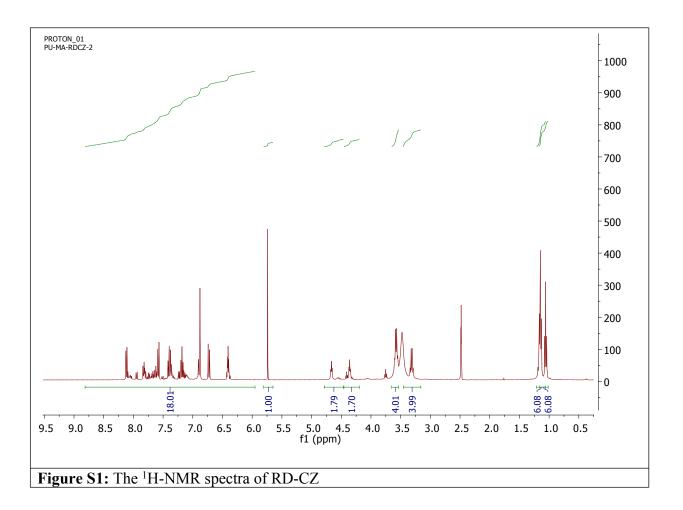
Rhodamine-Based Conjugate Polymer: Potentiometric, colorimetric and voltammetric sensing of mercury ion in aqueous medium.

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The ¹H-NMR spectra of RD-CZ have the characteristic peaks of the structure, as illustrated in Figure S1. ¹H-NMR (400 MHz, DMSO-d₆) δ H/ppm: 1.04 (t; 6H), 1.14 (t; 6H), 3.31 (m; 4H), 3.57 (m; 4H), 4.35 (t; 2H), 4.66 (t; 2H), 5.72 (s; 1H), 6.33-8.43 (m; 18H-Arm).



In FTIR spectrum of the RD-CZ monomer which is shown in Figure S2 (black line), the following absorption bands were observed at 682 cm⁻¹ (C–H_a out of plane bending of thiophene), 1181 cm⁻¹ (C-N stretching), 1466 cm⁻¹ (C-H bending), 1588 cm⁻¹ (aromatic C=C stretching), 1727 cm⁻¹ (C=O stretching). In Figure S2 (red line), the FTIR spectra of electrochemically synthesized P(RD-CZ) was showed the characteristic peaks of the monomer and new strong absorption peak at 1057 cm⁻¹ due to incorporation dopant ion (PF₆⁻) into polymer film. Then, after interaction of Hg²⁺ ions, (green line) stretching vibration of C-O appears at 1044 cm⁻¹ in P(RD-CZ) which shifts to the 1079 cm⁻¹. This result indicate that the binding of the –C-O group on rhodamine moieties of P(RD-CZ) with Hg²⁺ ions.

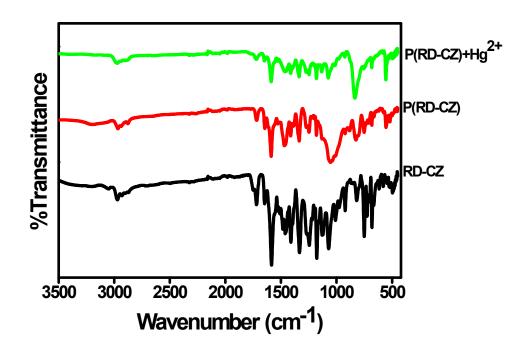


Figure S2: FTIR spectra of RD-CZ, P(RD-CZ) and P(RD-CZ)+Hg²⁺

In FTIR spectrum of the RD-CZ monomer which is shown in Figure S2 (black line), the following absorption bands were observed at 682 cm⁻¹ (C–H_a out of plane bending of thiophene), 1181 cm⁻¹ (C-N stretching), 1466 cm⁻¹ (C-H bending), 1588 cm⁻¹ (aromatic C=C stretching), 1727 cm⁻¹ (C=O stretching). In Figure S2 (red line), the FTIR spectra of electrochemically synthesized P(RD-CZ) was showed the characteristic peaks of the monomer and new strong absorption peak at 1057 cm⁻¹ due to incorporation dopant ion (PF₆⁻) into polymer film. Then, after interaction of Hg²⁺ ions, (green line) stretching vibration of C-O appears at 1044 cm⁻¹ in P(RD-CZ) which shifts to the 1079 cm⁻¹. This result indicate that the binding of the –C-O group on rhodamine moieties of P(RD-CZ) with Hg²⁺ ions.