ELECTRONIC SUPPLEMENTARY INFORMATION

Zirconium arene triple-decker sandwich complexes: Synthesis, electronic structure and bonding

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1. Experimental details

1.1. General procedures and instrumentation

All manipulations were carried out using standard Schlenk techniques under N_2 ,¹ or in an MBraun UNIIab glovebox under N_2 . All glassware was dried at 160 °C overnight prior to use. Benzene and toluene were dried and degassed using an MBraun SPS-800 solvent purification system.² Dried solvents were collected, degassed under partial vacuum and stored over N_2 in K mirrored ampoules. THF was distilled from purple Na/benzophenone, degassed under partial vacuum and stored over activated 4 Å molecular sieves. Deuterated solvents (C₆D₆, C₇D₈) were purchased from Goss Scientific, degassed by three freeze-pump-thaw cycles, dried by refluxing over K for 3 days, vacuum distilled into ampoules and stored under N_2 .

Cumene, *o*-xylene and *m*-xylene were purchased from Sigma-Aldrich, stored over activated 4 Å molecular sieves and degassed by three freeze-pump-thaw cycles before use. Reference complexes Cp₂ZrCl₂ and Cp*₂ZrCl₂ were purchased from Sigma-Aldrich and used as received. The following compounds were synthesised according to the literature procedures: KC_{8} ,^{3,4} [ZrPn*(μ -Cl)_{3/2}]₂(μ -Cl)₂Li.THF_x,⁵ Pn*ZrCpCl,⁶ Cp₂Zr(CO)₂⁷ and Cp*₂Zr(CO)₂.⁷ The permethylpentalene ligand precursor, Li₂Pn*(TMEDA)_x,⁸ was kindly donated by Dr S. C. Binding.

Solution samples for NMR spectroscopy were prepared in the glovebox, using 5 mm J. Young tap NMR tubes. Spectra were recorded either on a Bruker Avance III HD nanobay 400 MHz NMR spectrometer, or a Bruker Avance III 500 MHz NMR spectrometer and were referenced internally to the residual protic solvent (¹H) or the signals of the solvent (¹³C). Spin simulations of ¹H NMR spectra were carried out using MestReNova⁹ and gNMR.¹⁰

Mass spectrometry measurements were carried out by Dr A. Abdul-Sada at the University of Sussex using a VG Autospec Fisons instrument (electron impact at 70 eV).

Elemental analyses for were carried out by Mr. S. Boyer at the Elemental Analysis Service, London Metropolitan University.

Electrochemical studies were carried out using a Princeton Applied Research AMETEK VersaSTAT 3 potentiostat under computer control. CV experiments were performed in a Saffron Omega Scientific glovebox under N₂ using a three-electrode configuration with a Au disc (2.0 mm²) or glassy carbon disc (7.0 mm²) as the working electrode, a Pt wire as the counter electrode and a Ag wire as the pseudo-reference electrode. Sample solutions were prepared by dissolving the analyte (*ca*. 5 mM) in THF (5.0 cm³) followed by addition of a supporting electrolyte [^{*n*}Bu₄N][PF₆]. The reported mid-peak potentials are referenced internally to that of the FeCp₂^{+/0} redox couple, which was measured by adding ferrocene (*ca*. 1 mg) to the sample solution.

Single crystal X-ray diffraction data for 1–5 were collected at Chemical Crystallography (University of Oxford) on a Bruker-Nonius Kappa CCD area detector diffractometer with a sealed-tube source ($\lambda_{Mo K\alpha}$), in ω scanning mode with ψ and ω scans to fill the Ewald sphere. The data were collected at 150 K using an Oxford Cryosystems low temperature device. Data were processed using Collect,¹¹ Scalepack, and Denzo,¹² and unit cell parameters were refined against all data. Absorption was corrected for by Multi-Scan methods from symmetry-related measurements using *SORTAV*.¹³ The structure was solved using SUPERFLIP¹⁴ and refined on F_o² by full-

matrix least-squares refinements using SHELXL-2013.¹⁵ Solutions and refinements were performed using the OLEX2¹⁶ and WinGX¹⁷ software packages. All non-hydrogen atoms were refined with anisotropic displacement parameters. Hydrogen atoms treated by a mixture of independent and constrained refinement.

Zr K-edge (17.998 keV) X-ray Absorption Near-Edge Structure (XANES) spectroscopy data were collected on the B18 beamline of the Diamond Light Source, Didcot, UK, with the assistance of beamline scientists Dr G. Cibin, Dr S. Parry and Mr D. Gianolio.

Measurements were performed using a QEXAFS set up with a fast-scanning Si (111) double crystal monochromator. The resolution of the spectra reported herein was 186.74 s/spectrum ($k_{max} = 18$, step size 1.0 eV), on average 10 scans were acquired to improve the signal to noise level of the data for transmission measurements.

Complex samples (*ca.* 15 mg) were mixed with dry boron nitride (*ca.* 65 mg) pressed into a cylindrical pellet of 8 mm diameter with a thickness (*ca.* 4 mm) chosen to give a total absorbance (μ x) of about 2.0. Pellets were positioned vertically in a Nalgene vial under Ar sealed with Parafilm, before being measured in transmission mode using ion chamber detectors. All XANES spectra were acquired concurrently with a Zr foil placed between I_t and I_{ref} . XANES data processing was performed using IFEFFIT¹⁸ with the Horae package (Athena).¹⁹ The edge energy in the Zr K-edge energy was determined from the first peak of the 1st derivative of the XANES, which was fit with a Gaussian curve using OriginPro2017 software.¹⁹

1.2. Synthesis and characterisation of $[(\eta^8 - Pn^*)Zr]_2(\mu:\eta^6,\eta^6 - C_6H_6)$ (1).

To an ampule charged with $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x}$ (331 mg, 0.405 mmol) in benzene (10 mL) was added KC₈ (241 mg, 1.78 mmol). A color change from light brown to dark red was observed and the mixture was allowed to stir for 48 h at room temperature. The resultant suspension was filtered and dried *in vacuo*. The red filtrate was concentrated to *ca*. 1 mL and after cooling to 5 °C, deposited dark red crystals that were isolated by decantation and dried *in vacuo*. Total yield: 84 mg (33% with respect to $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x})$).

¹H NMR (C₆D₆, 400.2 MHz, 298 K): $\delta_{\rm H}$ 2.82 (6H, s, C₆H₆), 2.19 (24H, s, Pn* NWT CH₃), 1.58 (12H, s, Pn* WT CH₃). ¹³C{¹H} NMR (C₆D₆, 100.6 MHz, 298 K): $\delta_{\rm C}$ 133.06 (Pn* ring *C*), 132.39 (Pn* ring *C*), 68.13 (C₆H₆), 12.46 (Pn* NWT CH₃), 10.21 (Pn* WT CH₃), Pn* bridgehead *C* signals were not observed. ¹³C NMR (C₆D₆, 100.6 MHz, 298 K): $\delta_{\rm C}$ 68.10 (d, ¹J_{CH} = 171.15 Hz, C₆H₆). ²H NMR (C₆H₆, 61.43 MHz, 298 K): (for **1-**d₆) $\delta_{\rm D}$ 2.79 (C₆D₆). EI-MS: *m*/*z* = 627–637 (principal peak 629, 60%), [M]⁺; 462–465 (principal peak 462, 30%), [M – Zr – C₆H₆]⁺. Anal. found (calcd. for C₃₄H₄₂Zr₂): C, 64.63 (64.50); H, 6.72 (6.69) %.

1.3. Synthesis and characterisation of $[(\eta^8 - Pn^*)Zr]_2(\mu:\eta^6,\eta^6 - C_6H_5CH_3)$ (2).

To an ampule charged with $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x}$ (297 mg, 0.363 mmol) in toluene (20 mL) was added KC₈ (216 mg, 1.60 mmol). A color change from light brown to dark red was observed and the mixture was allowed to stir for 48 h at room temperature. The resultant suspension was filtered and dried *in vacuo*. The red filtrate was concentrated to *ca*. 1 mL and after cooling to -35 °C, deposited dark red crystals that were isolated by decantation and dried *in vacuo*. Total yield: 71 mg (30% with respect to $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x})$).

¹H NMR (C₆D₆, 400.2 MHz, 298 K): $\delta_{\rm H}$ 3.13 (1H, m, Ph *p*-*H*), 2.58 (4H, apparent d, Ph *o/m-H*), 2.18 (24H, s, Pn* NWT CH₃), 1.59 (12H, s, Pn* WT CH₃), 1.49 (3H, s, Ph-CH₃). ¹³C {¹H} NMR (C₆D₆, 100.6 MHz, 298 K): $\delta_{\rm C}$ 132.75 (Pn* NWT C), 132.10 (Pn* WT C), 77.65 (Ph *i*-C), 72.48 (Ph *o/m*-CH), 68.90 (Ph *o/m*-CH), 68.57 (Ph *p*-CH), 19.42 (Ph-CH₃), 12.39 (Pn* NWT CH₃), 10.25 (Pn* WT CH₃). ²H NMR (C₆H₆, 61.43 MHz, 298 K): (for **2-d**₈) $\delta_{\rm D}$ 3.51 (Ph *p*-D), 2.52 (Ph *o/m*-D), 1.40 (Ph-CD₃). EI-MS: *m/z* = 637–644 (principal peak 641, 25%), [M – 2H]⁺; 602–610 (principal peak 604, 30%), [M – H – C₃H₃]⁺; 587–594 (principal peak 590, 20%), [M – H – C₄H₅]⁺. Anal. found (calcd. for C₃₅H₄₄Zr₂): C, 64.96 (64.68); H, 6.85 (6.99) %.

1.4. Synthesis and characterisation of $[(\eta^8-Pn^*)Zr]_2(\mu:\eta^6,\eta^6-C_6H_5)^iPr)$ (3).

To an ampoule charged with $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x}$ (100 mg, 0.122 mmol) in cumene (1 mL) was added KC₈ (132 mg, 0.978 mmol). A colour change from light brown to dark red was observed and the mixture was allowed to stir for 48 h at room temperature. The resultant suspension was filtered and dried *in vacuo*. The red residue was dissolved in benzene (*ca.* 1 mL) and concentrated by slow evaporation at room temperature to afford the product as dark red crystals that were isolated by decantation and dried *in vacuo*. Total yield: 29 mg (35% with respect to $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x})$).

¹H NMR (C₆D₆, 400.2 MHz, 298 K, selected data): $\delta_{\rm H}$ 3.07 (1H, m, Ph *p*-*H*), 2.88 (4H, m, Ph *o/m*-*H*), 2.33 (1H, m, ⁱPr-C*H*), 2.20 (24H, s, Pn* NWT CH₃), 1.59 (12H, s, Pn* WT CH₃), 1.30 (1H, br s, ⁱPr-CH₃), 0.92 (1H, br s, ⁱPr-CH₃). ¹³C {¹H} NMR (C₆D₆, 100.6 MHz, 298 K, selected data): $\delta_{\rm C}$ 132.40 (Pn* WT C), 132.16 (Pn* NWT C), 100.69 (Pn* bridgehead C), 99.93 (Pn* bridgehead C), 94.03 (Ph *i*-C), 70.19 (Ph *o/m*-CH), 69.10 (Ph *p*-CH), 67.15 (Ph *o/m*-CH), 23.14 (ⁱPr CH), 14.39 (ⁱPr CH₃), 13.63 (ⁱPr CH₃), 12.80 (Pn* NWT CH₃), 10.18 (Pn* WT CH₃). EI-MS: *m/z* = 668–678 (principal peak 672, 35%), [M]⁺; 462–466 (principal peak 462, 55%), [M – Zr – PhⁱPr]⁺.

1.5. Synthesis and characterisation of $[(\eta^8 - Pn^*)Zr]_2(\mu:\eta^6,\eta^6 - C_6H_4\{1,2-CH_3\}_2)$ (4).

To an ampoule charged with $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x}$ (100 mg, 0.122 mmol) in *o*xylene (1 mL) was added KC₈ (132 mg, 0.976 mmol). The mixture was allowed to stir for 48 h at room temperature, and then filtered. The filtrate was stripped to dryness and the red residue dissolved in benzene (5 mL). This solution was frozen at -78 °C, exposed to dynamic vacuum, then removed from the cold bath to allow the benzene to sublime over 3 h to afford the product as a red powder. Single crystals suitable for X-ray diffraction were grown from slow evaporation at ambient temperature of a saturated *o*-xylene solution. Total yield: 15 mg (31% with respect to $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}Li.THF_{x})$).

¹H NMR (C₆D₆, 400.2 MHz, 298 K, selected data): $\delta_{\rm H}$ 2.99 (2H, m, Ar-*H*), 2.17 (24 H, br s, Pn* NWT CH₃), 2.99 (2H, m, Ar-*H*), 1.59 (12 H, s, Pn* WT CH₃), 1.51 (6H, s, Ar-CH₃). ¹³C{¹H} NMR (C₆D₆, 400.2 MHz, 298 K, selected data): $\delta_{\rm C}$ 132.46 (Pn* ring *C*), 131.85 (Pn* ring *C*), 73.91 (Ar-CH), 69.97 (Ar-CH), 17.45 (Ar-CH₃), 12.32 (br s, Pn* NWT CH₃), 10.34 (Pn* WT CH₃). EI-MS: m/z = 627-637 (principal peak 629, 60%), [M]⁺; 462-465 (principal peak 462, 30%), [M – Zr – C₆H₆]⁺.

1.6. Synthesis and characterisation of $[(\eta^8 - Pn^*)Zr]_2(\mu:\eta^6,\eta^6 - C_6H_4\{1,3-CH_3\}_2)$ (5).

To an ampoule charged with $[\eta^8 - Pn^*Zr(\mu-Cl)_{3/2}]_2Li$. THF_x (113 mg, 0.138 mmol) in mxylene (2 mL) was added KC₈ (130 mg, 0.962 mmol) and the mixture was allowed to stir for 48 h at room temperature. The mixture was allowed to stir for 48 h at room temperature, and then filtered. The filtrate was stripped to dryness and the red residue dissolved in benzene (5 mL). This solution was frozen at -78 °C, exposed to dynamic vacuum, then removed from the cold bath to allow the benzene to sublime over 3 h to afford the product as a red powder. Single crystals suitable for X-ray diffraction were grown from slow evaporation at ambient temperature of a saturated *m*-xylene solution. Total yield: 23 mg (25% with respect to $[\eta^8-Pn^*Zr(\mu-Cl)_{3/2}]_2$ -Li.THF_x)). ¹H NMR (C₆D₆, 400.2 MHz, 298 K, selected data): $\delta_{\rm H}$ 3.13 (2H, m, Ar-H), 2.17 (24 H, br s, Pn* NWT CH₃), 2.04 (1H, m, Ar-H), 1.80 (1H, br s, Ar-H), 1.59 (12 H, s, Pn* WT CH₃), 1.44 (6H, s, Ar-CH₃). ¹³C{¹H} NMR (C₆D₆, 400.2 MHz, 298 K, selected data): δ_C 133.32 (Pn* ring C), 131.79 (Pn* ring C), 100.71 (Ar-C), 98.83 (Pn* ring C), 77.05 (Ar-CH), 73.06 (Ar-CH), 70.12 (Ar-CH), 19.16 (Ar-CH₃), 12.31 (br s, Pn* NWT CH₃), 10.32 (Pn* WT CH₃). EI-MS: m/z = 657-667 (principal peak 662, 30%), $[M]^+$; 462-465 (principal peak 464, 30%), $[M - Zr - C_6H_2Me_2]^+$.

Note: Compounds 1–5 are highly air and moisture sensitive and solid state samples were unstable to prolonged (*ca.* 10 h) dynamic vacuum. 1–5 are very soluble in aromatic hydrocarbon solvents, each giving dark red solutions that were stable for *ca.* 7 days. However, aliphatic hydrocarbon (methylcyclohexane- d_{14}) and ethereal (tetrahydrofuran- d_8) solutions of 1–2 showed decomposition by ¹H NMR spectroscopy after *ca.* 24 h.

2. Additional characterising data

2.1. Cyclic voltammetry data



Figure S1 CV scans (2 cycles) of $[\eta^8$ -Pn*Zr(μ -Cl)_{3/2}]₂(μ -Cl)₂Li·THF_x in THF/0.1 M [ⁿBu₄N][PF₆], scan rate 100 mV s⁻¹.



Figure S2 CV scans (1 cycle) of 1 in THF/0.1 M [$^{n}Bu_{4}N$][PF₆], scan rate 100 mV s⁻¹.

Table S1 Peak potentials (E_p vs FeCp₂^{+/0}) and limiting currents (i_p) for CV scans in THF / 0.1 M [n Bu₄N][PF₆] at scan rate 100 mV s⁻¹.

Complex	$E_{\rm pa}/{ m V}$	$E_{\rm pc}/{ m V}$	$\Delta E_{\rm pp} / {\rm mV}$	i _{pa} /μA	<i>i</i> _{pc} / μA	$ i_{\rm pa}/i_{\rm pc} $
$[Pn*_2Zr_2Cl_5]Li \cdot THF_x$	-2.444	-2.627	183	1.019	-3.871	0.263
1	-1.335	n/a	n/a	7.660	n/a	n/a





Figure S4 ²H NMR spectrum (C₆H₆, 61.43 MHz, 298 K) of 1-*d*₆.



Figure S5 ${}^{13}C{}^{1}H$ NMR spectrum (C₆D₆, 125.7 MHz, 298 K) of 1.



Figure S6 Selected region of 13 C NMR spectrum (C₆D₆, 100.6 MHz, 298 K) of 1.



Figure S7 ^{13}C - ^{1}H HSQC NMR spectrum (C₆D₆, 298 K) of 1.



Figure S8 ${}^{13}C-{}^{1}H$ HMBC NMR spectrum (C₆D₆, 298 K) of 1.



Figure S9 Selected region of variable-temperature ¹H NMR spectra (C₇D₈) of 1 from 183 to 298 K. Asterisk denotes residual protio solvent.



Figure S10 Eyring plot of the exchange rate for the Pn* ligand twisting process in 1, as determined from VT ¹H NMR spectra in C₇D₈.

Table S2 Activation parameters for Pn* ligand twisting process in 1, calculated fromfitting the exchange rate for VT ¹H NMR spectra in C_7D_8 .^aCompound $\Delta H^{\ddagger} / kJ mol^{-1} \Delta S^{\ddagger} / J K^{-1} mol^{-1} \Delta G^{\ddagger}_{285} / kJ mol^{-1}$

1	37(2)	-65(9)	56(3)			
^{<i>a</i>} Errors in parentheses.						

 $[(\eta^{8}-Pn^{*})Zr]_{2}(\mu:\eta^{6},\eta^{6}-C_{6}H_{5}CH_{3})(2).$



Figure S12 Selected region of the ¹H NMR spectrum (C_6D_6 , 400.2 MHz, 298 K) of 2: a) experimental spectrum; b) AA'BB'X spectrum simulated using typical aromatic J coupling values.²⁰



Figure S13 Selected region of the ${}^{1}H-{}^{1}H$ COSY NMR spectrum (C₆D₆, 298 K) of 2.



Figure S14 ${}^{13}C{}^{1}H$ NMR spectrum (C₆D₆, 100.6 MHz, 298 K) of 2.



Figure S15 $^{13}C^{-1}H$ HSQC NMR spectrum (C₆D₆, 298 K) of 2.



Figure S16 ^{13}C - ^{1}H HMBC NMR spectrum (C₆D₆, 298 K) of 2.



Figure S17 Thermal displacement ellipsoid drawing (50% probability) of **1**. Hydrogen atoms are omitted for clarity. Primed atoms are generated by symmetry.



Figure S18 Thermal displacement ellipsoid drawing (50% probability) of **2**. Hydrogen atoms are omitted for clarity. Primed atoms are generated by symmetry.



Figure S19 Thermal displacement ellipsoid drawing (50% probability) of **3**. Hydrogen atoms are omitted for clarity.



Figure S20 Thermal displacement ellipsoid drawing (50% probability) of **4**. Hydrogen atoms are omitted for clarity.



Figure S21 Thermal displacement ellipsoid drawing (50% probability) of 5. Hydrogen atoms are omitted for clarity.



Figure S22 Pentalene numbering scheme and definition of structural parameters. The Pn* fold angle (FA) is defined as the angle between the wing-tip carbons (C2, C6) and the bridgehead (C4, C8) centroid. The twist angle (θ) for inverted sandwich structures is defined as the average torsion angle between the two opposite bridgehead carbon atoms and two zirconium atoms.

Parameter	1	2	3	4	5
Zr–Zr	3.8437(6)	3.84765(12)	3.8648(6)	3.8478(4)	3.8427(4)
Zr-Ct _{arene}	1.9219(3)	1.90667(6)	1.9328(5)	1.9241(3)	1.9216(17)
Zr-Carene	2.4013	2.4043	2.419(4)	2.4168(3)	2.4140(3)
C–C _{arene}	1.4640(20)	1.4680(17)	1.4629(26)	1.4713(16)	1.4701(21)
Puckering Q_{arene}	0.226(4)	0.210(3)	0.220(5)	0.241(3)	0.236(4)
Puckering φ_{arene}	330.0(10)	17.0(10)	148.4(13)	29.2(7)	331.4(10)
Pn* twist angle	45.3(4)	47.199(2)	49.6(8)	47.5597(8)	47.8(8)
Pn* fold angle	33.08(13)	31.2133(3)	31.58(16)	32.7895(6)	33.3(2)

Table S3 Selected mean distances (Å), angles (°) and parameters (defined in Fig. S22) for molecular structures 1–5 determined by X-ray crystallography. Ct denotes the η⁶-centroid of the bridging arene ring.

Complex	1	2	3	4	5		
Crystal data							
CCDC	1573277	1573278	1573279	1573280	1573281		
Chemical formula	$C_{34}H_{42}Zr_2$	$C_{35}H_{44}Zr_2$	$C_{37}H_{48}Zr_2$	$C_{36}H_{46}Zr_2$	$C_{36}H_{46}Zr_2$		
$M_{ m r}$	633.11	647.14	675.19	661.17	661.17		
Crystal system, space group	Orthorhombic, Pbnb	Orthorhombic, Pbnb	Triclinic, P^{-1}	Triclinic, P ⁻¹	Monoclinic, $P2_1/c$		
<i>a</i> , <i>b</i> , <i>c</i> (Å)	13.4361(3), 14.4172(4), 14.9673(4)	13.1736(5), 14.3455(6), 15.6258(8)	9.4522(1), 15.4296(2), 23.2350(3)	13.3867(2), 14.8682(2), 16.7564(2)	14.4103(3), 13.6833(3), 16.5470(4)		
α, β, γ (°)	90, 90, 90	90, 90, 90	89.097(1), 88.321(1), 72.418(1)	96.391(1), 109.561(1), 90.021(1)	90, 107.418(1), 90		
$V(Å^3)$	2899.33(13)	2953.0(2)	3228.91(7)	3120.59(8)	3113.14(12)		
Ζ	4	4	4	4	4		
μ (mm ⁻¹)	0.74	0.73	0.67	0.69	0.69		
Crystal size (mm)	$0.20\times0.20\times0.10$	$0.1 \times 0.1 \times 0.05$	$0.24 \times 0.20 \times 0.10$	$0.30 \times 0.20 \times 0.10$	$0.15 \times 0.10 \times 0.10$		
		Dat	ta collection				
T_{\min}, T_{\max}	0.951, 1.000	0.849, 1.000	0.914, 1.000	0.849, 1.000	0.900, 1.000		
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	3305, 3305, 2213	6359, 3379, 1932	26410, 14721, 9962	25997, 14243, 10296	13842, 7076, 5151		
R _{int}	0.035	0.071	0.046	0.035	0.039		
$(\sin \theta / \lambda)_{max} (\text{\AA}^{-1})$	0.649	0.650	0.650	0.651	0.649		
		R	efinement				
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.038, 0.097, 0.99	0.054, 0.131, 1.06	0.048, 0.114, 1.04	0.038, 0.095, 1.04	0.038, 0.100, 0.96		
No. of reflections	3305	3379	14721	14243	7076		
No. of parameters	169	206	762	745	369		
No. of restraints	0	132	0	0	0		
H-atom treatment	Constrained	Constrained	Mixed	Mixed	Mixed		
$\Delta \rho_{\text{max}}, \Delta \rho_{\text{min}} (e \text{ Å}^{-3})$	0.94, -0.57	0.98, -0.54	0.91, -0.65	0.74, -0.56	0.66, -0.65		

Table S4 Selected experimental crystallographic data.

All experiments were carried out at 150 K with Mo $K\alpha$ radiation using a KappaCCD diffractometer. Absorption was corrected for by multi-scan methods from symmetry-related measurements using SORTAV.¹³

2.4. Additional XANES data



Figure S23 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of Cp₂ZrCl₂ (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S24 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of Cp₂Zr(CO)₂ (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S25 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of Cp*₂ZrCl₂ (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S26 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of Cp*₂Zr(CO)₂ (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S27 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of $[\eta^{8}-Pn^{*}Zr(\mu-Cl)_{3/2}]_{2}(\mu-Cl)_{2}Li \cdot THF_{x}$ (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S28 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of Pn*ZrCpCl (blue) and its first derivative (red). The position of the edge energy could not be determined.



Figure S29 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of 1 (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.



Figure S30 Solid state XANES Zr K-edge spectrum from 17960 eV to 18060 eV of 2 (blue) and its first derivative (red). The position of the edge energy is denoted by a vertical dotted line.

3. DFT calculations

3.1. Computational details

Density functional calculations were carried using the Amsterdam Density Functional package (version ADF2016.107).²¹ The Slater-type orbital (STO) basis sets were of triple- ζ quality augmented with a one polarisation function (ADF basis TZP). Core electrons were frozen (C 1s; Zr 3d) in the model of the electronic configuration for each atom. The local density approximation (LDA) by Vosko, Wilk and Nusair (VWN)²² was used together with the exchange correlation corrections of Becke and Perdew (BP86).^{23,24} In addition to the calculations on the experimental compounds calculations the related model compounds with no methyl substituents on the penatene ligands were carried out. They are denoted by Roman numerals.

3.2. Additional computational data

Parameter	Ι	1	II	2	III	3	IV	4	V	5
1 ur uniteter										
Zr–Zr	3.944	3.963	3.943	3.964	3.939	3.972	3.946	3.964	3.941	3.969
Zr-Ct _{arene}	1.972	1.982	1.972	1.987	1.920	1.987	1.973	1.982	1.971	1.985
av. Zr–C _{arene}	2.456	2.464	2.454	2.468	2.456	2.468	2.460	2.466	2.459	2.467
av. C–C _{arene}	1.474	1.474	1.476	1.475	1.478	1.476	1.476	1.477	1.476	1.475
Puckering Q_{arene}	0.252	0.253	0.257	0.258	0.254	0.260	0.264	0.264	0.250	0.253
Puckering φ_{arene}	90	90	88.128	86.941	86.994	88.035	90.096	90	85.752	83.274
Pn* twist angle	44.04	44.87	44.06	44.06	43.26	46.4	44.01	43.56	45.38	44.19
Pn* fold angle	29.91	29.41	29.77	29.01	29.63	28.98	29.70	28.88	29.60	28.63
Short C–C _{arene}	1.472	1.473	1.471	1.473	1.47	1.473	1.47	1.473	1.469	1.47
Long C–C _{arene}	1.477	1.477	1.483	1.478	1.488	1.479	1.484	1.483	1.483	1.478
$\Sigma(C-C_{arene})$	8.842	8.846	8.855	8.852	8.867	8.855	8.857	8.864	8.857	8.851
$\Delta(C-C_{arene})$	0.005	0.004	0.012	0.005	0.018	0.006	0.014	0.010	0.014	0.008

Table S5 Selected distances (Å), angles (°) and parameters for DFT optimised structures

(Roman numerals refer to the Pn structures; Arabic to the Pn*).

 $\Sigma(C-C_{arene})$ is the sum of the six $C-C_{arene}$ bond distances, $\Delta(C-C_{arene})$ is the difference between the longest and shortest $C-C_{arene}$ distances.



Figure S31 Energy levels and isosurfaces for the Kohn-Sham frontier orbitals of I.

Table S6 Hirshfeld charges on fragments and binding energies (kJ mol⁻¹) for calculated structures **1/I**, **2/II**, **3/III**, **4/IV** and **5/V** (Roman numerals refer to the Pn structures Arabic to the Pn*). The binding energies are for the fragments with *S*=0 and structures identical with those found for the optimised compound.

Structure	ZrPn	Arene	Binding energy
1	0.09	-0.18	-723
Ι	0.15	-0.30	-766
2	0.08	-0.16	-719
II	0.14	-0.28	-766
3	0.07	-0.14	-723
III	0.12	-0.26	-769
	0.14		
4	0.07	-0.13	-713
IV	0.13	-0.25	-766
5	0.07	-0.14	-706
\mathbf{V}	0.13	-0.26	-757
6	0.005	-0.005	-345
VI	0.039	-0.039	-374

1

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С	-0.67405814	0.27472132	4.17512917
С	0.67405814	-0.27472132	4.17512917
C	1 58966073	0 78052287	3 78678583
C	0 79641007	1 96483406	3 61873704
C	-0 59120671	1 6755/113	3 81072243
C	0.59120671	1 67554112	2 01072243
	1 5000072	-1.07554115	3.010/2243
	-1.58966073	-0.78052287	3./86/8583
C	-0./964100/	-1.96483406	3.618/3/04
H	3.4/14844/	-0.25236528	3.4/485101
H	2.28924518	3.28046151	2.75963321
Н	-1.47322233	3.54562756	3.16765110
Н	2.65130250	-2.25658079	3.45220393
H	-3.50424262	-0.93232925	4.79076426
Н	-0.64793230	-3.95786303	2.77905362
С	-0.67405814	-0.27472132	-4.17512917
С	0.67405814	0.27472132	-4.17512917
С	0.59120671	1.67554113	-3.81072243
С	-0.79641007	1.96483406	-3.61873704
С	-1.58966073	0.78052287	-3.78678583
С	1.58966073	-0.78052287	-3.78678583
С	-0.59120671	-1.67554113	-3.81072243
С	0 79641007	-1 96483406	-3 61873704
н	1 89203519	3 07902838	-4 82526511
н	-0 64793230	3 95786303	-2 77905362
и П	-3 47148447	-0 25236528	-3 47485101
11	2 50424262	0.23230320	1 70076426
п	1 47222222	-0.95252925	2 16765110
п	-1.4/322233	-3.34362736	-3.16/65110
н	1.58555832	-3.85887031	-4.2/849229
Zr	0.0000000	0.0000000	-1.98160642
Zr	0.0000000	0.0000000	1.98160642
С	0.0000000	1.43089506	0.0000000
С	-1.28745692	0.72752169	0.12670811
С	-1.28745692	-0.72752169	-0.12670811
С	0.0000000	-1.43089506	0.0000000
С	1.28745692	-0.72752169	0.12670811
С	1.28745692	0.72752169	-0.12670811
Н	-2.20831833	1.29073439	-0.01649025
Н	-2.20831833	-1.29073439	0.01649025
Н	0.0000000	-2.51825422	0.0000000
Н	2.20831833	-1.29073439	-0.01649025
Н	2.20831833	1.29073439	0.01649025
Н	0.0000000	2.51825422	0.0000000
С	3.09336583	0.72955598	3.78602424
Н	3.52510465	1.47577453	3.10422189
Н	3.50424262	0.93232925	4.79076426
С	1.35650460	3.33146579	3.33734245
Н	0.64793230	3.95786303	2.77905362
н	1.58555832	3.85887031	4.27849229
C	1.70592483	-2.68489705	3.81058577
ч	1 89203519	-3 07902838	4 82526511
н Ц	1 4732203313	-3 54562756	3 16765110
C	-1 35650160	-3 331/6570	3 3373/0/5
$\overline{}$	T. JJ0J0400	J.JJII0J/J	5.55/54245

Н	-1.58555832	-3.85887031	4.27849229
ц	-2 29024519	-3 29046151	2 75063321
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C	-1.70592483	2.68489705	3.810585//
Н	-2.65130250	2.25658079	3.45220393
Н	-1.89203519	3.07902838	4.82526511
С	-3 09336583	-0 72955598	3 78602424
	2 47140447	0.25236530	2 47405101
н	-3.4/14844/	0.25256528	3.47485101
Н	-3.52510465	-1.4/5//453	3.10422189
С	1.70592483	2.68489705	-3.81058577
Н	1.47322233	3.54562756	-3.16765110
u	2 65130250	2 25658079	-3 45220393
11 C	2.00130230	2.23030079	3.43220393
C	3.09336583	-0.72955598	-3./8602424
H	3.47148447	0.25236528	-3.47485101
Н	3.52510465	-1.47577453	-3.10422189
С	1 35650460	-3 33146579	-3 33734245
U U	2 28024518	-3 28046151	-2 75963321
11	2.20924010	3.20040131	2.75905521
Н	0.64/93230	-3.95/86303	-2.//905362
С	-1.70592483	-2.68489705	-3.81058577
Н	-2.65130250	-2.25658079	-3.45220393
Н	-1 89203519	-3.07902838	-4 82526511
C	1 25650460	2 22146570	2 22724245
C	-1.55650460	5.55140579	-3.33/34243
Н	-1.58555832	3.85887031	-4.27849229
Н	-2.28924518	3.28046151	-2.75963321
С	-3.09336583	0.72955598	-3.78602424
ц	-3 52510465	1 47577453	-3 10422189
11	2 50424262	1.1/3//1999	4 70076406
Н	-3.50424262	0.93232925	-4./90/6426
I			
10			
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42			
42 C	-0.80803288	0.29037100	4.15982859
42 C C	-0.80803288 0.41017694	0.29037100 -0.51523838	4.15982859 4.12224231
42 C C C	-0.80803288 0.41017694 1.50299079	0.29037100 -0.51523838 0.35871523	4.15982859 4.12224231 3.76828909
42 C C C	-0.80803288 0.41017694 1.50299079 0.96043613	0.29037100 -0.51523838 0.35871523 1.66912460	4.15982859 4.12224231 3.76828909 3.64102817
42 C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613	0.29037100 -0.51523838 0.35871523 1.66912460	4.15982859 4.12224231 3.76828909 3.64102817
42 C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237
42 C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179
42 C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632
42 С С С С С С С С С С С С С С С С С С	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022
42 C C C C C C C C C C C H H	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892 2.55906655	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678
42 C C C C C C C C C C C C H H H	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892 2.55906655 2.51153776	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956
42 C C C C C C C C C C C C C C H H H	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892 2.55906655 2.51153776 -2.69971492	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892 2.55906655 2.51153776 -2.69971492 -0.28162743	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047	0.29037100 -0.51523838 0.35871523 1.66912460 1.65223034 -1.84927734 -0.55450134 -1.85236267 0.08967892 2.55906655 2.51153776 -2.69971492 -0.28162743 -2.72245646	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047 -0.58861092	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646 - 0.00633145	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668 -4.24584938
42 C C C C C C C C C C C C C C C C C C C	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047 -0.58861092 0.84473313	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646 - 0.00633145 0.22929579	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668 -4.24584938 -4.10658432
42 ССССССС ССССС Н Н Н Н Н Н Н Н ССС СССС	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047 -0.58861092 0.84473313 1.02841788	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646 - 0.00633145 0.22929579 1.60225563	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668 -4.24584938 -4.10658432 -3.70647713
42 ССССССС ССССС Н Н Н Н Н Н Н Н СССС СССС СССССС	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047 -0.58861092 0.84473313 1.02841788 -0.26345480	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646 - 0.00633145 0.22929579 1.60225563 2.18554045	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668 -4.24584938 -4.10658432 -3.70647713 -3.61483166
42 ССССССС Н Н Н Н Н Н СССС СССССС СССССССС	-0.80803288 0.41017694 1.50299079 0.96043613 -0.44543720 0.04477554 -1.90345246 -1.36117854 2.54962294 1.54158343 -1.10795758 0.70836404 -2.95249595 -1.94424047 -0.58861092 0.84473313 1.02841788 -0.26345480	0.29037100 - 0.51523838 0.35871523 1.66912460 1.65223034 - 1.84927734 - 0.55450134 - 1.85236267 0.08967892 2.55906655 2.51153776 - 2.69971492 - 0.28162743 - 2.72245646 - 0.00633145 0.22929579 1.60225563 2.18554045 1.22312676	4.15982859 4.12224231 3.76828909 3.64102817 3.84936237 3.71316179 3.74527871 3.51591632 3.66157022 3.40365678 3.79919956 3.58655968 3.67536660 3.21678668 -4.24584938 -4.10658432 -3.70647713 -3.61483166 -3.9016061
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5 84 C C C C C C C C C	4.18430152 4.17509253 3.85725178 3.75155271 3.90457691 3.74073919 3.73439469	-0.21788305 0.33399229 -0.73792632 -1.93288457 -1.63735174 1.71314122 0.81384169	-0.90668137 0.44059084 1.36158282 0.57390512 -0.81685450 0.36020226 -1.81965618
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5 84 C C C C C C C C C C C C C C C C C C	4.18430152 4.17509253 3.85725178 3.75155271 3.90457691 3.74073919 3.73439469 3.51494271 3.53583285 2.98032700 3.46864406 3.39858729 4.69771419	-0.21788305 0.33399229 -0.73792632 -1.93288457 -1.63735174 1.71314122 0.81384169 1.98933736 0.27997006 -3.30440060 -3.58404159 2.28402699 1.01766321	-0.90668137 0.44059084 1.36158282 0.57390512 -0.81685450 0.36020226 -1.81965618 -1.02661964 3.24997582 2.06562120 -1.66119398 2.42623837 -3.74980328
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5 84 C C C C C C C C C C C C C C C C C C	4.18430152 4.17509253 3.85725178 3.75155271 3.90457691 3.74073919 3.73439469 3.51494271 3.53583285 2.98032700 3.46864406 3.39858729 4.69771419 2.57352582 -4.14730830	-0.21788305 0.33399229 -0.73792632 -1.93288457 -1.63735174 1.71314122 0.81384169 1.98933736 0.27997006 -3.30440060 -3.58404159 2.28402699 1.01766321 3.93551277 0.32488481	-0.90668137 0.44059084 1.36158282 0.57390512 -0.81685450 0.36020226 -1.81965618 -1.02661964 3.24997582 2.06562120 -1.66119398 2.42623837 -3.74980328 -0.86749782 -0.92614549
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C	0.69245607	-0 32352519	4 15200307
C	1 60633303	0.73119900	3 70025732
C	0 02201722	1 01017065	2 64645562
C	-0 54629597	1 64717137	3 95606473
C	-0.54020507	1 70651514	2 75010507
	0.57819108	-1.70651514	3.75910597
	-1.5/612/12	-0./9189313	3.78869839
C	-0.805/9645	-1.96985333	3.56991046
H	2.68499317	0.65/86023	3.68499450
H	1.2446/624	2.89696621	3.40329301
H	-1.35502766	2.37041738	3.80543327
H	1.38546605	-2.42382867	3.64410427
Н	-2.65742308	-0.71571343	3.71851465
H	-1.22011832	-2.93602624	3.28518226
С	-0.66888881	-0.29676708	-4.16964562
С	0.68610252	0.24705716	-4.16929597
С	0.59993937	1.64882167	-3.83752231
С	-0.77536537	1.94510914	-3.64968763
С	-1.56847966	0.77095695	-3.80747193
С	1.58347401	-0.81049147	-3.76868457
С	-0.58692562	-1.68613435	-3.78840891
С	0.78757834	-1.97620920	-3.57769981
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Н	-1 17133264	2 93087574	-3 40969788
н	-2 65041625	0 71875650	-3 72757618
ц	2 66504449	-0 75612768	-3 68710246
ц	-1 40913938	-2 38971001	-3 69570623
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11 7 m	0 01051676	0.00564090	1 07102026
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21	-0.01/64898	0.00860465	1.96930846
C	-0.00945491	1.4/552469	-0.00219512
C	-1.284/1442	0.75276118	0.09501467
C	-1.2/93360/	-0./0688301	-0.11/05644
С	0.01/2426/	-1.401/0218	-0.01396346
С	1.30790153	-0.69972921	0.16831899
С	1.27269155	0.75531411	-0.11461366
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С	2.59982501	-1.46566824	-0.01296625
Н	3.43638207	-0.91526981	0.43859489
Н	2.19003264	1.32807971	0.02788612
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Н	-0.80292784	3.39264604	-0.64145220
Н	-0.21476873	3.38813737	1.02932879
		-	

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