

## Electronic Supporting Information

# **Effective determination of Cd(II) and Pb(II) simultaneously using square wave anodic stripping voltammetry based on aluminum silicon carbide-reduced graphene oxide nanocomposite modified electrode**

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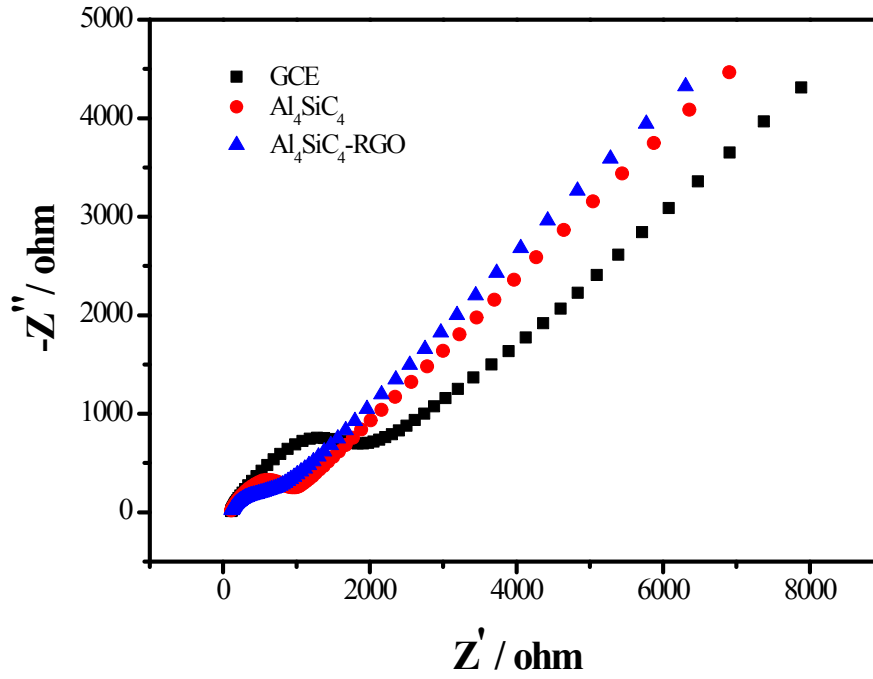
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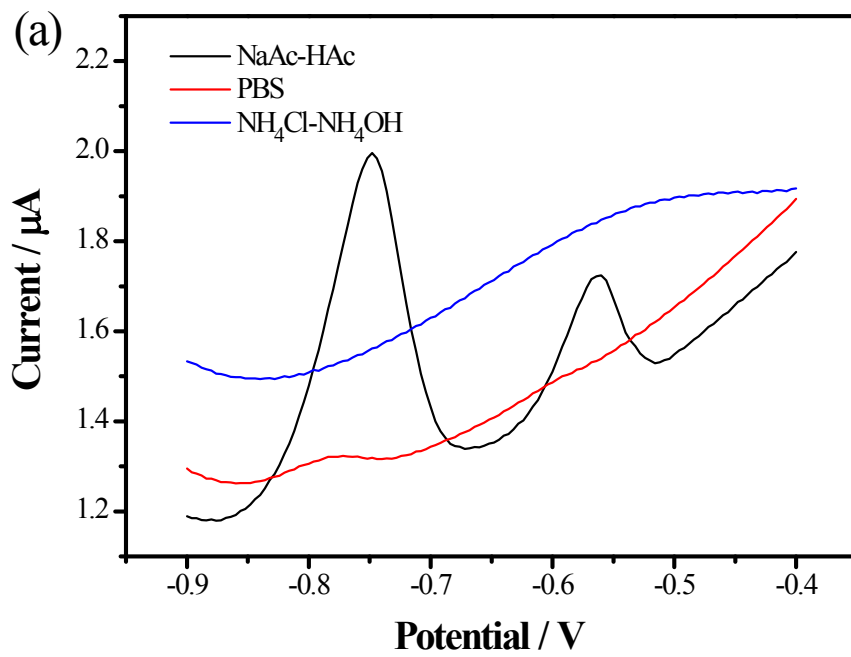
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**Fig. S1** Nyquist diagram of electrochemical impedance spectra for bare ,  $\text{Al}_4\text{SiC}_4$  nanoparticle, and  $\text{Al}_4\text{SiC}_4$ -RGO nanocomposite modified GCE in the solution of 5 mmol/L  $[\text{Fe}(\text{CN})_6]^{3-/4-}$  containing 0.1 mol/L KCl.



**Fig. S2 (a)**

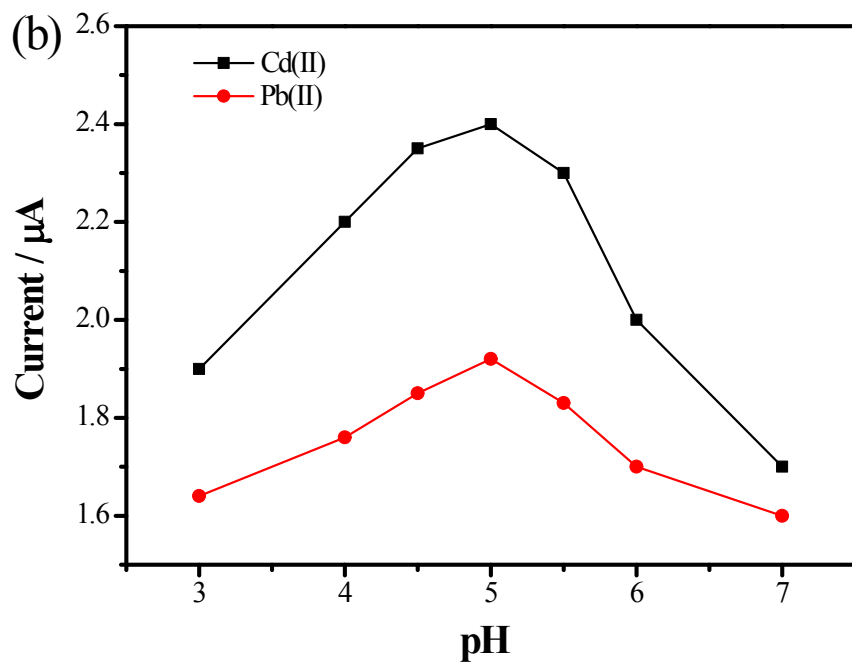


Fig. S2 (b)

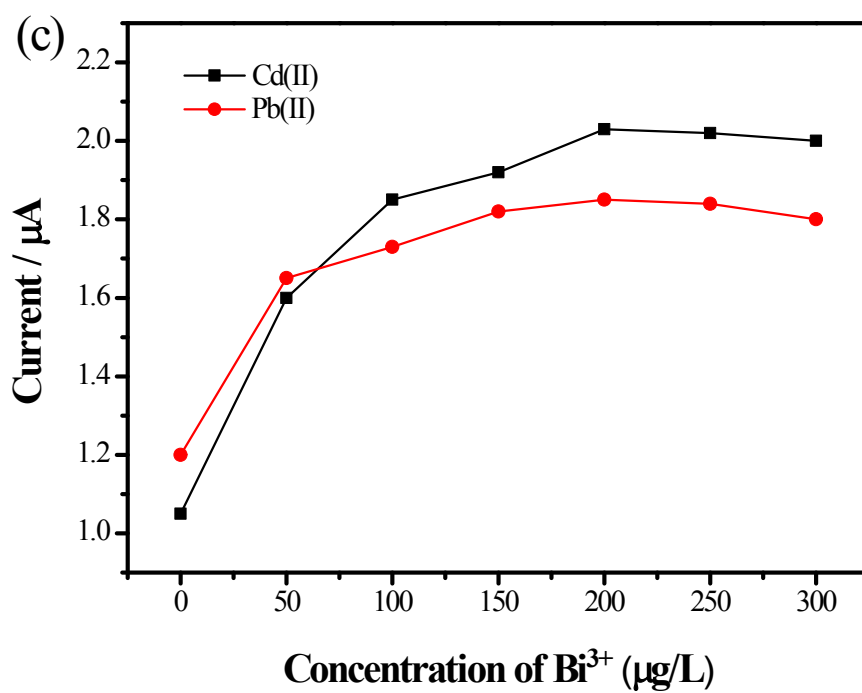


Fig. S2 (c)

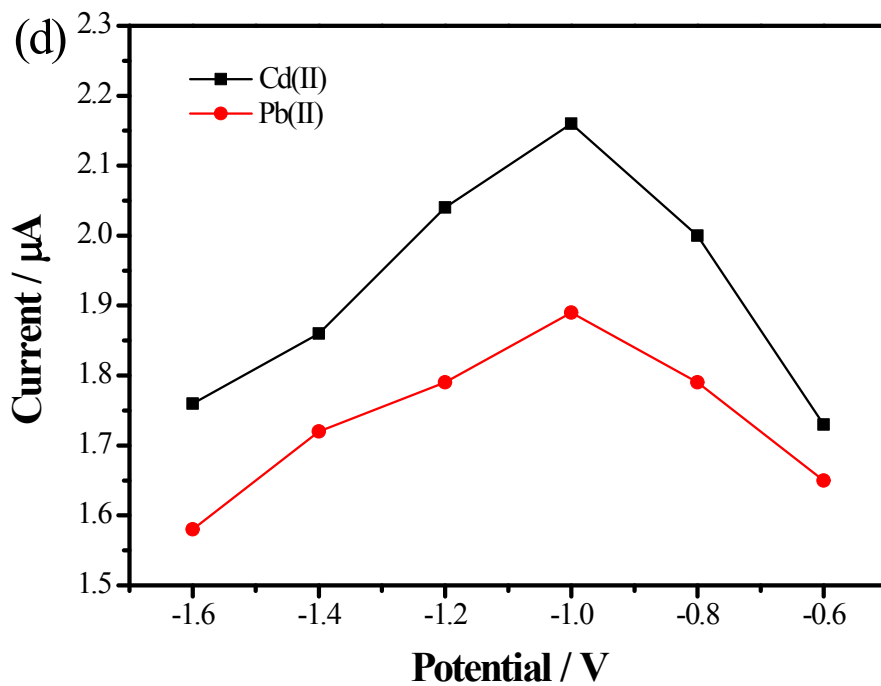


Fig. S2 (d)

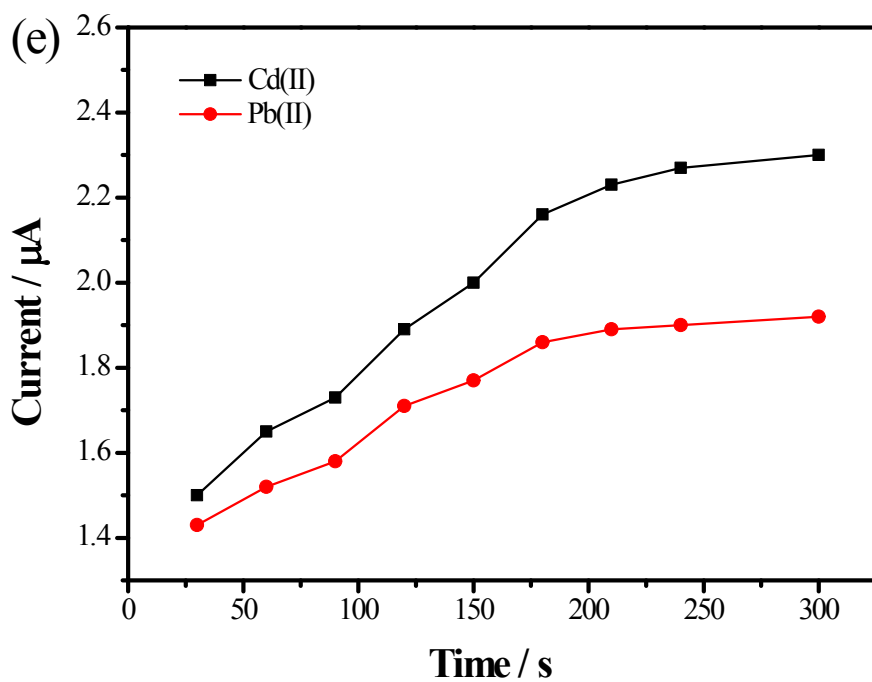
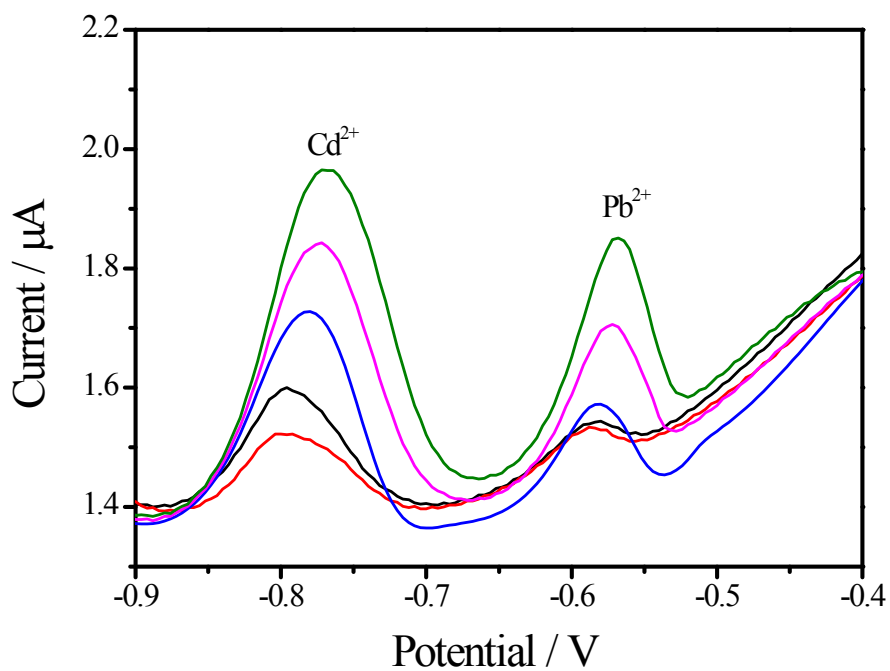


Fig. S2 (e)

**Figs. S2(a-e)** Optimization of experimental conditions. Influence of (a) supporting electrolytes; (b) pH value; (c) Concentrations of  $\text{Bi}^{3+}$ ; (d) deposition potential; and (e) deposition time on the voltammetric response of the  $\text{Al}_4\text{SiC}_4\text{-RGO}$  nanocomposite modified GCE. Data were evaluated by SWASV of 150  $\mu\text{g/L}$  each of Cd(II) and Pb(II).



**Fig.S3** SWASV response of the  $\text{Al}_4\text{SiC}_4\text{-RGO}$  nanocomposite modified GCE for the simultaneous analysis of  $\text{Cd(II)}$  and  $\text{Pb(II)}$  with different concentrations, from LOQ to  $150 \mu\text{g/L}$  ( $10 \mu\text{g/L}$ ,  $40 \mu\text{g/L}$ ,  $70 \mu\text{g/L}$ ,  $100 \mu\text{g/L}$ ,  $120 \mu\text{g/L}$ ,  $150 \mu\text{g/L}$ ).  $0.1 \text{ mol/L}$  acetate buffer (pH 5.0); Deposition potential,  $-1.0 \text{ V}$ ; deposition time,  $180 \text{ s}$ ; concentrations of  $\text{Bi}^{3+}$ ,  $200 \mu\text{g/L}$ .