Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2017

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Figure S4. Images of coloured complex in organic extract where 1<sup>st</sup> vial is blank [Bi (III) absent], 2<sup>nd</sup> vial is Bi (III)-I<sup>-</sup> complex and 3<sup>rd</sup> vial is Bi (III)-I<sup>-</sup>-N-OAA complex

Figure S1.

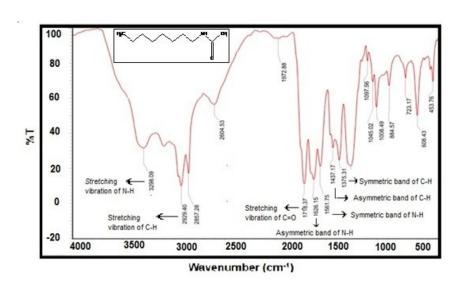


Figure S2.

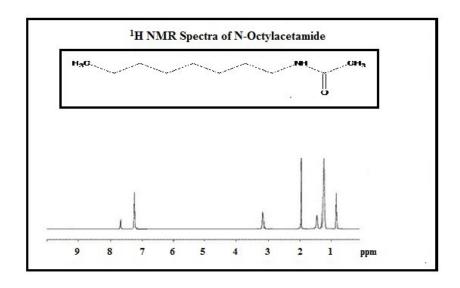


Figure S3.

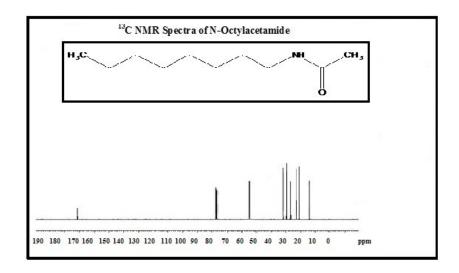


Figure S4.

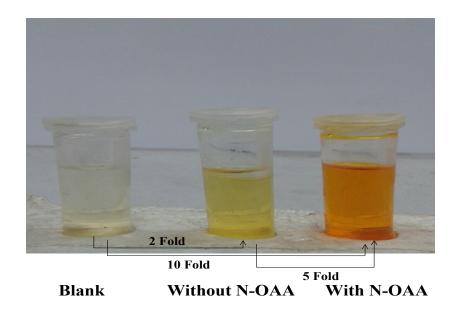


Table S1. FTIR Characterization details of N-OAA.

Table S2. Effect of different amides at  $\lambda_{\text{max}}$  365 nm.

Table S3. Statistical data calculation.

Table S1

S.	Functional Group	Band(cm <sup>-1</sup> )
No.		
1	Amide Carbonyl(C=O)	1718.37
2	Stretching Vibration of N-H	3298.09
3	Symmetric /Asymmetric band of N-H	1561.75-1626.15
4	Stretching Vibration of C-H	2829.40
5	Symmetric/Asymmetric band of C-H	1375.31-1437.17
	Symmetric rasymmetric band of C 11	

Table S2

S. no.	Amides	Molecular Formula	Molar Absorptivity  E (L/mol/cm)
1.	N-butylacetamide	C <sub>4</sub> H <sub>9</sub> NHCOCH <sub>3</sub>	2.0X10 <sup>3</sup>
2.	N-phenylacetamide	C <sub>6</sub> H <sub>5</sub> NHCOCH <sub>3</sub>	1.7X10 <sup>4</sup>
3.	N-hexylacetamide	C <sub>6</sub> H <sub>13</sub> NHCOCH <sub>3</sub>	4.4X10 <sup>3</sup>
4.	N-octylacetamide	C <sub>8</sub> H <sub>17</sub> NHCOCH <sub>3</sub>	7.2X10 <sup>5</sup>

Table S3

No. of	Concentration of	Absorbance at	Statistical Data
determination	Bi(III) μg/mL	$\lambda_{max\ 365nm}$	
	1.0	0.342	Mean Abs.= 0.345
		0.347	St. Dev.= 0.005
6		0.352	$RSD = \pm 1.44\%$
0		0.341	Dt. Limit=36.0 μg/L
		0.350	
		0.343	