## A simple and fast method based on functionalized magnetic nanoparticles for

## determination of Ag(I), Au(III) and Pd(II) in mine stone, road dust and water samples

Sara Karami, Homeira Ebrahimzadeh\*, Ali Akbar Asgharinezhad

Faculty of Chemistry, Shahid Beheshti University, G.C., Evin, Tehran, Iran

\* Corresponding author. Tel.: +98 21 29902891; fax: +98 21 22403041.

E-mail address: h-ebrahim@sbu.ac.ir (H. Ebrahimzadeh)



Fig. 1S: TGA/DTA curves of Fe<sub>3</sub>O<sub>4</sub>@Mu nanosorbent.

## Table 1S

The tolerance limit of potentially interfering ions in the determination of Ag(I), Au(III) and Pd(II) ions.

Interfering Ions	Tolerable concentration ratio X/Ag(I), Au(III) - and Pd(II)	Recovery (%)			
		Ag(I)	Au(III)	Pd(II)	
Na <sup>+</sup>	10000	$98.0\pm2.0$	$97.3\pm2.7$	99.1 ± 1.9	
$\mathrm{K}^+$	10000	$96.5\pm1.6$	$94.8\pm2.0$	$99.3\pm2.5$	
Ca <sup>2+</sup>	1000	$97.8\pm2.3$	$96.6 \pm 3.0$	$98.4 \pm 2.1$	
$Mg^{2+}$	1000	$98.5\pm2.0$	$97.3 \pm 2.8$	98.1 ± 1.8	
Pb <sup>2+</sup>	1000	$99.0\pm2.0$	$97.2 \pm 3.5$	$98.3 \pm 2.6$	
$Mn^{2+}$	1000	$97.4 \pm 2.8$	98.3 ± 1.4	$98.8\pm2.0$	
$Zn^{2+}$	1000	$98.8 \pm 1.7$	99.0 ± 1.0	$97.5 \pm 2.8$	
Ni <sup>2+</sup>	1000	$97.2 \pm 2.4$	$97.9\pm2.8$	$98.0\pm3.2$	
Co <sup>2+</sup>	1000	$98.7\pm3.0$	$96.5 \pm 2.0$	$97.3 \pm 2.0$	
$Cu^{2+}$	1000	$99.2 \pm 1.9$	$99.3 \pm 3.2$	$97.6 \pm 2.4$	
Cr <sup>3+</sup>	1000	$96.0\pm1.0$	95.3 ± 1.1	95.3 ± 1.1	
$Pt^{2+}$	100	$94.6\pm2.0$	$93.7\pm2.0$	95.6 ± 2.1	
$\mathrm{Hg}^{2+}$	100	97.6 ± 3.1	94.1 ± 2.2	$96.8 \pm 2.0$	
Br	5000	96.7 ± 1.8	97.2 ± 2.0	95.2 ± 2.6	
NO <sub>3</sub> -	2000	$97.0\pm1.5$	$98.8\pm2.0$	$98.2 \pm 2.0$	
CrO <sub>4</sub> <sup>2-</sup>	100	$94.1\pm3.5$	$95.9\pm2.3$	95.0 ± 1.6	
AsO <sub>4</sub> <sup>3-</sup>	100	$96.0 \pm 1.7$	95.5 ± 2.0	94.1±2.4	

## Table 2S

Comparison of the proposed method with other methods applied for the extraction and determination of target ions.

Detection system	Sorbent	Analytes	LOD a	LDR	Sorption time (min)	SC <sup>b</sup>	RSD (%)	Ref.
ICP-OES <sup>c</sup>	Polythiophene-coated Fe <sub>3</sub> O <sub>4</sub> NPs	Ag, Au, Pd	0.2-2.0	0.75-100	12	-	4.2	[1]
FAAS	SBA <sup>d</sup> -15-NH <sub>2</sub>	Ag	4.0	-	-	137	1.7	[3]
FI <sup>d</sup> -FAAS	Silica gel based chelating sorbent	Ag, Au, Pd	1.3-21	-	40	24.5-50.9	3.0	[26]
FI-FAAS	Carboxylated pillar[5]arene	Au, Pd	15.9-16	50-1000	-	12-16	< 0.8	[27]
FAAS	Fe <sub>3</sub> O <sub>4</sub> @murexide NPs	Ag, Au, Pd	0.15-0.6	0.5-400	5.0	34-50	5.0-7.5	This work

 $a \mu g L^{-1}$ 

<sup>b</sup> Sorption capacity (mg g<sup>-1</sup>)

<sup>c</sup> Inductively coupled plasma optical emission spectrometry

<sup>d</sup> Santa Barbara Amorphous type material

<sup>e</sup>Flow injection

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