

Two Fluorescein-based Chemosensors for the Fast Detection of 2, 4, 6-Trinitrophenol (TNP) in Water

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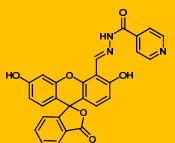
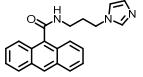
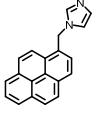
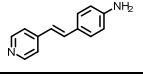
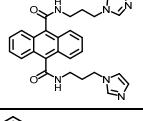
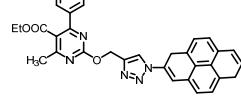
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1. Some reported work for the detection of TNP.

Table S1. A comparison about ours and some reported work for the detection of TNP.

Probe		Solvent	Ex (nm)	Reference
F1		Ethanol	500	<i>This work</i>
		HEPES buffer(pH=7.0)	400	
F2		Ethanol	517	<i>This work</i>
		HEPES buffer(pH=7.0)	400	
A		EtOH	362	(1)
B		Toluene Acetonitrile	342	(2)
C		THF	350	(3)
D		EtOH-H ₂ O (1/1, v/v, pH=7.0)	366	(4)
E		THF-HEPES (1/9, v/v, pH=6.99)	342	(5)

- (1) A. Pandith, A. Kumar, J.-Y. Lee, H.-S. Kim, *Tetrahedron Lett.*, 2015, **56**, 7094–7099.
- (2) R. Sodkhomkhum, M. Masik, S. Watchasit, C. Suksai, J. Boonmak, S. Youngme, N. Wanichachacheva, V. Ervithayasuporn, *Sens. Actuators, B*, 2017, **245**, 665–673.
- (3) J. Pan, F. Tang, A. Ding, L. Kong, L. Yang, X. Tao, Y. Tian, J. Yang, *RSC Adv.*, 2015, **5**, 191–195.
- (4) A. Pandith, A. Kumar and H.S. Kim, *RSC Adv.*, 2016, **6**, 68628–68637.
- (5) R. Chopra, P. Kaur, K. Singh, *Anal. Chim. Acta*, 2015, **864**, 55–63.

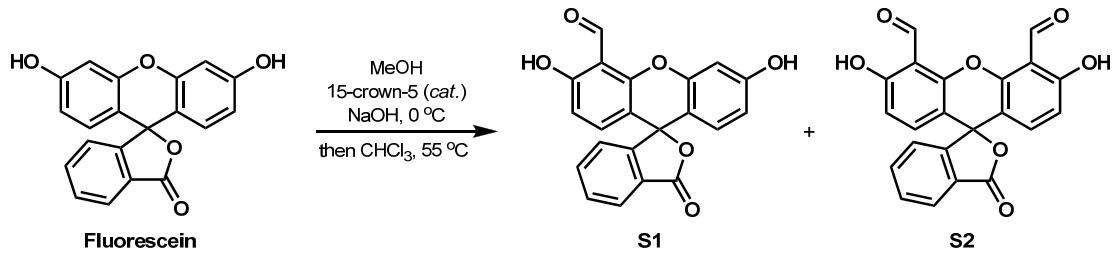
2. General information.

The commercially available chemicals were used without further purification. All of solvents used in experiments were analytical-reagent grade. ^1H and ^{13}C NMR spectra were measured on the Bruker ADVANCE III 400 MHz, JEOL JNM-ECS M Hz or Varian Mercury plus 300 MHz instruments using TMS as an internal standard. The single-crystal X-ray was performed on a Bruker smart CCD area detector diffractometer by the use of graphite-monochromated Mo K α radiation ($\lambda = 0.71072 \text{ \AA}$). Lorentz-polarization and absorption corrections were applied for the compounds. The structures were solved with direct methods and refined by full-matrix least-squares on F2 by the use of the SHELXL-97 program package. Mass spectra were determined on a Bruker esquire 6000 spectrometer. IR spectra were recorded on Nicolet FT-170SX instrument using KBr discs in the 400–4000 cm^{-1} region. UV-vis spectra were performed on an Agilent Carry 5000 UV-Vis-NIR spectrophotometer. Fluorescence spectra were recorded on a Hitachi F-7000 spectrophotometer equipped with quartz cuvettes of 1 cm path length. All pH measurements were made with a pH-10C digital pH meter.

Various stock solutions of NACs such as P (phenol), TNP (2,4,6-trinitrophenol), 3, 5-DMP (3,5-dimethoxyphenol), 4-NB (4-nitrobenzene), 1, 3-DNB (1,3-dinitrobenzene), 4-NT (4-nitrotoluene), 2,4-DNT (2,4-dinitrotoluene), 3, 5-DMT (3,5-methylcatechol), TNT (2,4,6-trinitrotoluene), PAP (*p*-aminophenol), 2-IP (2-iodophenol), 4-MP (4-methoxyphenol), 2-CP (2-chlorophenol) and 4-BP (4-bromophenol) in DMSO or ethanol were prepared. Stock solutions of **F1** and **F2** (10.0 mM) were also prepared in DMSO. Test solutions were prepared by placing 2.0 μL of the probe stock solution into a test tube, and then diluting to 2.0 mL, followed by the addition of an appropriate aliquot of NACs's stock solution. Fluorescence quantum yields were determined in solution, using rhodamin 6G ($\Phi = 0.76$ in H_2O and $\Phi = 0.88$ in ethanol) as a standard.⁶

(6) J. Olmsted, *J. Phys. Chem.*, 1979, **83**, 2581–2584.

3. Synthesis of fluorescein-derived aldehydes **S1** and **S2**.



This procedure was adapted from a known literature.⁷ To a 250 mL three-neck round-bottom flask were added fluorescein (5.0 g, 15.0 mmol) and MeOH (12 mL) at room temperature. The whole system was then cooled to 0 °C, followed by the addition of a mixture of NaOH aqueous solution (40 g, 50%) and 15-crown-5 (60 µL) within 5 min. The resulting mixture was stirred for 10 min, and then allowed to warm gradually in an oil bath. CHCl₃ (20 mL) was added dropwise while the reaction temperature was maintained at 55 °C. The reaction mixture was further stirred for 10 h at this temperature, and then cooled to room temperature. The mixture is acidified with H₂SO₄ (5 mL, 10 M), and the purple-black precipitate appeared. This solid was filtered and dried and purified by flash column chromatography (DCM/EtOAc = 85:15) on silica gel to afford the crude products, which could be recrystallized in acetone (10 mL) to give pure sample (**S1**, 708 mg, 13% yield) as a pale-yellow solid and (**S2**, 136 mg, 2.3% yield) as a white solid.

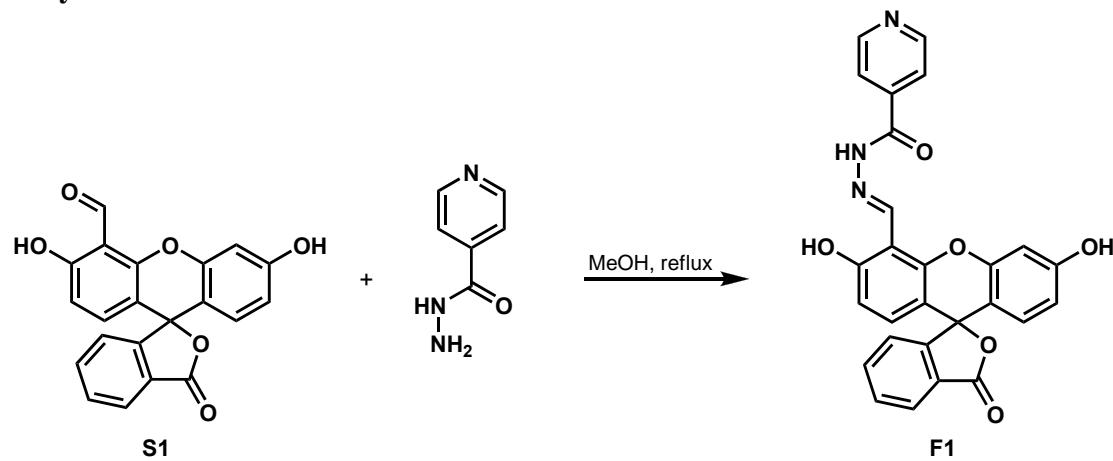
S1. Mp: 272–276 °C. ¹H NMR (DMSO-*d*₆, 300 MHz): δ = 11.89 (s, 1H), 10.63 (s, 1H), 10.29 (s, 1H), 8.02 (d, *J* = 7.5 Hz, 1H), 7.81 (d, *J* = 7.2 Hz, 1H), 7.73 (d, *J* = 7.2 Hz, 1H), 7.32 (d, *J* = 7.5 Hz, 1H), 6.95 (d, *J* = 9.0 Hz, 1H), 6.85 (s, 1H), 6.71 (d, *J* = 9.0 Hz, 1H), 6.61 (brs, 2H) ppm. ¹³C NMR (DMSO-*d*₆, 75 MHz) δ = 192.9, 168.5, 162.9, 159.6, 152.4, 152.2, 150.9, 136.5, 135.8, 130.3, 129.0, 125.9, 124.8, 124.0, 113.5, 113.4, 109.7, 109.2, 109.1, 102.7, 81.8 ppm. ESI-MS: *m/z* 361.1 [M + H]⁺.

S2. Mp: 311–313 °C. ¹H NMR (DMSO-*d*₆, 400 MHz): δ = 11.85 (s, 2H), 10.68 (s, 2H), 8.04 (d, *J* = 7.6 Hz, 1H), 7.82 (td, *J* = 1.2, 7.6 Hz, 1H), 7.73–7.77 (m, 1H), 7.37 (d, *J* = 7.6 Hz, 1H), 6.99 (d, *J* = 8.8 Hz, 2H), 6.79 (d, *J* = 9.2 Hz, 2H) ppm. ¹³C NMR (DMSO-*d*₆, 100 MHz) δ = 192.3 (2C), 168.5, 163.3 (2C), 152.0, 151.0 (2C), 136.2

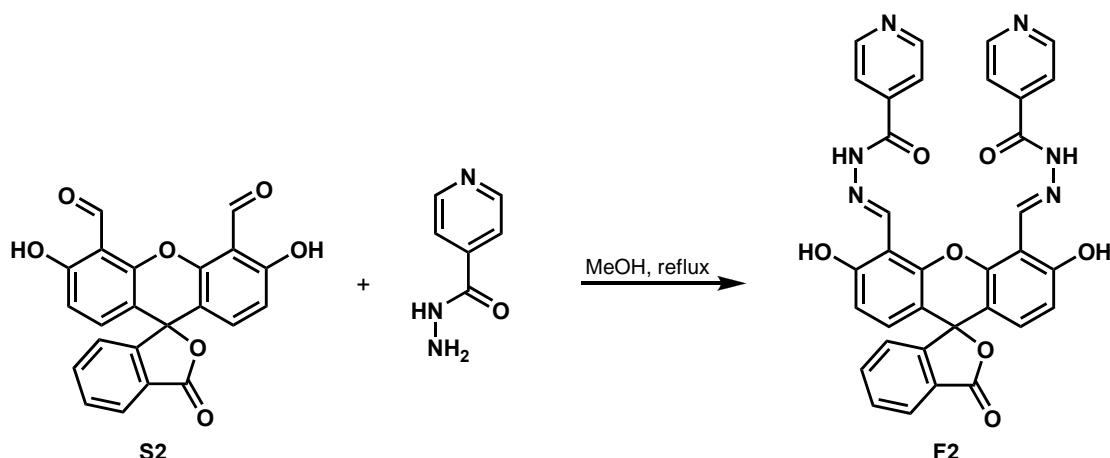
(2C), 136.1, 130.7, 125.8, 125.1, 124.1, 114.4 (2C), 109.7 (2C), 109.4 (2C), 80.8 ppm.
ESI-MS: m/z 389.0454 [M + H]⁺.

(7) W. Wang, O. Rusin, X. Xu, K. K. Kim, J. O. Escobedo, S. O. Fakayode, K. A. Fletcher, M. Lowry, C. M. Schowalter, C. M. Lawrence, F. R. Fronczek, I. M. Warner, R. M. Strongin, *J. Am. Chem. Soc.*, 2005, **127**, 15949–15958.

4. Synthesis of **F1** and **F2**.



The isonicotinohydrazide (206 mg, 1.5 mmol) was added to a solution of aldehyde **S1** (360 mg, 1.0 mmol) in methanol (15 mL). The reaction mixture was then stirred for 8 h at reflux. The resulting solid was filtered and washed with methanol, dried over vacuum to afford **F1**. (256 mg, 53.4%). Mp: 350–353°C. ¹H NMR (DMSO-*d*₆, 400 MHz): δ = 12.71 (brs, 1H), 12.60 (brs, 1H), 10.31 (brs, 1H), 9.26 (s, 1H), 8.86 (d, *J* = 5.6 Hz, 2H), 8.02 (d, *J* = 7.6 Hz, 1H), 7.91 (d, *J* = 5.6 Hz, 2H), 7.81 (t, *J* = 7.6 Hz, 1H), 7.74 (t, *J* = 7.6 Hz, 1H), 7.34 (d, *J* = 7.6 Hz, 1H), 6.75 (d, *J* = 8.8 Hz, 2H), 6.72 (d, *J* = 9.2 Hz, 1H), 6.62 (d, *J* = 0.8 Hz, 2H) ppm. ¹³C NMR (DMSO-*d*₆, 100 MHz): δ = 168.6, 161.3, 160.1, 159.7, 152.2, 151.2 (2C), 150.52, 150.50, 149.7, 145.5, 139.5, 139.4, 135.7, 131.1, 130.3, 129.2, 126.1, 124.8, 124.1, 121.5, 113.4, 109.6, 109.4, 105.6, 102.1, 82.4 ppm. ESI-MS: m/z 480.1362 [M + H]⁺. IR (KBr): ν_{max} = 3480, 3157, 3055, 2943, 2837, 1749, 1654, 1632, 1609, 1579, 1507, 1441, 1357, 1309, 1244, 1158, 1119, 1092, 1030, 839, 763, 703, 692 cm⁻¹.



The isonicotinohydrazide (412 mg, 3.0 mmol) was added to a solution of aldehyde **S2** (388 mg, 1.0 mmol) in methanol (15 mL). The reaction mixture was then stirred for 8 h at reflux. The resulting solid was filtered and washed with methanol, dried over vacuum to afford **F2** (428 mg, 68.3%). Mp: 296–298 °C. ^1H NMR (DMSO-*d*₆, 400 MHz): δ = 12.43 (brs, 2H), 12.05 (brs, 2H), 9.59 (s, 2H), 8.70 (d, *J* = 5.6 Hz, 4H), 8.04 (d, *J* = 7.6 Hz, 1H), 7.85 (d, *J* = 5.6 Hz, 4H), 7.84 (t, *J* = 7.6 Hz, 1H), 7.76 (t, *J* = 7.6 Hz, 1H), 7.41 (d, *J* = 7.6 Hz, 1H), 6.78 (d, *J* = 8.8 Hz, 2H), 6.74 (d, *J* = 8.8 Hz, 2H) ppm. ^{13}C NMR (DMSO-*d*₆, 100 MHz): δ = 168.7, 162.2 (2C), 160.2 (2C), 152.3, 150.3 (4C), 149.3 (2C), 147.0 (2C), 139.7 (2C), 135.9, 131.1 (2C), 130.4, 126.1, 124.9, 124.3, 121.7 (4C), 113.4 (2C), 109.5 (2C), 106.9 (2C), 82.4 ppm. ESI-MS: *m/z* 627.5 [M + H]⁺. IR (KBr): ν_{max} = 3435, 3251, 3047, 2849, 1744, 1665, 1638, 1611, 1542, 1466, 1434, 1353, 1290, 1229, 1162, 1117, 1065, 1014, 839, 823, 756, 692 cm⁻¹.

5. ^1H , ^{13}C NMR, IR and ESI-MS copies of compounds.

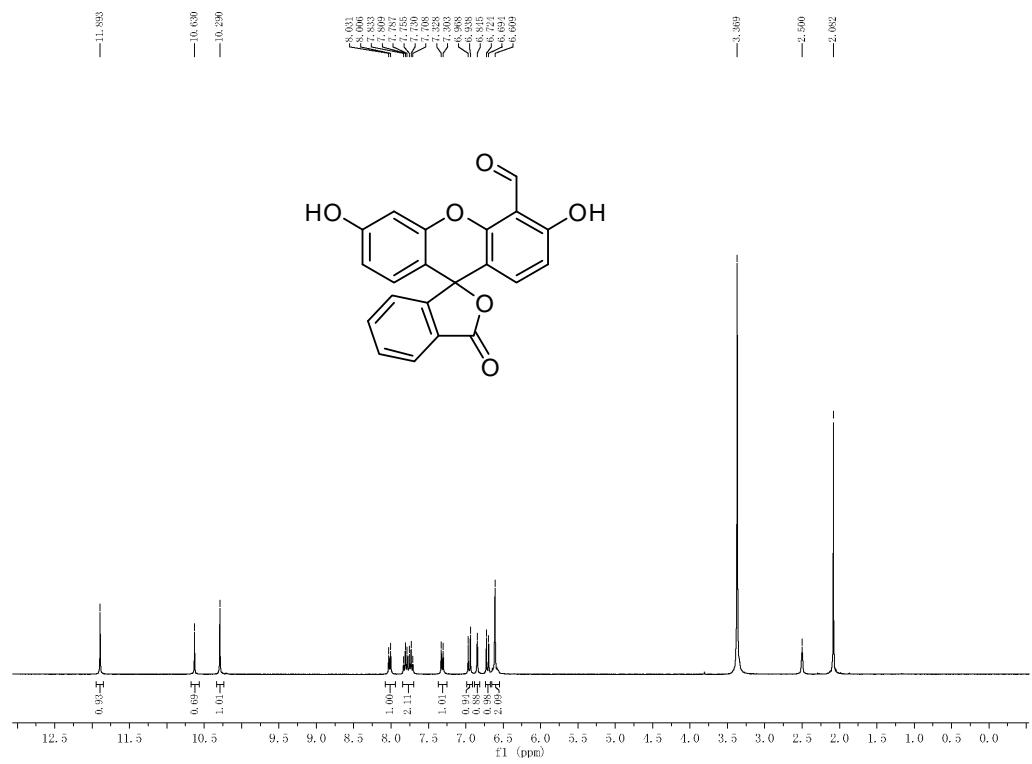


Figure S1. ^1H NMR spectrum (300 MHz, $\text{DMSO}-d_6$) of **S1**.

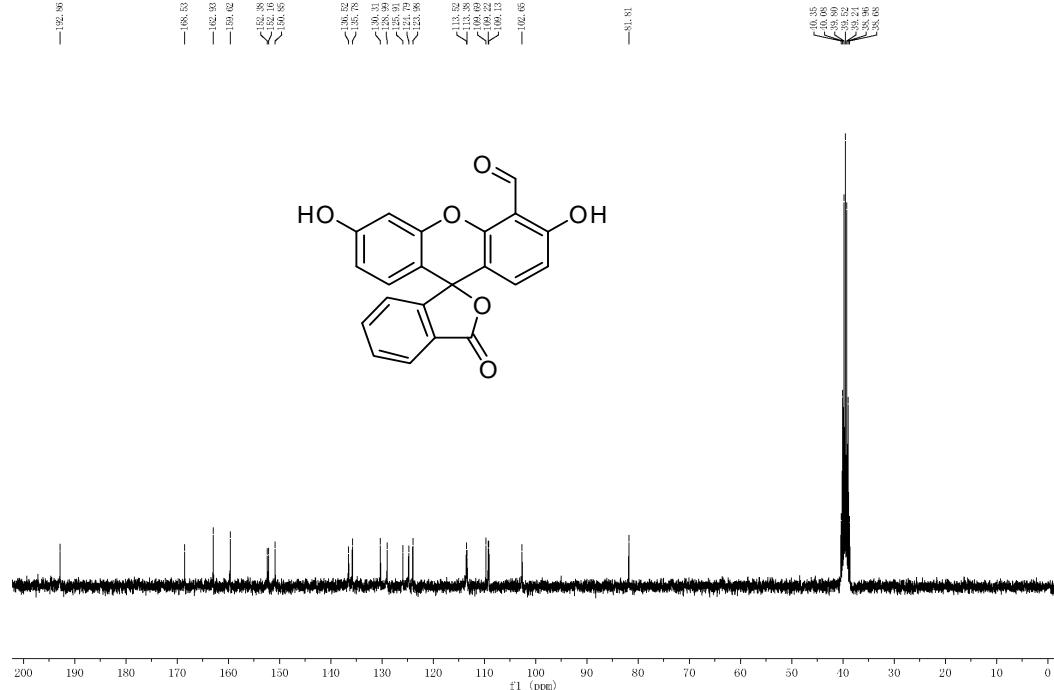


Figure S2. ^{13}C NMR spectrum (75 MHz, $\text{DMSO}-d_6$) of **S1**.

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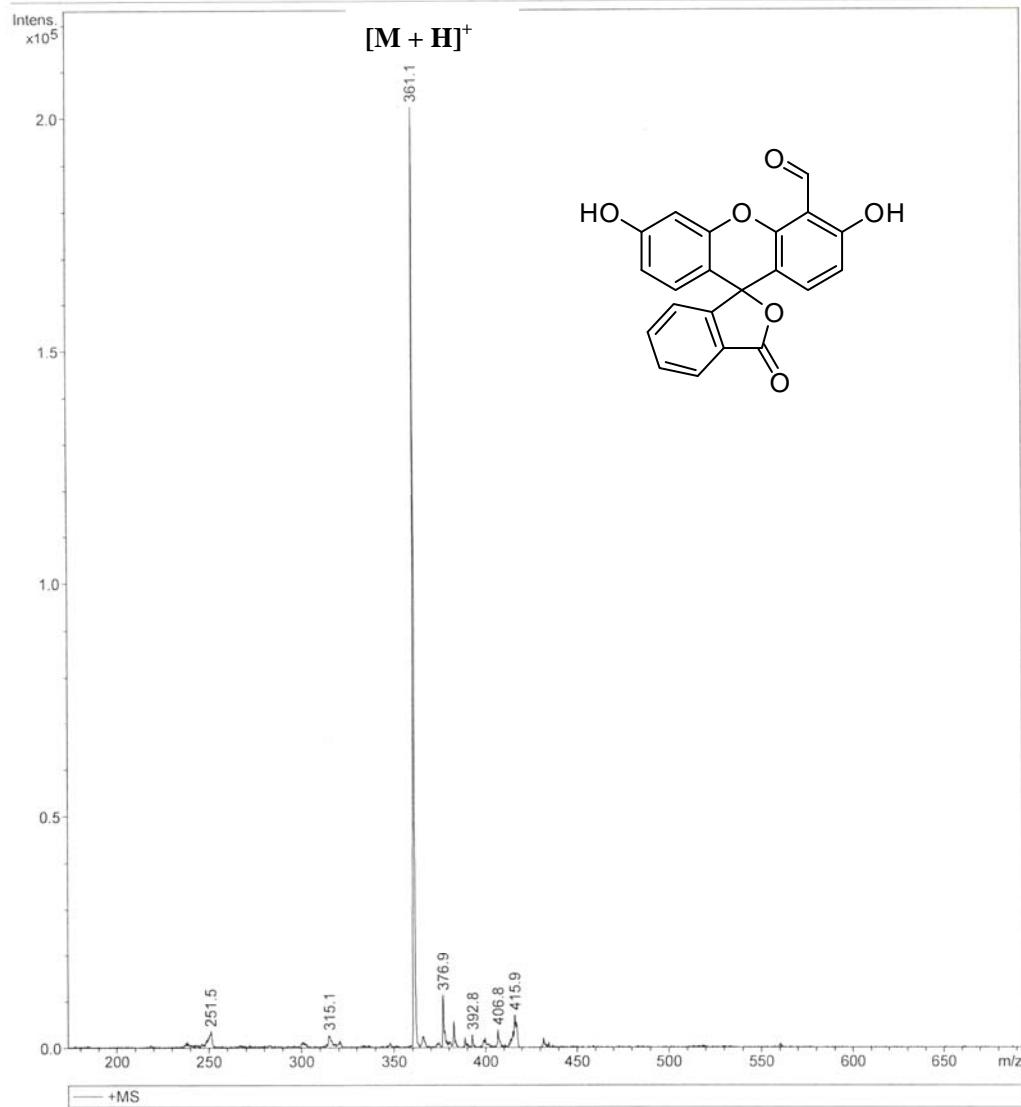


Figure S3. ESI-MS spectrum of monoaldehyde -functionalized fluorescein.

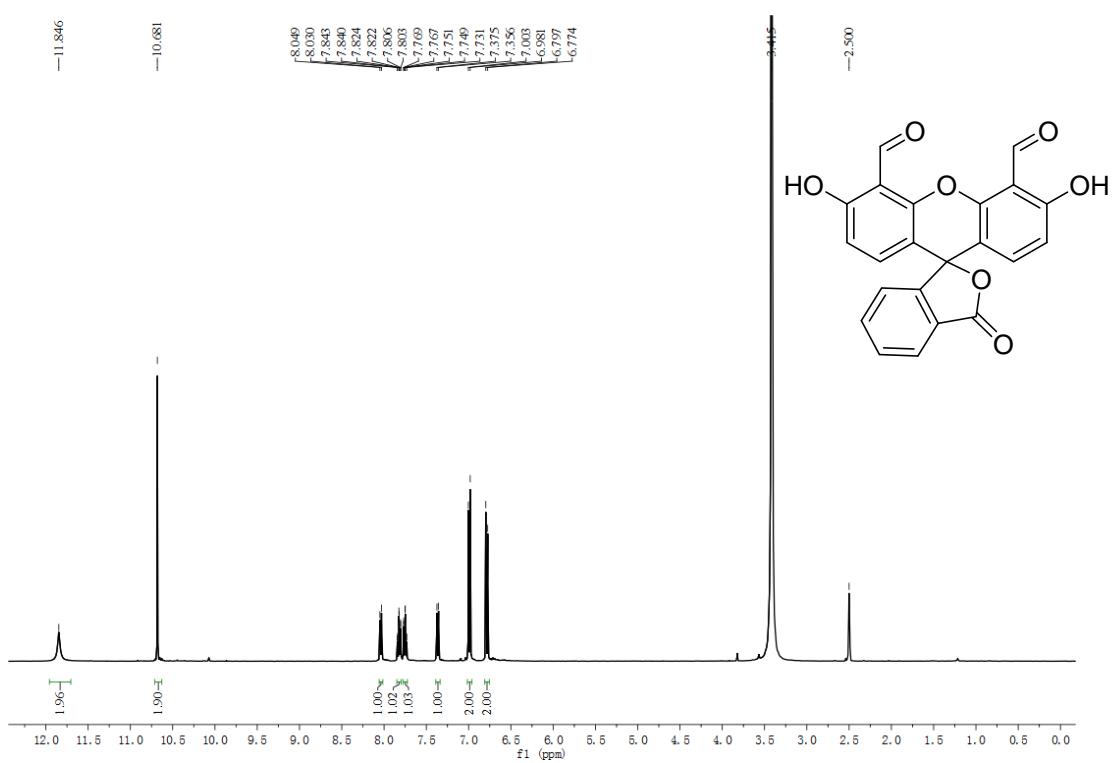


Figure S4. ¹H NMR spectrum (400 MHz, DMSO-*d*₆) of S2.

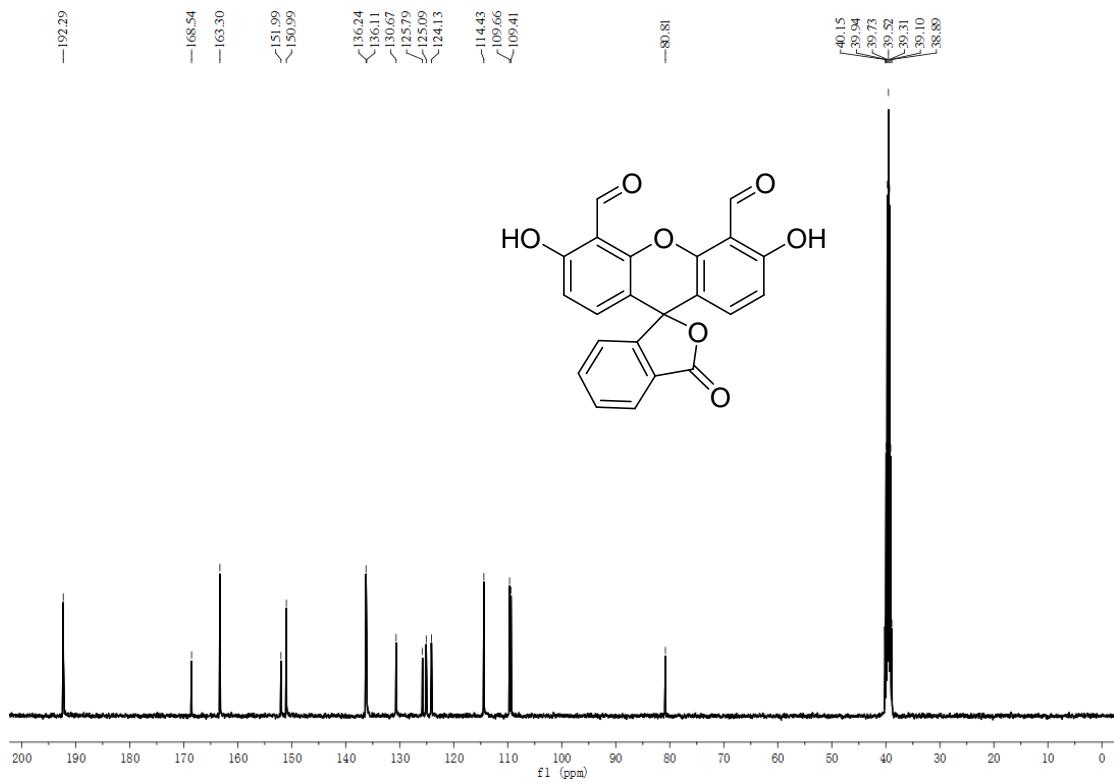


Figure S5. ¹³C NMR spectrum (100 MHz, DMSO-*d*₆) of S2.

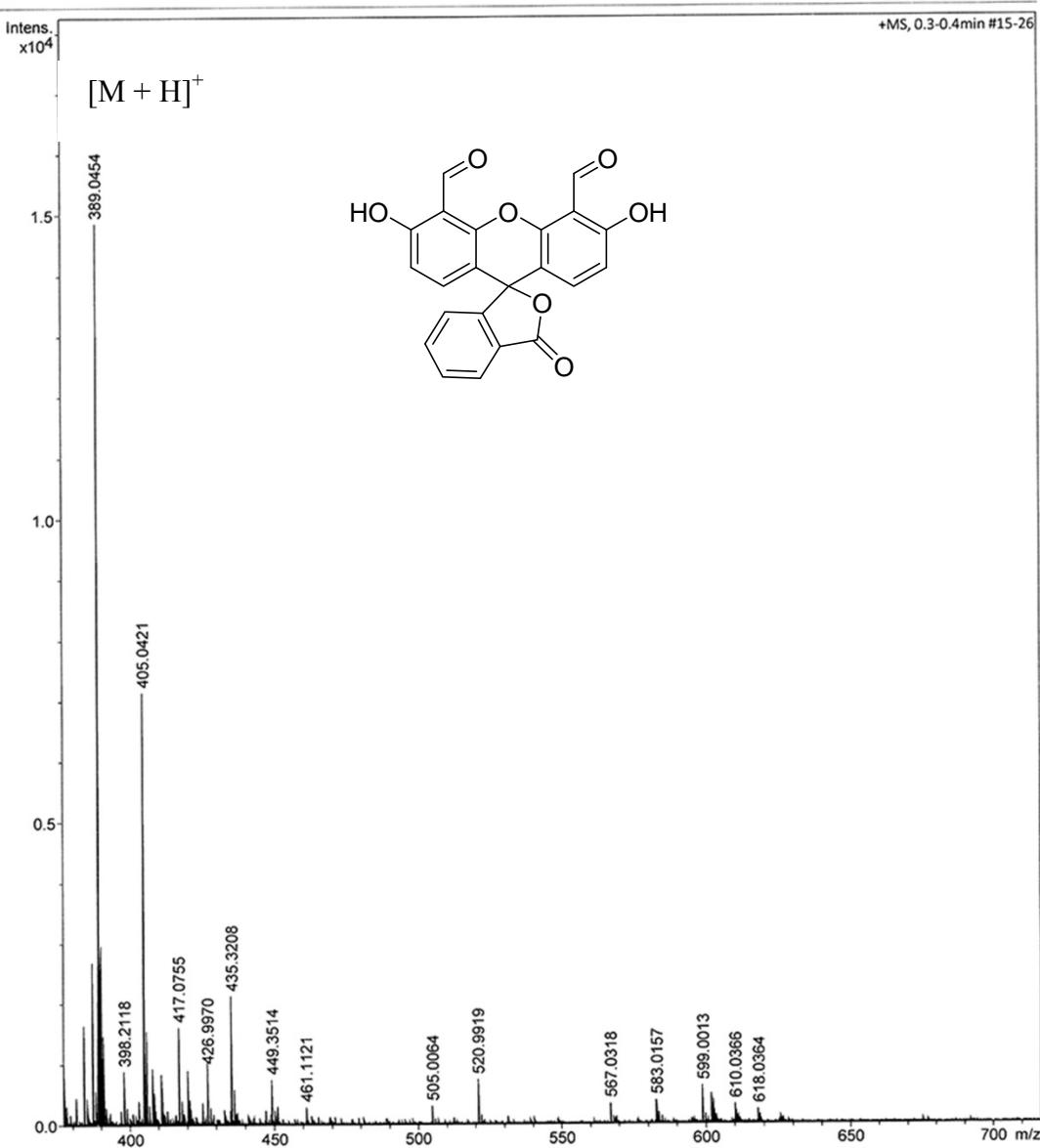
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Instrument micrOTOF



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Figure S6. ESI-MS spectrum of dialdehyde-functionalized fluorescein.

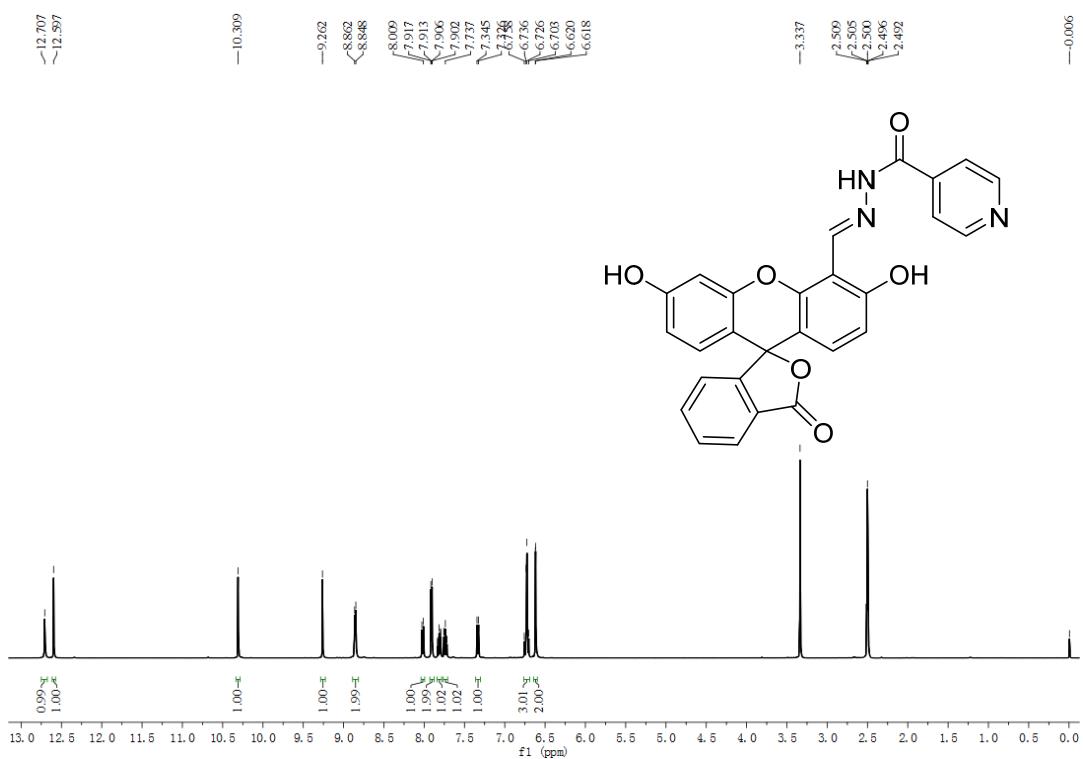


Figure S7. ^1H NMR spectrum (400 MHz, DMSO- d_6) of **F1**.

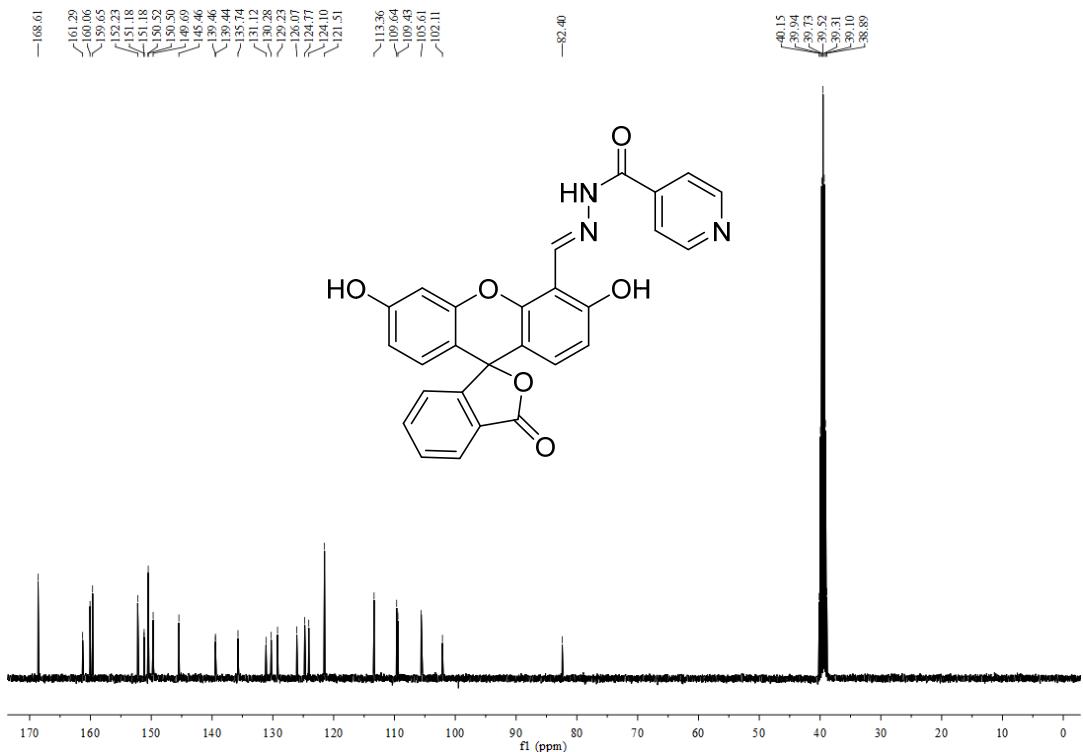


Figure S8. ^{13}C NMR spectrum (100 MHz, DMSO- d_6) of **F1**.

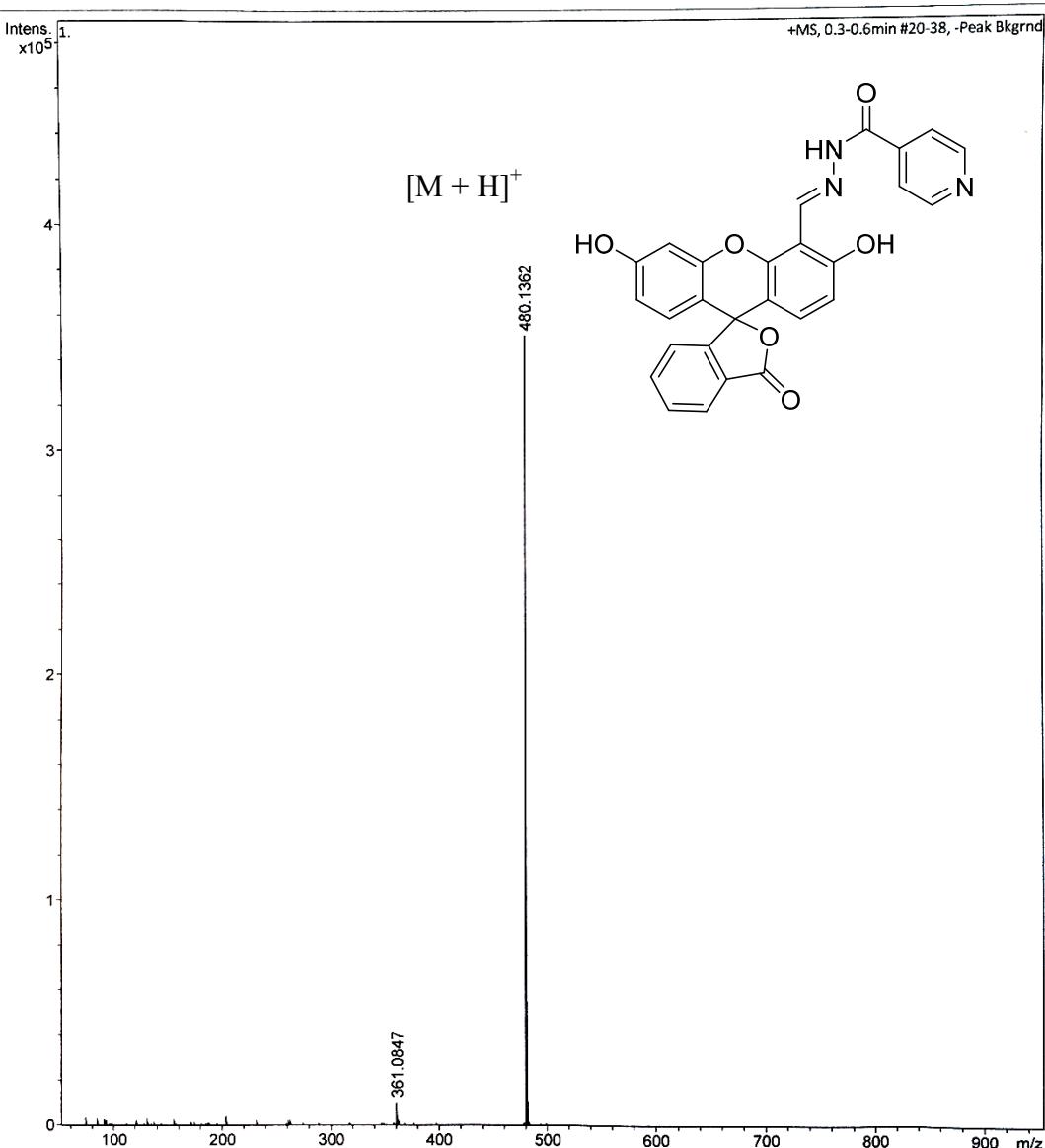
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Figure S9. ESI-MS spectrum of **F1**.

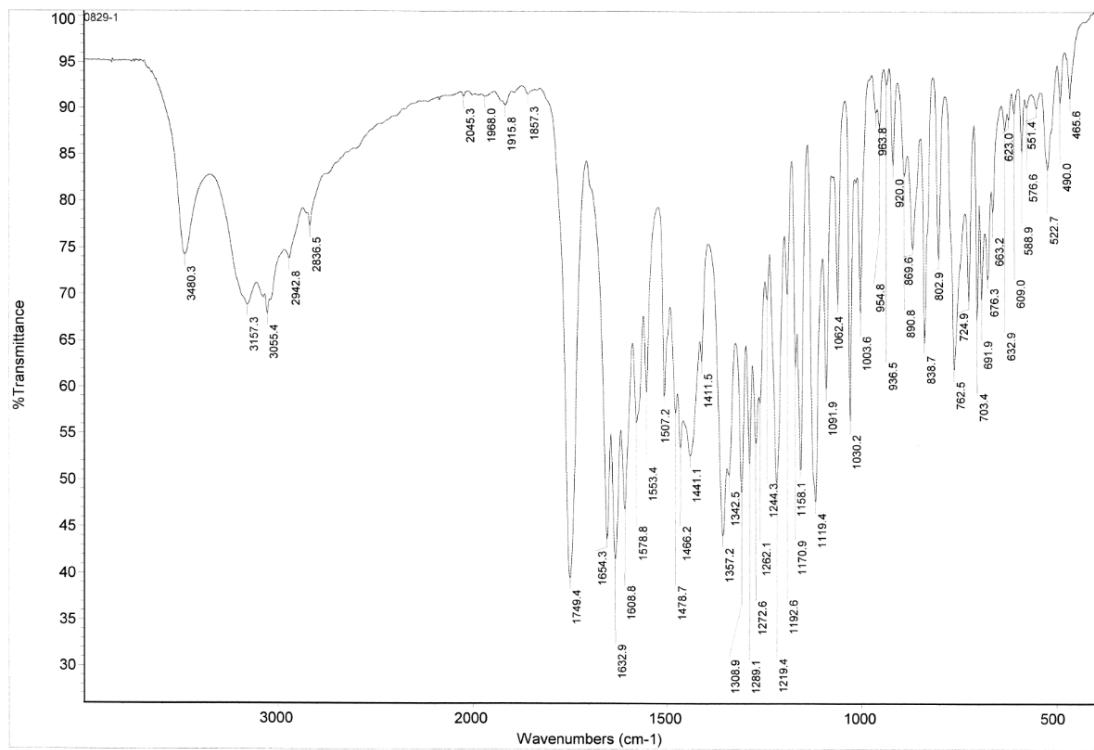


Figure S10. FTIR spectrum of **F1**.

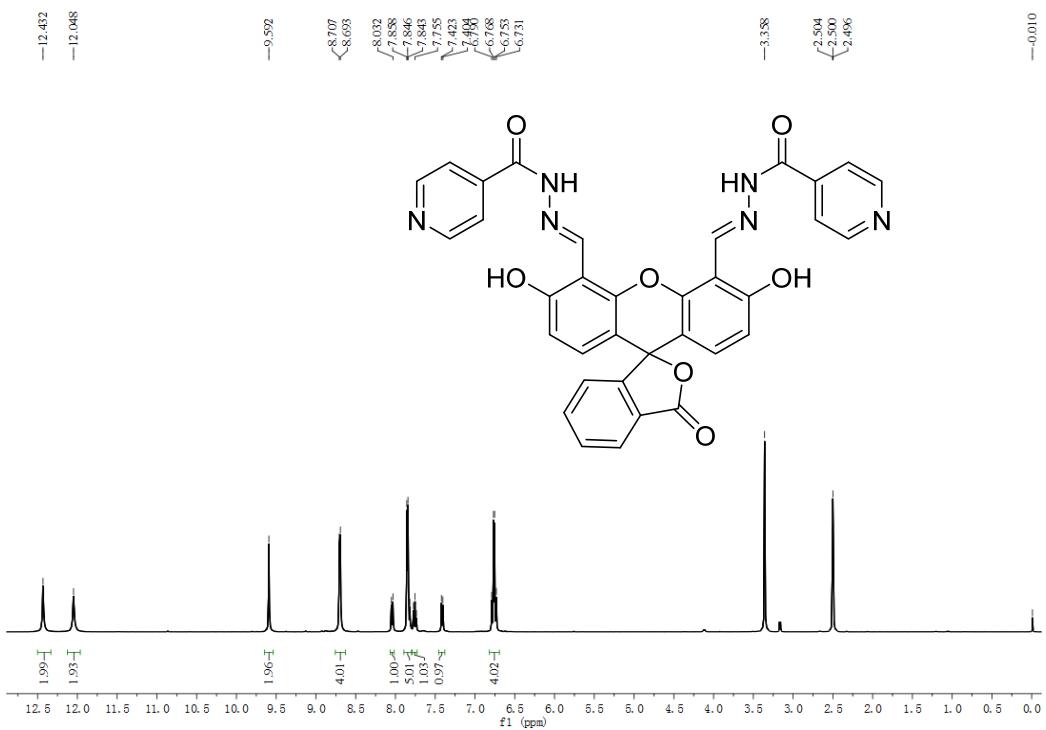


Figure S11. ^1H NMR spectrum (400 MHz, $\text{DMSO}-d_6$) of **F2**.

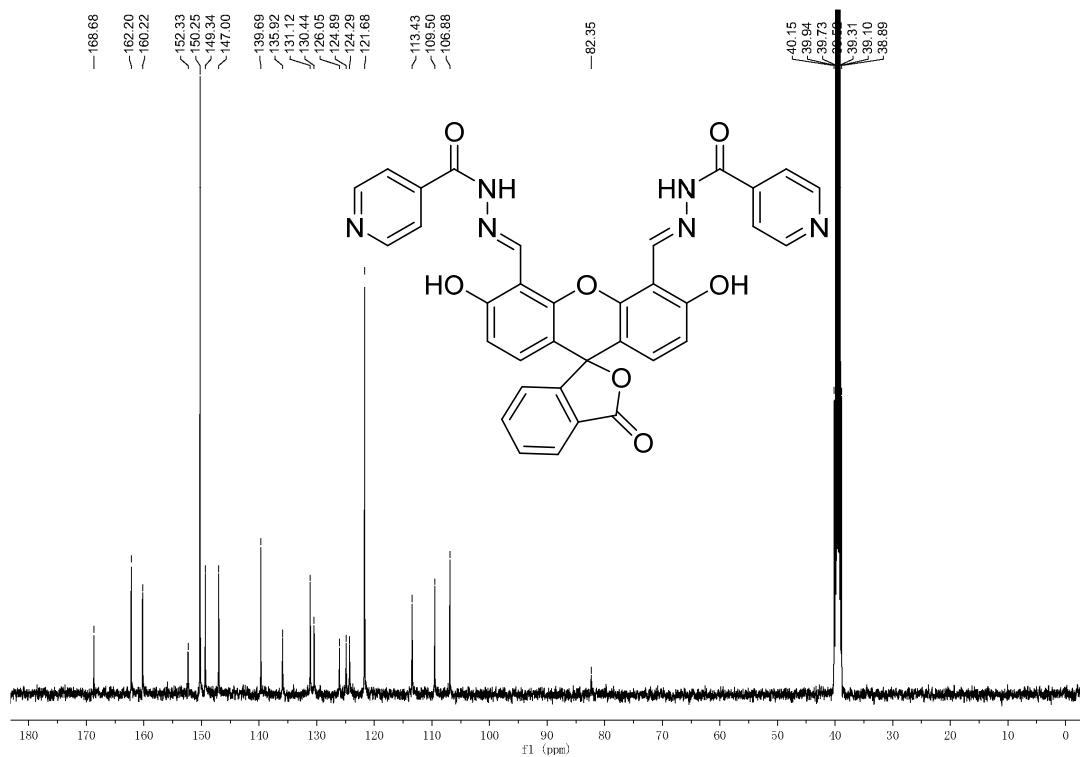


Figure S12. ^{13}C NMR spectrum (100 MHz, $\text{DMSO}-d_6$) of **F2**.

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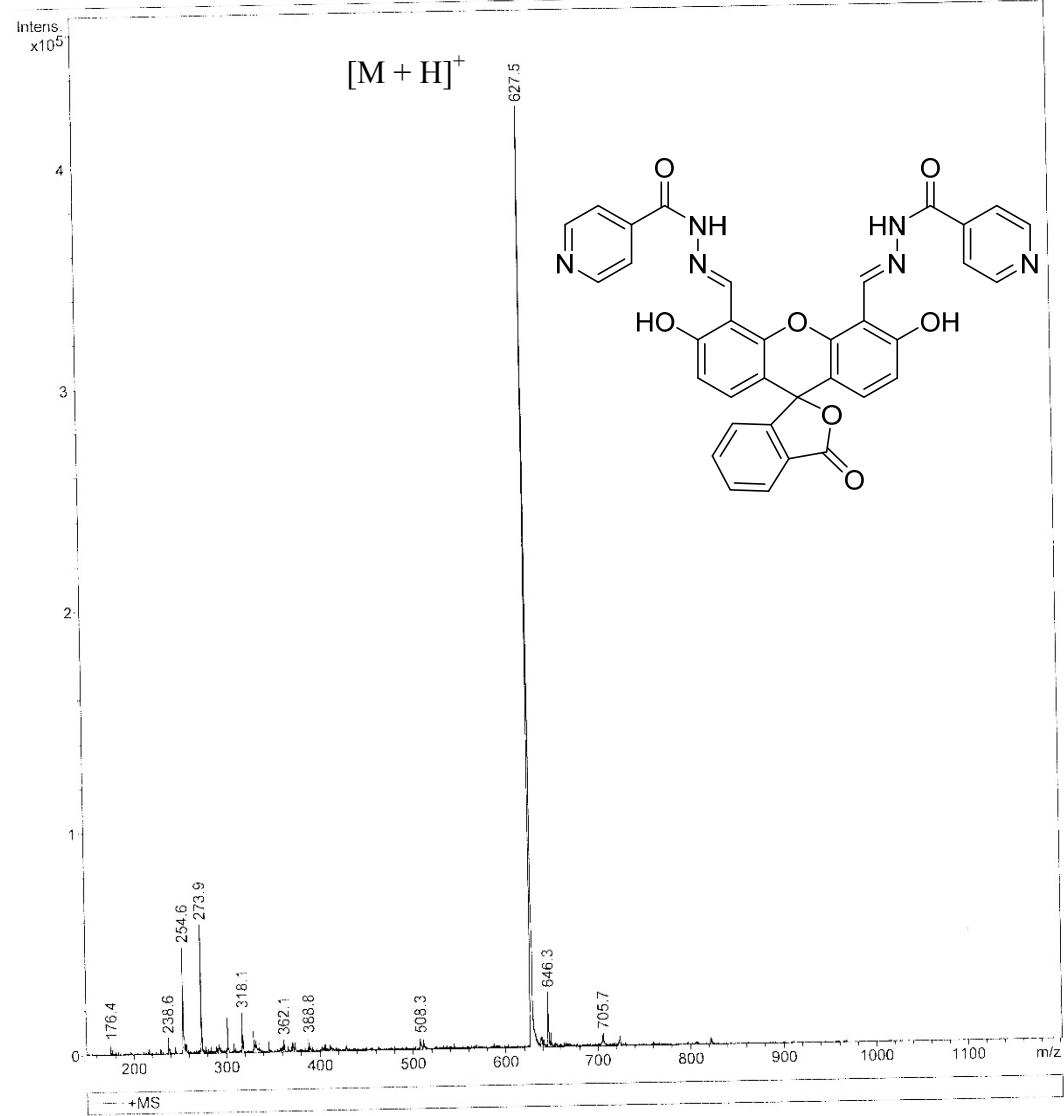


Figure S13. ESI-MS spectrum of F2.

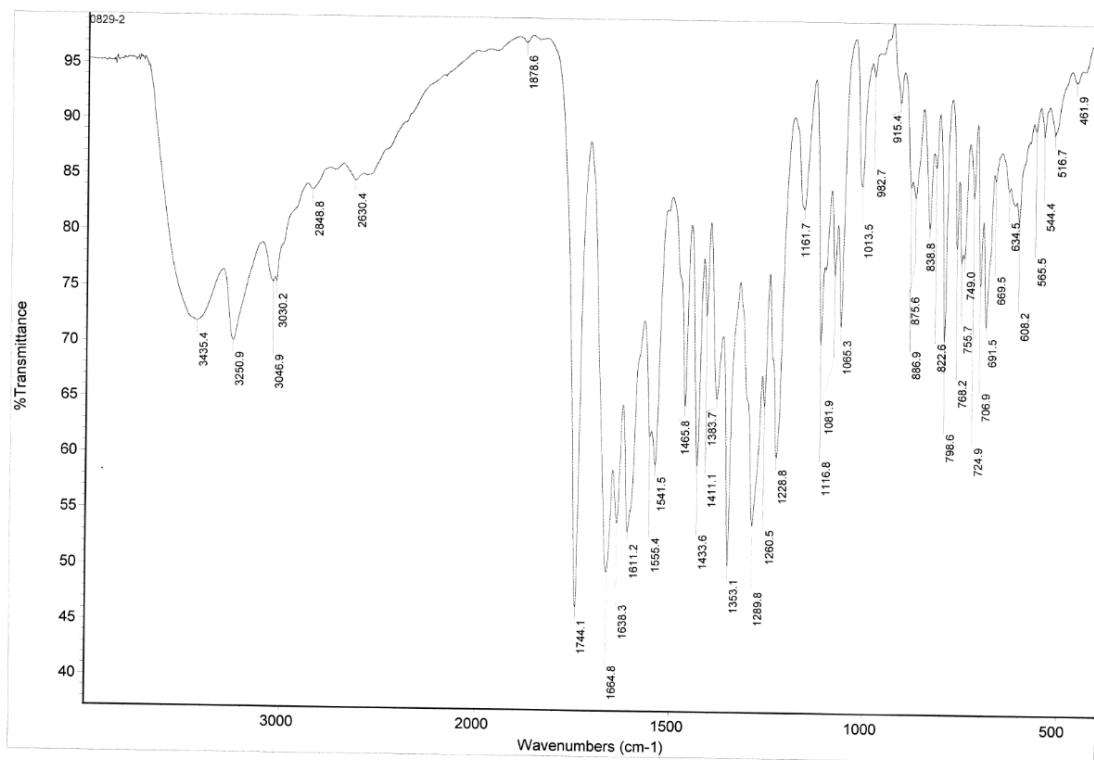


Figure S14. FTIR spectrum of F2.

6. The crystal data and structure refinement parameters.

Table S2. Crystal data and structure refinement parameters for **F1** and **F2**

Compound	F1	F2
Empirical formula	C ₂₉ H ₂₃ N ₃ O ₇ S	C ₃₈ H ₃₄ N ₆ O ₉ S ₂
Temperature/K	293 (2)	293 (2)
<i>M</i>	557.56	782.83
Crystal system	Triclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1
<i>a</i> /Å	7.7595(16)	8.2956(8)
<i>b</i> /Å	13.126(4)	15.1849(14)
<i>c</i> /Å	14.358(3)	16.4901(16)
<i>a</i> /°	106.59(2)	83.140(8)
<i>β</i> /°	94.664(19)	81.044(8)
<i>γ</i> /°	105.99(2)	87.483(8)
<i>V</i> /Å ³	1326.8(6)	2036.5(3)
<i>Z</i>	2	2
<i>D</i> _v /kg m ⁻³	1.396	1.277
μ /mm ⁻¹	0.176	0.190
<i>F</i> (000)	580.0	816.0
Crystal size/mm	0.15 × 0.12 × 0.07	0.23 × 0.21 × 0.19
θ Range for data collection/°	3.41-26.022	3.392-26.014
Index ranges, <i>hkl</i>	-9 ≤ <i>h</i> ≤ 9, -9 ≤ <i>k</i> ≤ 16, -17 ≤ <i>l</i> ≤ 16	-10 ≤ <i>h</i> ≤ 9, -18 ≤ <i>k</i> ≤ 18, -20 ≤ <i>l</i> ≤ 19
Reflections collected/unique	8550/5225	13764/7998
Data/restraints/params	5225/0/365	7998/0/509
Goodness-of-fit on <i>F</i> ²	1.054	0.933
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0941, <i>wR</i> 2 = 0.2012	<i>R</i> 1 = 0.0811, <i>wR</i> 2 = 0.1995
<i>R</i> indices (all data)	<i>R</i> 1 = 0.1973, <i>wR</i> 2 = 0.2879	<i>R</i> 1 = 0.1648, <i>wR</i> 2 = 0.2539

7. UV-vis spectra and images of F1 and F2 in different solution.

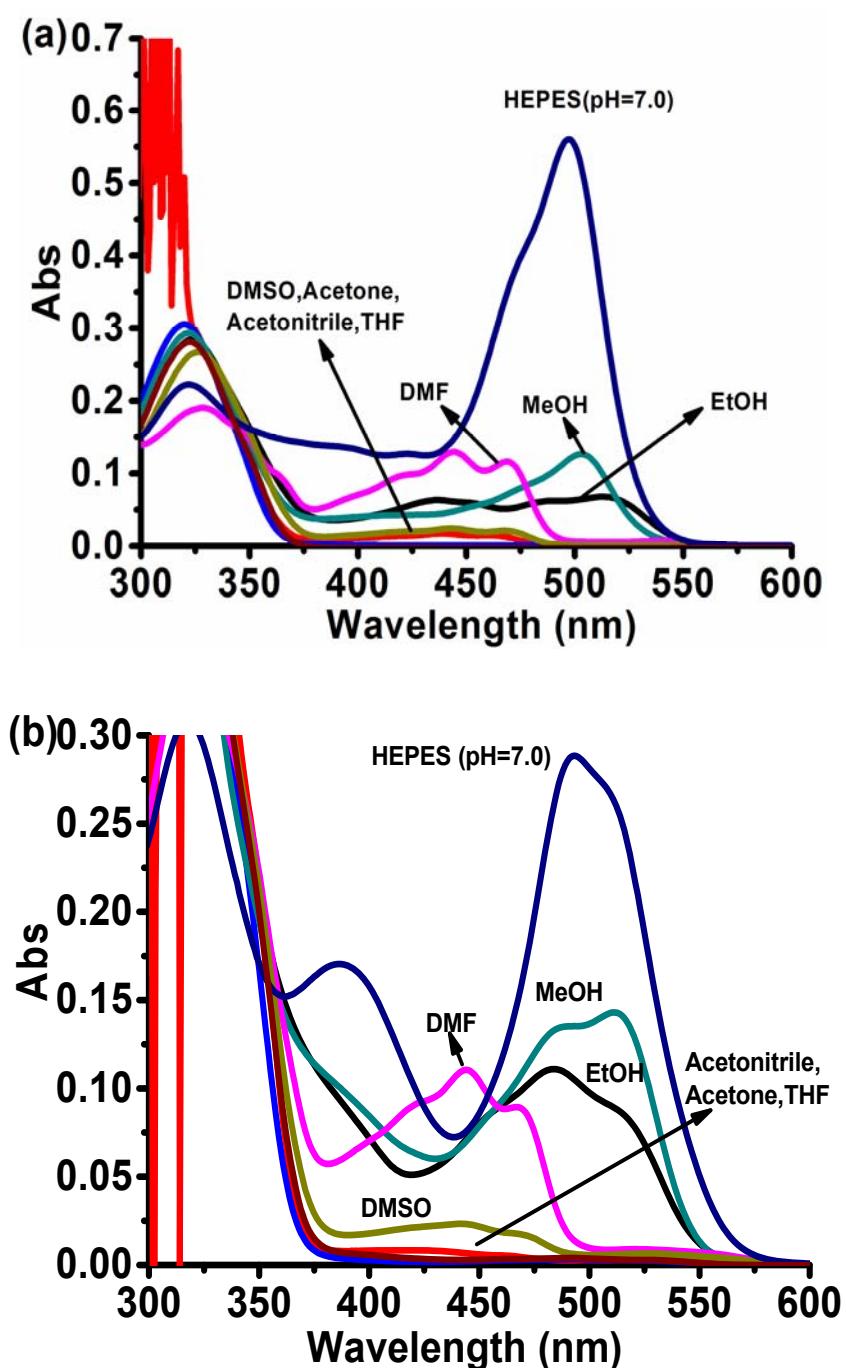


Figure S15. The UV-vis spectra of **F1** and **F2** ($10.0 \mu\text{M}$) in different solution. (a) for **F1**; (b) for **F2**.

8. UV-vis spectra and images of F1 and F2 with different NACs in ethanol.

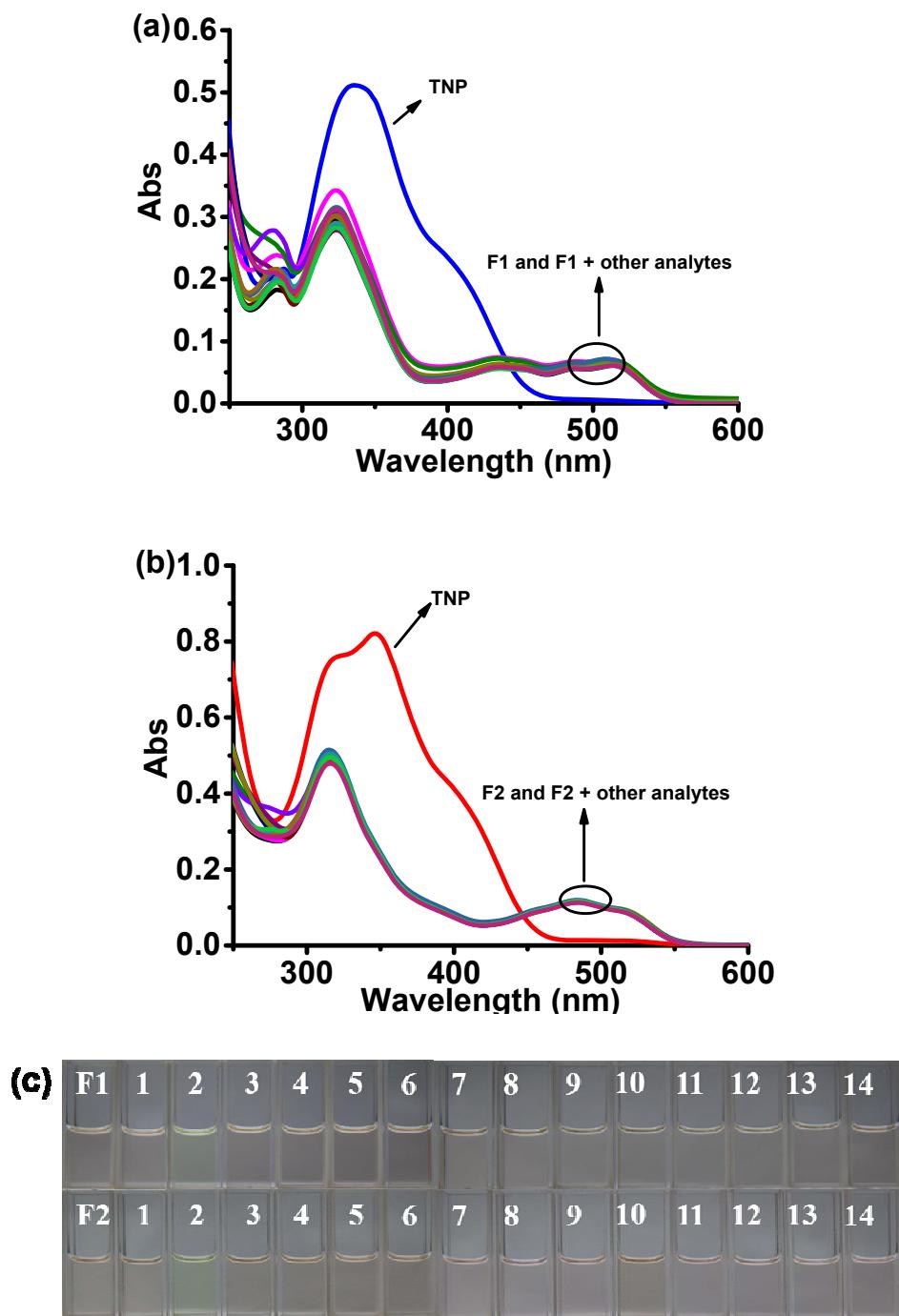


Figure S16. The UV-vis spectra and images of **F1** (a) and **F2** (b) ($10.0 \mu\text{M}$) toward different NACs ($10.0 \mu\text{M}$ or $20.0 \mu\text{M}$) in ethanol. (c) From 1 to 14: P, TNP, 3,5-DMP, 4-NB, 1,3-DNB, 4-NT, 2,4-DNT, 3,5-DMT, TNT, PAP, 2-IP, 4-MP, 2-CP, 4-BP.

9. UV-vis spectra and images of F1 and F2 with different NACs in HEPES buffer (pH = 7.0).

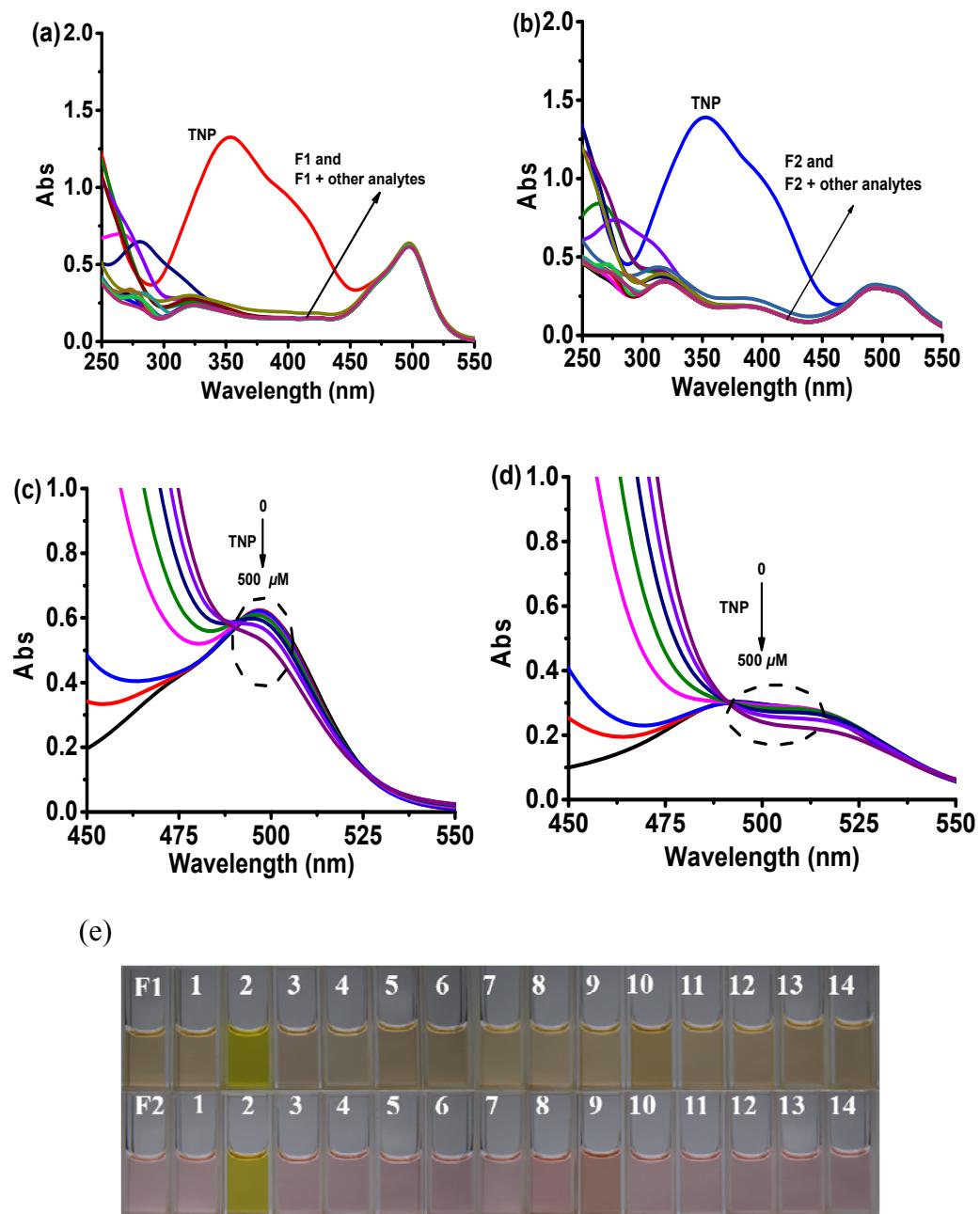


Figure S17. (a) (b) The UV-vis spectra of $10.0 \mu\text{M}$ **F1** and **F2** toward different NACs ($200 \mu\text{M}$) in HEPES buffer ($\text{pH} = 7.0$). (c)(d) The UV-vis spectra of $10.0 \mu\text{M}$ **F1** and **F2** toward TNP (0 - $500 \mu\text{M}$) in HEPES buffer ($\text{pH} = 7.0$). (e) The images observed from 1 to 14: P, TNP, 3,5-DMP, 4-NB, 1,3-DNB, 4-NT, 2,4-DNT, 3,5-DMT, TNT, PAP, 2-IP, 4-MP, 2-CP, 4-BP.

10. The fluorescence selectivity of F1 and F2 in ethanol.

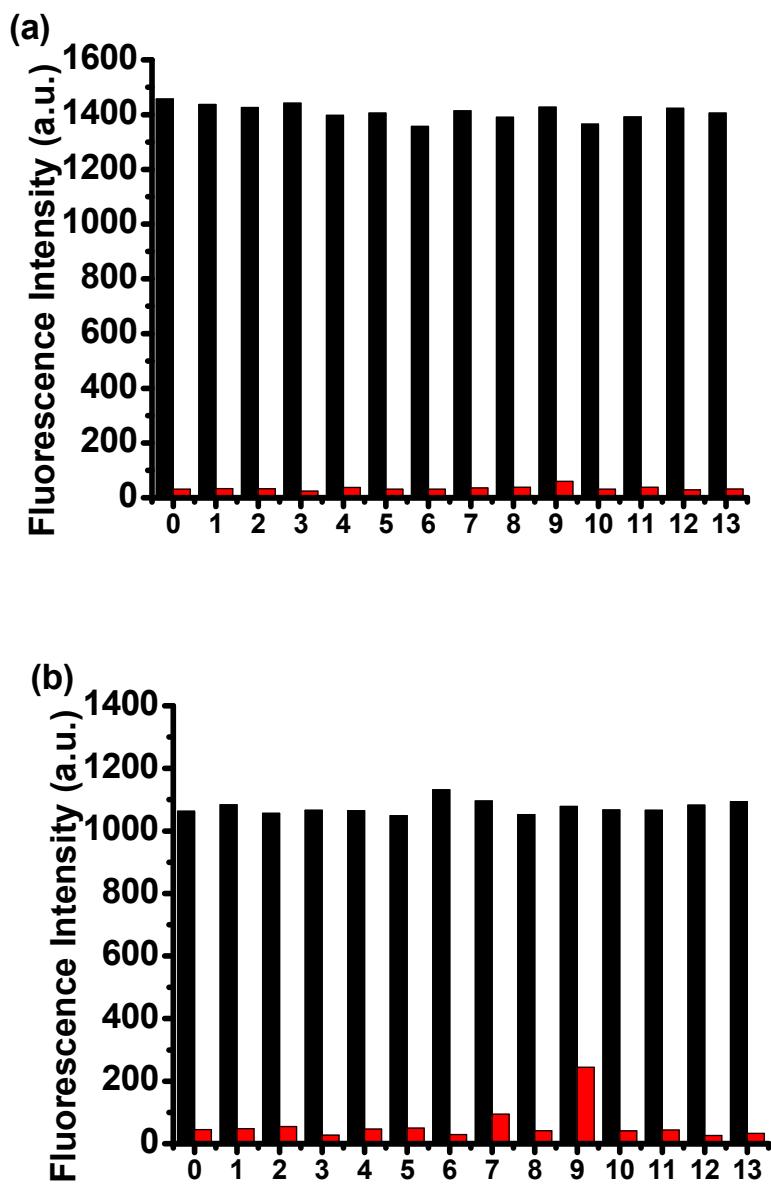


Figure S18. The fluorescence selectivity of chemosensors (10.0 μ M), the black bars represent the emission intensity of chemosensors in the presence of NACs (10.0 μ M or 200.0 μ M), the red bars represent the emission intensity that occurs upon the subsequent addition of 10.0 μ M or 200.0 μ M of TNP to the above solution. From 0 to 15: none, P, TNP, 3,5-DMP, 4-NB, 1,3-DNB, 4-NT, 2,4-DNT, 3,5-DMT, TNT, PAP, 2-IP, 4-MP, 2-CP, 4-BP. (a) F1 in ethanol, (b) F2 in HEPES buffer (pH=7.0).

11. The fluorescence spectral changes of F1 and F2 with the addition of TNP in ethanol.

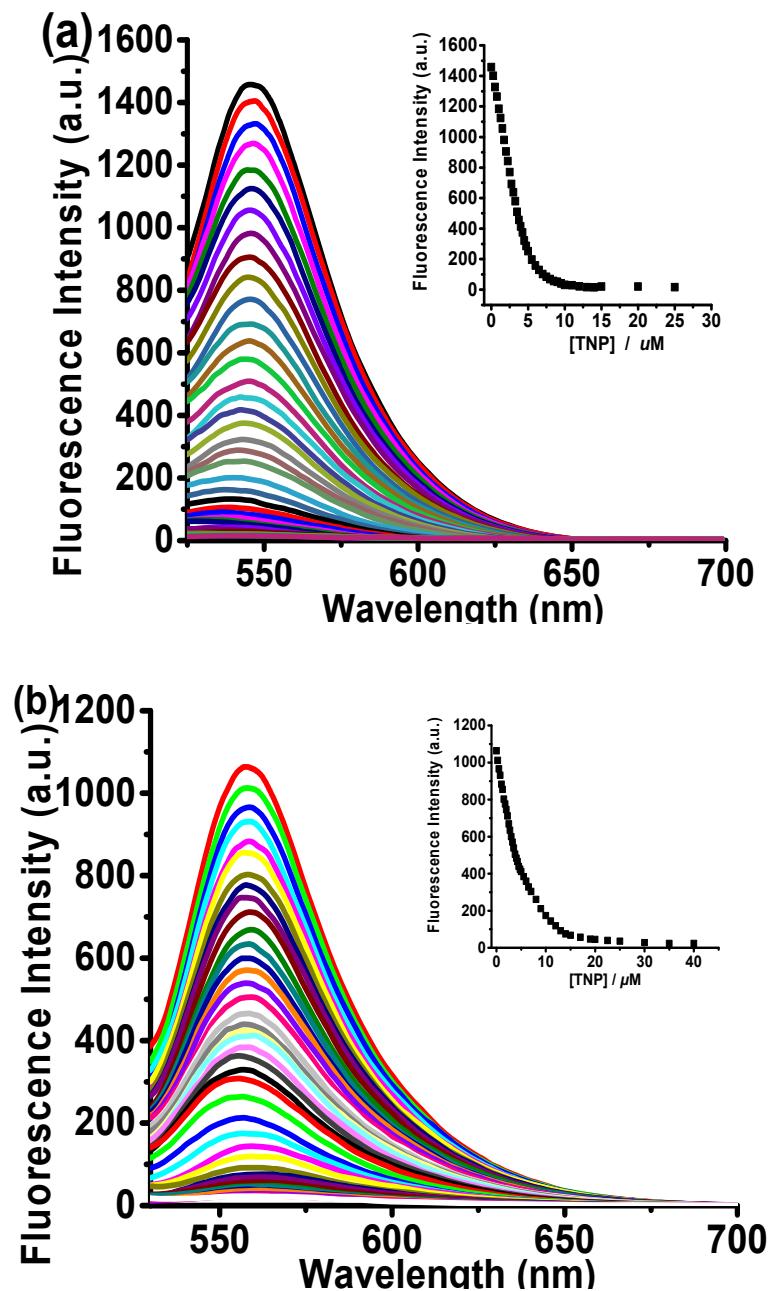


Figure S19. Fluorescence spectral changes of chemosensors ($10.0 \mu\text{M}$) with the addition of TNP in ethanol. (a) **F1**, Inset: Fluorescence intensity changes at 545 nm upon the addition of TNP ($\lambda_{\text{ex}} = 500 \text{ nm}$). (b) **F2**, Inset: Fluorescence intensity changes at 558 nm upon the addition of TNP ($\lambda_{\text{ex}} = 517 \text{ nm}$).

12. Plot of fluorescence intensity against time for F1 and F2 upon addition of TNP in ethanol.

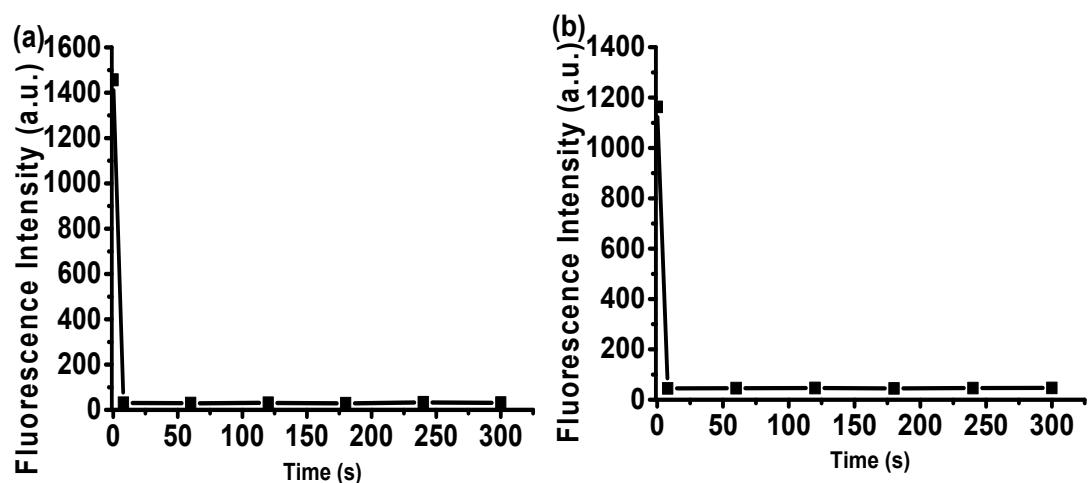


Figure S20. Plot of fluorescence intensity against time for chemosensors upon addition of TNP in ethanol. (a) F1; (b) F2.

13. The pH dependence of the fluorescence intensity change.

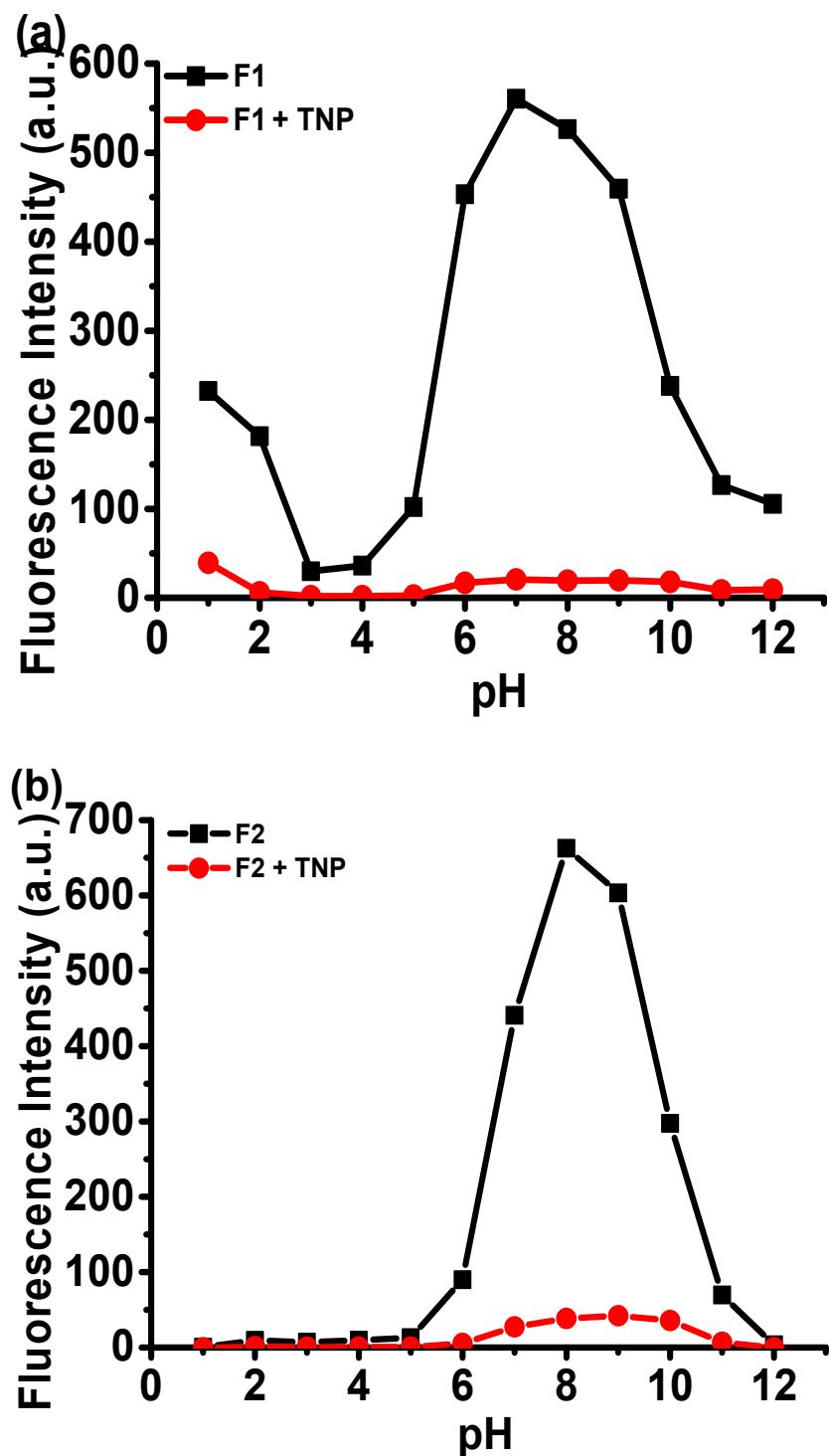


Figure S21. Fluorescence intensity of chemosensors ($10.0 \mu\text{M}$) in HEPES solution of different pH in the absence and presence of $200.0 \mu\text{M}$ TNP ($\lambda_{\text{ex}} = 400 \text{ nm}$).

14. The fluorescence spectral changes of F1 and F2 with the addition of TNP in HEPES buffer ($\text{pH} = 7.0$) under different excitation wavelength.

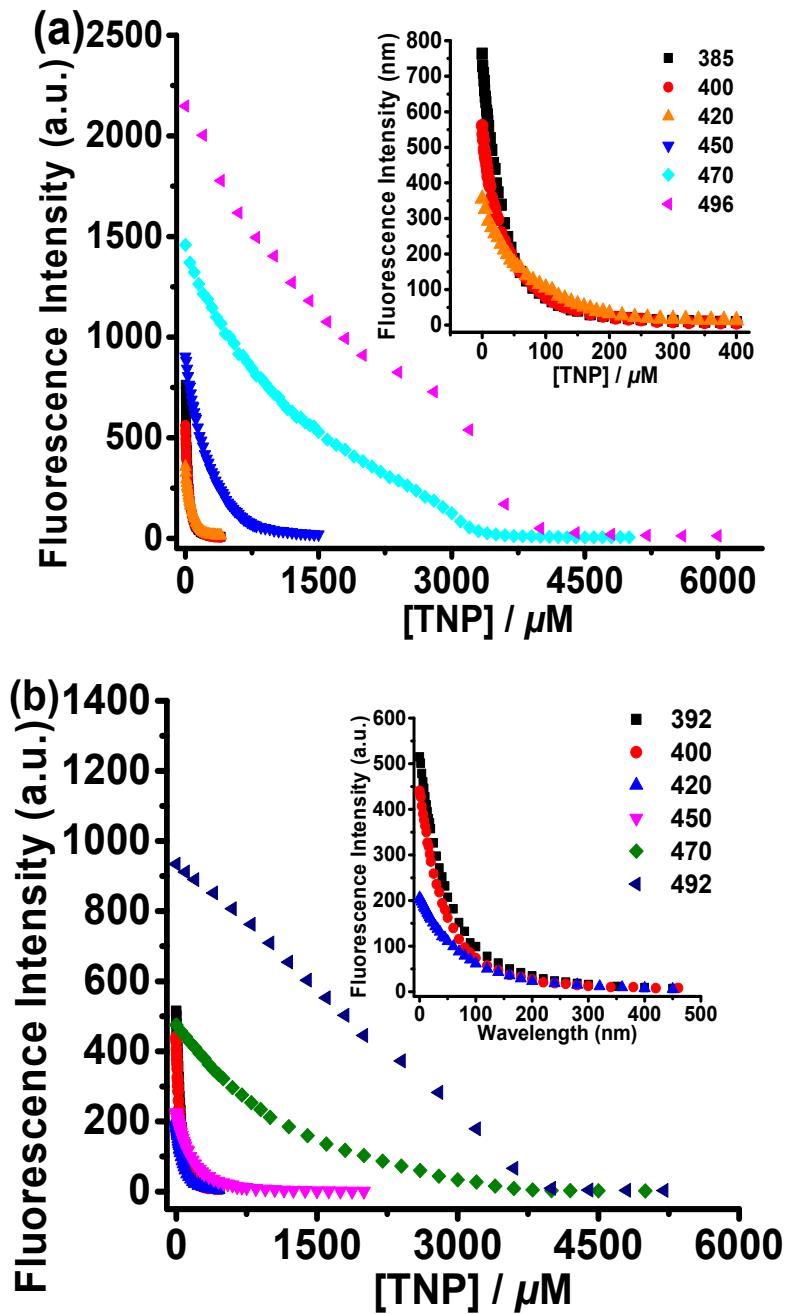


Figure S22. The fluorescence spectral changes of F1 and F2 with the addition of TNP in HEPES buffer ($\text{pH}=7.0$) under different excitation wavelength. (a) F1: $\lambda_{\text{em}} = 542 \text{ nm}$; (b) F2: $\lambda_{\text{em}} = 555 \text{ nm}$.

15. The fluorescence spectral changes of F1 and F2 with the addition of TNP in HEPES buffer (pH = 7.0).

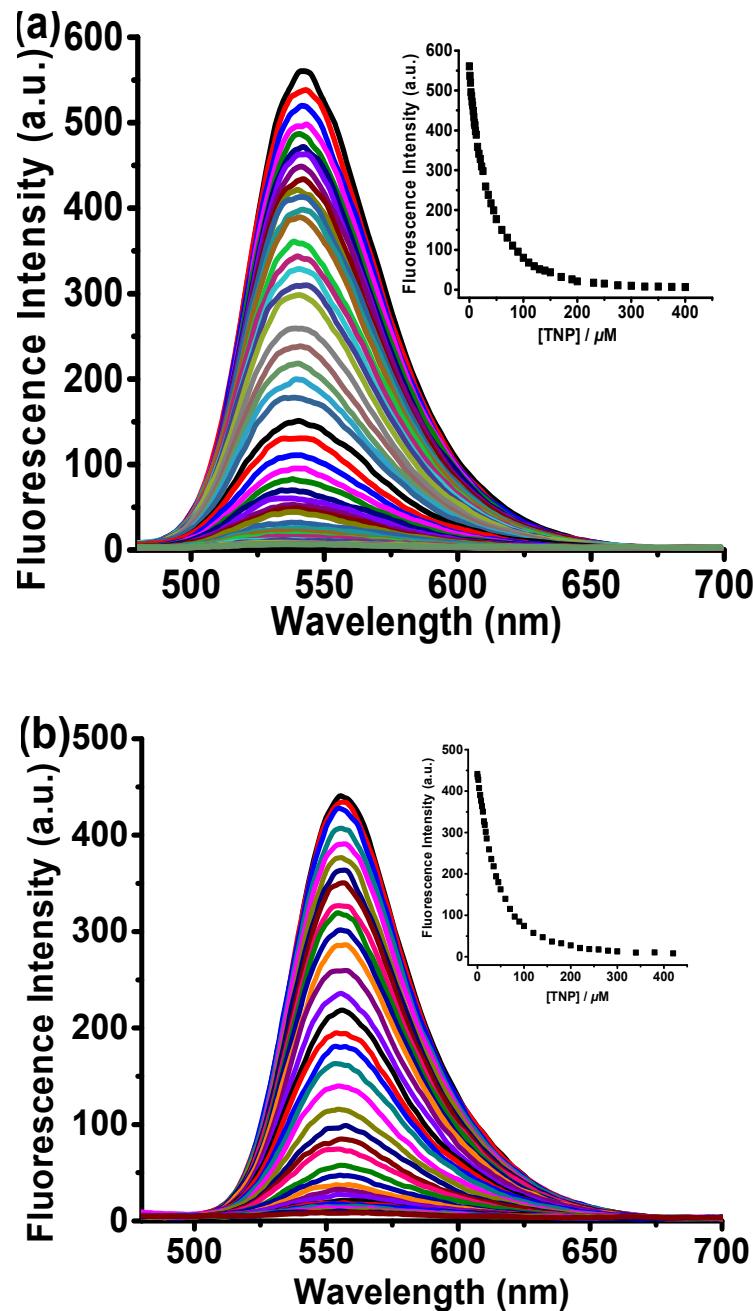


Figure S23. Fluorescence spectral changes of chemosensors ($10.0 \mu\text{M}$) with the addition of TNP in HEPES buffer ($\text{pH}=7.0$). (a) **F1**, Inset: Fluorescence intensity changes at 542 nm upon the addition of TNP ($\lambda_{\text{ex}}= 400 \text{ nm}$). (b) **F2**. Inset: Fluorescence intensity changes at 555 nm upon the addition of TNP ($\lambda_{\text{ex}} = 400 \text{ nm}$).

16. The fluorescence selectivity of F1 and F2 in HEPES buffer (pH=7.0).

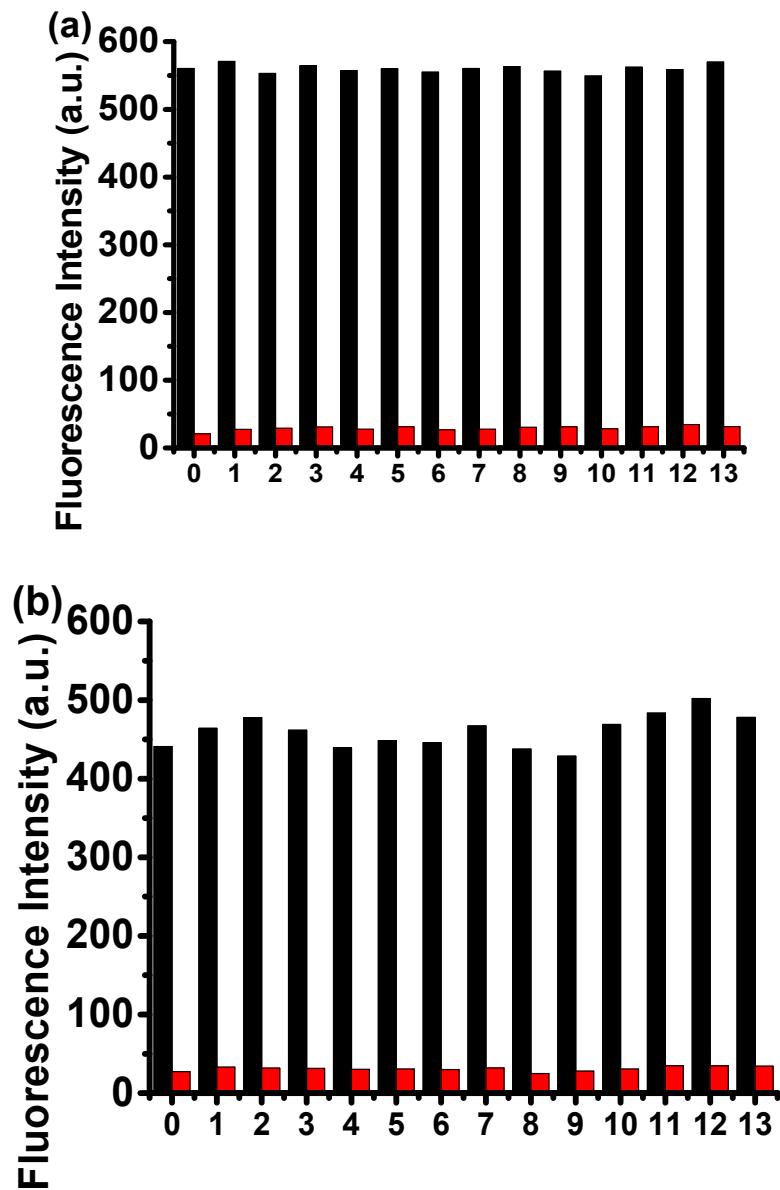


Figure S24. The fluorescence selectivity of chemosensors (10.0 μ M) in HEPES buffer (pH = 7.0), the black bars represent the emission intensity of chemosensors in the presence of NACs (200.0 μ M), the red bars represent the emission intensity that occurs upon the subsequent addition of 200.0 μ M of TNP to the above solution. (a) for F1; (b) for F2. From 0 to 13: none, P, 3,5-DMP, 4-NB, 1,3-DNB, 4-NT, 2,4-DNT, 3,5-DMT, TNT, PAP, 2-IP, 4-MP, 2-CP, 4-BP.

17. Stern-Volmer plots of F1 and F2 in ethanol.

The fluorescence quenching process can be analyzed by Stern-Volmer equation:

$$F_0/F - 1 = K_{sv} [Q]$$

Where F_0 is the fluorescence intensity in the absence of TNP, F is the fluorescence intensity upon the addition of TNP and $[Q]$ is the concentration of TNP.

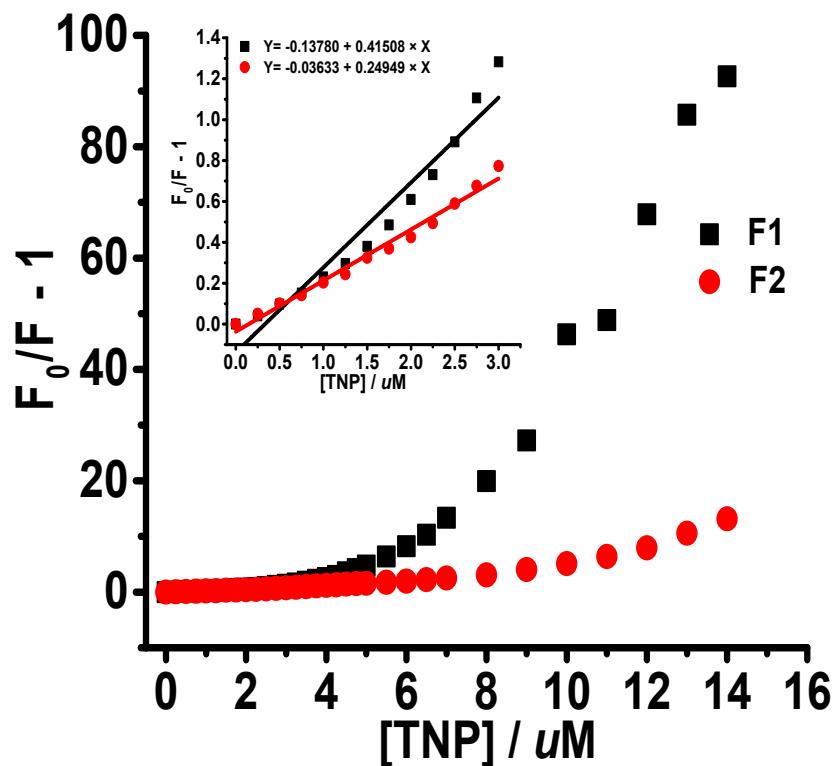


Figure S25. Stern-Volmer plots of F1 and F2 for TNP in ethanol.

18. The detection limit of F1 and F2 with TNP in ethanol.

The detection limit was calculated accordingly to literature⁸

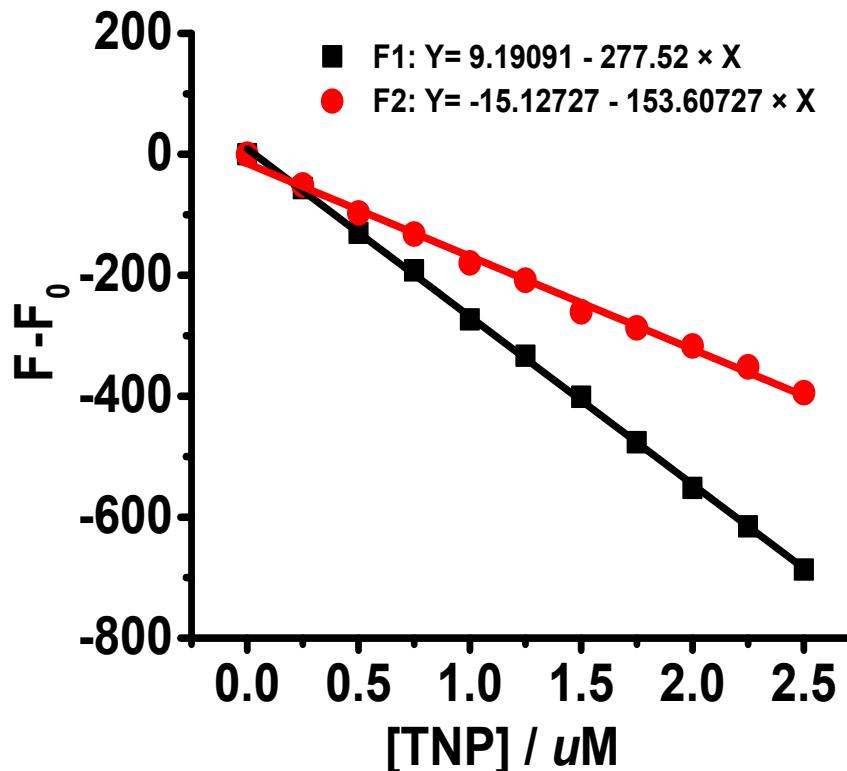


Figure S26. Plot of the intensity a mixture of chemosensors and TNP in ethanol.

Table S3. The Stern-Volmer quenching constants and detection limits in ethanol.

	F1	F2
TNP		TNP
LOD/ μM	0.075	0.110
K_{sv} / M^{-1}	4.15×10^5	2.5×10^5

(8) Z.-H. Fu, L.-B. Yan, X. Zhang, F.-F. Zhu, X.-L. Han, J. F, Y.-W. Wang, Y. Peng, *Org. Biomol. Chem.*, 2017, **15**, 4115–4121.

19. Stern-Volmer plots of F1 and F2 in HEPES buffer (pH = 7.0).

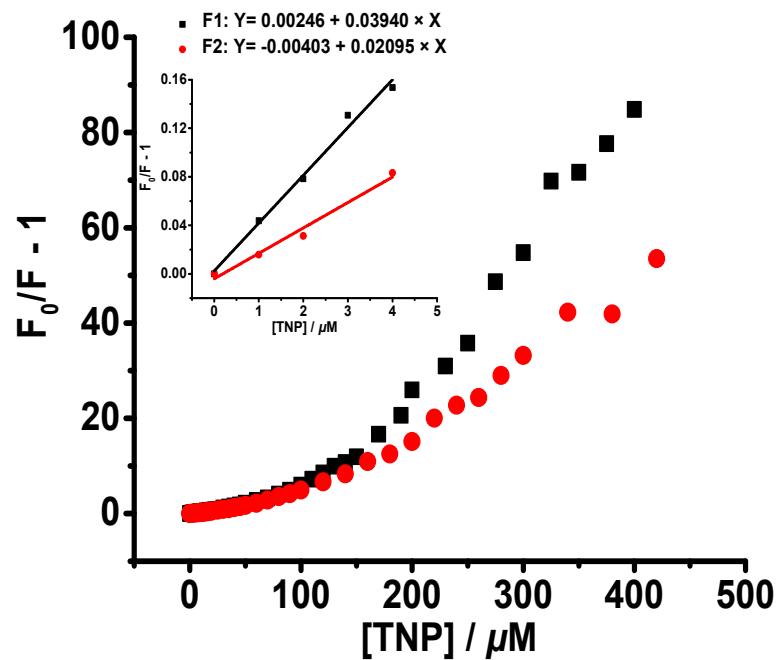


Figure S27. Stern-Volmer plots of F1 and F2 in HEPES buffer (pH = 7.0).

20. The detection limit of F1 and F2 with TNP in HEPES buffer (pH = 7.0).

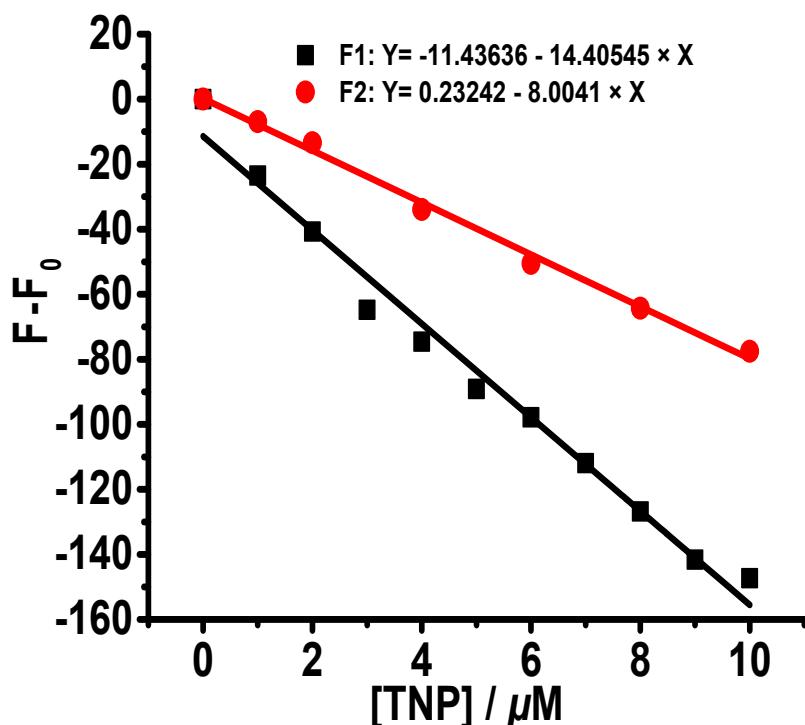


Figure S28. Plot of the intensity a mixture of chemosensors and TNP in HEPES buffer (pH = 7.0).

Table S4. The Stern-Volmer quenching constants and detection limits in HEPES buffer (pH = 7.0).

	F1	F2
TNP	TNP	TNP
LOD/ μM	0.57	0.93
K_{sv} / M^{-1}	3.94×10^4	2.1×10^4

21. Optimized structures, cartesian coordinates and computed total energies of F1, F1+TNP, F2 and F2+TNP.

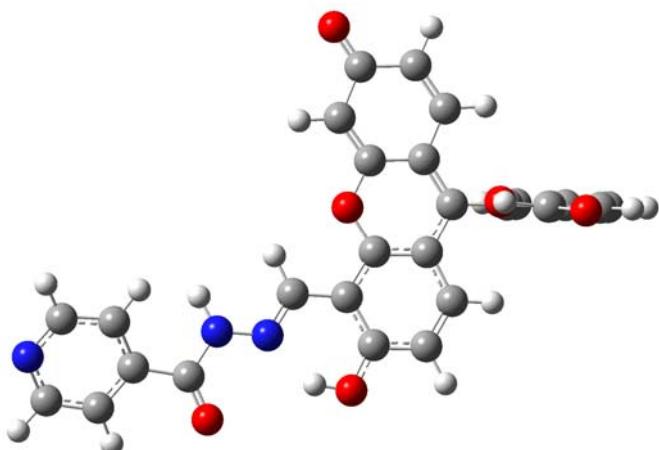


Figure S29. Optimized structure of F1.

B3LYP/6-31G(d) in gas phase, E = -1654.6897019 a.u.

	charge = 0	Multiplicity = 1	
O	2.085330087	5.5547045071	-0.1064262301
O	0.3387702364	1.1844536977	-0.1553870583
O	-1.6609356449	-3.1425208947	-0.2432275419
H	-2.4906651369	-2.6015434234	-0.1713007968
O	5.3195509389	-1.6375590798	2.8479825886
O	-5.4609931534	-2.2314038754	-0.1715707789
N	-3.0801773736	-0.9371089169	-0.0500432237
N	-4.2696634125	-0.2911153889	0.041672068
H	-4.2861781951	0.7240528898	-0.0116392294
N	-9.1328163765	1.2077393496	0.2046395561
C	1.4239429112	2.0343776146	-0.1904448768
C	1.177803487	3.3671175384	-0.1310768762
H	0.1632275706	3.743834889	-0.0617275686
C	2.2718541023	4.3346146245	-0.1563543502
C	3.6292496972	3.7588770435	-0.2429203311
H	4.4503340214	4.4695071489	-0.2533364401
C	3.8484581834	2.4251905006	-0.3025639331
H	4.859569143	2.0350792144	-0.3617855565
C	2.7581327122	1.4721410225	-0.2876998676
C	2.9200441951	0.1051620966	-0.3393660652
C	1.7584118314	-0.7539482957	-0.3194581581
C	1.8128712669	-2.165661056	-0.3938765194
H	2.7810647376	-2.647799513	-0.4756747449
C	0.6710763134	-2.937651603	-0.366503156
H	0.7162738907	-4.0196306756	-0.4233153363

C	-0.5983156947	-2.3343628229	-0.2637051916
C	-0.7069001992	-0.9180233983	-0.1909323092
C	0.4851330935	-0.1645048249	-0.2223792031
C	4.6610805062	-1.0954406905	1.982908225
C	5.0497173426	-1.0628101884	0.5389055774
C	6.2988477873	-1.6335745619	0.2477033648
H	6.8667132864	-2.050496386	1.0722225824
C	6.7948117674	-1.6603947563	-1.0506379533
H	7.7645612495	-2.105214572	-1.2532291456
C	6.0382461834	-1.1096138667	-2.0852412523
H	6.4113729486	-1.1198291009	-3.1055165319
C	4.7948353975	-0.541924192	-1.810803731
H	4.2052741523	-0.1163728895	-2.6175117459
C	4.2773282377	-0.5057713993	-0.5069208661
C	-1.991360867	-0.2471319338	-0.0937204349
H	-1.9979781413	0.8455757741	-0.0614152288
C	-5.4470707045	-1.0213611066	-0.0286867185
C	-6.7007973339	-0.2004015868	0.0742399428
C	-6.7798689142	1.0459418453	0.7067206037
H	-5.92291916	1.496301837	1.1998454573
C	-8.012433588	1.7004075585	0.7435679456
H	-8.1016424279	2.6669933594	1.2361564325
C	-9.0487219852	0.0082454703	-0.3872982897
H	-9.9720985843	-0.3791700895	-0.813505975
C	-7.8721210069	-0.733139383	-0.4730476653
H	-7.8477696985	-1.7070951644	-0.9494805629
O	3.5024951294	-0.4571621459	2.2637449142
H	3.3743221909	-0.5540072002	3.2265731859

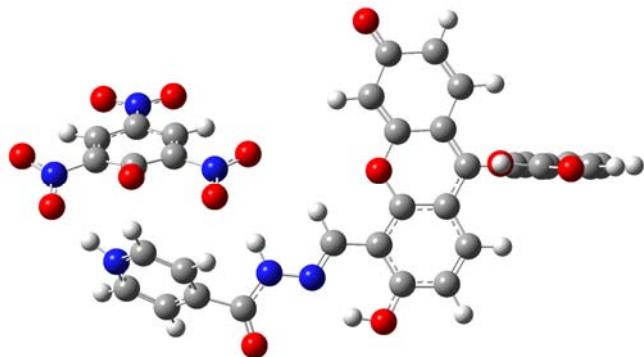


Figure S30. Optimized structure of **F1+TNP**.

B3LYP/6-31G(d) in gas phase, E = -2575.6494594 a.u.

charge = 0 Multiplicity = 1

C	-3.9039997178	-3.1953220797	1.5240868847
H	-3.1760964409	-3.2966748921	2.3200926151
C	-5.1280263779	-2.6099011069	1.7924174015
H	-5.4067509744	-2.2189731622	2.7602909344
C	-5.7702184524	-2.8217562004	-0.471593132
H	-6.5470604009	-2.605746353	-1.1952718239
C	-4.5637296857	-3.4153705977	-0.790617504
H	-4.3516764109	-3.7194893408	-1.8092014413
C	-3.5824683737	-3.5255210842	0.2044802867
C	-2.1738471742	-3.9078189116	-0.1944953924
C	0.7678504187	-1.9887075682	-0.0417791167
H	0.3480057687	-1.0397788598	0.3003654516
N	-1.2849559963	-2.9286594779	0.1684490322
H	-1.6341018282	-2.0178124173	0.496083329
N	0.0334244332	-3.045599443	-0.1739909797
N	-6.0062586938	-2.4143334558	0.7925480141
H	-6.7765271705	-1.7352177988	0.9448258618
O	-1.9173750024	-4.9367989398	-0.7948165135
C	-4.0108698471	1.273620054	0.1412934988
C	-3.9598033703	2.3273721552	-0.7563805801
H	-3.0068871686	2.6771765752	-1.1331125297
C	-5.1419270196	2.9152427068	-1.2012488111
C	-6.3881649836	2.4291918425	-0.7970933651
H	-7.2942829807	2.8665444471	-1.1974333565
C	-6.4491485733	1.3712559095	0.0873404967
C	-5.2670373621	0.7941442118	0.7431318312
N	-2.7622119979	0.6128684711	0.4356158153
N	-5.0805468595	4.0076569151	-2.1596660067
N	-7.7678284874	0.8485192211	0.3869611877
O	-5.3433184655	0.0324078137	1.7169678041

O	-2.8079554265	-0.5315584874	0.9313534616
O	-1.6985596376	1.1633596652	0.1392917201
O	-3.9662372324	4.3914908071	-2.5217454
O	-6.1485009812	4.4831168017	-2.5541158584
O	-8.7334369231	1.5982850649	0.2813066402
O	-7.9022411725	-0.3585544	0.6902379684
O	8.3665343193	-1.0194656961	2.2587765263
O	2.2711003946	4.7351599761	1.5941391867
O	2.3364507908	0.220280532	0.2658526677
O	2.146716751	-4.3567081804	-1.042139016
H	1.1927175613	-4.2268884109	-0.8009160345
C	7.4426168164	-0.5923586058	1.5953138142
C	7.5970594112	-0.0300468564	0.2180139431
C	8.9201413593	0.049522893	-0.2447005202
H	9.7060892382	-0.3022045321	0.4145820601
C	9.2154919832	0.5653779461	-1.50129931
H	10.2463422391	0.6192520149	-1.8385901527
C	8.1781905774	1.0142022568	-2.31908727
H	8.3905604338	1.4231416728	-3.3029455311
C	6.8594398813	0.939151145	-1.8731122404
H	6.0526583169	1.2865756614	-2.5117793444
C	6.5412142117	0.4211984565	-0.6082236209
C	5.087636667	0.3662979892	-0.2511762594
C	4.4230299338	1.4831092766	0.202768662
C	5.0547750139	2.7609362002	0.4601311448
H	6.121829949	2.8430951143	0.2791990218
C	4.3503019738	3.822487301	0.9140356411
H	4.8272481305	4.7785223565	1.1087014178
C	2.8974958948	3.7559294386	1.1781595111
C	2.266026404	2.4633911355	0.9199535275
C	2.9976336828	1.4135920973	0.471988533
C	2.9781671646	-0.8825614326	-0.1895387553
C	2.1809418896	-2.0360171136	-0.3602408303
C	2.8124600031	-3.2149388544	-0.8435066445
C	4.1899349532	-3.2056527104	-1.13537456
H	4.639312441	-4.1205225476	-1.5054681368
C	4.9346733375	-2.0588480172	-0.9519273956
H	5.994594374	-2.0695714839	-1.1826058064
C	4.355408851	-0.8618239735	-0.4705313669
H	1.1999856669	2.3633978607	1.0938231819
H	6.2204433104	-0.9689747595	2.9588029903
O	6.1738446422	-0.584986236	2.062609542

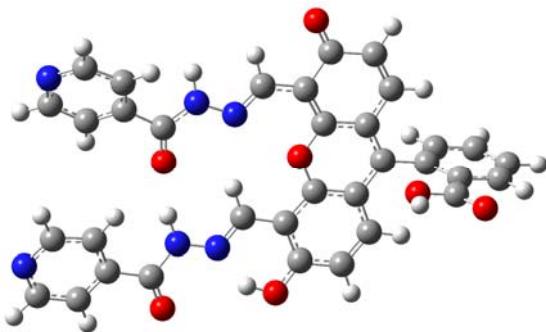


Figure S31. Optimized structure of **F2**.

B3LYP/6-31G(d) in gas phase, E = -2163.921766 a.u.

	charge = 0	Multiplicity = 1	
O	-6.8181430195	-0.7738262328	-2.6971447473
O	-1.9538569322	5.0482417718	0.4315923339
O	-1.2301534898	0.3841108378	0.3099839307
O	-0.251445666	-4.2661963279	0.2117141803
H	0.6827813431	-3.9107263648	0.1401960801
O	3.2637270128	0.7570466465	0.038850523
O	3.4899777672	-4.3590792974	-0.0148012291
N	0.8643400197	2.1586614337	0.1405786229
N	2.1093679661	2.7347624786	0.0499938201
H	2.1869327934	3.7481081887	0.1208214945
N	6.9563919481	4.1784258917	-0.2737774512
N	1.6086068309	-2.4587667297	0.072657438
N	2.9229094971	-2.1468533782	-0.0267275178
H	3.1940300602	-1.1629690942	0.0051568405
N	8.0249601587	-2.2418590526	-0.4980391711
C	-6.0086087531	-0.5653274261	-1.8162413383
C	-6.3396264075	-0.5966505934	-0.3576860934
C	-7.6842651256	-0.8670942017	-0.0586311187
H	-8.3614646459	-1.0304560828	-0.889758388
C	-8.1339882496	-0.9230904365	1.2555327702
H	-9.1785910762	-1.133869157	1.4642419015
C	-7.2335750435	-0.7055014574	2.2987128076
H	-7.567998581	-0.7443998958	3.3315711462
C	-5.8951797158	-0.4353700255	2.0167866085
H	-5.195064313	-0.2670299425	2.8298881207
C	-5.4241933046	-0.3748867363	0.6965998262
C	-3.9635638032	-0.0924189251	0.5178215814
C	-3.4887976059	1.2108118927	0.5075525981
C	-4.34169787	2.3753143256	0.5868929615
H	-5.4119036465	2.2156214387	0.6680102819

C	-3.8349459128	3.6284667226	0.5619740301
H	-4.4741370883	4.5037559013	0.6212710488
C	-2.3949691186	3.8920278415	0.4507888571
C	-1.5090874091	2.7101189586	0.3624198798
C	-2.071944605	1.4465268324	0.3924979775
C	-1.6684539866	-0.9018725103	0.337735951
C	-0.6725135802	-1.8892303639	0.257110987
C	-1.1001633298	-3.2453603408	0.2820802462
C	-2.4776112189	-3.5544278526	0.3874736279
H	-2.7579698712	-4.6020212126	0.4042251584
C	-3.4201329203	-2.5564266115	0.4659777533
H	-4.4709845535	-2.8114466789	0.5483742211
C	-3.0444238603	-1.1887694754	0.4431489059
C	-0.0998276486	3.0133682542	0.2516106651
H	0.0962854498	4.0926803182	0.2671571665
C	3.247201707	1.9886959514	-0.0075497629
C	4.5152728046	2.7873613844	-0.1145343207
C	5.6791818289	2.2700703313	0.4641331578
H	5.6531884255	1.317392156	0.981788847
C	6.8604888672	3.0031349988	0.3628232496
H	7.7767534106	2.628566153	0.8151502269
C	5.8429414644	4.6565715096	-0.8395911699
H	5.9414332925	5.6035635194	-1.3670957544
C	4.6062799005	4.0106391616	-0.7889990675
H	3.7522781328	4.4450777809	-1.3009986925
C	0.7326276095	-1.5135584559	0.1545781056
H	1.0020209482	-0.4556359072	0.1496065352
C	3.831299474	-3.1873138707	-0.0831145823
C	5.2745526023	-2.7892712687	-0.2315586186
C	6.2353345743	-3.7793177887	-0.0004632057
H	5.9158489515	-4.7771372592	0.2787781549
C	7.5828268353	-3.4557764647	-0.141004096
H	8.3482405436	-4.2083231648	0.0401544627
C	7.0974882802	-1.3054972877	-0.7289338342
H	7.469593085	-0.3268926814	-1.0284238767
C	5.7225349624	-1.5183814101	-0.6116133505
H	5.039967691	-0.7017922478	-0.821128439
O	-4.7147043349	-0.2807438763	-2.0872959122
H	-4.6373535512	-0.294887594	-3.0603323457

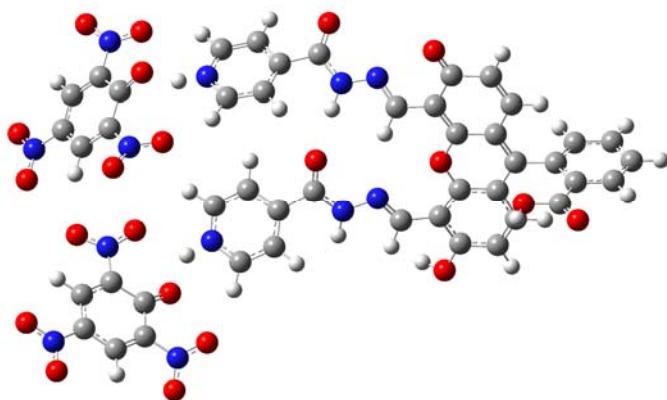


Figure S32. Optimized structure of **F2+TNP**.

B3LYP/6-31G(d) in gas phase, E = -4005.8135635 a.u.

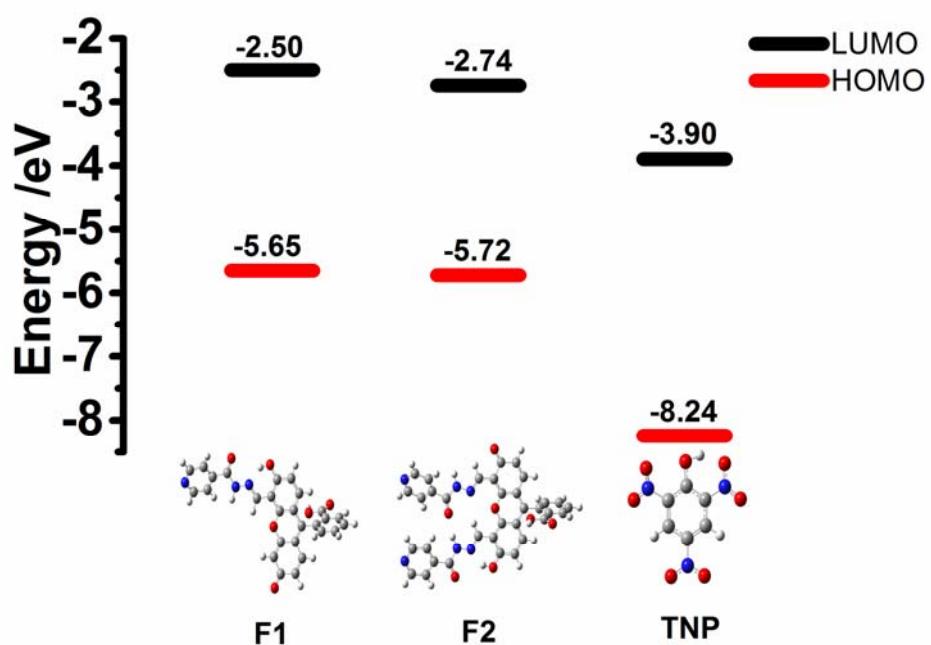
charge = 0 Multiplicity = 1

O	10.9515423038	3.3460182515	-1.8825476229
O	6.338567952	-3.5103034646	-2.8574445184
C	6.0422343095	-1.8246536469	-1.1815493955
C	4.1098024059	1.7216488385	1.7373158239
C	5.5552415464	1.7057394994	1.8913761561
C	6.4541400198	0.9267177702	1.117153171
C	4.5983970826	-1.9807897064	-1.0499494256
C	6.6930076234	-0.8405879277	-0.4789540569
C	1.1145537484	0.513513868	0.3579256968
C	8.894742578	-1.4950616061	-1.3768246352
C	7.8471074973	1.1331985561	1.1569839492
C	10.8975633158	1.503844871	-0.3542909638
C	8.3483334496	2.114976773	2.0341870399
C	6.8384128865	-2.6780706689	-2.1077761774
C	8.6917230814	0.3325663166	0.2954752145
C	8.1261573589	-0.6348590694	-0.5009142104
C	10.1766887144	0.5169400861	0.3583591564
C	10.3062561852	2.4947385145	-1.3060268538
C	6.1305550985	2.6281135913	2.8061248238
C	7.5097946001	2.8420927943	2.8578250323
C	-0.5943875245	2.417918661	0.5759469942
C	-0.2925343719	1.0515292984	0.481020885
C	-1.3416733888	0.1244946698	0.4582081152
C	-0.2028168636	-2.9815957489	-1.8212532853
C	8.2970756803	-2.448598967	-2.1186931819
C	1.9559749776	-4.2660895774	-1.2832575861
C	-2.6504527361	0.5693481789	0.5747949503
C	10.8855987859	-0.3475691338	1.2062975933
C	-1.9199914299	2.8114549857	0.6925460011

C	0.4514329758	-4.0914563101	-1.266929997
C	12.268906529	-0.2497292414	1.3528786563
C	12.2897463748	1.5903669309	-0.1965578552
C	-1.5881230108	-2.9343132322	-1.8033860244
C	12.9755615996	0.7251360621	0.6484857981
C	-1.6970627112	-5.0276531996	-0.7251897341
C	-0.3160075423	-5.137602469	-0.7376760453
O	-9.8478043527	4.7880606333	-2.1695684679
O	-3.4199572997	5.8112691337	0.2496445182
C	-7.7956535213	3.6823224043	-0.7709664071
O	-8.9131918057	6.7617381295	-2.1614319374
C	-6.7162689919	3.1696230277	-0.0667334
C	-7.8128824794	5.0297751843	-1.1142632618
O	-4.8166165628	7.4774067225	0.2571059484
C	-5.678954735	5.3642140005	-0.0779385179
C	-6.7560465101	5.8764367173	-0.7710032961
O	-7.8300632057	1.1652394527	0.1557004484
O	-5.6918316284	1.1717145979	0.5384436356
O	-4.6812744707	3.586931932	1.1410158378
C	-5.6012644716	3.9882052969	0.3810543312
O	-9.7930093306	-3.5392243256	2.6398376898
O	-4.7176169837	-4.4764052268	-0.8573710537
O	-6.0910341647	-0.8518557859	-1.6359056317
C	-8.0219539364	-3.1508639068	1.2174039945
C	-6.1754984496	-2.676943586	-0.2242784843
C	-7.5650146303	-4.4589597634	1.3972988863
C	-5.659067362	-4.0326102087	-0.1466808854
C	-6.412296557	-4.8732230357	0.7614338973
O	-4.2089951764	-1.7411736313	-0.9992416706
C	-7.3343466146	-2.2509973629	0.4063934364
O	-6.7830190861	-7.0641283714	1.3836259158
H	4.5749758281	2.9147583091	3.9205357704
H	1.7416196129	2.0461097676	1.5818282573
H	2.1876028265	-2.2261787987	-0.9106396663
H	3.6173756397	2.5625556257	2.246628691
H	-3.8837500617	2.2869272356	0.8408424075
H	4.0357002736	-1.117598413	-0.6815692403
H	9.9688595821	-1.349384157	-1.4172690275
H	9.4189093463	2.2801454475	2.0790787454
H	7.8944252783	3.5677476303	3.5653626778
H	0.1739995765	3.1819791563	0.5276943901
H	-1.1320651587	-0.9319484887	0.3491148424
H	0.3517107905	-2.1649655965	-2.2666814483
H	-3.3676882379	-3.9104921566	-1.1772066792

H	8.8618206773	-3.0927340241	-2.7853974048
H	-3.5117313416	-0.086861696	0.54245681
H	10.3375514401	-1.1079105874	1.755208306
H	-2.2640902656	3.8399195909	0.761832234
H	12.7891190646	-0.9362140726	2.0147703106
H	12.8139982402	2.3544992935	-0.7598749079
H	-2.1756374253	-2.1160553968	-2.2016567553
H	14.0529415068	0.8090686543	0.7542688471
H	-2.3749270656	-5.7637807645	-0.3050408523
H	0.1789662353	-6.0180992149	-0.345096536
H	-8.6158014039	3.0361793485	-1.0561035539
H	-6.7765357425	6.9190457668	-1.0613029709
H	-8.1056678368	-5.1398262453	2.0421174247
H	-7.6721714007	-1.2247222809	0.3031628902
O	-4.729230942	-6.4523493838	0.9977648678
O	-9.5971146202	-1.5548749562	1.7498195863
N	-5.9404974245	-6.2267082855	1.0535190265
N	-9.2224293728	-2.7162798503	1.9182191039
N	-5.460639143	-1.6989039865	-1.014463383
N	-6.7491421502	1.7494511217	0.2342876298
N	-8.9386404465	5.5637307277	-1.868789186
N	-4.5639724939	6.2747650269	0.174957877
N	3.3834580045	0.9240499364	1.0290783109
N	4.0159038611	-3.097614205	-1.3248657233
N	2.6561155688	-3.0882136608	-1.1979312609
N	2.0528627985	1.256953795	1.016674043
N	-2.8919073013	1.8832121619	0.706364533
N	-2.2877513102	-3.9369352897	-1.2485933428
O	5.9184251932	-0.0221082443	0.3219230829
O	5.3811469372	3.3845099223	3.6601295058
O	1.3436144933	-0.5130355799	-0.2693105565
O	2.4367873758	-5.3819673125	-1.3700825776
H	8.7261283548	3.063551231	-2.1276850828
O	8.9697978642	2.3705063226	-1.4845530297

22. HOMO and LUMO energy-level diagram of F1, F2 and TNP.



Firgue S33. HOMO and LUMO energy-level diagram of **F1**, **F2** and TNP.

23. The images of paper sensors absorbed different concentration of F1 and F2 under visible light and UV light.

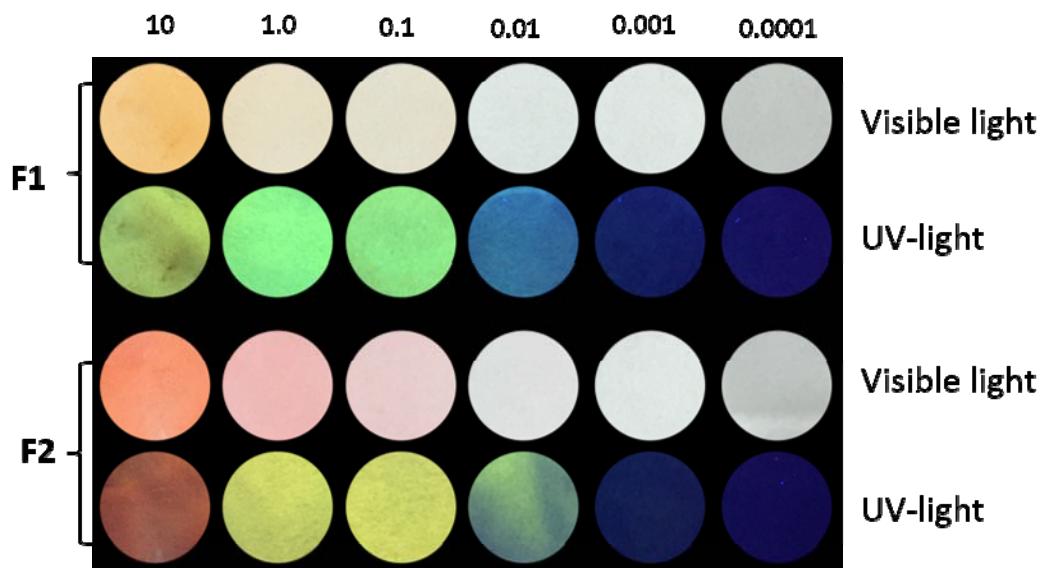


Figure S34. The images of paper sensors absorbed different concentration of **F1** and **F2** under visible light and UV light (365 nm). [TNP]= 10.0 mM, 1.0 mM, 1.0×10^{-1} mM, 1.0×10^{-2} mM, 1.0×10^{-3} mM, 1.0×10^{-4} mM.