

Electronic Supplementary Information (ESI)

Light-controlled reversible self-assembly of nanorod suprastructures

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S1 Materials and instruments

Reagent grade β -cyclodextrin (β -CD) was purchased from Sigma and recrystallized twice from water followed by dehydration in vacuum at 95 °C. 4,4'-dipyridine (DPD) and zinc chloride (ZnCl_2) were commercially available and used without further purification. 2,3,3-trimethylindolenine, propane sultone, and 2-hydroxybenzaldehyde were commercially obtained and used without further purification. All other chemicals used in this study were of analytical grade.

The pH values were collected on a commercially available instrument. ^1H NMR spectra were recorded in deuterium oxide at 298 ± 0.1 K on a Bruker AV600 spectrometer. Fourier transform infrared spectra (FT-IR) were performed on a Bio-rad FTS6000 with samples prepared as KBr pellets. The UV-vis absorption spectral studies were carried out on a ThermoFisher EVO300 PC spectrophotometer. Transmission electron microscope (TEM) experiments were performed with a Tecnai G2 F20 to analyse the morphology of supramolecular assemblies. Dynamic light scattering (DLS) was conducted with Nanobrook 173plus (Brookhaven Co. USA) using a vertically polarized light at $\lambda = 488$ nm.

S2 Synthesis and characterization of compounds

Synthesis of MEH

The synthesis of MEH was carried out following a literature procedure.¹ Starting material 2,3,3-trimethyl-1-(3-sulfonatepropyl)-3H-indolium was synthesized by the reaction of 2,3,3-trimethylindolenine with propane sultone. To an anhydrous ethanol were added 2-hydroxybenzaldehyde and 2,3,3-trimethyl-1-(3-sulfonatepropyl)-3Hindolium. The reaction mixture was refluxed overnight, filtered at room temperature give orange solid MEH.

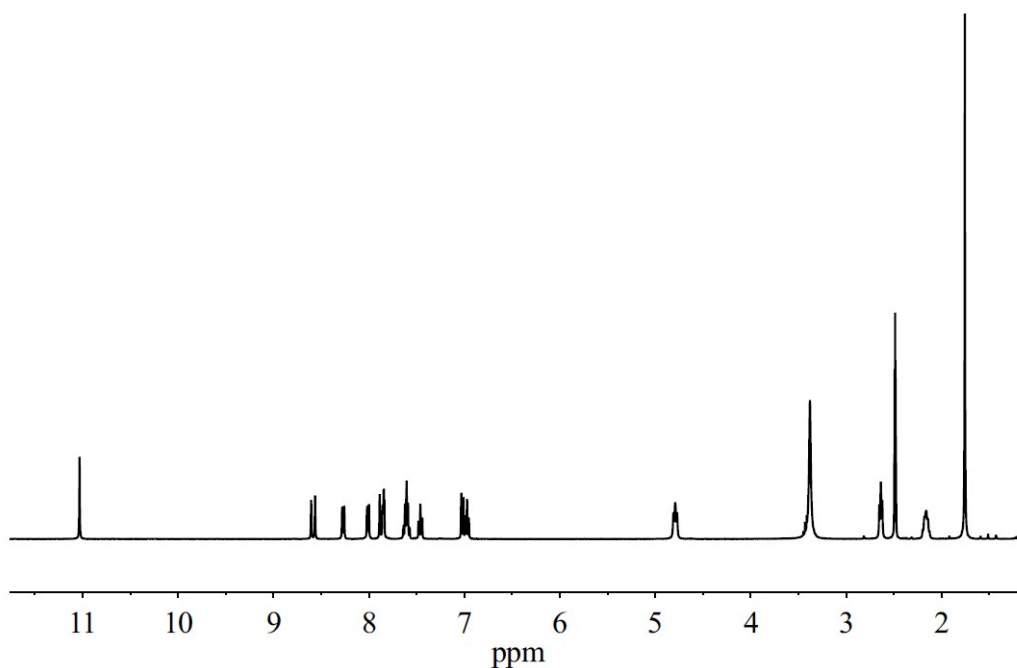


Figure S1. ^1H NMR spectrum of MEH (400 MHz, D_2O , 298 K).

Preparation of polypseudorotaxane 1

Complex $\beta\text{-CD/DPD}$ was synthesized by the reaction of DPD with $\beta\text{-CD}$ in aqueous solution. As for the synthesis of polypseudorotaxane **1**, an aqueous solution of ZnCl_2 was slowly added to an aqueous solution of $\beta\text{-CD/DPD}$. The reaction mixture was then stirred at 40 °C for 7 h. The resulting solution was concentrated under a reduced pressure. After filtration the solid was recrystallized from water and dried under vacuum.

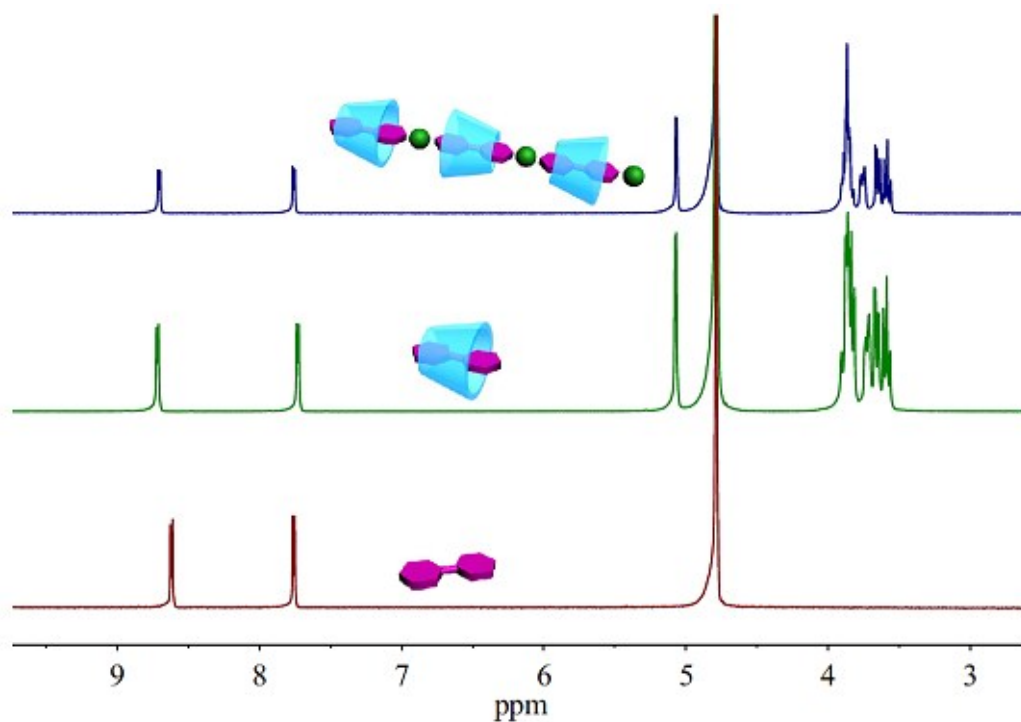


Figure S2. ^1H NMR spectra of polypseudorotaxane **1**, complex $\text{DPD} \subset \beta\text{-CD}$ and DPD (400 MHz, D_2O , 298 K).

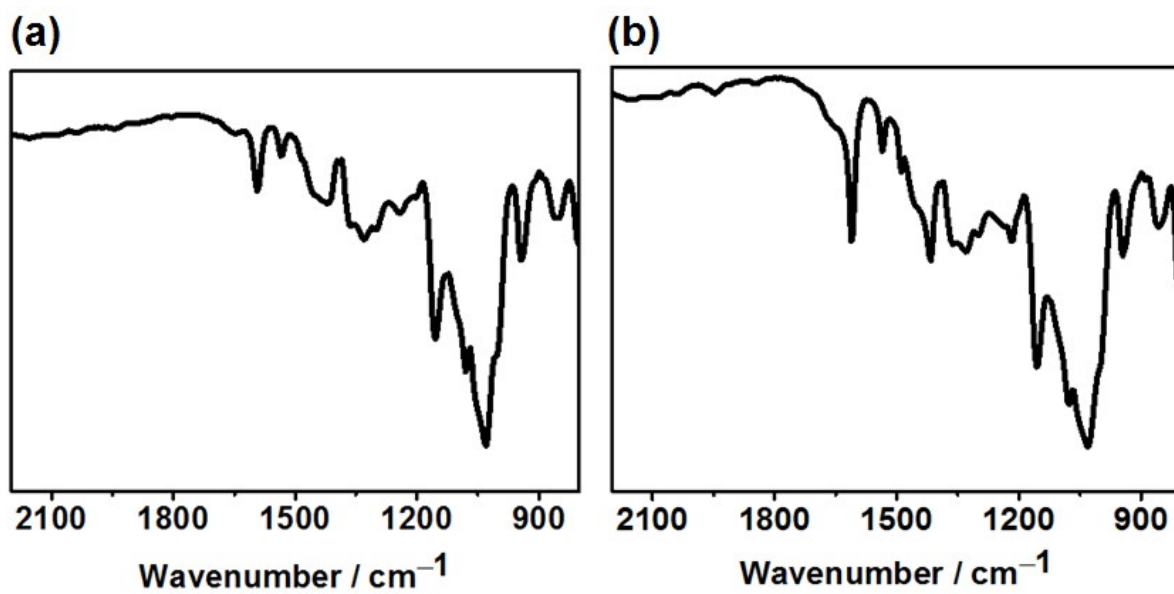


Figure S3. FT-IR of (a) complex $\text{DPD} \subset \beta\text{-CD}$ and (b) polypseudorotaxane **1**.

S3 Protonation of DPD and DPD@polypseudorotaxane by Hydrochloric Acid (HCl)

Adding of DPD with HCl to adjust the pH to 3 was performed in D₂O. ¹H NMR changes were monitored during the adding process because of the protonation of DPD. In addition, the protonation of polypseudorotaxane **1** at different pH values was carried out by adding a series of DCl to a D₂O solution of **1** and detected by ¹H NMR.

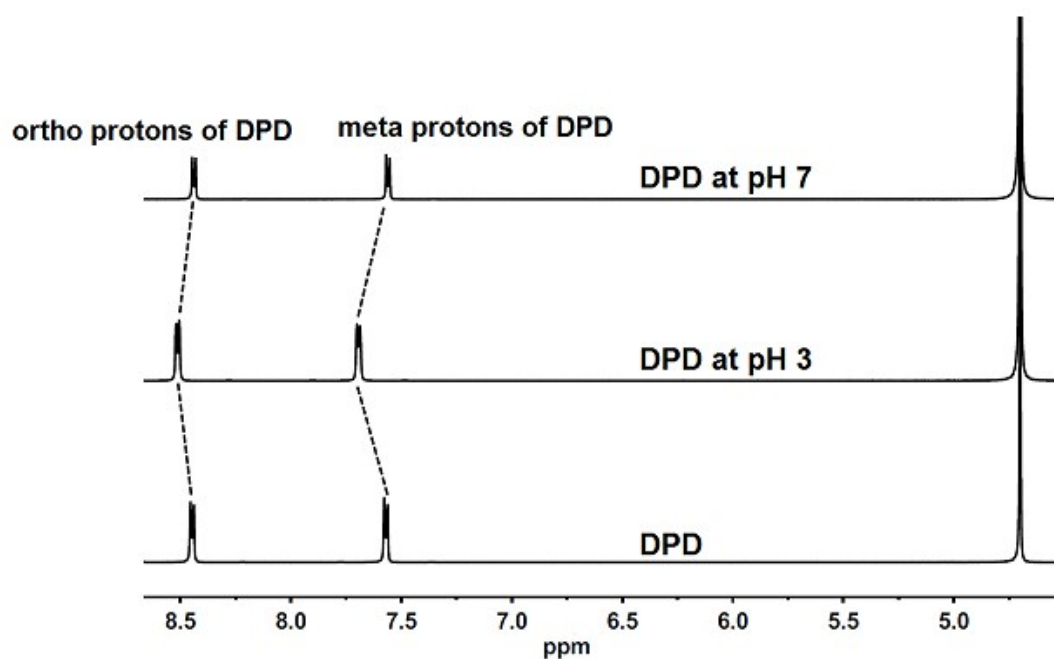


Figure S4. ¹H NMR of DPD upon the sequential addition of HCl and NaOH (400 MHz, D₂O, 298 K).

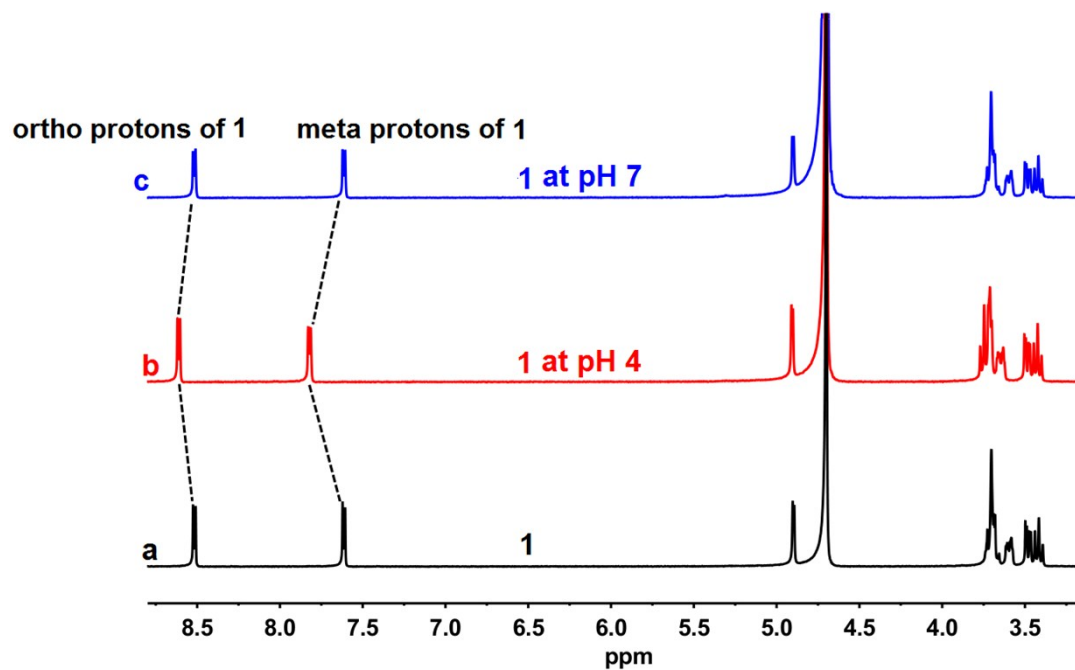


Figure S5. ^1H NMR of (a) **1** (2.0×10^{-4} M) (b) **1** (2.0×10^{-4} M) + HCl with pH 4 (c) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7 (400 MHz, D_2O , 298 K).

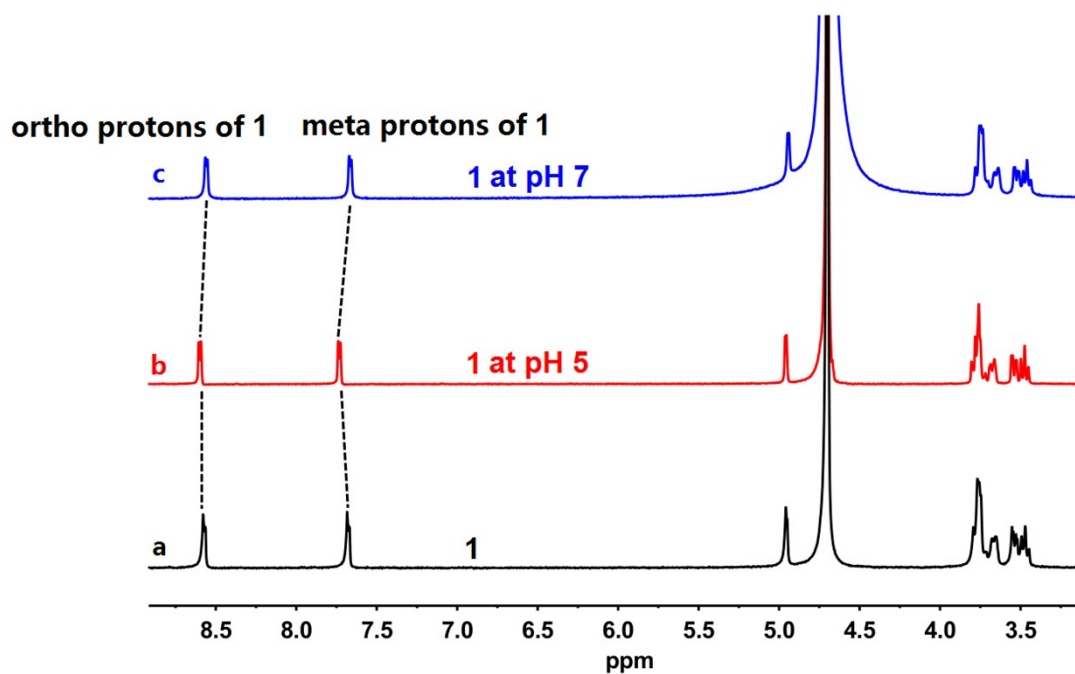


Figure S6. ^1H NMR of (a) **1** (2.0×10^{-4} M) (b) **1** (2.0×10^{-4} M) + HCl with pH 5 (c) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7 (400 MHz, D_2O , 298 K).

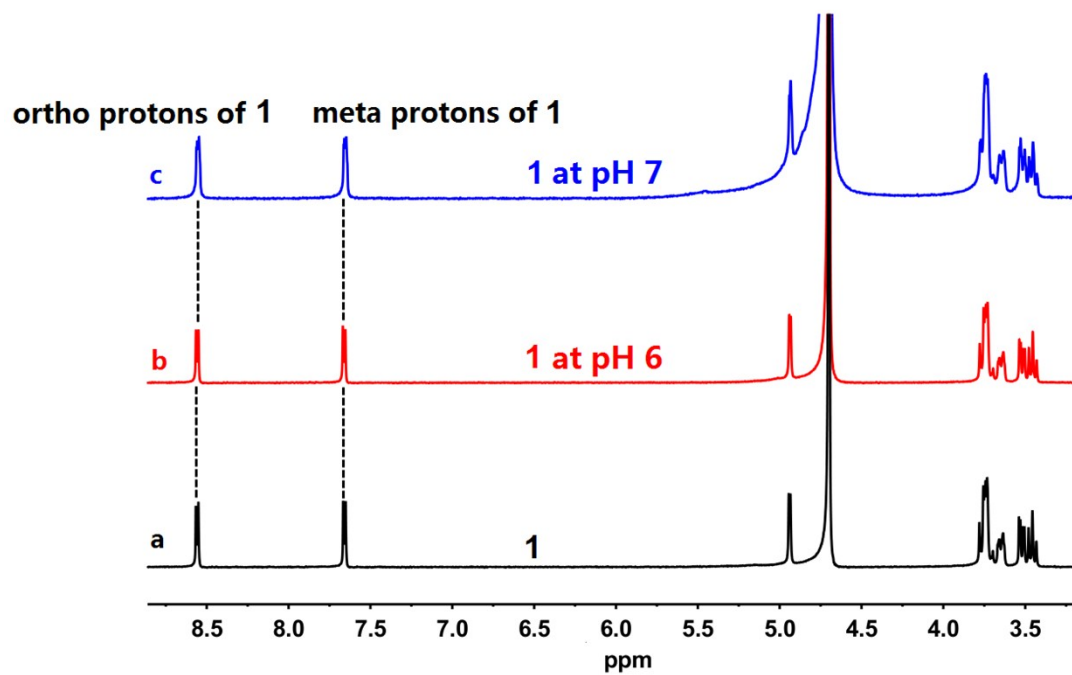


Figure S7. ^1H NMR of (a) **1** (2.0×10^{-4} M) (b) **1** (2.0×10^{-4} M) + HCl with pH 6 (c) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7 (400 MHz, D_2O , 298 K).

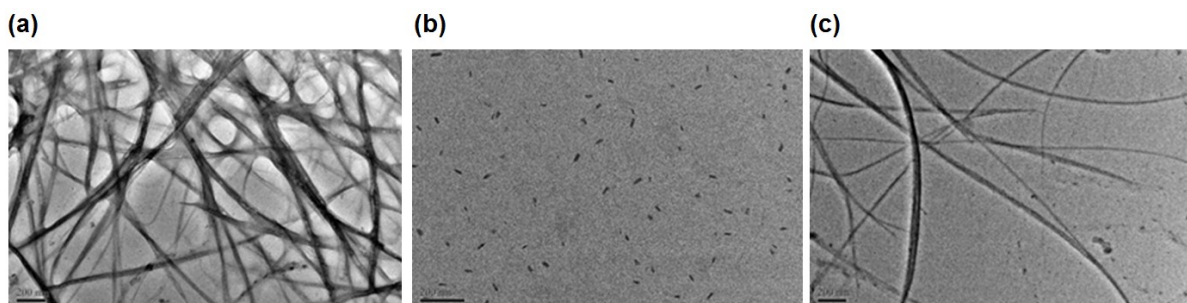


Figure S8. TEM of (a) **1**, (a) **1** (2.0×10^{-4} M) + HCl with pH 3, (b) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7.

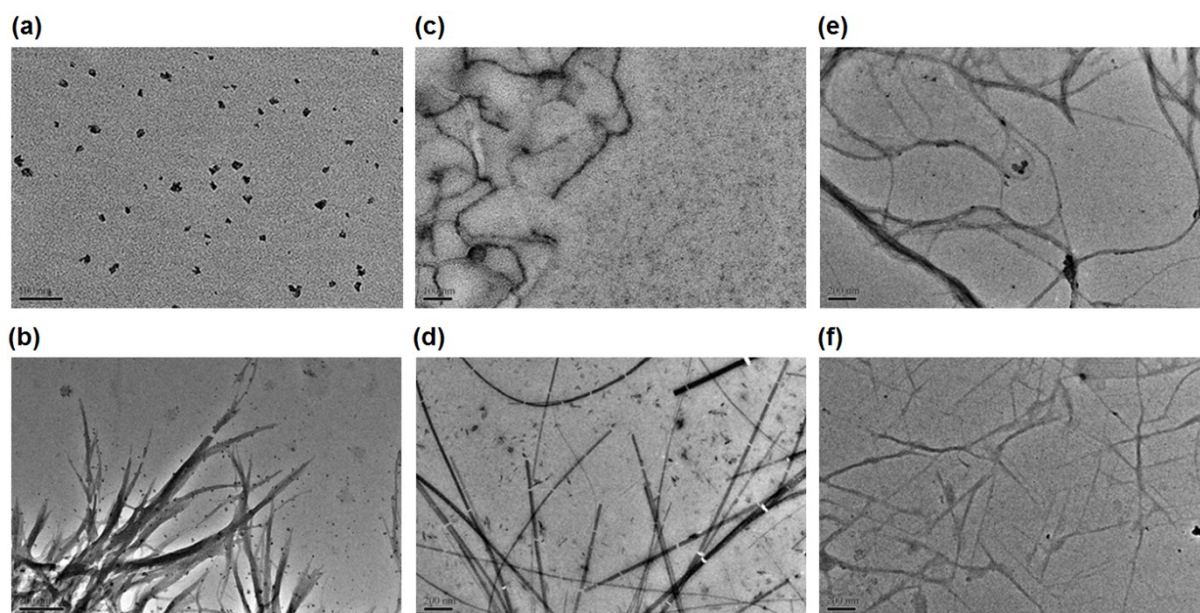


Figure S9. TEM of (a) **1** (2.0×10^{-4} M) + HCl with pH 4, (b) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7, (c) **1** (2.0×10^{-4} M) + HCl with pH 5, (d) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7, (e) **1** (2.0×10^{-4} M) + HCl with pH 6, (f) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7.

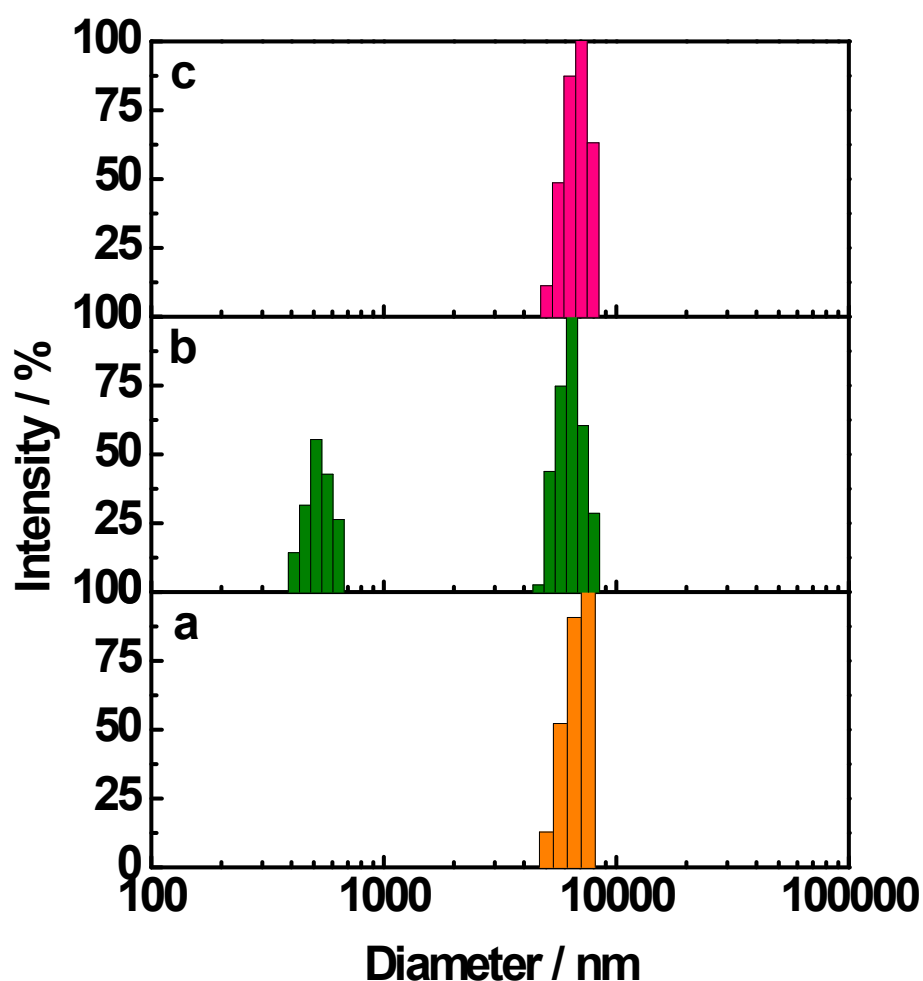


Figure S10. DLS of (a) **1** (2.0×10^{-4} M), (b) **1** (2.0×10^{-4} M) + HCl with pH 3, (c) **1** (2.0×10^{-4} M) + HCl + NaOH with pH 7.

S4 Protonation of DPD@1 by light

In a typical experiment, an aqueous solution of polypseudorotaxane **1** with MEH was exposed to visible light for 5 min to realize proton transfer from photoacid to DPD@1. Keeping the solution in the dark resulted in the rethreading of polypseudorotaxane **1**. pH measurements, UV-vis spectra, TEM images and DLS studies were recorded to monitor the threading-dethreading process.

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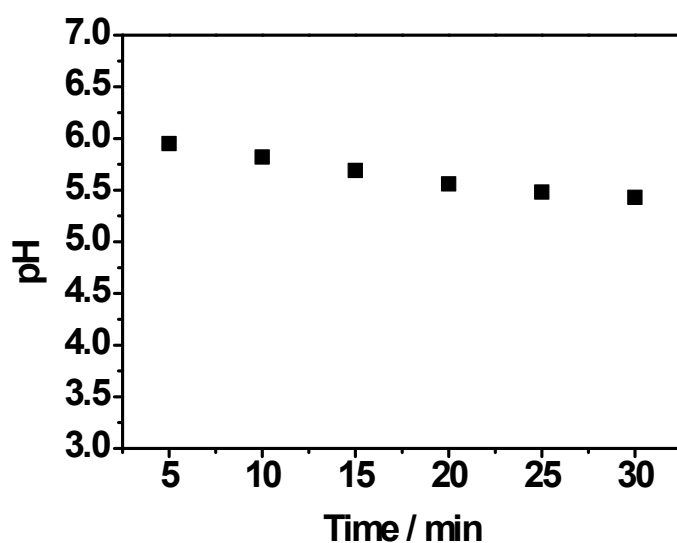


Figure S11. pH of **1** (2.0×10^{-4} M) + MEH (6.0×10^{-4} M) with different irradiation time.

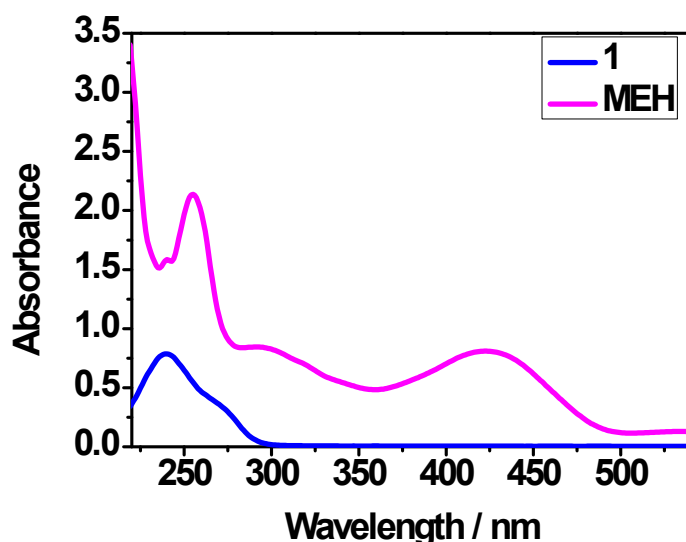


Figure S12. UV-vis absorption of MEH (2×10^{-4} M) and **1** (2×10^{-6} M), respectively

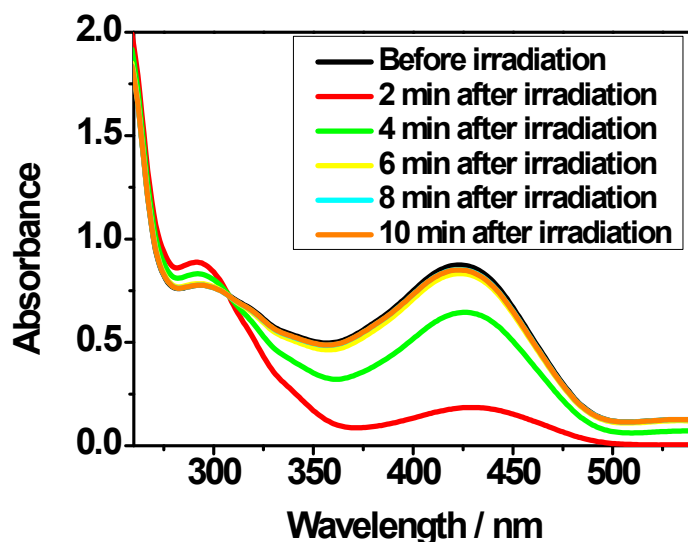


Figure S13. UV-vis absorption of MEH (2×10^{-4} M) before and after irradiation. (irradiation at 420 nm for 5 min, and spectra were collected every 2 min after irradiation)

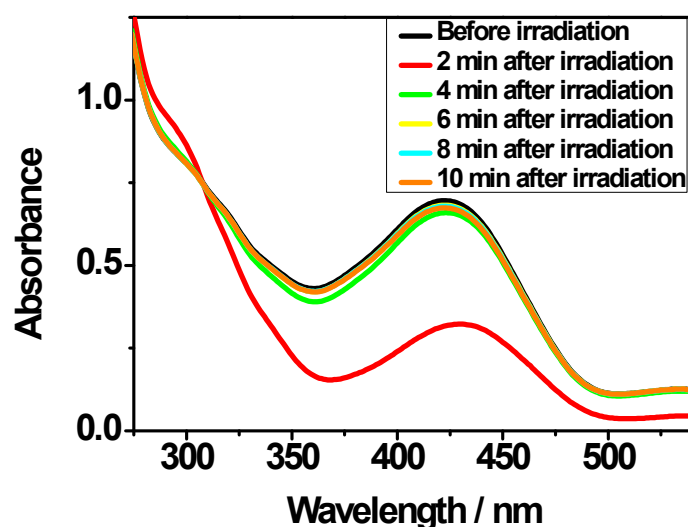


Figure S14. UV-vis absorption of MEH (2×10^{-4} M) + **1** (2×10^{-6} M) before and after irradiation. (irradiation at 420 nm for 5 min, and spectra were collected every 2 min after irradiation)

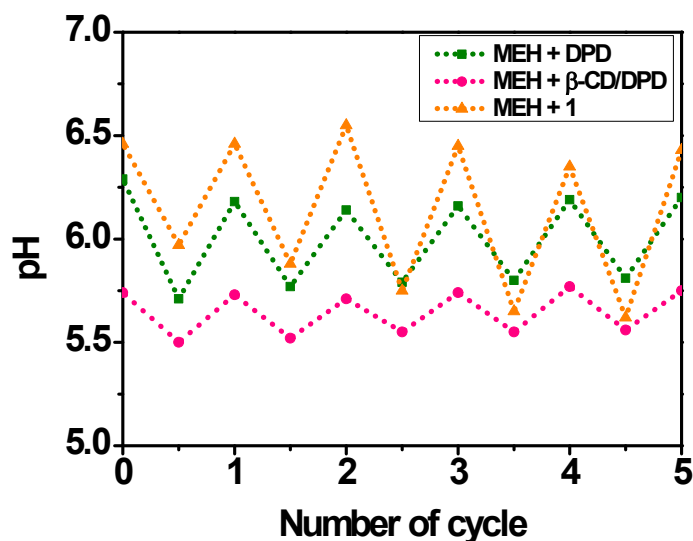


Figure S15. Cycles of pH change under irradiation and in the dark. For each cycle, the solution of DPD (2.95×10^{-3} M) + MEH (6.0×10^{-4} M) (red), and β -CD/DPD (2.95×10^{-3} M) + MEH (6.0×10^{-4} M) (blue) and **1** (2.0×10^{-4} M) + MEH (6.0×10^{-4} M) (red) were irradiated with 420 nm for 5 min and kept in the dark for 15 min.

References:

1. Pires, M. M.; Przybyla, D. E.; Rubert Pérez, C. M.; Chmielewski, J. *J. Am. Chem. Soc.* 2011, **133**, 14699–14703.