# **Electronic Supplementary Information**

## Five-component trigonal nanoprism with six dynamic corners

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### **Table of Contents**

#### 1. Synthesis

	1.1 General information	S02
	1.2 Synthesis of ligands	S03
	1.3 Synthesis of metal complexes	S12
2.	NMR spectra	S18
3.	DOSY spectrum	S40
4.	Energy minimised structure (MM <sup>+</sup> force field)	S40
5.	ESI-MS spectra	S41
6.	UV-vis spectra	S47
7.	References	S49

#### 1. Synthesis

#### 1.1 General information

All reagents were obtained from commercial suppliers and used without further purification. Technical grade solvents were distilled prior to use. Tetrahydrofuran (THF) was predried over basic alumina and then distilled over potassium. Dimethylformamide (DMF) and triethylamine ( $Et_3N$ ) were distilled on calcium hydride. Diethyl ether ( $Et_2O$ ) was predried over calcium hydride and then distilled over sodium.

Melting points of compounds were measured using a Büchi SMP-11 instrument. <sup>1</sup>H, <sup>13</sup>C, and <sup>1</sup>H<sup>-1</sup>H COSY NMR spectra were recorded on a Bruker Avance 400 at 298 K. DOSY NMR was recorded on Varian VNMR-S600 MHz. Chemical shifts refer to the residual protiated fraction of the NMR solvent (CHCl<sub>3</sub>:  $\delta_{\rm H}$  = 7.26 ppm,  $\delta_{\rm C}$  = 77.0 ppm; CHDCl<sub>2</sub>:  $\delta_{\rm H}$  = 5.32 ppm,  $\delta_{\rm C}$  = 53.8 ppm). Abbreviations were used in <sup>1</sup>H NMR assignments to describe splitting patterns (s: singlet, d: doublet, t: triplet, dd: doublet of doublet, ddd: doublet of doublet of doublet, bs: broad singlet, td: triplet of doublets, quint: quintet, m: multiplet), the value of coupling constant(s) is reported in hertz (Hz) and the number of protons are implied. Numbering of the carbon atoms is not in accordance with IUPAC nomenclature. UV-vis spectra were measured on Cary Win 50. Electrospray ionisation-mass spectra (ESI-MS) were recorded on a Thermo-Quest LCQ Deca instrument. Infrared spectra were recorded using a Perkin Elmer Spectrum-Two FT-IR spectrometer. Column chromatography was performed on silica gel 60 (60-230 mesh) or on neutral alumina (0.05-0.15 mm, Brockmann Activity 1). Thin layer chromatography (TLC) was performed using Merck silica gel (60 F254) or on neutral Al<sub>2</sub>O<sub>3</sub> (150 F254) sheets. Size exclusion chromatography was performed on BioRads Biobeads-SX3 using toluene or THF as an eluents. Compounds 5A,<sup>1</sup>7,<sup>2</sup>10,<sup>3</sup>11,<sup>4</sup>13,<sup>5</sup>20,<sup>6</sup>21<sup>7</sup> and 23<sup>8</sup> were synthesised according to known protocols and in some cases modified procedures.



S2

#### 1.2 Synthesis of ligands

 $((3,5-Dibromophenyl)ethynyl)trimethylsilane (10)^{3}$ 



A mixture of 1,3,5-tribromobenzene (**9**, 5.00 g, 15.9 mmol), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (600 mg, 860 µmol), and CuI (300 mg, 1.58 mmol) was placed in a tube. The tube was evacuated and filled with N<sub>2</sub> (3×), then freshly distilled Et<sub>3</sub>N (60 mL) and benzene (20 mL) were added. Trimethyl-silylacetylene (3.60 mL, 25.5 mmol) was added, and the reaction mixture was stirred at 55 °C for 36 h under inert atmosphere (TLC control). The mixture was cooled to 25 °C, and solvents were removed under reduced pressure. The residue was extracted in DCM (75 mL) and washed with deionised water (75 mL) and brine (30 mL). The organic layer was dried over anhydrous MgSO<sub>4</sub> and evaporated. The column chromatographic purification of crude product on silica gel ( $\phi = 5$  cm, *l* = 20 cm) using *n*-hexane ( $R_f = 0.40$ , SiO<sub>2</sub>, *n*-hexane) afforded compound **10** as colourless oil (4.61 g, 12.2 mmol, 87%). IR (neat): 3073, 3056, 2958, 2897, 2108, 1761, 1761, 1736, 1577, 1554, 1540, 1418, 1359, 1262, 1102, 1078, 900, 843, 759, 669, 575 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.61 (t, <sup>4</sup>*J* = 1.8 Hz, 1H, a-H), 7.53 (d, <sup>4</sup>*J* = 1.8 Hz, 2H, b-H), 0.24 (s, 9H, c-H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  134.0, 133.9, 125.0, 121.6, 102.4, 96.6, –0.2 ppm.

Compound 12



A solution of 4'-(4-ethynylphenyl)-2,2':6',2"-terpyridine<sup>4</sup> (**11**, 350 mg, 1.05 mmol) and ((3,5-dibromophenyl)ethynyl)trimethylsilane<sup>3</sup> (**10**, 1.56 g, 4.70 mmol) in a mixture of distilled anhydrous DMF (20 mL) and Et<sub>3</sub>N (40 mL) was subjected to freeze-pump-thaw cycles ( $3\times$ ). After addition of Pd(PPh<sub>3</sub>)<sub>4</sub> (121 mg, 105 µmol), the reaction mixture was heated to 70 °C for 18

h (TLC). The reaction mixture was allowed to cool to 25 °C and the solvents were evaporated in vacuo. The crude product was extracted in DCM (50 mL) then washed with deionised water (60 mL  $\times$  2) and a saturated brine solution (30 mL). The organic layer was dried over anhydrous MgSO<sub>4</sub> and evaporated. The crude product was purified by column chromatography on silica gel ( $\phi$  = 3.5 cm, l = 10 cm) using DCM as an eluent furnishing compound **12** as off white solid (599 mg, 1.02 mmol, 98%). Mp: 207 °C. IR (KBr): 3055, 3012, 2957, 2923, 2856, 2223, 2152, 1603, 1584, 1566, 1546, 1512, 1424, 1410, 1264, 1247, 1216, 1109, 1088, 1075, 1038, 989, 966, 895, 862, 840, 790 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  8.76 (s, 2H, e-H), 8.73 (ddd, <sup>3</sup>J = 4.7 Hz, <sup>4</sup>J = 1.8 Hz,  ${}^{5}J$  = 1.0 Hz, 2H, a-H), 8.67 (ddd,  ${}^{3}J$  = 7.6 Hz,  ${}^{4}J$  = 1.2 Hz,  ${}^{5}J$  = 1.0 Hz, 2H, d-H), 7.92 (d,  ${}^{3}J = 8.4$  Hz, 2H, f-H), 7.88 (td,  ${}^{3}J = 7.6$  Hz,  ${}^{4}J = 1.8$  Hz, 2H, c-H), 7.65 (d,  ${}^{3}J = 8.4$  Hz, 2H, g-H), 7.65 (t,  ${}^{4}J = 1.4$  Hz, 1H, i-H), 7.60 (t,  ${}^{4}J = 1.4$  Hz, 1H, h-H), 7.58 (t,  ${}^{4}J = 1.4$  Hz, 1H, i-H), 7.34 (ddd,  ${}^{3}J = 7.6$  Hz,  ${}^{3}J = 4.7$  Hz,  ${}^{4}J = 1.2$  Hz, 2H, b-H), 0.26 (s, 9H, k-H) ppm.  ${}^{13}C$  NMR (CDCl<sub>3</sub>, 100 MHz): δ 156.1, 149.2 (2C), 138.7, 136.9, 134.4, 134.1, 133.6, 132.3, 132.2, 127.3, 125.2, 125.1, 123.9, 123.2, 121.8, 121.4, 118.7, 102.4, 96.8, 90.9, 88.5, -0.2 ppm. ESI-MS: Calcd for  $[C_{34}H_{26}BrN_{3}Si \cdot H]^{+} = [12 \cdot H]^{+}$ , m/z = 586.1; Found:  $[12 \cdot H]^{+}$ , m/z (%) = 586.3 (100). Anal. calcd for C<sub>34</sub>H<sub>26</sub>BrN<sub>3</sub>Si: C, 69.86; H, 4.48; N, 7.19. Found, C, 69.84; H, 4.61; N, 6.79.

Compound 14



A solution of compounds **12** (150 mg, 257  $\mu$ mol) and zinc(II)-5-(4-ethynylphenyl)-10,15,20trimesitylporphyrin<sup>5</sup> (**13**, 425 mg, 513  $\mu$ mol) in a mixture of freshly distilled anhydrous DMF (30 mL) and anhydrous Et<sub>3</sub>N (20 mL) was degassed using freeze-pump-thaw cycles (3 ×). After addition of Pd(PPh<sub>3</sub>)<sub>4</sub> (30.5 mg, 25.9  $\mu$ mol), the resulting mixture was heated to 80 °C for 24 h

(TLC). Solvents were removed under reduced pressure. The crude product was extracted in DCM (20 mL) then washed successively with deionised water (30 mL  $\times$  2) and saturated brine solution (30 mL  $\times$  2). The organic layer was removed and the aqueous layer was re-extracted with DCM (30 mL). The combined organic layers were dried over anhydrous MgSO<sub>4</sub> and evaporated in *vacuo*. The column chromatographic purification of crude product on silica gel ( $\phi$ = 3.5 cm, l = 15 cm) using DCM provided compound 14 as purple solid (255 mg, 191  $\mu$ mol, 75%). Mp > 250 °C. IR (KBr): 2955, 2915, 2854, 2204, 2152, 1806, 1605, 1582, 1567, 1523, 1510, 1477, 1465, 1440, 1385, 1334, 1249, 1203, 1062, 1038, 998, 975, 878, 850, 831, 811, 794, 739, 723, 681, 660, 619 cm<sup>-1</sup>. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz):  $\delta$  8.89 (d, <sup>3</sup>J = 4.6 Hz, 2H, β-H), 8.81 (s, 2H, e-H), 8.76 (d,  ${}^{3}J = 4.6$  Hz, 2H,  $\beta$ -H), 8.74 (ddd,  ${}^{3}J = 4.9$  Hz,  ${}^{4}J = 1.8$  Hz,  ${}^{5}J = 1.0$  Hz, 2H, a-H), 8.71–8.73 (m, 6H, d-,  $\beta$ -H), 8.26 (d,  ${}^{3}J$  = 8.2 Hz, 2H, 1-H), 7.96 (d,  ${}^{3}J$  = 8.2 Hz, 2H, m-H), 7.94 (d,  ${}^{3}J = 8.4$  Hz, 2H, f-H), 7.92 (td,  ${}^{3}J = 7.6$  Hz,  ${}^{4}J = 1.8$  Hz, 2H, c-H), 7.84 (t,  ${}^{4}J = 1.5$ Hz, 1H, j-H), 7.77 (d,  ${}^{3}J$  = 8.4 Hz, 2H, g-H), 7.75 (t,  ${}^{4}J$  = 1.5 Hz, 1H, h-H), 7.69 (t,  ${}^{4}J$  = 1.5 Hz, 1H, i-H), 7.39 (ddd,  ${}^{3}J = 7.6$  Hz,  ${}^{3}J = 4.9$  Hz,  ${}^{4}J = 1.0$  Hz, 2H, b-H), 7.30 (s, 4H, o-H), 7.29 (s, 2H, r-H), 2.62 (2 s, 9H, p,s-H), 1.84 (s, 6H, q-H), 1.83 (s, 12H, n-H), 0.31 (s, 9H, k-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz): δ 156.5, 156.3, 150.3 (2C), 150.1, 150.0, 149.5 (2C), 144.0, 139.6, 139.5 (2C), 139.3 (2C), 139.1, 137.9, 137.2, 134.9, 134.8, 134.7, 132.7, 132.2, 135.5, 131.4, 131.0, 130.3, 128.0 (2C), 127.7, 124.6, 124.5, 124.4, 124.3 (2C), 123.8, 122.1, 121.5, 119.5, 119.3, 119.2, 118.9, 103.4, 96.4, 91.1, 90.6, 89.4, 88.8, 21.8, 21.7 (2C), 21.5, -0.1 ppm. ESI-MS: Calcd for  $[C_{89}H_{71}N_7SiZn \cdot H]^+ = [14 \cdot H]^+$ , m/z = 1332.5; Found:  $[14 \cdot H]^+$ , m/z (%) = 1332.5 (100). Anal. calcd for C<sub>89</sub>H<sub>71</sub>N<sub>7</sub>SiZn: C, 80.25; H, 5.37; N, 7.36. Found, C, 80.26; H, 5.30; N, 7.11.

#### Compound 15



To a solution of compound 14 (160 mg, 120 µmol) in THF-MeOH (v/v, 30 mL, 2:1) was added a solution of KOH (34.1 mg, 601 µmol in 5 mL of H<sub>2</sub>O). The reaction mixture was stirred at 25 °C for 3 h. The solvent was evaporated in vacuo. The residue was extracted in DCM (20 mL) and washed with deionised water (20 mL  $\times$  3). The organic layer was dried over anhydrous MgSO<sub>4</sub>. The evaporation of solvent yielded compound 15 as a purple solid (132 mg, 105 µmol, 87%). Mp > 200 °C. IR (KBr): 2956, 2914, 2853, 2731, 2222, 1689, 1605, 1581, 1566, 1478, 1466, 1441, 1410, 1383, 1333, 1298, 1263, 1226, 1062, 1038, 997, 878, 850, 810, 794, 739, 723, 680, 660, 621 cm<sup>-1</sup>. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz): δ 8.89 (d,  ${}^{3}J$  = 4.6 Hz, 2H, β-H), 8.81 (s, 2H, e-H), 8.76 (d,  ${}^{3}J = 4.6$  Hz, 2H,  $\beta$ -H), 8.73 (ddd,  ${}^{3}J = 4.8$  Hz,  ${}^{4}J = 1.6$  Hz,  ${}^{5}J = 0.8$  Hz, 2H, a-H), 8.69–8.72 (m, 6H, d-,  $\beta$ -H), 8.25 (d,  ${}^{3}J$  = 8.2 Hz, 2H, 1-H), 7.97 (d,  ${}^{3}J$  = 8.2 Hz, 2H, m-H), 7.95 (d,  ${}^{3}J$  = 8.4 Hz, 2H, f-H), 7.92 (td,  ${}^{3}J$  = 7.6 Hz,  ${}^{4}J$  = 1.6 Hz, 2H, c-H), 7.88 (t,  ${}^{4}J$  = 1.5 Hz, 1H, j-H), 7.79 (d,  ${}^{4}J = 1.5$  Hz, 1H, h-H), 7.77 (d,  ${}^{3}J = 8.4$  Hz, 2H, g-H), 7.73 (t,  ${}^{4}J = 1.5$  Hz, 1H, i-H), 7.39 (ddd,  ${}^{3}J$ = 7.6 Hz,  ${}^{4}J$  = 4.8 Hz,  ${}^{4}J$  = 1.0 Hz, 2H, b-H), 7.30 (s, 4H, o-H), 7.29 (s, 2H, r-H), 3.28, (s, 1H, k-H), 2.62 (2 s, 9H, p,s-H), 1.84 (s, 6H, q-H), 1.83 (s, 12H, n-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz): δ 156.5, 156.3, 150.3 (2C), 150.1, 150.0, 149.6, 149.5 (2C), 144.0, 139.5 (2C), 139.3 (2C), 139.2, 137.9, 137.2, 135.1 (2C), 135.0, 134.9, 132.7, 132.2, 131.5, 131.4, 131.0, 130.3, 128.0 (2C), 127.7, 124.7, 124.5, 124.3, 123.8, 123.5, 122.0, 121.5, 119.5, 119.3, 119.2, 118.9, 91.3, 90.7, 89.3, 88.6, 82.2, 78.9, 21.8, 21.7 (2C), 21.5 ppm. ESI-MS: Calcd for  $[C_{86}H_{63}N_7Zn \cdot H]^+ = [15 \cdot H]^+, m/z = 1258.5;$  Found:  $[15 \cdot H]^+, m/z$  (%) = 1258.6 (100). Anal. calcd. for C<sub>86</sub>H<sub>63</sub>N<sub>7</sub>Zn: C, 81.99; H, 5.04; N, 7.78. Found, C, 81.95; H, 4.93; N, 7.48.

4-((4-Iodo-2,3,5,6-tetramethylphenyl)ethynyl)-2,6-dimethylpyridine (17)



A solution of 4-ethynyl-2,6-dimethylpyridine<sup>1</sup> (5A, 150 mg, 1.14 mmol) and 1,4-diiodo-2,3,5,6tetramethylbenzene<sup>9</sup> (16, 2.20 g, 5.70 mmol) in a mixture of freshly distilled anhydrous DMF (20 mL), anhydrous C<sub>6</sub>H<sub>6</sub> (20 mL) and Et<sub>3</sub>N (20 mL) was degassed using freeze-pump-thaw cycles  $(3 \times)$ . Catalyst Pd(PPh<sub>3</sub>)<sub>4</sub> (66 mg, 57 µmol) was added and the reaction mixture was heated to 60 °C for 20 h (TLC). The reaction mixture was concentrated in vacuo. The residue was dissolved in DCM (30 mL), then successively washed with deionised water (30 mL  $\times$  2) and saturated brine solution (50 mL). The organic layer was dried over anhydrous MgSO<sub>4</sub> and the solvent was evaporated. The crude product was purified by column chromatography on silica gel ( $\phi = 2.5$  cm, l = 25 cm) using 10% EtOAc in DCM. Compound 17 was afforded as a light gray solid (210 mg, 539 µmol, 47%). Mp: 213 °C. IR (KBr): 3049, 2990, 2922, 2861, 2735, 2401, 2232, 1699, 1594, 1543, 1524, 1446, 1411, 1343, 1225, 1213, 1155, 1013, 868, 854, 720, 605, 549 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.05 (s, 2H, b-H), 2.54 (s, 6H, c-H), 2.49 (s, 12H, a,d-H) ppm. <sup>13</sup>C NMR  $(CDCl_3:CD_2Cl_2 = 6:4, 100 \text{ MHz}): \delta 157.7, 137.6, 136.2, 131.8, 122.5, 121.5, 113.1, 95.3, 91.3, 122.5, 121.5, 113.1, 122.5, 121.5, 113.1, 122.5, 121.5, 113.1, 122.5, 121.5, 113.1, 122.5, 121.5, 113.1, 122.5, 121.$ 27.4, 24.1, 20.0 ppm. ESI-MS: Calcd for  $[C_{19}H_{20}IN \bullet H]^+ = [17 \bullet H]^+$ , m/z = 390.1; Found: m/z (%)  $[17 \cdot H]^+ = 390.1 (100)$ . Anal. calcd for C<sub>19</sub>H<sub>20</sub>IN: C, 58.62; H, 5.18; N, 3.60. Found: C, 58.55, H, 5.11, N, 3.52.

Compound 6



A solution of compounds 15 (100 mg, 79.3 µmol) and 4-((4-iodo-2,3,5,6-tetramethylphenyl)ethynyl)-2,6-dimethylpyridine (17, 124 mg, 319 µmol) in a mixture of freshly distilled anhydrous DMF (25 mL) and anhydrous Et<sub>3</sub>N (15 mL) was degassed using freeze-pump-thaw cycles (3  $\times$ ). Catalyst Pd(PPh<sub>3</sub>)<sub>4</sub> (9.50 mg, 7.93 µmol) wad added under inert atmosphere, then the reaction mixture was heated to 80 °C for 24 h (TLC). The solvent was evaporated under reduced pressure. The residue was dissolved in DCM (25 mL), then subsequently washed with deionised water (30 mL  $\times$  2) and a saturated brine solution (15 mL  $\times$  2). The organic layer was dried over anhydrous  $MgSO_4$  and evaporated in *vacuo*. The purple residue was purified by column chromatography on silica gel ( $\phi = 3$  cm, l = 15 cm) using 20% EtOAc in DCM to yield compound 6 as a purple solid. The crude product was further subjected to size-exclusion chromatography ( $\phi = 2$  cm, l = 55 cm) on Bio-beads-SX3 using toluene-DCM (v/v, 9:1) as an eluent. All fractions were analysed by <sup>1</sup>H NMR. Pure fractions were combined, and the solvent was evaporated in *vacuo* to furnish the title compound **6** as a purple solid. (39.0 mg, 25.6 µmol,32%). Mp >250 °C. IR (KBr): 2942, 2916, 2872, 2214, 1601, 1581, 1566, 1550, 1521, 1477, 1466, 1440, 1410, 1383, 1332, 1202, 1015, 994, 871, 851, 830, 793, 738, 722 cm<sup>-1</sup>. <sup>1</sup>H NMR  $(CD_2Cl_2, 400 \text{ MHz})$ :  $\delta 8.87 \text{ (d, }^3J = 4.6 \text{ Hz}, 2\text{H}, \beta_{\text{por}}\text{-H}), 8.81 \text{ (s, 2H, e-H)}, 8.72-8.75 \text{ (m, 4H, a-, a-)}$  $\beta_{por}$ -H), 8.69 (dt,  ${}^{3}J$  = 7.5 Hz,  ${}^{4}J$  = 1.2 Hz, 2H, d-H), 8.68 (s, 4H,  $\beta_{por}$ -H), 8.24 (d,  ${}^{3}J$  = 8.2 Hz,

2H, p-H), 7.98 (d,  ${}^{3}J = 8.2$  Hz, 2H, o-H), 7.97 (d,  ${}^{3}J = 8.5$  Hz, 2H, f-H), 7.92 (td,  ${}^{3}J = 7.5$  Hz,  ${}^{4}J = 1.2$  Hz, 2H, c-H), 7.87 (2 t,  ${}^{4}J = 1.2$  Hz, 2H, j-, h-H), 7.81 (t,  ${}^{4}J = 1.2$  Hz, 1H, i-H), 7.79 (d,  ${}^{3}J = 8.5$  Hz, 2H, g-H), 7.39 (ddd,  ${}^{3}J = 7.5$  Hz,  ${}^{3}J = 4.8$  Hz,  ${}^{4}J = 1.2$  Hz, 2H, b-H), 7.29 (s, 4H, r-H), 7.28 (s, 2H, u-H), 7.11 (s, 2H, m-H), 2.62 (9H, s-,v-H), 2.59 (s, 6H, 1-H), 2.54 (s, 6H, k-H), 2.50 (s, 6H, n-H), 1.83 (s, 6H, t-H), 1.82 (s, 12H, q-H) ppm.  ${}^{13}$ C NMR (CDCl<sub>3</sub>:CD<sub>2</sub>Cl<sub>2</sub> = 4:6, 100 MHz):  $\delta$  157.7, 155.9, 155.8, 149.7, 149.6, 149.5, 149.3, 149.0 (2C), 143.5, 138.9 (3C), 138.8 (2C), 138.5, 137.3, 136.7, 136.0, 135.9, 134.3, 133.9, 133.8, 133.7, 132.1, 131.7, 131.5, 130.9, 130.7, 130.4, 129.7, 127.4 (2C), 127.2, 124.5, 124.1, 123.9, 123.8, 123.3 (2C), 122.6, 121.5, 121.4, 121.0, 118.7, 118.6, 118.4 (2C), 96.6, 95.8, 91.6, 90.7, 90.1, 89.6, 89.0, 88.4, 24.0, 21.4, 21.3 (2C), 21.1, 18.2, 18.1 ppm. ESI-MS: Calcd for [C<sub>105</sub>H<sub>82</sub>N<sub>8</sub>Zn•H]<sup>+</sup> = [**6**·H]<sup>+</sup>, *m/z* = 1521.7; Found: [**6**·H]<sup>+</sup>, *m/z* (%) = 1521.6 (100). Calcd for [C<sub>105</sub>H<sub>82</sub>N<sub>8</sub>Zn•H]<sup>+</sup> = [**6**·H]<sup>+</sup>, *m/z* = 761.8; Found: [**6**·2H]<sup>2+</sup>, *m/z* (%) = 761.7 (32). Anal. calcd. for C<sub>105</sub>H<sub>82</sub>N<sub>8</sub>Zn•1/2CH<sub>2</sub>Cl<sub>2</sub>: Found, C, 81.03; H, 5.35; N, 7.17; Zn, 4.18. Found: C, 81.30; H, 5.45; N, 7.30.

Compound  $7^2$ 



1,4-Diiodo-2,3,5,6-tetramethylbenzene (**16**, 500 mg, 1.30 mmol), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (45.0 mg, 389 µmol) and CuI (75.0 mg, 394 µmol) were placed in a Schlenk tube. The tube was evacuated and filled with N<sub>2</sub> (3×). Then a pre-degassed solution of 4-ethynylpyridine hydrochloride (**18**, 900 mg, 6.50 mmol) in a mixture of anhydrous C<sub>6</sub>H<sub>6</sub> (40 mL) and anhydrous Et<sub>2</sub>NH (40 mL) was added under N<sub>2</sub>. The reaction mixture was heated to 80 °C for 18 h (TLC). After evaporation of solvents, the greenish brown residue was extracted in DCM (30 mL). The organic layer was subsequently washed with deionised water (25 mL × 2) and saturated brine solution (50 mL). The combined organic layers were removed and dried over anhydrous MgSO<sub>4</sub>. After evaporation of the solvent, the crude product was subjected to chromatographic purification on silica gel ( $\phi$  = 3.5 cm, *l* = 15 cm) using 10% EtOAc in DCM yielding **7** as yellow solid (228 mg, 550 µmol, 89%). Mp: 239 °C. IR (KBr): 3069, 3025, 2924, 2863, 2214, 2190, 1954, 1688, 1586, 1532, 1491, 1417, 1407, 1377, 1320, 1263, 1218, 1204, 1078, 1012, 987, 856, 819, 713, 665, 648, 557,

540 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  8.62 (d, <sup>3</sup>*J* = 4.4 Hz, 4H, a-H), 7.40 (d, <sup>3</sup>*J* = 4.4 Hz, 4H, b-H), 2.50 (s, 12H, c-H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz, 298 K):  $\delta$  149.8, 136.3, 131.7, 125.3, 123.1, 95.6, 92.9, 18.4 ppm.

1,4-Diiodo-2,5-bis(octyloxy)benzene  $(20)^6$ 



ICl (7.96 g, 49.0 mmol) was added dropwise to 100 mL of precooled MeOH (< 10 °C). Thereafter, 1,4-bis(octyloxy)benzene<sup>6</sup> (**19**, 4.10 g, 12.3 mmol) was added below 10 °C and the reaction mixture was heated to reflux for 3 h (TLC). The resultant mixture was allowed to cool to 25 °C and filtered. The solid was washed with cold methanol (25 mL × 4) and dried under vacuum to furnish **20** as a white powder (6.25 g, 10.7 mmol, 87%). Mp: 57 °C. IR (KBr): 2939, 2918, 2862, 2847, 1672, 1486, 1464, 1446, 1388, 1351, 1263, 1234, 1143, 1068, 1050, 1018, 1012, 997, 909, 900, 846, 834, 786, 746, 721 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.17 (s, 2H, a-H), 3.92 (t, <sup>3</sup>*J* = 6.4 Hz, 4H, b-H), 1.79 (tt, <sup>3</sup>*J* = 6.4 Hz, <sup>3</sup>*J* = 6.8 Hz, 4H, c-H), 1.49 (quint, <sup>3</sup>*J* = 6.8 Hz, 4H, d-H), 1.32 (m, 16H, e-, f-, g-, h-H), 0.89 (t, <sup>3</sup>*J* = 6.9 Hz, 6H, i-H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  152.9, 122.8, 86.3, 70.4, 31.8, 29.2 (2C), 29.1, 26.0, 22.7, 14.1 ppm.

3-((4-Iodo-2,5-bis(octyloxy)phenyl)ethynyl)-2,9-dimesityl-1,10-phenanthroline (22)



3-Ethynyl-2,9-dimesityl-1,10-phenanthroline<sup>7</sup> (**21**, 300 mg, 681  $\mu$ mol) and 1,4-diiodo-2,5bis(octyloxy)benzene<sup>6</sup> (**20**, 798 mg, 1.36 mmol) were placed in a Schlenk tube. Freshly distilled anhydrous DMF (30 mL) and Et<sub>3</sub>N (10 mL) were added and the mixture was subjected to freezepump-thaw cycles (3×). After addition of Pd(PPh<sub>3</sub>)<sub>4</sub> (79.0 mg, 68.1  $\mu$ mol), the reaction mixture

was heated to 70 °C for 16 h. The solvents were removed under reduced pressure and the residue was extracted in DCM. The crude product was purified by column chromatography on silica gel using 7% EtOAc in *n*-hexane affording compound 22 as light yellow solid (475 mg, 528 µmol, 77%). Mp: 140 °C. IR (KBr): 2924, 2853, 2202, 1614, 1579, 1536, 1485, 1461, 1410, 1377, 1355, 1266, 1211, 1106, 1065, 1027, 964, 925, 887, 860, 847, 775, 719, 635, 615 cm<sup>-1</sup>. <sup>1</sup>H NMR  $(CD_2Cl_2, 400 \text{ MHz})$ :  $\delta 8.47$  (s, 1H, 7-H), 8.32 (d,  ${}^{3}J = 8.2 \text{ Hz}$ , 1H, 4-H), 7.91 (d,  ${}^{3}J = 8.8 \text{ Hz}$ , 1H, 5-H), 7.86 (d,  ${}^{3}J$  = 8.8 Hz, 1H, 6-H), 7.56 (d,  ${}^{3}J$  = 8.2 Hz, 1H, 3-H), 7.28 (s, 1H, g-H), 6.97 (s, 2H, e-H), 6.95 (s, 2H, b-H), 6.33 (s, 1H, h-H), 3.92 (t,  ${}^{3}J = 6.8$  Hz, 2H, i-H), 3.85 (t,  ${}^{3}J = 6.4$  Hz, 2H, i'-H), 2.34 (s, 3H, f-H), 2.34 (s, 3H, a-H), 2.05 (s, 6H, d-H), 2.04 (s, 6H, c-H), 1.78 (2 x quint,  ${}^{3}J = 6.8$  Hz,  ${}^{3}J = 6.4$  Hz, 4H, j-, j'-H), 1.51 (2 x quint,  ${}^{3}J = 6.8$  Hz,  ${}^{3}J = 6.4$  Hz, 4H, k-, k'-H), 1.29–1.42 (m, 16H, l-, l'-, m-, m'-, n-, n'-, o-, o'-H), 0.90 (t,  ${}^{3}J = 7.0$  Hz, 3H, p-H), 0.86 (t,  ${}^{3}J$ = 7.2 Hz, 3H, p'-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz):  $\delta$  161.8, 160.9, 154.2, 152.1, 146.4, 145.4, 138.8, 138.4, 137.8, 137.6 (2C), 136.4 (2C), 136.1, 128.6 (2C), 128.2, 127.9, 127.3 (2C), 126.0, 125.1, 124.3, 120.1, 116.5, 113.3, 91.6, 88.3, 70.3, 70.2, 32.2 (2C), 29.8 (2C), 29.7, 29.6 (2C), 29.5, 26.4, 26.5, 23.1, 21.4, 21.2, 20.4 (2C), 20.1, 14.3, 14.2 ppm. ESI-MS: Calcd for  $[C_{54}H_{63}IN_2O_2 \bullet H]^+ = [22 \bullet H]^+, m/z = 899.4;$  Found:  $[22 \bullet H]^+, m/z$  (%) = 899.4 (100). Anal. calcd for C<sub>54</sub>H<sub>63</sub>IN<sub>2</sub>O<sub>2</sub>: C, 72.14; H, 7.06; N, 3.12. Found, C, 72.38; H, 6.97; N, 3.07.

2-(4-Bromo-2,3,5,6-tetramethylphenyl)-3-((4-((2,9-dimesityl-1,10-phenanthrolin-3 yl)ethynyl)-2,5-bis(octyloxy)phenyl)ethynyl)-9-(2,4,6-trimethoxyphenyl)-1,10-phenanthroline (**8**)



2-(4-Bromo-2,3,5,6-tetramethylphenyl)-3-ethynyl-9-(2,4,6-trimethoxyphenyl)-[1,10]phenanthroline<sup>8</sup> (**23**, 101 mg, 112  $\mu$ mol) and 3-((4-iodo-2,5-bis(octyloxy)phenyl)ethynyl)-2,9dimesityl-1,10-phenanthroline (**22**, 50.0 mg, 85.9  $\mu$ mol) in a mixture of anhydrous Et<sub>3</sub>N (30 mL)

and anhydrous DMF (30 mL) were degassed using freeze-pump-thaw cycles (3  $\times$ ). Pd(PPh<sub>3</sub>)<sub>4</sub> (10.0 mg, 8.60 µmol) was added and the reaction mixture was stirred at 55 °C for 19 h (TLC). The solvent was evaporated in *vacuo* and purified by column chromatography on silica gel ( $\phi$  = 2.5 cm, l = 15 cm) using 20% EtOAc in DCM. The pure fractions were combined and evaporated to furnish the title compound 8 as yellow solid (131 mg, 96.8 µmol, 86%). Mp: 249 °C. IR (KBr): 2923, 2854, 2208, 1609, 1584, 1500, 1457, 1411, 1383, 1355, 1334, 1274, 1213, 1205, 1154, 1127, 1104, 1061, 1026, 989, 948, 885, 865, 809, 825, 636, 610 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  8.46 (2 s, 2H, 7,7'-H), 8.32 (d,  ${}^{3}J$  = 8.2 Hz, 1H, 4'-H), 8.27 (d,  ${}^{3}J$  = 8.2 Hz, 1H, 4-H), 7.92 (d,  ${}^{3}J = 8.8$  Hz, 1H, 6'-H), 7.90 (d,  ${}^{3}J = 8.8$  Hz, 1H, 6-H), 7.86 (d,  ${}^{3}J = 8.8$  Hz, 1H, 5'-H), 7.85 (d,  ${}^{3}J = 8.8$  Hz, 1H, 5-H), 7.58 (d,  ${}^{3}J = 8.2$  Hz, 1H, 3'-H), 7.56 (d,  ${}^{3}J = 8.1$  Hz, 1H, 3-H), 6.97 (s, 2H, q'-H), 6.95 (s, 2H, t'-H), 6.37 (s, 1H, g'-H), 6.25 (s, 2H, b'-H), 6.19 (s, 1H, f'-H), 3.87 (s, 3H, a'-H), 3.85 (t,  ${}^{3}J = 6.8$  Hz, 2H, h"-H), 3.82 (t,  ${}^{3}J = 6.8$  Hz, 2H, h'-H), 3.69 (s, 6H, c'-H), 2.47 (s, 6H, d'-H), 2.35 (s, 3H, p'-H), 2.33 (s, 3H, u'-H), 2.04 (2 s, 12H, r',s'-H), 2.00 (s, 6H, e'-H), 1.76–1.87 (2 x quint,  ${}^{3}J = 6.8$  Hz,  ${}^{3}J = 6.8$  Hz, 4H, i'-, i"-H), 1.51–1.61 (m, 4H, i'-, i"-H), 1.26–1.48 (m, 16H, k'-, l'-, m'-, n'-, k"-, l"-, m"-, n"-H), 0.88 (t,  ${}^{3}J = 6.8$  Hz, 3H, o"-H), 0.87 (t,  ${}^{3}J$ = 6.8 Hz, 3H, o'-H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  162.5, 162.0 161.9, 160.9, 159.3, 156.1, 153.5, 153.3, 146.4, 146.3, 145.4, 139.9, 138.8, 138.7, 138.4, 137.8, 137.6, 136.4 (2C), 136.1, 135.8, 134.1, 133.9, 129.0, 128.6, 128.3, 128.2, 127.9, 127.6, 127.3 (2C), 127.0, 126.0, 125.9, 125.1, 120.2, 120.1, 117.6, 117.0, 114.1, 114.0, 113.0, 92.6, 92.5, 92.2, 91.9, 91.2, 69.7, 56.3, 55.8, 32.3 (2C), 29.8 (4C), 29.6, 26.6, 26.5, 23.1, 21.4, 21.2, 21.1 (2C), 20.4 (2C), 20.1 (2C), 18.6, 14.3 ppm. ESI-MS: Calcd for  $[C_{87}H_{91}BrN_4O_5 \bullet H]^+ = [8 \bullet H]^+$ , m/z = 1353.6; Found  $[\mathbf{8}\cdot\mathbf{H}]^+$ , m/z (%) = 1353.6 (28). Calcd for  $[C_{87}H_{91}BrN_4O_5 \cdot 2H]^+ = [\mathbf{8}\cdot 2H]^{2+}$ , m/z = 677.8; Found:  $[8 \cdot 2H]^{2+}$ , m/z (%) = 677.8 (100). Anal. calcd for C<sub>87</sub>H<sub>91</sub>BrN<sub>4</sub>O<sub>5</sub>: C, 77.25; H, 6.78; N, 4.14. Found: C, 77.33; H, 6.76; N, 4.04.

#### 1.3 Synthesis of metal complexes

Synthesis of complex  $C1 = [Zn(1)(4)](OTf)_2$ 



Zn(OTf)<sub>2</sub> (910 µg, 2.50 µmol) was added to solution of 2-(4-iodo-2,3,5,6-tetramethylphenyl)-9-(2,4,6-trimethoxyphenyl)-1,10-phenanthroline (1, 1.51 mg, 2.50 µmol) in a mixture of CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN (9:1) that was sonicated at 40 °C for 5 min. Thereafter, 2,2':6',2"-terpyridine (4, 583 µg, 2.50 µmol) was added. The <sup>1</sup>H NMR was measured without further purification. Yield: Quantitative. Mp: 163 °C. IR (KBr): 3077, 2942, 1605, 1583, 1454, 1417, 1341, 1275, 1224, 1206, 1156, 1128, 1029, 871, 779, 637, 572, 516 cm<sup>-1</sup>. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN = 9:1, 400) MHz):  $\delta$  9.00 (d,  ${}^{3}J$  = 8.3 Hz, 1H, 7-H), 8.96 (d,  ${}^{3}J$  = 8.4 Hz, 1H, 4-H), 8.63 (t,  ${}^{3}J$  = 7.4 Hz, 1H, F-H), 8.46–8.54 (m, 6H, D-, E-, 5-, 6-H), 8.22 (td,  ${}^{3}J$  = 7.6 Hz,  ${}^{4}J$  = 1.5 Hz, 2H, C-H), 8.10 (d,  ${}^{3}J$ = 8.3 Hz, 1H, 8-H), 7.88 (d,  ${}^{3}J$  = 8.4 Hz, 1H, 3-H), 7.67 (ddd,  ${}^{3}J$  = 5.2 Hz,  ${}^{4}J$  = 1.5 Hz,  ${}^{5}J$  = 0.9 Hz, 2H, A-H), 7.55 (ddd,  ${}^{3}J = 7.6$  Hz,  ${}^{3}J = 5.2$  Hz,  ${}^{4}J = 0.9$  Hz, 2H, B-H), 5.61 (s, 2H, B'-H), 3.52 (s, 3H, A'-H), 2.95 (s, 6H, C'-H), 1.97 (s, 6H, D'-H), 0.83 (s, 6H, E'-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz): δ 164.0, 161.1, 158.4, 157.5, 149.0, 147.3, 146.8, 145.6, 142.9, 142.7, 141.9, 141.3, 141.0, 138.6, 137.9, 131.5, 130.9, 129.3 (2C), 128.5, 128.3, 128.1, 127.5, 123.3, 123.2, 114.5, 107.7, 90.7, 56.0, 55.6, 27.2, 18.7 ppm. ESI-MS: Calcd for  $[Zn(1)(4)-OTf]^+ m/z =$ 1050.1 and  $[Zn(1)(4)-2OTf]^+ m/z = 450.6$ . Found:  $[Zn(1)(4)-OTf]^+ m/z$  (%) = 1049.8 (100) and  $[Zn(1)(4)-2OTf]^{2+}$  m/z (%) = 450.9 (22). Anal. calcd for C<sub>48</sub>H<sub>40</sub>F<sub>6</sub>IN<sub>5</sub>O<sub>9</sub>S<sub>2</sub>Zn•1/10 CH<sub>2</sub>Cl<sub>2</sub>: C, 47.75; H, 3.35; N, 5.79, S, 5.30. Found: C, 47.46; H, 3.02; N, 5.57, S, 5.30.

Synthesis of complex  $C2 = [Cu(2)(5A)]B(C_6F_5)_4$ 



2,9-Dimesityl-1,10-phenanthroline (**2**, 538 μg, 1.29 μmol), 4-ethynyl-2,6-dimethylpyridine (**5A**, 169 μg, 1.29 μmol) and [Cu(CH<sub>3</sub>CN)<sub>4</sub>]B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub> (1.17 mg, 1.29 μmol) were placed in an NMR tube and dissolved in CD<sub>2</sub>Cl<sub>2</sub>. The resultant complex **C2** was analysed by <sup>1</sup>H NMR spectroscopy. Yield: Quantitative. Mp: 219 °C. IR (KBr): 2919, 2859, 2116, 1615, 1584, 1552, 1509, 1481, 1439, 1379, 1362, 1146, 1026, 869, 840, 557 cm<sup>-1</sup>. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>: CD<sub>3</sub>CN = 9:1, 400 MHz):  $\delta$  8.60 (d, <sup>3</sup>*J* = 8.3 Hz, 2H, 4'-, 7'-H), 8.08 (s, 2H, 5'-, 6'-H), 7.81 (d, <sup>3</sup>*J* = 8.3 Hz, 2H, 3'-, 8'-H), 7.01 (s, 2H, M-H), 6.95 (s, 4H, O'-, R'-H), 3.31 (s, 1H, L-H), 2.40 (s, 6H, S'-, P'-H), 2.32 (s, 6H, N-H), 1.94 (s, 12H, N'-, Q'-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz):  $\delta$  161.2, 156.7, 141.4, 139.9, 139.7, 136.8, 135.4, 131.9, 128.5, 128.3, 127.5, 127.2, 124.3, 116.9 (CH<sub>3</sub>CN), 84.1, 80.4, 25.9, 21.1, 20.5, 2.1 (<u>CH<sub>3</sub>CN</u>) ppm. ESI-MS: Calcd for [Cu(**2**)(**5A**)–B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>]<sup>+</sup> *m/z* = 610.2; Found: [Cu(**2**)(**5A**)–B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>]<sup>+</sup>, *m/z* (%) = 610.4 (100). Anal. calcd for C<sub>63</sub>H<sub>37</sub>BCuF<sub>20</sub>N<sub>3</sub>•2/5CH<sub>2</sub>Cl<sub>2</sub>: C, 59.89; H, 4.82; N, 5.32. Found: C, 60.15; H, 4.52; N, 5.41.

Synthesis of complex C3 = [(3)(5B)]



Zinc(II)-5,10,15,20-tetraphenylporphyrin (**3**, 873 µg, 1.29 µmol) and 4-iodopyridine (**5B**, 264 µg, 1.29 µmol) were placed in an NMR tube and dissolved in  $CD_2Cl_2$ . The resultant complex **C3** was directly analysed by <sup>1</sup>H NMR spectroscopy. Yield: Quantitative. Mp: 190 °C (decomposition). IR (KBr): 3065, 2998, 1593, 1575, 1483, 1476, 1406, 1336, 1215, 1173, 1065, 1000, 992, 797, 746, 717, 701, 659, 673 cm<sup>-1</sup>. <sup>1</sup>H NMR ( $CD_2Cl_2 : CD_3CN = 9:1$ , 400 MHz):  $\delta$  8.84 (s, 8H,  $\beta$ "-H),

8.16–8.18 (m, 8H, c'-H), 7.76–7.70 (m, 12H, a'-, b'-H), 6.64 (bs, 2H, β'-H), 4.52 (bs, 2H, α'-H) ppm. <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz): δ 150.4, 145.0, 143.6, 134.9, 132.5, 132.1, 127.7, 126.8, 121.2, 106.1 ppm. Anal. calcd for C<sub>49</sub>H<sub>32</sub>IN<sub>5</sub>Zn:C, 66.64; H, 3.65; N, 7.93. Found: C, 66.83; H, 3.49; N, 7.62.

Synthesis of tweezer complex T



In an oven-dried 50 mL flask, 2-(4-bromo-2,3,5,6-tetramethylphenyl)-3-((4-((2,9-dimesityl-1,10-phenanthrolin-3yl)ethynyl)-2,5-bis(octyloxy)phenyl)ethynyl)-9-(2,4,6-trimethoxyphenyl)-1,10 phenanthroline (**8**, 1.38 mg, 1.02  $\mu$ mol), ligand **6** (1.55 mg, 1.02  $\mu$ mol), Zn(OTf)<sub>2</sub> (370  $\mu$ g, 1.02  $\mu$ mol) and [Cu(CH<sub>3</sub>CN)<sub>4</sub>B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] (924  $\mu$ g, 1.02  $\mu$ mol) were refluxed in a mixture of CH<sub>3</sub>CN-CH<sub>2</sub>Cl<sub>2</sub> (3:1, v/v, 20 mL) for 3.5 h. The reaction was cooled to 25 °C and evaporated under reduced pressure. The resultant complex **T** was subjected to characterisation without further purification. Yield: 95%.

<sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN = 9:1, 400 MHz): δ 8.96 (s, 2H, 7-H), 8.94 (d,  ${}^{3}J$  = 8.5 Hz, 2H, 4-H), 8.78–8.85 (m, 8H, d-, β<sub>Por</sub>-H), 8.76 (s, 4H, e-H), 8.70 (d,  ${}^{3}J$  = 4.6 Hz, 4H, β<sub>Por</sub>-H), 8.66 (s, 2H, 7'-

H), 8.63 (s, 8H,  $\beta_{Por}$ -H), 8.59 (d,  ${}^{3}J$  = 8.4 Hz, 2H, 4'-H), 8.43 (d,  ${}^{3}J$  = 9.1 Hz, 4H, p-H), 8.35 (d,  ${}^{3}J = 9.1$  Hz, 4H, o-H), 8.32 (td,  ${}^{3}J = 7.6$  Hz,  ${}^{4}J = 1.5$  Hz, 4H, c-H), 8.27 (d,  ${}^{3}J = 8.8$  Hz, 4H, f-H), 8.20 (d,  ${}^{3}J = 8.8$  Hz, 2H, 5/6-H), 8.07 (d,  ${}^{3}J = 8.8$  Hz, 2H, 5'/6'-H), 8.07 (d,  ${}^{3}J = 8.5$  Hz, 2H, 3-H), 8.00 (d,  ${}^{3}J = 8.8$  Hz, 2H, 6<sup>1</sup>/5<sup>1</sup>-H), 7.95 (d,  ${}^{3}J = 8.8$  Hz, 4H, g-H), 7.93 (d,  ${}^{3}J = 8.8$  Hz, 2H, 6/5-H), 7.88 (2t,  ${}^{3}J$  = 1.5 Hz,  ${}^{3}J$  = 1.5 Hz, 4H, h-, j-H), 7.90 (t,  ${}^{3}J$  = 1.5 Hz, 2H, i-H), 7.81 (d,  ${}^{3}J$  = 8.4 Hz, 2H, 3'-H), 7.66 (ddd,  ${}^{3}J = 5.3$  Hz,  ${}^{4}J = 1.5$  Hz, 4H, a-H), 7.52 (ddd,  ${}^{3}J = 7.6$  Hz,  ${}^{3}J = 5.3$ Hz,  ${}^{4}J = 0.8$  Hz, 4H, b-H), 7.25 (2s, 12H, r-, u-H), 7.14 (s, 4H, m-H), 6.87 (s, 4H, q'-H), 6.80 (s, 4H, t'-H), 6.26 (s, 2H, g'-H), 6.24 (s, 2H, f'-H), 5.60 (s, 4H, b'-H), 3.64 (t,  ${}^{3}J = 6.3$  Hz, 4H, h"-H), 3.58 (t,  ${}^{3}J = 6.7$  Hz, 4H, h'-H), 3.46 (s, 6H, a'-H), 2.93 (s, 12H, c'-H), 2.59 (s, 6H, v-H), 2.58 (s, 12H, s-H), 2.52 (s, 12H, k-H), 2.42 (s, 12H, 1-H), 2.41 (s, 12H, n-H), 2.24 (s, 6H, p'-H), 2.21 (s, 6H, u'-H), 1.92 (s, 12H, e'-H), 1.91 (2s, 24H, r'-, s'-H), 1.77-1.81 (2s, 36H, q-, t-H), 1.63-1.66 (m, 4H, i'-H), 1.35–1.45 (m, 20H, i"-, j"-, k'-, j'-, k"-H), 1.20–1.27 (m, 24H, l'-, m'-, n'-, l"-, m"-, n"-H), 0.94 (s, 12H, d'-H), 0.78–0.85 (m, 12H, o'-, o"-H) ppm. ESI-MS: Calcd for  $[ZnCu(6)(8)]^{3+}$ m/z = 1001.4; Found:  $[ZnCu(6)(8)]^{3+} m/z$  (%) = 1001.6 (97). Calcd for  $[Zn_2Cu_2(6)_2(8)_2(OTf)]^{5+}$ m/z = 1231.2; Found;  $[Zn_2Cu_2(6)_2(8)_2(OTf)]^{5+}$  m/z (%) = 1231.3 (100). Calcd for  $[Zn_2Cu_2(6)_2(8)_2(OTf)_2]^{4+}$  m/z = 1576.3; Found:  $[Zn_2Cu_2(6)_2(8)_2(OTf)_2]^{4+}$  m/z (%) = 1576.2 (45).

Synthesis of complex prism P



In an oven-dried 50 mL flask, 2-(4-bromo-2,3,5,6-tetramethylphenyl)-3-((4-((2,9-dimesityl-1,10phenanthrolin-3yl)ethynyl)-2,5-bis(octyloxy)phenyl)ethynyl)-9-(2,4,6-trimethoxyphenyl)-1,10 phenanthroline (8, 863 µg, 0.639 µmol), tri-substituted ligand 6 (971 µg, 0.639 µmol), Zn(OTf)<sub>2</sub> (232 µg, 0.639 µmol), [Cu(CH<sub>3</sub>CN)<sub>4</sub>B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] (579 µg, 0.639 µmol) and bipyridine ligand 7 (107 µg, 0.319 µmol) were refluxed in a mixture of CH<sub>3</sub>CN-CH<sub>2</sub>Cl<sub>2</sub> (3:1, v/v, 20 mL). The reaction was cooled to 25 °C and evaporated under reduced pressure. The resultant complex P was subjected to characterisation without further purification. Yield: 95%. IR (KBr): 2922, 2854, 2201, 1641, 1604, 1512, 1463, 1375, 1336, 1275, 1257, 1223, 1205, 1158, 1084, 1030, 995, 851, 637 cm<sup>-1</sup>. <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN = 9:1, 400 MHz):  $\delta$  9.00 (s, 2H, 7-H), 8.98 (d, <sup>3</sup>J = 8.4 Hz, 2H, 4-H), 8.74–8.84 (m, 8H, d-,  $\beta_{Por}$ -H), 8.76 (s, 4H, e-H), 8.68 (d,  ${}^{3}J$  = 4.6 Hz, 4H,  $\beta_{Por}$ -H), 8.64 (s, 2H, 7'-H), 8.62 (s, 8H,  $\beta_{Por}$ -H), 8.52 (d,  ${}^{3}J$  = 8.2 Hz, 2H, 4'-H), 8.48 (d,  ${}^{3}J$  = 9.0 Hz, 4H, p-H), 8.38 (d,  ${}^{3}J = 9.0$  Hz, 4H, o-H), 8.32 (td,  ${}^{3}J = 7.6$  Hz,  ${}^{4}J = 1.6$  Hz, 4H, c-H), 8.23–8.28 (m, 6H, 5/6-, f-H), 8.09 (d,  ${}^{3}J = 8.4$  Hz, 2H, 3-H), 8.08 (d,  ${}^{3}J = 8.8$  Hz, 2H, 5'/6'-H), 8.00 (d,  ${}^{3}J = 8.8$  Hz, 2H, 6'/5'-H), 7.93–7.95 (m, 6H, 6/5-, g-H), 7.90 (t,  ${}^{3}J$  = 1.6 Hz, 2H, h-H), 7.89 (t,  ${}^{3}J$  = 1.6 Hz, 2H, j-H), 7.84 (t,  ${}^{3}J$  = 1.6 Hz, 2H, i-H), 7.80 (d,  ${}^{3}J$  = 8.2 Hz, 2H, 3'-H), 7.67 (ddd,  ${}^{3}J$  = 5.3 Hz,  ${}^{4}J$ = 1.6 Hz, 4H, a-H), 7.53 (ddd,  ${}^{3}J$  = 7.6 Hz,  ${}^{3}J$  = 5.3 Hz,  ${}^{4}J$  = 0.8 Hz, 4H, b-H), 7.27 (s, 8H, r-H), 7.27 (s, 4H, u-H), 7.11 (s, 4H, m-H), 6.90 (s, 4H, q'-H), 6.86 (s, 4H, t'-H), 6.31 (s, 2H, g'-H), 6.25 (s, 2H, f'-H), 6.14 (bs, 4H,  $\beta\beta$ -H), 5.61 (s, 4H, b'-H), 3.67 (t,  ${}^{3}J$  = 6.8 Hz, 4H, h"-H), 3.65 (bs, 4H,  $\alpha\alpha$ -H), 3.60 (t,  ${}^{3}J$  = 6.8 Hz, 4H, h'-H), 3.48 (s, 6H, a'-H), 2.94 (s, 12H, c'-H), 2.595 (s, 6H, v-H), 2.591 (s, 12H, s-H), 2.586 (s, 12H, k-H), 2.53 (s, 12H, 1-H), 2.47 (s, 12H, n-H), 2.29 (s, 6H, p'-H), 2.25 (s, 6H, u'-H), 2.05 (s, 12H,  $\gamma\gamma$ -H), 1.91 (s, 12H, e'-H), 1.89 (2s, 24H, r'-, s'-H), 1.77–1.81 (m, 44H, i'-, i"-, q-, t-H), 1.63–1.76 (m, 8H, j'-, j"-H), 1.37–1.46 (m, 8H, k'-, k"-H), 1.22-1.35 (m, 24H, l'-, m'-, n'-, l"-, m"-, n"-H), 0.96 (s, 12H, d'-H), 0.79-0.84 (m, 12H, o'-, o"-H) ppm. ESI-MS: Calcd for  $[ZnCu(6)(8)]^{3+}$ , m/z = 1000.7; Found:  $[ZnCu(6)(8)]^{3+}$ , m/z (%) = 1000.2 (75). Calcd for  $[Zn_2Cu_2(6)_2(8)_2(OTf)]^{5+}$ , m/z = 1231.8; Found: m/z (%) = 1231.2 (100). Calcd for  $\mathbf{P} = [Zn_2Cu_2(6)_2(7)(8)_2(OTf)]^{5+}$ : m/z = 1298.1; Found: m/z (%) = 1298.3 (87).<sup>10</sup> Calcd for  $[Zn_2Cu_2(6)_2(8)_2(OTf)_2]^{4+}$  m/z = 1575.8; Found: m/z (%) = 1575.5 (100). Calcd for  $[ZnCu(6)(7)(8)(OTf)]^{2+}$  m/z = 1743.1; Found: m/z (%) = 1741.9 (66).<sup>10</sup> Anal. calcd. for C<sub>460</sub>H<sub>366</sub>B<sub>2</sub>Br<sub>2</sub>Cu<sub>2</sub>F<sub>52</sub>N<sub>26</sub>O<sub>22</sub>S<sub>4</sub>Zn<sub>4</sub>•3CH<sub>2</sub>Cl<sub>2</sub>: C, 65.03; H, 4.38; N, 4.26, S, 1.50. Found: C, 64.80; H, 4.13; N, 4.15, S, 1.23.

## 2. NMR spectra



Figure S1. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, 400 MHz) of compound 10.



Figure S2. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, 100 MHz) of compound 10.





Figure S4. <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum (CDCl<sub>3</sub>, 400 MHz) of compound **12**.



200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 Figure S5.  $^{13}$ C NMR spectrum (CDCl<sub>3</sub>, 100 MHz) of compound 12.



Figure S6. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 14.



Figure S7. <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 14.



Figure S8. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of compound 14.



Figure S9.  $^{1}$ H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 15.



Figure S10. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of compound 15.



Figure S11.  $^{1}$ H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 17.



Figure S12. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>:CD<sub>2</sub>Cl<sub>2</sub>, 6:4, 100 MHz) of compound 17.



Figure S13. <sup>1</sup>H NMR spectrum ( $CD_2Cl_2$ , 400 MHz) of compound 6.



Figure S14. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>:CD<sub>2</sub>Cl<sub>2</sub>,4:6, 100 MHz) of compound 6.



Figure S15. <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 6.





Figure S17. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, 100 MHz) of compound 7.



Figure S19. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, 100 MHz) of compound 20.



Figure S20. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 22.



Figure S21.  $^{1}$ H- $^{1}$ H COSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of compound 22.



Figure S22. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of compound 22.



Figure S23. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 400 MHz) of compound 8.



Figure S24. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of compound 8.



 $[Zn(1)(4)](OTf)_2.$ 



Figure S26. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of complex  $C1 = [Zn(1)(4)](OTf)_2$ .



Figure S27. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of complex C2 =  $[Cu(2)(5A)]B(C_6F_5)_4$ .



Figure S28. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of complex  $C2 = [Cu(2)(5A)]B(C_6F_5)_4$ .



Figure S29. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of complex C3 = [(3)(5B)].



Figure S30. <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 100 MHz) of complex C3 = [(3)(5B)].



**Figure S31**. Comparison of partial <sup>1</sup>H NMR spectra ( $CD_2Cl_2:CD_3CN$ , 9:1, 400 MHz) of complex (a)  $C1 = [Zn(1)(4)](OTf)_2$ , (b)  $C2 = [Cu(2)(5A)]B(C_6F_5)_4$  and (c) an equimolar mixture of C1, 2, 5A and  $[Cu(CH_3CN)_4B(C_6F_5)_4]$  after equilibration.



(a)  $C2 = [Cu(2)(5A)]B(C_6F_5)_4$ , (b) C3 = [(3)(5B)], and (c) an equimolar mixture of C2, 3, 5B after equilibration.



9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 Figure S33. Comparison of partial <sup>1</sup>H NMR spectra (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of complex (a)  $C1 = [Zn(1)(4)](OTf)_2$ , (b) C3 = [(3)(5B)] and (c) an equimolar mixture of C1, 3, and 5B after equilibration.



**Figure S34**. Comparison of partial <sup>1</sup>H NMR spectra ( $CD_2Cl_2:CD_3CN$ , 9:1, 400 MHz) of complex (a) C3 = [(3)(5B)] and (b) an equimolar mixture of C3 and 5A.



Figure S35. Comparison of partial <sup>1</sup>H NMR spectra (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of complex (a) C1 =  $[Zn(1)(4)](OTf)_2$ , (b) C2 =  $[Cu(2)(5A)]B(C_6F_5)_4$ , (c) C3 = [(3)(5B)] and (d) an equimolar mixture of 1, 2, 3, 4, 5A, 5B,  $[Zn(OTf)_2]$  and  $[Cu(CH_3CN)_4B(C_6F_5)_4]$  after 3 h reflux in DCM-acetonitrile (1:3).



Figure S36. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of tweezer complex T.



Figure S37. <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of tweezer T.





Figure S38. <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of prism P.





**Figure S39**.  ${}^{1}\text{H}-{}^{1}\text{H}$  COSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>:CD<sub>3</sub>CN, 9:1, 400 MHz) of prism **P**. Expanded parts of the aromatic region are shown at the top.

## 3. DOSY NMR spectrum of nanoprism P



Figure S40. DOSY NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 600 MHz) of prism P.

## 4. Energy minimised structure (MM<sup>+</sup> force field)



Figure S41. Energy minimised structure of the supramolecular nanoprism P (counteranions, hydrogens and alkyl chains are omitted for clarity).



Figure S42. ESI-MS spectrum of compound 1 in DCM.



Figure S43. ESI-MS spectrum of compound 12 in DCM.



Figure S44. ESI-MS spectrum of compound 14 in DCM.



Figure S45. ESI-MS spectrum of compound 15 in DCM.



Figure S46. ESI-MS spectrum of compound 17 in DCM.



Figure S47. ESI-MS spectrum of compound 6 in DCM.



Figure S48. ESI-MS spectrum of compound 22 in DCM.



Figure S49. ESI-MS spectrum of compound 8 in DCM.



Figure S50. ESI-MS spectrum of compound  $C1 = [Zn(1)(4)](OTf)_2$  in DCM-CH<sub>3</sub>CN.



Figure S51. ESI-MS spectrum of compound  $C2 = [Cu(2)(5A)]B(C_6F_5)_4$  in DCM-CH<sub>3</sub>CN.



**Figure S52**. ESI-MS spectrum of equimolar mixture of 1, 2, 3, 4, 5A, 5B,  $[Zn(OTf)_2]$  and  $[Cu(CH_3CN)_4B(C_6F_5)_4]$  after 3 h reflux in DCM-acetonitrile (1:3).



Figure S53. ESI-MS spectrum of tweezer  $T = [Zn_2Cu_2(6)_2(8)_2](B(C_6F_5)_4)_2(OTf)_4$  in DCM: CH<sub>3</sub>CN = 9:1.

#### 6. UV-vis spectra



**Figure S54.** UV-vis titration of complex  $[Cu(2)]^+$  (1.20 × 10<sup>-5</sup> M) vs. ligand **5A** (8.38 × 10<sup>-5</sup> M) in CH<sub>2</sub>Cl<sub>2</sub> at 298 K. Wavelength region 220-350 nm was analysed using SPECFIT/32 global analysis system (Spectrum Software Associates, Marlborough, MA). Result: log  $K = 4.50 \pm 0.21$ .



Figure S55. Comparison of UV-vis spectra of ligand 6, C = C1+C2+C3, tweezer complex T and of nanoprism P in DCM (concentrations of ligand 6 and of complexes  $C \sim 10^{-6}$  M. Concentration of complex T and P,  $0.50 \times 10^{-6}$  M) at 298 K.



Figure S56. Comparison of UV-vis spectra of C3 = [(3)(5B)] and of C = C1+C2+C3 in DCM (both concentrations,  $c = 6 \times 10^{-4}$  M) at 298 K.



**Figure S57**. UV-vis titration of tweezer complex T ( $0.50 \times 10^{-4}$  M) vs. ligand 7 ( $1.25 \times 10^{-3}$  M) furnishing P (in CH<sub>2</sub>Cl<sub>2</sub> at 298 K). The full wavelength region of 500-800 nm was analyzed using SPECFIT/32 global analysis system (Spectrum Software Associates, Marlborough, MA). Result: log *K* = 9.29 ± 0.04.

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