

## Supporting Information

# Polymer-mediated metallophilic interactions for gram-scale production, high-yield (~90%) synthesis of ultrathin bismuth nanowires

Tzu-Lun Kao and Hsing-Yu Tuan\*

*Department of Chemical Engineering, National Tsing Hua University, Hsinchu,*

*Taiwan 300.*

*AUTHOR EMAIL ADDRESS: toms00280@hotmail.com*

*\* Corresponding author*

*Phone: (886)3-571-5131 ext. 42509*

*Email: hytuan@che.nthu.edu.*

## ***Materials***

All chemicals were used as received without any further purification, including bismuth(III) 2-ethylhexanoate ( $C_{24}H_{45}BiO_6$ ) from Alfa Aesar, oleylamine (OLA, 97%) from Acros Organics, trioctylamine (TOA, 98%) from Sigma-Aldrich, gold(III) chloride trihydrate ( $\geq 99.9\%$ ) from Sigma-Aldrich, and polyvinylpyrrolidone/hexadecane (PVP-HDE) from ISP. Toluene (ACS reagent, 99.5%) and ethanol (absolute, 99.8%) for the centrifugation or dispersion were also purchased from Sigma-Aldrich.

## ***Characterization***

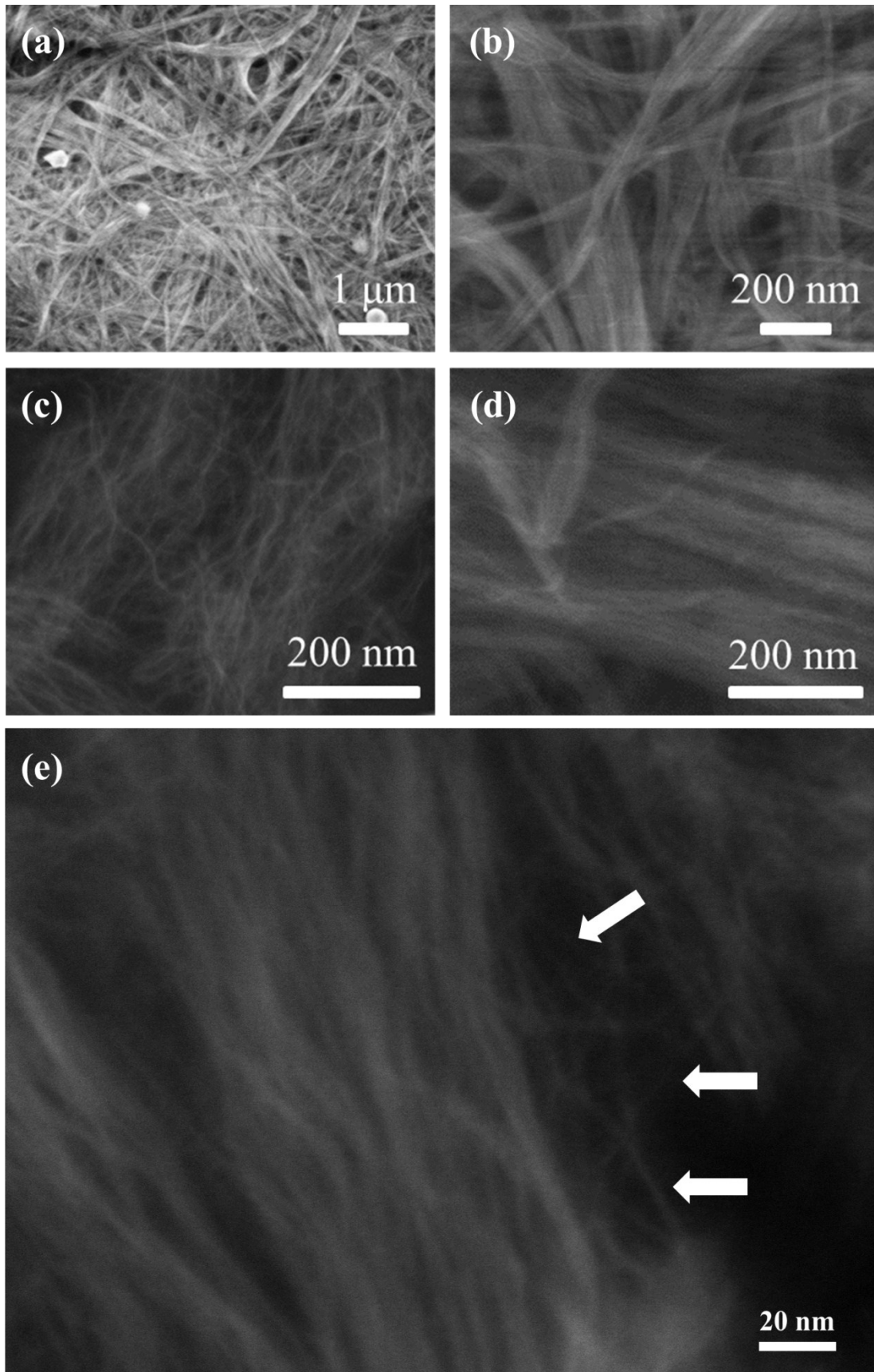
Ultrathin bismuth nanowires were characterized by the following instruments, including scanning electron microscope (SEM), transmission electron microscope (TEM), X-ray diffractometer (XRD), and X-ray photoelectron spectroscope (XPS). TEM images were obtained on JEOL JEM-1200 at an accelerating voltage of 120 kV for low resolution images and on a JEOL JEM-3000F at an accelerating voltage of 300 kV, respectively. SEM images were obtained on JSM-6500. XRD data were obtained by Rigaku Ultima IV X-ray diffractometer using a Cu radiation source ( $\lambda = 1.54 \text{ \AA}$ ). XPS analysis was performed on ULVAC-PHI XPS.

### ***Synthesis of ultrathin bismuth nanowire in heating-up style***

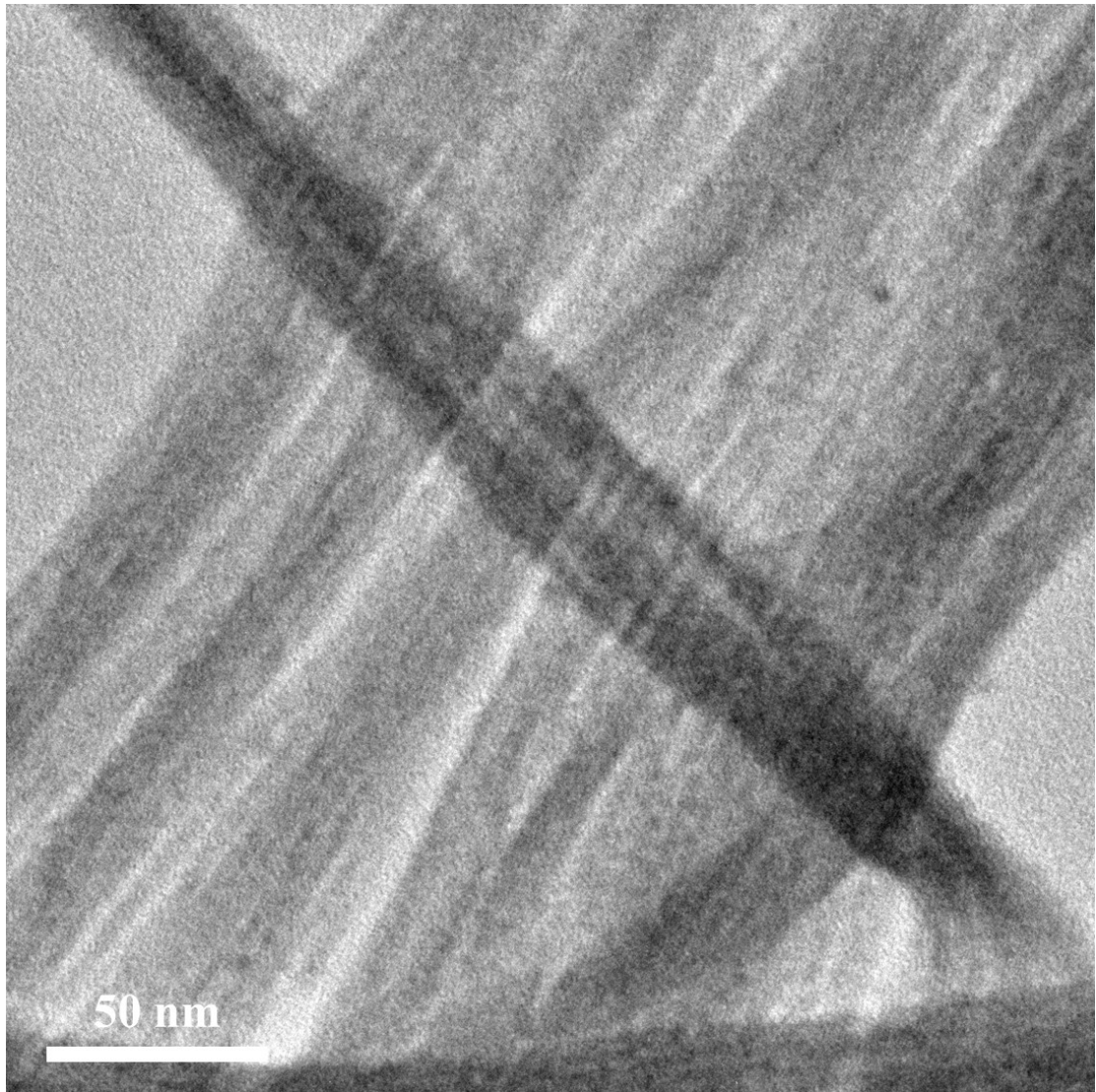
In a typical synthesis of ultrathin bismuth nanowires in heating-up style, 0.6386 g  $C_{24}H_{45}BiO_6$ , 2 g PVP-HDE, and 10 ml oleylamine (OLA) were added into a 50 ml three-necked flask inside a glovebox with both water and oxygen  $<0.1$  ppm, and the flask was then sealed with a stopcock valve at the middle neck for the input of argon and with a glass tube for the temperature sensor and a rubber septum at two side necks before taken out from the glovebox. The three-necked flask was placed on a heating mantle, connected to a Schlenk line system, and purged with continuous argon flow for 30 min to ensure an oxygen-free environment to prevent the oxidization of bismuth in the heating process. The mixture was then rapidly heated to 200 °C and kept at 200 °C for 30 min with vigorous stirring. The heating mantle was then removed, and the flask was cooled down with a cool water bath, followed by the addition of 10 ml toluene and 15 ml methanol to the flask. The product was washed by centrifugation at 6000 rpm for 5 min several times. The supernatant was discarded to obtain purified ultrathin bismuth nanowires, which was stored dry under argon prior to characterization. For the large-scale production of Bi ultrathin nanowires, all the reactants are 50 times in volume and gram as the recipe shown above with a 1 L three-necked flask.

### ***Gram-scale synthesis of ultrathin bismuth nanowire in heating-up style***

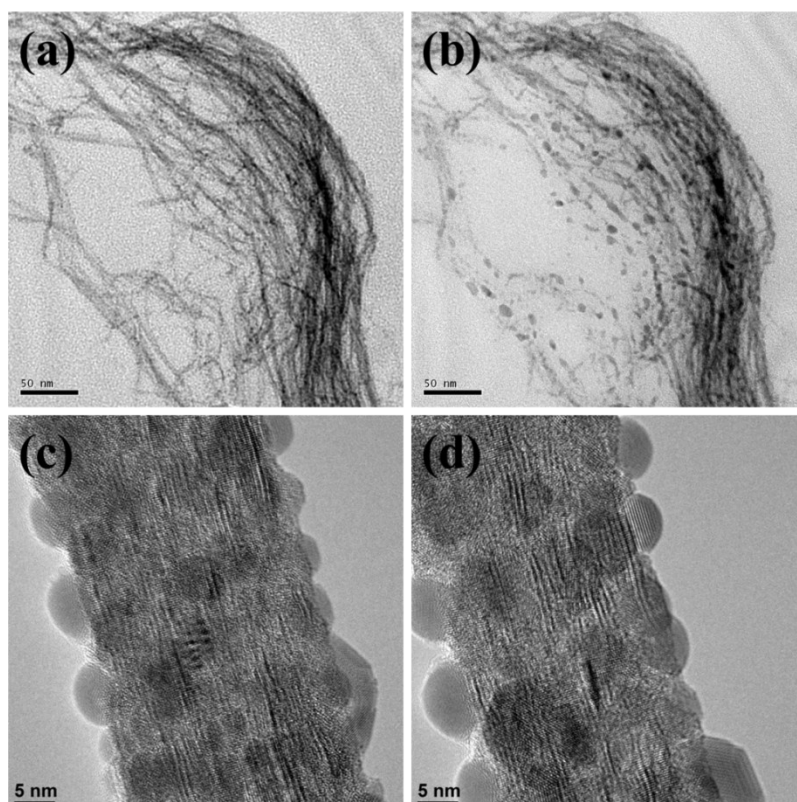
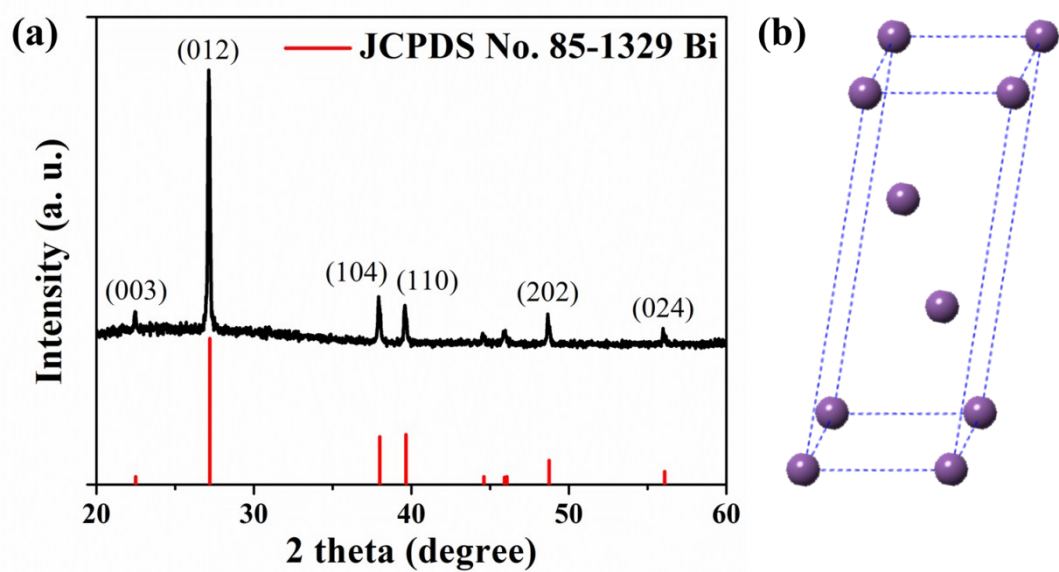
31.93 g  $C_{24}H_{45}BiO_6$ , 5 g PVP-HDE, and 500 ml oleylamine (OLA) were added into a 1 liter three-necked flask inside a glovebox with both water and oxygen  $<0.1$  ppm, and the flask was then sealed with a stopcock valve at the middle neck for the input of argon and with a glass tube for the temperature sensor and a rubber septum at two side necks before taken out from the glovebox. The three-necked flask was placed on a heating mantle, connected to a Schlenk line system, and purged with continuous argon flow for 30 min. The mixture was then rapidly heated to 200 °C and kept at 200 °C for 1 day with vigorous stirring. The heating mantle was then removed, and the flask was cooled down with a cool water bath. The resulting product was purified by centrifugation at 6000 rpm for 5 min several times. The supernatant was discarded to obtain purified ultrathin bismuth nanowires, which was stored dry under argon prior to characterization.

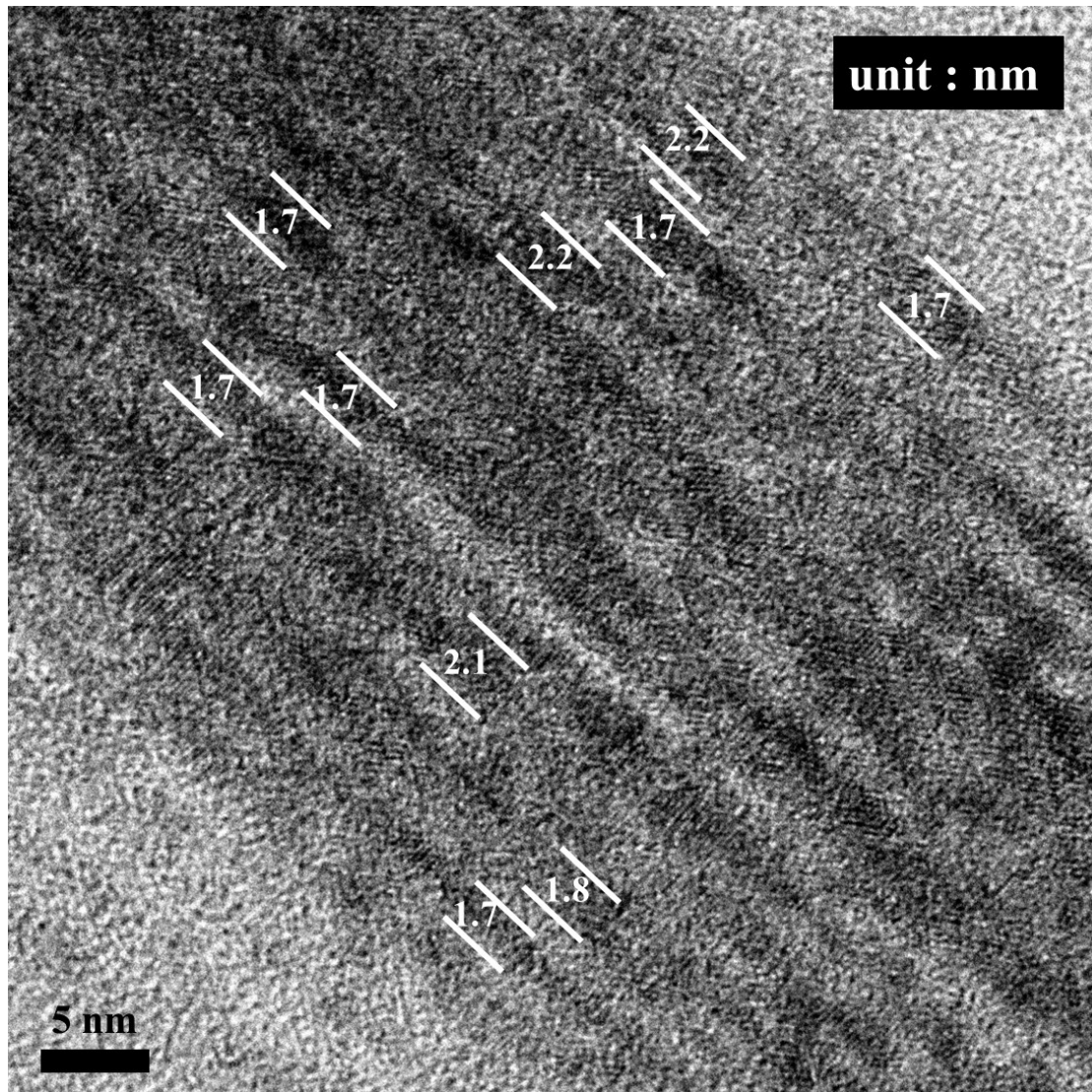


**Figure S1** (a-e) SEM images of as-synthesized Bi UNWs at various magnifications.



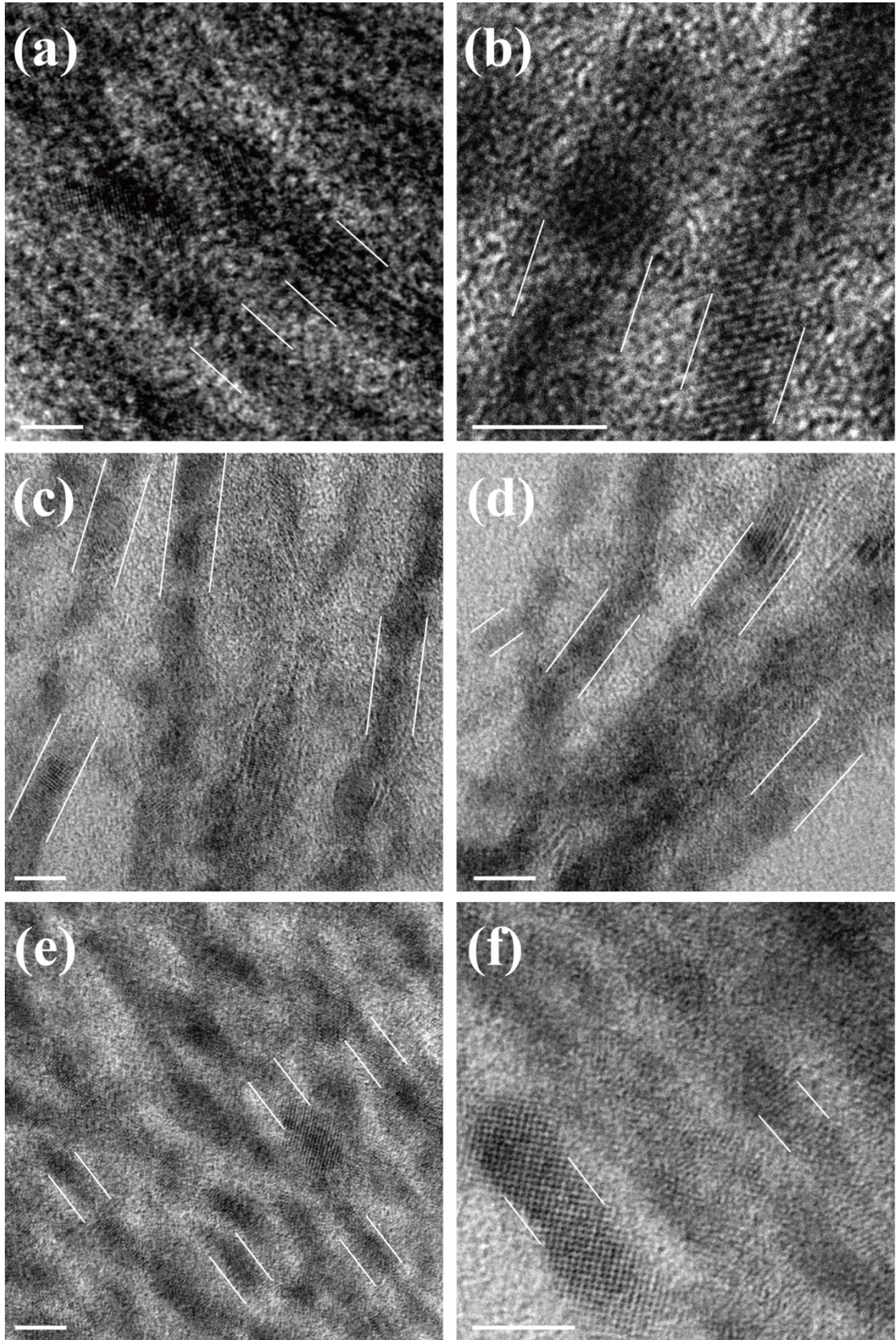
**Figure S2** TEM image of as-synthesized Bi UNWs.



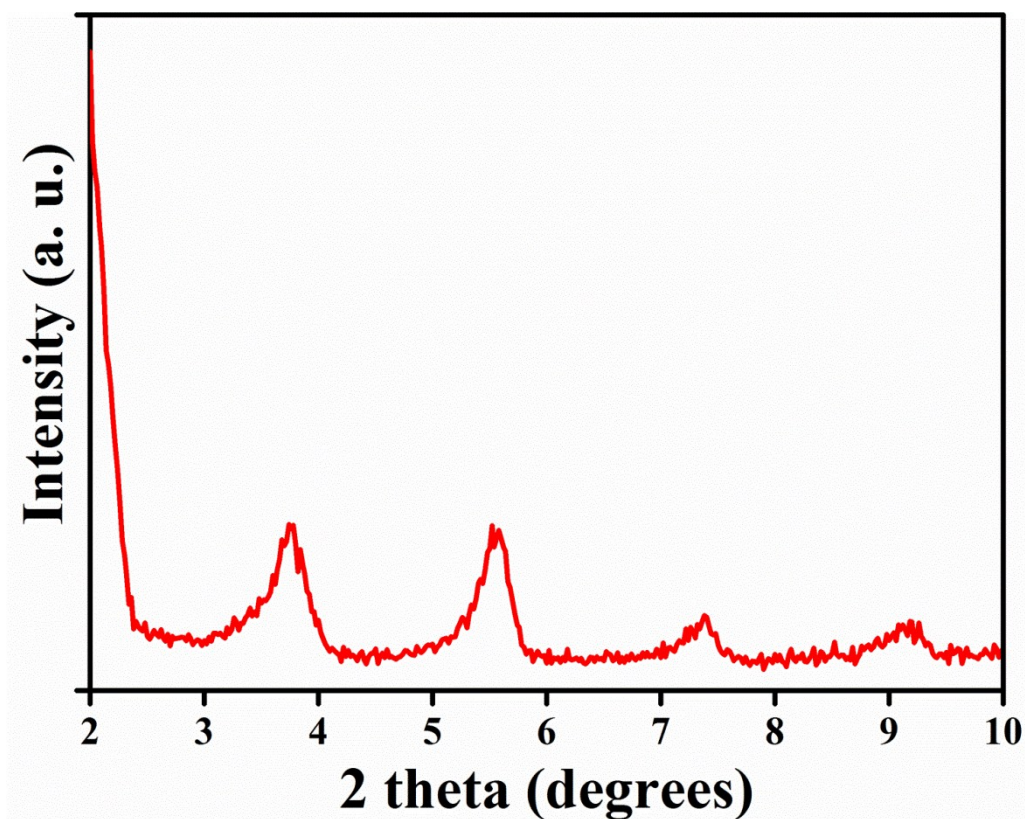


**Figure S5** TEM image of as-synthesized Bi UNWs with the diameter of some UNWs measured in nm.

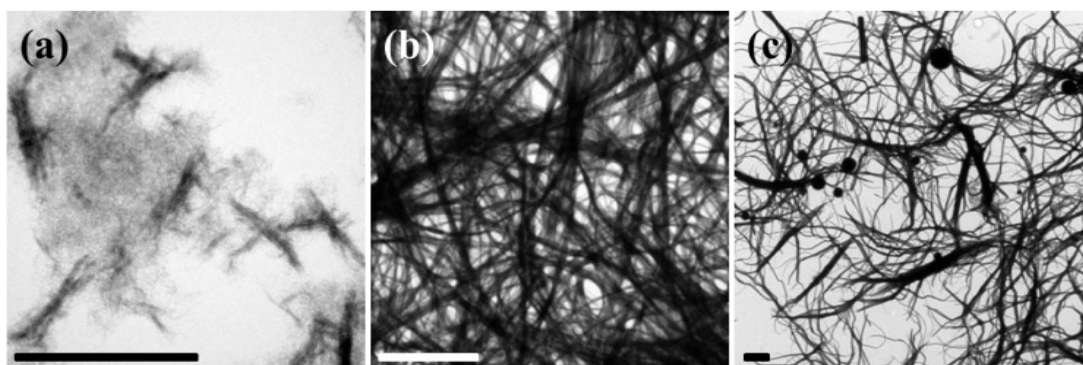




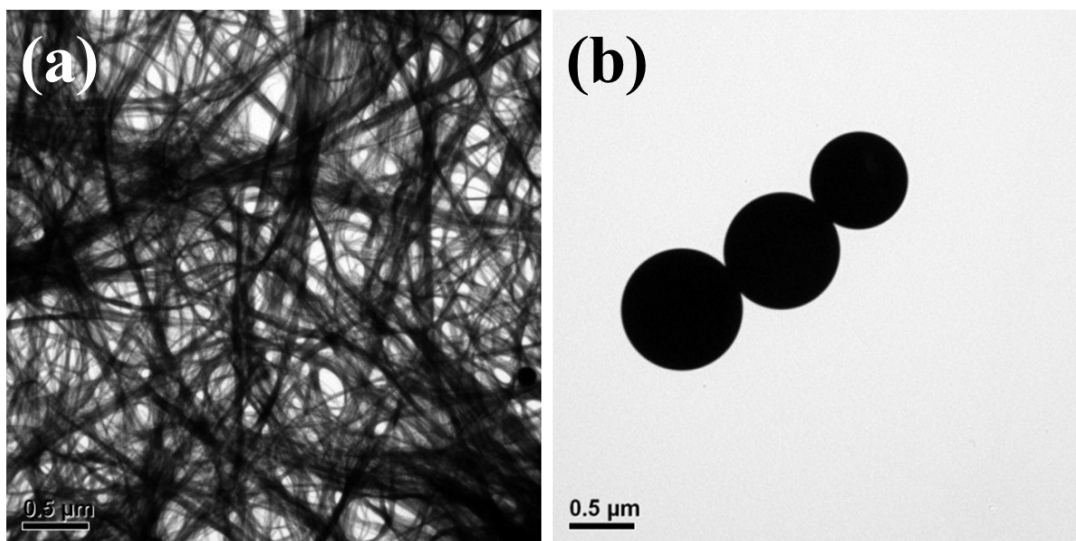
**Figure S6** (a-f) TEM images of as-synthesized Bi UNWs with the scale bar equivalent to 5 nm.



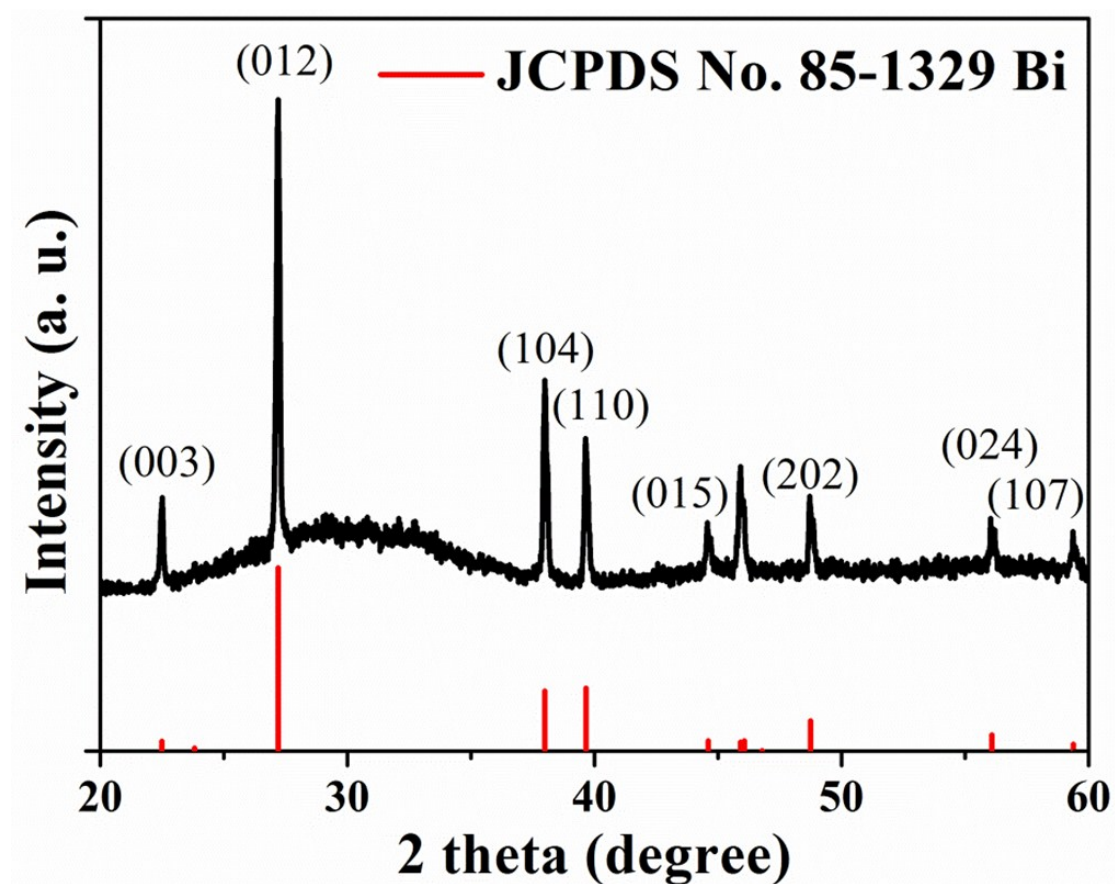
**Figure S7** XRD spectrum of Au<sup>+</sup> complex gel prepared by continuously stirring the mixture of 0.2 mmol gold(III) chloride trihydrate and 20 ml oleylamine at room temperature for 2 days.<sup>1</sup>



**Figure S8** TEM images with a scale bar of 1  $\mu\text{m}$  of the resulting product obtained from the synthesis (a) at room temperature for 1 day (b) at 200  $^{\circ}\text{C}$  for 30 min (c) at 250  $^{\circ}\text{C}$  for 30 min.



**Figure S9** TEM images of the corresponding product resulted from the synthesis using (a) oleylamine (b) trioctylamine as the reducing agent at 200 ° C for 30 minutes.



**Figure S10** XRD spectrum of the as-synthesized Bi UNWs in gram-scale synthesis.

## References

1. Z. Huo, C.-k. Tsung, W. Huang, X. Zhang, and P. Yang, *Nano Lett.*, 2008, **8**, 2041-2044.