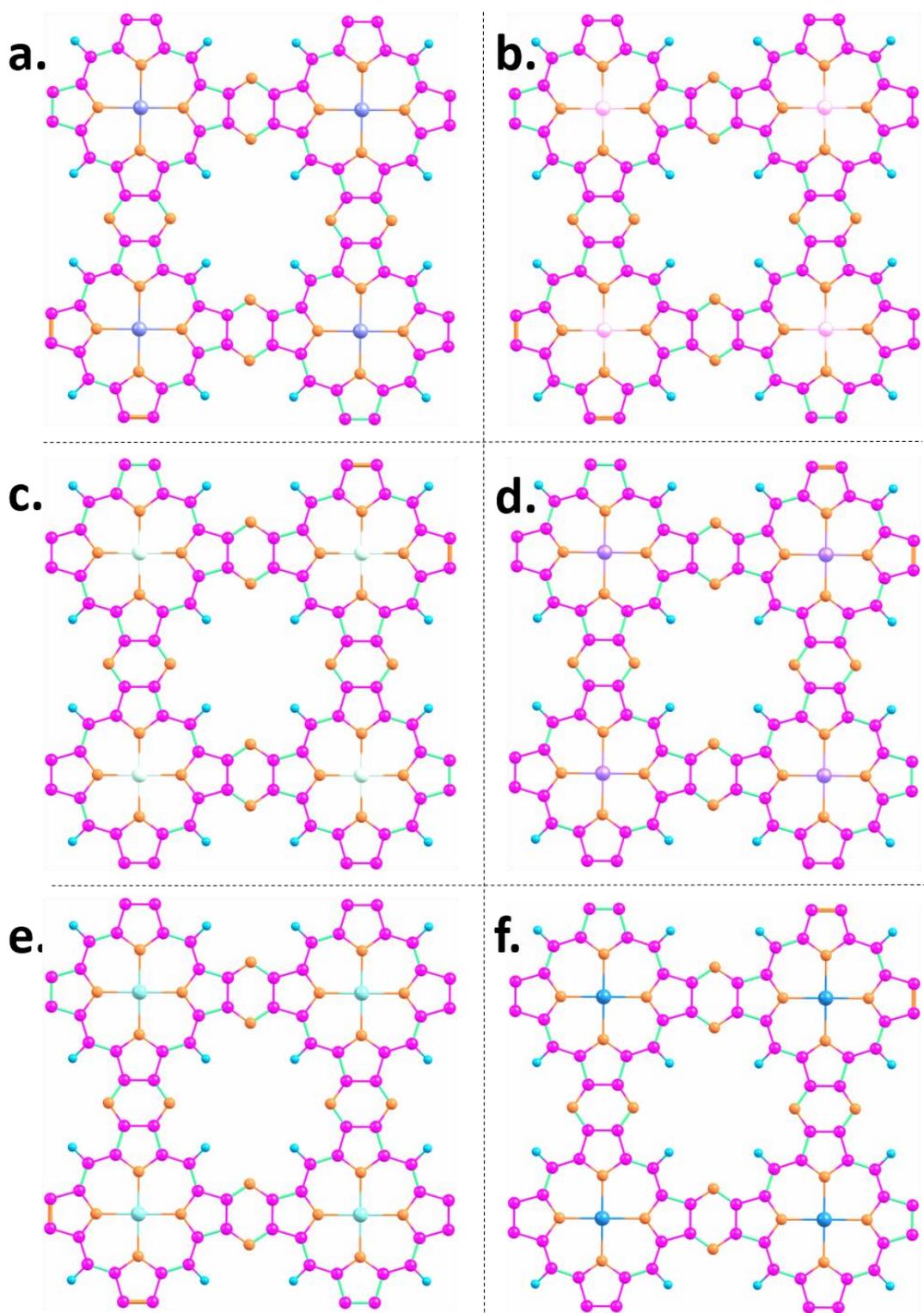
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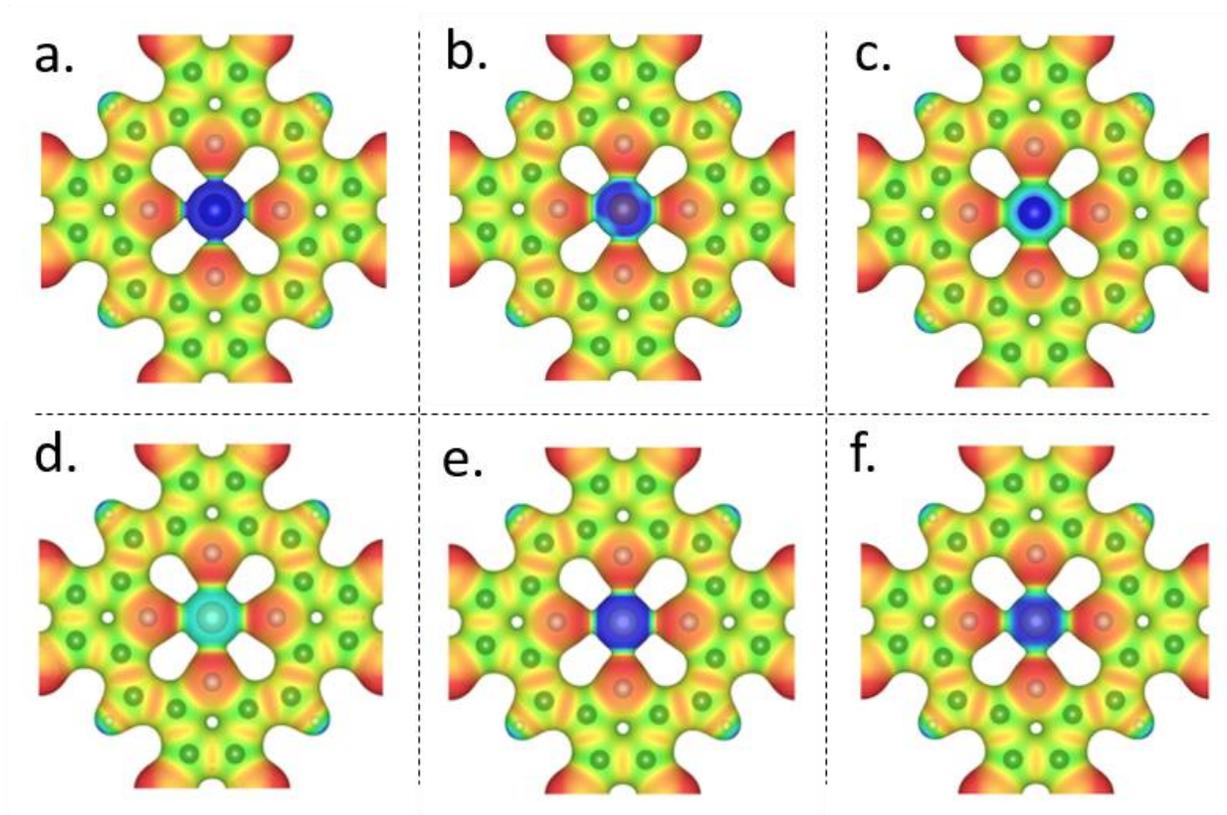
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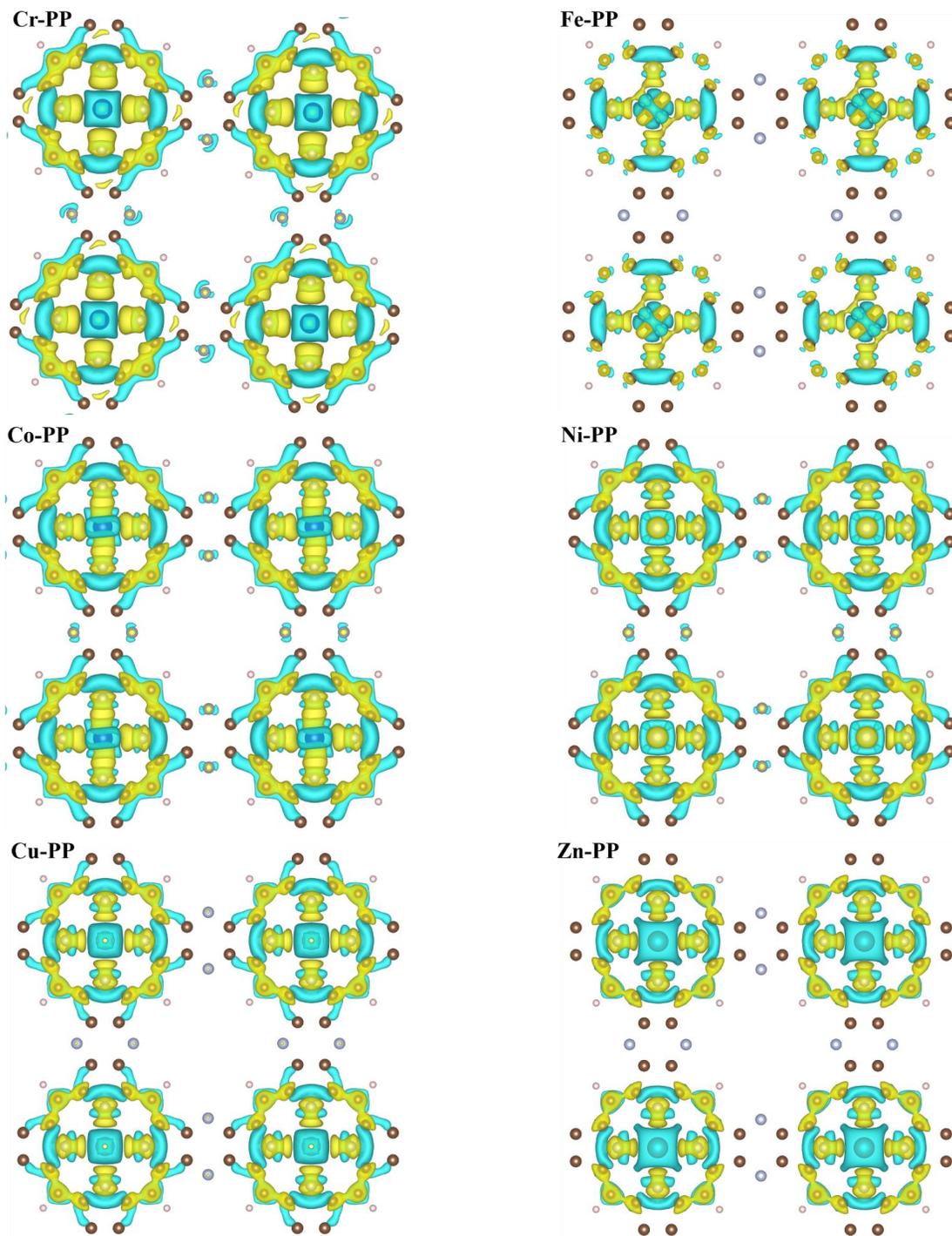
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**Table S1.** Net effective charges on N and TM atoms form Bader charge analysis.

TM-PP	Net effective charges								
	Metal	Pyrazine linker Nitrogen				Porphyrin moiety Nitrogen			
		N1	N2	N3	N4	N5	N6	N7	N8
No Metal		-1.17	-1.11	-1.17	-1.11	-1.01	-1.01	-1.01	-1.00
Cr	1.26	-1.18	-1.15	-1.18	-1.15	-1.15	-1.15	-1.17	-1.17
Mn	1.25	-1.15	-1.15	-1.15	-1.14	-1.13	-1.13	-1.15	-1.15
Fe	1.15	-1.10	-1.14	-1.10	-1.14	-1.15	-1.15	-1.10	-1.10
Co	1.09	-1.10	-1.14	-1.10	-1.14	-1.15	-1.15	-1.12	-1.12
Ni	0.98	-1.10	-1.14	-1.10	-1.14	-1.14	-1.14	-1.12	-1.12
Cu	0.91	-1.18	-1.14	-1.18	-1.14	-1.10	-1.10	-1.13	-1.13
Zn	1.12	-1.18	-1.11	-1.18	-1.11	-1.13	-1.13	-1.20	-1.19



**Figure S3.** Charge density difference plot for all the TM-PPs. (Isosurface value  $0.004 \text{ eV/\text{Å}^3}$ ). The yellow color shows accumulation of charge and cyan color shows depletion of charge.

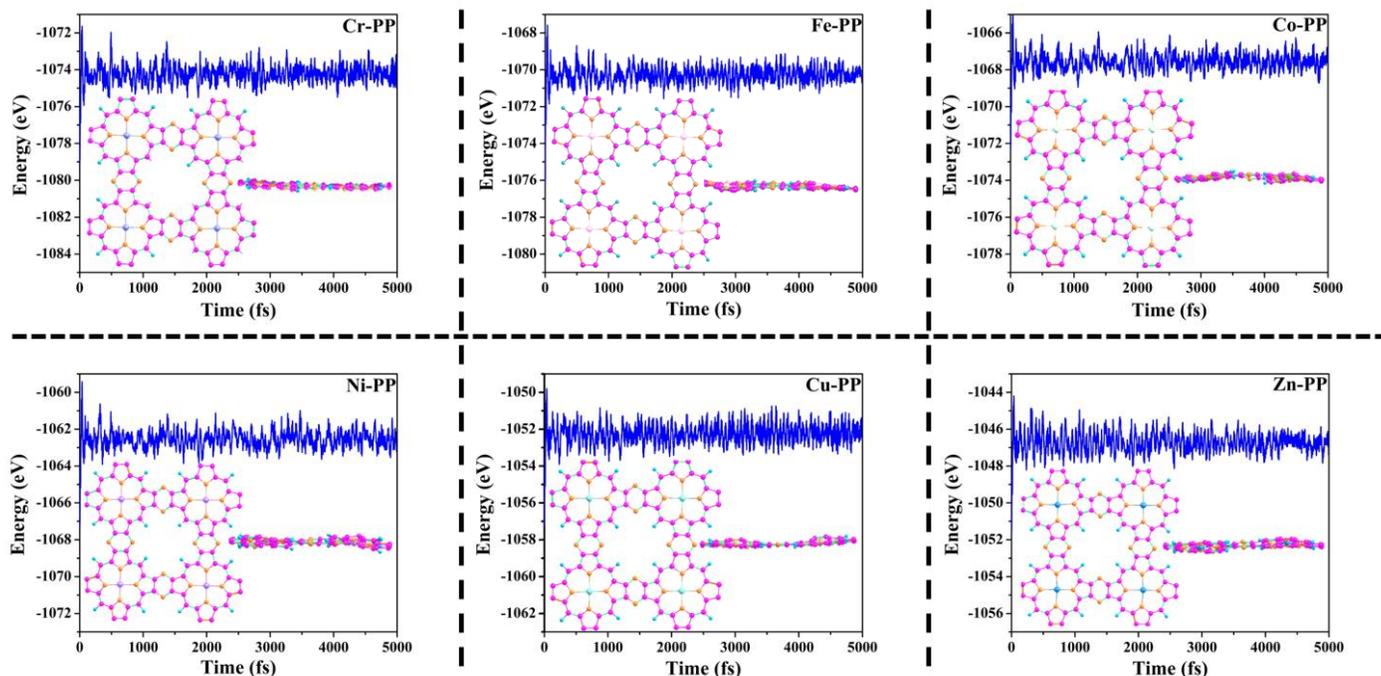
## Section S1. Charge density difference

We have plotted the charge density difference for the better understanding of charge transfer between the transition metal and Nitrogen of porphyrin cavity.

The definition of charge density difference is defined by:

$$\Delta\rho = \rho_{\text{PP-TM}} - \rho_{\text{PP}} - \rho_{\text{TM}}$$

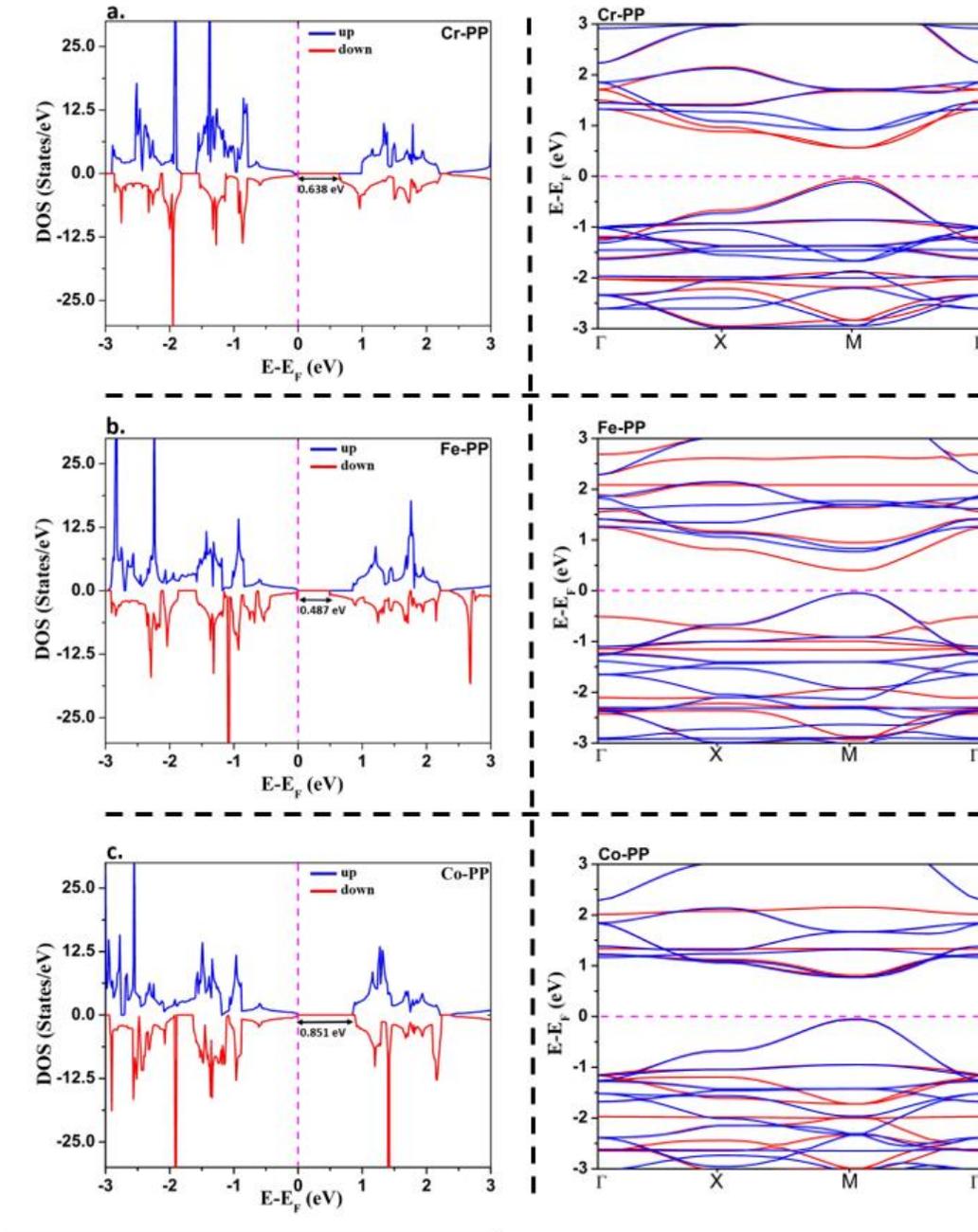
Where  $\rho_{\text{Por-M}}$ ,  $\rho_{\text{Por}}$  and  $\rho_{\text{M}}$  are the charge densities of TM-PPs, porphyrin moiety without transition metal incorporation and the transition metal. From the analysis of charge density difference. It clearly shows the accumulation of charge on N-atom (in porphyrin moiety) and depletion of charge on the transition metal. <sup>1</sup>

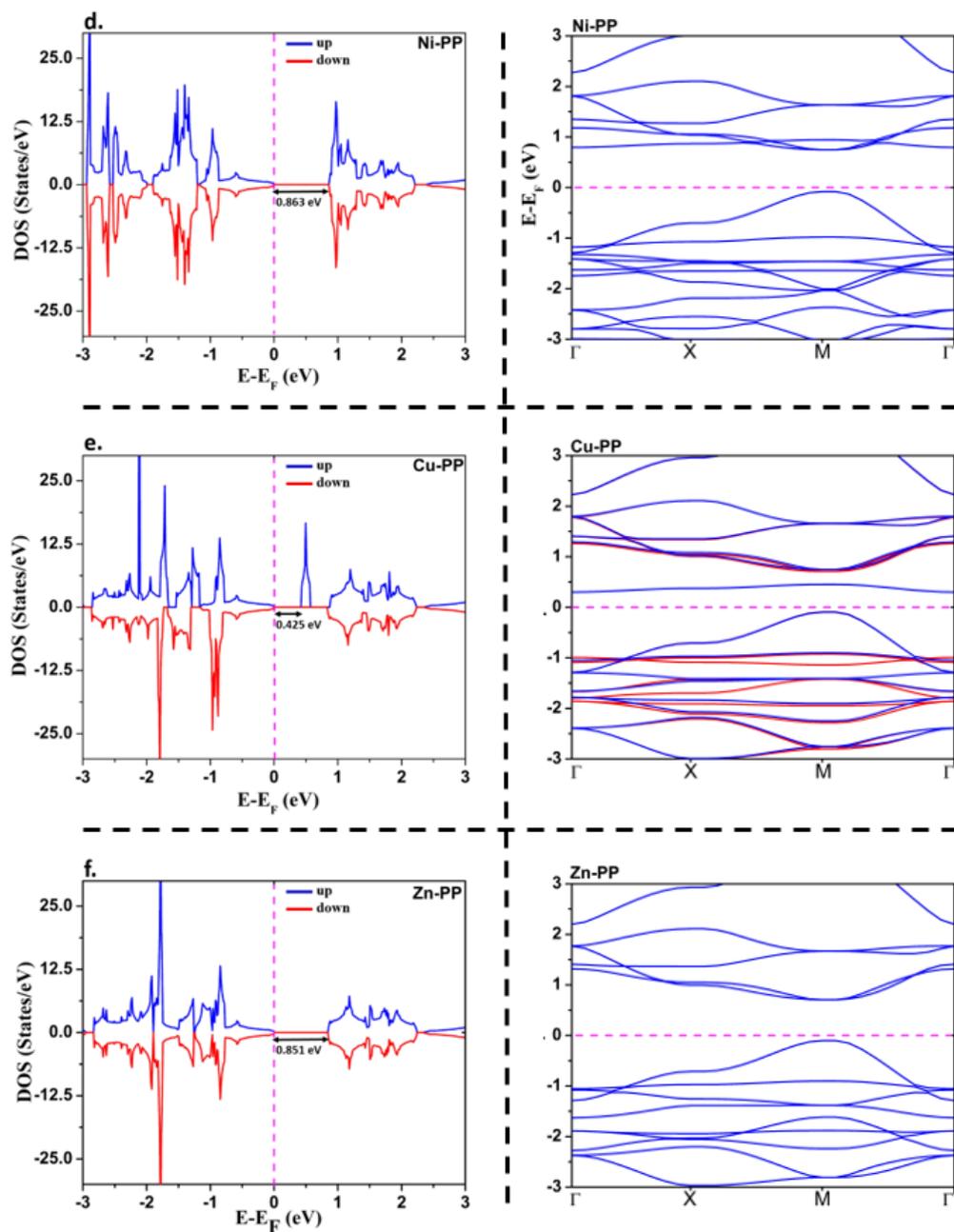


**Figure S4.** AIMD simulations of 2 × 2 TM-PP sheets for 5 ps with a time step of 1 fs at 300 K

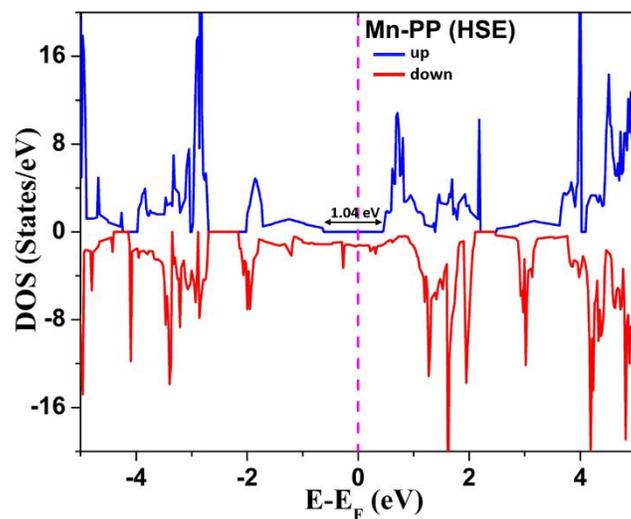
**Table S2.** Summary of elastic constants ( $C_{ij}$  in N/m) and Poisson's ratio for all the TM-PPs.

Moduli	Cr-PP	Mn-PP	Fe-PP	Co-PP	Ni-PP	Cu-PP	Zn-PP
$C_{11}$	148	114	106	111	106	97	96
$C_{12}$	37	24	39	31	36	44	52
$C_{44}$	57	45	33	40	35	26	22
$\nu$	0.18	0.17	0.27	0.22	0.25	0.31	0.35

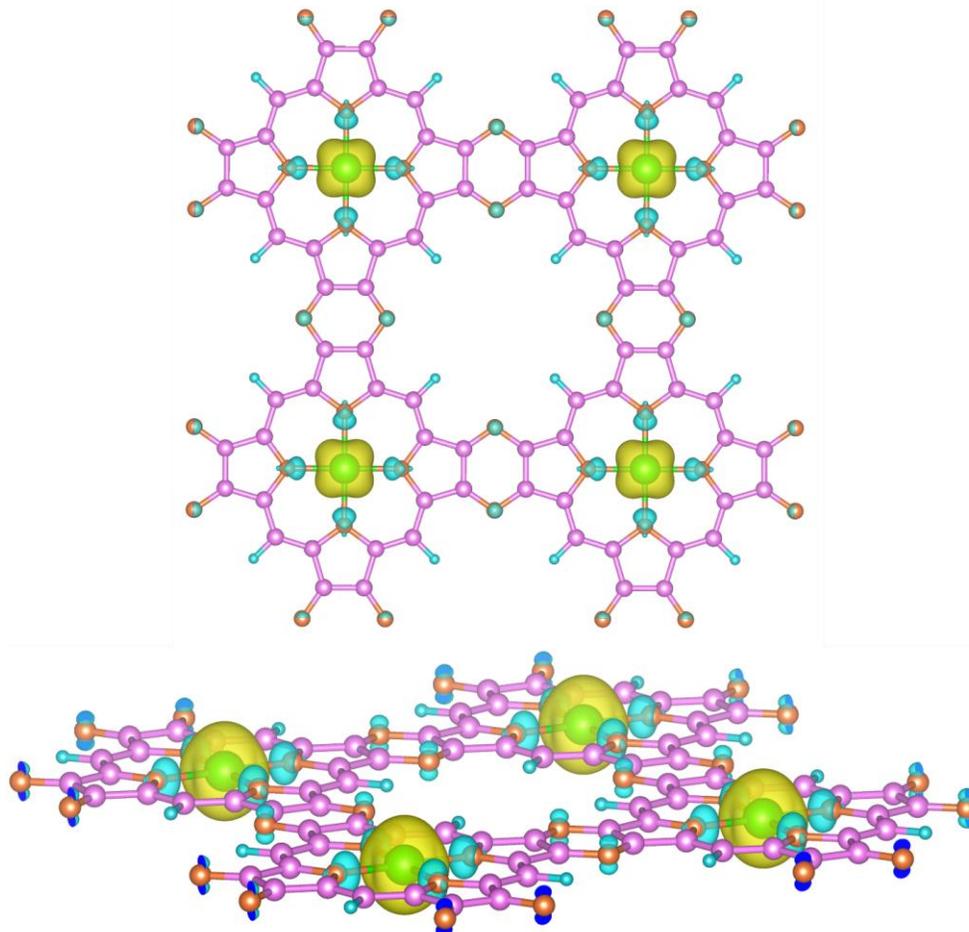




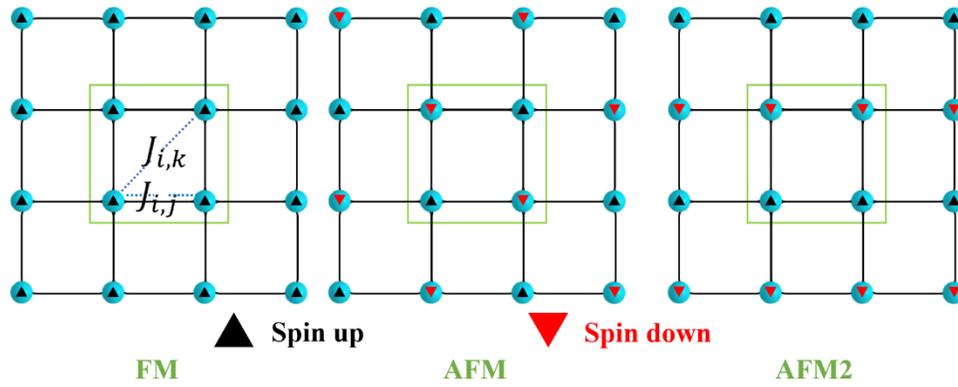
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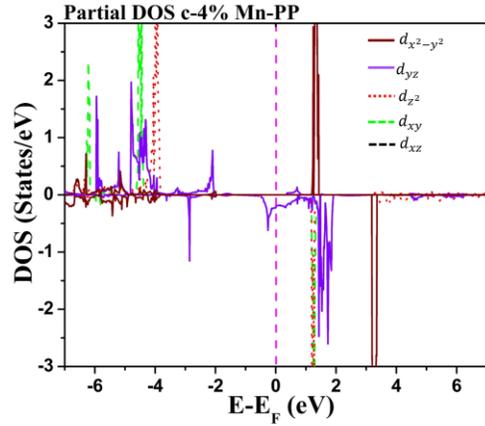
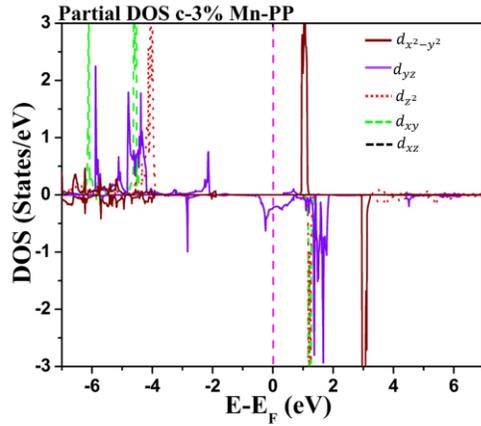
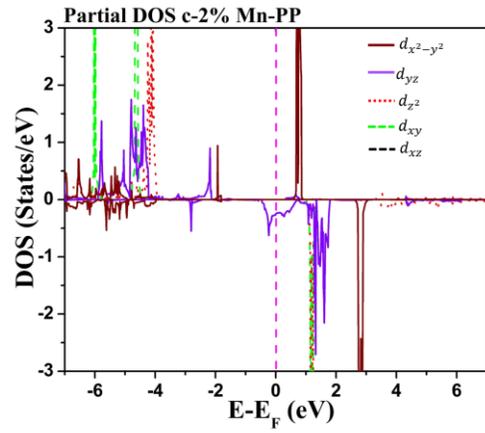
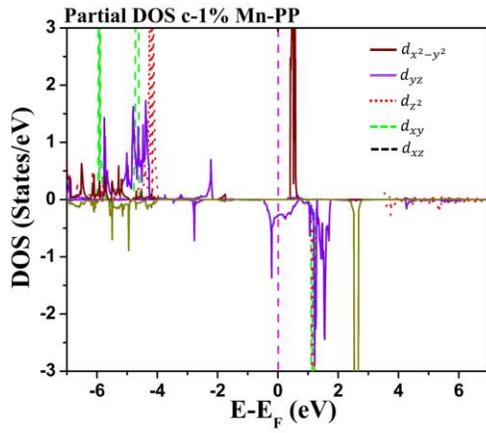
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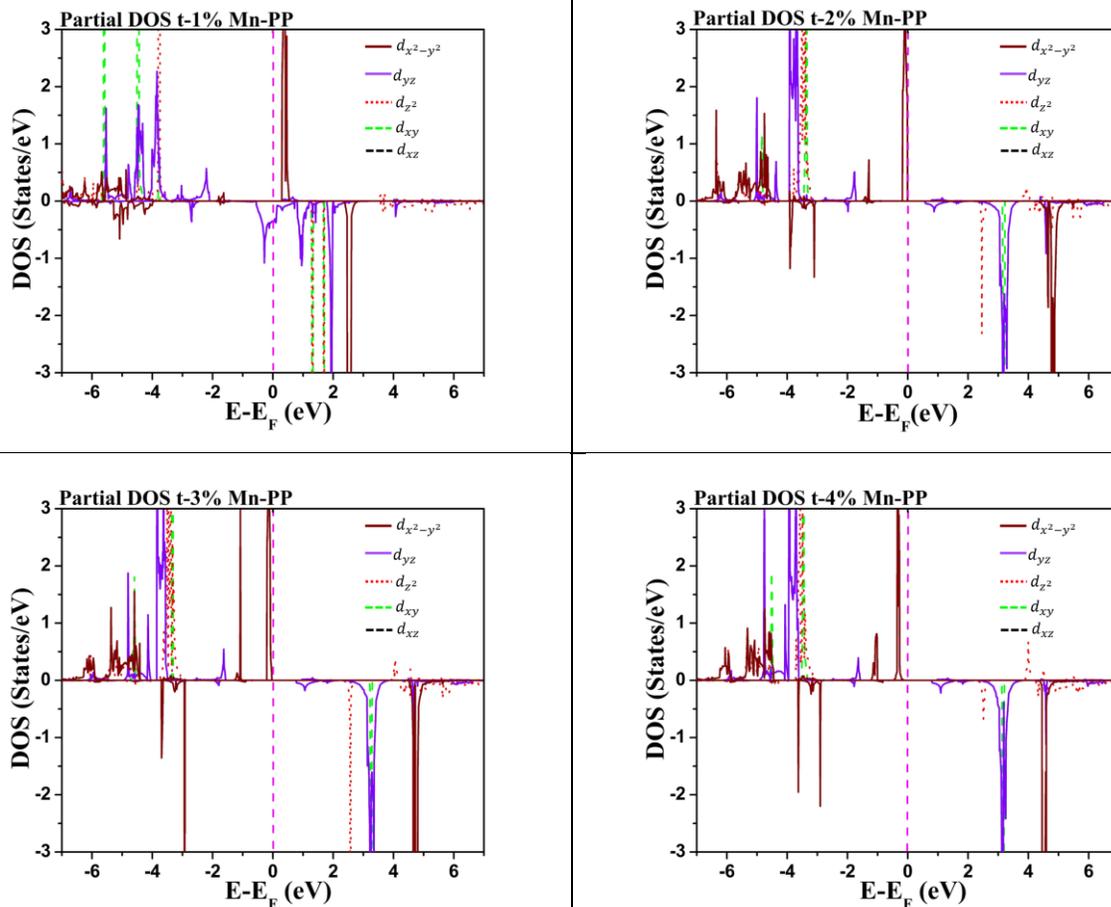


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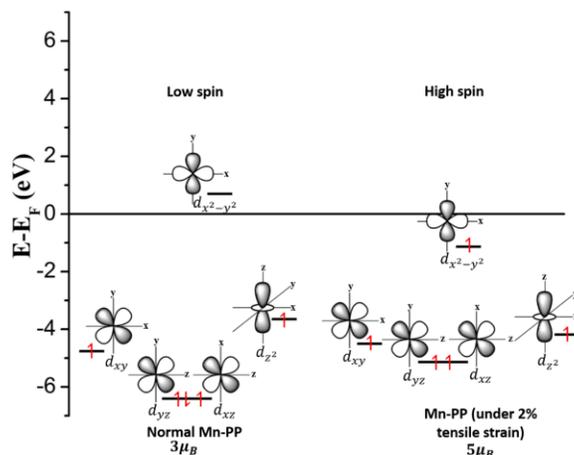


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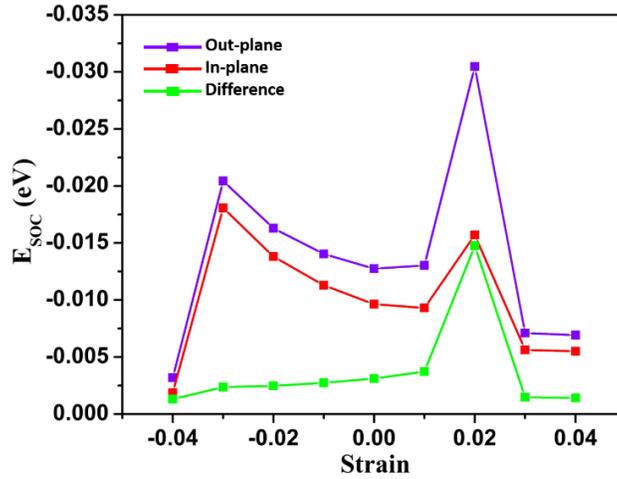
**Figure S9.** Partial DOS Mn-PP under compressive and tensile strain.



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		Unstrained	C-4%	C-3%	C-2%	C-1%	T-1%	T-2%	T-3%	T-4%
$E_{\text{SOC}}$	001	-0.0127509	-0.0031846	-0.0204375	-0.0162915	-0.0140363	-0.0130289	-0.0304598	-0.0071021	-0.0069149
	100	-0.0096328	-0.0018703	-0.0180670	-0.0138112	-0.0112889	-0.0093012	-0.0157078	-0.0056189	-0.0054952
	diff <sub>001-100</sub>	-0.0031181	-0.0013143	-0.0023705	-0.0024803	-0.0027474	-0.0037277	-0.014752	-0.0014832	-0.0014197



**Figure S11.** Variation of  $E_{\text{SOC}}$  under the strain for perpendicular (out-plane) and parallel (in-plane) direction of Mn-PP.

## References

1. S. Deng, L. Wang, T. Hou, and Y. Li, *J. Phys. Chem. C*, 2015, **119**, 28783–28788.