

Supporting Information

Controlling of the Electronic Properties of WS₂ by Graphene Oxide

Substrate from First-principles calculations

Mingye Yang, Lu Wang*, Tingjun Hou, Youyong Li*

Functional Nano & Soft Materials Laboratory (FUNSOM) and Collaborative Innovation Center of Suzhou Nano Science and Technology Jiangsu Key Laboratory for Carbon-Based Functional Materials & Devices, Soochow University, Suzhou, Jiangsu 215123, China.

E-mail: lwang22@suda.edu.cn (Dr. Lu Wang); yyli@suda.edu.cn (Prof. Youyong Li)

Table S1. Band gaps of pristine WS₂ and some GO/WS₂ heterostructures with and without considering SOC effect.

Band Gap/eV	WS ₂	WS ₂ /GO-h ₂ e (8%)	WS ₂ /GO-h ₂ e (40%)	WS ₂ /GO-h ₂ e (56%)	WS ₂ /GO-h ₂ e (80%)
PBE	1.92	0.32	1.29	1.72	1.91
PBE+SOC	1.58	0.33	1.15	1.48	1.57

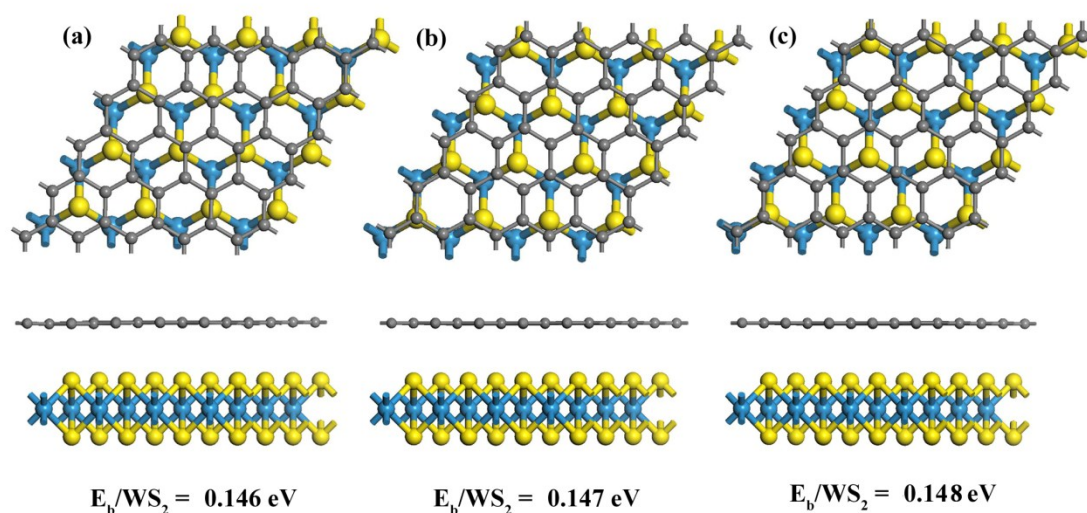


Fig. S1. Different staking patterns and the binding energies per WS₂ unit of graphene and WS₂ hybrids.

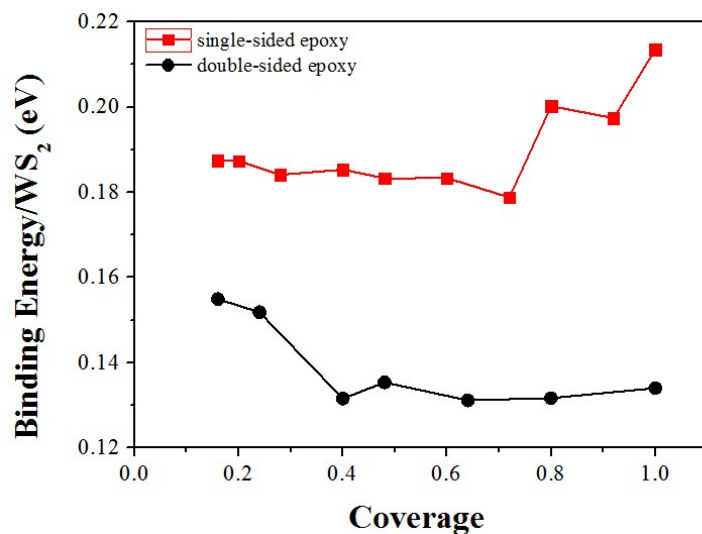


Fig. S2. Binding energies of WS₂ and GO with epoxy distributed on single side and on both sides heterostructures.

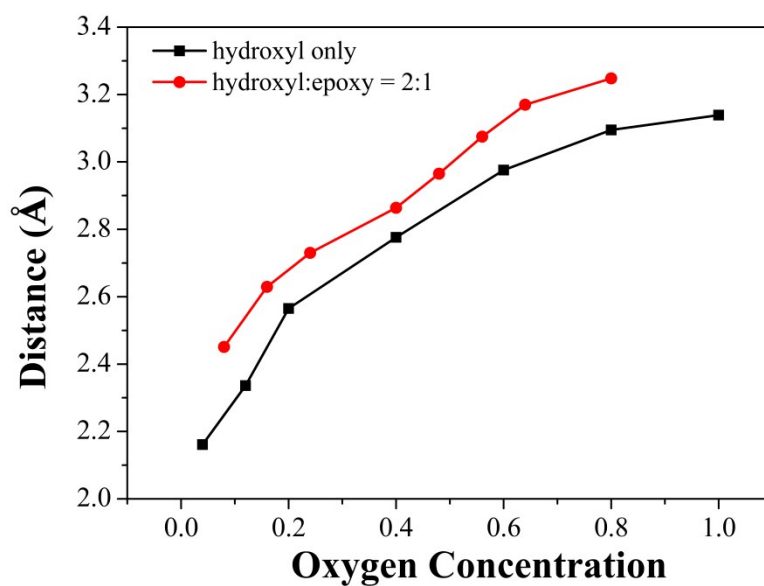


Fig. S3. The distances between S atoms in WS₂ and O atoms in GO as the function of oxygen concentration.

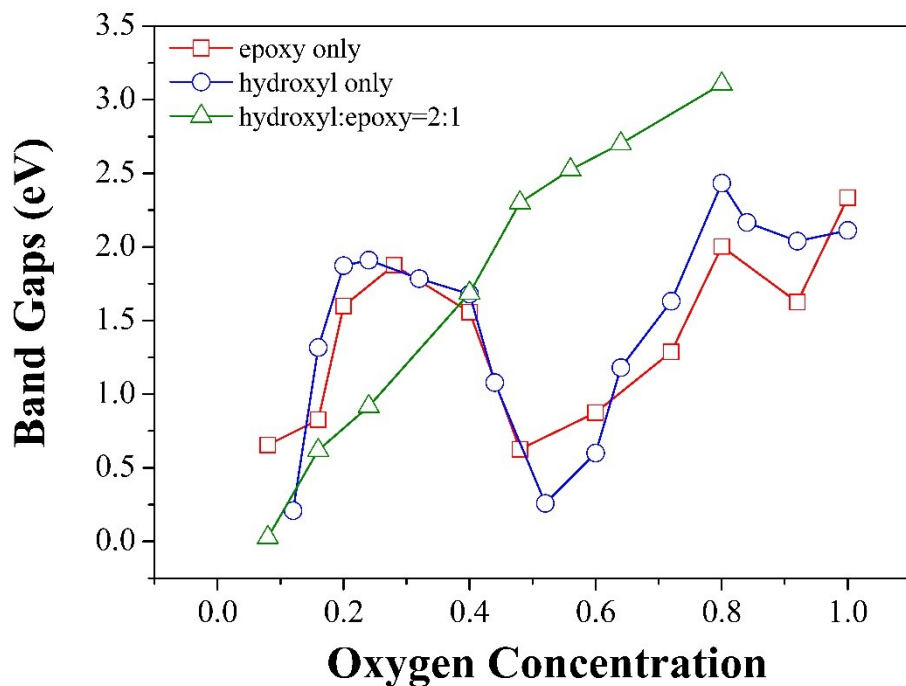


Fig. S4. Band gaps of the free-standing graphene oxide as the oxygen concentration increasing.

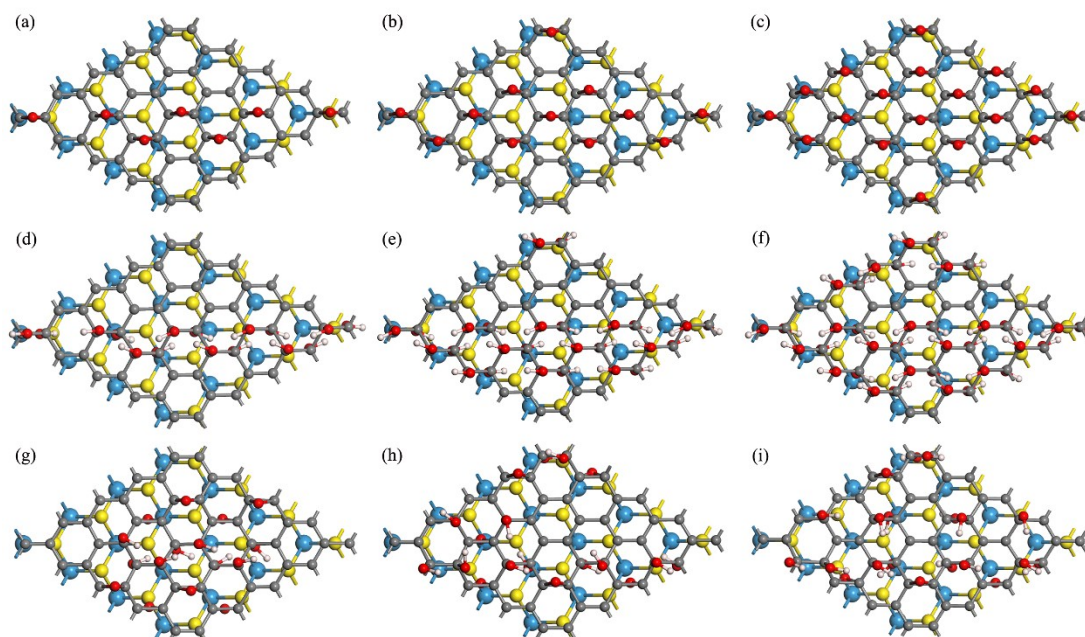


Fig. S5. Optimized Structures of GO and WS₂ heterostructures. (a) to (d) are the structures of e-GO/WS₂ (h₂-GO/WS₂) at the first peak, the peak valley and the second peak of the band gaps. Their oxygen concentrations are 0.28 (0.32), 0.48 (0.52) and 0.72 (0.72), respectively. (g) to (i) are the heterostructures with the same oxygen concentration of 0.48 but with different hydroxyl to epoxy ratios of 1:1, 2:1 and 4:1, respectively.