

Supporting Information for:

CaCO₃ nanoparticles as the tags for potentiometric detection of cardiac biomarker using calcium ion-selective electrode

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Experimental

S1.1. Reagents and chemicals

Monoclonal mouse anti-human myoglobin capture antibody (designated as mAb₁, 0.29 mg mL⁻¹ in PBS containing 0.09% sodium azide, w/v), polyclonal rabbit anti-human myoglobin detection antibody (designated as pAb₂, application: 1 : 500 – 1 : 1000), myoglobin (Mb) standards, myc-oncogene (MYC), squamous cell carcinoma antigen (SCCA), alpha-fetoprotein (AFP), human IgG, carcinoembryonic antigen (CEA), thyroid-stimulating hormone (TSH), troponin T (TPT), troponin I (TPI), CK-MBmass (CKMB) and human myoglobin ELISA kit were purchased from Abcam (Shanghai, China). N-(3-Dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC), N-hydroxysuccinimide (NHS), thioglycolic acid, bovine serum albumin (BSA, 96-99%) and chitosan (high purity, MW 60,000 – 120,000) were acquired from Sigma-Aldrich. All other chemicals were of analytical grade and used as received. Ultrapure water obtained from a Millipore water purification system at ~18.2 MΩ·cm⁻¹ (Milli-Q, Millipore) was used in all runs. A pH 9.6 carbonate buffer (1.69 g Na₂CO₃ and 2.86 g NaHCO₃), a pH 9.8 carbonate buffer (2.33 g Na₂CO₃ and 2.35 g NaHCO₃), and a pH 7.4 phosphate-buffered saline (PBS, 0.01 M) (2.9 g Na₂HPO₄·12H₂O, 0.24 g KH₂PO₄, 0.2 g KCl and 8.0 g NaCl) were prepared by adding the corresponding chemicals in 1000-mL distilled water, respectively. The blocking buffer and washing buffer were obtained by adding 1.0 wt% BSA and 0.05% Tween 20 (v/v) in PBS, respectively.

S1.2. Preparation of BSA-functionalized CaCO₃ nanospheres (BSA-CaCO₃)

BSA-functionalized CaCO₃ nanoparticles (denoted as BSA-CaCO₃) are prepared similar to the literature [J. Zhou, J. Tang, G. Chen and D. Tang, *Biosens. Bioelectron.*, 2014, **54**, 323-328]. Briefly, CaCl₂ aqueous solution (5.0 mL, 50 mM) was initially added into BSA aqueous solution (10 mL, 3.0 mg mL⁻¹) under vigorous stirring at room temperature (note: During this process, the positively charged Ca²⁺ ions were adsorbed onto the negatively charged BSA molecules because the isoelectronic point of BSA is ~4.7. Meanwhile, partial Ca²⁺ ions could be chelated onto the BSA by the residual –COOH groups of BSA molecules). Following that, 5.0 mL of 50 mM Na₂CO₃ aqueous solution was dropped into the resulting mixture under the same conditions. After reaction, the resultant suspension was aged overnight under ambient conditions. Subsequently, the suspension was separated and purified by centrifugation for 15 min at 10,000g. Finally, the obtained pellets (i.e.,

BSA-CaCO₃) were used for further use.

S1.3. Conjugation of BSA-CaCO₃ with pAb₂ antibody (pAb₂-CaCO₃)

Polyclonal rabbit anti-human myoglobin detection (pAb₂) were conjugated with BSA-CaCO₃ nanospheres (denoted as pAb₂-CaCO₃) through a typical carbodiimide coupling [W. Lai, J. Zhuang, J. Tang, G. Chen and D. Tang, *Microchim. Acta*, 2012, **178**, 357-365], schematically illustrated in Fig. S1. Initially, BSA-CaCO₃ nanospheres (25 mg) were dispersed into 1.0 mL of ultrapure water. Then, NHS (11 mg) and EDC (15 mg) were thrown in the resultant mixture under gentle stirring for 12 h min at 4 °C (note: The -COOH groups of BSA molecules were activated during this process). Following that, 500 μL of 1.0 mg mL⁻¹ pAb₂ antibody (dispersed in 10 mM PBS, pH 7.4) was added into the mixture with gentle stirring (~150 rpm), and left overnight at 4 °C. After completion of the reaction, the conjugates were centrifuged for 10 min at 8000 g. Finally, the obtained pAb₂-CaCO₃ nanoconjugates were washed with pH 7.4 PBS several times, and dispersed into pH 7.4 PBS (1.0 mL, 10 mM) containing 1.0 wt % BSA and 0.1 wt % sodium azide, with a final concentration of ~25 mg mL⁻¹, and stored at 4 °C until use.

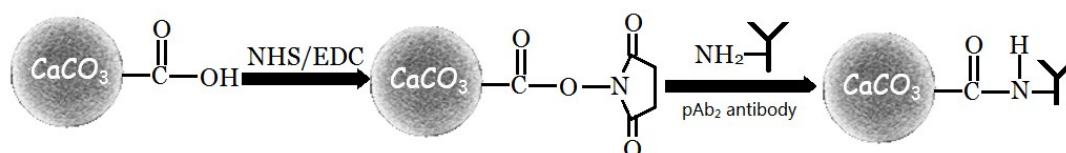


Fig. S1 Schematic diagram of BSA-CaCO₃ conjugation with pAb₂ antibody.

S1.4. Potentiometric measurement with Ca²⁺ ion-selective electrode

Fig. 1 represents potentiometric immunoassay protocol and measurement method. A high-binding polystyrene 96-well microplate was coated overnight at 4 °C with mAb₁ (50 μL per well, 10 μg mL⁻¹) in 0.05 M sodium carbonate buffer (pH 9.6). The microplate was covered with adhesive plastics plate sealing film. On the following day, the microplate was washed three times with washing buffer, and then incubated with 300 μL per well of blocking buffer for 60 min at 37 °C with shaking. The microplate was then washed as before. Following that, 50 μL of myoglobin standards in pH 7.4 PBS was added into the microplate, and incubated for 60 min at 37 °C under shaking. After washing, 50 μL of pAb₂-CaCO₃ solution was added into the well and incubated for 60 min at 37 °C with shaking. The microplate was washed three times with washing buffer and one

time with water. Afterward, a 50- μ L aliquot of 0.1 M HCl was added into each well, and reacted for 10 min (optimized) at room temperature. The resulting solution containing the released Ca^{2+} ions was transferred into 2.0 mL of 1.0 mM NaNO_3 solution for potentiometric measurement by using Ca^{2+} ion-selective electrode (Ca-ISE) (perfectIONTM, Mettler-Toledo, Switzerland). The obtained signal was registered as the sensing response to target myoglobin. According to the Nernstian equation, the reaction temperature will affect the potentiometric response of the Ca-ISE. To consider the possible application in future, however, all potentiometric measurements were carried out at room temperature (25 ± 1.0 °C).

Table S1 Comparison of the assayed results by using the developed potentiometric immunoassay and commercial human myoglobin ELISA kit for human serum samples

Sample no.	Method; concentration [ng mL^{-1} , mean \pm SD, n = 3]		
	Potentiometric immunoassay	ELISA kit	RSD (%) ^a
1	23.4 \pm 2.12	22.2 \pm 1.97	3.72
2	0.67 \pm 0.05	0.72 \pm 0.06	5.09
3	12.3 \pm 1.03	12.4 \pm 1.01	0.57
4	34.7 \pm 3.34	32.1 \pm 2.98	5.51
5	68.4 \pm 6.43	70.2 \pm 5.53	1.84
6	19.7 \pm 1.65	19.5 \pm 1.42	0.72

^aRSD was calculated according to the individual average value of two methods per sample.