

Supporting information for

The first fluorogenic sensor for sphingosine-1-phosphate lyase activity in intact cells

Pol Sanllehí,^{1,2} Mireia Casasampere,^{1,2} José Luís Abad,¹ Gemma Fabriàs,¹ Olga López,³
Jordi Bujons,⁴ Josefina Casas,^{*1} and Antonio Delgado^{*1,2}

Table of contents

General synthetic procedures	4
General procedure 1: Ru-catalyzed olefin cross metathesis reactions	4
General procedure 2: Regioselective phosphorylation of 1,3-diols	4
General procedure 3: TMSBr-mediated deprotection of dimethyl phosphates and N-Boc amino groups.....	5
General procedure 4: fluoride-mediated deprotection of TBS groups	5
Synthesis of probes RBM77 and RBM148	5
Synthesis of building blocks 5 and 7.....	5
Scheme S1	5
(S)-tert-butyl (1-((tert-butyldimethylsilyl)oxy)-3-oxopent-4-en-2-yl)carbamate (4)	6
tert-butyl ((2S,3R)-1-((tert-butyldimethylsilyl)oxy)-3-hydroxypent-4-en-2-yl) carbamate (5).....	6
tert-butyl ((2S,3R)-1-((tert-butyldimethylsilyl)oxy)-3-hydroxyhex-5-en-2-yl) carbamate (7).....	7
Configurational assignment of 5 and 7.....	8
Scheme S2.....	8
Synthesis of MPA esters.	8
Synthesis of precursors 12 and 14.....	10
tert-butyl ((2S,3R,E)-1-((tert-butyldimethylsilyl)oxy)-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-4-en-2-yl)carbamate (12).....	10
tert-butyl ((2S,3R,E)-1-((tert-butyldimethylsilyl)oxy)-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-5-en-2-yl)carbamate (14).....	10
Conversion of 12 and 14 into probes RBM148 and RBM77	11
Scheme S3.....	11

tert-butyl ((2S,3R,E)-1,3-dihydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-4-en-2-yl) carbamate (13).....	11
tert-butyl ((2S,3R,E)-1-((dimethoxyphosphoryl)oxy)-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-4-en-2-yl)carbamate (1)	12
(2S,3R,E)-2-ammonio-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-4-en-1-yl hydrogen phosphate (RBM77)	12
tert-butyl ((2S,3R,E)-1,3-dihydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-5-en-2-yl) carbamate (15).....	13
tert-butyl ((2S,3R,E)-1-((dimethoxyphosphoryl)oxy)-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-5-en-2-yl)carbamate (2)	13
(2S,3R,E)-2-ammonio-3-hydroxy-7-((2-oxo-2H-chromen-7-yl)oxy)hept-5-en-1-yl hydrogen phosphate (RBM148)	14
Synthesis of aldehyde 17	14
Scheme S4.....	14
(E)-7-((5-hydroxypent-3-en-1-yl)oxy)-2H-chromen-2-one (16).....	14
(E)-5-((2-oxo-2H-chromen-7-yl)oxy)pent-2-enal (17).....	15
Fluorometric characterization of probes RBM77 and RBM148	16
Figure S1	16
Table S1. Spectral characterization of the probes	16
Release of umbelliferone from 17	17
Figure S2.....	17
Attempts to synthesize aldehyde 22	18
Scheme S5.....	18
(But-3-en-1-yloxy)(tert-butyl)diphenylsilane (18).....	18
(E)-7-((5-((tert-butyldiphenylsilyl)oxy)pent-2-en-1-yl)oxy)-2H-chromen-2-one (19)....	18
(E)-7-((5-hydroxypent-2-en-1-yl)oxy)-2H-chromen-2-one (21).....	19
S1PL activity with recombinant hS1PL	19
a) Fluorogenic assay with RBM77 as substrate.....	19
b) Fluorogenic assay with RBM148 as substrate	20
Validation of the probes as hS1PL substrates	21
Figure S3	21
Figure S4.....	22
Computational Methods	22
General data	22

Docking studies	23
Molecular dynamics	24
Figure S5	25
Figure S6	26
Figure S7	27
Cell assays.....	28
Overexpression of Sgpl1	28
Cell lysates	28
Determination of protein concentration	28
S1PL activity in cell lysates	28
S1PL activity in intact cells.....	29
Liposome preparations	29
Measurement of liposomes size distribution	30
Figure S8	30
Determination of encapsulation efficiency.....	30
Figure S9	31
Effect of liposomes and encapsulated RBM148 on cell viability.....	31
Figure S10.....	31
Figure S11	32
S1PL activity in intact cells with encapsulated RBM77	33
Figure S12.....	33
Figure S13	33
References.....	34
NMR SPECTRA	37

General synthetic procedures

Unless otherwise stated, reactions were carried out under argon atmosphere. Dry solvents were obtained by passing through an activated alumina column on a Solvent Purification System (SPS). Methanol and ethanol were dried over CaH_2 and distilled prior to use. Commercially available reagents and solvents were used with no further purification. All reactions were monitored by TLC analysis using ALUGRAM[®] SIL G/UV₂₅₄ precoated aluminum sheets (Machery–Nagel). UV light was used as the visualizing agent and a 5% (w/v) ethanolic solution of phosphomolybdic acid as the developing agent. Flash column chromatography was carried out with the indicated solvents using flash–grade silica gel (37–70 μm). Yields refer to chromatographically and spectroscopically pure compounds, unless otherwise stated. . High Resolution Mass Spectrometry analyses were carried out on an Acquity UPLC system coupled to a LCT Premier orthogonal accelerated time–of–flight mass spectrometer (Waters) using electrospray ionization (ESI) technique. Optical rotations were measured at room temperature on a Perkin Elmer 341 polarimeter.

General procedure 1: Ru–catalyzed olefin cross metathesis reactions

To a stirred solution of the starting olefins (1.5 and 6 mmol, respectively) in degassed CH_2Cl_2 (20 mL), Grubbs catalyst 2nd generation (0.03 equiv/mol) was added portionwise at rt. The resulting mixture was refluxed in the dark for 2 h, cooled down to rt and concentrated *in vacuo* to afford a crude, which was purified as indicated for each compound.

General procedure 2: Regioselective phosphorylation of 1,3–diols

A solution of the starting diol (0.5 mmol) in CH_2Cl_2 (15 mL) at 0 °C was treated successively with *N*–methylimidazole (1.5 equiv/mol) and dimethyl chlorophosphate (1.2 equiv/mol). The reaction mixture was stirred at rt for 1 h, cooled down to 0 °C and quenched by the dropwise addition of saturated aqueous NH_4Cl (10 mL). The resulting mixture was extracted with CH_2Cl_2 (3 x 15 mL) and the combined organic layers were washed with brine (2 x 20 mL), dried over anhydrous MgSO_4 and filtered. Evaporation of the solvent afforded a crude mixture, which was purified as indicated for each compound.

General procedure 3: TMSBr-mediated deprotection of dimethyl phosphates and *N*-Boc amino groups.

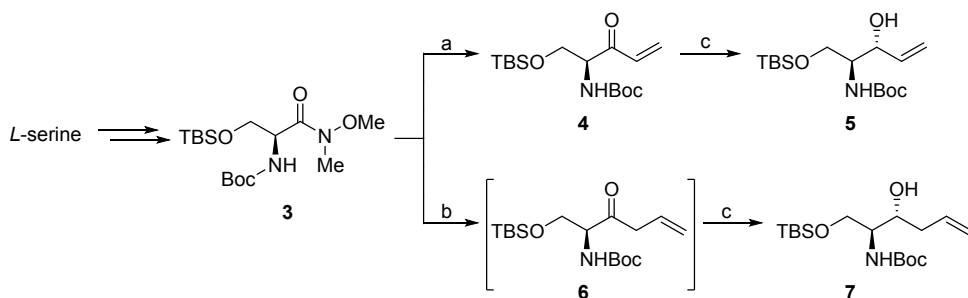
To an ice cooled solution of the starting dimethyl phosphate (0.2 mmol) in dry CH₃CN (8 mL), was added dropwise TMSBr (5 equiv/mol). After stirring for 3 h at rt, the reaction mixture was concentrated under reduced pressure. The residue was then redissolved in MeOH/H₂O (95:5, same reaction volume) and stirred for an additional hour at rt. Evaporation of the solvent afforded a crude mixture, which was purified as indicated for each compound.

General procedure 4: fluoride-mediated deprotection of TBS groups

To a solution of the corresponding TBS-protected alcohol (1 mmol) in THF (10 mL) was added dropwise TBAF (1 M in THF, 2 equiv/mol) at 0 °C. After stirring at the same temperature for 30 min, the reaction was quenched with saturated aqueous NH₄Cl (10 mL) and the resulting mixture was extracted with Et₂O (3 x 15 mL). The combined organic layers were washed with brine (2 x 20 mL), dried over anhydrous MgSO₄, filtered, and evaporated to give the crude products. Purification by flash chromatography on silica gel (from 0 to 3 % MeOH in CH₂Cl₂) afforded the required alcohols.

Synthesis of probes RBM77 and RBM148

Synthesis of building blocks 5 and 7



Scheme S1. Reagents and conditions: (a) vinylmagnesium bromide, THF, 0 °C to rt (71 %). (b) allylmagnesium chloride, THF, -20 °C to rt. (c) LiAlH(Ot-Bu)₃, EtOH, -78 °C (for **5**: 90 %, dr = 98:2 *anti/syn*; for **7**: 78 % over 2 steps, dr = 99:1 *anti/syn*)

(*S*)-*tert*-butyl (1-((*tert*-butyldimethylsilyl)oxy)-3-oxopent-4-en-2-yl)carbamate (**4**)

Vinylmagnesium bromide (82.8 mL, 57.9 mmol, 0.70 M in THF) was added dropwise to a solution of **3**¹ (6.0 g, 16.6 mmol) in THF (50 mL) at 0 °C. After stirring at rt for 1 h,

the reaction mixture was added dropwise *via* canula to a 1 M aq. solution of HCl (50 mL) at 0 °C and the resulting mixture was extracted with EtOAc (3 x 50 mL). The organic layers were combined, washed with brine (2 x 50 mL), dried over anhydrous MgSO₄, filtered and concentrated *in vacuo* to give the crude product. Flash chromatography (from 0 to 5 % EtOAc in hexane) gave **4** (3.85 g, 71 %) as a colourless oil.

$[\alpha]^{20}_{\text{D}} = +62.2$ (*c* 1.0, CHCl₃) [lit.¹ $[\alpha]^{24}_{\text{D}} = +63.3$ (*c* 0.62, CHCl₃)]. ¹H NMR (400 MHz, CDCl₃) δ 6.56 (dd, *J* = 17.5, 10.6 Hz, 1H), 6.35 (dd, *J* = 17.4, 1.3 Hz, 1H), 5.83 (d, *J* = 10.6 Hz, 1H), 5.52 (br d, *J* = 7.0 Hz, 1H), 4.60 (dt, *J* = 7.6, 3.9 Hz, 1H), 4.01 (dd, *J* = 10.3, 3.3 Hz, 1H), 3.85 (dd, *J* = 10.3, 4.4 Hz, 1H), 1.45 (s, 9H), 0.84 (s, 9H), 0.01 (s, 3H), 0.00 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 197.0, 155.4, 133.3, 129.5, 79.9, 63.6, 59.7, 28.5, 25.9, 18.3, -5.5, -5.5. HRMS calcd. for C₁₆H₃₁NO₄SiNa ([M + Na]⁺): 352.1920, found: 352.1937.

***tert*-butyl ((2*S*,3*R*)-1-((*tert*-butyldimethylsilyl)oxy)-3-hydroxypent-4-en-2-yl) carbamate (**5**)**

A solution of **4** (1.01 g, 3.08 mmol) in ethanol (12 mL) was added dropwise to a suspension of lithium tri-*tert*-butoxyaluminum hydride (1.72 g, 6.78 mmol) in ethanol (30 mL) at -78 °C. After stirring at the same temperature for 30 min, the reaction mixture was allowed to warm to 0 °C and was quenched with 10 % (w/v) aqueous citric acid (20 mL). The resulting mixture was extracted with EtOAc (3 x 50 mL) and the combined organic layers were washed with brine (2 x 50 mL), dried over anhydrous MgSO₄, filtered, and concentrated *in vacuo* to give a 98:2 *anti/syn* crude mixture of diastereomers. Flash chromatography of the residue (from 0 to 11 % EtOAc in hexane) gave pure *anti*-**5** (922 mg, 90 %) as a colourless oil.

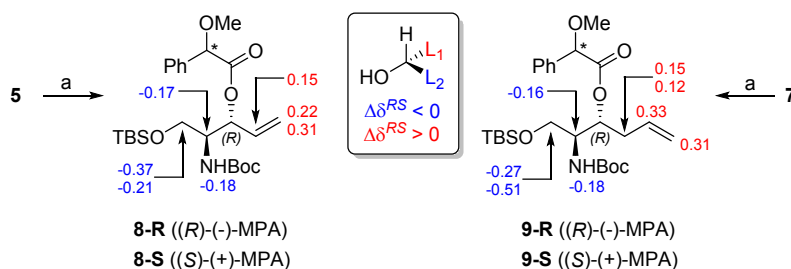
$[\alpha]^{20}_{\text{D}} = +28.1$ (*c* 1.0, CHCl₃) [lit.¹ $[\alpha]^{23}_{\text{D}} = +24.8$ (*c* 1.45, CHCl₃)]. ¹H NMR (400 MHz, CDCl₃) δ 5.93 (ddd, *J* = 17.2, 10.6, 4.9 Hz, 1H), 5.39 (dt, *J* = 17.2, 1.7 Hz, 1H), 5.32 – 5.19 (m, 2H), 4.27 (s, 1H), 3.93 (dd, *J* = 10.4, 2.9 Hz, 1H), 3.76 (dd, *J* = 10.3, 2.6 Hz, 1H), 3.63 (dd, *J* = 7.3, 3.5 Hz, 2H), 3.44 (br s, 1H), 1.45 (s, 9H), 0.90 (s, 9H), 0.07 (s, 3H), 0.07 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 156.0, 138.0, 116.0, 79.7, 75.0, 63.6, 54.2, 28.5, 25.9, 18.3, -5.5, -5.5. HRMS calcd. for C₁₆H₃₄NO₄Si ([M + H]⁺): 332.2257, found: 332.2267.

***tert*-butyl ((2*S*,3*R*)-1-((*tert*-butyldimethylsilyl)oxy)-3-hydroxyhex-5-en-2-yl) carbamate (7)**

Allylmagnesium chloride (6.9 mL, 13.8 mmol, 2.0 M in THF) was added dropwise to a solution of **3** (2.50 g, 6.90 mmol) in THF (30 mL) at -20 °C. The reaction mixture was allowed to warm to 0 °C over 5 h with stirring and was then quenched with saturated aqueous NH₄Cl (25 mL). The resulting mixture was extracted with EtOAc (3 x 50 mL) and the combined organic layers were washed with brine (2 x 50 mL), dried over anhydrous MgSO₄, filtered, and concentrated under reduced pressure. The resulting residue, containing crude **6**, was dissolved in ethanol (15 mL) and was added dropwise to a suspension of lithium tri-*tert*-butoxyaluminum hydride (3.86 g, 15.2 mmol) in ethanol (60 mL) at -78 °C. After stirring at the same temperature for 30 min, the reaction mixture was allowed to warm to 0 °C and was quenched with 10 % (w/v) aqueous citric acid (40 mL). The resulting mixture was carefully concentrated under reduced pressure and then extracted with EtOAc (3 x 50 mL). The organic layers were combined and were washed with brine (2 x 50 mL), dried over anhydrous MgSO₄, filtered, and concentrated *in vacuo* to give a 99:1 *anti/syn* crude mixture of diastereomers. Flash chromatography of the residue (from 0 to 9 % EtOAc in hexane) gave pure *anti*-**7** (1.85 g, 78 %) as a colourless oil.

$[\alpha]_{\text{D}}^{20} = +32.6$ (*c* 1.0, CHCl₃) [lit.² $[\alpha]_{\text{D}} = +32.2$ (*c* 1.32, CHCl₃)]. ¹H NMR (400 MHz, CDCl₃) δ 5.85 (ddt, *J* = 17.2, 10.2, 7.0 Hz, 1H), 5.23 (br d, *J* = 7.9 Hz, 1H), 5.19 – 5.06 (m, 2H), 3.98 (dd, *J* = 10.6, 2.9 Hz, 1H), 3.84 – 3.76 (m, 1H), 3.76 – 3.67 (m, 1H), 3.60 – 3.48 (m, 1H), 3.02 (br s, 1H), 2.42 – 2.27 (m, 2H), 1.45 (s, 9H), 0.90 (s, 9H), 0.08 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 155.8, 134.7, 117.9, 79.6, 73.0, 63.4, 53.8, 39.5, 28.5, 26.0, 18.3, -5.5, -5.5. HRMS calcd. for C₁₇H₃₅NO₄NaSi ([M + Na]⁺): 368.2233, found: 368.2235.

Configurational assignment of 5 and 7



Scheme S2. Reagents and conditions: (a) (*R*)-(-)- or (*S*)-(+)-MPA, EDC, DMAP, CH₂Cl₂, 0 °C to rt. (64 % for **8-R**; 82 % for **8-S**; 36 % for **9-R**; 68 % for **9-S**).

Synthesis of MPA esters.

DMAP (1.1 equiv/mol) was added portionwise at 0 °C to a solution of the starting alcohol (0.2 mmol) in CH₂Cl₂ (2 mL) containing (*R*)-(-)- or (*S*)-(+)-MPA (1.6 equiv/mol) and EDC (1.5 equiv/mol). After stirring for 6 h at rt, the reaction mixture was washed with saturated aq. NaHCO₃ (2 x 2 mL) and the organic phase was separated, dried over anhydrous MgSO₄, filtered and evaporated to dryness. The resulting residue was flash chromatographed on silica gel (from 0 to 8 % EtOAc in hexane) to give the corresponding MPA esters

(*R*)-(-)-MPA ester of alcohol **5** (**8-R**)

Compound **8-R** (colorless oil, 52 mg, 64 %) was obtained from **5** (56 mg, 0.17 mmol), (*R*)-(-)-MPA (45 mg, 0.27 mmol), EDC (49 mg, 0.25 mmol) and DMAP (23 mg, 0.19 mmol), according to the above procedure.

¹H NMR (400 MHz, CDCl₃) δ 7.46 – 7.40 (m, 2H), 7.40 – 7.29 (m, 3H), 5.82 (ddd, *J* = 17.3, 10.5, 6.8 Hz, 1H), 5.37 – 5.21 (m, 3H), 4.75 (s, 1H), 4.49 (br d, *J* = 9.4 Hz, 1H), 3.78 – 3.69 (m, 1H), 3.43 – 3.36 (m, 4H), 3.19 (dd, *J* = 10.3, 3.5 Hz, 1H), 1.40 (s, 9H), 0.81 (s, 9H), –0.10 (s, 3H), –0.13 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 169.4, 155.4, 136.4, 133.3, 128.9, 128.8, 127.3, 119.3, 82.6, 79.6, 74.2, 61.2, 57.5, 53.9, 28.5, 25.9, 18.3, –5.5, –5.6. HRMS calcd. for C₂₅H₄₂NO₆Si ([*M* + *H*]⁺): 480.2781, found: 480.2782.

(S)-(+)-MPA ester of alcohol 5 (8-S)

Compound **8-S** (colorless oil, 72 mg, 82 %) was obtained from **5** (61 mg, 0.184 mmol), (S)-(+)-MPA (49 mg, 0.29 mmol), EDC (53 mg, 0.28 mmol) and DMAP (25 mg, 0.20 mmol), according to the above procedure.

¹H NMR (400 MHz, CDCl₃) δ 7.45 – 7.39 (m, 2H), 7.38 – 7.29 (m, 3H), 5.68 (ddd, *J* = 17.1, 10.7, 6.3 Hz, 1H), 5.39 (t, *J* = 6.5 Hz, 1H), 5.07 (d, *J* = 10.7 Hz, 1H), 4.95 (d, *J* = 17.2 Hz, 1H), 4.77 (s, 1H), 4.67 (br d, *J* = 9.4 Hz, 1H), 3.95 – 3.84 (m, 1H), 3.62 (dd, *J* = 10.3, 3.5 Hz, 1H), 3.55 (dd, *J* = 10.3, 4.6 Hz, 1H), 3.43 (s, 3H), 1.43 (s, 9H), 0.87 (s, 9H), 0.00 (s, 3H), –0.01 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 169.4, 155.5, 136.3, 132.5, 128.8, 128.7, 127.3, 118.9, 82.7, 79.6, 74.1, 61.8, 57.6, 53.7, 28.5, 25.9, 18.3, –5.4, –5.5. HRMS calcd. for C₂₅H₄₂NO₆Si ([M + H]⁺): 480.2781, found: 480.2762.

(R)-(-)-MPA ester of alcohol 7 (9-R)

Compound **9-R** (yellow oil, 38 mg, 36 %) was obtained from **7** (75 mg, 0.22 mmol), (R)-(-)-MPA (58 mg, 0.35 mmol), EDC (62 mg, 0.33 mmol) and DMAP (29 mg, 0.24 mmol), according to the above procedure.

¹H NMR (400 MHz, CDCl₃) δ 7.45 – 7.38 (m, 2H), 7.38 – 7.28 (m, 3H), 5.83 – 5.69 (m, 1H), 5.09 – 4.98 (m, 3H), 4.70 (s, 1H), 4.53 (br d, *J* = 9.5 Hz, 1H), 3.73 – 3.63 (m, 1H), 3.40 (s, 3H), 3.31 (dd, *J* = 10.3, 3.7 Hz, 1H), 3.06 (dd, *J* = 10.4, 3.5 Hz, 1H), 2.55 – 2.45 (m, 1H), 2.38 – 2.28 (m, 1H), 1.41 (s, 9H), 0.81 (s, 9H), –0.12 (s, 3H), –0.15 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 169.7, 155.5, 136.5, 133.6, 128.9, 128.8, 127.3, 118.1, 82.6, 79.6, 72.5, 61.3, 57.5, 53.5, 36.0, 28.5, 25.9, 18.3, –5.5, –5.6. HRMS calcd. for C₂₆H₄₄NO₆Si ([M + H]⁺): 494.2938, found: 494.2926.

(S)-(+)-MPA ester of alcohol 7 (9-S)

Compound **9-S** (yellow oil, 73 mg, 68 %) was obtained from **7** (75 mg, 0.22 mmol), (S)-(+)-MPA (58 mg, 0.35 mmol), EDC (62 mg, 0.33 mmol) and DMAP (29 mg, 0.24 mmol), according to the above procedure.

¹H NMR (400 MHz, CDCl₃) δ 7.45 – 7.37 (m, 2H), 7.37 – 7.28 (m, 3H), 5.43 (ddt, *J* = 17.0, 10.2, 7.2 Hz, 1H), 5.01 (td, *J* = 7.5, 4.0 Hz, 1H), 4.81 – 4.65 (m, 4H), 3.89 – 3.79 (m, 1H), 3.57 (d, *J* = 3.6 Hz, 2H), 3.42 (s, 3H), 2.40 – 2.31 (m, 1H), 2.26 – 2.16 (m, 1H),

1.44 (s, 9H), 0.87 (s, 9H), 0.01 (s, 3H), -0.01 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 170.0, 155.5, 136.4, 132.9, 128.8, 128.6, 127.4, 118.0, 82.8, 79.7, 73.1, 61.8, 57.6, 53.2, 35.4, 28.5, 26.0, 18.3, -5.4, -5.5. HRMS calcd. for $\text{C}_{26}\text{H}_{44}\text{NO}_6\text{Si}$ ($[\text{M} + \text{H}]^+$): 494.2938, found: 494.2933.

Synthesis of precursors **12** and **14**

tert-butyl

((2*S*,3*R*,*E*)-1-((*tert*-butyldimethylsilyl)oxy)-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-4-en-2-yl)carbamate (12**)**

Compound **12** (inseparable 94:6 *E/Z* mixture, colourless oil, 860 mg, 63 %) was obtained from **5** (875 mg, 2.64 mmol), **10**³ (2.28 g, 10.6 mmol) and Grubbs catalyst 2nd generation (67 mg, 0.08 mmol), according to general procedure 1. The title compound was purified by flash chromatography on silica gel (from 0 to 1 % CH_3OH in CH_2Cl_2). Early-eluting fractions were independently collected to give a sample of pure *E*-alkene, from which the following data were acquired.

$[\alpha]^{20}_{\text{D}} = +11.1$ (*c* 1.0, CHCl_3) ^1H NMR (400 MHz, CDCl_3) δ 7.63 (d, *J* = 9.5 Hz, 1H), 7.36 (d, *J* = 8.6 Hz, 1H), 6.85 – 6.76 (m, 2H), 6.25 (d, *J* = 9.5 Hz, 1H), 5.92 – 5.81 (m, 1H), 5.71 (dd, *J* = 15.4, 5.4 Hz, 1H), 5.24 (br d, *J* = 8.2 Hz, 1H), 4.29 – 4.22 (m, 1H), 4.06 (t, *J* = 6.5 Hz, 2H), 3.94 (dd, *J* = 10.4, 2.9 Hz, 1H), 3.80 – 3.70 (m, 1H), 3.65 – 3.55 (m, 1H), 2.60 (app q, *J* = 6.6 Hz, 2H), 1.45 (s, 9H), 0.88 (s, 9H), 0.06 (s, 3H), 0.05 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 162.2, 161.3, 156.0, 155.9, 143.5, 132.9, 128.9, 127.5, 113.2, 113.0, 112.7, 101.5, 79.7, 74.5, 67.8, 63.5, 54.5, 32.1, 28.5, 25.9, 18.2, -5.5, -5.5. HRMS calcd. for $\text{C}_{27}\text{H}_{41}\text{NO}_7\text{NaSi}$ ($[\text{M} + \text{Na}]^+$): 542.2550, found: 542.2557.

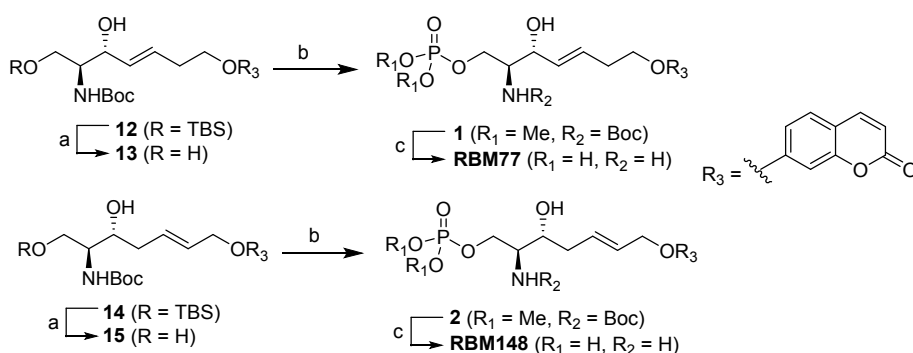
tert-butyl

((2*S*,3*R*,*E*)-1-((*tert*-butyldimethylsilyl)oxy)-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-5-en-2-yl)carbamate (14**)**

Compound **14** (93:7 *E/Z* mixture, colourless oil, 865 mg, 66 %) was obtained from **135** (868 mg, 2.51 mmol), **11**⁴ (2.03 g, 10.1 mmol) and Grubbs catalyst 2nd generation (64 mg, 0.08 mmol), according to general procedure 1. The title compound was purified by flash chromatography on silica gel (from 0 to 27 % EtOAc in hexane). Pure *E*-alkene was obtained after purification of the diastereomeric mixture by flash chromatography using the same gradient.

$[\alpha]_D^{20} = +16.5$ (c 1.0, CHCl_3). ^1H NMR (400 MHz, CDCl_3) δ 7.62 (d, $J = 9.5$ Hz, 1H), 7.36 (d, $J = 8.5$ Hz, 1H), 6.87 – 6.78 (m, 2H), 6.24 (d, $J = 9.5$ Hz, 1H), 6.00 – 5.90 (m, 1H), 5.80 (dt, $J = 15.5, 5.8$ Hz, 1H), 5.22 (br d, $J = 8.3$ Hz, 1H), 4.55 (d, $J = 5.8$ Hz, 2H), 3.97 (dd, $J = 10.6, 2.9$ Hz, 1H), 3.85 – 3.69 (m, 2H), 3.60 – 3.49 (m, 2H), 3.14 (br s, 1H), 2.46 – 2.32 (m, 2H), 1.45 (s, 9H), 0.90 (s, 9H), 0.08 (s, 6H). ^{13}C NMR (101 MHz, CDCl_3) δ 161.9, 161.3, 155.9, 155.8, 143.5, 132.1, 128.9, 126.9, 113.2, 113.1, 112.7, 101.8, 79.7, 73.1, 69.1, 63.4, 53.9, 37.9, 28.5, 25.9, 18.3, –5.5, –5.5. HRMS calcd. for $\text{C}_{27}\text{H}_{41}\text{NO}_7\text{NaSi}$ ($[\text{M} + \text{Na}]^+$): 542.2550, found: 542.2556.

Conversion of 12 and 14 into probes RBM148 and RBM77



Scheme S3. Reagents and conditions: (a) TBAF, THF, 0 °C; (b) $(\text{MeO})_2\text{POCl}$, 1-methylimidazole, CH_2Cl_2 , 0 °C to rt; (c) (i) TMSBr, CH_3CN , 0 °C to rt. (ii) $\text{CH}_3\text{OH}/\text{H}_2\text{O}$ (95% v/v), rt;

tert-butyl

((2*S*,3*R*,*E*)-1,3-dihydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-4-en-2-yl) carbamate (13)

Diol 13 (colorless oil, 345 mg, 95 %) was obtained from 12 (465 mg, 0.90 mmol) and TBAF (1.79 mL, 1.79 mmol), according to general procedure 4.

$[\alpha]_D^{20} = -3.6$ (c 1.0, CHCl_3). ^1H NMR (400 MHz, CDCl_3) δ 7.63 (d, $J = 9.5$ Hz, 1H), 7.36 (d, $J = 8.5$ Hz, 1H), 6.86 – 6.76 (m, 2H), 6.25 (d, $J = 9.5$ Hz, 1H), 5.94 – 5.83 (m, 1H), 5.74 (dd, $J = 15.5, 6.0$ Hz, 1H), 5.33 (br d, $J = 7.2$ Hz, 1H), 4.40 – 4.35 (m, 1H), 4.07 (t, $J = 6.4$ Hz, 2H), 3.94 (dd, $J = 11.4, 3.5$ Hz, 1H), 3.73 (dd, $J = 11.4, 3.7$ Hz, 1H), 3.65 (br s, 1H), 2.65 – 2.55 (m, 1H), 1.45 (s, 9H). ^{13}C NMR (101 MHz, CDCl_3) δ 162.1, 161.5, 156.4, 155.9, 143.6, 132.6, 128.9, 128.3, 113.1, 113.1, 112.7, 101.5, 80.0, 74.4, 67.8,

62.5, 55.4, 32.1, 28.5. HRMS calcd. for C₂₁H₂₇NO₇Na ([M + Na]⁺): 428.1685, found: 428.1691.

tert-butyl

((2*S*,3*R*,*E*)-1-((dimethoxyphosphoryl)oxy)-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-4-en-2-yl)carbamate (1**)**

Compound **1** (colorless oil, 105 mg, 69 %) was obtained from alcohol **13** (120 mg, 0.30 mmol), *N*-methylimidazole (35 μ L, 0.44 mmol) and dimethyl chlorophosphate (38 μ L, 0.36 mmol), according to general procedure 2. The title compound was purified by flash chromatography on silica gel (from 0 to 100 % EtOAc in hexane).

[α]_D²⁰ = +6.4 (*c* 0.9, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 7.63 (d, *J* = 9.5 Hz, 1H), 7.36 (d, *J* = 8.6 Hz, 1H), 6.87 – 6.77 (m, 2H), 6.24 (d, *J* = 9.5 Hz, 1H), 5.91 – 5.81 (m, 1H), 5.70 (dd, *J* = 15.5, 6.4 Hz, 1H), 5.09 (br d, *J* = 7.8 Hz, 1H), 4.35 (ddd, *J* = 10.8, 8.2, 4.7 Hz, 1H), 4.20 (t, *J* = 6.3 Hz, 1H), 4.16 – 4.08 (m, 1H), 4.05 (t, *J* = 6.5 Hz, 2H), 3.88 – 3.73 (m, 7H), 2.62 – 2.53 (m, 2H), 1.42 (s, 9H). ¹³C NMR (101 MHz, CDCl₃) δ 162.2, 161.33, 156.0, 155.8, 143.5, 132.0, 128.9, 113.2, 113.0, 112.7, 101.5, 80.0, 72.2, 67.8, 66.7 (d, *J*_{C-P} = 5.7 Hz), 55.0 (br d, *J*_{C-P} = 4.4 Hz), 54.7 (d, *J*_{C-P} = 5.9 Hz), 32.1, 28.5. ³¹P NMR (162 MHz, CDCl₃) δ 2.65. HRMS calcd. for C₂₃H₃₂NO₁₀NaP ([M + Na]⁺): 536.1662, found: 536.1670.

((2*S*,3*R*,*E*)-2-ammonio-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-4-en-1-yl hydrogen phosphate (RBM77**)**

Compound **RBM77** (white solid, 49 mg, 85 %) was obtained from dimethyl phosphate **1** (77 mg, 0.15 mmol) and TMSBr (99 μ L, 0.75 mmol), according to general procedure 3. The crude reaction mixture was dissolved in methanol and loaded on an Amberlite[®] XAD4 column (10 g), which had been washed thoroughly with acetone and then equilibrated with water. Elution with a linear gradient from 0 to 50 % CH₃CN in H₂O provided pure **RBM77**.

[α]_D²⁰ = +14.7 (*c* 0.7, DMSO). ¹H NMR (400 MHz, CD₃OD) δ 7.88 (d, *J* = 9.5 Hz, 1H), 7.54 (d, *J* = 8.6 Hz, 1H), 6.99 – 6.89 (m, 2H), 6.25 (d, *J* = 9.5 Hz, 1H), 6.06 – 5.93 (m, 1H), 5.68 (dd, *J* = 15.4, 6.6 Hz, 1H), 4.34 (t, *J* = 5.6 Hz, 1H), 4.20 – 4.07 (m, 3H), 4.00 (dt, *J* = 11.8, 7.9 Hz, 1H), 3.43 – 3.36 (m, 1H), 2.67 – 2.58 (m, 2H). ¹³C NMR (101 MHz, DMSO-d₆) δ 161.6, 160.3, 155.4, 144.3, 131.0, 129.5, 128.4, 112.7, 112.4, 112.3, 101.1, 69.0, 67.5, 61.5 (br d, *J*_{C-P} = 2.7 Hz), 56.2 (br d, *J*_{C-P} = 2.7 Hz), 31.3. ³¹P NMR (162

MHz, DMSO- d_6) δ 1.84. HRMS calcd. for $C_{16}H_{21}NO_8P$ ($[M + H]^+$): 386.1005, found: 386.0997.

***tert*-butyl**

((2*S*,3*R*,*E*)-1,3-dihydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-5-en-2-yl) carbamate (15)

Diol **15** (colorless oil, 427 mg, 88 %) was obtained from **14** (620 mg, 1.19 mmol) and TBAF (2.39 mL, 2.39 mmol), according to general procedure 4.

$[\alpha]^{20}_D = +4.5$ (c 1.0, $CHCl_3$). 1H NMR (400 MHz, $CDCl_3$) δ 7.63 (d, $J = 9.5$ Hz, 1H), 7.37 (d, $J = 8.5$ Hz, 1H), 6.87 – 6.78 (m, 2H), 6.25 (d, $J = 9.5$ Hz, 1H), 5.99 – 5.88 (m, 1H), 5.83 (dt, $J = 15.5, 5.5$ Hz, 1H), 5.31 (br d, $J = 10.4$ Hz, 1H), 4.56 (d, $J = 5.6$ Hz, 2H), 4.01 (dd, $J = 11.4, 3.4$ Hz, 1H), 3.92 – 3.83 (m, 1H), 3.77 (dd, $J = 11.4, 3.4$ Hz, 1H), 3.61 – 3.50 (m, 1H), 2.46 – 2.33 (m, 3H), 1.45 (s, 9H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 161.9, 161.5, 156.2, 155.8, 143.7, 131.7, 128.9, 127.4, 113.3, 113.1, 112.7, 101.8, 80.0, 72.9, 69.0, 62.5, 54.8, 37.5, 28.5. HRMS calcd. for $C_{21}H_{27}NO_7Na$ ($[M + Na]^+$): 428.1685, found: 428.1682.

***tert*-butyl**

((2*S*,3*R*,*E*)-1-((dimethoxyphosphoryl)oxy)-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-5-en-2-yl)carbamate (2)

Compound **2** (colorless oil, 232 mg, 78 %) was obtained from alcohol **15** (236 mg, 0.58 mmol), *N*-methylimidazole (70 μ L, 0.87 mmol) and dimethyl chlorophosphate (75 μ L, 0.70 mmol), according to general procedure 2. The title compound was purified by flash chromatography on silica gel (from 0 to 100 % EtOAc in hexane).

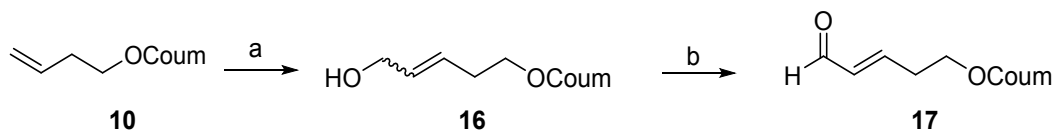
$[\alpha]^{20}_D = +14.1$ (c 1.0, $CHCl_3$). 1H NMR (400 MHz, $CDCl_3$) δ 7.63 (d, $J = 9.5$ Hz, 1H), 7.36 (d, $J = 8.5$ Hz, 1H), 6.87 – 6.78 (m, 2H), 6.24 (d, $J = 9.5$ Hz, 1H), 6.02 – 5.91 (m, 1H), 5.82 (dt, $J = 15.5, 5.8$ Hz, 1H), 5.05 (br d, $J = 8.1$ Hz, 1H), 4.56 (d, $J = 5.7$ Hz, 2H), 4.50 – 4.41 (m, 1H), 4.15 – 4.05 (m, 1H), 3.81 (d, $J = 3.4$ Hz, 3H), 3.78 (d, $J = 3.4$ Hz, 3H), 3.71 (br s, 2H), 3.34 (br s, 1H), 2.53 – 2.40 (m, 1H), 2.31 (dt, $J = 14.1, 6.9$ Hz, 1H), 1.44 (s, 9H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 161.9, 161.3, 155.9, 155.6, 143.6, 131.9, 128.9, 127.3, 113.2, 113.2, 112.7, 101.8, 80.1, 70.1, 69.1, 67.0 (d, $J_{C-P} = 5.7$ Hz), 54.8 (d, $J_{C-P} = 6.0$ Hz), 54.8 (d, $J_{C-P} = 6.0$ Hz), 54.7 (br d, $J_{C-P} = 5.4$ Hz), 36.8, 28.5. ^{31}P NMR (162 MHz, $CDCl_3$) δ 2.96. HRMS calcd. for $C_{23}H_{32}NO_{10}NaP$ ($[M + Na]^+$): 536.1662, found: 536.1660.

(2*S*,3*R*,*E*)-2-ammonio-3-hydroxy-7-((2-oxo-2*H*-chromen-7-yl)oxy)hept-5-en-1-yl hydrogen phosphate (RBM148)

Compound **RBM148** (white solid, 45 mg, 82 %) was obtained from dimethyl phosphate **2** (73 mg, 0.14 mmol) and TMSBr (94 μ L, 0.71 mmol), according to general procedure 3. The title compound was purified as described above for **RBM77**.

$[\alpha]^{20}_{\text{D}} = +4.2$ (c 1.0, DMSO). ^1H NMR (400 MHz, CD_3OD) δ 7.89 (d, $J = 9.5$ Hz, 1H), 7.54 (d, $J = 8.5$ Hz, 1H), 6.98 – 6.90 (m, 2H), 6.25 (d, $J = 9.5$ Hz, 1H), 6.01 – 5.84 (m, 2H), 4.65 (d, $J = 4.9$ Hz, 2H), 4.18 (ddd, $J = 10.8, 6.9, 3.7$ Hz, 1H), 4.08 (dt, $J = 11.5, 7.9$ Hz, 1H), 3.88 (dt, $J = 9.4, 4.8$ Hz, 1H), 3.39 (dt, $J = 8.2, 3.9$ Hz, 1H), 2.45 – 2.29 (m, 2H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 161.4, 160.3, 155.3, 144.3, 131.7, 129.5, 126.6, 112.8, 112.4, 112.3, 101.4, 68.8, 68.1, 61.4 (d, $J_{\text{C-P}} = 5.7$ Hz), 55.8 (d, $J_{\text{C-P}} = 3.4$ Hz), 35.6. ^{31}P NMR (162 MHz, $\text{DMSO}-d_6$) δ 2.08. HRMS calcd. for $\text{C}_{16}\text{H}_{21}\text{NO}_8\text{P}$ ($[\text{M} + \text{H}]^+$): 386.1005, found: 386.0994.

Synthesis of aldehyde 17



Scheme S4. Reagents and conditions: (a) allyl alcohol, Grubbs cat. 2nd gen., CH_2Cl_2 , reflux (33 %, dr = 85:15 *E/Z*). (b) DMP, CH_2Cl_2 , 0 $^\circ\text{C}$ to rt (64 %).

(*E*)-7-((5-hydroxypent-3-en-1-yl)oxy)-2*H*-chromen-2-one (16)

Compound **16** (inseparable 85:15 *E/Z* mixture, brownish oil, 114 mg, 33 %) was obtained from **10**⁴ (300 mg, 1.39 mmol), allyl alcohol (377 μ L, 5.55 mmol) and Grubbs catalyst 2nd generation (35 mg, 42.0 μ mol), according to general procedure 1. The title compound was obtained in 33% yield after flash chromatography (from 0 to 1.5 % MeOH in CH_2Cl_2).

^1H NMR (400 MHz, CDCl_3 , major isomer) δ 7.63 (d, $J = 9.5$ Hz, 1H), 7.36 (d, $J = 8.6$ Hz, 1H), 6.86 – 6.76 (m, 2H), 6.24 (d, $J = 9.5$ Hz, 1H), 5.87 – 5.74 (m, 2H), 4.15 (br s, 2H), 4.06 (t, $J = 6.6$ Hz, 2H), 2.66 – 2.52 (m, 2H). ^{13}C NMR (101 MHz, CDCl_3 , major isomer) δ 162.2, 161.4, 156.0, 143.6, 132.2, 128.9, 127.6, 113.2, 113.1, 112.7, 101.5, 68.0, 63.5, 32.0. HRMS calcd. for $\text{C}_{14}\text{H}_{15}\text{O}_4$ ($[\text{M} + \text{H}]^+$): 247.0970, found: 247.0961.

(*E*)-5-((2-oxo-2*H*-chromen-7-yl)oxy)pent-2-enal (17**)**

Dess–Martin periodinane (530 mg, 1.25 mmol) was added to a solution of allylic alcohol **16** (205 mg, 0.83 mmol) in CH₂Cl₂ (4 mL) at 0°C. After stirring for 2 h at rt, the reaction mixture was filtered through a plug of Celite and the filtrate was evaporated to dryness. The residue was purified by flash chromatography (from 0 to 50 % EtOAc in hexane) to afford aldehyde **17** (130 mg, 64 %) as a white solid. Late-eluting fractions were independently collected to give a 95:5 *E*:*Z* mixture of **17**, from which the following data were acquired.

¹H NMR (400 MHz, CDCl₃, major isomer) δ 9.57 (d, *J* = 7.8 Hz, 1H), 7.64 (d, *J* = 9.5 Hz, 1H), 7.39 (d, *J* = 8.5 Hz, 1H), 6.93 (dt, *J* = 15.7, 6.7 Hz, 1H), 6.88 – 6.78 (m, 2H), 6.31 – 6.21 (m, 2H), 4.20 (t, *J* = 6.1 Hz, 2H), 2.91 – 2.83 (m, 2H). ¹³C NMR (101 MHz, CDCl₃, major isomer) δ 193.7, 161.7, 161.2, 156.0, 153.0, 143.4, 134.9, 129.0, 113.6, 113.0, 101.5, 66.2, 32.3. HRMS calcd. for C₁₄H₁₃O₄ ([M + H]⁺): 245.0814, found: 245.0810.

Fluorometric characterization of probes RBM77 and RBM148

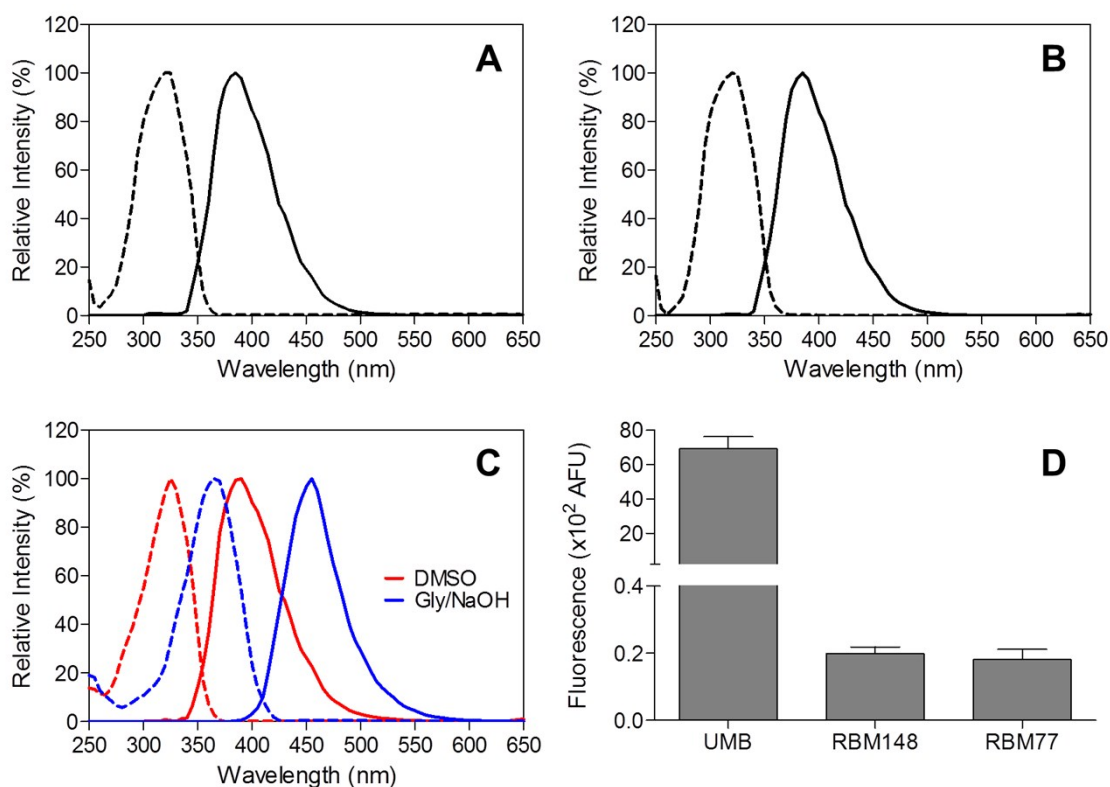


Figure S1. Excitation (dashed line) and emission (bold line) spectra of compounds **RBM148** (A) and **RBM77** (B) in DMSO. $\lambda_{\text{ex}}/\lambda_{\text{em}}$ were 325/385 nm for both compounds. (C) Excitation (dashed line) and emission (bold line) spectra of Umbelliferone in DMSO (red) or in Gly/NaOH buffer pH 10.6 (blue). $\lambda_{\text{ex}}/\lambda_{\text{em}}$ were 325/385 (DMSO) and 365/455 nm (Gly/NaOH buffer pH 10.6), respectively. (D) Fluorescence intensity ($\lambda_{\text{ex}}/\lambda_{\text{em}} = 365/455$ nm) of 10 μM solutions of umbelliferone, **RBM148** or **RBM77** in Gly/NaOH buffer pH 10.6.

Table S1. Spectral characterization of the probes

Compound	DMSO		Gly/NaOH buffer pH 10.6	
	$\lambda_{\text{ex}}/\lambda_{\text{em}}$ (nm)	ϵ ($\text{M}^{-1}\cdot\text{cm}^{-1}$)	$\lambda_{\text{ex}}/\lambda_{\text{em}}$ (nm)	ϵ ($\text{M}^{-1}\cdot\text{cm}^{-1}$)
Umbelliferone	325/385	17996	365/455	25976
RBM148	325/385	13456	325/390	14122
RBM77	325/385	11154	325/390	11540

Release of umbelliferone from **17**

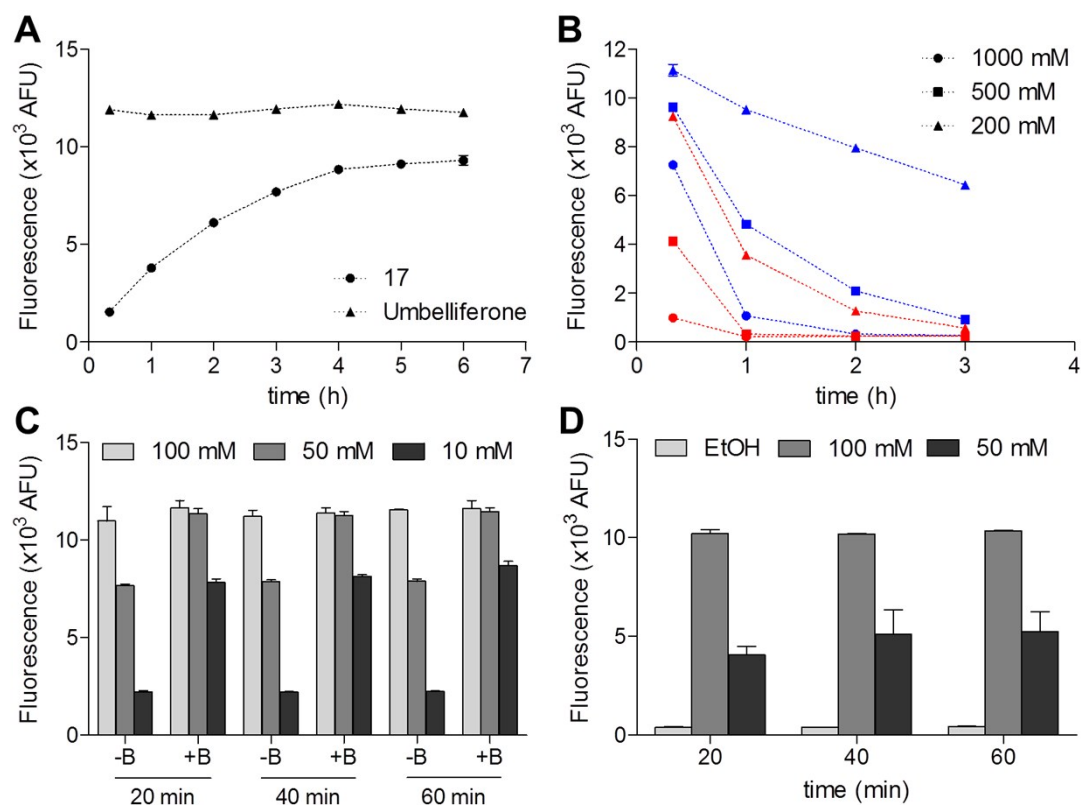
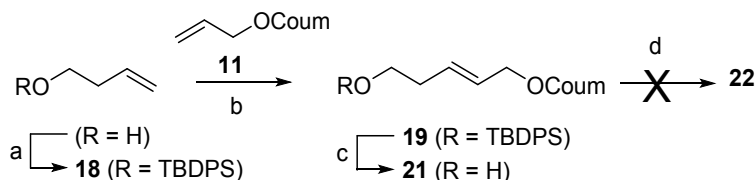


Figure S2. Optimization of the conditions for the base-mediated release of umbelliferone. A 5 μ M solution (100 μ L) of either **17** or umbelliferone in a 100 mM HEPES buffer pH 7.4, containing 0.1 mM EDTA, 0.05 % Triton X-100, 0.01 % Pluronic F127 (Biotium), and 100 μ M pyridoxal 5'-phosphate were successively treated with the corresponding basic solution (at the indicated concentrations) and/or solvent and incubated at 37 $^{\circ}$ C for the indicated times before fluorescence reading ($\lambda_{\text{ex/em}}$ = 355/460 nm). In all cases, umbelliferone and **17** were added from 100 μ M stock solutions in DMSO (5 μ L). (A) Umbelliferone and **17** were treated with 50 μ L of MeOH and 100 μ L of a 200 mM glycine/NaOH buffer solution (pH 10.6). (B) Umbelliferone was treated with 50 μ L of MeOH and 100 μ L of KOH in EtOH (blue) or aq. NaOH (red) at the indicated concentrations. (C) Umbelliferone was treated with 50 μ L of MeOH and 100 μ L of KOH in EtOH, at the indicated concentrations, and incubated at 37 $^{\circ}$ C for the indicated times without further treatment (-B) or treated with 100 μ L of KOH in EtOH, at the indicated concentrations, incubated at 37 $^{\circ}$ C for the indicated times and further treated with 50 μ L of a 200 mM glycine/NaOH buffer solution immediately before reading (+B). (D) Aldehyde **17** was treated with 100 μ L of KOH in EtOH at the indicated concentrations,

incubated at 37 °C for the indicated times and further treated with 50 μ L of a 200 mM glycine/NaOH buffer solution immediately before reading. Data are means \pm SD of one representative experiment with triplicates.

Attempts to synthesize aldehyde 22



Scheme S5. Reagents and conditions: (a) TBDPSCl, imidazole, DMF, rt (quant. yield). (b) **11**, Grubbs cat. 2nd gen., CH₂Cl₂, reflux (58 %, dr = 91:9 *E/Z*). (c) TBAF, THF, 0 °C to rt (96 %). (d) DMP, CH₂Cl₂, 0 °C to rt.

(But-3-en-1-yloxy)(*tert*-butyl)diphenylsilane (**18**)

A solution of 3-buten-1-ol (500 mg, 6.93 mmol) in DMF (10 mL) was treated successively with imidazole (1.04 g, 15.3 mmol) and TBDPSCl (1.98 mL, 7.63 mmol). The resulting mixture was stirred overnight at rt, diluted with Et₂O (10 mL) and poured into 1 M aqueous HCl (10 mL). The mixture was extracted with Et₂O (3 x 10 mL) and the combined organic layers were washed with brine (2 x 15 mL), dried over anhydrous MgSO₄, filtered and evaporated *in vacuo*. Flash chromatography of the residue (from 0 to 1 % EtOAc in hexane) gave **18** (2.17 g, quant. yield) as a colorless oil.

¹H NMR (400 MHz, CDCl₃) δ 7.72 – 7.64 (m, 4H), 7.46 – 7.33 (m, 6H), 5.83 (ddt, *J* = 17.1, 10.2, 6.9 Hz, 1H), 5.09 – 4.99 (m, 2H), 3.71 (t, *J* = 6.7 Hz, 2H), 2.32 (qt, *J* = 6.8, 1.3 Hz, 2H), 1.05 (s, 9H). ¹³C NMR (101 MHz, CDCl₃) δ 135.7, 135.6, 134.1, 129.7, 127.7, 116.5, 63.7, 37.4, 27.0, 19.4. HRMS calcd. for C₂₀H₂₇OSi ([M + H]⁺): 311.1831, found: 311.1834.

(*E*)-7-((5-((*tert*-butyldiphenylsilyl)oxy)pent-2-en-1-yl)oxy)-2*H*-chromen-2-one (**19**)

Compound **19** (inseparable 91:9 *E/Z* mixture, colourless oil, 347 mg, 58 %) was obtained from **11** (250 mg, 1.24 mmol), **18** (1.54 g, 4.95 mmol) and Grubbs catalyst 2nd generation (32 mg, 0.04 mmol), according to general procedure 1 (see main text). The title compound was purified by flash chromatography on silica gel (from 0 to 16 % EtOAc in hexane).

Late-eluting fractions were independently collected to give a 94:6 *E:Z* mixture of **19**, from which the following data were acquired.

¹H NMR (400 MHz, CDCl₃, major isomer) δ 7.69 – 7.60 (m, 4H), 7.46 – 7.32 (m, 6H), 6.86 – 6.76 (m, 2H), 6.25 (d, *J* = 9.5 Hz, 1H), 5.93 – 5.82 (m, 1H), 5.73 (dt, *J* = 15.5, 5.9 Hz, 1H), 4.51 (d, *J* = 5.8 Hz, 2H), 3.74 (t, *J* = 6.5 Hz, 2H), 2.42 – 2.31 (m, 2H), 1.04 (s, 9H). ¹³C NMR (101 MHz, CDCl₃, major isomer) δ 162.0, 161.4, 156.0, 143.5, 135.7, 134.0, 133.1, 129.8, 128.9, 127.8, 125.8, 113.2, 112.7, 101.8, 69.3, 63.3, 35.8, 27.0, 19.4. HRMS calcd. for C₃₀H₃₂O₄NaSi ([M + Na]⁺): 507.1968, found: 507.1972.

(*E*)-7-((5-hydroxypent-2-en-1-yl)oxy)-2*H*-chromen-2-one (21**)**

To an ice cooled solution of **19** (175 mg, 0.36 mmol) in THF (3.5 mL) was added dropwise TBAF (1 M in THF, 722 μL, 0.72 mmol). After completion of addition the ice bath was removed and the mixture was stirred for 2 h at rt. The reaction was then quenched with saturated aqueous NH₄Cl (5 mL) and the resulting mixture was extracted with Et₂O (3 x 10 mL). The combined organic layers were washed with brine (2 x 15 mL), dried over anhydrous MgSO₄, filtered, and evaporated to give the crude product. Purification by flash chromatography on silica gel (from 0 to 70 % EtOAc in hexane) afforded homoallylic alcohol **21** (85 mg, 96 %) as a colourless oil. Late-eluting fractions were independently collected to give a 98:2 *E:Z* mixture of **21**, from which the following data were acquired.

¹H NMR (400 MHz, CDCl₃, major isomer) δ 7.63 (d, *J* = 9.5 Hz, 1H), 7.36 (d, *J* = 8.5 Hz, 1H), 6.88 – 6.78 (m, 2H), 6.24 (d, *J* = 9.5 Hz, 1H), 5.95 – 5.86 (m, 1H), 5.81 (dt, *J* = 15.5, 5.6 Hz, 1H), 4.55 (dd, *J* = 5.5, 0.7 Hz, 2H), 3.72 (t, *J* = 6.3 Hz, 2H), 2.46 – 2.35 (m, 2H), 1.61 (s, 1H). ¹³C NMR (101 MHz, CDCl₃, major isomer) δ 161.9, 161.4, 155.8, 143.6, 132.4, 128.9, 126.5, 113.2, 113.0, 112.6, 101.7, 69.1, 61.7, 35.7. HRMS calcd. for C₁₄H₁₄O₄Na ([M + Na]⁺): 269.0790, found: 269.0785.

S1PL activity with recombinant hS1PL

a) Fluorogenic assay with RBM77 as substrate

hS1PL (50 μL from stock solutions in a 100 mM HEPES buffer pH 7.4, containing 0.1 mM EDTA, 0.05 % Triton X-100, 0.01 % Pluronic F127 (Biotium), and 100 μM pyridoxal 5'-phosphate., final concentration: 0.8 μg/mL) was added to a mixture of

RBM77 (final concentration: 125 μ M) and putative inhibitors (at the indicated concentrations) in the same buffer solution (final volume: 100 μ L). The mixture was incubated at 37 °C for 1 h and the enzymatic reaction was stopped by the addition of 100 μ L of KOH/EtOH (100 mM). After incubation at 37 °C for 20 min, the resulting mixture was treated with 50 μ L of a 200 mM glycine–NaOH buffer, pH 10.6, and the amount of umbelliferone formed was determined on either a SpectraMax M5 (Molecular Devices) or Synergy 2 (BioTek) microplate readers ($\lambda_{\text{ex/em}}$ = 355/460 nm), using a calibration curve. In all cases, the fluorescence of a blank solution, without enzyme, was subtracted from the total fluorescence.

b) Fluorogenic assay with RBM148 as substrate

hS1PL (50 μ L from stock solutions in a 100 mM HEPES buffer pH 7.4, containing 0.1 mM EDTA, 0.05 % Triton X–100, 0.01 % Pluronic F127 (Biotium), and 100 μ M pyridoxal 5'–phosphate., final concentration: 0.8 μ g/mL) was added to a mixture of **RBM148** (final concentration: 125 μ M) and putative inhibitors (at the indicated concentrations) in the same buffer solution (final volume: 100 μ L). The mixture was incubated at 37 °C for 1 h and the enzymatic reaction was stopped by the addition of 50 μ L of MeOH. Finally, 100 μ L of a 200 mM glycine–NaOH buffer, pH 10.6, were added to the resulting solution and the mixture was incubated for 20 additional min at 37 °C in order to complete the β –elimination reaction. The amount of umbelliferone formed was determined as described for **RBM77** (see above).

Kinetic parameters of the enzyme reaction were obtained by fitting the data to the Michaelis–Menten equation in Prism 5 (GraphPad Software, La Jolla). IC₅₀ values were determined by plotting percent activity versus log [I] and fitting the data to the log(inhibitor) vs. response equation in Prism 5 (GraphPad Software, La Jolla). In both cases, settings for curve adjustments were kept with their default values.

Validation of the probes as hS1PL substrates

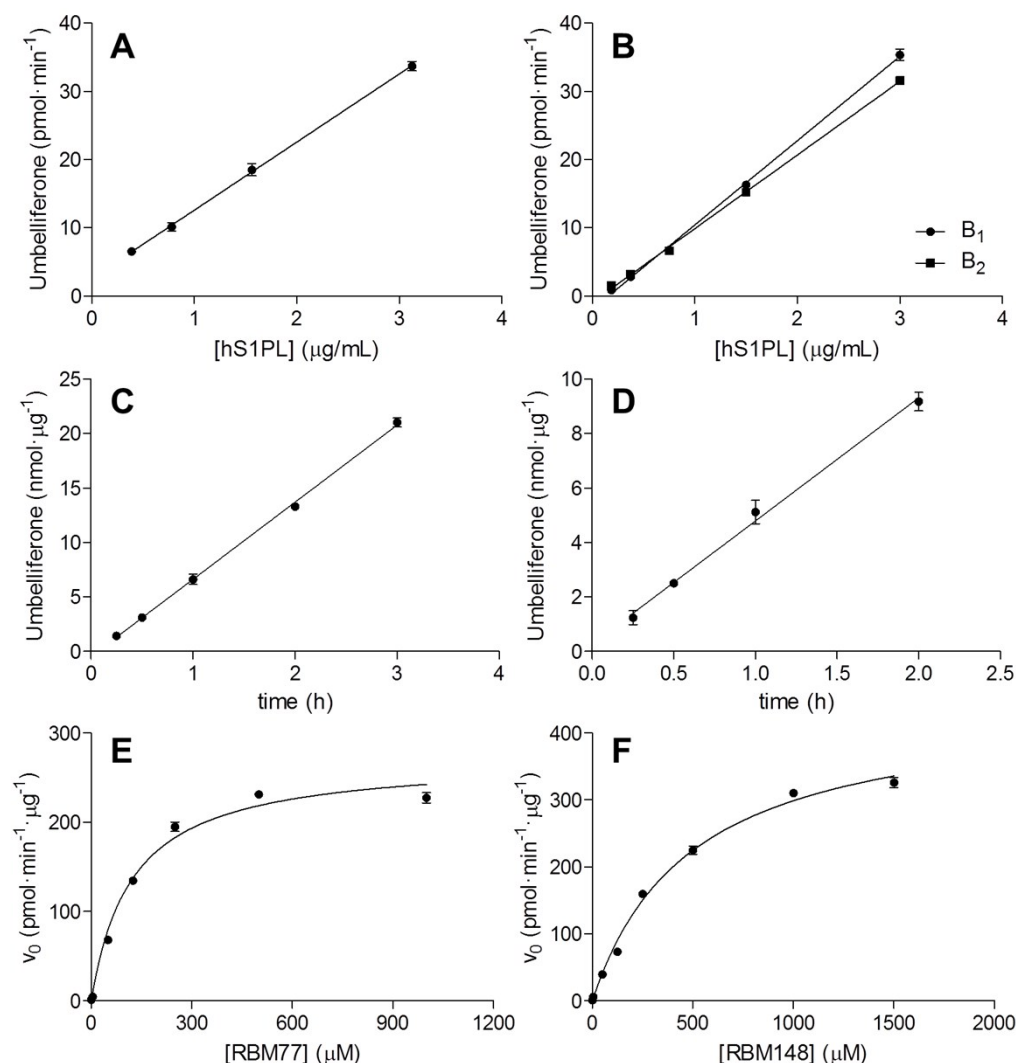


Figure S3. Enzyme concentration (A,B) and time (C,D) dependence of the hS1PL catalyzed reaction using **RBM77** (A,C) and **RBM148** (B,D) as substrates at a final concentration of 125 μM. A and B: Incubation time was 60 min. C and D: hS1PL concentration was 0.8 μg/mL. In B, the resulting enzymatic reaction was stopped with MeOH (50 μL) and treated with 100 μL of 200 mM glycine/NaOH buffer (pH 10.6) (B₁) or alternatively, treated with 100 mM KOH/EtOH (100 μL), incubated for 20 min and the pH was further adjusted with the addition of 50 μL of a 200 mM glycine/NaOH buffer solution (pH 10.6) immediately before reading (B₂). Data are means ± SD of one representative experiment with triplicates. In E and F, the substrate concentration dependence of the initial velocity (V₀, expressed as pmol of umbelliferone formed per minute and per μg of enzyme) of the hS1PL catalyzed reaction is represented. In both cases, hS1PL (0.8 μg/mL) was incubated for 60 min with **RBM77** (E) or **RBM148** (F) at

graded concentrations. Data correspond to the mean \pm SD of one representative experiment out of three performed with triplicates.

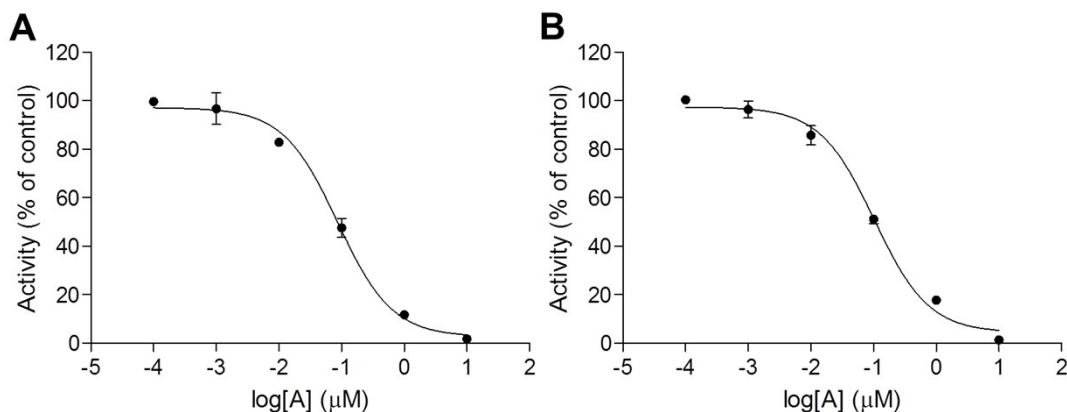


Figure S4. Activity (% of control) of hS1PL (final concentration: 0.8 μ g/mL) at graded concentrations of **A**⁵ using **RBM77** (A) or **RBM148** (B) as substrate (final concentration: 125 μ M). Data correspond to the mean \pm SD of one representative experiment out of three performed with triplicates

Computational Methods

General data

All the molecular modelling was carried out with the package Schrödinger Suite 2016,⁶ through its graphical interface Maestro.⁷ The program Macromodel⁸ with the OPLS3 force-field,⁹ a modified version of the OPLS-AA force-field,^{10,11} and GB/SA water solvation conditions¹² were used for energy minimization. Molecular dynamics simulations were performed with the program Desmond^{13–16} using the OPLS3 force-field.

Coordinates of hS1PL (PDB code 4Q6R)⁵ were obtained from the Protein Data Bank¹⁷ at Brookhaven National Laboratory. The protein X-ray structure was prepared using the Protein Preparation Wizard^{18,19} included in Maestro to remove solvent molecules, ligands and ions, adding hydrogens, setting protonation states²⁰ and minimizing the energies. To model the aldimines, the imine bond between the essential Lys353 and the PLP cofactor was cleaved and structures **20a–c** were manually built on the resulting PLP moiety. The structures were then minimized considering the 3-hydroxypyridine and imino groups of the aldimine-bound PLP prosthetic group in their ionized state. From these, simulation systems for molecular dynamics were built using the System Builder of the Maestro-

Desmond interface,²¹ which automatically assigns parameters to all atom. Each protein-alimine complex was immersed in a 120 x 120 x 120 Å cubic box of TIP3P water with enough Cl⁻ anions to achieve neutrality (~52000 water molecules, ~170000 atoms in total). Systems were relaxed by minimization (initial steepest descent followed by LBFGS minimization), first with the solute restrained and then without restrains, until a gradient threshold of 0.1 kcal mol⁻¹Å⁻¹ was reached. Then, they were heated stepwise up to 300 K with short MD runs under periodic boundary conditions (PBC) (25 ps at 0.1, 10, 100 and 300 K), and equilibrated for 2 ns at the same temperature and 1.0 bar, in the NPT ensemble. Production MD simulations (50 ns, 2 fs timestep) were performed under the same conditions (PBC, NPT ensemble, 300 K and 1.0 bar) using the Nose-Hoover thermostat method^{22,23} with a relaxation time of 1.0 ps and the Martyna-Tobias-Klein barostat method²⁴ with isotropic coupling and a relaxation time of 2 ps. Integration was carried out with the RESPA integrator²³ using time steps of 2.0, 2.0, and 6.0 fs for the bonded, short range and long range interactions, respectively. A cut-off of 9.0 Å was applied to van der Waals and short-range electrostatic interactions, while long-range electrostatic interactions were computed using the smooth particle mesh Ewald method with an Ewald tolerance of 10⁻⁹.^{25,26} Bond lengths to hydrogen atoms were constrained using the Shake algorithm.²⁷ Coordinates were saved every 50 ps, hence 1000 snapshots were obtained from each MD run. The Simulation Event Analysis application included in the Desmond-Maestro interface was used to analyze the trajectories.

Docking studies

Substrates **RBM13**, **RBM77** and **RBM148** were modelled in the form of external aldimines **20a–c** (Figure 1, paper) bound to the hS1PL (PDB 4Q6R) active site. In all cases, the reactive C3–OH group can establish hydrogen bond interactions with the phosphate group of the PLP cofactor and with the ε-NH of the imidazole side chain of H242. Comparison of the spatial arrangement of aldimines **20b** and **20c** revealed that the position of the double bond does not have a significant influence in the disposition of both the linker and the coumarin group.

Molecular dynamics

To further asses the plausibility of such structures, the modelled bound aldimines were used as starting point to run molecular dynamics simulations in explicit water (50 ns, 300

K, periodic boundary conditions, NPT ensemble; see above for complete details). Figure S6 illustrates the results obtained for aldimine **20c**. The simulations showed that the structures were mostly stable after an initial slight rearrangement of the whole protein, with an RMSD around 1.4 Å relative to the starting structure. Analysis of the hydrogen-bond interactions established by aldimine **20c** showed that the hydrogen bond between the C3-OH and the PLP-phosphate was kept for more than 95 % of the simulation time whilst that between the C3-OH and H242 was established only intermittently (Figure S6B). Similar results were obtained with the complexes from aldimines **20b** and **20a** (Figures S5 and S7, respectively), although in those cases an additional intermittent hydrogen bond interaction with T387 was also observed. Thus, these simulations would suggest that the PLP-phosphate group could be the base responsible for the C3-OH deprotonation that is previous to the C-C bond cleavage in the enzymatic cycle. The only other candidate base that apparently is close enough to the reactive C3-OH would be H242. However, this seems less likely, since the crystal structure of hS1PL shows that the δ -N of its imidazole ring accepts a hydrogen bond from the backbone amide of A244, thus forcing the ϵ -N-atom to be protonated and becoming unable to act as acceptor of another proton. Instead, as observed in our simulations, the ϵ -NH of the imidazole is suitable to act as a hydrogen-bond donor if an acceptor group is placed appropriately, such as the C3-OH present in aldimines **20a-c**. Therefore, it seems reasonable to propose that the hydrogen-bond interaction between H242 and the C3-OH of the substrate, as well as that formed with Y387, could serve to increase the acidity of this reactive OH, thus promoting the cleavage of the C-C bond and the concerted release of the corresponding aldehyde, both essential steps of the catalytic cycle of S1PL.

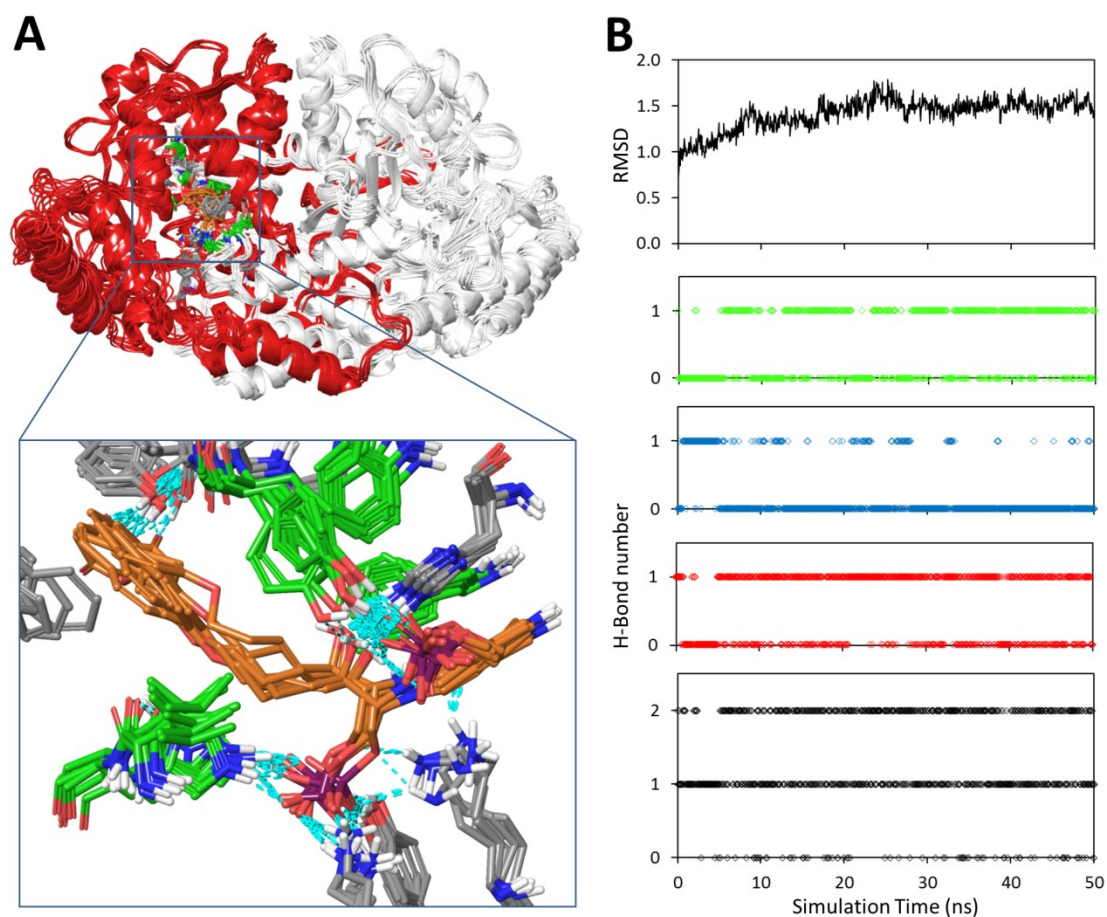


Figure S5: (A) Snapshots from the 50 ns MD simulation of the complex between aldimine **20b** (orange) and hS1PL (gray, residues forming a hydrophobic patch close to the PLP-binding site are highlighted in green). (B) RMSD plot and dependence of the number of hydrogen bond interactions established between the C3-OH of aldimine **20b** and the rest of the protein (black), the PLP phosphate group (red), His242 (blue) or Tyr387 (green), with simulation time.

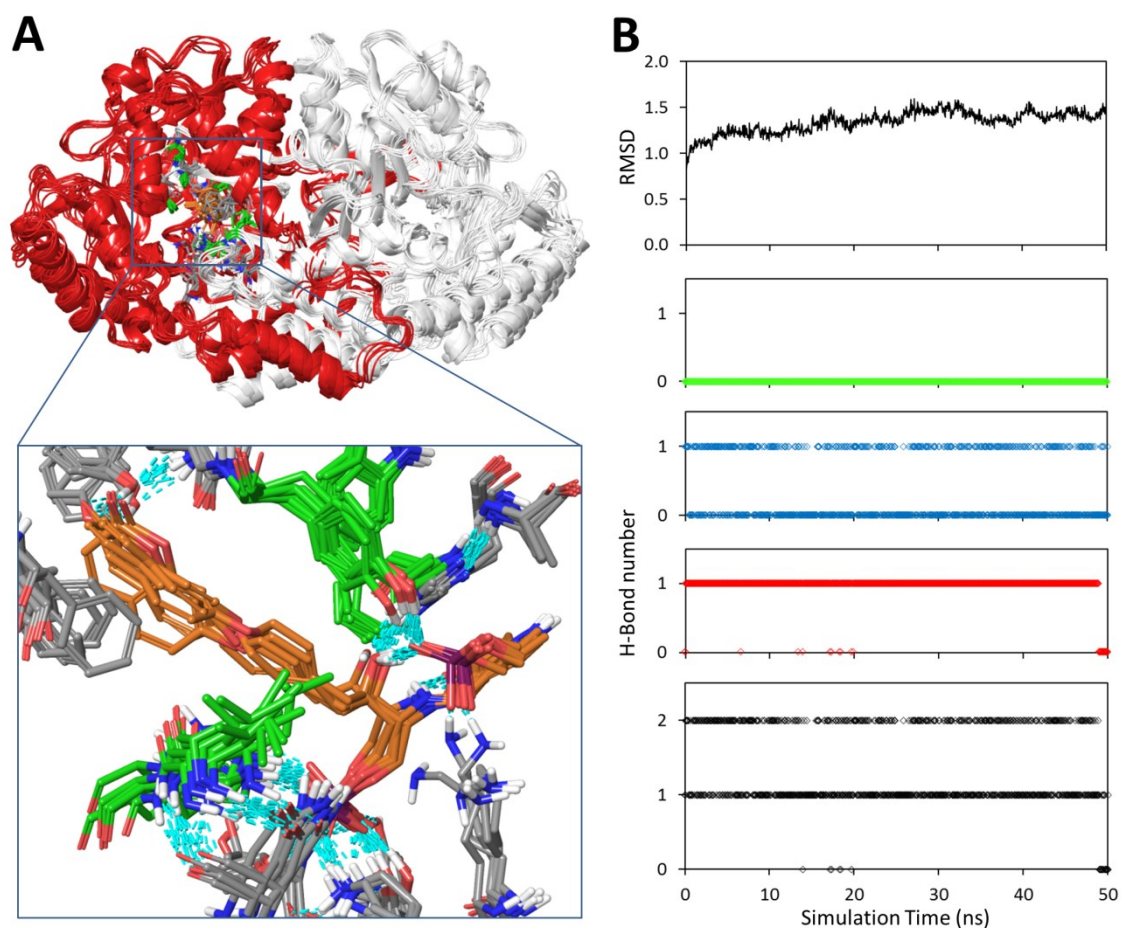


Figure S6. (A) Snapshots from the 50 ns MD simulation of the complex between aldimine **20c** (orange) and hS1PL (gray, residues forming a hydrophobic patch close to the PLP-binding site are highlighted in green). (B) RMSD plot and time dependence of the number of hydrogen bond interactions between the C3-OH of aldimine **20c** and the rest of the protein (black), the PLP phosphate group (red), H242 (blue) or Y387 (green)..

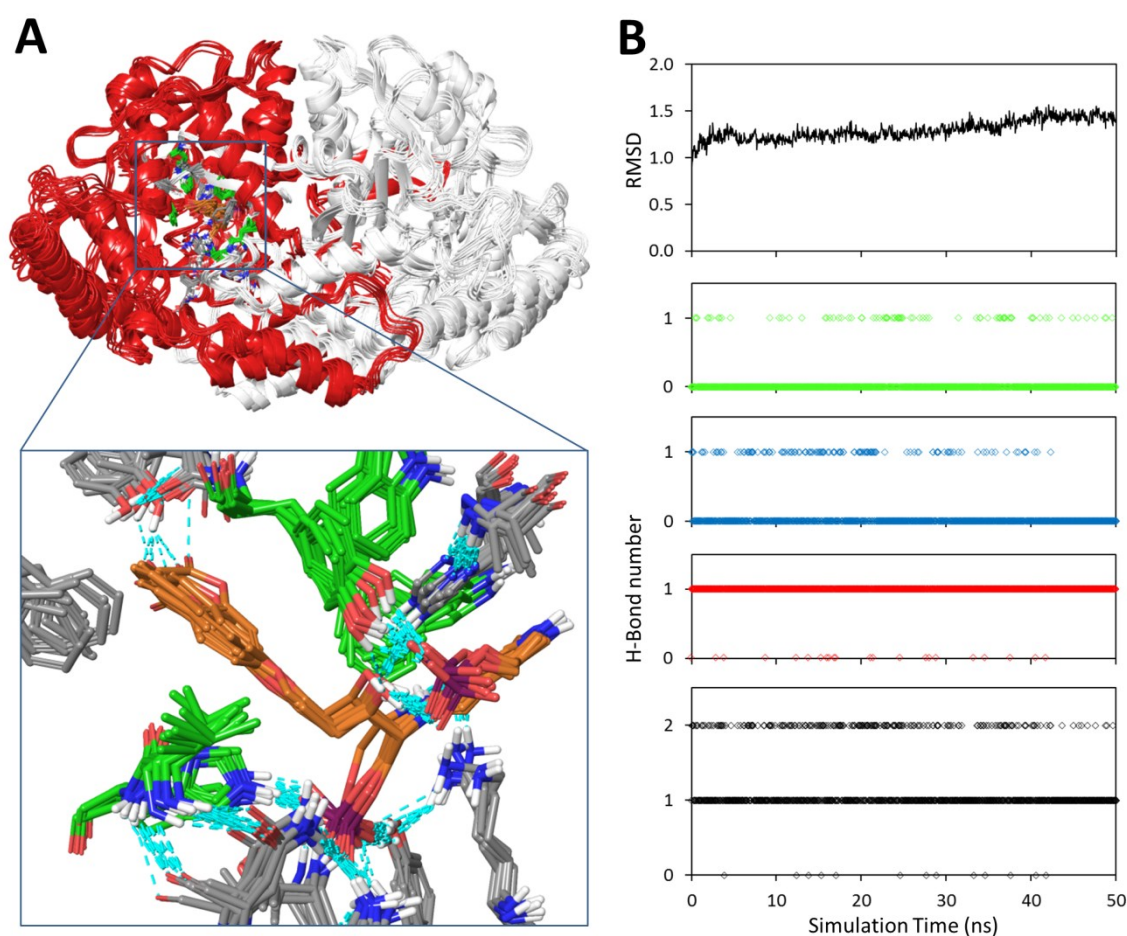


Figure S7: (A) Snapshots from the 50 ns MD simulation of the complex between aldimine **20a** (orange) and hS1PL (gray, residues forming a hydrophobic patch close to the PLP-binding site are highlighted in green). (B) RMSD plot and dependence of the number of hydrogen bond interactions established between the C3-OH of aldimine **20a** and the rest of the protein (black), the PLP phosphate group (red), His242 (blue) or Tyr387 (green), with simulation time.

Cell assays

Dulbecco's modified Eagle's medium, fetal bovine serum, penicillin/streptomycin solution, Hank's balanced salt solution were from Sigma. Opti-MEM and Lipofectamine 2000 were from Invitrogen. *N*-[1-(2,3-dioleoyloxy)propyl]-*N,N,N*-trimethylammonium methyl sulfate (DOTAP) and Lipoid S-100, whose main component (>94%) is soybean phosphatidylcholine, were from Lipoid GmbH. pCMV6 vector harboring the mouse *Sgpl1* gene was from OriGene.

Human embryonic kidney cells HEK293T and mouse embryo fibroblasts MEF were cultured at 37°C in 5% CO₂ in Dulbecco's modified Eagle's medium supplemented with 10% fetal bovine serum and 100 ng/mL each of penicillin and streptomycin.

Overexpression of *Sgpl1*

HEK29T cells, plated in 12-well plates (3x10⁵ cells per well) for 24 h, were transfected with 1 µg/well of pCMV6 vector harbouring the mouse *Sgpl1* gene using opti-MEM/Lipofectamine 2000 (L2000), following the manufacturer's instructions. Control cells were treated only with the transfection reagent L2000. Test compounds were added 48 h after transfection.

Cell lysates

Cell pellets were resuspended in the appropriate volume of a 25 µM Na₃VO₄ solution in 0.2 M phosphate buffer (pH 7.4). The suspension was submitted to three cycles of a 5 s sonication (probe) at 10 watts/5 s resting on ice. The cell lysate was centrifuged at 600 g for 5 min. The supernatant was collected and protein concentration was determined as specified below.

Determination of protein concentration

Protein concentrations were determined with BSA as a standard using a BCA protein determination kit (Thermo Scientific) according to the manufacturer's instructions.

S1PL activity in cell lysates

S1PL activity determination in cell lysates was done in 96-well plates at a final volume of 100 µL/well. The reaction mixture contained per well 100 µg of cell lysate, 125 µM of substrate (5 µL/well from 2.5 mM stock solution), 250 µM of pyridoxal 5'-phosphate (5 µL/well from 5 mM stock solution) and up to 100 µL of phosphate buffer. The reaction

mixture was incubated at 37°C for 1 h. The enzymatic reaction was stopped with 50 µL/well of methanol and then 100 µL of 200 mM glycine-NaOH buffer (pH 10.6) were added. After incubation at 37°C for 20 min in the dark, fluorescence was measured spectrophotometrically at excitation and emission wavelengths of 355 and 460 nm, respectively. The same reaction mixture without cell lysate was used as blank. Using **RBM77** as substrate, the reaction was stopped with 100 µL/well of 100 mM KOH-MeOH solution and the plate was incubated at 37°C for 30 min in the dark. Then 50 µL of 200 mM glycine-NaOH buffer (pH 10.6) were added and the fluorescence was measured at the same wavelengths.

S1PL activity in intact cells

Cells ($2-3 \times 10^5$ per well) were seeded in a 12-well plate. MEF cells were incubated for 24 h at 37°C and 5% CO₂. HEK293T cells were transfected with the *Sgpl1* encoded plasmid as mentioned above. Medium was replaced by 300 µL of HBSS to which the required volume of a 3.5 mM stock solution of the probes liposomes had been added to obtain the desired probe concentration. The plate was incubated for 3 h at 37°C and 5% CO₂. The reaction was stopped with 250 µL/well of methanol and then 500 µL of 200 mM glycine-NaOH buffer (pH 10.6) were added. After incubation at 37°C for 30 min in the dark, fluorescence was measured spectrophotometrically at excitation and emission wavelengths of 355 and 460 nm, respectively. The same reaction mixture without cells was used as blank.

Liposome preparations

Cationic liposomes were formulated as a 1:1 mixture of phosphatidylcholine and DOTAP at 3.5 mM total lipid concentration. The lipids were dissolved and mixed in chloroform. Once the lipids were mixed in the organic solvent, the solvent was removed by rotary evaporation yielding a thin lipid film on the sides of a round bottom flask. The lipid film was thoroughly dried to remove residual organic solvent by placing the flask on a vacuum pump overnight. The film was hydrated with 1 or 2 mM HBSS solutions of the probes. Three dilutions of liposomes were made in each case to give final total lipid concentrations of 350 µM, 700 µM and 1000 µM, containing probe concentrations of 100 µM, 200 µM and 290 µM, respectively (from the initial 1 mM probe solution), and 200 µM, 400 µM and 580 µM, respectively (from the initial 2 mM probe solution).

The lipid suspension was then placed on a rotatory evaporation system without vacuum and kept turning in warm water with gentle agitation. The product of hydration is a

dispersion of large vesicles but with cycles of 5 min sonication smaller liposomes were obtained.

Measurement of liposomes size distribution

The average size and size distribution of liposomes were measured by dynamic light scattering (DLS) using a Zetasizer Nano ZS (Malvern Instruments). The measurements by DLS were performed at room temperature (see Figure S8).

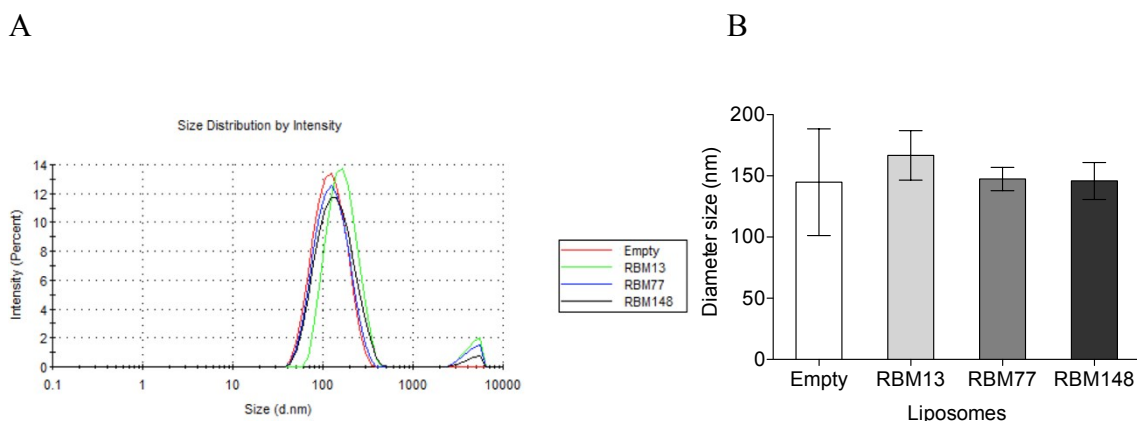


Figure S8. Particle size distribution curves of liposome suspension analysed by intensity.

A. Size distribution curves of different liposome formulation (empty and encapsulating **RBM13**, **RBM77** and **RBM148** probes) were measured using a Zetasizer Nano ZS instrument. Curves correspond to one representative measurement. **B.** Average diameters of liposomes. Data are the mean \pm SD of three experiments with triplicates. There is no statistical difference in diameters between liposome preparations: empty (145 \pm 40 nm), **RBM13** (167 \pm 20 nm), **RBM77** (147 \pm 9 nm) and **RBM148** (146 \pm 15 nm).

Determination of encapsulation efficiency

The encapsulation efficiency (EE) of liposomes was determined referring to the following equation $EE (\%) = (T-F) \times 100/T$, where T is the total probe and F is the free probe amount. Separation of liposome-incorporated and free probe was accomplished by centrifugal ultrafiltration. Liposomal suspension containing encapsulated probes as well as free probes were loaded into Amicon 3K (Millipore) centrifuge ultrafiltration units and centrifuged at 21°C for 30 min at 4500 g. An aliquot from total preparation before centrifugation and from the filtrate (after centrifugation) were collected and analyzed by UPLC-MS and the amount of RBM148 and its dephosphorylated derivative was quantified (see Figure S9).

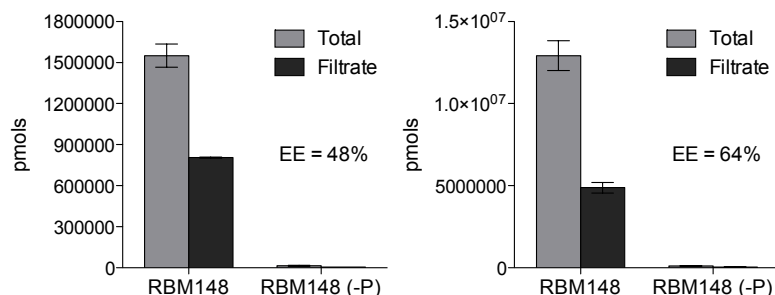


Figure S9. Entrapment efficiency analysis of **RBM148** encapsulated liposomes. Encapsulation efficiency of two different **RBM148** liposomes composition was calculated. 3.5 mM (A) or 7 mM (B) of lipid both containing 2 mM of **RBM148**. Each preparation was analyzed by UPLC-MS. Data are the mean \pm SD of three experiments with duplicates.

Effect of liposomes and encapsulated RBM148 on cell viability

a) Empty liposomes: Cell viability was determined by cell counting. MEF SGPL1^{+/+} cells were seeded in 12--well plates at a density of 2×10^5 cells per well and HEK293T cells were used after 48 h of *Sgpl1* transfection. Cells were exposed to different empty liposome dilutions for 3h. At the end of the treatment, cells were washed with 200 μ L PBS and harvested with 200 μ L Trypsin-EDTA and 300 μ L of medium. Cells were counted using a Neubauer chamber (see Figure S10)

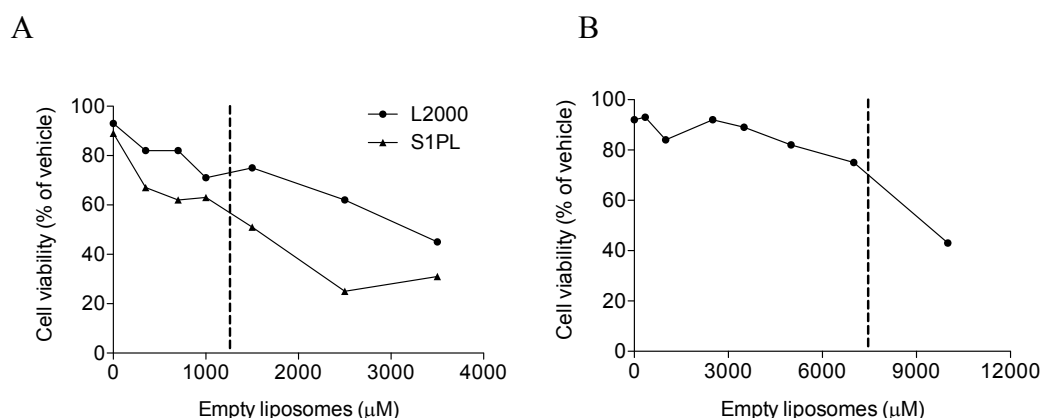


Figure S10. Effect of liposomes on cell viability. HEK293T cells overexpressing S1PL (A) and MEF SGPL1^{+/+} cells (B) were incubated with empty liposomes for 3 h and cell viability was determined by cell counting. Dotted line showed the maximum allowable

liposome concentration (based on toxicity assay) for each cell line; L2000: Lipofectamine 2000.

Due to the transfection agent (lipofectamine 2000), HEK293T cells were more sensitive to liposomes, reaching higher cytotoxicity levels. On the other hand, on MEF SGPL1^{+/+} cells, cell viability was not affected up to a liposome concentration of 7 mM. This result was extrapolated to other MEF cells used in this work (SGPL1^{+/-} and SGPL1^{-/-}).

b) Encapsulated RBM148 and comparison of cell viability with empty liposomes. No significant effect on cell viability is observed between empty and loaded liposomes.

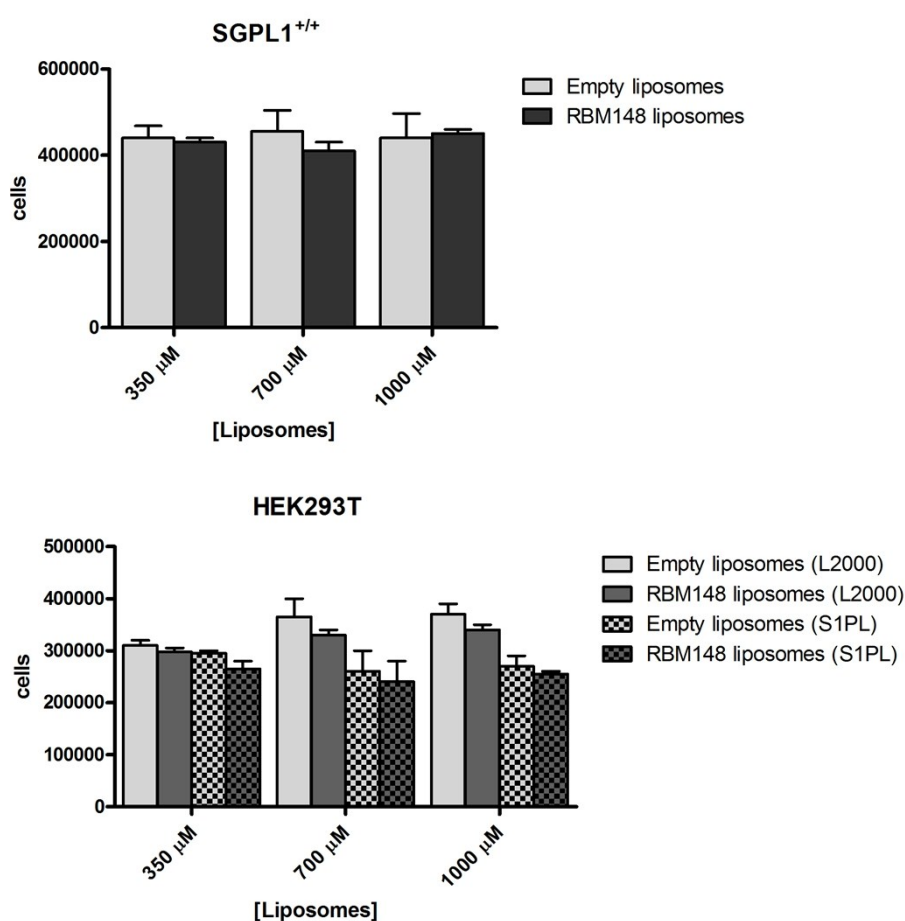


Figure S11. Effect of empty and RBM148 encapsulated liposomes on cell viability. HEK293T overexpressing S1PL and S1PL^{+/+} MEF cells were incubated with 350 μ M, 700 μ M and 1000 μ M of empty liposomes or liposomes containing RBM148 200 μ M, 400 μ M and 580 μ M, respectively, for 3 h. Cell viability was determined by cell counting. Data are the average \pm SD of two experiments.

S1PL activity in intact cells with encapsulated RBM77

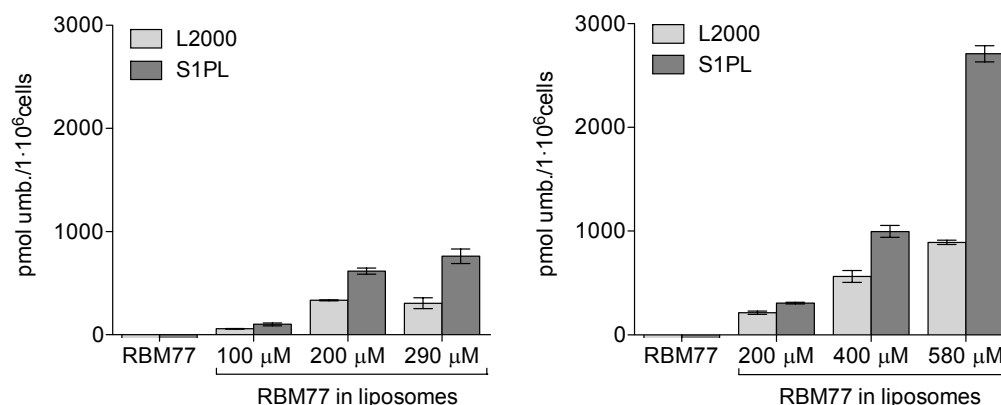


Figure S12. S1PL activity determination in intact cells using **RBM77** encapsulated liposomes. HEK293T overexpressing S1PL were incubated 48 h after transfection with different dilution of encapsulated liposomes obtained from 1 mM (**left**) or 2 mM (**right**) **RBM77** solution for 3 h. Non-encapsulated **RBM77** at 200 μM was also tested. Data are the average \pm SD of three experiments with duplicates.

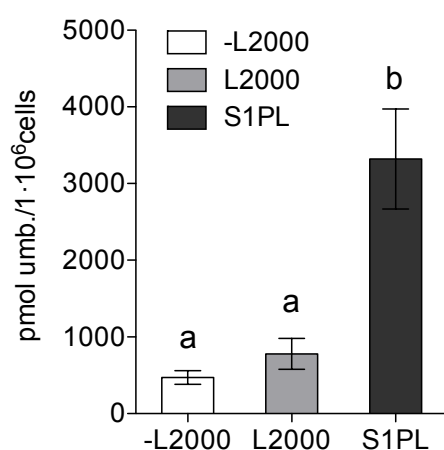


Figure S13. Amounts of umbelliferone released in non-transfected (L2000) and S1PL overexpressing (SPL) HEK293T cells incubated with 200 μM **RBM148** encapsulated in liposomes for 3 h. -L2000, cells used as a control without lipofectamine 2000[®]. Data are the average \pm SD of two experiments with duplicates. Data were analyzed by one way ANOVA following by Bonferroni's multiple comparison post-test if ANOVA $P < 0.05$. Different letters atop each denote statistical significance ($P < 0.05$).

References

- (1) Yamamoto, T.; Hasegawa, H.; Hakogi, T.; Katsumura, S. Versatile Synthetic Method for Sphingolipids and Functionalized Sphingosine Derivatives via Olefin Cross Metathesis. *Org Lett* **2006**, 8 (24), 5569–5572.
- (2) Fransson, R.; Mccracken, A. N.; Chen, B.; Mcmonigle, R. J.; Edinger, A. L.; Hanessian, S. Design, Synthesis, and Antileukemic Activity of Stereochemically Defined Constrained Analogues of FTY720 (Gilenya). *ACS Med. Chem. Lett.* **2013**, 4, 969–973.
- (3) Magolan, J.; Coster, M. J. Total Synthesis of (+)-Angelmanin. *J. Org. Chem.* **2009**, 74 (14), 5083–5086.
- (4) Wahler, D.; Badalassi, F.; Crotti, P.; Reymond, J. L. Enzyme Fingerprints by Fluorogenic and Chromogenic Substrate Arrays. *Angew Chem Int Ed Engl* **2001**, 40 (23), 4457–4460.
- (5) Weiler, S.; Braendlin, N.; Beerli, C.; Bergsdorf, C.; Schubart, A.; Srinivas, H.; Oberhauser, B.; Billich, A. Orally Active 7-Substituted (4-Benzyl-Phthalazin-1-Yl)-2-Methyl-Piperazin-1-Yl]-Nicotinonitriles as Active-Site Inhibitors of Sphingosine-1-Phosphate Lyase for the Treatment of Multiple Sclerosis. *J. Med. Chem.* **2014**, 57 (12), 5074–5084.
- (6) Schrödinger. Release 2016-1. Schrödinger, LLC: New York, NY 2016.
- (7) Schrödinger. Release 2016-1: Maestro. Schrödinger, LLC: New York, NY 2016.
- (8) Schrödinger. Release 2016-1: MacroModel. Schrödinger, LLC: New York, NY 2016.
- (9) Harder, E.; Damm, W.; Maple, J.; Wu, C.; Reboul, M.; Xiang, J. Y.; Wang, L.; Lupyan, D.; Dahlgren, M. K.; Knight, J. L.; Kaus, J. W.; Cerutti, D. S.; Krilov, G.; Jorgensen, W. L.; Abel, R.; Friesner, R. A. OPLS3: A Force Field Providing Broad Coverage of Drug-like Small Molecules and Proteins. *J Chem Theory Comput* **2016**, 12 (1), 281–296.
- (10) Jorgensen, W. L.; Maxwell, D. S.; TiradoRives, J. Development and Testing of the OPLS All-Atom Force Field on Conformational Energetics and Properties of

Organic Liquids. *J. Am. Chem. Soc.* **1996**, *118* (45), 11225–11236.

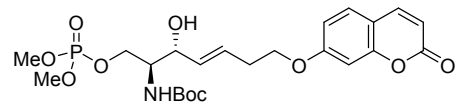
- (11) Kaminski, G. A.; Friesner, R. A.; Tirado-Rives, J.; Jorgensen, W. L. Evaluation and Reparametrization of the OPLS-AA Force Field for Proteins via Comparison with Accurate Quantum Chemical Calculations on Peptides. *J. Phys. Chem. B* **2001**, *105*, 6474–6487.
- (12) Still, W. C.; Tempczyk, A.; Hawley, R. C.; Hendrickson, T. Semianalytical Treatment of Solvation for Molecular Mechanics and Dynamics. *J. Am. Chem. Soc.* **1990**, *112* (16), 6127–6129.
- (13) Schrödinger. Release 2016-1: Desmond Molecular Dynamics System. D. E. Shaw Research: New York, NY 2016.
- (14) Bowers, K. J.; Chow, E.; Xu, H.; Dror, R. O.; Eastwood, M. P.; Gregersen, B. A.; Klepeis, J. L.; Kolossváry, I.; Moraes, M. A.; Sacerdoti, F. D.; Salmon, J. K.; Shan, Y.; Shaw, D. E. Scalable Algorithms for Molecular Dynamics Simulations on Commodity Clusters. In *Proceedings of the ACM/IEEE Conference on Supercomputing (SC06)*; Tampa, Florida, 2006.
- (15) Guo, Z.; Mohanty, U.; Noehre, J.; Sawyer, T. K.; Sherman, W.; Krilov, G. Probing the Alpha-Helical Structural Stability of Stapled p53 Peptides: Molecular Dynamics Simulations and Analysis. *Chem Biol Drug Des* **2010**, *75* (4), 348–359.
- (16) Shivakumar, D.; Williams, J.; Wu, Y.; Damm, W.; Shelley, J.; Sherman, W. Prediction of Absolute Solvation Free Energies Using Molecular Dynamics Free Energy Perturbation and the OPLS Force Field. *J Chem Theory Comput* **2010**, *6* (5), 1509–1519.
- (17) Berman, H. M.; Westbrook, J.; Feng, Z.; Gilliland, G.; Bhat, T. N.; Weissig, H.; Shindyalov, I. N.; Bourne, P. E. The Protein Data Bank. *Nucleic Acids Res* **2000**, *28* (1), 235–242.
- (18) Schrödinger. Release 2016-1: Schrödinger Suite 2016-1 Protein Preparation Wizard; Epik, Schrödinger, LLC, New York, NY, 2016; Impact, Schrödinger, LLC, New York, NY, 2016; Prime. Schrödinger, LLC: New York, NY 2016.
- (19) Sastry, G. M.; Adzhigirey, M.; Day, T.; Annabhimoju, R.; Sherman, W. Protein

and Ligand Preparation: Parameters, Protocols, and Influence on Virtual Screening Enrichments. *J. Comput. Aided. Mol. Des.* **2013**, 27 (3), 221–234.

- (20) Olsson, M. H. M.; Søndergaard, C. R.; Rostkowski, M.; H., J. J.; Søndergaard, C. R.; Rostkowski, M.; Jensen, J. H. PROPKA3: Consistent Treatment of Internal and Surface Residues in Empirical pKa Predictions. *J. Chem. Theor. Comput.* **2011**, 7 (2), 525–537.
- (21) Schrödinger. Release 2016-1: Maestro-Desmond Interoperability Tools. D. E. Shaw Research: New York, NY 2016.
- (22) Evans, D. J.; Holian, B. L. The Nose–Hoover Thermostat. *J. Chem. Phys.* **1985**, 83 (8), 4069–4074.
- (23) Martyna, G. J.; Klein, M. L.; Tuckerman, M. Nosé–Hoover Chains: The Canonical Ensemble via Continuous Dynamics. *J. Chem. Phys.* **1992**, 97 (4), 2635–2643.
- (24) Martyna, G. J.; Tobias, D. J.; Klein, M. L. Constant Pressure Molecular Dynamics Algorithms. *J. Chem. Phys.* **1994**, 101 (5), 4177–4189.
- (25) Darden, T.; York, D.; Pedersen, L. Particle Mesh Ewald: An N·log(N) Method for Ewald Sums in Large Systems. *J. Chem. Phys.* **1993**, 98 (12), 10089–10092.
- (26) Essmann, U.; Perera, L.; Berkowitz, M. L.; Darden, T.; Lee, H.; Pedersen, L. G. A Smooth Particle Mesh Ewald Method. *J. Chem. Phys.* **1995**, 103 (19), 8577–8593.
- (27) Kräutler, V.; van Gunsteren, W. F.; Hünenberger, P. H. A Fast SHAKE Algorithm to Solve Distance Constraint Equations for Small Molecules in Molecular Dynamics Simulations. *J. Comput. Chem.* **2001**, 22 (5), 501–508.

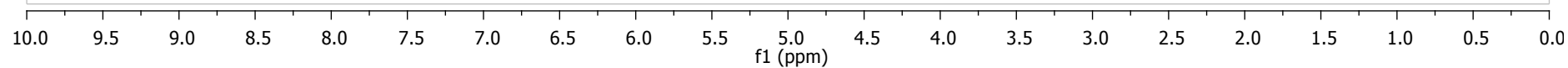
NMR SPECTRA

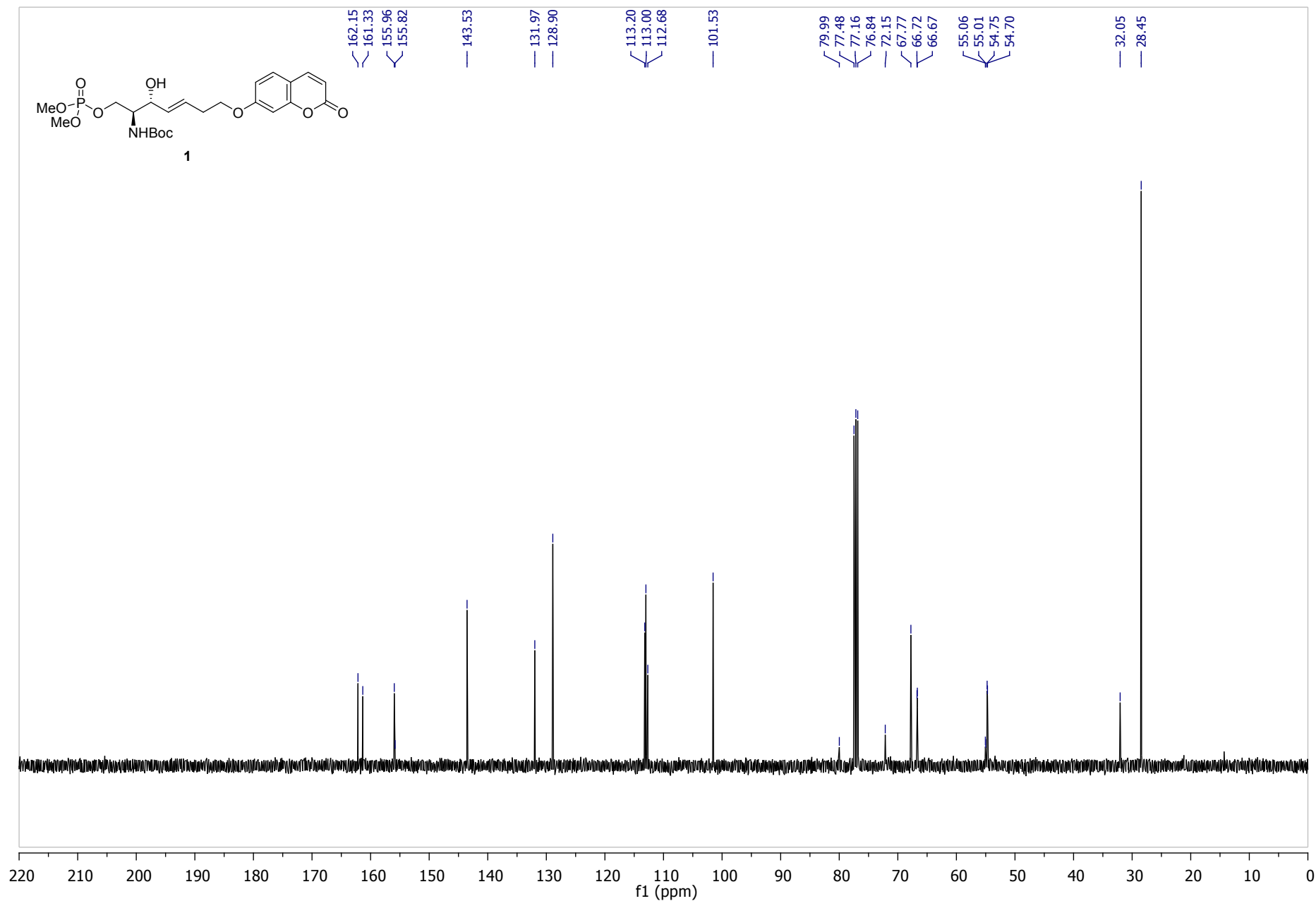
NMR spectra were recorded at room temperature on a Varian Mercury 400 instrument. The chemical shifts (δ) are reported in ppm relative to the solvent signal, and coupling constants (J) are reported in Hertz (Hz). ^{31}P chemical shifts are relative to a 85 % H_3PO_4 external reference (0 ppm). The following abbreviations are used to define the multiplicities in ^1H NMR spectra: s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, ddd = doublet of doublet of doublets, m = multiplet, br = broad signal and app = apparent

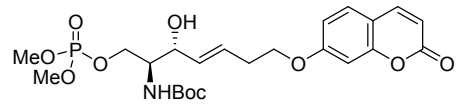


1

7.26

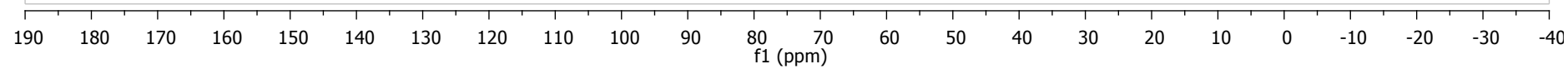


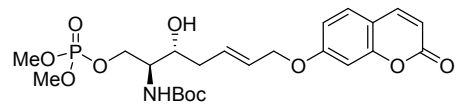




1

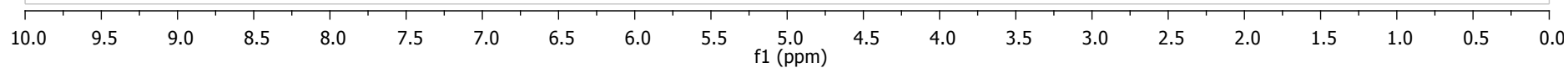
— 2.65

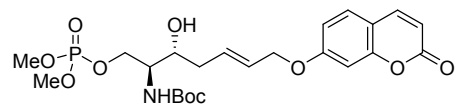




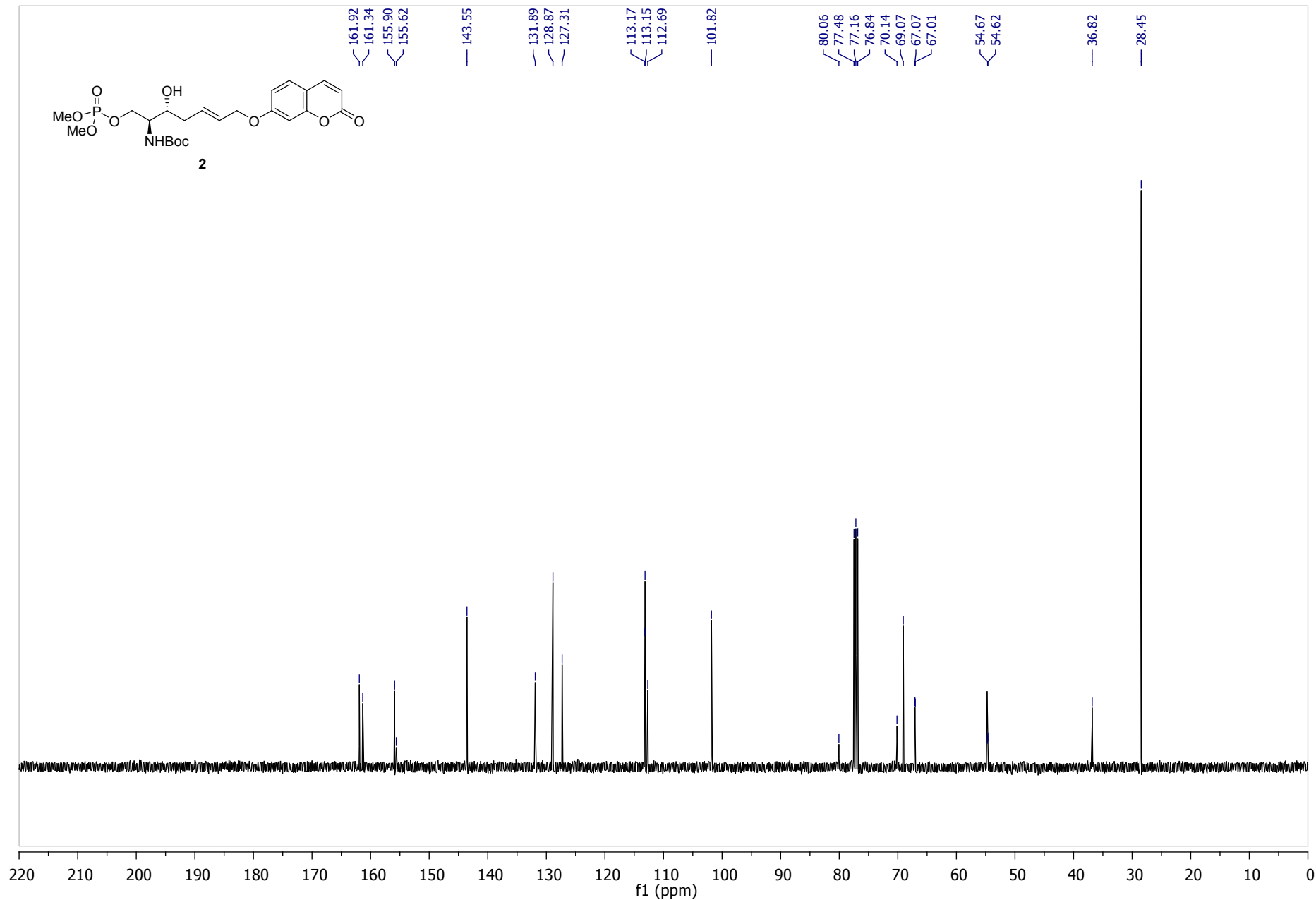
2

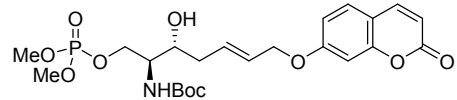
7.26





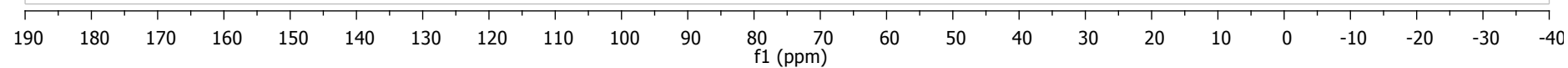
2

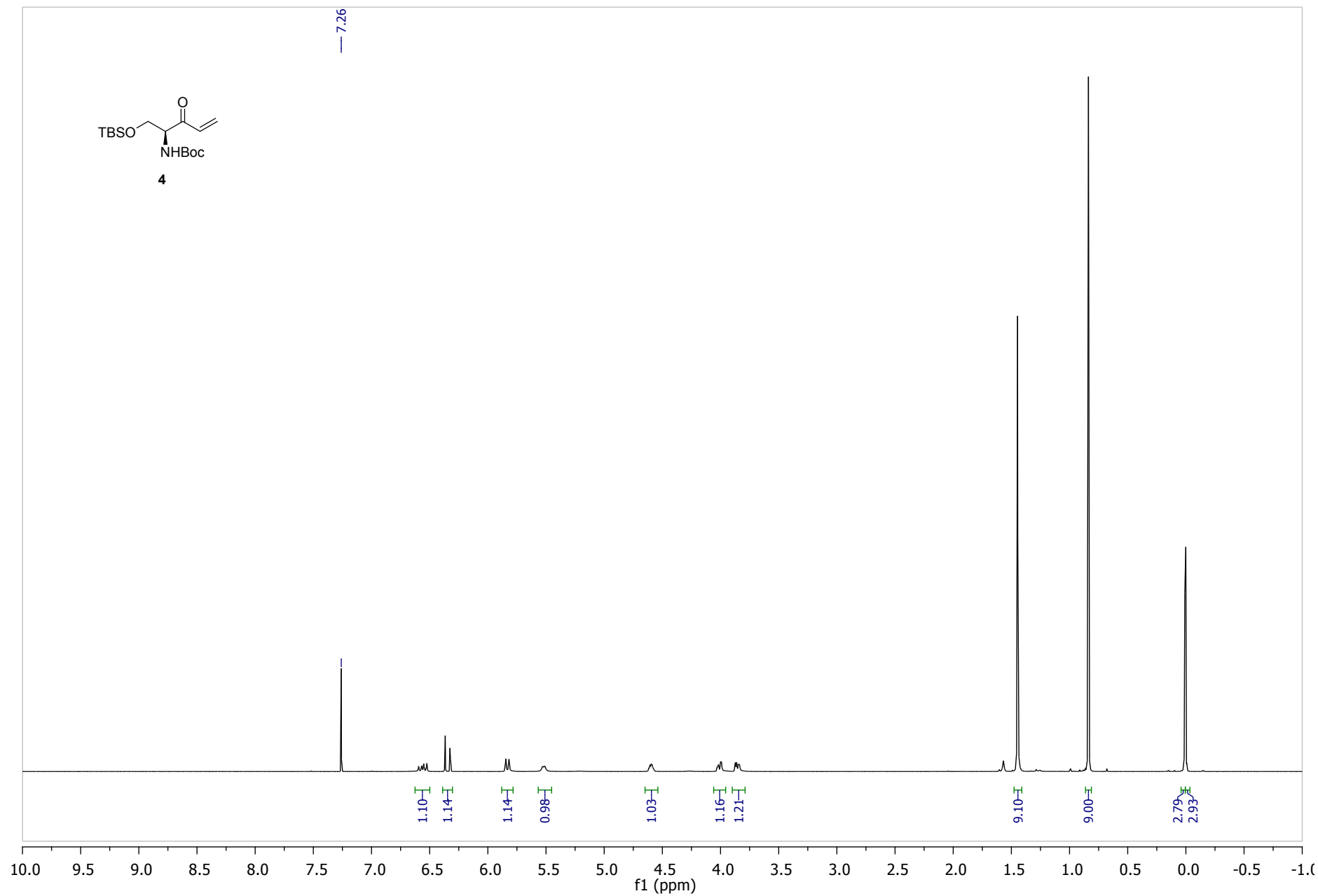
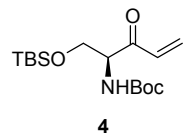


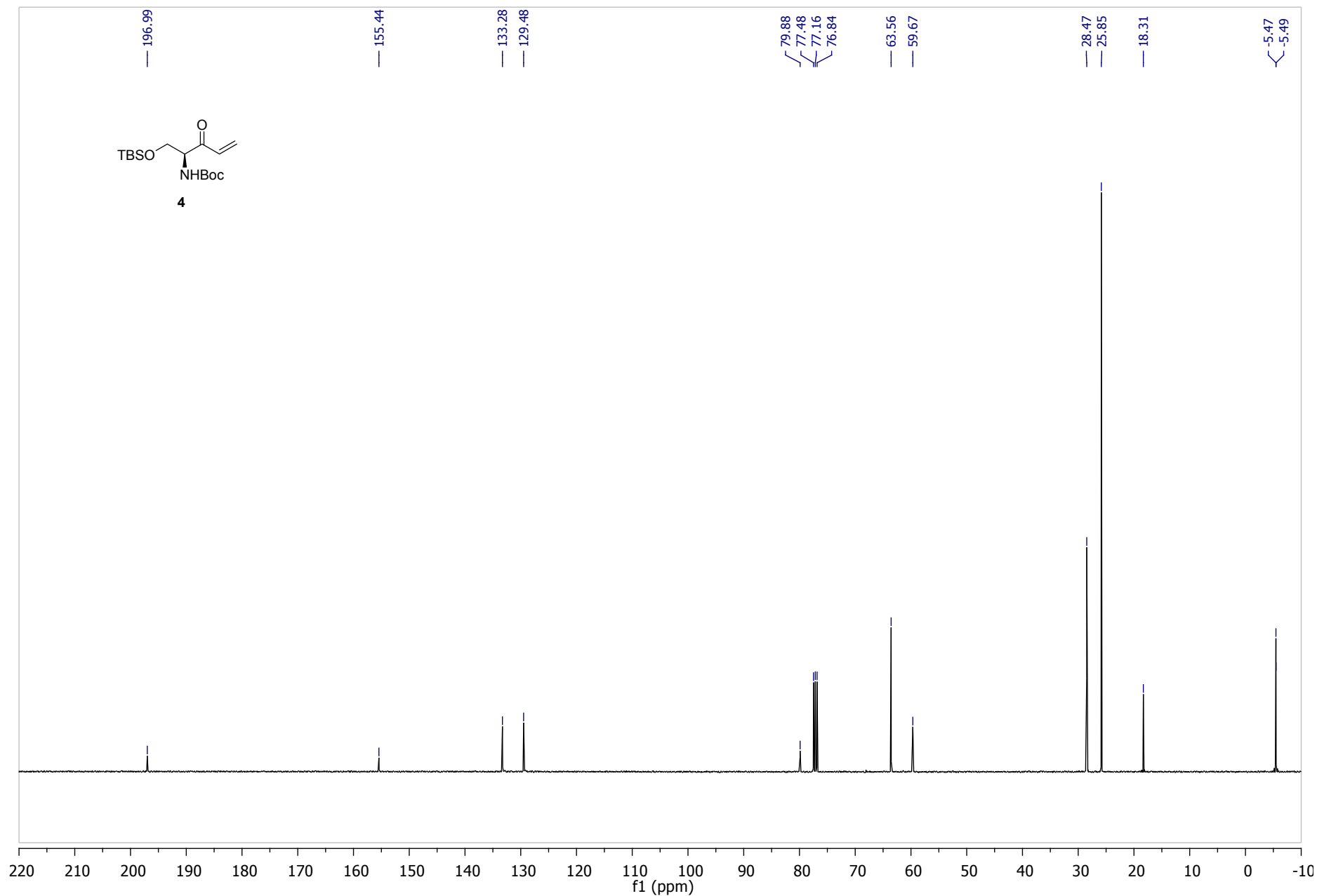
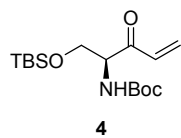


2

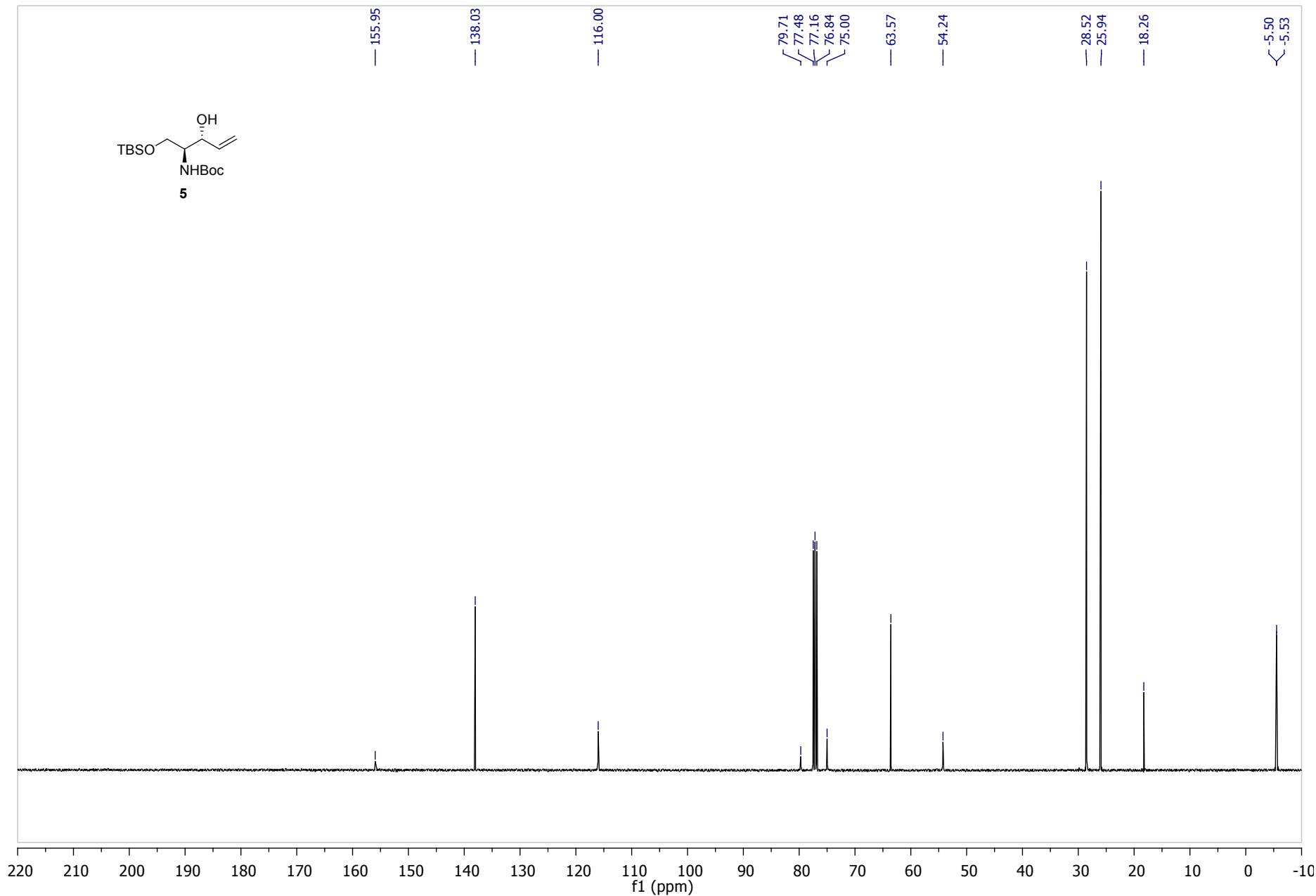
— 2.96

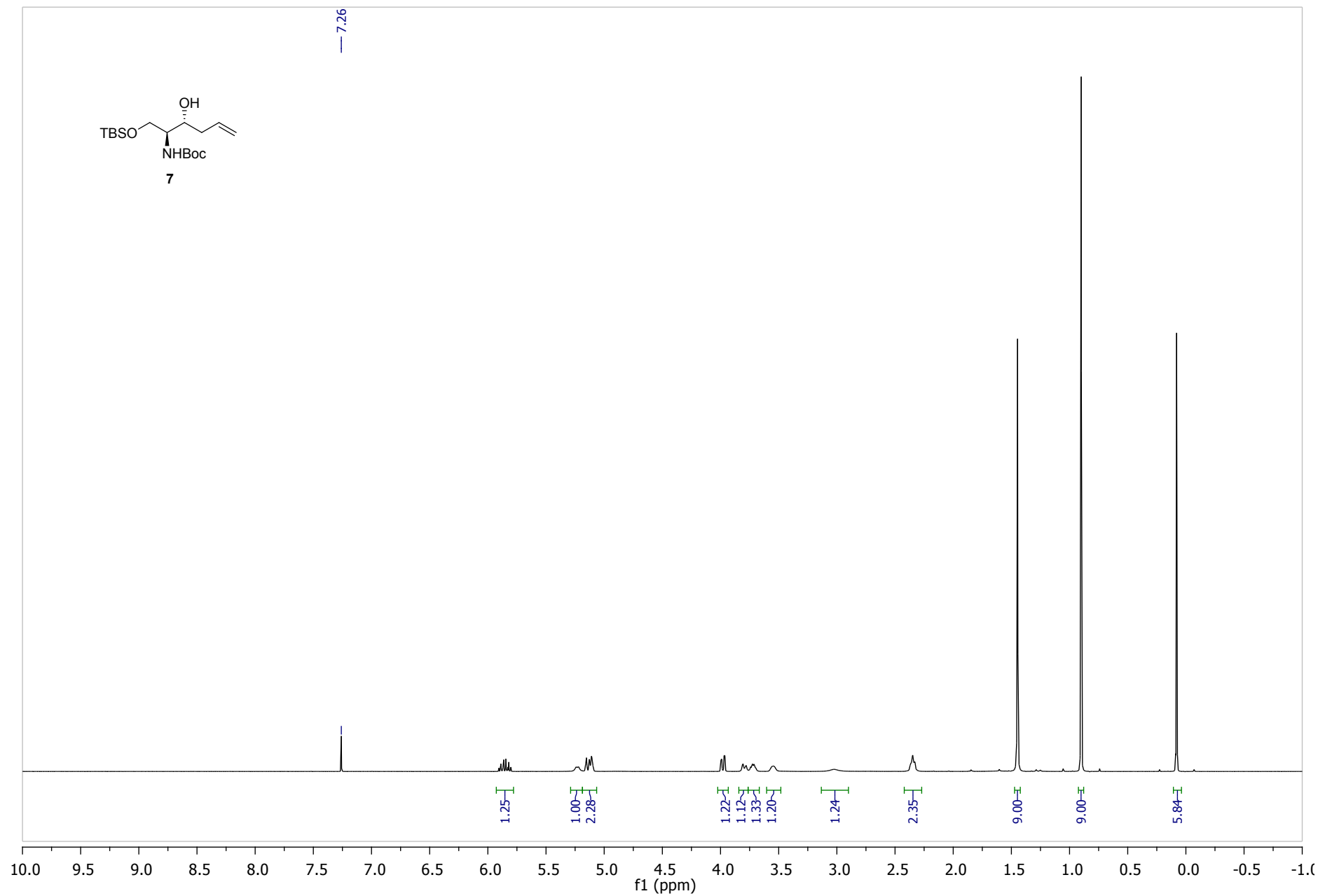
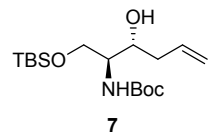














— 134.73

— 117.90

- 79.59

77.48

77.16

70.64
73.07

— 63.36

—53.81

— 39.49

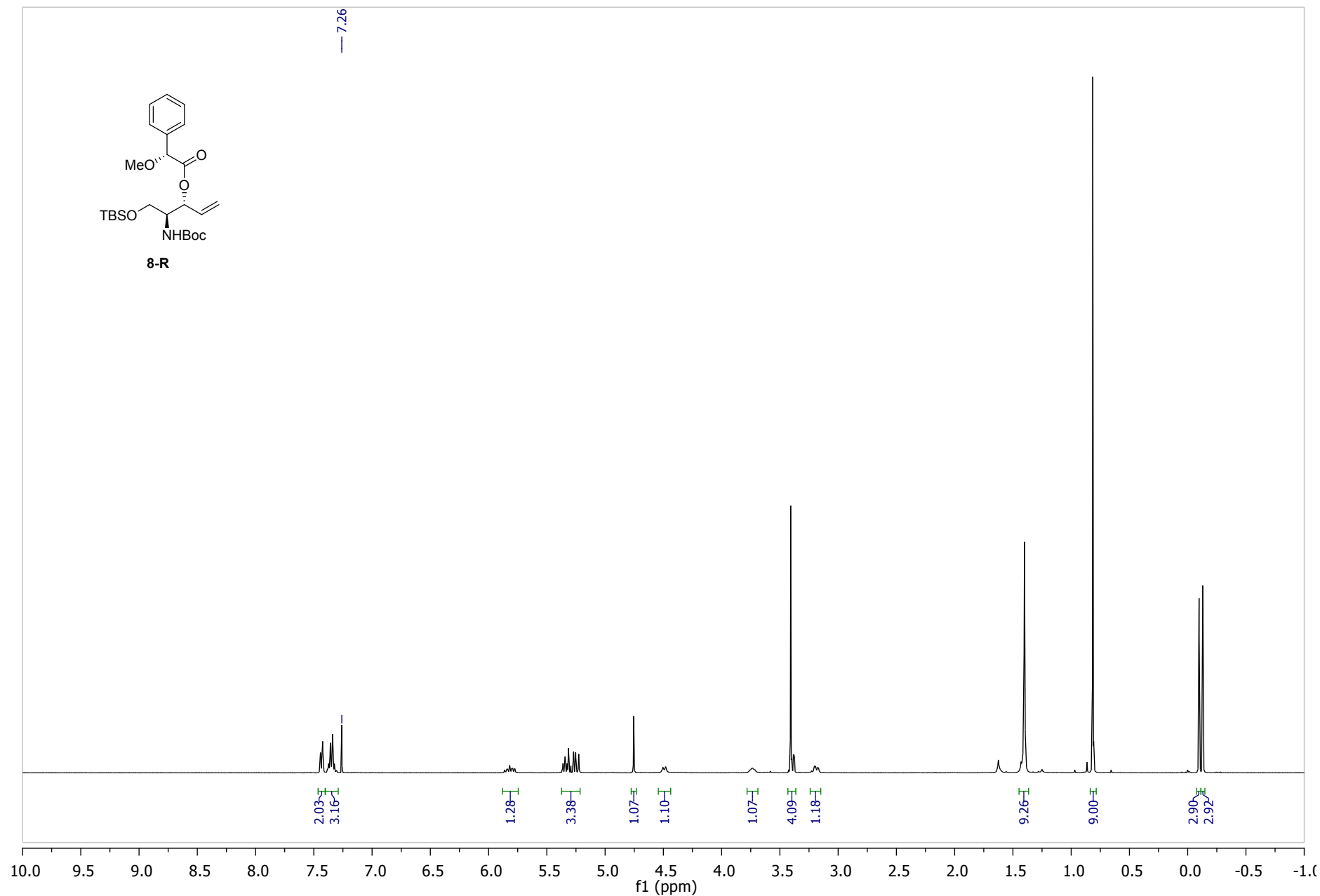
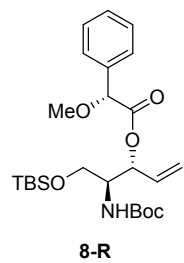
— 28.53

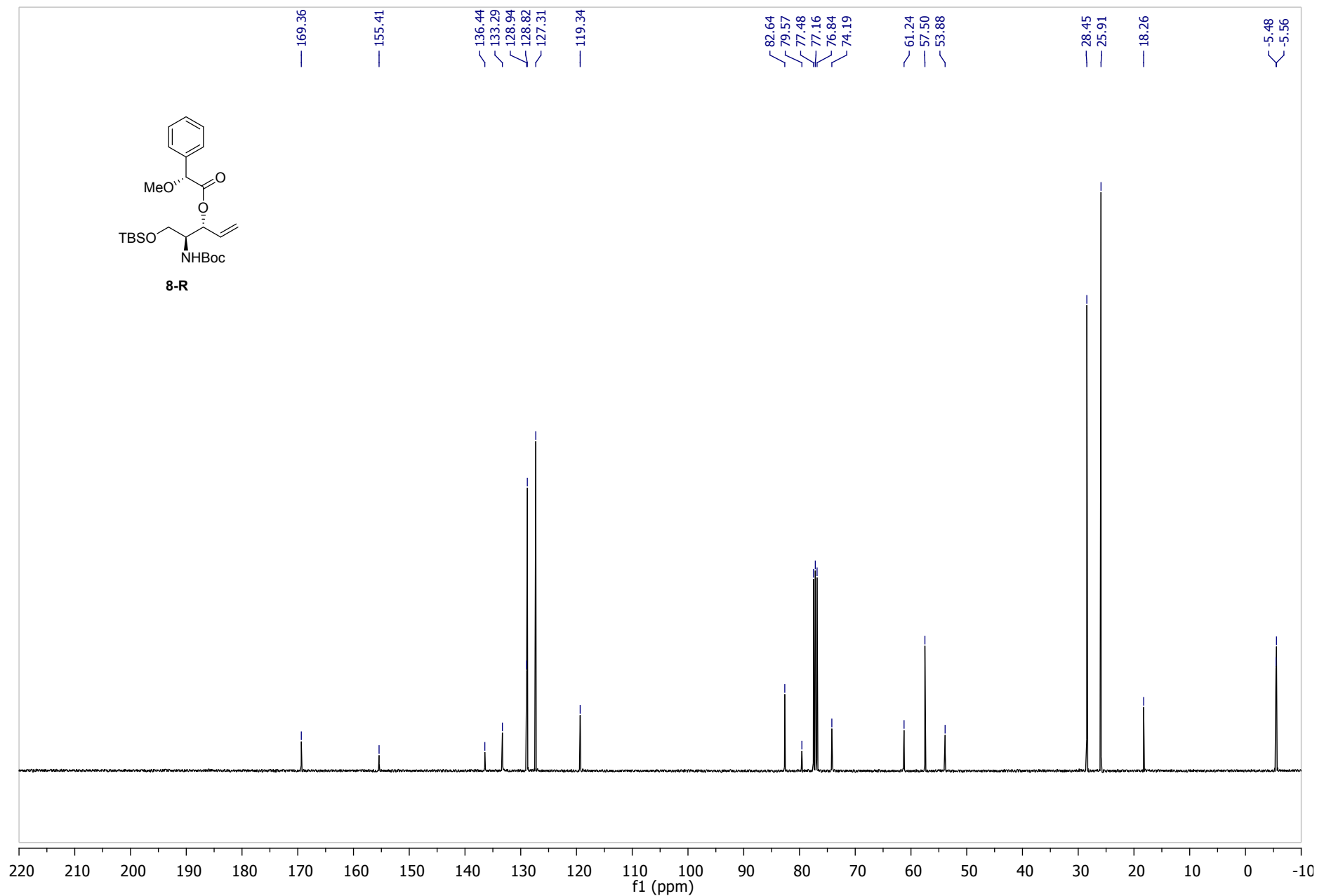
25.98

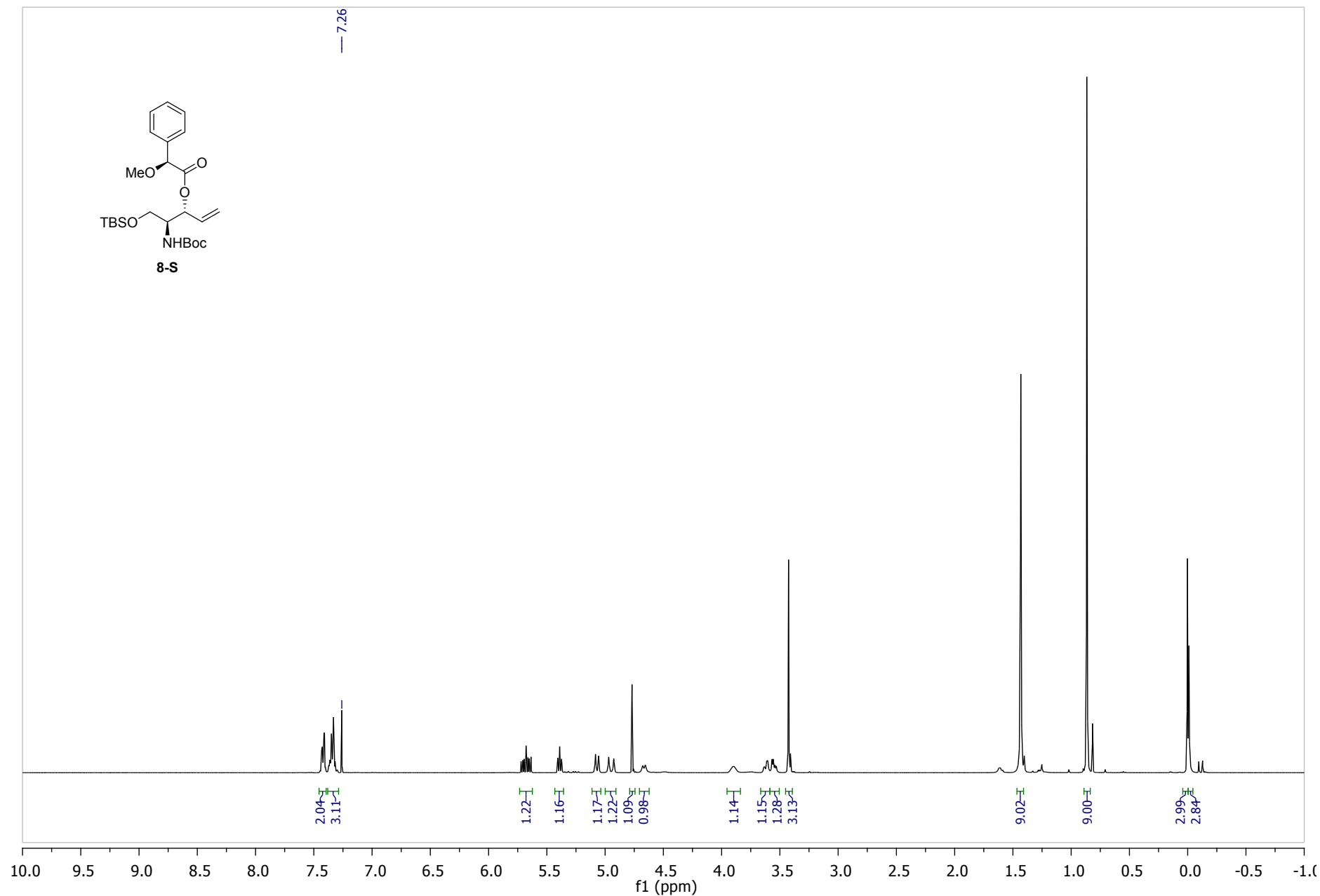
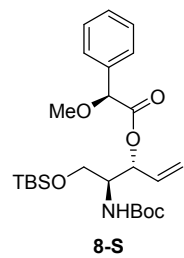
—18.30

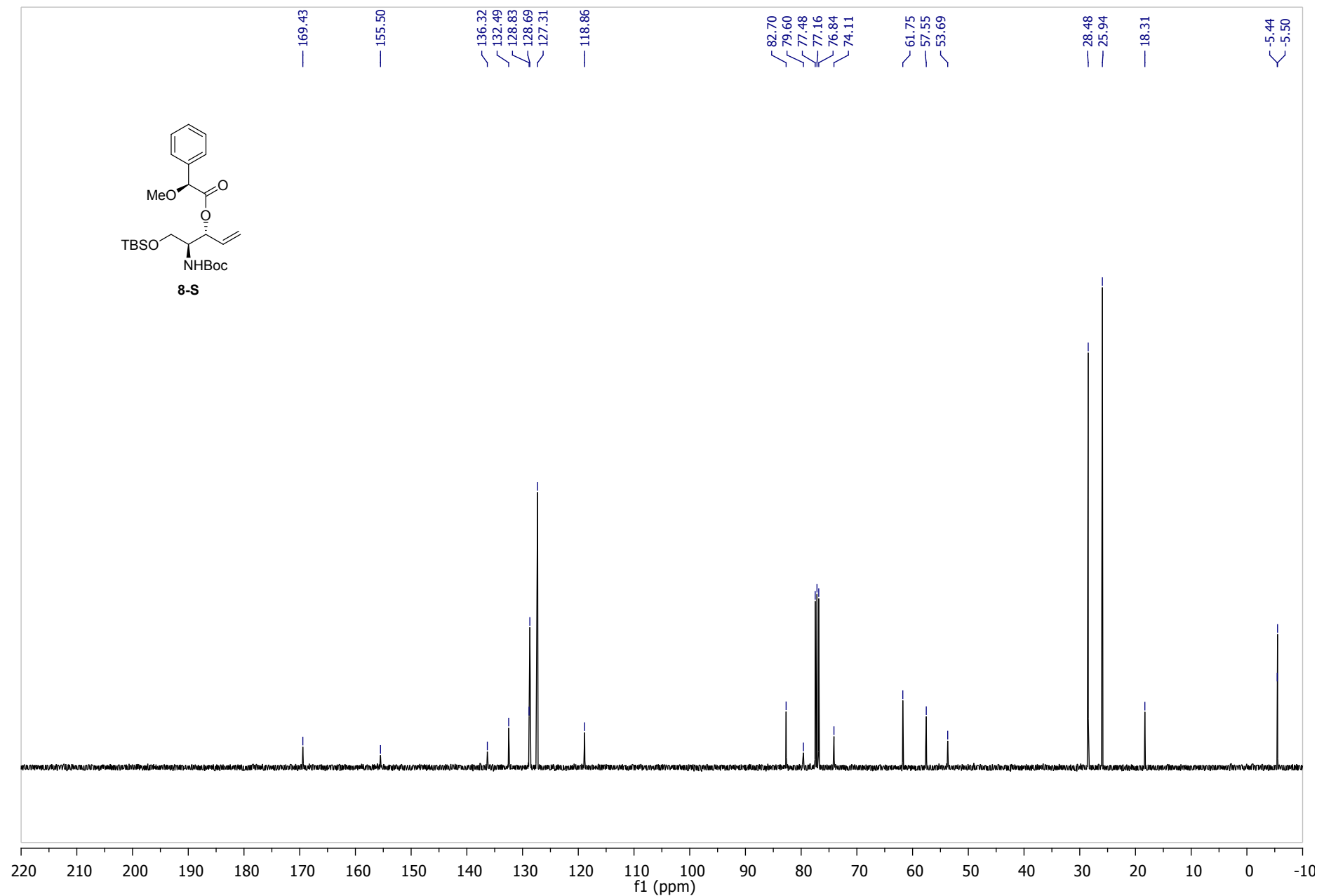
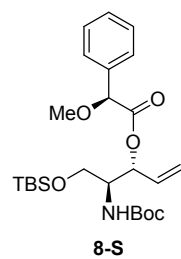
-5.45

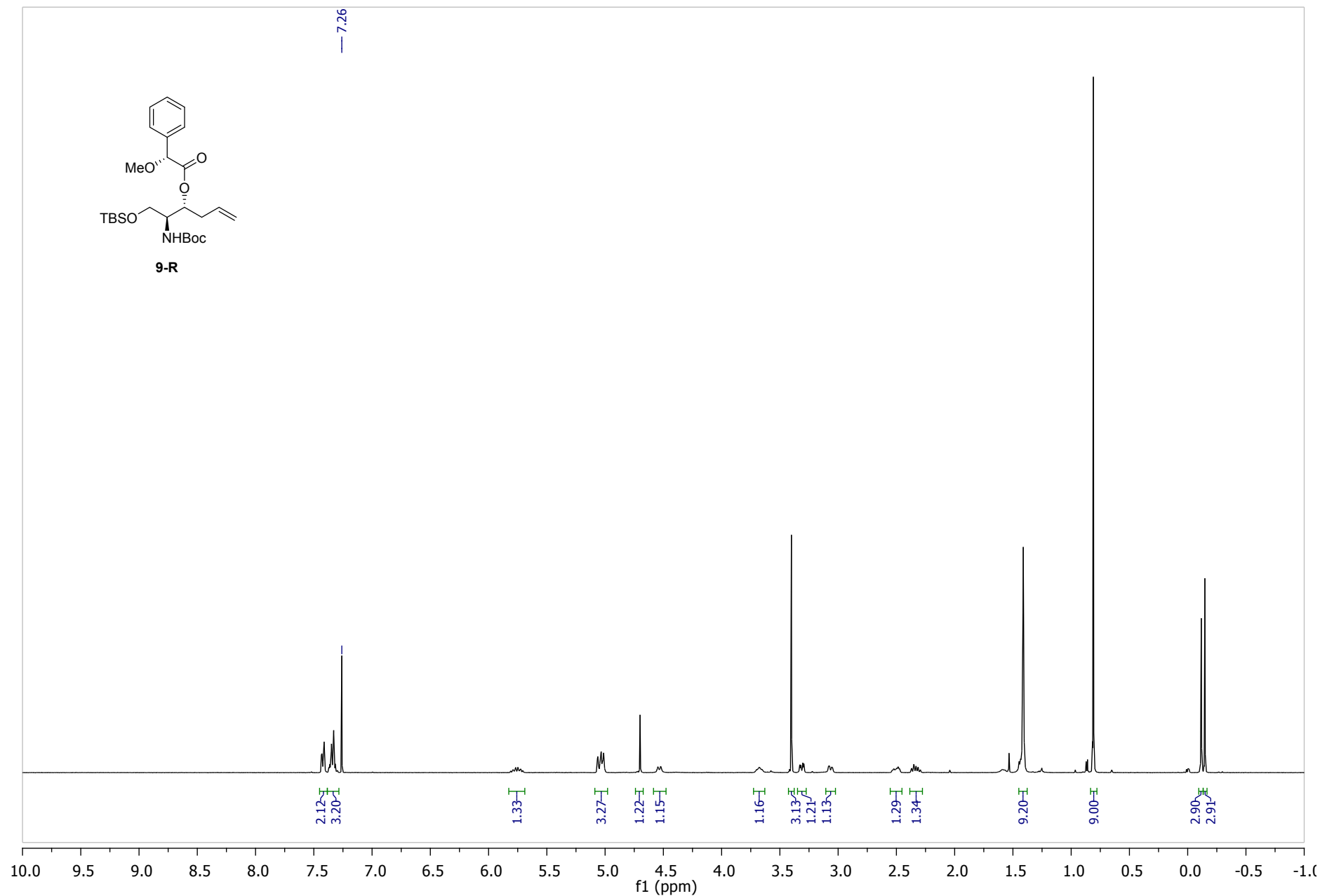
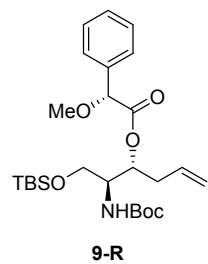
✓ -5.48

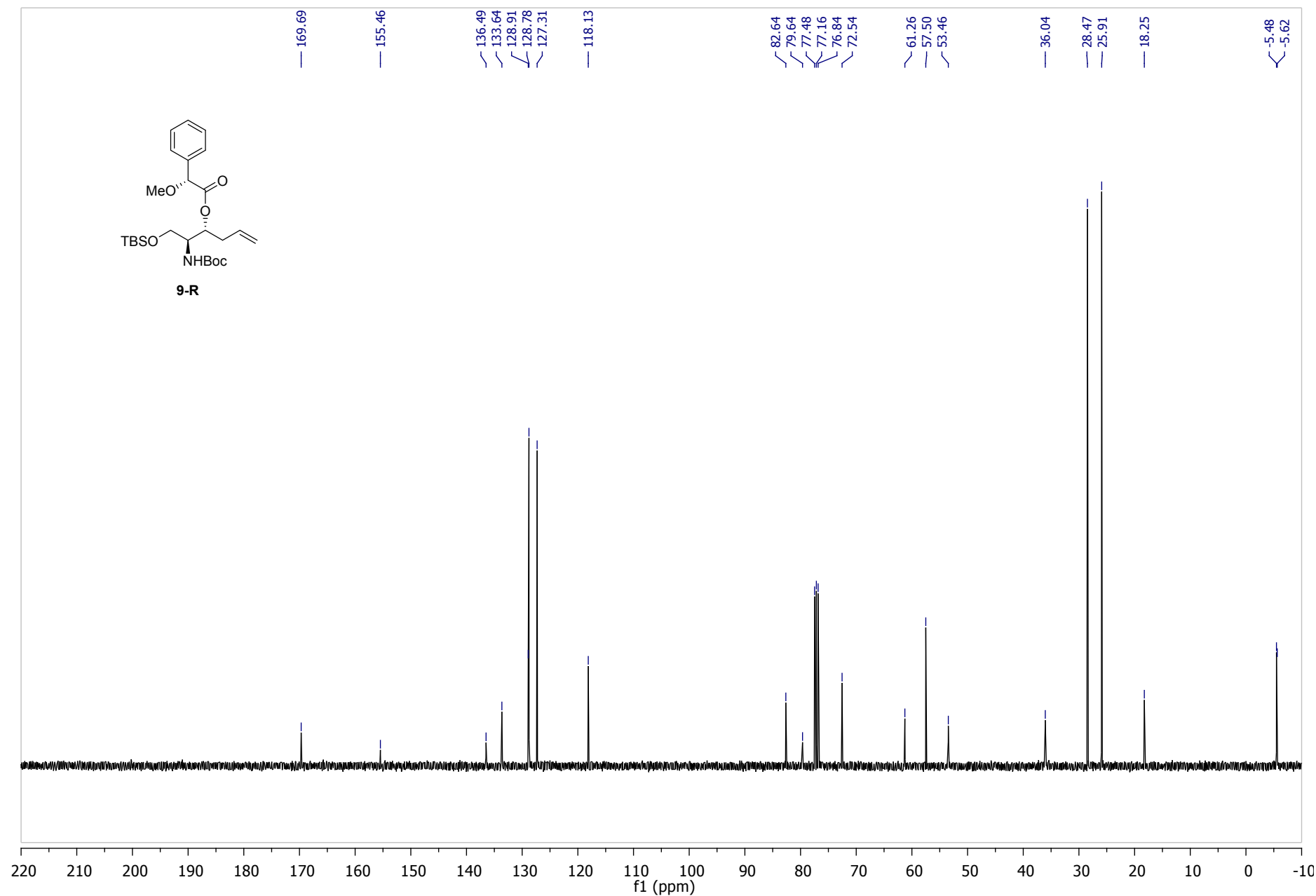
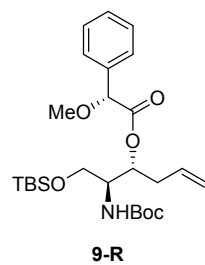


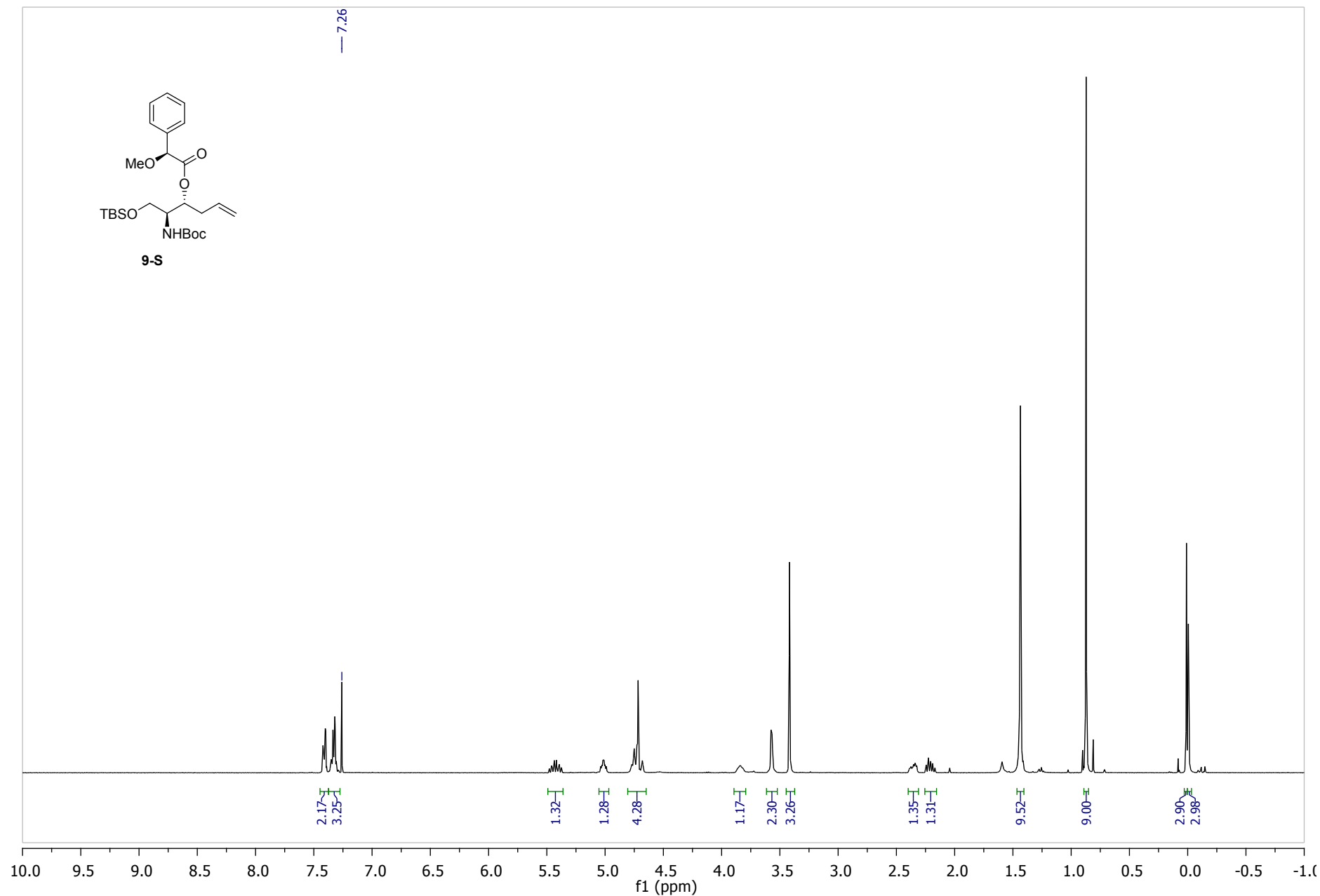
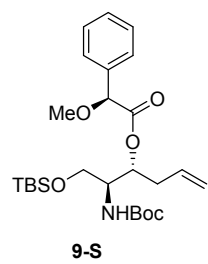


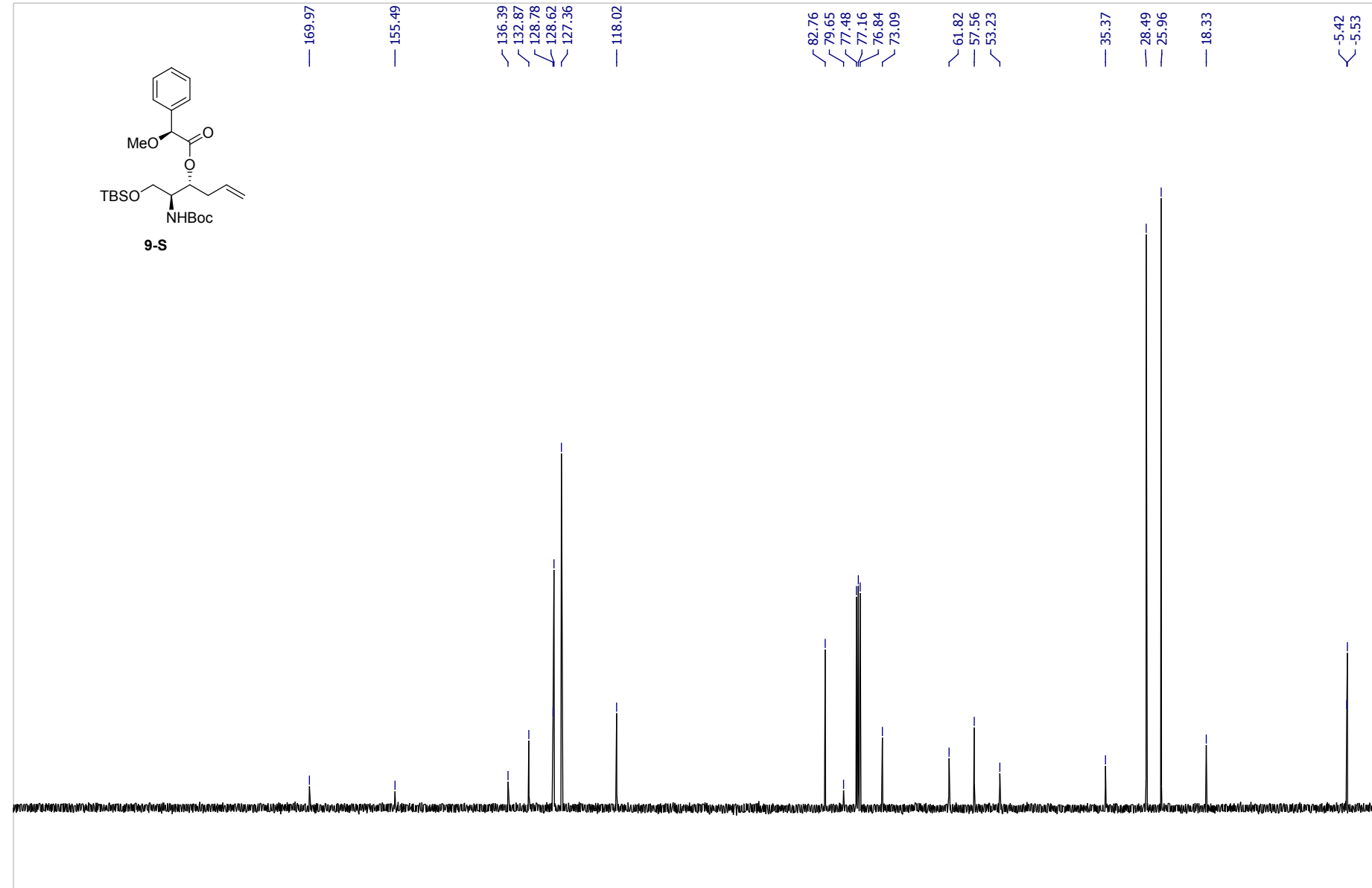
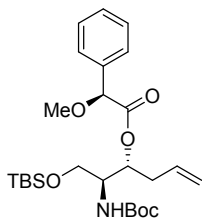


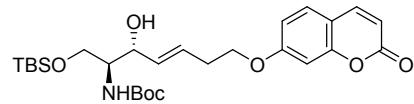




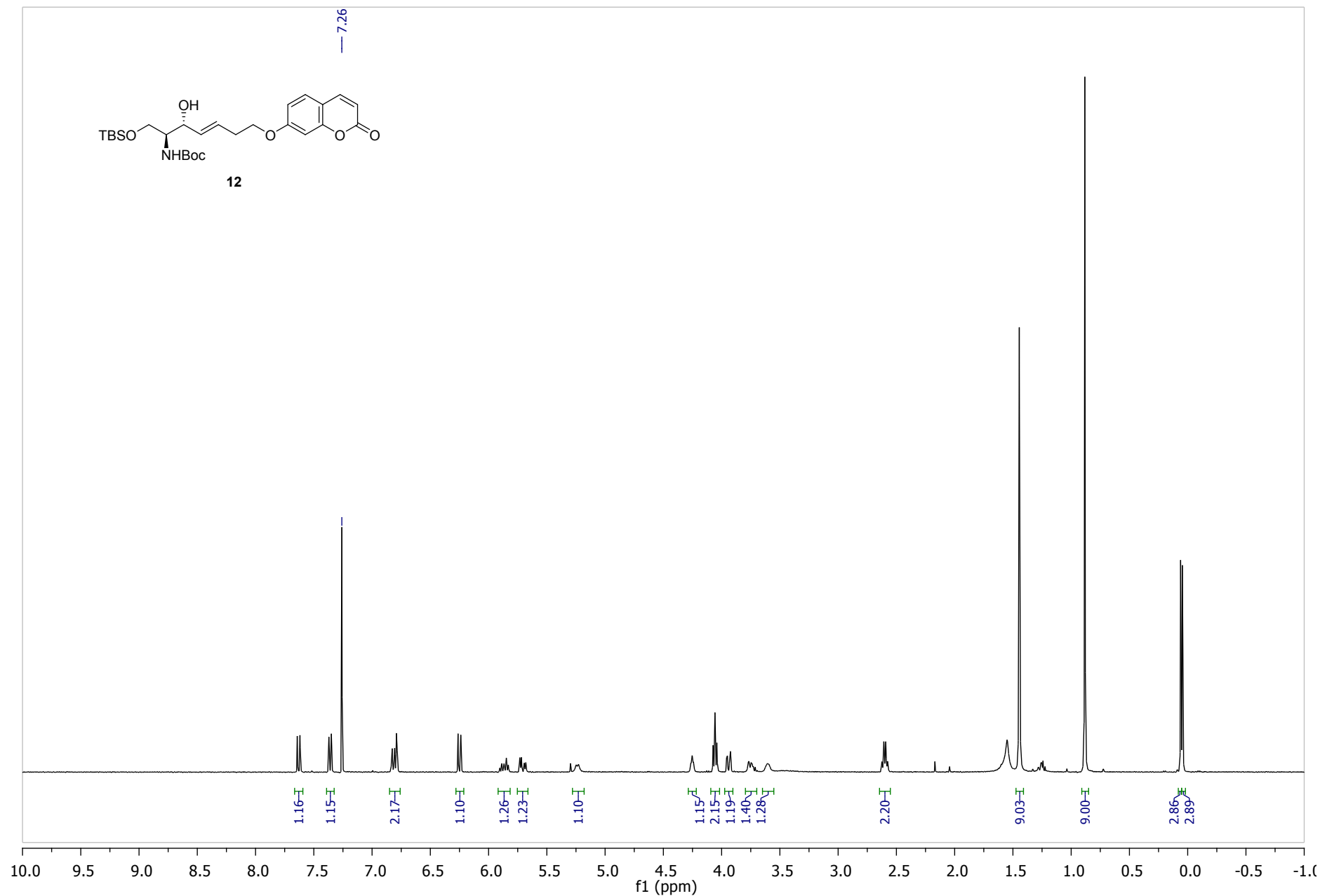


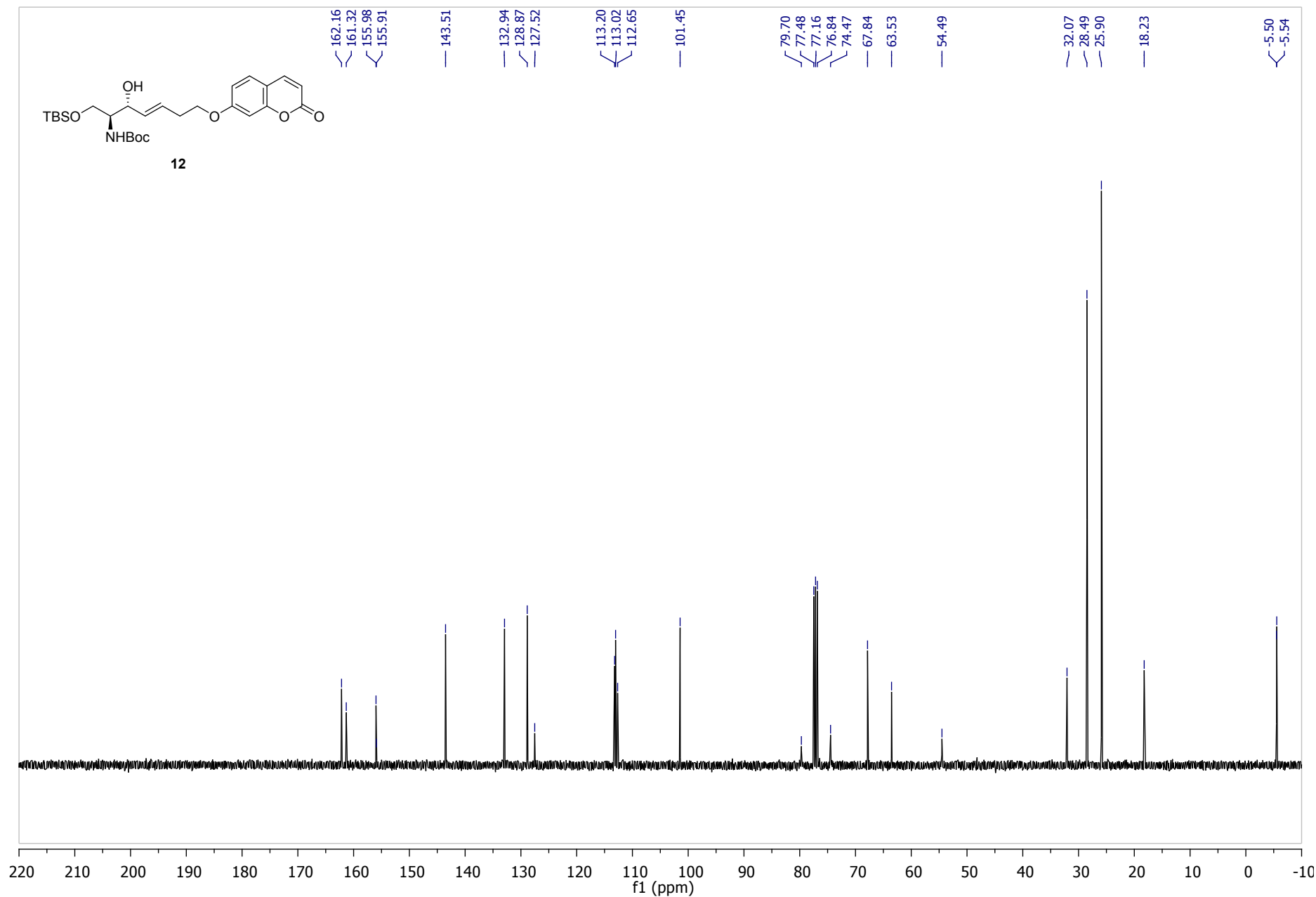


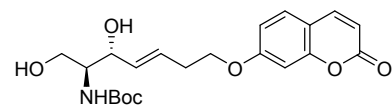




12

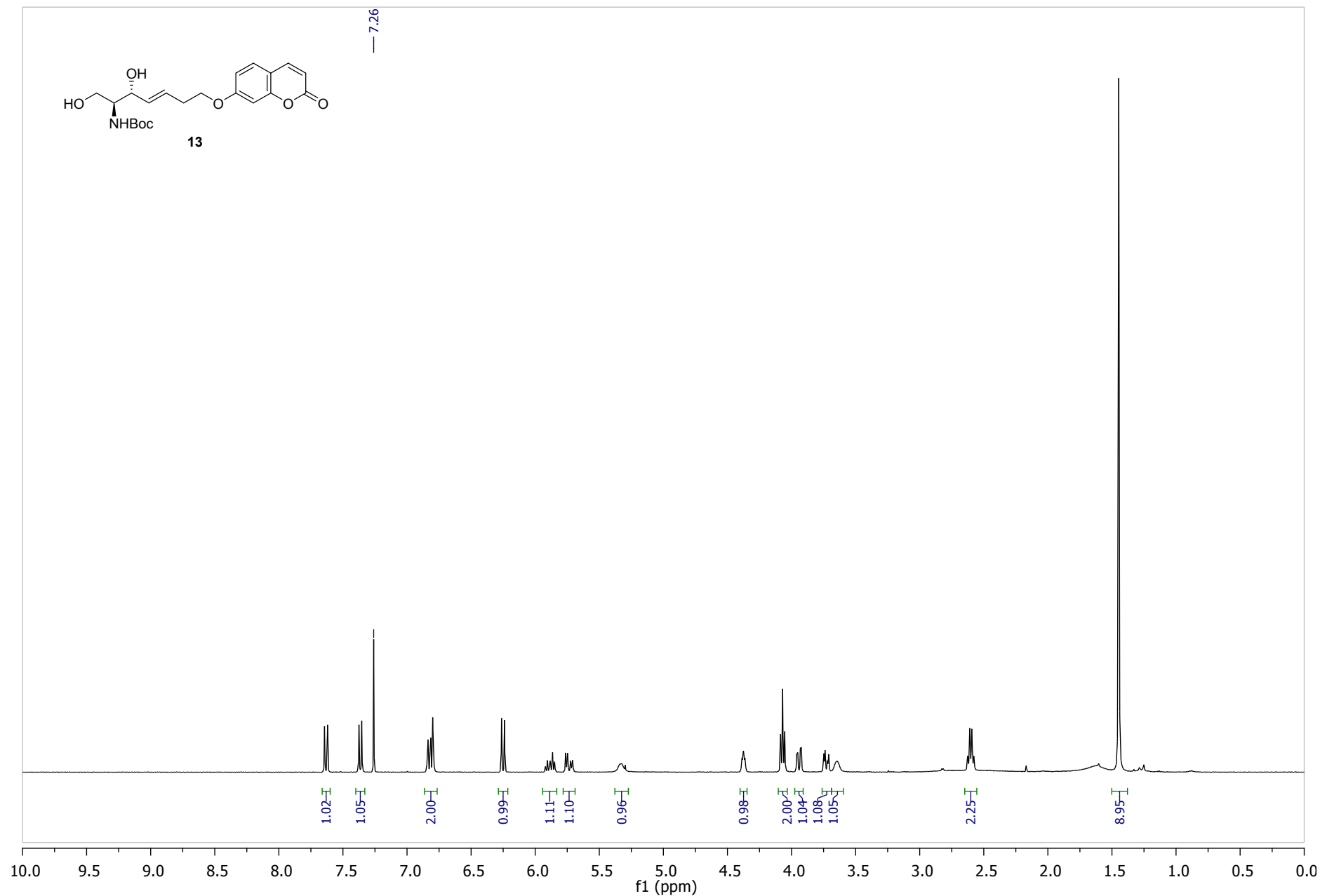


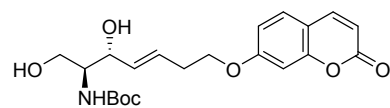




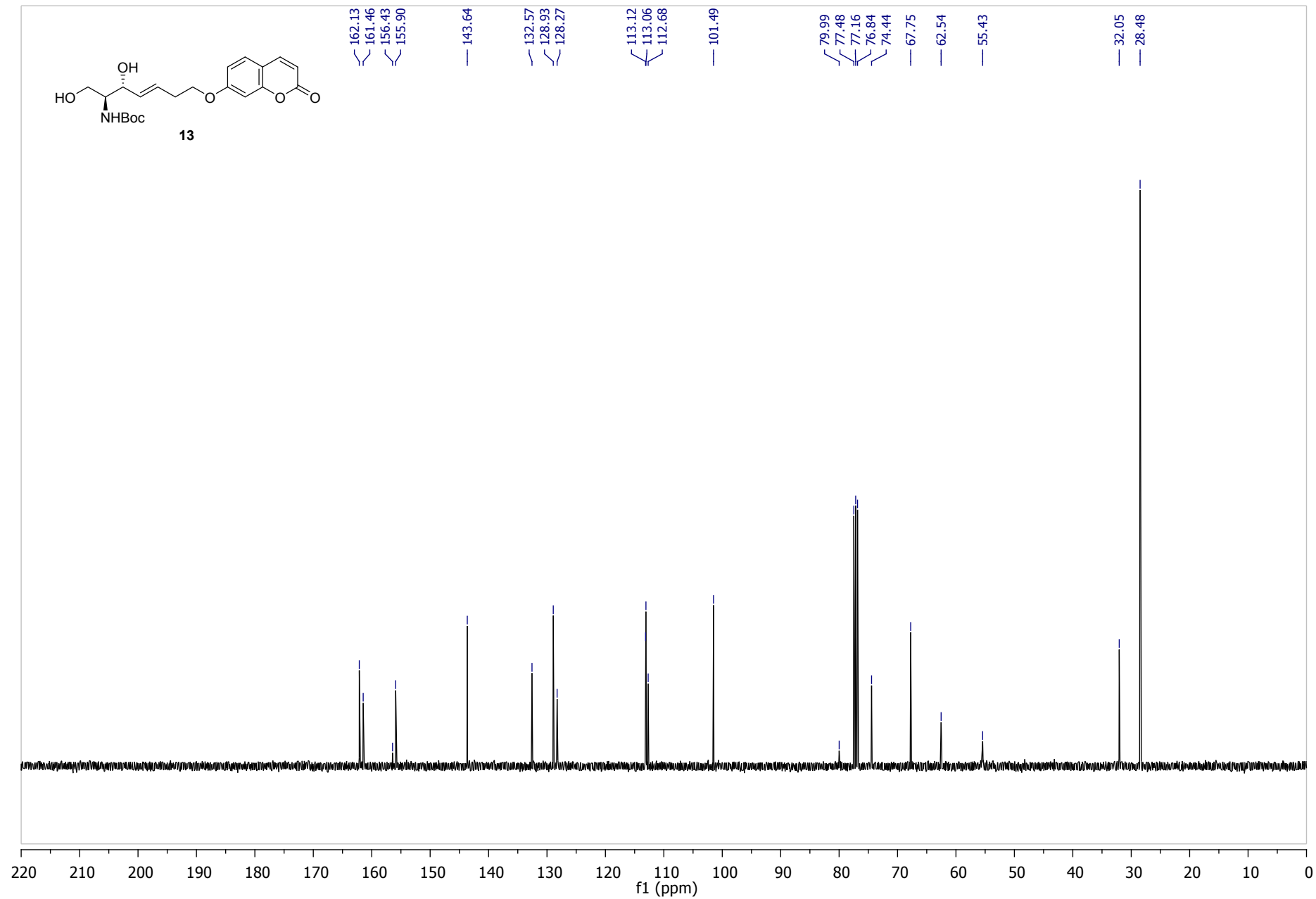
13

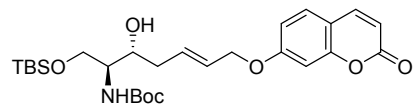
— 7.26





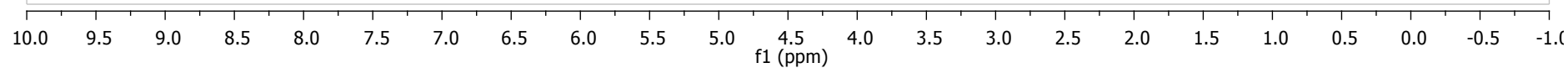
13

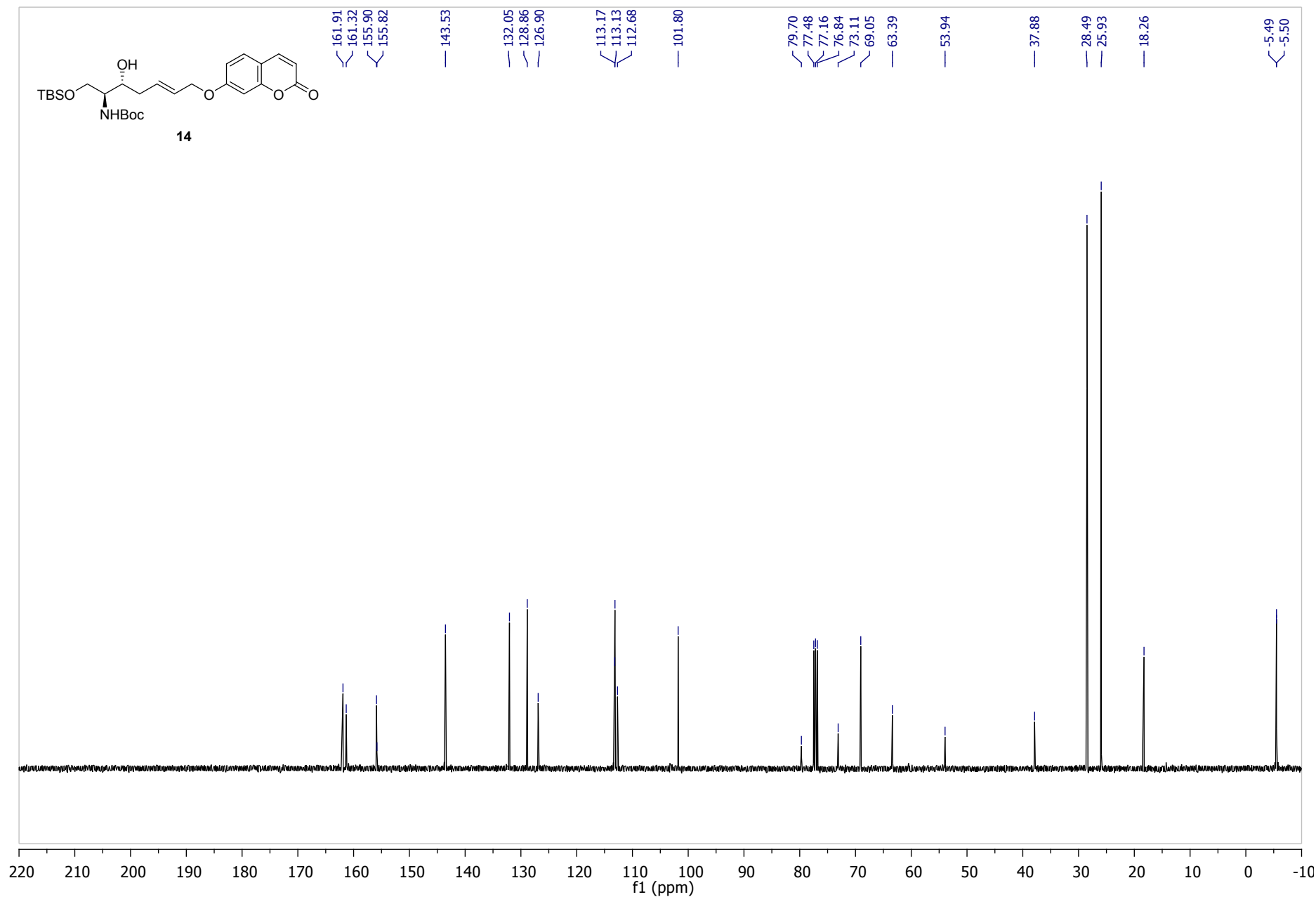


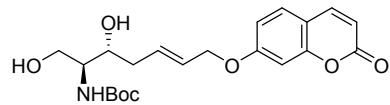


14

7.26

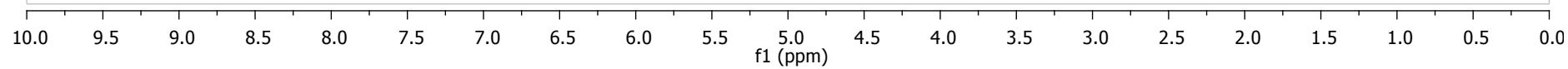


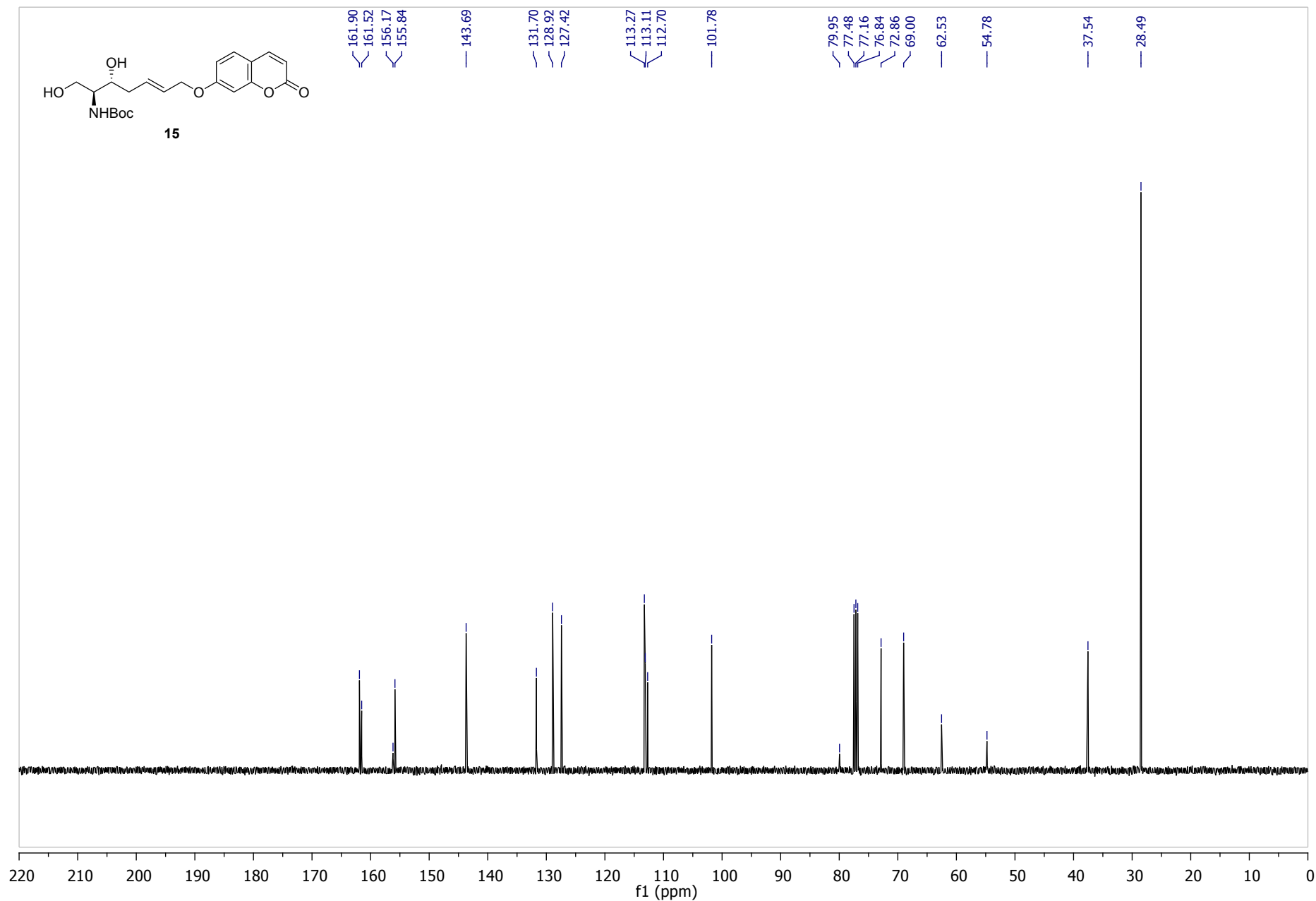


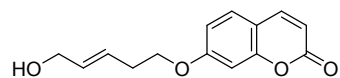


15

— 7.26

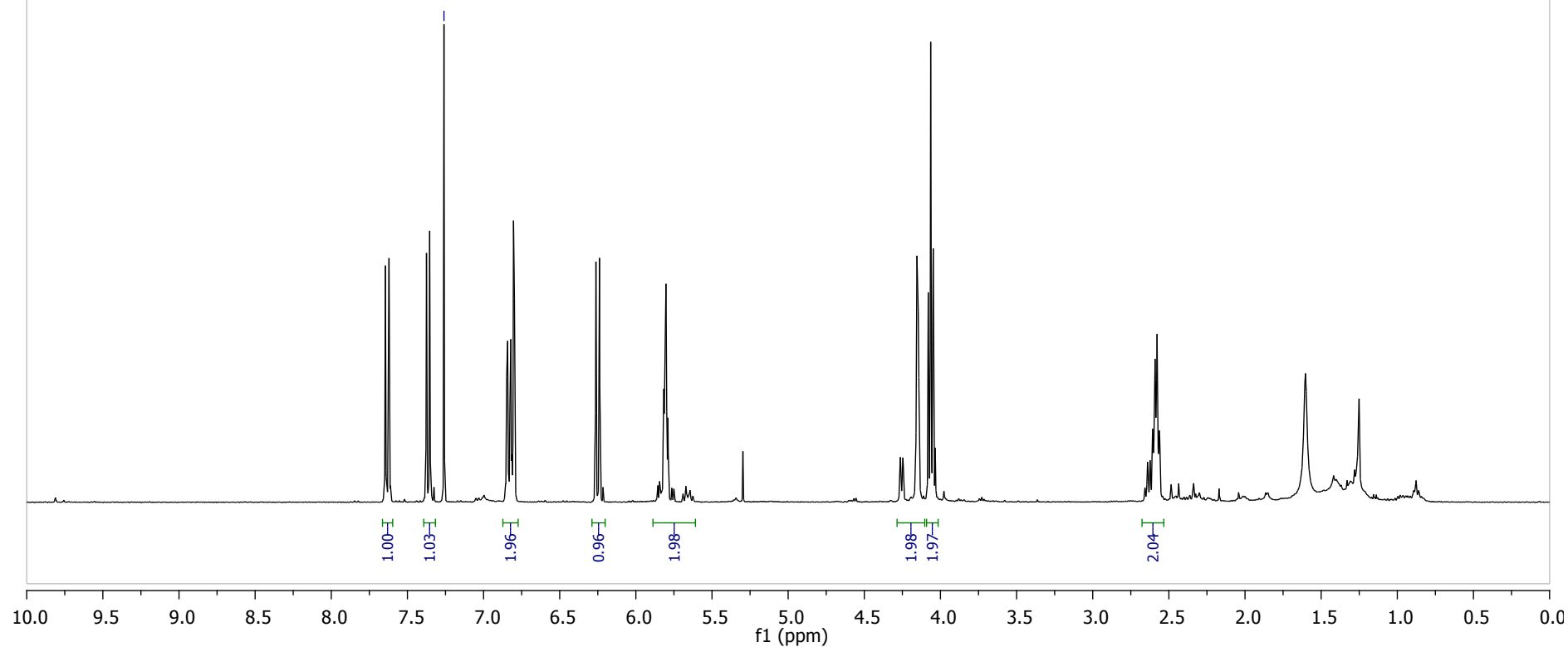


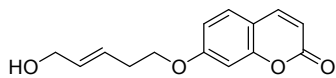




16

7.26





16

162.20

161.41

155.96

143.59

132.18

128.89

127.56

113.16

113.07

112.66

101.50

77.48

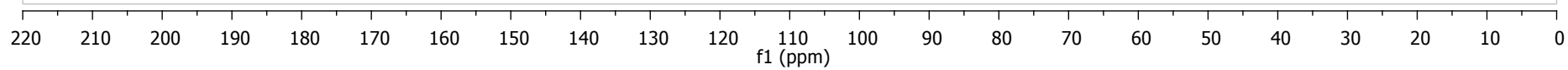
77.16

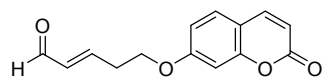
76.84

67.95

63.49

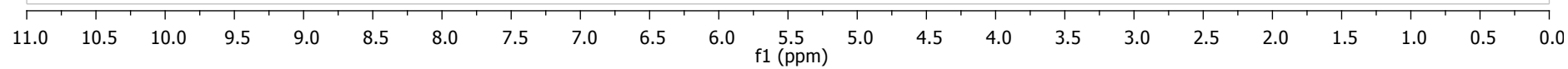
31.96

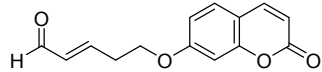




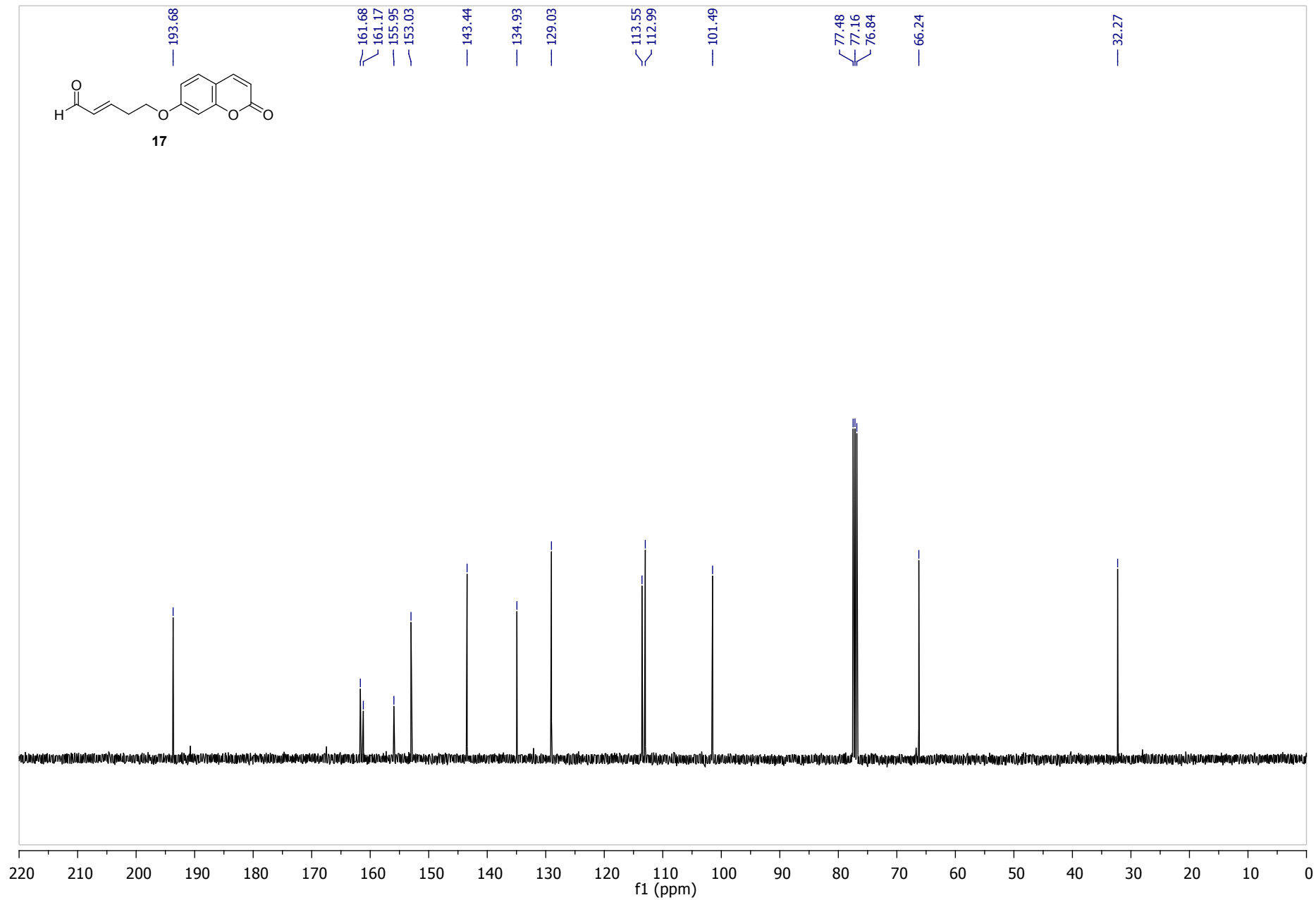
17

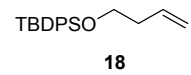
7.26



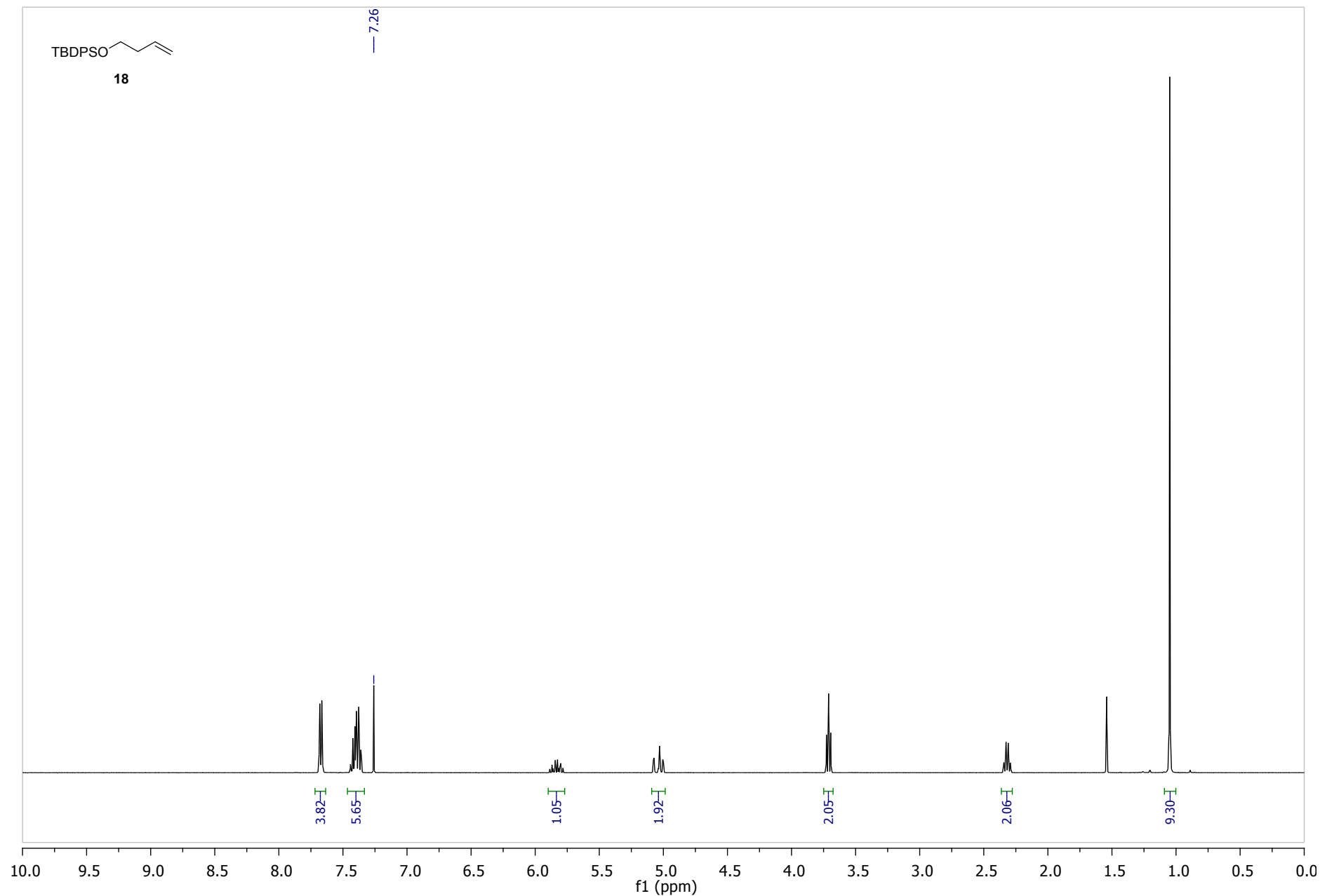


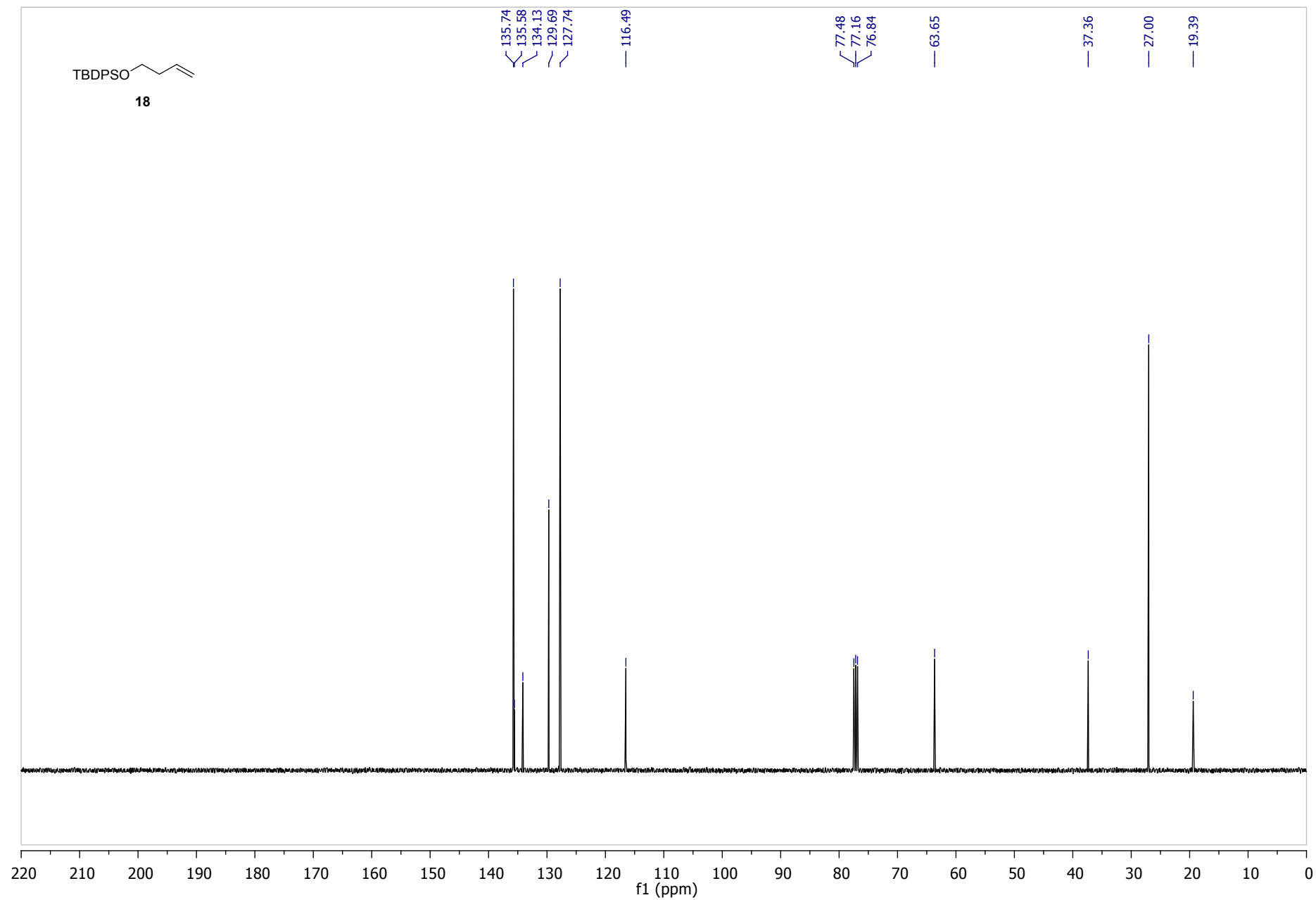
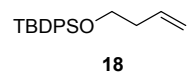
17

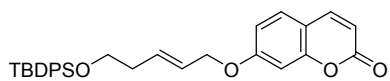




— 7.26

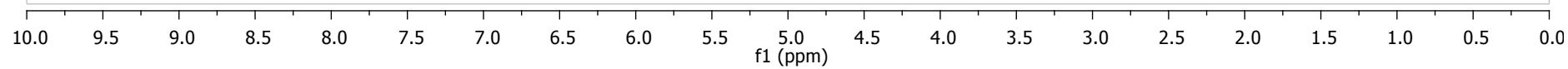


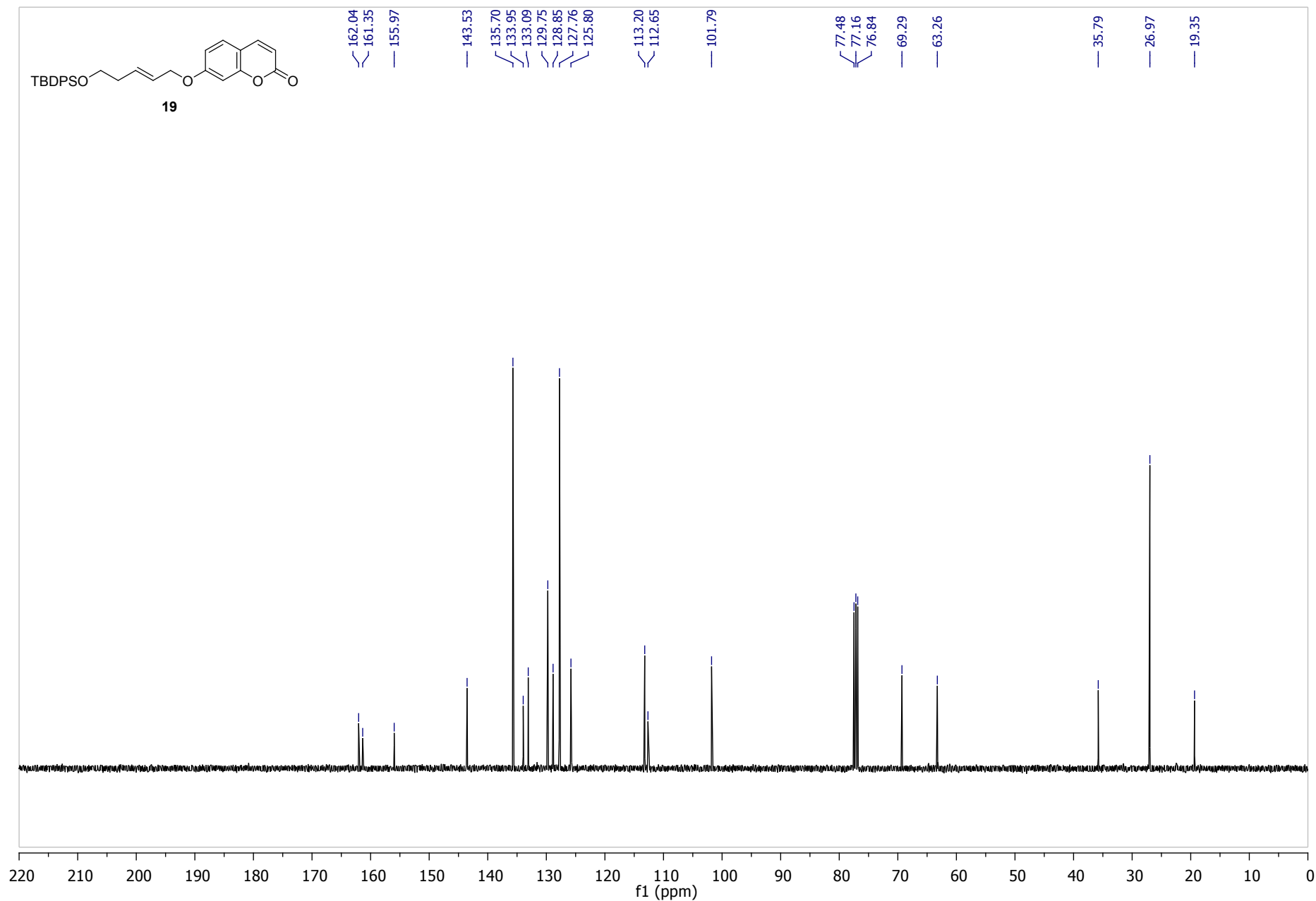


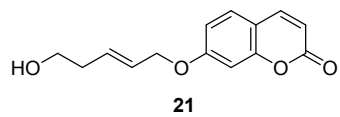


19

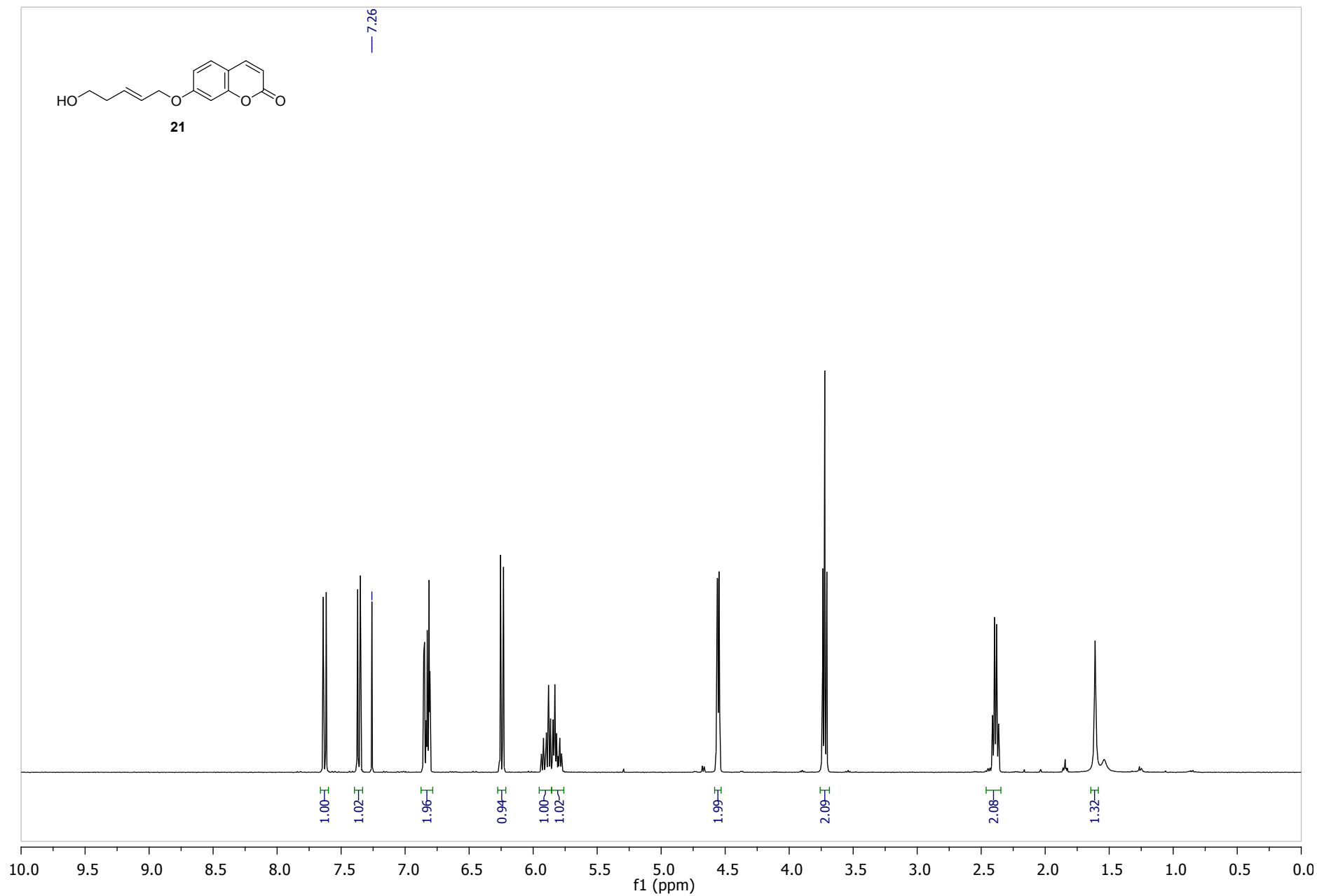
— 7.26

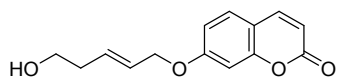




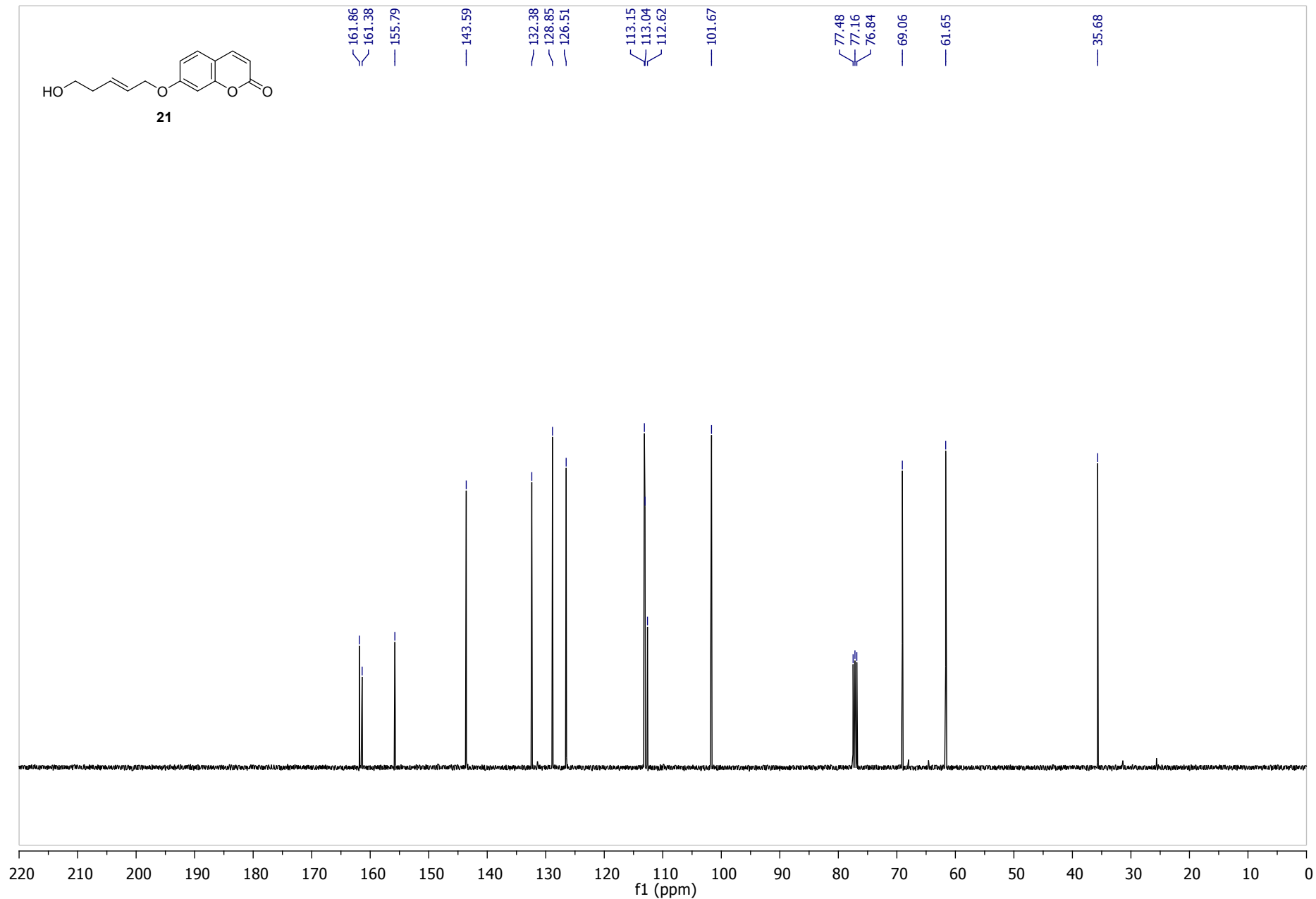


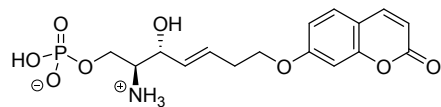
— 7.26



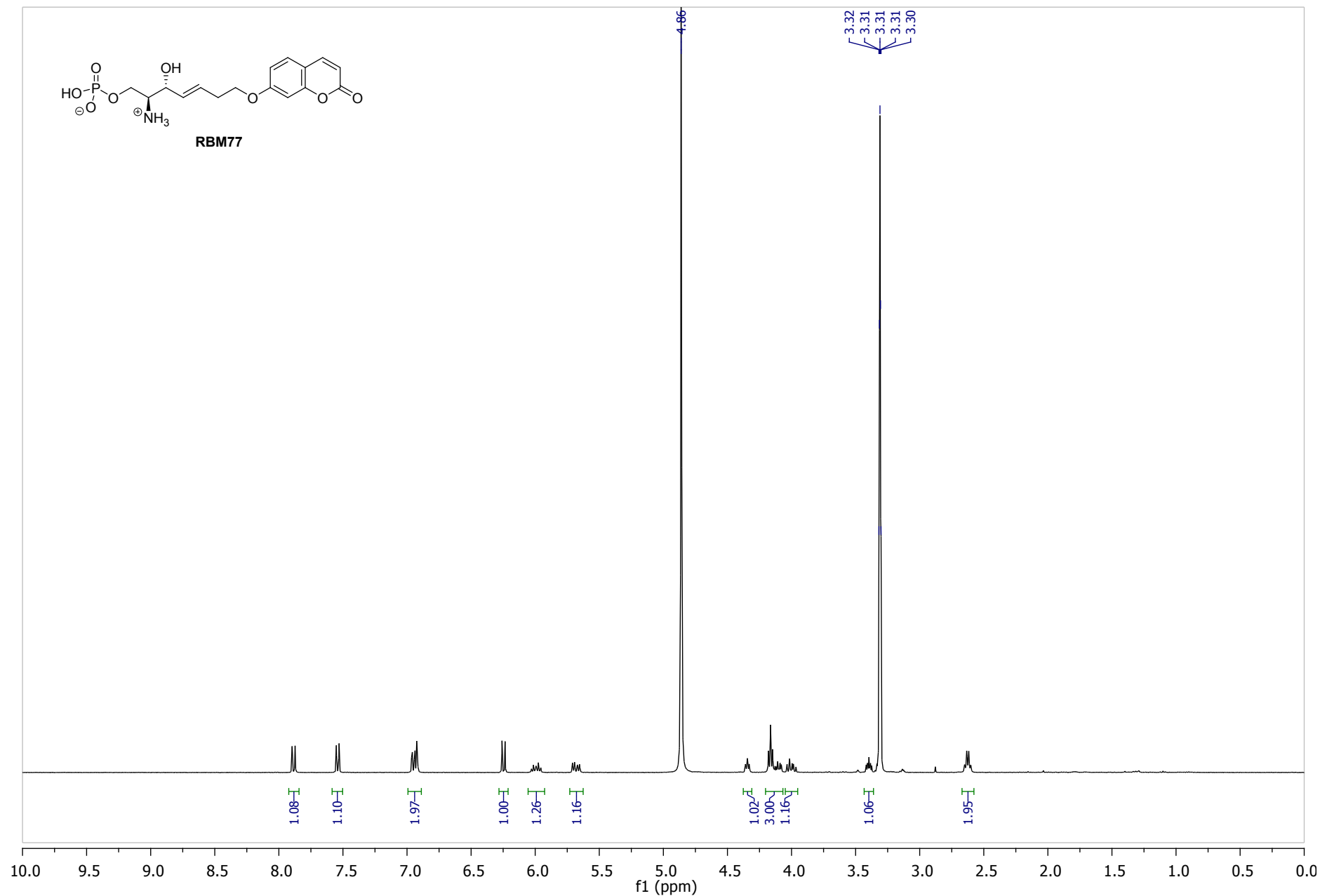


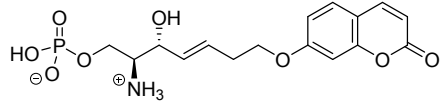
21



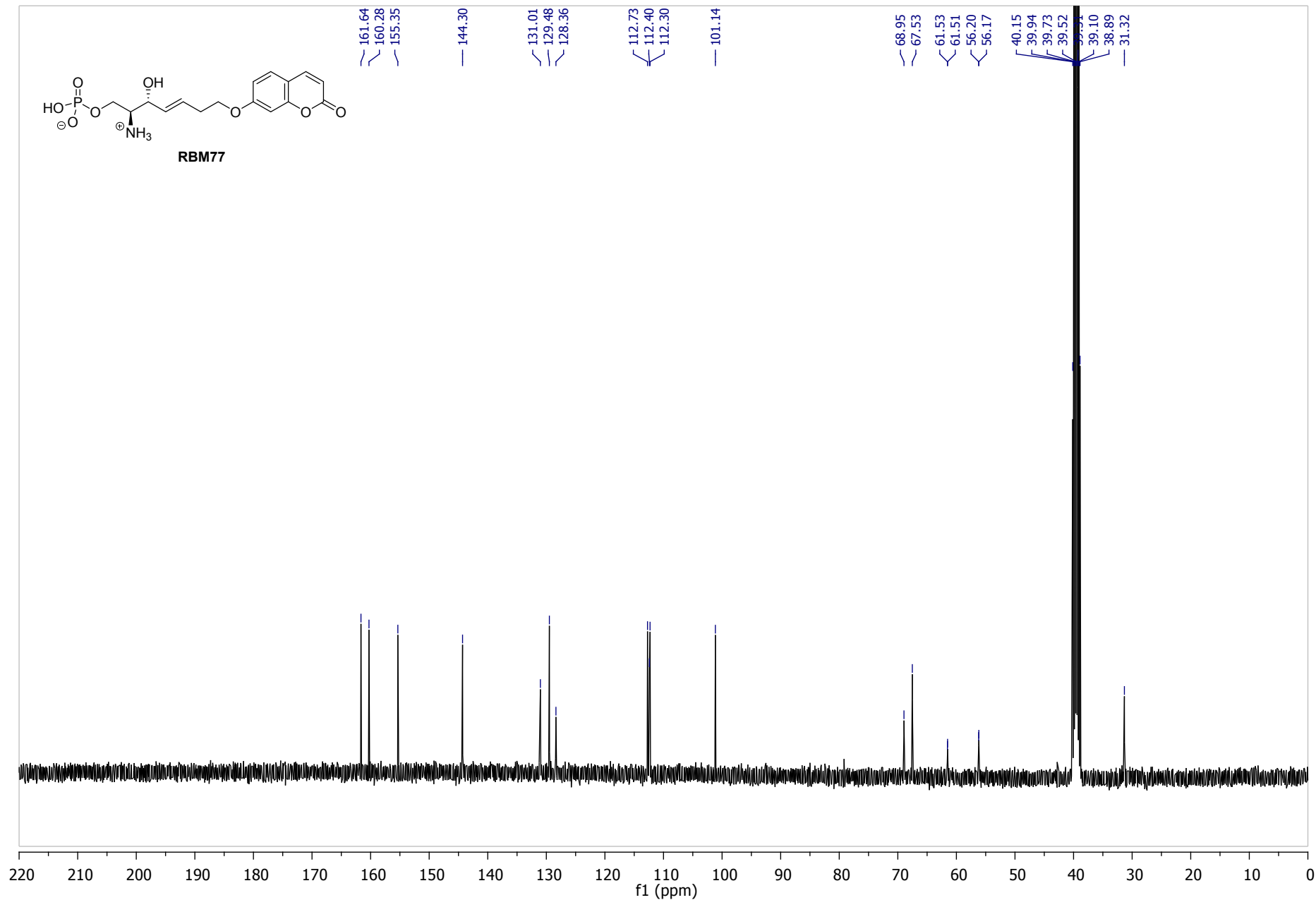


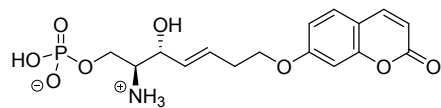
RBM77





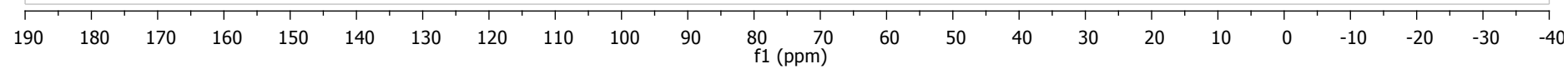
RBM77

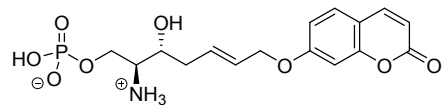




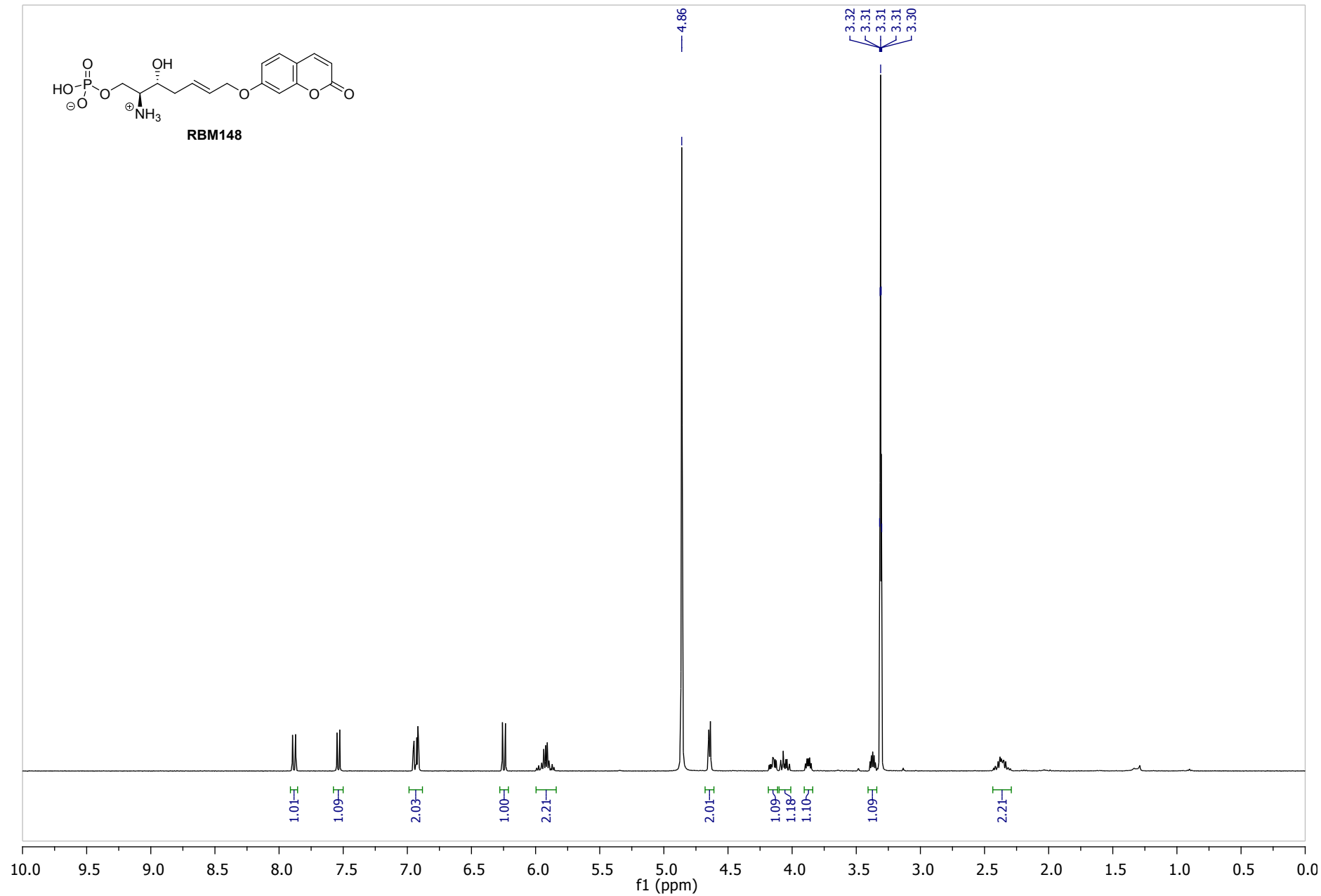
RBM77

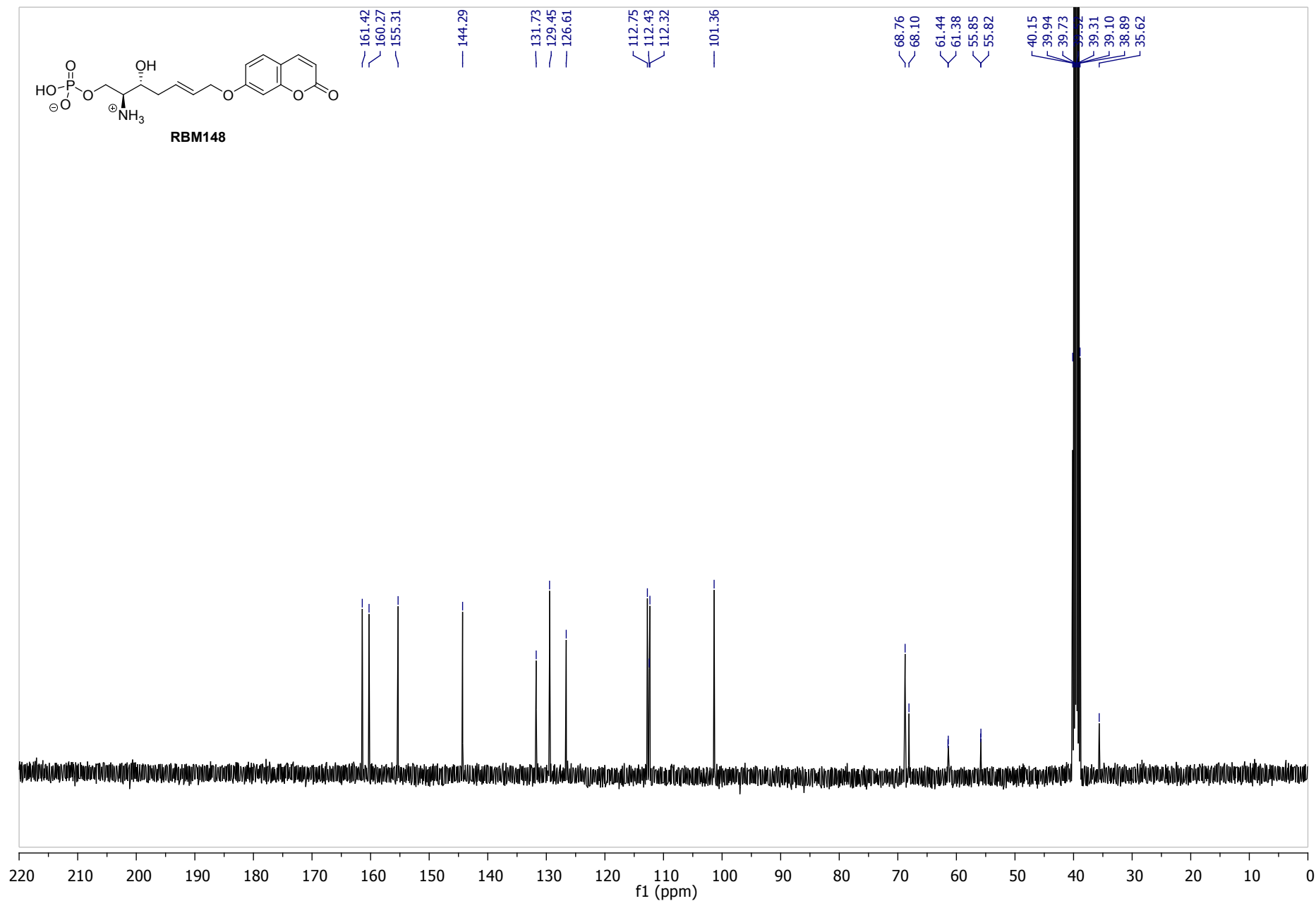
— 1.84

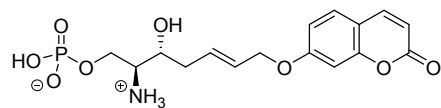




RBM148

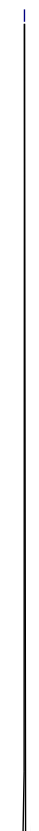






RBM148

— 2.08



190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40
f1 (ppm)