

Different sample preparation methods for analysis of suspension fertilizers combining LIBS and liquid-to-solid matrix conversion: determination of essential and toxic elements

Daniel Fernandes Andrade, Marco Aurélio Sperança, Edenir Rodrigues Pereira-Filho\*

Group of Applied Instrumental Analysis, Department of Chemistry, Federal University of São Carlos, Zip Code 13565-905, São Carlos, São Paulo State, Brazil

\*Corresponding author: [erpf@ufscar.br](mailto:erpf@ufscar.br)

Phone number: +55 16 3351-8092

Fax number: +55 16 3351-8350

## Supplementary material

**Table 1S** Instrumental parameters for ICP OES analysis.

<b>Instrument parameter</b>	<b>Operational conditions</b>
Integration time (s)	15 s for low and 5 s for high wavelengths
Sample introduction flow rate (mL min <sup>-1</sup> )	2.1
Pump stabilization time (s)	5
Radio frequency (RF) applied power (kW)	1.15
Argon gas flow rate (L min <sup>-1</sup> )	12
Auxiliary gas flow rate (L min <sup>-1</sup> )	0.50
Nebulizer gas flow rate (L min <sup>-1</sup> )	0.70
Viewing mode	Radial
Elements and wavelengths (nm), I and II are atomic and ionic lines	B I 249.773, Ca II 317.933, Cu I 324.754, Fe II 240.488, K I 766.490, Mg I 285.213, Mn II 294.920, Na I 589.592, P I 178.284 and Zn I 481.053

**Table 2S** Operating parameters used in ICP-MS measurements.

<b>Instrumental parameter</b>	<b>Operational conditions</b>
Radio frequency generator (MHz)	27
Radio frequency (RF) applied power (kW)	1.55
Argon gas flow rate (L min <sup>-1</sup> )	14.0
Auxiliary gas flow rate (L min <sup>-1</sup> )	0.8
Nebulizer gas flow rate (L min <sup>-1</sup> )	1.18
Sampling depth (mm)	5.0
Integration time (s)	3.0
Sampling cone (mm)	Nickel 0.8
Skimmer (mm)	Nickel 1.2
Nebulizer	Glass concentric (MicroMist)
Spray chamber	Cyclonic
Replicates	3
Mass/charge ratios monitored	<sup>52</sup> Cr, <sup>75</sup> As, <sup>111</sup> Cd and <sup>208</sup> Pb

**Table 3S** Concentrations (average, n = 3) and recoveries (%) for samples S1, S2 and S4 spiked with a standard solution of selected elements (Ba, Ca, Cu, Fe, K, Mg, Mn, Na, P and Zn) by ICP OES.

Emission line (nm)	Added (mg L <sup>-1</sup> )	Found (mg L <sup>-1</sup> )			Recovery (%)		
		S1	S2	S4	S1	S2	S4
B I 249.773	50	47	46	41	95	92	82
Ca II 317.933	40	35	33	41	88	83	102
Cu I 324.754	80	86	91	71	107	114	88
Fe II 240.488	20	20	23	17	100	114	83
K I 766.490	6.0	6.6	6.4	6.8	111	107	113
Mg I 285.213	25	29	32	24	116	127	98
Mn II 294.920	30	28	30	35	94	100	116
Na I 589.592	4.0	3.3	4.8	3.6	82	119	91
P I 178.284	10	10	9	11	100	88	106
Zn I 481.053	40	37	49	46	94	124	115

**Table 4S** Concentrations (average, n = 3) and recoveries (%) for samples S2, S3 and S7 spiked with a 20 µg L<sup>-1</sup> standard solution of selected elements (As, Cd, Cr and Pb) by ICP-MS.

Isotope	Found (µg L <sup>-1</sup> )			Recovery (%)		
	S2	S3	S7	S2	S3	S7
<sup>75</sup> As	17	16	20	84	80	98
<sup>111</sup> Cd	17	17	16	86	84	80
<sup>52</sup> Cr	19	20	22	94	99	112
<sup>208</sup> Pb	17	18	15	87	90	76

## Figure caption

**Figure 1S** Emission signal intensities of (a) Cu, (b) K, (c) Mg, (d) Mn and (e) Zn preliminarily monitored in immobilized standard solution prepared for Doehlert design experiments (experiment 1 up to 15, optimization of LIBS system).

**Figure 2S** LIBS spectra and the identification of the main signals monitored for the eight samples studied: (a) S1, (b) S2, (c) S3, (d) S4, (e) S5, (f) S6, (g) S7 and (h) S8.

**Figure 3S** Study of polymer stability for the 5 analytes determined: (a) Cu, (b) K, (c) Mg, (d) Mn and (e) Zn.

Figure 1Sa

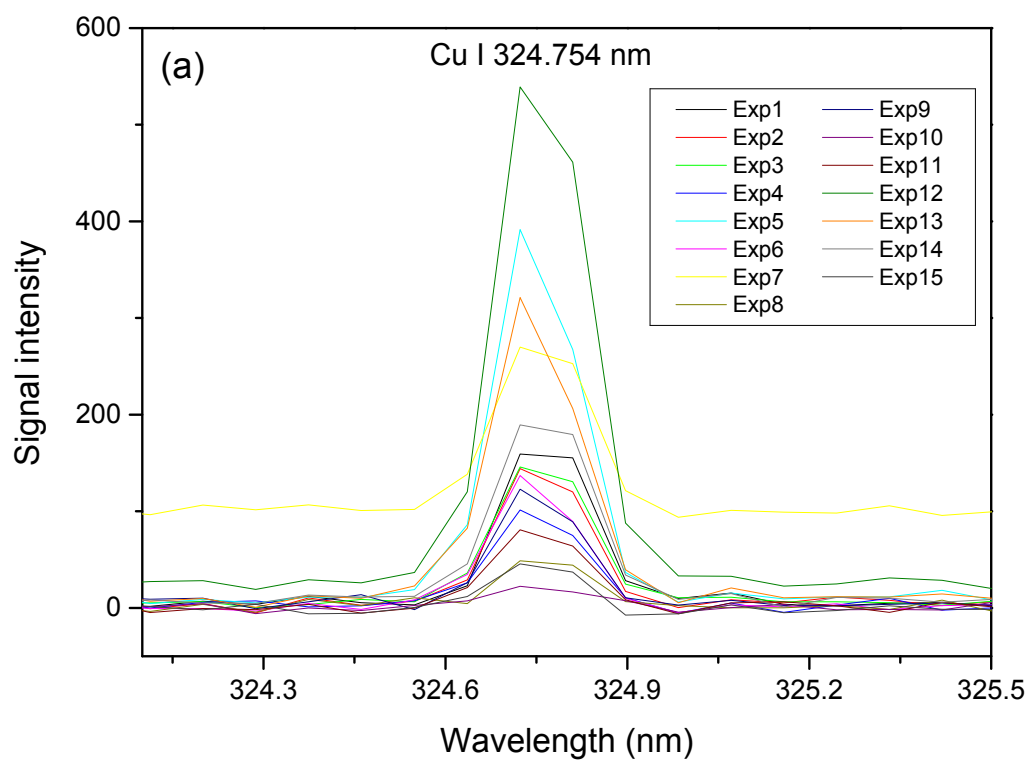


Figure 1Sb

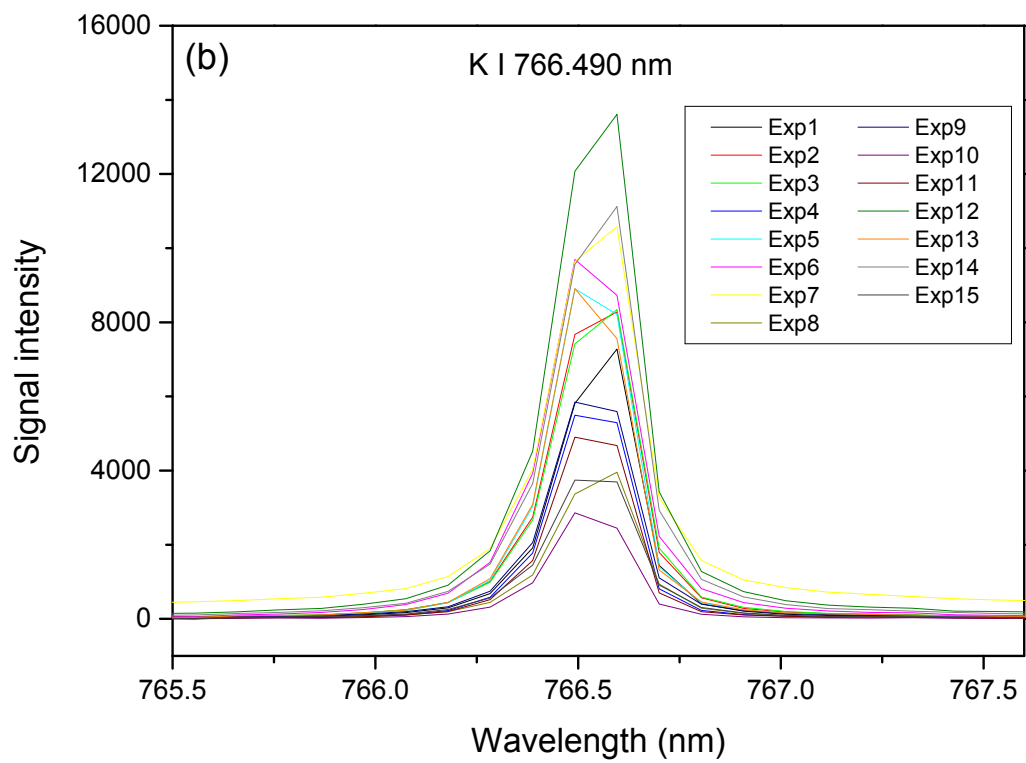




Figure 1Sc

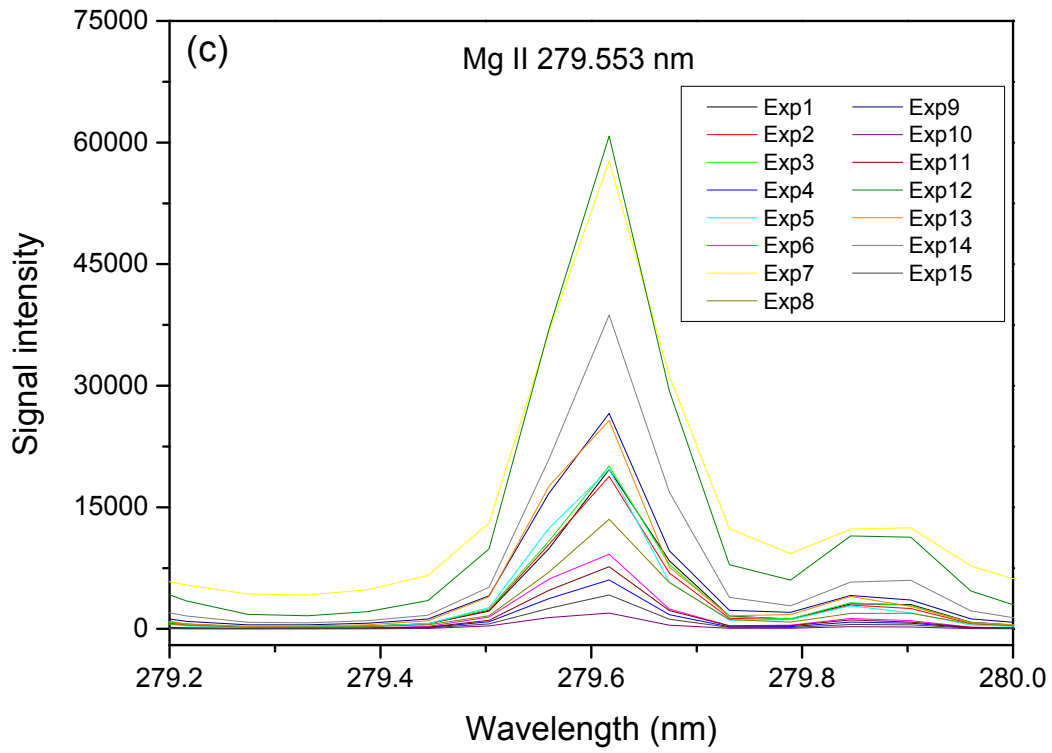


Figure 1Sd

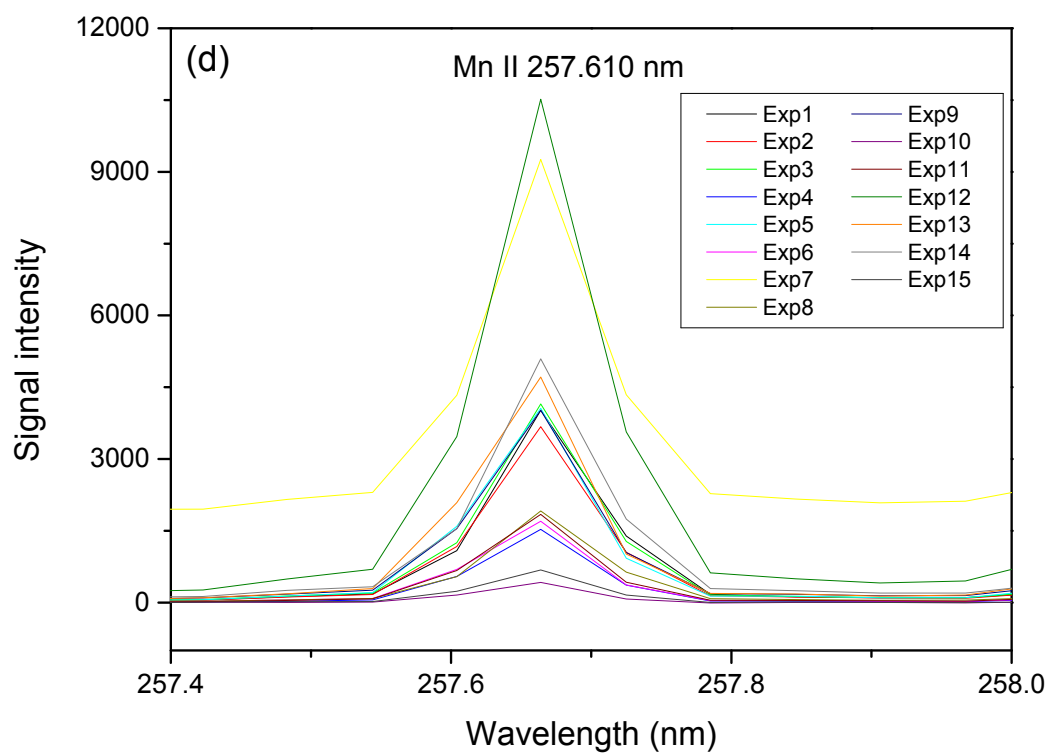


Figure 1Se

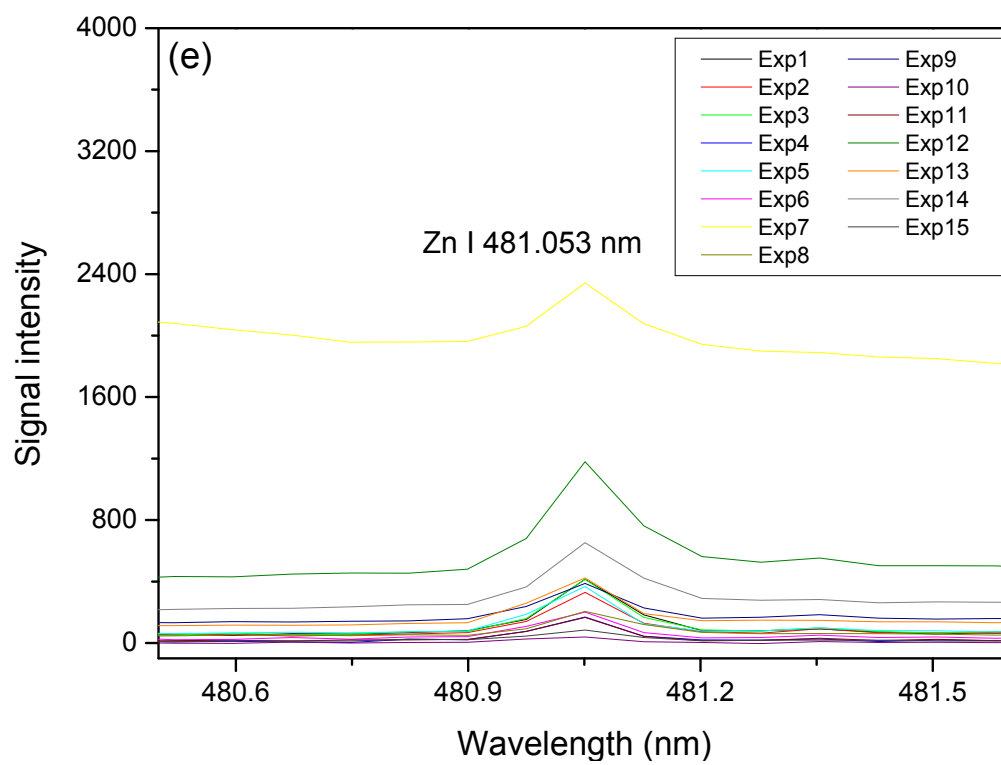


Figure 2Sa

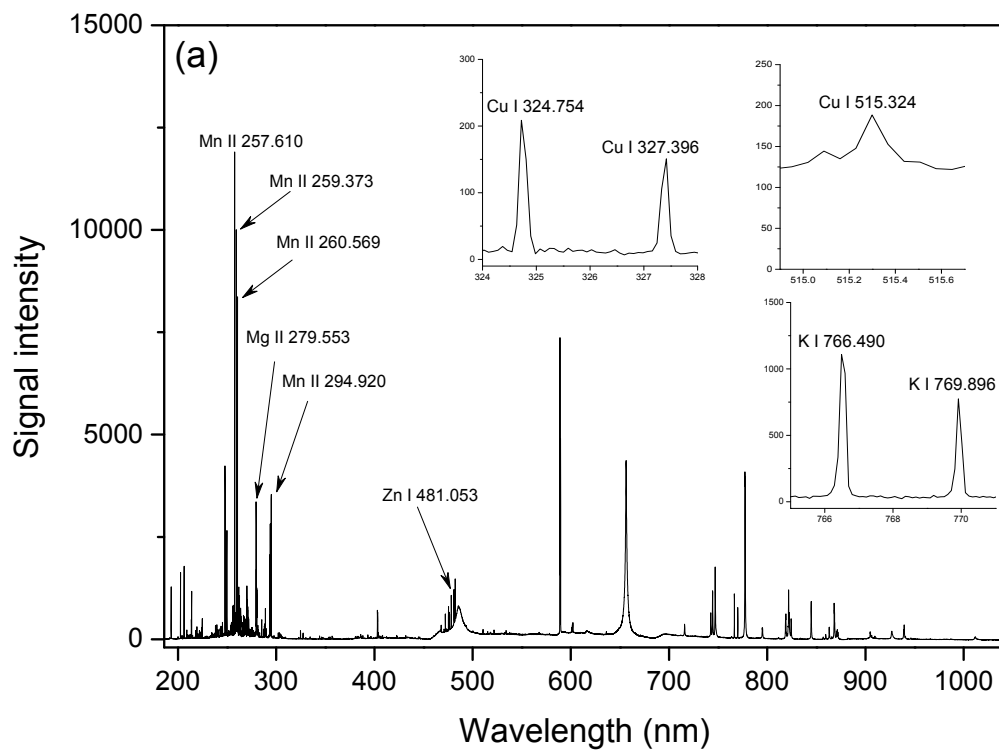


Figure 2Sb

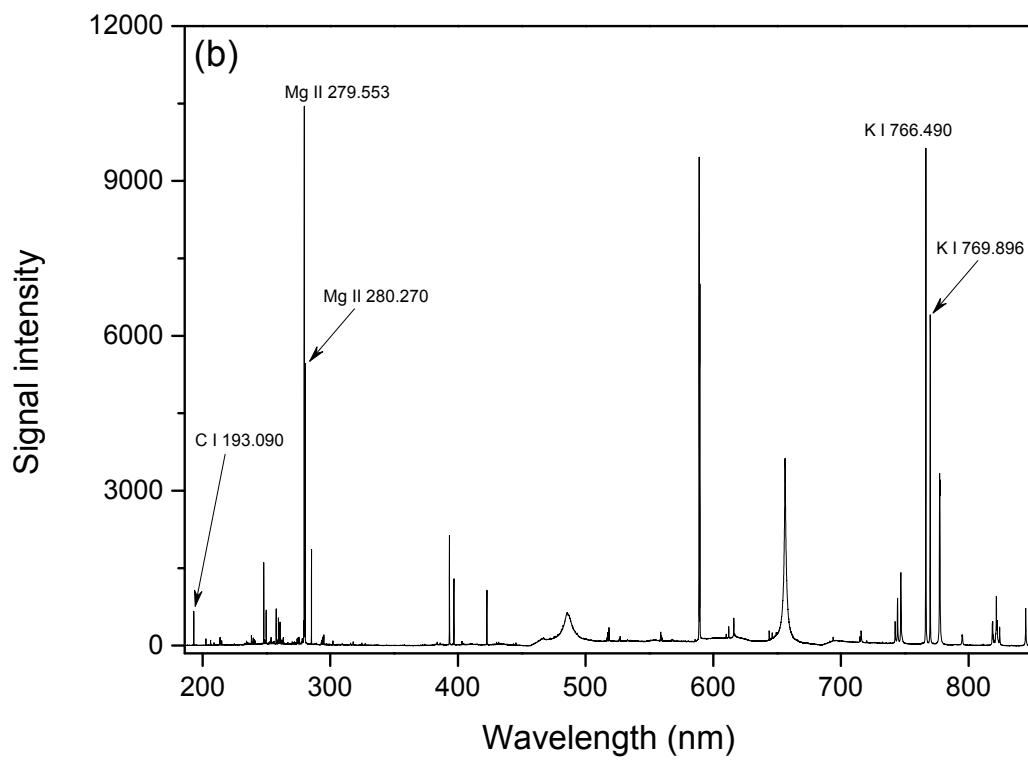


Figure 2Sc

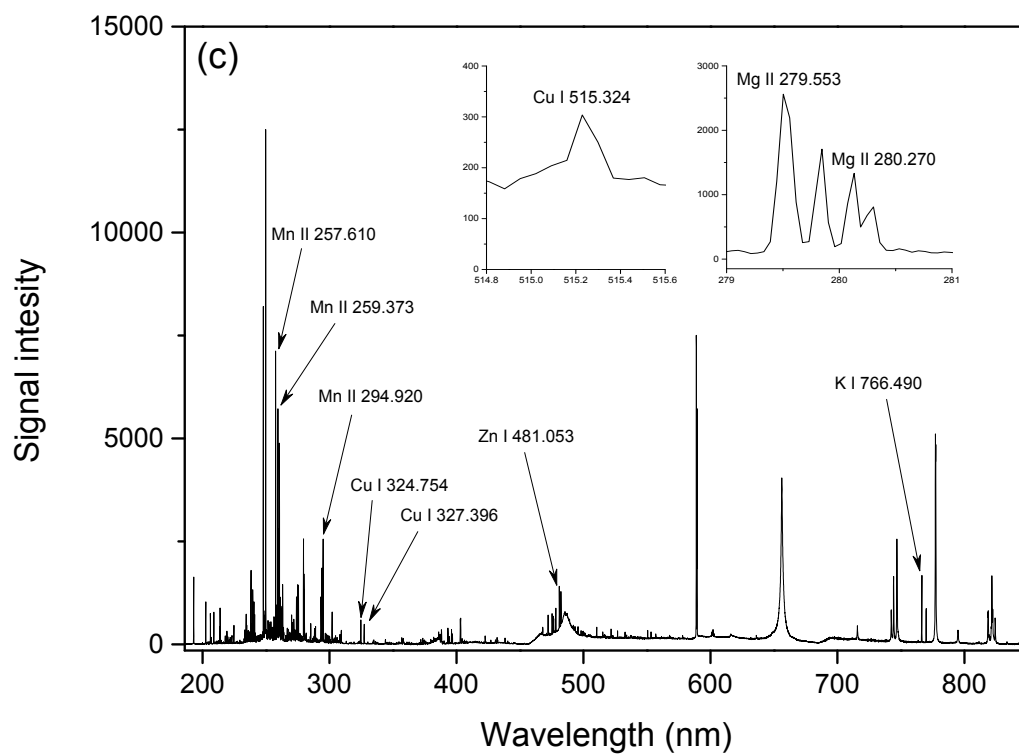


Figure 2Sd

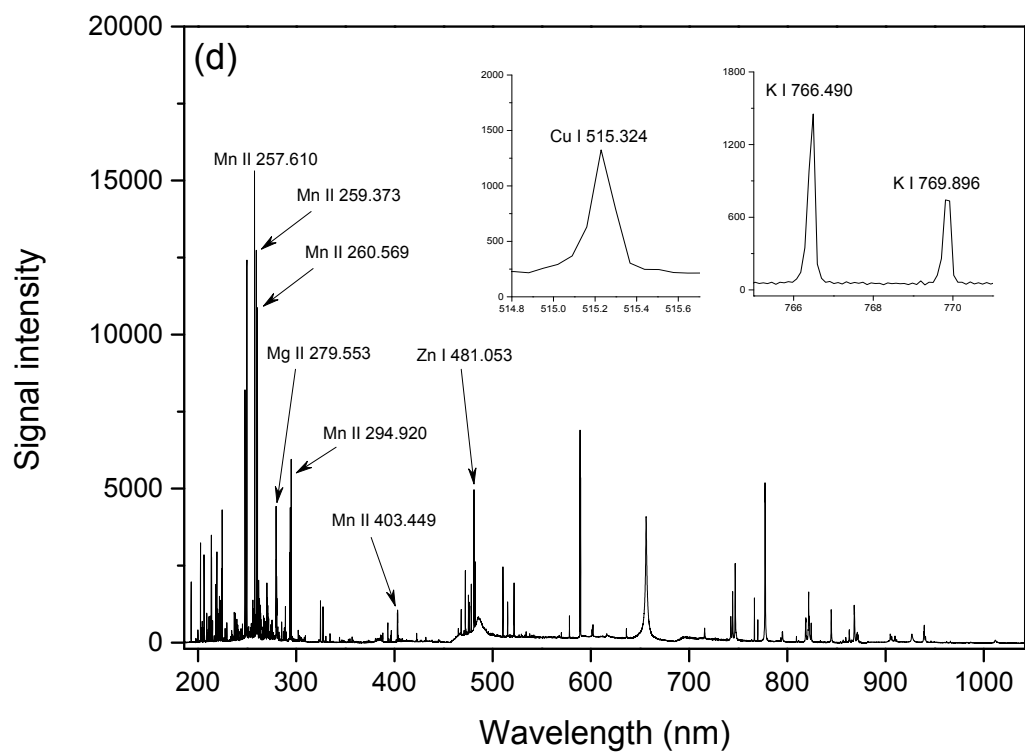


Figure 2Se

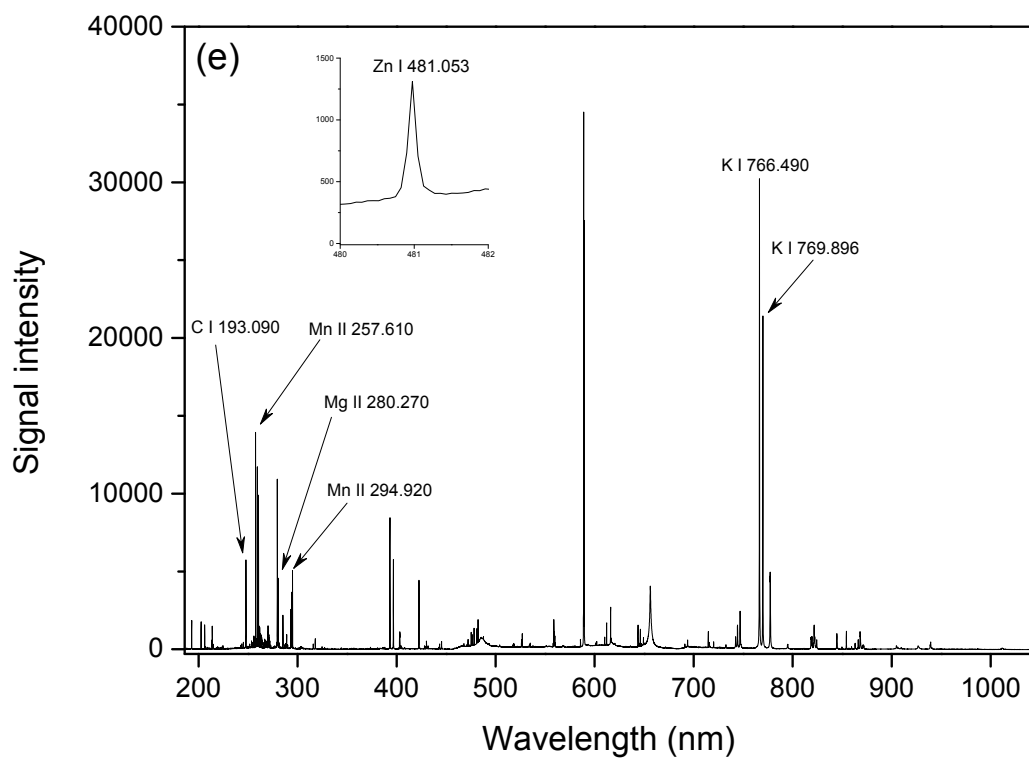




Figure 2Sf

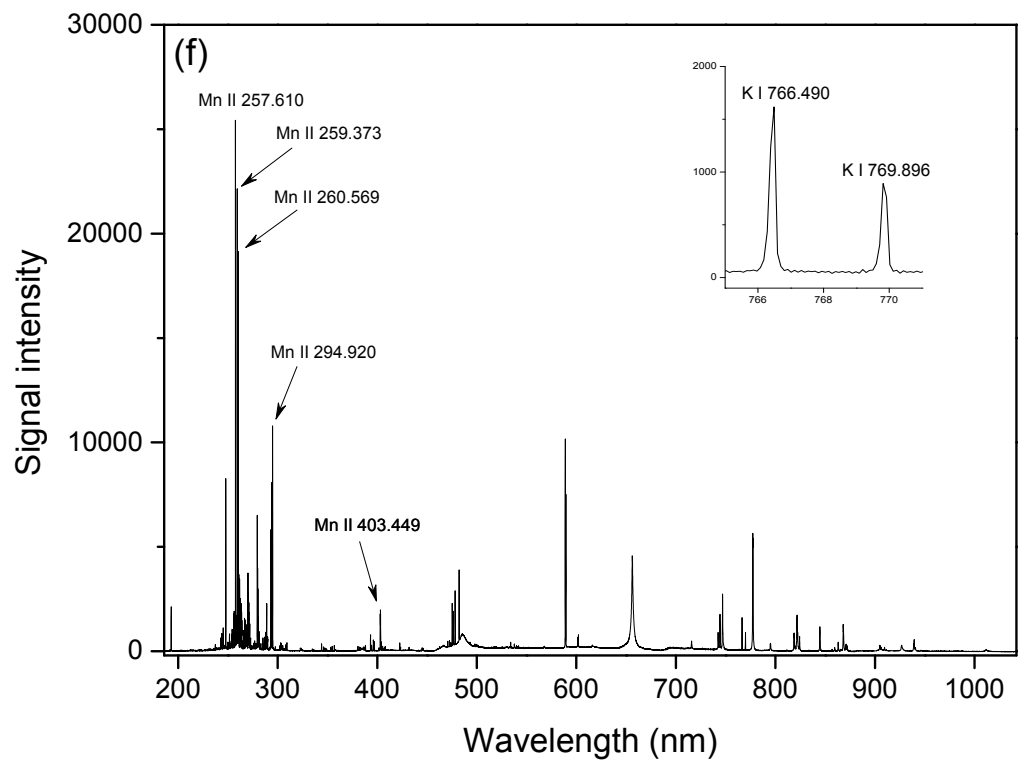


Figure 2Sg

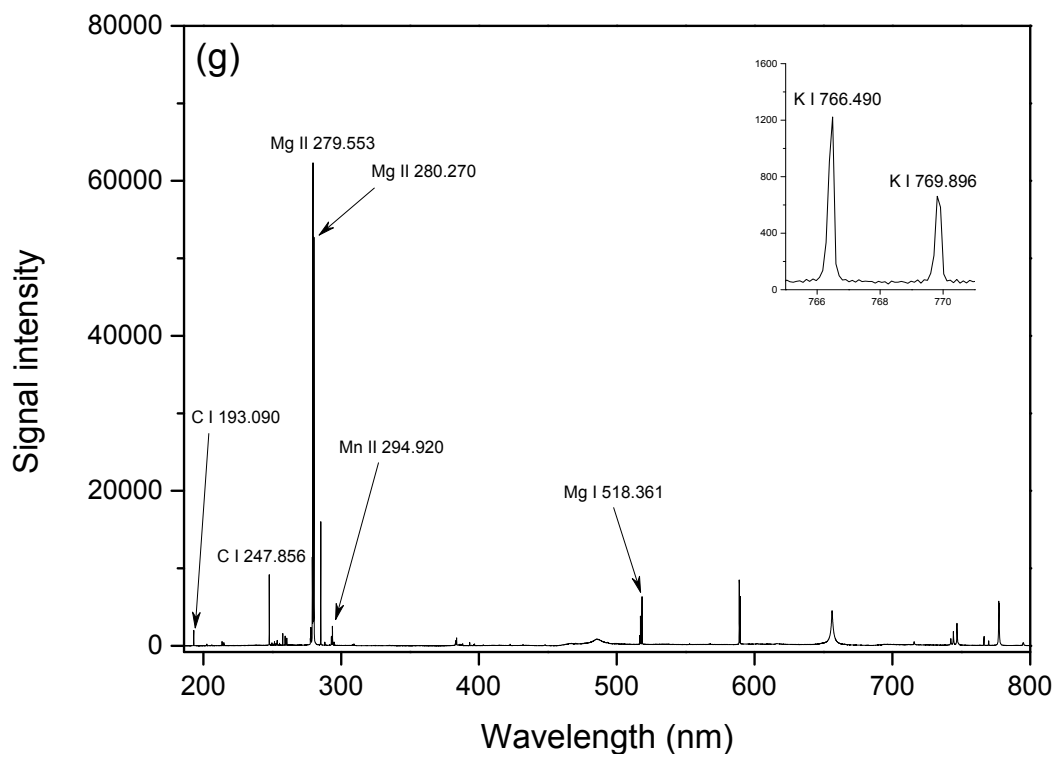


Figure 2Sh

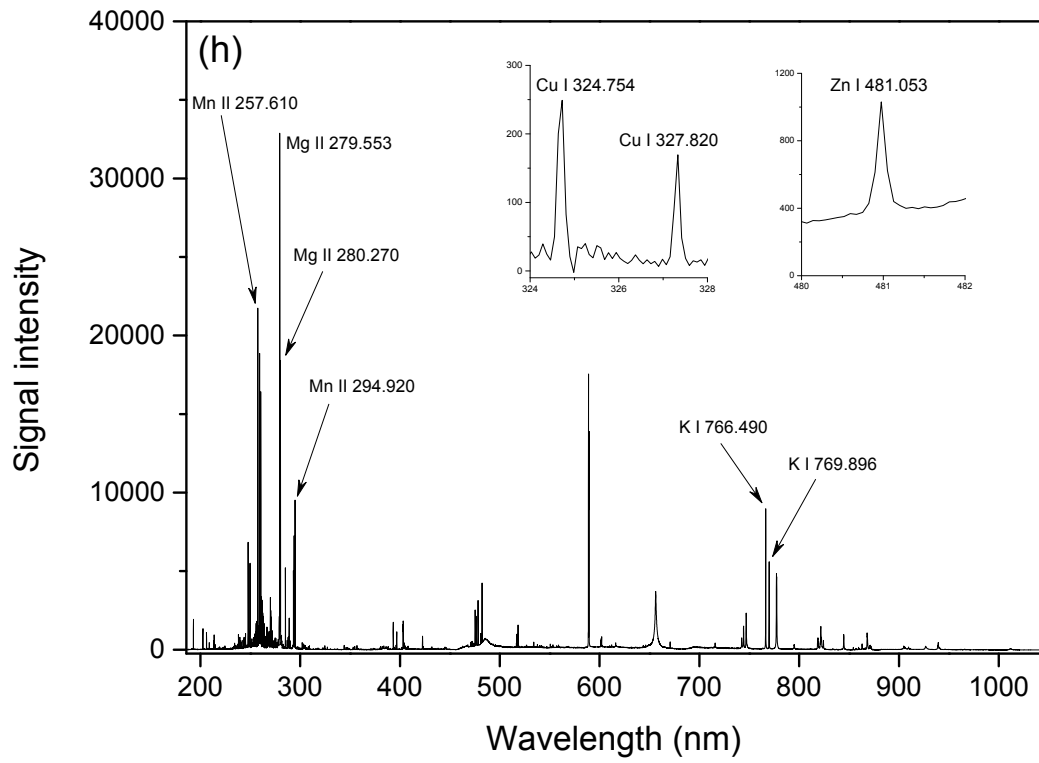


Figure 3S

