Electronic Supplementary Information for: <u>Capping the calix: solvation and recognition of</u> <u>Cs(I) through cation-pi interactions</u>

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ESI1: EXAFS Collection and Analysis of Cs-C4P-Br Solution in Toluene

n-Hexyl substituted C4P was prepared as previously reported in the literature.¹ A solution of CsBr in methanol (Sigma-Aldrich) was prepared by dropping a saturated aqueous solution of CsBr into methanol, and mixed with a solution of hexyl-substituted C4P in toluene (Sigma-Aldrich) (mole ratio CsBr:C4P was 1:1.1). The mixture was left over night and then evaporated on a rotary evaporator. The residue was dissolved in toluene to 25 mM CsBr. Collection and analysis of EXAFS data was conducted using the procedure reported previously in our publications.² Briefly, the Cs L3-edge and Br K-Edge spectra were collected in fluorescence mode at beamline 12-BM-B at the Advanced Photon Source (APS) at Argonne National Laboratory. A 13-element Ge detector (Canberra) was used to detect the fluorescence signal at a sample-to-detector distance of 1 cm (for Cs) and 40 cm (for Br). The solution was injected into 1 mm Kapton capillaries. Six 1 hr scans were averaged for each solution. Analyses of the k3 χ (k) EXAFS was performed using EXAFSPAK. Curve-fitting used theoretical phase and amplitude functions that were calculated with FEFF8.01.



Figure S1. Br K-edge EXAFS and FT-EXAFS for Cs-C4P-Br toluene solution.

ESI1: References

N. J. Williams, V. S. Bryanstev, R. Custelcean, C.A. Seipp and B. A. Moyer, *Supramol. Chem.* 2016, 28, 176-187.
R. J. Ellis, Y, Meridiano, R. Chiarizia, L. Berthon, J. Muller, L. Couston and M. R. Antonio, *Chem.-Eur. J.* 2013, 19, 2663-2675.

ESI2: Computational Details

Electronic structure calculations were carried out using the Gaussian 09¹ Revision D.01 software packages using the M06-2X² flavor of density functional. The 6-311+G(d,p) basis set was employed for hydrogen and main group elements. Cs was modeled using the 46MWB effective core potential (ECP46MWB) and the associated basis set,³ while Br was modeled using ECP28MWB⁴ and the VTZ (14s10p2d1f)/[3s3p2d1f]) basis set.⁵ The M06-2X density functional was chosen because it provides a relatively accurate prediction of interaction energies in noncovalent complexes.^{2,6} Additional calculations were performed with the ω B97X-D function,⁷ which is also known to provide accurate binding energies in noncovalent complexes.⁶ Using the gas phase geometries, implicit solvent corrections for toluene as a solvent were obtained with the SMD model⁸ in Gaussian 09.¹

All-atom molecular dynamics simulations were performed with the MedeA platform⁹ using the MedeA LAMMPS¹⁰ interface module. We employed the non-polarizable PCFF+ forcefield, which is based on the PCFF forcefield¹¹ with extension to additional molecular classes and refinements, in particular for non-bonded parameters, to reproduce densities and cohesive energies of molecular liquids in a fashion similar to that used in the development of the COMPASS forcefield.¹² The Waldman and Hagler 6th order combination rules were used for the off-diagonal L-J-9-6 parameters. In order to substantially improve the agreement between the computed PCFF+ and experimental crystal structure containing Cs-C4P-Br (LAMVAX ref code), as well as the PCFF+ and M06-2X structures of the Cs-C4P-Br complex in the gas phase, the non-bonded L-J-9-6 ε parameters for Cs and Br were changed from their default values of 0.1475 and 0.07993 kcal/mol to the values of 0.850 and 0.150 kcall/mol, respectively. A full list of forcefield parameters in the LAMMPS format is given at the end of the ESI.

The initial configuration for MD simulations was generated using the MedeA Amorphous Cell Builder⁹ by randomly distributing one Cs-C4P^L-Br complex and 353 toluene solvent molecules in a periodic cubic box of 40 Å length, which closely corresponds to the experimental density of toluene, $\rho = 0.867$ g/cm³, at T = 298 K. The dispersion interactions were evaluated using a nonbonded cutoff of 9.5-11 Å with an added long-range Van der Waals tail correction. The long-range electrostatic interaction was calculated by means of the Particle-Particle-Particle-Mesh (P3M) Ewald summation method¹³ with long-range precision smaller than 10⁻⁵ for the electrostatic energy. The initial structure of was first minimized with 1000 steps using the steepest descent method, followed by 2 ns of equilibration at 298 K in a canonical NVT ensemble and another 2 ns of equilibration at 298 K in an NPT ensemble. The average equilibrium volume from the NPT simulations was taken for the subsequent NVT simulations that were performed for 6 ns. The data for the last 5 ns were used for the analyses. The equations of motion were integrated with a time step of 1 fs. The temperature was controlled by the Nosé-Hoover thermostat and barostat with a temperature (Tdamp) and pressure damping (Pdamp) of 100 fs and a friction coefficient (drag) of 1.0 (NVT) and 0 (NPT). Graphical visualization of the complexes and the analyses of the trajectory were performed with VMD.14



Figure S2. Potential energy surfaces for displacing the Cs⁺ cation along the C₆ symmetry axis of benzene. The 6-311+G(d,p)/ECP46MWB(7s6p)/[5s4p](Cs) basis set was employed with M06-2X and ω B97X-D. The aug-cc-pVDZ/ECP46MDF(12s11p5d3f)/[8s8p5d3f](Cs)¹⁵ basis set was employed with MP2 and CCSD(T).



Figure S3. Snapshot of the Cs-C4P^L-Br complex in toluene from classical molecular dynamics simulations. Color scheme: Br, green; Cs, magenta; N, blue; H, white; C grey. Solvent molecules with the Cs–C distance in the range of 4-7 Å are shown with thin grey bonds.



Figure S4. Cs–Br (left) and Cs–N (right) radial distribution functions for the Cs-C4P^L-Br complex in toluene obtained from classical molecular dynamics simulations.

	oralled with D11 and post flatified 1 oek wave function methods				
Distance (Å)	M06-2X	ωB97X-D	MP2	CCSD(T)	
C(toluene)	3.530	3.603	3.415		
C(toluene)	3.530	3.607	3.415		
C(toluene)	3.533	3.645	3.418		
C(toluene)	3.554	3.665	3.432		
CH ₃ (toluene)	4.449	4.431	4.349		
C(benzene)	3.557	3.665	3.440	3.489	

Table S1. Comparison of the optimized Cs–C distances in Cs⁺-toluene and Cs⁺-benzene systems obtained with DFT and post-Hartree-Fock wave function methods^a

The 6-311+G(d,p)/ECP46MWB(7s6p)/[5s4p](Cs) basis set was employed with M06-2X and ω B97X-D. The aug-cc-pVDZ/ECP46MDF(12s11p5d3f)/[8s8p5d3f](Cs) basis set was employed with MP2 and CCSD(T).

Table S2. DFT optimized Cs-ligand and Cs-solvent distances in Cs-C4P-Br

Distance			M06-2X		ωB97X-D
	Cs-C4P ^L -Br	Cs-C4P-Br	Cs-C4P-Br	Cs-C4P-Br	Cs-C4P-B ⁻
(A)*		(toluene)	(toluene) ₂ -B	(toluene) ₂ -A	(toluene) ₂ -A
$C_{\alpha}(pyrrole)$	3.678	3.664	3.664	3.551 3.583 3.829 4.250	3.645 3.668 3.902
$C_{\beta}(pyrrole)$	3.466	3.451	3.460	3.473 3.482 3.532 3.766	4.080
C _{meso}	4.122	4.099	4.108	4.131 4.139 4.278 4.287	3.535 3.547 3.612
Ν	3.304	3.300	3.312	3.321 3.394 3.399 3.422	3.700
Br	4.724	4.597	4.615	4.825	4.210 4.223 4.271
C(toluene)		3.519	3.478 6.296	3.559 3.775	4.272
C(toluene)		3.520	3.500 6.404	3.590 4.061	3.404 3.430 3.444
C(toluene)		3.533	3.541 6.608	3.641 4.781	3.449
C(toluene)		3.543	3.576 6.807	3.678 5.246	4.850
C(toluene)		3.561	3.602 6.807	3.733 5.818	3.600 3.782
C(toluene)		3.565	3.624 7.008	3.747 6.014	3.626 4.086
CH ₃ (toluene		4.297	4.354 6.578	4.409 3.665	3.701 4.741
)					3.743 5.242
					3.808 5.764
					3.826 5.975
					4.296 3.752

^aDistances between Cs(I) and ligand atoms are averaged (i) over all pyrrole rings in Cs-C4P-Br(toluene) and Cs-C4P-Br(toluene)₂-B and (ii) for each pyrrole ring in Cs-C4P-Br(toluene)₂-A. The lipophilic *n*-hexyl substituted C4P is denoted as C4P^L.

Table S3. Comparison of the simulated (DFT) Cs-ligand distances in the Cs-C4P-Br complex in gas phase, solid crystalline (both simulations and experimental¹⁶) and solution (toluene) phases.

	Cs-C4P-Br, gas phase		Cs-C4P-Br, solid state			Cs-C4P ^L -Br	
Distance						solution	
(Å) ^a	M06-	default	modified	default	modified	Experimental	modified
	2X	pcff+	pcff+	pcff+	pcff+	X-ray ¹⁵	pcff+
$C_{\alpha}(\text{pyrrole})$	3.654	3.650	3.733	3.813	3.792	3.756	3.789±0.270
$C_{\beta}(pyrrole)$	3.452	3.211	3.414	3.496	3.597	3.595	3.480±0.154
C _{meso}	4.102	3.816	4.045	4.139	4.248	4.245	4.121±0.150
N	3.304	2.930	3.208	3.272	3.466	3.460	3.266±0.155
Br	4.600	3.564	4.424	4.694	5.063	4.946	4.654±0.225
Br (ion pair)				3.265	3.638	3.588	

^aDistances between Cs(I) and ligand atoms are averaged over all pyrrole rings. The lipophilic *n*-hexyl substituted C4P is denoted as C4P^L.

ESI2: References

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Cartesian coordinates and electronic energies of the studied complexes obtained at the M06-2X/6-311+G(d,p)/ ECP46MWB(Cs)/ECP28MWB VTZ(Br) level.

Cs-C4P-Br; E(RM062X) = -1341.08388679 A.U.

19.8941 17.2532 22.7657 Br Ν 21.5782 19.6868 21.0133 21.2004 18.8252 21.4112 Η С 21.0503 20.3486 19.9276 21.6107 19.6777 24.4992 Ν 21.2269 18.8168 24.1060 Η С 21.8079 21.4834 19.7377 Η 21.6714 22.2100 18.9526 С 22.8241 21.4974 20.7353 Η 23.6010 22.2370 20.8473 С 22.6643 20.3706 21.5116 С 23.4310 19.9159 22.7400 С 22.6869 20.3648 23.9839 С 22.8605 21.4878 24.7625 23.6341 22.2288 24.6388 Η С 21.8629 21.4685 25.7788 Η 21.7397 22.1920 26.5689 21.1025 20.3343 25.5975 С С 19.8819 19.8457 26.3556 С 24.8157 20.5770 22.7288 25.3626 20.2731 21.8344 Η Η 25.3783 20.2699 23.6122 С 19.8840 20.4780 27.7536 20.7825 20.1754 28.2942 Η Η 19.0050 20.1468 28.3094 С 23.6215 18.3899 22.7346 22.6777 17.8410 22.7419 Η Η 24.1752 18.0936 21.8408 С 19.9107 18.3162 26.5146 Η 19.0345 17.9909 27.0800 19.9094 17.7866 25.5597 Η Ν 18.1250 19.6179 24.5314 Η 18.5307 18.7707 24.1303 18.6302 20.2904 25.6215 С Ν 18.0931 19.6281 21.0456 Η 18.5068 18.7794 21.4350 17.8360 21.3990 25.8167 С Η 17.9485 22.1255 26.6057 С 16.8203 21.3851 24.8185 Η 16.0191 22.0988 24.7102

С 17.0173 20.2685 24.0362 С 16.2665 19.7943 22.8055 С 16.9937 20.2753 21.5634 С 16.7816 21.3951 20.7896 15.9821 22.1076 20.9155 Η 17.7795 21.4145 19.7736 С 17.8769 22.1444 18.9858 Η С 18.5782 20.3061 19.9499 С 19.8154 19.8646 19.1903 С 14.8592 20.4051 22.8204 Η 14.3242 20.0773 23.7135 14.3076 20.0820 21.9358 Η С 19.7918 20.5043 17.7958 18.9032 20.1754 17.2541 Η Η 20.6807 20.2051 17.2375 16.1306 18.2623 22.8026 С Η 17.0933 17.7469 22.7914 Η 15.5885 17.9414 23.6950 19.8403 18.3360 19.0229 С Η 20.7283 18.0441 18.4577 Η 19.8580 17.8016 19.9749 19.8659 21.5681 27.7074 Η Η 14.8889 21.4959 22.8229 19.7741 21.5940 17.8481 Η 24.7471 21.6660 22.7315 Η Η 24.1917 18.0893 23.6165 20.8099 18.0219 27.0605 Η Η 18.9529 18.0138 18.4733 Η 15.5708 17.9464 21.9194 Cs 19.8150 21.8521 22.7812 Cs-C4P-Br(toluene); E(RM062X) = -1612.60616447 A.U.

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 $Cs-C4P-Br(toluene)_2-A; E(RM062X) = -1884.12445097A.U.$

Br 19.6891 19.5024 20.1622 21.5252 21.9492 18.5935 Ν Η 20.9952 21.2011 19.0475 21.1821 22.5342 17.3934 С 21.5543 21.7545 21.9747 Ν 21.1307 20.9202 21.5661 Η 22.2220 23.3559 17.0275 С 22.2693 23.9599 16.1350 Η 23.2245 23.2593 18.0356 С 24.1730 23.7732 18.0441 Η С 22.7736 22.3773 18.9906 С 23.4333 21.8947 20.2670 С 22.6701 22.3923 21.4807 С 22.9026 23.4821 22.2919 Η 23.7227 24.1759 22.1962 21.9018 23.4909 23.3051 С 21.8176 24.1959 24.1170 Η С 21.0822 22.4032 23.0922 С 19.8569 21.9366 23.8563

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Č	18 1659	27 4808	20.0493
\sim	-0.1007	_/000	-0.0175

17.4906 26.8467 21.0937 С С 18.0659 26.7441 22.3591 С 19.3374 27.2578 22.5949 С 20.0200 27.8951 21.5597 С 17.5690 27.5892 18.6710 С 21.6029 26.6657 18.0514 С 22.4227 27.1741 19.0613 С 22.6771 28.5476 19.0815 С 22.1182 29.3924 18.1283 С 21.2911 28.8753 17.1334 21.0391 27.5078 17.0963 С 21.4295 25.5954 17.9867 Η 23.3237 28.9577 19.8513 Η 22.3295 30.4550 18.1592 Η Η 20.8580 29.5327 16.3889 20.4091 27.0909 16.3180 Η С 22.9877 26.2717 20.1265 24.0128 26.5508 20.3818 Η 22.3969 26.3404 21.0483 Η Η 22.9886 25.2285 19.8025 19.9620 28.5260 19.5082 Η 16.5040 26.4289 20.9176 Η Η 17.5241 26.2528 23.1564 19.7855 27.1726 23.5781 Η 21.0048 28.3177 21.7293 Η Η 16.4892 27.4337 18.6909 17.7721 28.5702 18.2369 Η 18.0075 26.8385 18.0068 Η 23.9902 20.0122 21.1934 Η С 19.8228 20.4007 23.9492 18.9552 20.0924 24.5370 Η 19.7519 19.9175 22.9725 Η Η 20.7286 20.0403 24.4425 18.7255 20.5615 15.9759 Η Η 15.4980 20.6251 19.2131

 $Cs-C4P-Br(toluene)_2-B; E(RM062X) = -1884.12107444A.U.$ 19.9427 29.2862 22.0080 Br Ν 18.3885 26.3066 22.7548 18.7612 27.2565 22.7216 Η 17.1806 25.9154 22.2219 С Ν 21.8113 26.3982 22.1131 21.3909 27.3236 22.2178 Η С 17.0631 24.5601 22.4402 Η 16.2200 23.9459 22.1625 С 18.2329 24.1261 23.1265

Н	18.4422	23.1223	23.4653
С	19.0375	25.2268	23.3128
С	20.4124	25.3265	23.9457
С	21.4716	25.2935	22.8604
С	22.2180	24.2381	22.3846
Н	22.2086	23.2339	22,7785
С	23 0331	24 7275	21 3240
H	23.7576	24.1624	20.7595
C	22 7658	26 0707	21 1766
C	23 3056	27 0675	20 1675
C	20 6124	24 1135	24 8632
H	19 8473	24 1118	25 6424
Н	21 5995	24 1594	25 3275
C	24.6376	26.5389	19.6195
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C	20.5355	26.6059	24,7907
H	21.5242	26.6471	25.2534
Н	20.4012	27.5174	24.2044
С	23.5585	28.4331	20.8283
H	22.6542	28.8761	21.2502
Н	24.2850	28.3195	21.6361
N	21.2071	27.9995	19.0765
Н	20.9411	28.5699	19.8809
С	22.3236	27.1971	19.0173
N	17.7754	27.9006	19.7206
Н	18.2865	28,4881	20.3821
С	22.2953	26.5534	17.7994
Η	23.0414	25.8681	17.4292
С	21.1291	26.9882	17.1072
Н	20.8252	26.6921	16.1158
С	20.4719	27.8873	17.9181
С	19.1551	28.6107	17.7060
С	18.0416	27.8233	18.3719
С	17.1783	26.8896	17.8421
Η	17.1139	26.6063	16.8034
С	16.3669	26.3986	18.9043
Η	15.5727	25.6738	18.8200
С	16.7533	27.0428	20.0593
С	16.2603	26.8700	21.4843
С	18.8719	28.6851	16.1998
Η	19.6698	29.2383	15.7012
Η	17.9223	29.1948	16.0273
С	14.8517	26.2636	21.4449
Η	14.8423	25.2961	20.9402
Η	14.1764	26.9334	20.9093

С	19.2212	30.0437	18.2597
Η	19.4195	30.0778	19.3329
Η	20.0167	30.5953	17.7535
С	16.1869	28.2270	22.2049
Н	17.1570	28.7231	22.2777
Н	15.5091	28.8924	21.6651
Н	24.5200	25.5663	19.1388
Н	18.8120	27.6935	15.7487
Н	14.4813	26.1222	22.4619
Н	20.5299	23.1730	24.3147
Cs	19.6584	25.1848	19.9115
С	16.3062	21.6013	24.6298
С	16.6371	20.8342	23.5101
С	17.9136	20.2737	23.4432
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С	18.4914	21.2325	25.5798
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Н	15.3219	22.0546	24.6923
Н	18.1887	19.6782	22.5781
Н	19.8218	20.0245	24.3988
Н	19.2084	21.3889	26.3771
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С	15.6328	20.6198	22.4056
Η	16.0730	20.0715	21.5711
Η	14.7723	20.0521	22.7680
Н	15.2563	21.5742	22.0263
С	19.7676	21.7848	20.6345
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С	20.1088	22.0113	18.2691
С	18.7420	22.1927	18.0833
С	17.8791	22.1702	19.1785
С	18.3973	21.9655	20.4527
Η	20.1576	21.6512	21.6399
Η	20.7761	22.0342	17.4133
Η	18.3502	22.3556	17.0859
Η	16.8148	22.3218	19.0400
Н	17.7445	21.9745	21.3162
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Η	22.4067	21.6639	20.7769
Η	19.7782	26.5986	25.5780
Η	15.8057	28.0834	23.2185
Η	18.2709	30.5509	18.0778
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Ca		E(DM0(2Y) = 2127.1(150590.4.1)
CS-	$C4P^2$ -Br;	E(RM062A) = -212/.10159580 A.U.
Br	-0.0005	-0.9152 -0.000/
IN II	1.008/	1.5/84 -1.8014
Н	1.2856	0.7264 -1.3925
C	1.1386	2.2381 -2.8885
N	1.7792	1.5855 1.6548
Н	1.3779	0.7198 1.2911
C	1.9298	3.3427 -3.1143
Н	1.8002	4.0616 -3.9074
C	2.9710	3.3373 -2.1415
H	3.7761	4.0503 -2.0626
С	2.7928	2.2296 -1.3432
С	3.5760	1.7524 -0.1338
С	2.8725	2.2330 1.1212
С	3.1045	3.3445 1.8990
Η	3.9022	4.0566 1.7586
С	2.1303	3.3560 2.9398
Η	2.0537	4.0764 3.7387
С	1.3264	2.2518 2.7727
С	0.1315	1.7818 3.5807
С	4.9817	2.3638 -0.1806
Η	5.5066	2.0127 -1.0714
Н	5.5488	2.0745 0.7051
С	0.1908	2.4274 4.9709
Η	1.1193	2.1577 5.4748
Η	-0.6549	2.0870 5.5718
С	3.6785	0.2115 -0.1535
Η	2.6757	-0.2266 -0.1479
Η	4.1245	-0.0757 -1.1111
С	4.4696	-0.4013 1.0080
Η	4.2943	0.1809 1.9206
Η	5.5442	-0.3468 0.8029
С	4.0653	-1.8564 1.2611
Н	4.6565	-2.2620 2.0905
Η	3.0156	-1.8778 1.5821
С	4.2249	-2.7681 0.0466
Η	5.2638	-2.7296 -0.3071
Η	3.5983	-2.4050 -0.7757
С	3.8404	-4.2149 0.3483
Н	4.4362	-4.5793 1.1938
Н	2.7934	-4.2394 0.6703
С	4.0291	-5.1359 -0.8541
Н	3.7470	-6.1646 -0.6210
Н	5.0722	-5.1398 -1.1819
Н	3.4168	-4.8029 -1.6961
С	0.1357	0.2417 3.7191

Η	-0.7414	-0.0447	4.3098
Η	-0.0005	-0.2193	2.7359
С	1.4061	-0.3454	4.3358
Η	1.4617	-0.1136	5.4047
Η	2.2852	0.1174	3.8706
С	1.4856	-1.8573	4.1316
Η	0.6592	-2.3506	4.6574
Η	1.3412	-2.0766	3.0664
С	2.8170	-2.4472	4.5888
Н	2.9199	-2.3410	5.6757
Н	3.6341	-1.8648	4.1452
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Η	2.8299	-4.0070	3.1153
С	4.3532	-4.4718	4.5670
Н	4.4511	-5.5210	4.2800
Η	4.5283	-4.4000	5.6439
Н	5.1478	-3.9122	4.0646
Ν	-1.6687	1.5796	1.7993
Н	-1.2859	0.7273	1.3906
С	-1.1383	2.2393	2.8862
Ν	-1.7792	1.5858	-1.6569
Н	-1.3782	0.7201	-1.2930
С	-1.9291	3.3443	3.1117
Н	-1.7992	4.0633	3.9047
С	-2.9703	3.3390	2.1389
Н	-3.7751	4.0523	2.0599
С	-2.7925	2.2310	1.3409
С	-3.5760	1.7539	0.1316
С	-2.8722	2.2339	-1.1235
С	-3.1039	3.3452	-1.9016
Н	-3.9013	4.0576	-1.7613
С	-2.1297	3.3561	-2.9424
Н	-2.0527	4.0762	-3.7415
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С	-0.1314	1.7809	-3.5829
С	-4.9814	2.3659	0.1782
Η	-5.5064	2.0152	1.0691
Η	-5.5486	2.0765	-0.7074
С	-0.1905	2.4262	-4.9733
Н	-1.1191	2.1567	-5.4771
Н	0.6550	2.0852	-5.5740
С	-3.6791	0.2130	0.1518
Н	-2.6764	-0.2256	0.1464
Н	-4.1252	-0.0737	1.1095
С	-4.4704	-0.3998	-1.0095

Н	-4.2948 0.1819 -1.9223	
Н	-5.5450 -0.3448 -0.8045	
С	-4.0668 -1.8552 -1.2621	
Η	-4.6582 -2.2608 -2.0913	
Η	-3.0171 -1.8772 -1.5832	
С	-4.2267 -2.7663 -0.0472	
Н	-5.2656 -2.7272 0.3065	
Η	-3.5999 -2.4032 0.7749	
С	-3.8429 -4.2134 -0.3483	
Н	-4.4389 -4.5780 -1.1936	
Н	-2.7959 -4.2385 -0.6705	
С	-4.0318 -5.1338 0.8545	
Н	-3.7502 -6.1627 0.6219	
Н	-5.0749 -5.1370 1.1825	
Н	-3.4192 -4.8006 1.6963	
С	-0.1362 0.2408 -3.7208	
Н	0.7407 -0.0461 -4.3115	
Н	-0.0001 -0.2200 -2.7375	
С	-1.4070 -0.3460 -4.3371	
Н	-1.4626 -0.1144 -5.4061	
Н	-2.2859 0.1172 -3.8719	
С	-1.4871 -1.8578 -4.1326	
Н	-0.6609 -2.3515 -4.6584	
Н	-1.3427 -2.0769 -3.0673	
С	-2.8187 -2.4473 -4.5896	
Н	-2.9216 -2.3414 -5.6765	
Н	-3.6356 -1.8643 -4.1462	
С	-2.9847 -3.9127 -4.1962	
Η	-2.1977 -4.5083 -4.6715	
Η	-2.8324 -4.0065 -3.1155	
С	-4.3558 -4.4712 -4.5671	
Н	-4.4542 -5.5202 -4.2798	
Η	-4.5309 -4.3996 -5.6440	
Η	-5.1502 -3.9110 -4.0649	
Η	0.1478 3.5161 4.9091	
Н	-4.9439 3.4559 0.2115	
Н	-0.1469 3.5149 -4.9118	
Η	4.9446 3.4539 -0.2141	
Cs	0.0004 3.8085 -0.0013	

LAMMPS parameters.dat file

bond_coeff	1 1.53 299.67 -501.77 679.81 #c c
bond_coeff	2 1.101 345 -691.89 844.6 # c h
bond_coeff	3 1.53 299.67 -501.77 679.81 #c c
bond_coeff	4 1.0982 372.83 -803.45 894.32 # cp h
bond_coeff	5 1.417 470.84 -627.62 1327.6 # cp cp
bond_coeff	6 1.53 299.67 -501.77 679.81 #c c
bond coeff	7 1.101 345 -691.89 844.6 # c h

bond coeff 8 1.501 321.9 -521.82 572.16 # c cp bond coeff 9 1.53 299.67 -501.77 679.81 # c bond coeff 10 1.0012 465.86 -1066.2 1496.6 # h* nn bond coeff 11 1.3912 447.04 -784.53 886.17 # cp nn bond coeff 12 1.501 321.9 -521.82 572.16 # c cp angle_coeff 1 111.87 40.837 -15.667 0 # cp nn h* 2 120 50 0 0 #* n angle_coeff angle coeff 3 121.46 61.065 -21.617 0 # cp cp nn angle coeff 4 120.05 44.715 -22.735 0 # c cp cp angle_coeff 5 120 65 0 0 #* cp angle coeff 6 117.94 35.156 -12.468 0 # ср ср h angle coeff 7 118.9 61.023 -34.993 0 # cp cp cp angle coeff 8 112.67 39.516 -7.443 -9.5583 # c с с 9 108.4 43.959 -8.3924 -9.3379 # c angle coeff с cp angle coeff 10 108.4 43.959 -8.3924 -9.3379 # c с cp angle_coeff 11 111 44.323 -9.4454 0 # cp c cp angle_coeff 12 110.77 41.453 -10.604 5.129 # c h с angle_coeff 13 107.66 39.641 -12.921 -2.4318 # h h с angle_coeff 14 110.77 41.453 -10.604 5.129 # c с h angle_coeff 15 110.77 41.453 -10.604 5.129 # c с h angle coeff 16 107.66 39.641 -12.921 -2.4318 # h с h angle coeff 17 112.67 39.516 -7.443 -9.5583 # c с с angle coeff 18 112.67 39.516 -7.443 -9.5583 # c с с angle_coeff 19 110.77 41.453 -10.604 5.129 # c с h angle coeff 20 112.67 39.516 -7.443 -9.5583 # c с с angle coeff 21 110.77 41.453 -10.604 5.129 # c с h 22 120.05 44.715 -22.735 0 angle coeff # c cp cp angle coeff 23 111 44.323 -9.4454 0 # cp c h dihedral coeff $1 \ 0 \ 0 \ 1.219 \ 0 \ 0 \ 0$ # cp cp h* nn # * dihedral coeff $2\ 0\ 0\ 2.25\ 0\ 0\ 0$ cp n dihedral coeff 3002.25000 # * * cp_ n_ # * dihedral coeff 4002.25000 cp_ n dihedral coeff 5001.3331000 # h cp cp nn dihedral coeff 6005.3826000 # cp cp cp nn # c cp dihedral coeff 7001.559000 h cp dihedral coeff 8 0 0 4.4072 0 0 0 # c cp cp cp dihedral_coeff 9 -0.2802 0 -0.0678 0 -0.0122 0 # c с cp cp dihedral coeff 10 -0.2802 0 -0.0678 0 -0.0122 0 # c с cp cp dihedral coeff 11 -0.2802 0 -0.0678 0 -0.0122 0 # cp с cp cp dihedral coeff 12 8.3667 0 1.1932 0 0 0 # cp cp cp cp dihedral coeff 13 0 0 3.9661 0 0 0 # cp h cp cp 14 0 0 1.8769 0 0 0 #h dihedral coeff h cp ср dihedral coeff 15 0 0 0.0316 0 -0.1681 0 # c с с h dihedral coeff 16 -0.0228 0 0.028 0 -0.1863 0 # cp c h с dihedral coeff 17 0 0 0.0316 0 -0.1681 0 # c h с с dihedral coeff 18 0 0 0.0514 0 -0.143 0 # c с с с dihedral_coeff 19 -0.0228 0 0.028 0 -0.1863 0 # cp h с с dihedral coeff 200000.1580 #* с с # c dihedral coeff 21 0 0 0.0316 0 -0.1681 0 с h dihedral_coeff 22 -0.1432 0 0.0617 0 -0.1083 0 # h с с h dihedral_coeff 23 0 0 0.0514 0 -0.143 0 # c с с с dihedral coeff 24 0 0 0.0316 0 -0.1681 0 # c с с h dihedral coeff 25 0 0 0.0514 0 -0.143 0 # c с с с dihedral coeff 26 0 0 0.0316 0 -0.1681 0 # c с с h dihedral coeff 27 0 0 0.0514 0 -0.143 0 # c с с с dihedral coeff 28 -0.1432 0 0.0617 0 -0.1083 0 # h c с h # c dihedral coeff 29 0 0 0.0316 0 -0.1681 0 с с h dihedral coeff 30 0 0 4.4072 0 0 0 # c cp cp cp dihedral coeff 31 0 0 1.559 0 0 0 # c cp h cp 32 -0.2801 0 -0.0678 0 -0.0122 0 # h dihedral coeff с cp cp # * improper coeff 1 0.1 0 n 2 10 0 # * improper coeff cp 3 4.8912 0 # cp cp improper coeff h cp improper_coeff 4 0.0 0.0 # c2 c0 c3 cp improper coeff 5 0.0 0.0 # c3 c0 cp cp improper coeff 6 0.0 0.0 # c2 c0cp cp

improper_coeff	$7\ 0.0\ 0.0$ # c0 c3 hc hc	
improper_coeff	8 0.0 0.0 # hc c3 hc hc	
improper_coeff	$9\ 0.0\ 0.0$ # c0 c2 c2 hc	
improper_coeff	$10\ 0.0\ 0.0$ # c0 c2 hc hc	
improper_coeff	$11\ 0.0\ 0.0$ # c2 c2 hc hc	
improper_coeff	$12\ 0.0\ 0.0$ # c2 c2 c2 hc	
improper_coeff	$13\ 0.0\ 0.0$ # c2 c2 c3 hc	
improper_coeff	14 0.0 0.0 # c3 c2 hc hc	
improper_coeff	15 0.0 0.0 # c2 c3 hc hc	
improper_coeff	16 7.8153 0 #с ср ср ср	
improper_coeff	17 0.0 0.0 # cp c3 hc hc	
angle_coeff	1 bb 4.5393 1.3912 1.0012 # cp nn h*	
angle coeff	2 bb 0.0 1.3912 1.3912 # * n *	
angle coeff	3 bb 46.951 1.417 1.3912 # cp cp nn	
angle coeff	4 bb 12.068 1.501 1.417 # c cp cp	
angle coeff	5 bb 0.0 1.501 1.3912 #* cp *	
angle coeff	6 bb 1.0795 1.417 1.0982 #cp cp h	
angle coeff	7 bb 68.286 1.417 1.417 # cp cp cp	
angle coeff	8 bb 0 1.53 1.53 # c c c	
angle coeff	9 bb 0 1.53 1.501 # c c cp	
angle coeff	10 bb 0 1.53 1.501 # c c cp	
angle coeff	11 bb 0.0 1.501 1.501 # cp c cp	
angle coeff	12 bb 3.3872 1.53 1.101 # c c h	
angle_coeff	$13 \text{ bb} 5 3316 1 101 1 101 \ \#\text{h} \text{c} \text{h}$	
angle_coeff	14 bb 3 3872 1 53 1 101 # c c h	
angle_coeff	15 bb 3 3872 1 53 1 101 # c c h	
angle_coeff	16 bb 5 3316 1 101 1 101 #h c h	
angle_coeff	$17 \text{ bb} 0.153153 \pm c$	
angle_coeff	17 bb 01.551.55 #c c c	
angle_coeff	10 bb 3 3872 1 53 1 101 # c c b	
angle_coeff	20 bb 0.153153 #c c c	
angle_coeff	200001.551.55 # C C C	
angle_coeff	21 bb 5.3672 1.35 1.101 # c c n	
angle_coeff	$22 \text{ bb} 12.008 1.501 1.417 \ \ \pi \text{c} \text{ cp} \text{ cp}$	
angle_coeff	25 00 2.9108 1.501 1.101 # cp C II 1 bo 28 57 16 552 1 2012 1 0012 # cp pp	ե*
angle_coeff	1 ba 38.57 10.552 1.5912 1.0012 # cp IIII 2 ba 0 0 0 0 1 2012 1 2012 # * p *	п.
angle_coeff	$2 \text{ ba } 0.0 0.0 1.3912 1.3912 \#^{\circ} \text{ II}$	
angle_coeff	5 ba 39.404 / 5.055 1.41 / 1.5912 # cp	IIII
angle_coeff	4 ba 4/.058 51.0// 1.501 1.41/# c cp cp	
angle_coeff	5 ba 0.0 0.0 1.501 1.3912 # cp m	1.
angle_coeff	6 ba 20.003 24.218 1.417 1.0982 # cp cp	n
angle_coeff	/ ba 28.8/1 28.8/1 1.41/1.41/7 cp cp cp	
angle_coeff	8 ba 8.016 8.016 1.55 1.55 # c	
angle_coeff	9 ba 0 0 1.53 1.501 #c c cp	
angle_coeff	10 ba 0 0 1.53 1.501 # c	
angle_coeff	11 ba 0.0 0.0 1.501 1.501 # cp c cp	
angle_coeff	12 ba 20.754 11.421 1.53 1.101 # c c h	
angle_coeff	13 ba 18.103 18.103 1.101 1.101 # h c h	
angle_coeff	14 ba 20.754 11.421 1.53 1.101 # c c n	
angle_coeff	15 ba 20.754 11.421 1.53 1.101 # c c n	
angle_coeff	16 ba 18.103 18.103 1.101 1.101 # h c h	
angle_coeff	1/ba 8.016 8.016 1.53 1.53 #c c c	
angle_coeff	18 ba 8.016 8.016 1.53 1.53 # c c c	
angle_coeff	19 ba 20.754 11.421 1.53 1.101 # c c h	
angle_coeff	20 ba 8.016 8.016 1.53 1.53 # c c c	
angle_coeff	21 ba 20.754 11.421 1.53 1.101 # c c h	
angle_coeff	22 ba 47.058 31.077 1.501 1.417 # c cp cp	
angle_coeff	23 ba 26.461 11.772 1.501 1.101 # cp c h	
dihedral_coeff	1 mbt 0 3.2085 0 1.3912 # cp cp nn	h*
dihedral_coeff	2 mbt 0.0 0.0 0.0 1.3912 #* cp_ n_	*
dihedral_coeff	3 mbt 0.0 0.0 0.0 1.3912 #* cp_ n_	*
dihedral_coeff	4 mbt 0.0 0.0 0.0 1.3912 #* cp_ n_	*
dihedral_coeff	5 mbt 0 2.2883 0 1.417 # h cp cp nn	
dihedral_coeff	6 mbt 0 -0.5693 0 1.417 # cp cp cp	nn
dihedral_coeff	7 mbt 0 3.9421 0 1.417 # c cp cp h	
dihedral_coeff	8 mbt 0 9.1792 0 1.417 # c cp cp cp	
dihedral_coeff	9 mbt 0 0 0 1.501 # c c cp cp	

dihedral coeff 10 mbt 0 0 0 1.501 # c с cp cp dihedral coeff 11 mbt 0.0 0.0 0.0 1.501 #cp c cp cp dihedral coeff 12 mbt 27.599 -2.312 0 1.417 # cp cp cp cp # cp cp dihedral coeff 13 mbt 0 -1.1521 0 1.417 h cp dihedral coeff 14 mbt 0 4.8228 0 1.417 # h h cp cp dihedral coeff 15 mbt -14.879 -3.6581 -0.3138 1.53 # c с с h dihedral coeff 16 mbt 0 0 0 1.53 # cp c h с dihedral coeff 17 mbt -14.879 -3.6581 -0.3138 1.53 # c с с h dihedral coeff 18 mbt -17.787 -7.1877 0 1.53 # c с с с dihedral_coeff 19 mbt 0 0 0 1.53 #cp c h с dihedral coeff 20 mbt 0.0 0.0 0.0 1.53 # * c с dihedral coeff 21 mbt -14.879 -3.6581 -0.3138 1.53 #cс с h dihedral coeff 22 mbt -14.261 -0.5322 -0.4864 1.53 # h h с с 23 mbt -17.787 -7.1877 0 1.53 # c dihedral coeff с c c dihedral coeff 24 mbt -14.879 -3.6581 -0.3138 1.53 # c h с с dihedral coeff 25 mbt -17.787 -7.1877 0 1.53 # c с 26 mbt -14.879 -3.6581 -0.3138 1.53 dihedral_coeff # c с с h dihedral_coeff 27 mbt -17.787 -7.1877 0 1.53 # c c с с dihedral_coeff 28 mbt -14.261 -0.5322 -0.4864 1.53 h # h с dihedral_coeff 29 mbt -14.879 -3.6581 -0.3138 1.53 # c h c с 30 mbt 0 9.1792 0 1.417 # c cp dihedral coeff cp cp dihedral coeff 31 mbt 0 3.9421 0 1.417 # c cp cp h # h dihedral coeff 32 mbt -5.5679 1.4083 0.301 1.501 с cp cp dihedral coeff 1 ebt 0 -3.7281 0 0 0.2849 0 1.417 1.0012 # cp cp nn h* dihedral coeff 2 ebt 0.0 0.0 0.0 0.0 0.0 0.0 1.501 1.0012 # * cp n dihedral coeff 3 ebt 0.0 0.0 0.0 0.0 0.0 0.0 1.417 1.3912 # * cp n # * 4 ebt 0.0 0.0 0.0 0.0 0.0 0.0 1.501 1.3912 dihedral coeff cp n dihedral_coeff 5 ebt 0 -2.6482 0 0 -1.6402 0 1.0982 1.3912 # h cp cp nn 6 ebt 0 -6.5404 0 0 -7.3477 0 1.417 1.3912 dihedral coeff # cp cp cp nn dihedral coeff 7 ebt 0 -1.797 0 0 -0.4879 0 1.501 1.0982 # c h cp cp dihedral_coeff 8 ebt 0 0.2421 0 0 -0.6918 0 1.501 1.417 # c cp cp cp 9 ebt 0 0 0 0 0 0 1.53 1.417 # c c dihedral coeff cp cp dihedral coeff 10 ebt 0 0 0 0 0 0 1.53 1.417 # c c cp cp 11 ebt 0.0 0.0 0.0 0.0 0.0 0.0 1.501 1.417 # cp c dihedral coeff cp cp 12 ebt -0.1185 6.3204 0 -0.1185 6.3204 0 1.417 1.417 # cp dihedral coeff cp cp cp dihedral coeff 13 ebt 0 -6.8958 0 0 -0.4669 0 1.417 1.0982 # cp cp cp h #h cp dihedral_coeff 14 ebt 0 -0.689 0 0 -0.689 0 1.0982 1.0982 cp h dihedral coeff 15 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 # c h с c dihedral coeff 16 ebt 0 0 0 0 0 0 1.501 1.101 # cp c с h 17 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 dihedral coeff # c h с с dihedral coeff 18 ebt -0.0732 0 0 -0.0732 0 0 1.53 1.53 # c c с с dihedral coeff 19 ebt 0 0 0 0 0 0 1.501 1.101 # cp c с h dihedral coeff 20 ebt 0.0 0.0 0.0 0.0 0.0 0.0 1.501 1.53 #* c с # c dihedral coeff 21 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 h c с dihedral coeff 22 ebt 0.213 0.312 0.0777 0.213 0.312 0.0777 1.101 1.101 # h с с h dihedral coeff 23 ebt -0.0732 0 0 -0.0732 0 0 1.53 1.53 #c c с с dihedral_coeff 24 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 # c с h dihedral coeff 25 ebt -0.0732 0 0 -0.0732 0 0 1.53 1.53 # c c с c dihedral coeff 26 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 # c h с dihedral_coeff 27 ebt -0.0732 0 0 -0.0732 0 0 1.53 1.53 #ccc с dihedral_coeff 28 ebt 0.213 0.312 0.0777 0.213 0.312 0.0777 1.101 1.101 # h h с с dihedral coeff 29 ebt 0.2486 0.2422 -0.0925 0.0814 0.0591 0.2219 1.53 1.101 # c c с h dihedral coeff 30 ebt 0 0.2421 0 0 -0.6918 0 1.501 1.417 # c cp cp cp dihedral coeff 31 ebt 0 -1.797 0 0 -0.4879 0 1.501 1.0982 # c cp h cp dihedral coeff 32 ebt 1.3997 0.7756 0 -0.5835 1.122 0.3978 1.101 1.417 # h с cp cp 1 at 0 1.2616 0 0 0.7744 0 121.46 111.87 h* dihedral coeff # cp cp nn * * dihedral coeff 2 at 0.0 0.0 0.0 0.0 0.0 0.0 120 111.87 # cp n 3 at 0.0 0.0 0.0 0.0 0.0 0.0 121.46 120 # * * dihedral coeff cp n dihedral coeff 4 at 0.0 0.0 0.0 0.0 0.0 0.0 120 120 #* cp n dihedral coeff 5 at 0 2.9813 0 0 0.2787 0 117.94 121.46 # h cp cp nn dihedral coeff 6 at 0 9.0901 0 0 -6.0882 0 118.9 121.46 # cp cp nn cp # c dihedral coeff 7 at 0 -0.1242 0 0 3.4601 0 120.05 117.94 cp cp h dihedral coeff 8 at 0 -4.4683 0 0 3.8987 0 120.05 118.9 # c cp cp cp 9 at 0 0 0 0 0 0 108.4 120.05 # c c dihedral_coeff cp cp dihedral coeff 10 at 0 0 0 0 0 0 108.4 120.05 # c c cp cp dihedral coeff 11 at 0.0 0.0 0.0 0.0 0.0 0.0 111 120.05 # cp с cp cp

dihedral coeff 12 at 1.9767 1.0239 0 1.9767 1.0239 0 118.9 118.9 # cp cp cp cp dihedral coeff 13 at 0 2.5014 0 0 2.7147 0 118.9 117.94 # cp cp cp h dihedral coeff 14 at 0 2.4501 0 0 2.4501 0 117.94 117.94 #h cp h cp dihedral coeff 15 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 # c с с h dihedral coeff 16 at 0 0 0 0 0 0 108.4 110.77 # cp c c h dihedral coeff 17 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 # c с с h dihedral coeff 18 at 0.3886 -0.3139 0.1389 0.3886 -0.3139 0.1389 112.67 112.67 # c с с с dihedral coeff 19 at 0 0 0 0 0 0 108.4 110.77 # cp c c h 20 at 0.0 0.0 0.0 0.0 0.0 0.0 108.4 112.67 #* dihedral coeff * с 21 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 # c dihedral_coeff с с h dihedral coeff 22 at -0.8085 0.5569 -0.2466 -0.8085 0.5569 -0.2466 110.77 110.77 # h с h c 23 at 0.3886 -0.3139 0.1389 0.3886 -0.3139 0.1389 112.67 112.67 # c dihedral coeff с с с dihedral coeff 24 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 #c c h 25 at 0.3886 -0.3139 0.1389 0.3886 -0.3139 0.1389 112.67 112.67 # c dihedral coeff с с с 26 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 dihedral coeff h #c c с dihedral coeff 27 at 0.3886 -0.3139 0.1389 0.3886 -0.3139 0.1389 112.67 112.67 # c c с с dihedral coeff 28 at -0.8085 0.5569 -0.2466 -0.8085 0.5569 -0.2466 110.77 110.77 # h с с h dihedral_coeff 29 at -0.2454 0 -0.1136 0.3113 0.4516 -0.1988 112.67 110.77 # c h с с dihedral_coeff 30 at 0 -4.4683 0 0 3.8987 0 120.05 118.9 #c cp cp cp dihedral coeff # c cp 31 at 0 -0.1242 0 0 3.4601 0 120.05 117.94 h cp dihedral coeff 32 at 4.6266 0.1632 0.0461 0.2251 0.6548 0.1237 111 120.05 # h с cp cp dihedral coeff 1 aat -7.1755 121.46 111.87 # cp cp h* nn dihedral coeff 2 aat 0.0 120 111.87 #* cp n * 3 aat 0.0 121.46 120 #* dihedral coeff n cp dihedral coeff 4 aat 0.0 120 120 #* cn n dihedral coeff 5 aat 0 117.94 121.46 # h cp cp nn 6 aat 0 118.9 121.46 # cp dihedral coeff cp cp nn dihedral_coeff 7 aat 4.4444 120.05 117.94 # c h cp cp 8 aat -14.41 120.05 118.9 dihedral coeff # c cp cp cp dihedral coeff 9 aat 0 108.4 120.05 # c с cp cp dihedral coeff 10 aat 0 108.4 120.05 # c с cp cp dihedral coeff 11 aat 0.0 111 120.05 # cp с cp cp dihedral coeff 12 aat 0 118.9 118.9 # cp cp cp cp dihedral coeff # cp 13 aat -4.8141 118.9 117.94 cp cp h dihedral coeff 14 aat 0.3598 117.94 117.94 # h cp cp h dihedral coeff 15 aat -16.164 112.67 110.77 # c с h с 16 aat 0 108.4 110.77 # cp dihedral_coeff с с h dihedral coeff 17 aat -16.164 112.67 110.77 # c h c с dihedral coeff 18 aat -22.045 112.67 112.67 # c c c с dihedral coeff 19 aat 0 108.4 110.77 # cp с с h dihedral coeff 20 aat 0.0 108.4 112.67 # * с c dihedral coeff 21 aat -16.164 112.67 110.77 # c с h c dihedral coeff 22 aat -12.564 110.77 110.77 #h h с с dihedral coeff 23 aat -22.045 112.67 112.67 # c с с с # c dihedral coeff 24 aat -16.164 112.67 110.77 с h с dihedral coeff 25 aat -22.045 112.67 112.67 # c с с с 26 aat -16.164 112.67 110.77 dihedral_coeff # c с с h dihedral coeff 27 aat -22.045 112.67 112.67 # c с c с dihedral coeff 28 aat -12.564 110.77 110.77 # h с с h dihedral_coeff 29 aat -16.164 112.67 110.77 # c h с с dihedral_coeff 30 aat -14.41 120.05 118.9 # c cp cp cp dihedral coeff 31 aat 4.4444 120.05 117.94 # c h cp cp dihedral coeff 32 aat -5.8888 111 120.05 # h с cp cp dihedral coeff # cp 1 bb13 4.2366 1.417 1.0012 h' cp nn dihedral coeff 2 bb13 0.0 1.501 1.0012 # * cp n dihedral coeff 3 bb13 0.0 1.417 1.3912 # * * cp n # * * dihedral coeff 4 bb13 0.0 1.501 1.3912 cp n 5 bb13 -1.0746 1.0982 1.3912 #h dihedral coeff cp cp nn dihedral coeff 6 bb13 -9.7999 1.417 1.3912 # cp cp nn cp dihedral coeff 7 bb13 0.8743 1.501 1.0982 # c h cp cp 8 bb13 2.5085 1.501 1.417 dihedral coeff # c cp cp cp dihedral coeff 9 bb13 0.0 1.53 1.417 # c с cp cp dihedral coeff 10 bb13 0.0 1.53 1.417 # c cp cp с dihedral_coeff 11 bb13 0.0 1.501 1.417 # cp c cp cp 12 bb13 53 1.417 1.417 # cp cp dihedral coeff cp cp dihedral coeff 13 bb13 -6.2741 1.417 1.0982 # cp cp cp h

dihedral_coeff	14 bb13 -1.7077 1.0982 1.0982 #h cp cp h			
dihedral_coeff	15 bb13 0.0 1.53 1.101 # c c c h			
dihedral_coeff	16 bb13 0.0 1.501 1.101 # cp c c h			
dihedral_coeff	17 bb13 0.0 1.53 1.101 # c c c h			
dihedral_coeff	18 bb13 0.0 1.53 1.53 #c c c c			
dihedral_coeff	19 bb13 0.0 1.501 1.101 # cp c c h			
dihedral_coeff	20 bb13 0.0 1.501 1.53 #* c_ c_ *			
dihedral_coeff	21 bb13 0.0 1.53 1.101 # c c c h			
dihedral_coeff	22 bb13 0.0 1.101 1.101 # h c c h			
dihedral_coeff	23 bb13 0.0 1.53 1.53 #c c c c			
dihedral coeff	24 bb13 0.0 1.53 1.101 # c c c h			
dihedral coeff	25 bb13 0.0 1.53 1.53 #c c c c			
dihedral coeff	26 bb13 0.0 1.53 1.101 # c c c h			
dihedral coeff	27 bb13 0.0 1.53 1.53 #c c c c			
dihedral coeff	28 bb13 0.0 1.101 1.101 # h c c h			
dihedral coeff	29 bb13 0.0 1.53 1.101 # c c c h			
dihedral coeff	30 bb13 2.5085 1.501 1.417 # c cp cp cp			
dihedral coeff	31 bb13 0.8743 1.501 1.0982 #c cp cp h			
dihedral coeff	32 bb13 -3.4826 1.101 1.417 #h c cp cp			
improper coeff	1 aa 0.0 0.0 0.0 120.0 120.0 120.0 #* n * *			
improper coeff	2 aa 0.0 0.0 0.0 120.0 120.0 120.0 #* cp * *	k		
improper coeff	3 aa 0 0 0 118.9 117.94 117.94 # cp cp cp h			
improper coeff	4 aa 0.0 0.0 0.0 120.0 120.0 120.0 # c2 c0 c3 c	cp		
improper coeff	5 aa 0.0 0.0 0.0 120.0 120.0 120.0 # c3 c0 cp o	cp		
improper coeff	6 aa 0.0 0.0 0.0 120.0 120.0 120.0 # c2 c0 cp	cp		
improper coeff	7 aa 0.2738 -0.4825 0.2738 110.77 110.77 107.66 #	c c	h	h
improper coeff	8 aa -0.3157 -0.3157 -0.3157 107.66 107.66 107.66 #	h c	h	h
improper coeff	9 aa -1.3199 -1.3199 0.1184 112.67 110.77 110.77 #	c c	с	h
improper_coeff	10 aa 0.2738 -0.4825 0.2738 110.77 110.77 107.66 #	ⁱ c c	h	h
improper coeff	11 aa 0.2738 -0.4825 0.2738 110.77 110.77 107.66 #	[!] c c	h	h
improper coeff	12 aa -1.3199 -1.3199 0.1184 112.67 110.77 110.77 #	¢c c	с	h
improper coeff	13 aa -1.3199 -1.3199 0.1184 112.67 110.77 110.77 #	¢c c	с	h
improper_coeff	14 aa 0.2738 -0.4825 0.2738 110.77 110.77 107.66 #	ⁱ c c	h	h
improper coeff	15 aa 0.2738 -0.4825 0.2738 110.77 110.77 107.66 #	¹ c c	h	h
improper coeff	16 aa 0 0 0 120.05 120.05 118.9 # c cp cp cp			
improper_coeff	17 aa 2.3794 3.0118 2.3794 111 111 107.66 # cp c	h	h	
pair_coeff	1 1 0.15 5.4135 # Br -			
pair coeff	2 2 0.007 3.75 # c0 -			
pair coeff	3 3 0.0634 3.7584 # c2 -			
pair coeff	4 0.071 3.884 # c3 -			
pair coeff	5 5 0.0668 3.9147 # cp -			
pair coeff	6 6 0.85 4.1214 # Cs+ -			
pair coeff	7 7 0.023 2.878 # h -			
pair coeff	8 8 0.008 1.087 # h* -			
pair_coeff	9 9 0.092 3.89 # nn -			