Synthesis and shift-reagent-assisted full NMR assignment of the bacterial (Z_8, E_2, ω) -undecaprenol

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Electronic Supplementary Information

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Experimental Section

Synthesis. Synthesis of compound **3** from E,E-farnesol was published previously by our laboratory.¹ Undecaprenol **2** was purchased from American Radiolabeled Chemicals, Inc.

Compound 7. Phosphorus tribromide (34 μ L, 0.36 mmol) in dry ether (0.8 mL) was added dropwise to a mixture of heptaprenol 3^2 (0.4 g, 0.8 mmol) and pyridine (6 µL) in dry ether (2 mL) at ice-water temperature. The resulting solution was stirred for 2 h at the same temperature. The brown mixture was poured into a mixture of ice and water and then washed with hexanes. The separated organic layer was washed with a series of aqueous solutions (water, saturated NaHCO₃, and brine), dried over anhydrous MgSO₄, filtered, and concentrated under reduced pressure. The residue, the resultant bromide intermediate, was purified by column chromatography on silica gel. This intermediate (0.3 g, 0.5 mmol) was dissolved in acetonitrile:DMF:ether (1:1:1, 3 mL) and the solution was added dropwise to a suspension of sodium *p*-toluenesulfinate (0.18 g, 1.0 mmol) in DMF (5 mL) in an ice-water bath. The mixture was brought to room temperature over a few minutes and was allowed to stir for 6 h. The mixture was diluted with ethyl acetate and was washed sequentially with water and brine, dried over anhydrous MgSO₄, filtered, and the filtrate was concentrated under reduced pressure. Compound 7 was purified by column chromatography on silica gel (0.3 g, 59% calculated from **3**). ¹H NMR (500 MHz, CDCl₃) δ 1.60, 1.61, 1.65, 1.68, 1.73 (8 × s, 24H), 1.75 - 2.14 (m, 24H), 2.44 (s, 3H, $CH_3C_6H_4$ -), 3.78 (d, J = 8.0 Hz, 2H, -SO₂ CH_2 -), 4.95 (t, J = 6.7 Hz, 1H), 5.06-5.16 (m, 5H), 5.20 (t, J = 7.6 Hz, 1H), 7.33 (d, J = 8.0 Hz, 2H), 7.75 (d, J = 8.2 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 16.19 (q), 16.20 (q), 17.9 (q), 21.8 (q, CH₃C₆H₄-), 23.5 (q), 23.64 (q), 23.67 (q), 23.75 (q), 25.9 (q), 26.0 (t), 26.5 (t), 26.6 (t), 26.8 (t), 26.9 (t), 32.18 (t), 32.26 (t), 32.32 (t), 32.41 (t), 39.92 (t), 39.94 (t), 56.2 (t, -SO₂CH₂-), 111.2 (d, -SO₂CH₂CH=), 124.2 (d, -CH₂CH=),

124.3 (d, -CH₂CH=), 124.4 (d, -CH₂CH=), 124.6 (d, -CH₂CH=), 125.0 (d, -CH₂CH=), 125.1 (d, -CH₂CH=), 128.6 (d), 129.8 (d), 131.5 (s), 135.1 (s), 135.4 (s), 135.57 (s), 135.61 (s), 136.1 (s), 136.2 (s), 144.6 (s), 146.0 (s); HRMS (ESI/Q-TOF) m/z [M+Na]⁺ Calcd 655.4519 for C₄₂H₆₄O₂SNa; found 655.4528.

Compound 8. The sulfone 7 (300 mg, 470 µmol) was dissolved in a mixture of anhydrous THF and hexamethylphosphoramide (HMPA) (4:1, 5 mL) and the resulting solution was cooled to -78 °C. n-Butyllithium (0.4 mL, 640 µmol, 1.6 M in hexane) was added dropwise to this solution, which was stirring for 1 h at the same temperature. A solution of bromide 5^2 (430 mg, 700 µmol) in anhydrous THF-HMPA (4:1, 2 mL) was added dropwise to the sulfone solution and the mixture was allowed to stir for an additional 3 h at -78 °C. After warming the mixture to iced-water temperature, it was poured into a mixture of ice and water and the mixture was washed with hexanes-ether (1:1). The separated organic layer was washed with water and brine, dried over anhydrous MgSO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to afford compound 8 (470 mg, 85%). ¹H NMR (500 MHz, CDCl₃) δ 1.57 - 1.78 (m, 38H), 1.91 - 2.10 (m, 30H), 2.35 -2.70 (m, 4H), 2.41, 2.44 (2 × s, 6H, $-SO_2C_6H_4CH_3$), 3.79 - 3.87 (m, 2H, $-CHSO_2C_6H_4CH_3$), 3.98 (d, J = 6.6 Hz, 2H, -CH₂OBn), 4.49 (s, 2H, -OCH₂Ph), 4.83 - 5.02 (m, 4H), 5.05 - 5.19 (m, 6H), 5.41 (t, J = 6.8 Hz, 1H, -CHCH₂OBn), 7.26 - 7.36 (m, 9H), 7.67 - 7.74 (m, 4H); ¹³C NMR $(126 \text{ MHz}, \text{CDCl}_3) \delta 16.2 \text{ (q)}, 17.9 \text{ (q)}, 21.80, 21.84 \text{ (}2 \times \text{q}, CH_3C_6H_4\text{)}, 23.48 \text{ (q)}, 23.50 \text{ (q)},$ 23.52 (q), 23.61 (q), 23.65 (q), 23.68 (q), 23.77 (q), 23.79 (q), 25.8 (t), 25.9 (q), 26.0 (t), 26.4 (t), 26.5 (t), 26.75 (t), 26.78 (t), 26.81 (t), 26.9 (t), 30.2 (t), 30.5 (t), 32.12 (t), 32.17 (t), 32.27 (t), 32.33 (t), 32.4 (t), 39.91 (t), 39.93 (t), 63.3, 63.4 ($2 \times d$, CHTs), 66.5 (t, CH₂OBn), 72.3 (t, OCH₂Ph), 117.9 (d, -CH₂CH=), 118.0 (d, -CH₂CH=), 122.4 (d, BnOCH₂CH=), 124.25 (d, -

CH₂CH=), 124.27 (d, -CH₂CH=), 124.4 (d, -CH₂CH=), 124.5 (d, -CH₂CH=), 124.96 (d, -CH₂CH=), 125.01 (d, -CH₂CH=), 127.7 (d), 127.9 (d, -CH₂CH=), 128.0 (d), 128.3 (d, -CH₂CH=), 128.5 (d), 129.3 (d), 129.4 (d), 129.53 (d), 129.61 (d, ArH), 130.63 (s), 130.9 (s), 131.4 (s), 135.05 (s), 135.14 (s), 135.4 (s), 135.57 (s), 135.64 (s), 135.9 (s), 138.7 (s), 140.2 (s), 144.5 (s), 144.55 (s), 144.62 (s), 144.8 (s), 145.1 (s); HRMS (ESI/Q-TOF) m/z [M+Na]⁺ Calcd for C₇₆H₁₀₈O₅S₂Na 1187.7530; found 1187.7547.

Compound 1. Ethylamine (50 mL) was added to a flask containing lithium (100 mg, 8.6 mmol) at -78 °C and ether (10 mL) was added. After stirring for 10 min at -78 °C, an ethereal solution of compound **8** (470 mg, 400 µmol in 10 mL) was added dropwise to the blue solution over 20 min. After stirring for 40 min at -78 °C, the reaction was quenched by the addition of isoprene (5 mL) and MeOH (10 mL). After addition of saturated NH₄Cl, the reaction mixture was washed with ethyl acetate. The combined ethyl acetate layer was washed with brine, dried over anhydrous MgSO₄, filtered through a layer of silica gel, and the filtrate was concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to afford the desired product **1** (190 mg, 61%). The complete assignment of ¹H and ¹³C NMR is given in Table S1. HRMS (ESI/Q-TOF) m/z [M+Na]⁺ Calcd 789.6884 for C₃₅H₉₀ONa; found 789.6842.

NMR Spectroscopy. The structures of undecaprenol **1** and **2** [10 mg of each was dissolved in 600 μ L C₆D₆ (100%, 99.96 atom % D) containing 10 mg Eu(hfc)₃] were determined by interpretation of the 2D homonuclear DQF-COSY, TOCSY, ROESY and heteronuclear ¹H-¹³C HSQC, HSQC-TOCSY, HMBC NMR spectra. All NMR spectra were recorded at 25 °C on a four-channel Bruker AVANCE II spectrometer at field strength of 18.79 T using a 5-mm inverse triple-resonance (TCI) ¹H/¹³C/¹⁵N, z-axis PFG cryoprobe, and running the TopSpin 3.2, pl6

software. The above spectra were measured by employing ordinary pulse sequences in the Bruker pulse sequence library. To enhance spectral resolution, linear prediction and zero filling were applied to the time domain data. Also, the squared shifted sine weighting window functions were used before 2D Fourier transformation. The homonuclear spectra and the ¹H dimension in heteronuclear spectra were referenced to the residual solvent signal (C_6D_6 , δ_H 7.15 ppm). The ¹³C dimension in the heteronuclear spectra was referenced indirectly.² Resonance signals in the measured spectra exhibited slight downfield shifts as the samples stayed in the magnet longer. Consequently, Table S1 contains values of the proton and carbon chemical shifts for carbons with directly attached protons corresponding to the positions of the crosspeaks in the HSQC spectra. The chemical-shift values for the quaternary carbons then correspond to the positions of the crosspeaks in the HMBC spectra.

References

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	$1 + Eu(hfc)_3$				$2 + \mathrm{Eu}(\mathrm{hfc})_3$				
unit	13	³ C δ [ppm] ^{<i>c,d</i>}	¹ H δ [ppm] ^c	HMBC connectivities	unit		13 C $\delta [ppm]^{c,d}$	¹ H δ [ppm] ^c	HMBC connectivities
CH ₂ -1			14.37		CH ₂ -1			18.02	
CH-2	Ζ	129.09	11.50		CH-2	Ζ	129.51	10.35	
C-3		143.62			C-3		142.98		
CH ₂ -4	·	35.42	4.83	2, 3, 5, 6, 55	CH ₂ -4		34.90	4.01	2, 3, 5, 6, 55
CH ₂ -5		28.33	3.77	3, 4, 6, 7	CH ₂ -5		28.21	3.48	3, 4, 6, 7
CH-6	Ζ	126.54	6.70	4, 5, 7, 8, 54	CH-6	Ζ	126.33	6.41	4, 5, 8, 54
C-7		137.08			C-7		137.14		
CH ₂ -8	·	33.35	2.88	6, 7, 9, 10, 54	CH ₂ -8	·	33.27	2.74	6, 7, 9, 10, 54
CH ₂ -9		27.41	2.70	7, 8, 10, 11	CH ₂ -9		27.34	2.61	7, 8, 10, 11
CH-10	Ζ	126.02	5.71	8, 9, 11, 12, 53	CH-10	Ζ	125.93	5.65	8, 9, 11, 12, 53
C-11		135.83			C-11		136.02		
CH ₂ -12	÷	32.97	2.45	10, 11, 13, 14, 53	CH ₂ -12	· · ·	32.90	2.40	10, 11, 13, 53
CH ₂ -13		27.19	2.41	11, 12, 14, 15	CH ₂ -13		27.14	2.40	11, 12, 14, 15
CH-14	Ζ	125.83	5.47	12, 13, 15, 16, 52	CH-14	Ζ	125.82	5.45	12, 13, 52
C-15		135.58			C-15		135.83		
CH ₂ -16		32.85	2.33	14, 15, 17, 18, 52	CH2-16		32.83	2.29	14, 17, 18, 52
CH ₂ -17		27.09	2.32	, -, -, -, -	CH ₂ -17		27.10	2.30	15, 16, 18, 19
CH-18	Ζ	125.79	5.39	16, 17, 19, 20, 51	CH-18	Ζ	125.75	5.37	16, 17, 19, 20, 51
C-19		135.46		- , , , , , , - , -	C-19		135.73		- 1 - 1 - 1 - 1 -
CH ₂ -20		32.77	2.25	21, 51	CH ₂ -20		32.76	2.23	e
CH ₂ -21		27.07	2.26	19, 20, 22, 23	CH ₂ -21		27.07	2.25	е
CH-22	Z	125.73	5 34	20 21 23 24 50	CH-22	Z	125.72	5 33	20 21 24 50
C-23	2	135.45	5.51	20, 21, 23, 21, 30	C-23	2	135.69	0.00	20, 21, 21, 30
CH ₂ -24	· ·	32.72	2 23	25 50	CH ₂ -24	· ·	32.75	2 23	e
CH ₂ -25		27.03	2.23	25, 50	CH ₂ -25		27.04	2.25	е
CH-26	7	125.72	5 32	24 25 27 28 49	CH-26	7	125.65	5 31	24 25 28 49
C-27	L	135.43	5.52	24, 25, 27, 20, 49	C-27	L	135.76	5.51	24, 25, 26, 49
CH2-28		32.74	2 30	е	CH28		32.45	2 19	е
CH ₂ -20		27.03	2.30	е	CH ₂ -20 CH ₂ -29		27.14	2.19	е
CH-30	7	125.60	5 29	28 29 31 32 48	CH-30	F	124.76	5 33	28 29 32 48
C-31	L	135.00	5.27	20, 27, 51, 52, 40	C-31	Ľ	135.69	5.55	20, 27, 52, 40
CH ₂₋ 32		32.46	2.18	е	<u>CH-32</u>		40.30	2 12	е
CH ₂ -32		27.14	2.10	е	CH ₂ -32		27.21	2.12	е
CH-34	F	124.78	5 32	32 33 35 36 47	CH-34	F	124.85	5 31	32 33 36 47
C-35	L	135 38	5.52	52, 55, 55, 50, 47	C-35	L	135 38	5.51	52, 55, 50, 47
CH-36		40.30	2.12	34 35 37 47	CH36		40.30	2.12	e
CH ₂ -30		27.19	2.12	54, 55, 57, 47	CH ₂ -30		27.21	2.12	е
CH-38	F	124.87	5 29	36 37 39 40 46	CH-38	F	124.90	5 30	36 37 40 46
C-39	L	135.12	5.27	50, 57, 57, 40, 40	C-39	L	135.25	5.50	50, 57, 40, 40
CH-40		40.30	2 10	38 30 11 12 16	CH-40		40.30	2 11	е
$CH_{2}-40$		40.30 27.32	2.10	50, 57, 41, 42, 40	$CH_{2}-40$		27.30	2.11	е
CH-42	Ø	125.02	5.25	40 41 43 44 45	CH-42	Ø	125.03	5 25	40 41 44 45
C-43	ω	131.21	5.25	+0, +1, +3, +7, +3	C-43	ω	131.46	5.25	-0, -1,, -5
CH-44	Ø	17.82	1 58	12 13 15	<u>CH-44</u>	Ø	25.91	1 60	12 13 15
$CH_{3}-44$	<i>w</i>	25.92	1.50	42, 43, 43	CH ₃ -44	0	17.82	1.09	42, 43, 45
CH ₃ -45	E E	16.20	1.62	38 39 40	CH ₃ -45	E E	16.20	1.50	38 39 40
CH ₃ -40	F	16.19	1.64	34 35 36	CH ₂ -40	E	16.20	1.62	34 35 36
CH ₂ -48	7	23.73	1.04	30 31 32	CH-48	F	16.20	1.65	30 31 32
CH ₂ -49	7	23.75	1.75	26 27 28	CH-40	7	23.76	1.05	26 27 28
CH ₂ -50	7	23.80	1 79	20, 27, 20 22, 23, 24	CH ₂ -50	7	23.78	1.70	20, 27, 20 22, 23, 24
CH ₂ -51	7	23.82	1.82	18 19 20	CH ₂ -51	7	23.81	1.70	18 19 20
CH ₂ -52	7	23.82	1.86	14 15 16	CH ₂ -52	7	23.85	1.86	14, 15, 16
CH ₂ -53	7	23.98	1.00	10 11 12	CH ₂ -52	7	23.05	1.00	10 11 12
CH ₂ -54	7	24.19	2.19	678	CH ₂ -54	7	24.12	2.11	6 7 8
CH ₃ -55	Z	25.73	3.11	2, 3, 4	CH ₃ -55	Z	25.35	2.82	2, 3, 4

Table S1. NMR spectroscopic data for **1** (synthetic sample) and **2** (from *Magnolin kobus*) in C₆D₆ at 298 K (800.13 MHz).^{*a,b*}

^{*a*}C, CH-, CH₂-, and CH₃- are colored in white, blue, pink, and gray. ^{*b*}The numbers for **2** that are significantly different from those for **1** are in bold. ^{*c*}The proton and carbon chemical shift values for carbons with directly attached protons represent positions of the corresponding crosspeaks in the HSQC spectra. ^{*d*}The chemical shift values of quaternary carbons represent positions of the corresponding crosspeaks in the HMBC spectra. ^{*e*}The connectivities could not be unambiguously established due to strong signal overlaps.



TOCSY spectra of compound $1 + Eu(hfc)_3$





HSQC-TOCSY spectra of compound $1 + Eu(hfc)_3$



TOCSY spectra of compound $\mathbf{2} + Eu(hfc)_{3}$

















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