

A Porous Porphyrin Organic Polymer (PPOP) for Visible Light Triggered Hydrogen Production

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Section S-1: General Information

General Remarks: Triformylphloroglucinol (**Tp**) was prepared from Phloroglucinol¹ and 5,10,15,20-Tetrakis(4-aminophenyl)-21H,23H-porphine (**Tph**) was synthesized by following the literature method². All other reagents and solvents were commercially available and used as received. Powder X-ray diffraction (PXRD) patterns were recorded on a Phillips PANalytical diffractometer for Cu K α radiation ($\lambda = 1.5406 \text{ \AA}$), with a scan speed of 1° min^{-1} and a step size of 0.02° in 2θ . Fourier transform infrared (FT-IR) spectra were taken on a Bruker Optics ALPHA-E spectrometer with a universal Zn-Se ATR (attenuated total reflection) accessory in the $600\text{-}4000 \text{ cm}^{-1}$ region or using a Diamond ATR (Golden Gate). Thermogravimetric analyses (TGA) were carried out on a TG50 analyzer (Mettler-Toledo) or a SDT Q600 TG-DTA analyzer under N₂ atmosphere at a heating rate of $10 \text{ }^\circ\text{C min}^{-1}$ within a temperature range of $30\text{-}900 \text{ }^\circ\text{C}$. SEM images were obtained with a Zeiss DSM 950 scanning electron microscope and FEI, QUANTA 200 3D Scanning Electron Microscope with tungsten filament as electron source operated at 10 kV. The samples were sputtered with Au (nano-sized film) prior to imaging by a SCD 040 Balzers Union. TEM images were recorded using FEI Tecnai G2 F20 X-TWIN TEM at an accelerating voltage of 200 kV. The TEM Samples were prepared by dropcasting the sample from isopropanol on copper grids TEM Window (TED PELLA, INC. 200 mesh). All gas adsorption experiments (up to 1 bar) were performed on a *Quantachrome Quadrasorb* automatic volumetric instrument. Solid state NMR (SSNMR) was taken in a Bruker 300 MHz NMR spectrometer. All measurements were carried out at room temperature on an (JobinYvon Horiba, France) using monochromatic radiation emitted by an AR-laser (514 nm) (NRS 1500 W) operating at 20 mW using 50x long distance objective. The experiment was repeated several times and at different positions to verify the consistency of the measurement. The samples were prepared simply by putting a drop of dispersion of the polymer materials in isopropanol on a clean piece of Silicon wafer.

Section S-2: Synthetic procedures

TpTph: A pyrex tube (o.d. \times i.d. = 10 \times 8 mm² and length 18 cm) is charged with Triformylphloroglucinol (**Tp**) (12.6 mg, 0.06 mmol) and tetra(*p*-amino-phenyl)porphyrin (**Tph**) (27.0 mg, 0.04 mmol) in presence of 6 M acetic acid (0.2 mL) using dichlorobenzene, tertiary butanol (1:1) as solvent combination (2 mL). This mixture was sonicated for few minutes in order to get a homogenous dispersion. The tube was then flash frozen at 77 K (liquid N₂ bath) and degassed by three freeze-pump-thaw cycles. The tube was sealed off and then heated at 120 °C for 3 days. After the reaction the material was filtered out, washed with ethanol and dried under vacuum at 150 °C for 12 hours to give purple colored powder in 85% (30 mg) isolated yield based on **Tph**. **FT-IR (powder):** 1617, 1591, 1572, 1454, 1287, 1180, 991, 964, 894, 796, 729 cm⁻¹.

Section S-3: Structure Modeling and Computational Details.

The calculation have been carried out with Turbomole 7.0^{3,4} using the SVP basis set⁵ and the PBE⁶ functional. Dispersion correction⁷ have been included in all the geometry optimization calculations. The resolution of identity (RI)⁸ along with the multipole accelerated RI (marij)⁹ approximations have been used for an accurate and efficient treatment of the electronic Coulomb term in the DFT calculations.

The modelled geometry has been optimized computationally. The optimized geometry is provided below. Specifically, four aldehyde linkers connected to the central porphyrin ring have been considered, with further porphyrin linkages at the 2, 4 and 6 positions of the aldehyde linker are taken into account. Further, continuation of this linked structure would lead to the experimentally obtained polymer. The current model therefore provides a truncated, optimized model for the real polymeric system, thereby providing insight into the nature of the 3D structure of the linked aldehyde geometry.

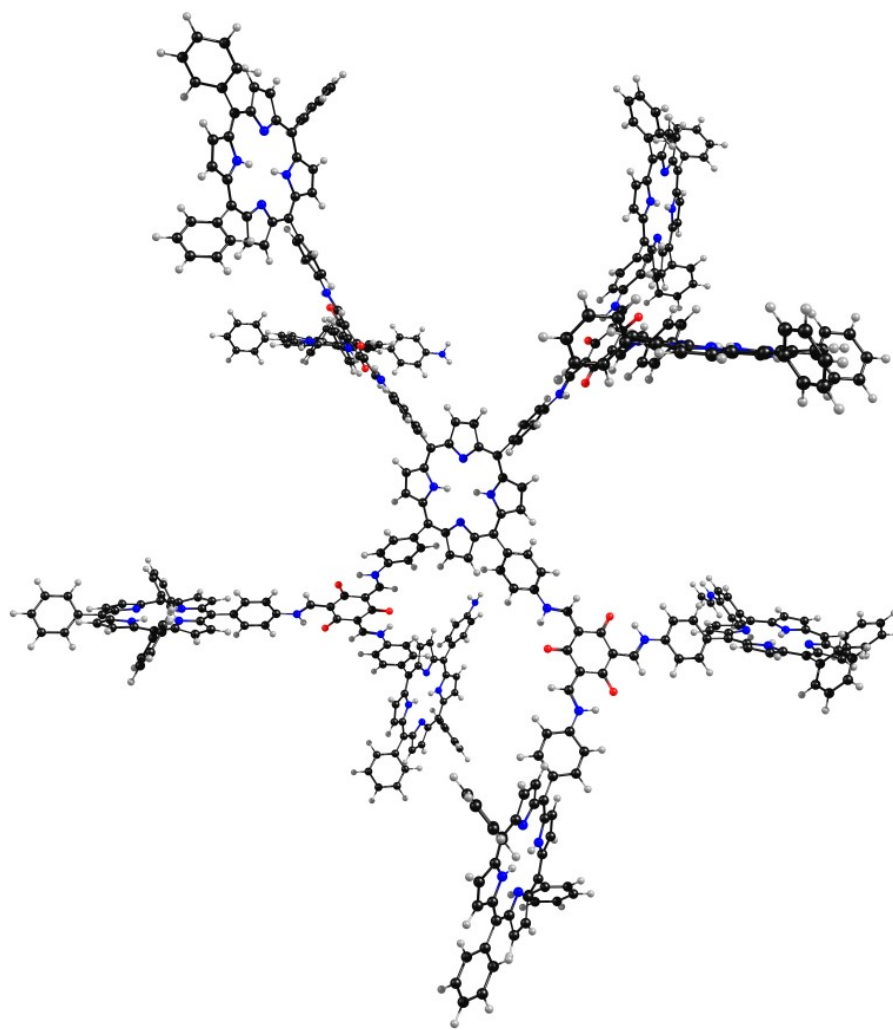


Figure S1: The optimized geometry having H-atom at the edge. The colour scheme is as follows: carbon: black, oxygen: red, nitrogen: blue, hydrogen: grey.

The optimized geometries of the structures reported in the manuscript (the atomic symbol followed by the three Cartesian coordinates, in Å).

C	23.570041	-10.029218	14.180546		C	3.701779	-17.040182	-8.378959	
C	23.536093	-8.625426	14.166991		C	3.722549	-17.059606	-9.745003	
C	22.788810	-7.947512	13.192646		C	4.899058	-17.845252	-10.123379	
C	22.062985	-8.660314	12.211705		N	5.560504	-18.310855	-9.015554	
C	22.106980	-10.072494	12.235969		C	5.264738	-18.073882	-11.477679	
C	22.853081	-10.749962	13.211639		C	6.319739	-18.912197	-11.907042	
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C	-3.340598	-2.636848	-0.907093		C	-11.840485	22.570732	-13.432025	
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H	-0.725292	0.786876	-0.917659		H	-11.529129	21.240183	-11.738756	
H	-4.620506	2.548343	-0.937001		H	-7.976196	23.122854	-13.345895	
H	-2.780202	4.534251	-0.894144		C	3.015492	-16.051514	-14.571209	
H	-2.442234	-4.657162	-0.945329		C	3.014505	-17.454039	-14.501016	
H	-4.420672	-2.816777	-0.887999		C	3.744504	-18.110796	-13.499190	
H	0.874436	-0.713511	-0.914802		C	4.487173	-17.376675	-12.547325	
C	20.094266	7.938086	-13.229090		C	4.479267	-15.966081	-12.628986	
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C	21.418225	9.093642	-11.376328		H	3.760890	-14.210430	-13.682780	
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C	21.388631	8.972095	-6.472699		C	16.047907	4.602590	-12.633676	
C	20.445171	8.008882	-6.045613		H	18.123332	2.565165	-15.239799	
C	20.012461	7.765787	-4.693329		H	19.290930	3.857950	-13.444152	
C	19.087897	6.740735	-4.721240		H	15.468362	5.182533	-11.899586	
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C	17.897109	4.971533	-7.946749		C	-3.590088	18.184021	-13.513099	
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C	17.142680	3.834169	-9.753352		C	-3.599987	15.383851	-13.656933	
C	17.124161	3.807761	-8.387345		H	-2.276242	18.121066	-15.239314	
C	18.156375	5.383028	-11.480095		H	-3.578329	19.282568	-13.446345	
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H	20.456004	8.531392	-14.074597		H	-3.611644	14.284403	-13.712002	
H	23.442157	10.722903	-7.514628		C	-15.977515	-3.090240	-14.590926	
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H	18.539516	6.316585	-3.874548		C	-17.304229	-4.549080	-12.559127	
H	20.358194	8.328644	-3.820769		C	-15.893660	-4.543249	-12.640607	
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H	16.659597	3.058574	-7.737450		H	-17.968328	-2.510819	-15.251244	
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C	-15.136418	4.009893	5.330155		C	23.382704	11.226452	-3.365178	
C	-16.545807	3.909731	5.369591		C	22.553300	11.879970	-4.291087	
C	-17.172860	3.088493	4.404079		C	21.907947	11.149254	-5.299513	
N	-14.232161	1.846005	2.466942		C	22.079088	9.750348	-5.399272	
C	-14.627924	1.058048	1.456651		C	22.914593	9.104782	-4.460581	
C	-13.736594	0.450295	0.569992		C	-23.291020	9.926240	3.178023	

C	-12.291076	0.651650	0.706418		C	-23.288692	8.523621	3.245620	
C	-11.414497	-0.012682	-0.268510		C	-22.564690	7.867017	4.251898	
C	-11.942118	-0.840448	-1.358128		C	-21.830976	8.601021	5.211044	
C	-13.397782	-1.019142	-1.439847		C	-21.842755	10.011858	5.131851	
C	-14.316408	-0.383312	-0.491204		C	-22.565448	10.667745	4.124534	
C	-10.038127	0.176198	-0.122975	155	H	-23.860598	7.934572	2.512180	559
N	-9.139065	-0.363232	-0.960115		H	-22.572364	6.768044	4.311552	
C	-7.743843	-0.278705	-0.907983		H	-21.267058	10.592061	5.868819	
C	-7.004279	-0.775099	-2.005649		H	-22.557368	11.767527	4.075868	
C	-5.607825	-0.716372	-2.006014		H	22.403128	12.968542	-4.225423	
C	-4.898159	-0.164993	-0.915736		H	21.251602	11.659276	-6.020727	
C	-5.648626	0.325547	0.177056		H	23.063130	8.016897	-4.536901	
C	-7.045825	0.271645	0.190399		H	24.215721	9.316883	-2.736868	
O	-11.811337	1.371123	1.627805		C	-9.949590	-23.390154	3.174429	
O	-15.568106	-0.540861	-0.579334		C	-8.547131	-23.374774	3.243264	
C	-13.887255	-1.826062	-2.470550		C	-7.897862	-22.636114	4.243773	
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C	-17.233459	-2.887404	-3.629871		C	-10.698432	-22.662985	4.113944	
C	-17.913523	-3.651882	-4.581197		C	-10.980882	23.369497	-3.395371	
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C	-15.801454	-4.344988	-5.538091		C	-10.900456	21.893848	-5.328940	
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H	-9.694847	0.804452	0.715728		C	-9.590520	23.547123	-3.480849	
H	-13.153525	-2.278510	-3.158168		H	-12.721747	22.390934	-4.259087	
H	-15.703304	0.871814	1.300797		H	-11.409580	21.237768	-6.051013	
H	-9.627366	-0.914455	-1.720292		H	-7.768891	23.046565	-4.559429	
H	-13.179138	1.929191	2.483615		H	-7.952372	-23.948242	2.515678	
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C	-7.873526	20.474771	-11.881347		H	-21.083559	-11.582524	-6.009814	
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C	-5.038544	18.074587	-6.588750		C	11.139081	-23.296334	-3.380829	
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C	-0.471097	-13.778078	0.561882		C	3.617005	16.215141	9.522958	
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C	20.179097	-7.010659	5.841359	H	-7.713199	-20.029220	13.846126
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C	17.325142	-3.057439	4.372709		C	2.910355	16.418205	14.274926	
N	14.385978	-1.810802	2.434817		H	4.151628	13.338790	13.423907	
C	14.779819	-1.010448	1.433482		H	5.245513	14.719837	11.646475	
C	13.887168	-0.393450	0.554153		H	3.345183	18.274969	13.227967	
C	12.441822	-0.596212	0.689194		H	2.248492	16.902477	15.009270	
C	11.564334	0.080963	-0.276635		C	-16.442749	2.685488	14.381368	
C	12.091149	0.919961	-1.358008		C	-15.030336	2.793624	14.401782	
C	13.546915	1.097922	-1.439635		C	-14.409141	3.564894	13.387044	
C	14.466111	0.452520	-0.497659		C	-15.164046	4.192133	12.395003	
C	10.188078	-0.107890	-0.131035	343	C	-16.574664	4.090888	12.367781	747
N	9.287817	0.440892	-0.960891		C	-17.187303	3.326535	13.386161	
C	7.892607	0.355579	-0.907372		N	-14.230342	2.182260	15.360894	
C	7.151433	0.852045	-2.003849		C	-14.619551	1.390851	16.426689	
C	5.755077	0.791631	-2.003017		C	-13.782724	0.807801	17.316451	
C	5.046980	0.238644	-0.912525		H	-15.707184	1.252336	16.519912	
C	5.798841	-0.251443	0.179365		H	-13.224838	2.331085	15.253644	

C	7.196094	-0.195921	0.191495	H	-13.312339	3.672966	13.387080
O	11.962555	-1.326776	1.601940	H	-14.654819	4.788917	11.623540
O	15.717548	0.612143	-0.584544	H	-18.282048	3.214145	13.382406
C	14.035872	1.912246	-2.464631	H	-16.970962	2.088393	15.137103
N	15.344606	2.146714	-2.639889	H	-23.810754	-12.069236	-15.075299
C	15.969385	2.932638	-3.613122	H	-15.461526	-2.522322	-15.380099
C	17.382279	2.977035	-3.621228	H	12.008943	-23.860817	-15.111071
C	18.062756	3.740937	-4.572730	H	2.442633	-15.536159	-15.357197
C	17.364642	4.483760	-5.551966	H	-11.851781	23.940545	-15.123856
C	15.951268	4.433541	-5.531068	H	-9.367181	24.292545	-15.064907
C	15.258387	3.677459	-4.581274	H	-2.287357	15.614760	-15.377904
O	11.322559	1.475791	-2.193824	H	23.956672	12.128999	-15.090466
H	9.845981	-0.745282	0.701232	H	15.616468	2.574505	-15.368050
H	13.301149	2.369315	-3.148136	H	-10.680249	-24.028401	14.932351
H	15.854719	-0.820405	1.279206	H	-2.272335	-13.222847	15.217771
H	9.774911	0.998836	-1.716782	H	-10.459038	-23.969404	2.389201
H	13.333576	-1.896242	2.451310	H	-23.716251	-11.722611	-2.559580
H	15.911169	1.630568	-1.909510	H	24.155399	-10.561427	14.945810
H	17.943831	2.392512	-2.876206	H	14.795169	-1.514272	15.936010
H	19.162823	3.751421	-4.574757	H	24.089356	-10.360668	2.402125
H	15.386473	5.015406	-6.274519	H	11.715940	-23.802591	-2.591880
H	14.159454	3.677574	-4.591161	H	-23.988426	10.684137	14.922638
H	7.685852	1.276407	-2.867793	H	-12.688758	0.923141	17.255490
H	5.194479	1.171853	-2.869810	H	-14.188390	0.195795	18.131739
H	5.269000	-0.667623	1.049497	H	-23.858722	10.441475	2.388200
H	7.741905	-0.563842	1.071750	H	-11.556158	23.877612	-2.606526
H	13.446344	-3.411727	4.294751	H	-9.072023	24.200682	-2.762696
H	14.776279	-4.640315	6.011403	H	10.822532	24.106382	14.937136
H	18.419238	-2.944483	4.400566	H	2.646777	14.426705	15.112859
H	17.117431	-1.715616	2.694546	H	23.889996	11.800318	-2.574766
C	6.175074	18.624412	4.502942	H	10.600009	24.045769	2.408013
C	5.793702	18.393941	5.872884				784
N	6.491426	19.303805	6.638243				
C	7.272261	20.112381	5.839478				
C	7.075324	19.671068	4.482524				
C	4.883099	17.420296	6.348662				
C	4.611646	17.156974	7.719871				
C	3.572798	16.221053	8.157236				
N	5.249154	17.724711	8.792502				
N	7.892043	20.841560	8.734664				
C	8.341175	21.521298	7.631554				
C	9.116208	22.702829	8.017899				
C	8.105497	21.169009	6.274902				
H	7.582196	20.097684	3.611505				
H	5.815481	18.039518	3.650759				
H	2.886653	15.668855	7.506271				
H	9.552284	23.437728	7.332594				
H	6.471787	19.339666	7.666124				
C	6.644611	-19.006903	-4.722755				
C	6.223233	-18.848220	-6.091195				

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N	7.023761	-19.673139	-6.852174	
C	7.909643	-20.362864	-6.051575	
C	7.669873	-19.931377	-4.698356	
C	5.190149	-18.008307	-6.571100	
C	4.866695	-17.813763	-7.942648	404

Section S-4: Powder X-Ray Diffraction Analysis (PXRD)

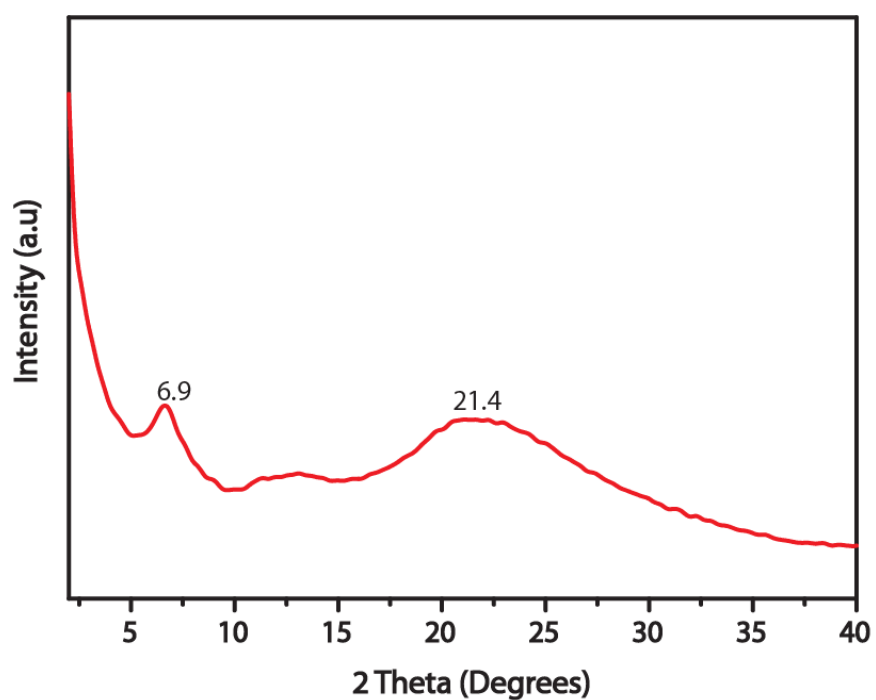


Figure S2: PXRD pattern of As-synthesized **TpTph** shows amorphous character.

Section S-5: FT-IR Spectra

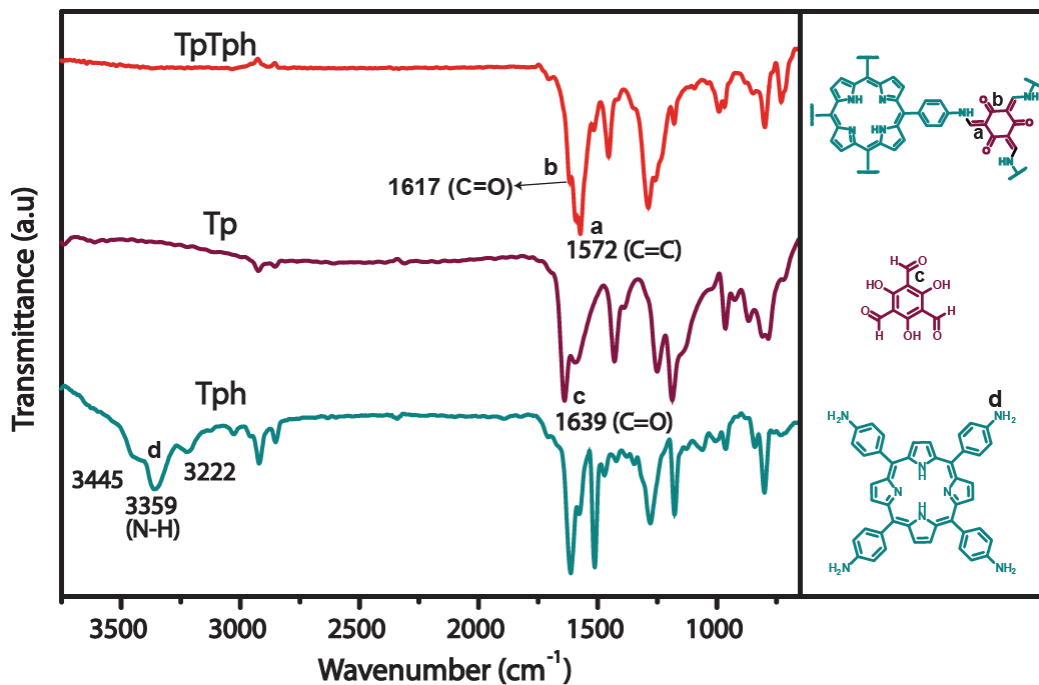


Figure S3: FT-IR spectra of **TpTph** (red), 1,3,5-triformylphloroglucinol (**Tp**) (red accent), and 5,10,15,20-Tetrakis(4-aminophenyl)-21H,23H-porphine (**Tph**) (green).

Section S-6: ^{13}C CP/MAS NMR Spectra

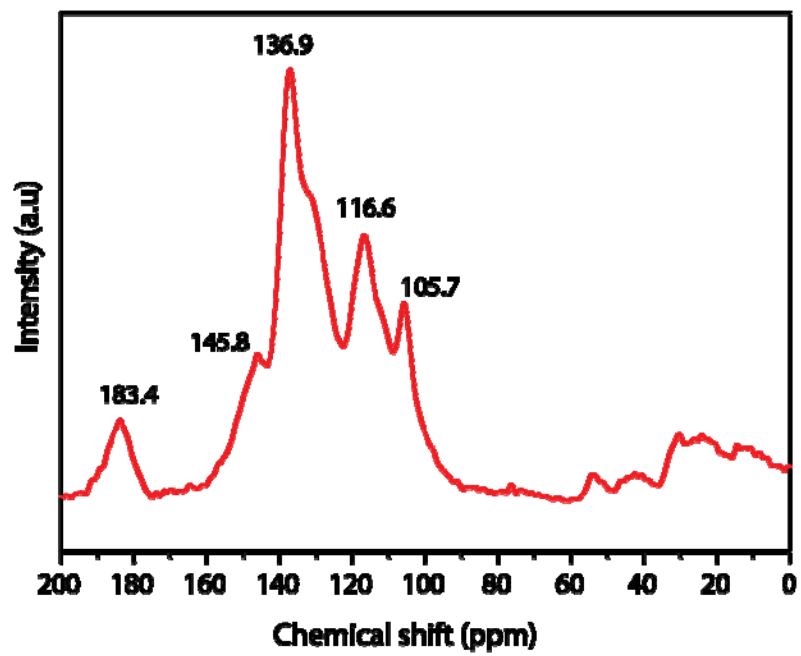


Figure S4: ^{13}C CP-MAS spectrum of TpTph.

Section S-7: Gas Adsorption Studies of TpTph

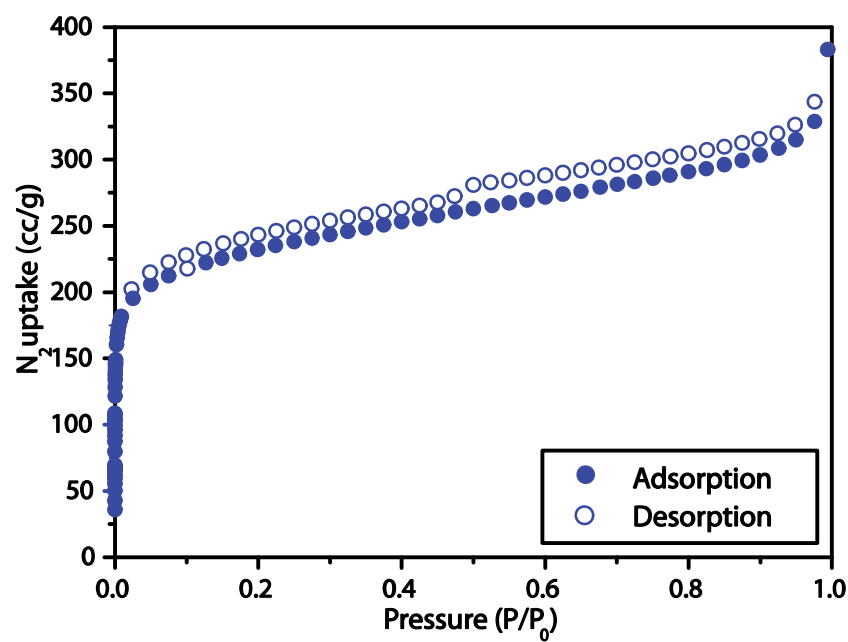


Figure S5: N₂ adsorption isotherm of **TpTph**.

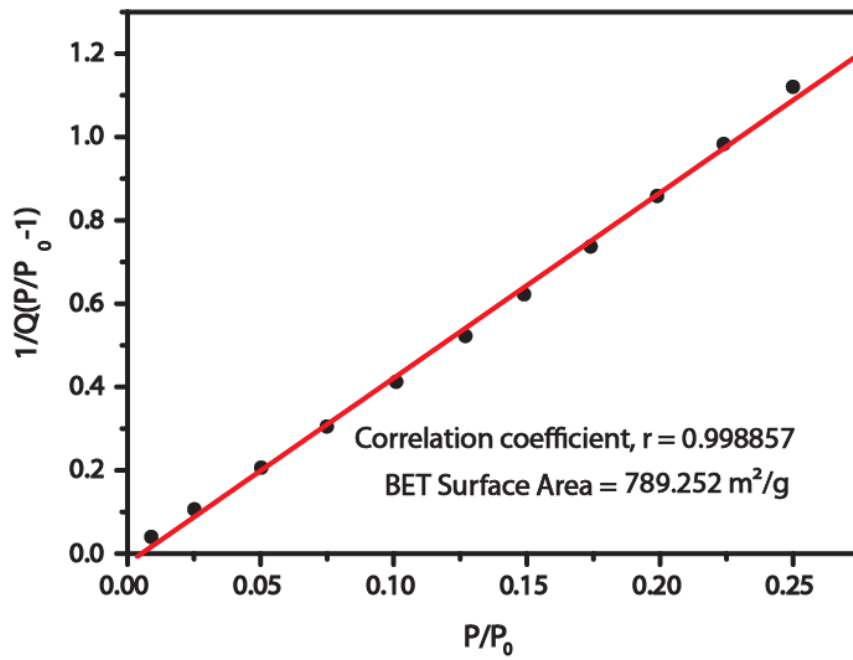


Figure S6: Langmuir surface area plot for **TpTph** calculated from the isotherm.

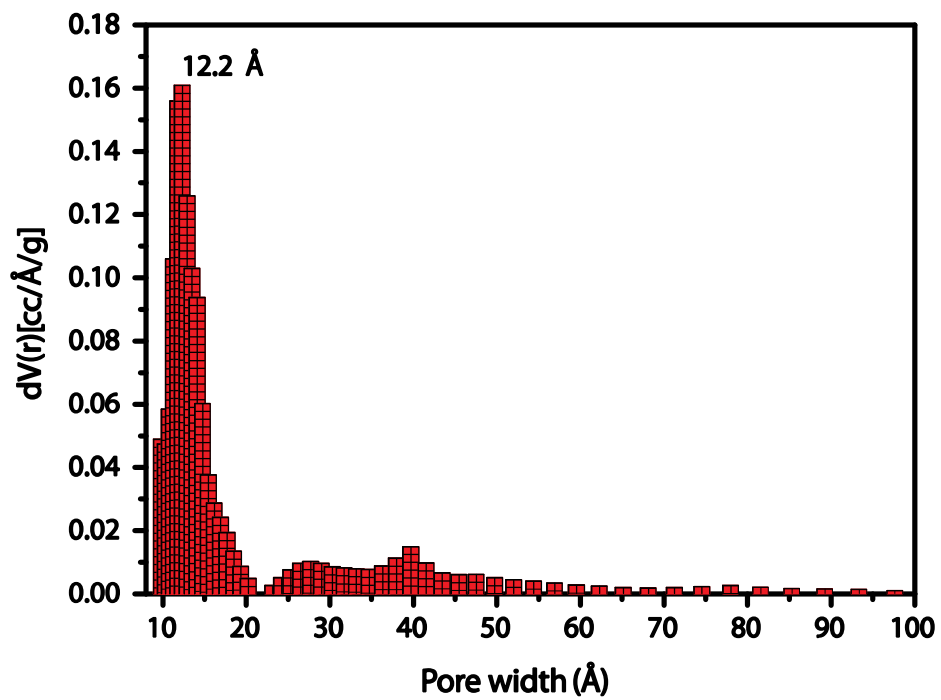


Figure S7: Pore size distribution of **TpTph**.

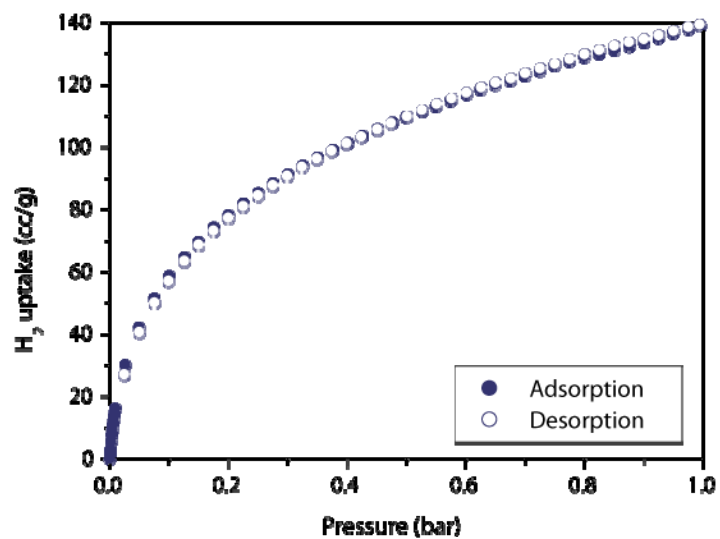


Figure S8: Hydrogen adsorption isotherm of **TpTph** at 77K.

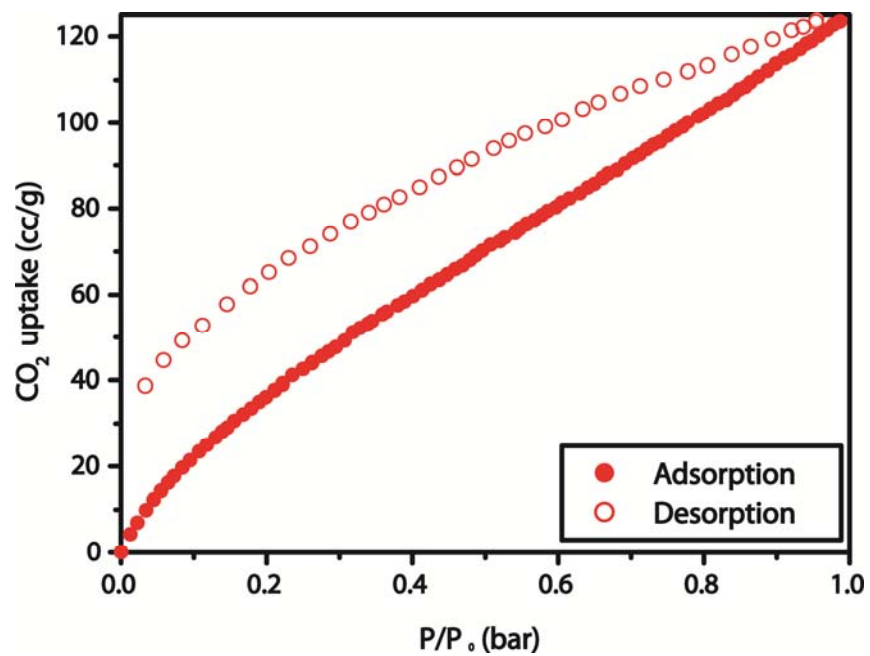


Figure S9: Carbon dioxide adsorption isotherms of **TpTph (Red)** at 273 K.

Section S-8: Vapor Adsorption Studies of TpTph

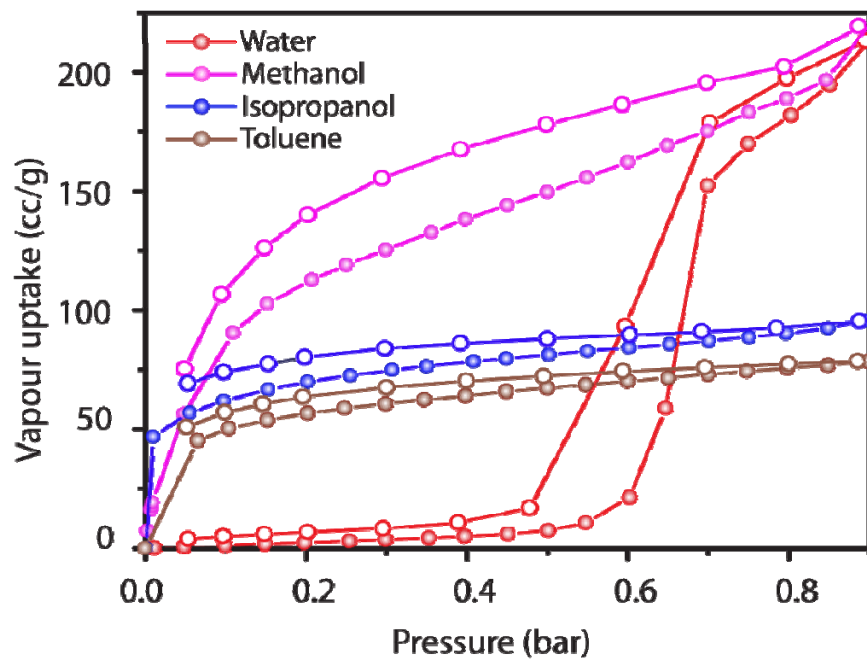


Figure S10: Water and vapor adsorption isotherms of TpTph (Red).

Section S-9: Thermogravimetric Analysis (TGA)

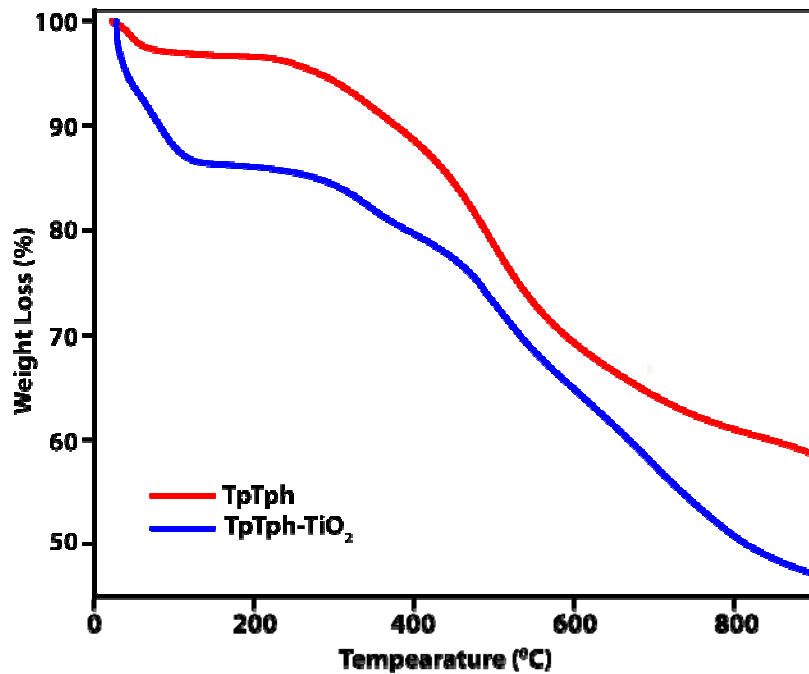


Figure S11: TGA data of activated **TpTph** under N₂ atmosphere.

Section S-10: Scanning Electron Micrographs (SEM)

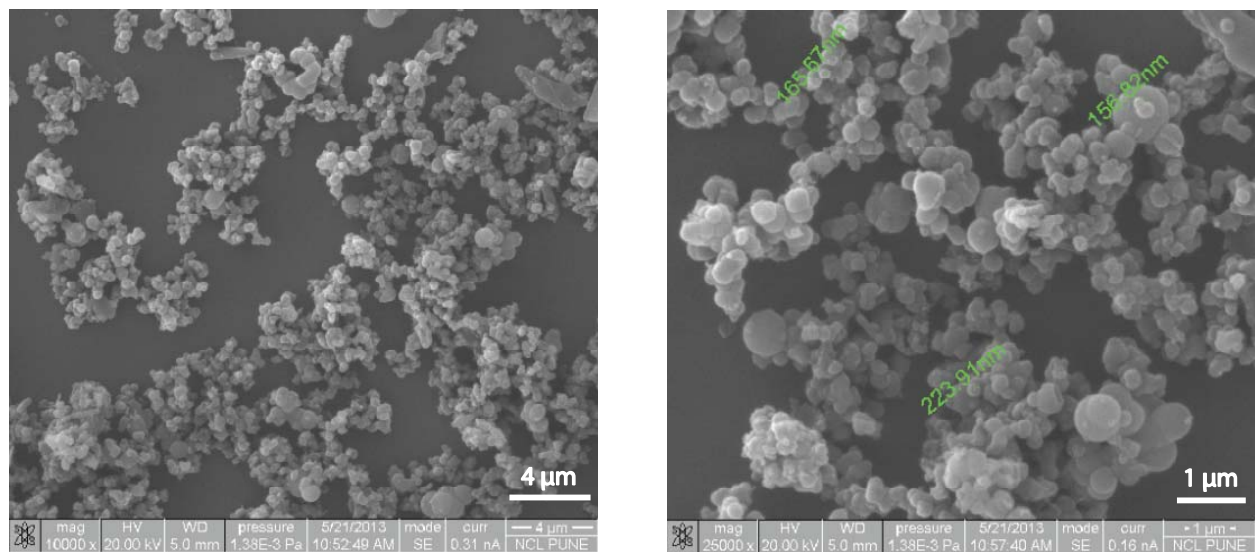


Figure S12: SEM images of TpTph.

Section S-11: Transmission Electron Micrographs (TEM)

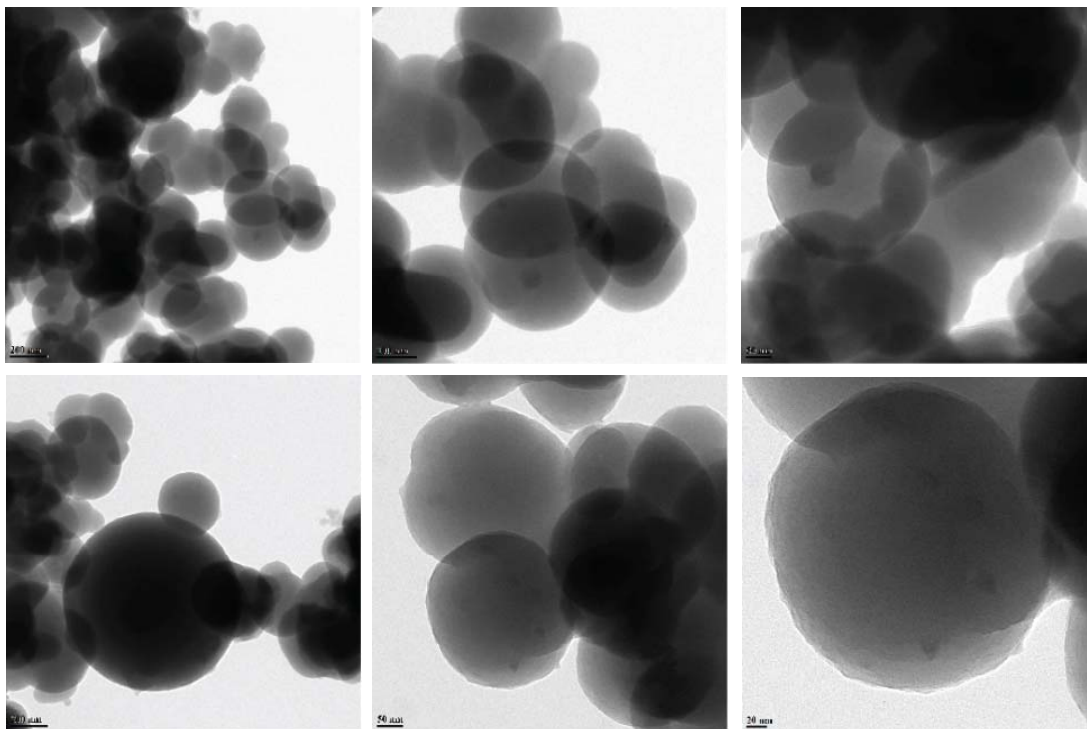


Figure S13: TEM images of TpTph.

Section S-12: UV-VIS Spectra

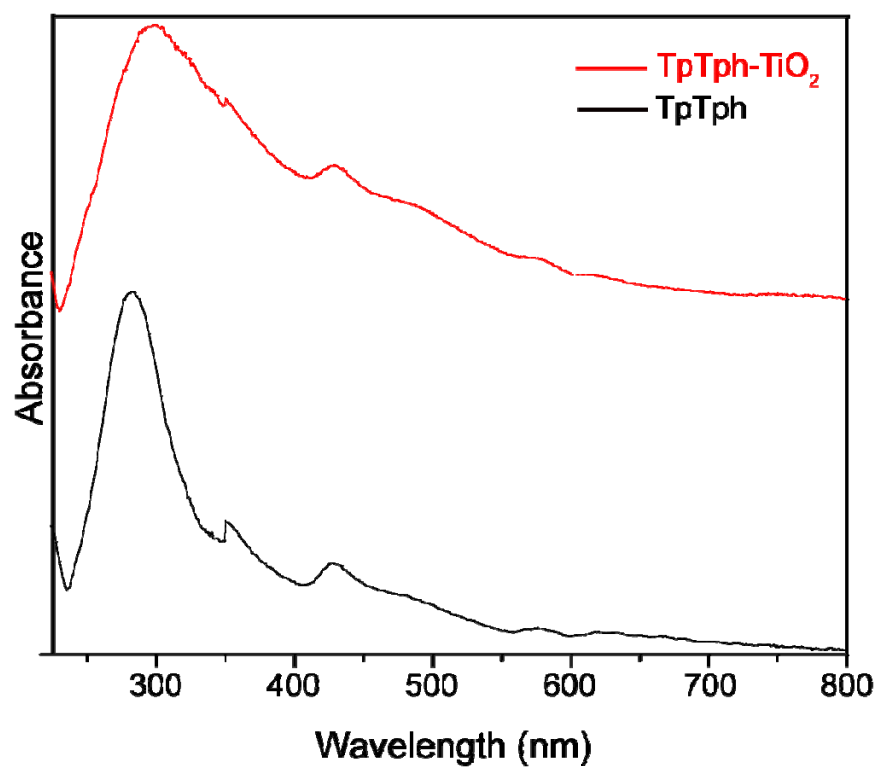


Figure S14: UV spectra of **TpTph** and **TpTph-TiO₂(2:1)** composite.

Section S-13: Stability test for TpTph in water, acid and base

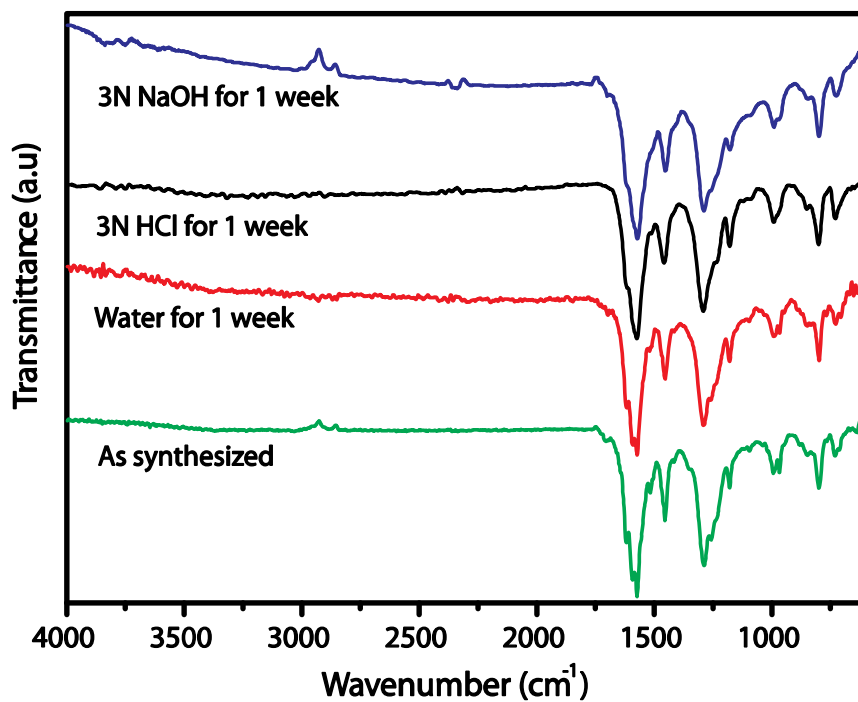


Figure S15: FT-IR spectrum of **TpTph** after treatment with water, 3(N) HCl, 3(N) NaOH for 1 week.

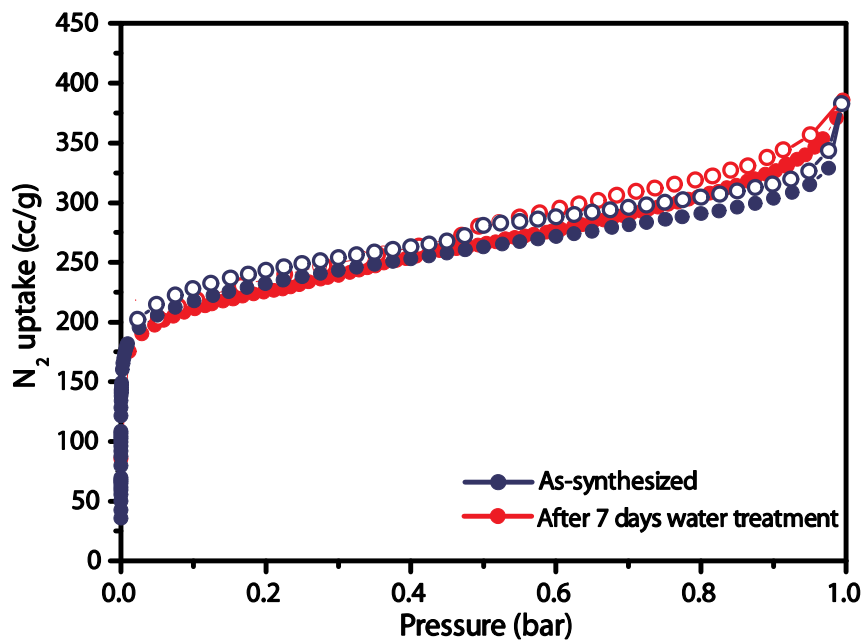


Figure S16: N₂ adsorption properties of **TpTph** before and after water treatment.

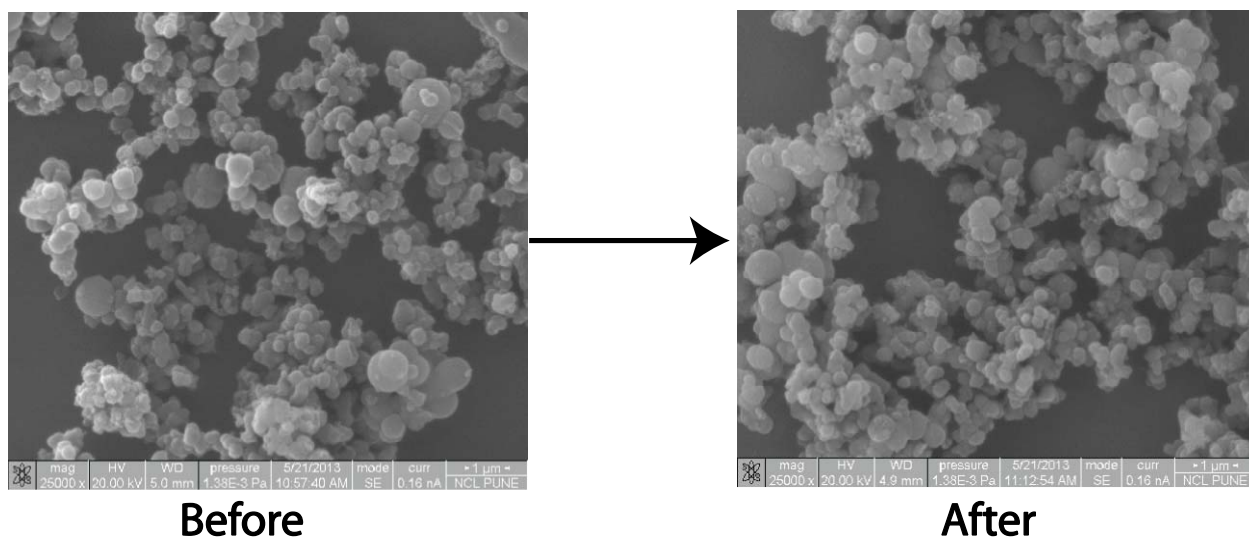


Figure S17: SEM of **TpTph** after treatment with 3(N) HCl for 1 week.

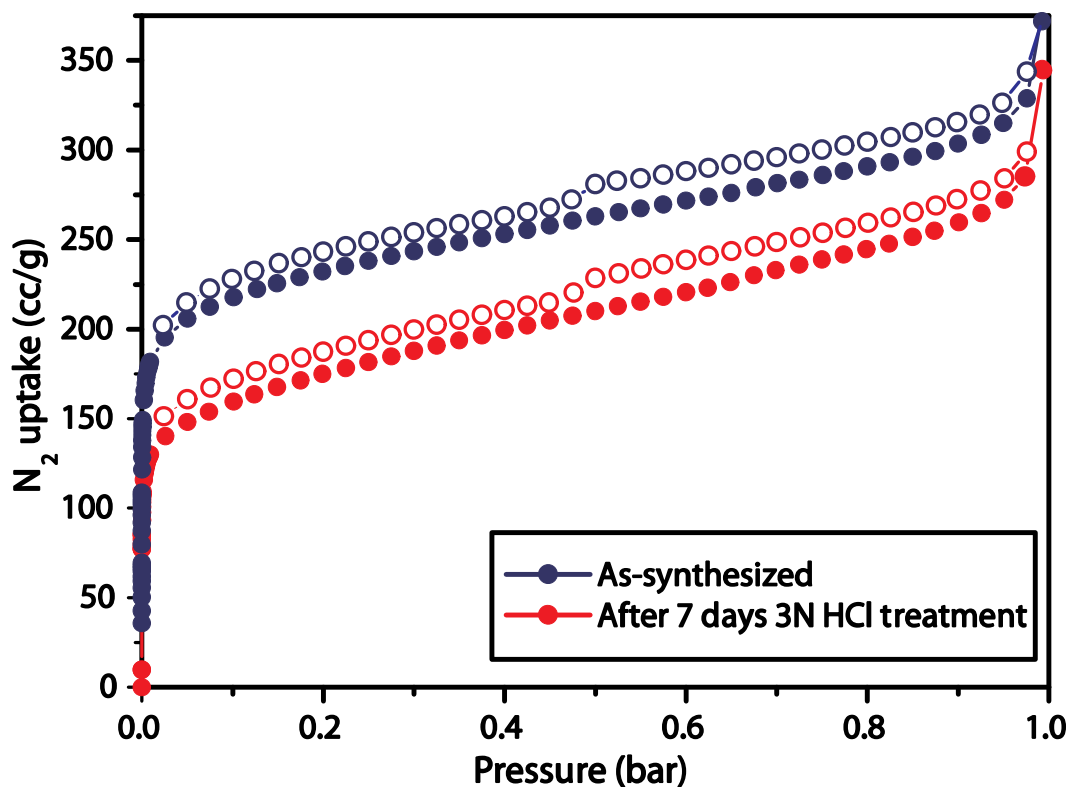


Figure S18: N₂ adsorption properties of **TpTph** before and after acid for 1 week.

Section S-14: References

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