## **Supporting Information for**

# A Nile Red-based Near-infrared Fluorescent Probe for Endogenous Hydrogen

## **Polysulfides in Living Cells**

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#### **S1. Additional Figures**



Fig. S1. HR-MS of the product of KB1+Na<sub>2</sub>S<sub>2</sub>



Fig. S2. Fluorescence spectra of compound 1 (5  $\mu$ M) and KB1+Na<sub>2</sub>S<sub>2</sub> (5  $\mu$ M)



**Fig. S3.** Cell viability of (a) MCF-7 and (b) Hela cells in the presence of **KB1** with indicated concentration determined by MTS

#### S2. Generation of reactive nitrogen and oxygen species (RNS / ROS)

**ONOO**<sup>-</sup>: The synthesis of peroxynitrite involved nitrosation of  $H_2O_2$  at pH  $\ge$ 12.0 by isoamyl nitrite (CAS: 110-46-3).

**NO:** Nitric oxide was generated from SNP (Sodium Nitroferricyanide (III) Dihydrate, CAS: 13755-38-9).

CIO-: NaCIO was dissolved in deionized water.

**H<sub>2</sub>O<sub>2</sub>:** H<sub>2</sub>O<sub>2</sub> was dissolved in deionized water.

**•OH:** Hydroxyl radicals were generated by the addition of  $Fe^{2+}$  and  $H_2O_2$  in 10 mM HEPES buffer (pH 7.4).

<sup>1</sup>**O**<sub>2</sub>: <sup>1</sup>**O**<sub>2</sub> was generated by hydrogen peroxide/sodium hypochlorite system.

 $O_2^-: O_2^-$  was generated by xanthine and xanthine oxidase. Xanthine oxidase was added first. After xanthine oxidase was dissolved, xanthine was added and the mixtures were stirred at 25 °C for 1 h. S3. Original spectral copy of new compounds



#### HR-ESI-MS of KB1



#### S4. The procedure for the preparation of Na<sub>2</sub>S<sub>2</sub> and Na<sub>2</sub>S<sub>4</sub>

$$Na_2S + (n-1)S = Na_2S_n$$

To a solution of sodium sulfide (0.78 g, 0.01 mmol) and sulfur (0.32 g, 0.01 mmol) in distilled water (100 mL), the mixture was refluxed for 1h, then 0.1 mol/L of  $Na_2S_2$  solution was obtained.  $Na_2S_4$  solution was obtained by using the same method with 3 equivalents of sulfur.