

Supporting Information

Manuscript title

Glutathione-S-Transferase catalyzed reaction of glutathione for electrochemical biosensing of Temephos, Fenobucarb and Dimethoate

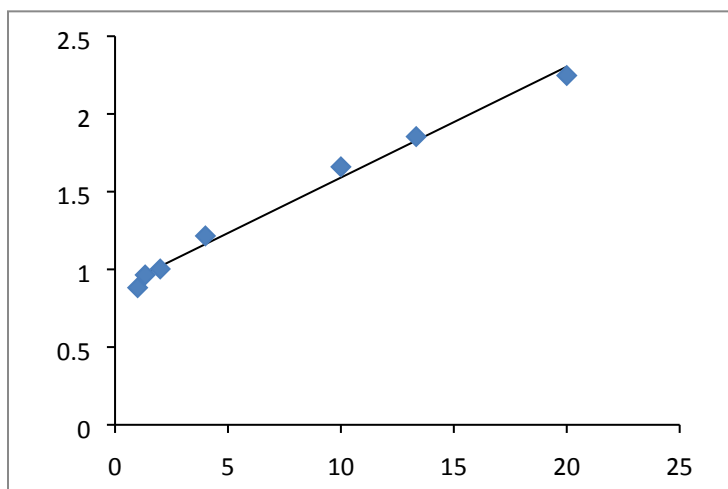
Authors

Himadri Borah^a, Rekha Rani Dutta^a, Sudarsan Gogoi^a, Tapas Medhi^b and Panchanan Puzari^{a*}

^aDepartment of Chemical Sciences, Tezpur University, Tezpur, Assam, India-784028

^b Department of Molecular Biology and Biotechnology, Tezpur University, Tezpur, Assam, India-784028

K_m and V_{max} for GSH-CDNB-GST (without inhibitor):



$$K_m = \text{slope}/\text{intercept} = 0.071/0.876 = 0.08$$

$$V_{max} = K_m/\text{slope} = 0.08/0.071 = 1.14$$

K_i calculation for Fenobucarb:

K_m 60 ppb= 0.219/0.875 = 0.2502

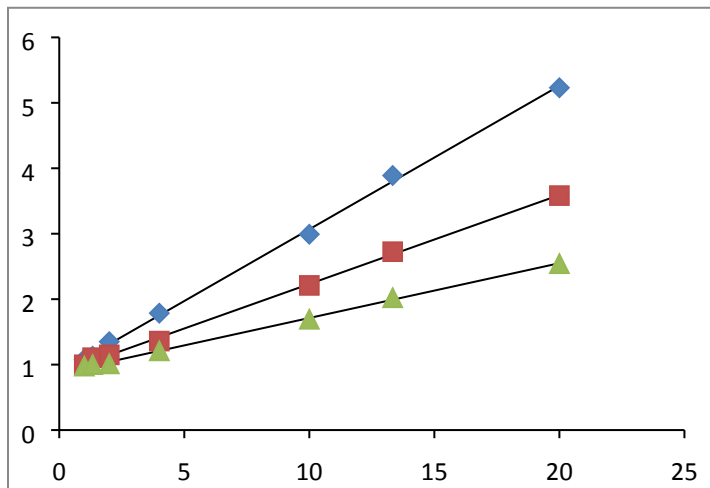
V_{max} = 0.2502/0.219 = 1.14

K_m 50 ppb= 0.135/0.872 = 0.1548

V_{max} = 0.1548/0.135= 1.14

K_m 40ppb = 0.083/0.875= 0.0948

V_{max} = 0.0948/0.083= 1.14



K_m (inhibitor) / K_m (without inhibitor) vs [I] = Slope = 1/ K_i

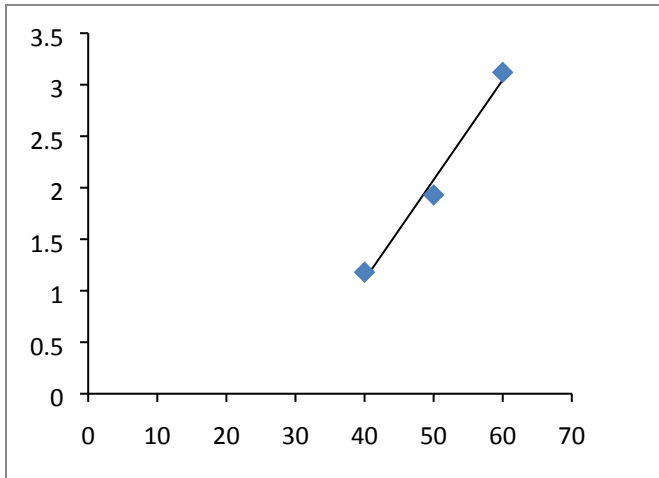
K_m (without inhibitor) = 0.08

$K_{60\text{ppb}}/K_m = 0.2502/0.08 = 3.12$

$K_{50\text{ppb}}/K_m = 0.1548/0.08 = 1.93$

$K_{40\text{ppb}}/K_m = 0.0948/ 0.08 = 1.18$

Concentration of fenobucarb	K_m value for each concentration
60 ppb	3.12
50 ppb	1.93
40 ppb	1.18



Slope = $1/K_i = 0.097$

Therefore, $K_i = 1/0.097 = 10.30$

K_i calculation for Temephos:

K_m 60 ppb = $0.104/1.278 = 0.0813$

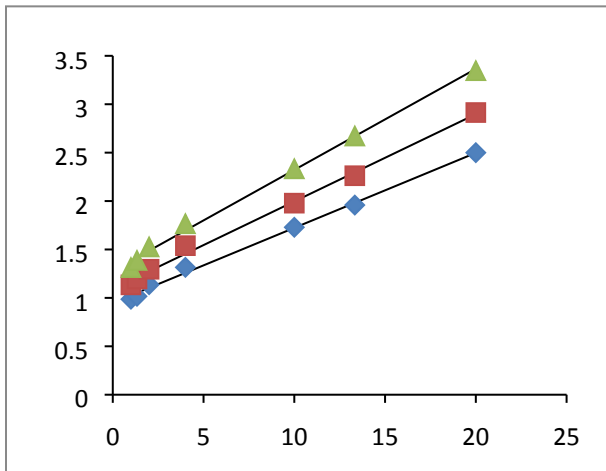
$V_{max} = 0.0813/0.104 = 0.7817$

K_m 50ppb = $0.090/1.094 = 0.0822$

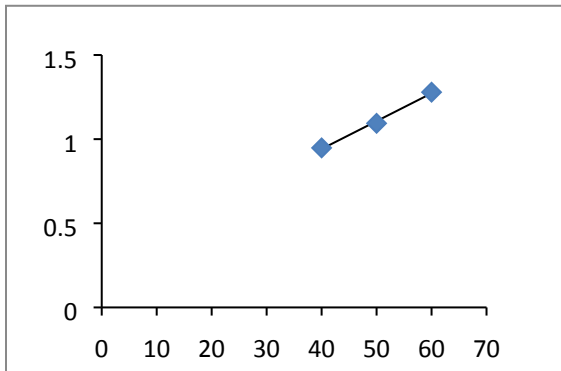
$V_{max} = 0.0822/0.090 = 0.9133$

K_m 40 ppb = $0.077/0.948 = 0.0812$

$V_{max} = 0.0812/0.077 = 1.054$



Concentration of temephos	$1/V_{max}$ value for each concentration
60 ppb	1.279
50 ppb	1.094
40 ppb	0.948



$$K_i = 1/(\text{slope} \times V_{\max}) = 1/(0.016 \times 1.14) = 54.82$$

[V_{\max} (without inhibitor) = 1.14]

K_i calculation for Dimethoate:

$$K_m \text{ 60 ppb} = 0.139/0.231 = 0.6017$$

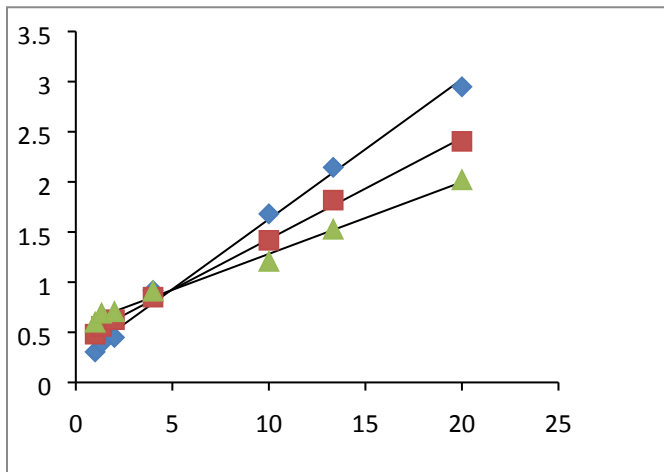
$$V_{\max} \text{ 60 ppb} = 0.6017/0.139 = 4.32$$

$$K_m \text{ 50 ppb} = 0.1/0.42 = 0.2380$$

$$V_{\max} \text{ 50 ppb} = 0.2380/0.1 = 2.38$$

$$K_m \text{ 40 ppb} = 0.071/0.564 = 0.1258$$

$$V_{\max} \text{ 40 ppb} = 0.1258/0.071 = 1.77$$



Comparison of the method with other methods

Table S1 shows a comparison of our method with the other methods available in literature. The table reveals that the detection limits of all the four pesticides obtained by our method is within the range of those reported methods. It is evident from the table that for the two pesticides fenobucarb and temephos no reliable biosensing method other than our one available in literature. It is the most promising aspect of the present method.

Table T₁: Comparison of the present method with other methods

Sl. No.	Pesticide	Method of detection	Limit of detection	Linear range	Reference
1.	Fenobucarb	Flow Amperometry	2 ppb	-	54
2.	-do-	Liquid chromatography– tandem mass spectrometry	1.5 ppb	-	55
3.	-do-	Gas chromatography/Mass spectrometry	4 ppb	-	56
4.	-do-	GST biosensor	2 ppb	2-50 ppb	This work
5.	Temephos	Liquid Chromatography/Diode Array	9 ppb	-	57
6.	-do-	-do-	0.02-0.2 ppb	-	58
7.	-do-	Liquid chromatography–ion trap- mass spectrometry		-	59
8.	-do-	Thermospray liquid chromatography- mass spectrometry	0.038 ppb	-	60
9.	-do-	GST biosensor	4 ppb	4-30 ppb	This work
10.	Dimethoate	Acetylcholinesterase biosensor	0.4 ppb	2.29-5.5x10 ² ppb	61
11.	-do-	-do-	2.29 ppb	11.45-22 ppb	62
12.	-do-	-do-	389 ppb		63
13.	-do-	GST biosensor	5 ppb	5-25 ppb	This work
14.	* Cypermethrin	Surface plasmon resonance	0.5 ppb	-	64
15.	-do-	Nucleic acid sensor	2.5 ppb	2.5–2x10 ³ ppb	65
16.	-do-	Enzyme-linked immunosorbent assay	0.5 ppb	-	66
17.	-do-	DNA biosensor	2.5 ppb	-	67
18.	-do-	*GST biosensor	2 ppb	2-25 ppb	53

*Cypermethrin was analyzed by the same method in our previous work (Reference 53)

