Fluorescence light-up detection of cyanide in water based on cyclization reaction followed by ESIPT and AIEE

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Fig. S1 Time-dependent absorption spectra changes of 1 (50 μ M, DMSO) in the presence of 100 equiv. of cyanide. The arrows indicate the changes over time. Inset: the time-dependent changes of absorbance at 332 nm and 455 nm.

Fig. S2 Time-dependent fluorescence ($\lambda_{ex} = 398$ nm) spectra changes of 1 (50 μ M, DMSO) in the presence of 100 equiv. of cyanide. The arrow indicates the changes over time. Inset: the time-dependent changes of emission intensity at 476 nm.

Fig. S3. (a) Absorption spectra and (b) colorimetric responses of 1 (50 μ M) before and after addition of various anions (100 equiv.) in DMSO. The spectra and pictures were obtained immediately.

Fig. S4. (a) Fluorescence spectra (λ_{ex} = 398 nm) and (b) fluorimetric (excitation at 365 nm) responses of 1 (50 µM) before and after addition of various anions (100 equiv.) in DMSO. The spectra were obtained after 120 min.

Fig. S5. The (a) absorption and (b) fluorescence intensity ($\lambda_{ex} = 398$ nm) changes of 1 (50 μ M) in the presence of 100 equiv. of CN⁻ at 25 °C, 50 °C and 70 °C were recorded in DMSO.

Fig. S6. ESI mass spectrometry of 1 after the addition of TBACN (5 equiv).

Fig. S7. ¹³C NMR spectra of (a) 1 (2.5 mM, DMSO- d_6) after the addition of TBACN (DMSO- d_6 solution, 5 equiv) and (b) the synthetic product 2.

Fig. S8. ¹H NMR spectra of **1** (2.5 mM, DMSO- d_6) before (a) and after (b) the addition of TBACN (DMSO- d_6 solution, 15 equiv). (c) ¹H NMR spectra of the synthetic product **2**.

Fig. S9. Time-dependent UV-Vis spectral changes of **1** (50 μ M) upon the addition of 100 equiv. of CN⁻ in water with CTAB (a) 1 min to 15 min (b) 15 min to 150 min.

Fig. S10. Time-dependent fluorescence changes of 1 (50 μ M) upon the addition of 100 equiv. of CN⁻ in water with CTAB ($\lambda_{ex} = 347$ nm)) from 1 min (yellow line) to 15 min (blue line) and 150 min (green line). Inset: Time-dependent changes of emission intensity at 508 nm.

Fig. S11. Colorimetric responses of 1 (50 μ M) before and after addition of various anions (5 equiv., immediately) in water with CTAB.

Fig. S12. Colorimetric response of 1 (50 μ M) to CN⁻ (5 equiv.) in the presence of various anions (5 equiv.) in water with CTAB. The spectra were obtained immediately. **Table. S1.** Chemodosimeters for cyanide determination in water.

Fig. S13. Fluorescence intensity spectra of 1 (50 μ M) in the presence of different concentration of CN⁻ (0-100 equiv.) in water with CTAB ($\lambda_{ex} = 347$ nm). Inset: fluorescence intensities at 508 nm as a function of CN⁻ concentration. The spectra were obtained after 150 min.

Fig. S14. Fuorescent response of **1** (50 μ M) to CN⁻ (100 equiv.) in the presence of various anions (100 equiv.) in water with CTAB ($\lambda_{ex} = 347$ nm). The spectra were obtained after 150 min.

Fig. S15. pH dependence of fluorescence intensity at 510 nm of **1** with or without CN⁻ (100 equiv.).

Fig. S16. ¹H NMR spectrum of 2 in CDCl₃.

Fig. S17. ¹³C NMR spectrum of 2 in DMSO.

Fig. S18. ESI-MS spectrum of 2 in DMSO/MeOH.

Fig. S19. FT-IR spectrum of 2.

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Fig. S11. Colorimetric responses of 1 (50 μ M) before and after addition of various anions (5 equiv., immediately) in water with CTAB.







Chemodosimeter	Medium	LOD (M)	Reference in maintext
Br Br	HEPES buffer	$1.9 imes 10^{-8} \mathrm{M}$	65
	HEPES buffer	$5.52 \times 10^{-8} \text{ M}$	66
-N - C - C - C - C - C - C - C - C - C -	HEPES-DMSO (250 : 1, v/v))	2.86×10^{-7} M.	67
	DMSO-H ₂ O (5 : 95 v/v)	$1.5 imes 10^{-6} \mathrm{M}$	68
$O_2N \longrightarrow H \to O_2$	Aqueous CTABr	$4.93 \times 10^{-5} \text{ M}$	69
	Water with Triton X-100	1.56 × 10 ⁻⁶ M	70
	Water with CTAB	5.92 × 10 ⁻⁷ M	Our work

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