

Supporting Information

Cobalt(II)-Catalyzed Chelation-Assisted C–H Iodination of Aromatic Amides with I₂

Yadagiri Kommagalla, Ken Yamazaki, Takuma Yamaguchi and Naoto Chatani*

Department of Applied Chemistry, Faculty of Engineering, Osaka University,

Suita, Osaka 565-0871, Japan

chatani@chem.eng.osaka-u.ac.jp

Contents

I. General Information	S2
II. Materials	S2
III. Synthesis of Starting Amides	S2
IV. General Procedure for the Preparation of Starting Amides	S2
V. Spectroscopic Data for Starting Amides	S3
VI. General Procedure for the C-H Iodination	S3
VII. Spectroscopic Data for Iodinated Products	S4-S9
VIII. Deuterium Labelling Experiments (Scheme 3a)	S9
IX. Reaction with Radical Scavengers (Scheme 3b)	S9
X. Competition Experiment for Kinetic Isotope Effect (Scheme 4)	S9
XI. Optimization Study	S10
XII. Transition-State Geometry of the CMD Pathway	S11
XIII. Energy Surface for CMD Pathway	S11
XIV. Spin Densities of Proposed Intermediates in SET Pathway	S12
XV. Computational Details	S13
XVI. References	S13
XVII. Copies of ¹ H and ¹³ C NMR Spectra	S14-S47
XVIII. Calculated Cartesian Coordinates, Energies and Spin Densities	S48-S59

I. General Information

^1H NMR and ^{13}C NMR spectra were recorded on a JEOL ECS-400 spectrometer in CDCl_3 with tetramethylsilane as the internal standard. Data are reported as follows: chemical shifts in ppm (δ), multiplicity (s = singlet, d = doublet, t = triplet, dd = doublet of doublet, q = quartet, brs = broad singlet, and m = multiplet), coupling constant (Hz), and integration. Infrared spectra (IR) were obtained using a JASCO FT/IR-4200 spectrometer; absorptions are reported in reciprocal centimeters with the following relative intensities: s (strong), m (medium), or w (weak). Mass spectra and high resolution mass spectra (HRMS) were obtained using a JEOL JMS-700 spectrometer. Melting points were determined using a Stanford Research Systems apparatus. Column chromatography was performed with SiO_2 (Silicycle SiliaFlash F60 (230-400 mesh)). Some compounds were purified by LC-908 HPLC (GPC).

II. Materials

$\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ (CAS 6147-53-1) was purchased from Wako Pure Chemical Industries, Ltd. Ag_2CO_3 (CAS 534-16-7) was purchased from Nacalai Tesque. Iodine (CAS 7553-56-2) and 8-aminoquinoline (CAS 578-66-5) were purchased from Tokyo Chemical Industry Co., Ltd. KOTf (CAS 2926-27-4) and DCE (CAS 107-06-2) were purchased from Sigma-Aldrich. These reagents were used as received. 5-Chloroquinolin-8-amine (CAS 5432-09-7) and 5-methoxyquinolin-8-amine (CAS 30465-68-0) were prepared by using literature procedures.^{1, 2}

III. Synthesis of Starting Amides

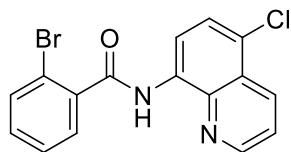
All amides bearing an 5-chloro-8-aminoquinolinyl moiety were prepared by reacting the corresponding acid or acid chloride with 5-chloro-8-aminoquinoline.³ The amides for **1a-1e**, **1g**, and **1i-1o** were prepared by using literature procedures.²

VI. General Procedure for the Preparation of Starting Amide

To an oven dried two neck round bottom flask, the acid chloride (5.0 mmol) dissolved in CH_2Cl_2 (12 mL) and the resulting reaction mixture was cool to 0 °C. The solution of 8-amino-5-chloroquinoline (5.5 mmol) and Et_3N (10 mmol) in CH_2Cl_2 (5 mL) was added dropwise to the reaction mixture at 0 °C. The resulting reaction mixture was allowed to warm to rt and continued for stirring for overnight. The crude reaction mixture was quenched with sat. NaHCO_3 and the aqueous layer was extracted with DCM (2 x 15 mL). The combined organic layers were washed with 1 M HCl aq. (30 mL) and brine (30 mL), dried over MgSO_4 , filtered and evaporated *in vacuo*. The resulting crude amide was purified by column chromatography on silica gel (eluant: hexane/ EtOAc = 10/1) to afford the desired amide.

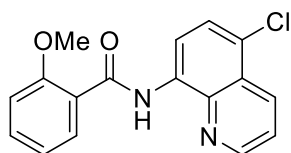
V. Spectroscopic Data for Starting Amides

2-bromo-*N*-(5-chloroquinolin-8-yl)benzamide (**1f**)



R_f 0.52 (hexane/EtOAc = 4/1). Yellowish green colour solid. MP = 166 °C. ¹H-NMR (400 MHz, Chloroform-D) δ: 7.34-7.39 (m, 1H), 7.46 (td, *J* = 7.6, 1.1 Hz, 1H), 7.58 (dd, *J* = 8.5, 4.1 Hz, 1H), 7.67 (d, *J* = 8.5 Hz, 1H), 7.71 (td, *J* = 8.0, 1.2 Hz, 2H), 8.59 (dd, *J* = 8.5, 1.6 Hz, 1H), 8.84 (dd, *J* = 4.4, 1.6 Hz, 1H), 8.90 (d, *J* = 8.5 Hz, 1H), 10.27 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ: 116.78, 119.65, 122.44, 124.99, 125.99, 127.22, 127.68, 129.61, 131.63, 133.42, 133.56, 133.73, 138.02, 139.14, 148.83, 165.85. IR (neat) 3334 w, 2995 w, 1752 s, 1521 w, 1479 w, 1368 s, 1239 s, 1055 w. MS *m/z* (relative intensity, %) 360 (M⁺, 36), 281 (10), 253 (5), 185 (97), 183 (100), 157 (20), 155 (21). HRMS calcd for C₁₆H₁₀BrClN₂O: 359.9665; found: 359.9669.

N-(5-chloroquinolin-8-yl)-2-methoxybenzamide (**1h**)



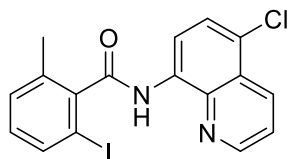
R_f 0.32 (hexane/EtOAc = 4/1). Off white colour solid. MP = 146 °C. ¹H-NMR (400 MHz, Chloroform-D) δ: 4.19 (s, 3H), 7.07 (d, *J* = 8.3 Hz, 1H), 7.14 (dt, *J* = 10.2, 3.9 Hz, 1H), 7.49-7.54 (m, 1H), 7.56 (dd, *J* = 8.5, 4.1 Hz, 1H), 7.63 (d, *J* = 8.5 Hz, 1H), 8.34 (dd, *J* = 7.8, 1.8 Hz, 1H), 8.56 (dd, *J* = 8.5, 1.6 Hz, 1H), 8.90 (dd, *J* = 4.1, 1.6 Hz, 1H), 8.98 (d, *J* = 8.5 Hz, 1H), 12.33 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ: 56.09, 111.56, 117.15, 121.32, 122.01, 122.12, 124.08, 125.97, 127.39, 132.35, 133.20, 133.28, 135.04, 139.75, 148.64, 157.70, 163.59. IR (neat) 3262 w, 1714 s, 1654 m, 1536 s, 1481 m, 1360 m, 1219 m, 1022 w. MS *m/z* (relative intensity, %) 313 (M⁺, 35), 178 (25), 136 (9), 135 (100), 92 (7). HRMS calcd for C₁₇H₁₃ClN₂O₂: 312.0666; found: 312.0663.

VI. General Procedure for the C-H Iodination

To an oven-dried 5 mL screw-capped vial, *N*-(5-chloroquinolin-8-yl)-2-methylbenzamide **1a** (45 mg, 0.15 mmol), iodine (76 mg, 0.3 mmol), Co(OAc)₂·4H₂O (7.6 mg, 0.03 mmol), Ag₂CO₃ (42 mg, 0.15 mmol), KOTf (57 mg, 0.3 mmol) and DCE (0.5 mL) were added. The mixture was stirred for 6 h at 120 °C followed by cooling. The resulting mixture was filtered through a celite pad and concentrated in vacuo. The residue was purified by column chromatography on silica gel (eluent: hexane/EtOAc = 10/1) to afford the desired iodinated product **2a** (51 mg, 80%) as a white solid. The compounds **2a-2e**, **2g**, **2i-2n** were known.^{2b}

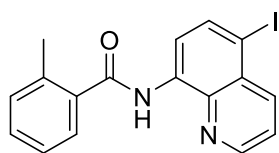
VII. Spectroscopic Data for Iodinated Products

N-(5-chloroquinolin-8-yl)-2-iodo-6-methylbenzamide (2a).



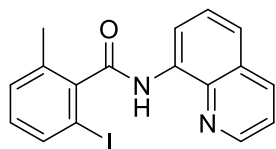
¹H-NMR (400 MHz, Chloroform-D) δ : 2.45 (s, 3H), 7.04 (t, J = 7.8 Hz, 1H), 7.25 (d, J = 7.1 Hz, 1H), 7.56 (dd, J = 8.2, 4.3 Hz, 1H), 7.67 (d, J = 8.2 Hz, 1H), 7.71 (d, J = 8.2 Hz, 1H), 8.58 (dd, J = 8.5, 1.6 Hz, 1H), 8.80 (dd, J = 4.1, 1.8 Hz, 1H), 8.90 (d, J = 8.2 Hz, 1H), 9.87 (s, 1H).

N-(5-iodoquinolin-8-yl)-2-methylbenzamide (1b')



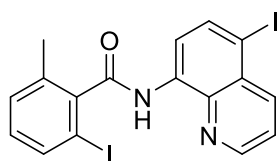
¹H-NMR (400 MHz, Chloroform-D) δ : 2.60 (s, 3H), 7.33 (t, J = 7.3 Hz, 2H), 7.44-7.40 (m, J = 4.0 Hz, 1H), 7.54 (dd, J = 8.7, 4.1 Hz, 1H), 7.68 (d, J = 7.8 Hz, 1H), 8.14 (d, J = 8.7 Hz, 1H), 8.40 (dd, J = 8.7, 1.4 Hz, 1H), 8.72 (d, J = 8.2 Hz, 1H), 8.75 (dd, J = 4.1, 1.8 Hz, 1H), 10.23 (s, 1H).

2-iodo-6-methyl-*N*-(quinolin-8-yl)benzamide (2b)



¹H-NMR (400 MHz, Chloroform-D) δ : 2.48 (s, 3H), 7.05 (t, J = 7.8 Hz, 1H), 7.26 (t, J = 3.7 Hz, 2H), 7.47 (dd, J = 8.2, 4.1 Hz, 1H), 7.65-7.58 (m, 2H), 7.73 (d, J = 7.8 Hz, 1H), 8.20 (dd, J = 8.2, 1.8 Hz, 1H), 8.77 (dd, J = 4.4, 1.6 Hz, 1H), 8.97 (dd, J = 7.1, 1.6 Hz, 1H), 9.93 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ : 20.15, 92.94, 117.13, 121.70, 122.25, 127.42, 128.04, 130.06, 130.57, 134.12, 136.44, 136.60, 136.80, 138.48, 143.16, 148.33, 168.18.

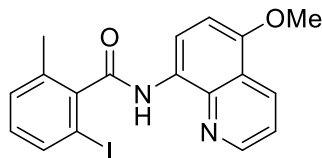
2-iodo-*N*-(5-iodoquinolin-8-yl)-6-methylbenzamide (2b')



R_f 0.52 (hexane/EtOAc = 4/1). Off white colour solid. MP = 173 °C. ¹H-NMR (400 MHz, Chloroform-D) δ : 2.46 (s, 3H), 7.05 (t, J = 8.0 Hz, 1H), 7.26 (t, J = 3.9 Hz, 1H), 7.54 (dd, J = 8.7, 4.1 Hz, 1H), 7.73 (d, J = 7.8 Hz, 1H), 8.16 (d, J = 8.3 Hz, 1H), 8.40 (dd, J = 8.5, 1.6 Hz, 1H), 8.75-8.73 (m, 2H), 9.93 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ : 20.13, 76.68, 77.00, 77.32, 90.28, 92.82, 118.33, 123.26, 129.70, 130.10, 130.68, 135.01, 136.61, 136.79, 138.21, 139.07,

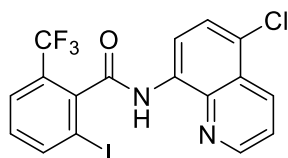
140.74, 142.92, 148.94, 168.17. IR (neat) 3001 w, 1710 s, 1513 w, 1421 w, 1358 s, 1219 s, 1092 w, 906 w. MS m/z (relative intensity, %) 514 (M^+ , 58), 387 (6), 270 (4), 260 (9), 245 (100), 217 (15), 119 (15). HRMS calcd for $C_{17}H_{12}I_2N_2O$: 513.9039; found: 513.9040.

2-iodo-*N*-(5-methoxyquinolin-8-yl)-6-methylbenzamide (2c).



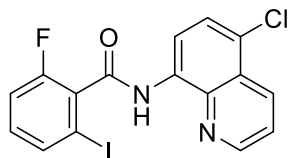
1H -NMR (400 MHz, Chloroform- D) δ : 2.45 (s, 3H), 4.01 (s, 3H), 6.89 (d, J = 8.7 Hz, 1H), 7.01 (t, J = 7.8 Hz, 1H), 7.23 (d, J = 8.4 Hz, 1H), 7.42 (dd, J = 8.5, 4.4 Hz, 1H), 7.70 (d, J = 7.8 Hz, 1H), 8.58 (dd, J = 8.5, 1.6 Hz, 1H), 8.75 (q, J = 4.4, 1.6 Hz, 1H), 8.86 (d, J = 8.2 Hz, 1H), 9.66 (s, 1H).

***N*-(5-chloroquinolin-8-yl)-2-iodo-6-(trifluoromethyl)benzamide (2d).**



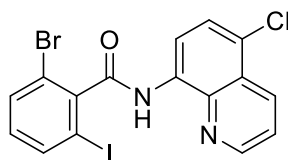
1H -NMR (400 MHz, Chloroform- D) δ : 7.30 (t, J = 8.0 Hz, 1H), 7.58 (dd, J = 8.7, 4.1 Hz, 1H), 7.69 (d, J = 8.2 Hz, 1H), 7.76 (d, J = 8.2 Hz, 1H), 8.13 (d, J = 7.8 Hz, 1H), 8.60 (dd, J = 8.7, 1.4 Hz, 1H), 8.81 (dd, J = 4.4, 1.6 Hz, 1H), 8.87 (d, J = 8.7 Hz, 1H), 9.95 (s, 1H).

***N*-(5-chloroquinolin-8-yl)-2-fluoro-6-iodobenzamide (2e).**



1H -NMR (400 MHz, Chloroform- D) δ : 7.14-7.23 (m, 2H), 7.59 (dd, J = 8.5, 4.4 Hz, 1H), 7.69 (d, J = 8.7 Hz, 1H), 7.73 (dd, J = 7.3, 1.4 Hz, 1H), 8.61 (dd, J = 8.5, 1.6 Hz, 1H), 8.84 (dd, J = 4.1, 1.4 Hz, 1H), 8.90 (d, J = 8.2 Hz, 1H), 10.05 (s, 1H).

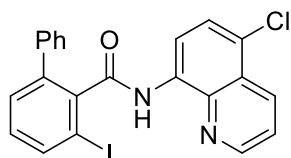
***N*-(5-chloroquinolin-8-yl)-2-bromo-6-iodobenzamide (2f).**



R_f 0.52 (hexane/EtOAc = 4/1). white colour solid. MP = 222 °C. 1H -NMR (400 MHz, Chloroform- D) δ : 7.02 (t, J = 8.0 Hz, 1H), 7.59 (dd, J = 8.7, 4.1 Hz, 1H), 7.64 (dd, J = 8.2, 0.9 Hz, 1H), 7.69 (d, J = 8.7 Hz, 1H), 7.86 (dd, J = 7.8, 0.9 Hz, 1H), 8.60 (dd, J = 8.5, 1.6 Hz, 1H), 8.84 (dd, J = 4.4, 1.6 Hz, 1H), 8.89 (d, J = 8.7 Hz, 1H), 9.93 (s, 1H). ^{13}C -NMR (101 MHz, Chloroform- D) δ : 93.27,

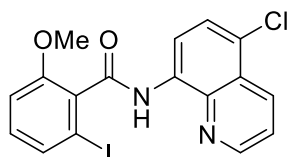
117.17, 119.61, 122.49, 125.39, 126.06, 127.21, 131.84, 132.65, 133.17, 133.47, 138.29, 139.10, 143.60, 148.91, 166.20. IR (neat) 3328 w, 2992 w, 1767 m, 1675 m, 1520 s, 1479 m, 1386 m, 1243 s, 913 s, 745 s. MS m/z (relative intensity, %) 486 (M^+ , 32), 407 (11), 357 (100), 311 (71), 309 (73), 205 (17), 150 (11). HRMS calcd for $C_{16}H_9BrClIN_2O$: 485.8631; found: 485.8625.

***N*-(5-chloroquinolin-8-yl)-3-iodo-[1,1'-biphenyl]-2-carboxamide (2g).**



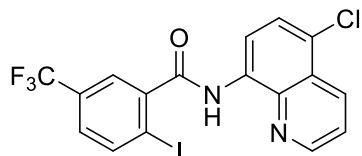
1H -NMR (400 MHz, Chloroform- D) δ : 7.16 (t, J = 7.3 Hz, 1H), 7.28 (t, J = 7.6 Hz, 6H), 7.46 (dd, J = 8.7, 4.1 Hz, 1H), 7.48-7.53 (m, 4H), 7.56-7.60 (m, 2H), 7.92 (d, 1H), 8.48 (dd, J = 8.5, 1.6 Hz, 1H), 8.55 (dd, J = 4.4, 1.6 Hz, 1H), 8.76 (d, J = 8.7 Hz, 1H), 9.70 (s, 1H).

***N*-(5-chloroquinolin-8-yl)-2-iodo-6-methoxybenzamide (2h).**



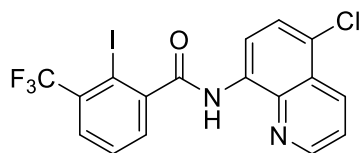
R_f 0.23 (hexane/EtOAc = 4/1). Yellow solid. MP = 224 °C. 1H -NMR (400 MHz, Chloroform- D) δ : 3.83 (s, 3H), 6.98 (d, J = 8.2 Hz, 1H), 7.11 (t, J = 8.0 Hz, 1H), 7.49 (dd, J = 7.8, 0.9 Hz, 1H), 7.57 (dd, J = 8.2, 4.1 Hz, 1H), 7.67 (d, J = 8.2 Hz, 1H), 8.59 (dd, J = 8.5, 1.6 Hz, 1H), 8.82 (dd, J = 4.4, 1.6 Hz, 1H), 8.93 (d, J = 8.7 Hz, 1H), 9.95 (s, 1H). ^{13}C -NMR (101 MHz, Chloroform- D) δ : 56.16, 94.13, 111.10, 117.04, 122.44, 124.91, 126.10, 127.38, 131.39, 131.75, 132.64, 133.47, 133.81, 139.19, 148.81, 157.06, 166.07. IR (neat) 3334 w, 2987 w, 1739 s, 1669 m, 1588 w, 1517 s, 1451 w, 1372 m, 1239 s, 1045 m. MS m/z (relative intensity, %) 438 (M^+ , 25), 262 (9), 261 (100), 203 (6), 166 (45), 135 (14). HRMS calcd for $C_{17}H_{12}ClIN_2O_2$: 437.9632; found: 437.9637.

***N*-(5-chloroquinolin-8-yl)-2-iodo-5-(trifluoromethyl)benzamide (2i).**



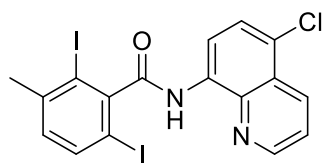
1H -NMR (400 MHz, Chloroform- D) δ : 7.44 (dd, J = 8.2, 2.3 Hz, 1H), 7.61 (dd, J = 8.2, 4.1 Hz, 1H), 7.69 (d, J = 8.2 Hz, 1H), 7.85 (d, J = 1.8 Hz, 1H), 8.12 (d, J = 8.2 Hz, 1H), 8.62 (dd, J = 8.5, 1.6 Hz, 1H), 8.85-8.87 (m, 2H), 10.14 (s, 1H).

***N*-(5-chloroquinolin-8-yl)-2-iodo-3-(trifluoromethyl)benzamide (2i').**



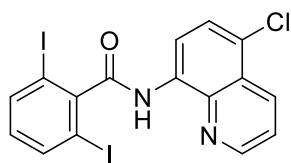
R_f 0.52 (hexane/EtOAc = 4/1). White colour solid. MP = 134 °C. ¹H-NMR (400 MHz, Chloroform-D) δ: 7.05 (d, *J* = 8.2 Hz, 1H), 7.45 (dd, *J* = 8.0, 3.9 Hz, 1H), 7.65 (d, *J* = 8.2 Hz, 1H), 7.87 (d, *J* = 7.8 Hz, 1H), 8.00 (s, 1H), 8.29 (d, *J* = 8.7 Hz, 1H), 8.39 (d, *J* = 8.2 Hz, 1H), 8.53 (d, *J* = 4.1 Hz, 1H), 10.21 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ: 36.70, 95.57, 116.26, 122.11, 124.24, 124.57, 124.96, 126.33, 127.75, 131.66, 132.84, 132.93, 136.69, 138.02, 148.45, 165.58. IR (neat) 3328 w, 2995 w, 1711 s, 1677 m, 1528 s, 1481 m, 1327 m, 1218 m, 1087 w. MS *m/z* (relative intensity, %) 476 (M⁺, 2), 350 (50), 207 (8), 205 (24), 173 (100), 145 (48). HRMS calcd for C₁₇H₉ClF₃N₂O: 475.9400; found: 475.9409.

N-(5-chloroquinolin-8-yl)-2,6-diiodo-3-methylbenzamide (2j)



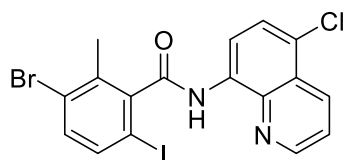
R_f 0.49 (hexane/EtOAc = 4/1). White colour solid. MP = 229 °C. ¹H-NMR (400 MHz, Chloroform-D) δ: 2.48 (s, 3H), 7.03 (d, *J* = 7.8 Hz, 1H), 7.59 (dd, *J* = 8.3, 4.1 Hz, 1H), 7.70 (d, *J* = 8.3 Hz, 1H), 7.76 (d, *J* = 8.3 Hz, 1H), 8.61 (dd, *J* = 8.5, 1.6 Hz, 1H), 8.84 (dd, *J* = 4.4, 1.6 Hz, 1H), 8.90 (d, *J* = 8.7 Hz, 1H), 9.89 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ: 28.72, 76.68, 77.00, 77.21, 77.32, 88.07, 99.53, 117.16, 122.48, 125.29, 126.06, 127.22, 131.42, 133.24, 133.46, 138.76, 139.10, 142.93, 147.83, 148.89, 168.52. IR (neat) 3335 w, 2922 w, 1677 m, 1517 s, 1477 m, 1386 w, 1320 w, 947 w. MS *m/z* (relative intensity, %) 548 (M⁺, 44), 371 (100), 343 (7), 216 (10), 89 (9). HRMS calcd for C₁₇H₁₁ClI₂N₂O: 547.8649; found: 547.8647.

N-(5-chloroquinolin-8-yl)-2,6-diiodobenzamide (2k)



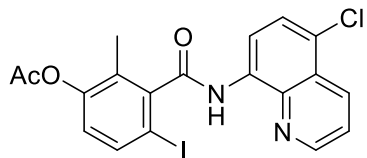
¹H-NMR (400 MHz, Chloroform-D) δ: 6.84 (t, *J* = 8.0 Hz, 1H), 7.60 (dd, *J* = 8.5, 4.4 Hz, 1H), 7.70 (d, *J* = 8.3 Hz, 1H), 7.89 (d, *J* = 8.3 Hz, 2H), 8.61 (dd, *J* = 8.7, 1.4 Hz, 1H), 8.85 (dd, *J* = 4.4, 1.6 Hz, 1H), 8.89 (d, *J* = 8.3 Hz, 1H), 9.91 (s, 1H).

3-bromo-N-(5-chloroquinolin-8-yl)-6-iodo-2-methylbenzamide (2l).



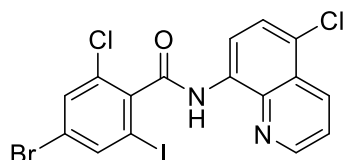
$^1\text{H-NMR}$ (400 MHz, Chloroform-D) δ : 2.52 (s, 3H), 7.35 (d, $J = 8.2$ Hz, 1H), 7.57-7.61 (m, 2H), 7.69 (d, $J = 8.2$ Hz, 1H), 8.61 (dd, $J = 8.5, 1.6$ Hz 1H), 8.83 (dd, $J = 4.1, 1.4$ Hz, 1H), 8.89 (dd, $J = 8.2, 6.0$ Hz, 1H), 9.88 (s, 1H).

3-((5-chloroquinolin-8-yl)carbamoyl)-4-iodo-2-methylphenyl acetate (2m).



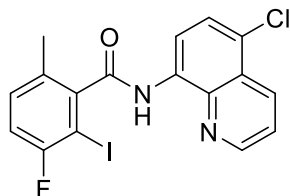
$^1\text{H-NMR}$ (400 MHz, Chloroform-D) δ : 2.29 (s, 3H), 2.35 (s, 3H), 6.89 (d, $J = 8.2$ Hz, 1H), 7.59 (dd, $J = 8.7, 4.1$ Hz, 1H), 7.68 (d, $J = 8.2$ Hz, 1H), 7.75 (d, $J = 8.7$ Hz, 1H), 8.60 (dd, $J = 8.7, 1.8$ Hz, 1H), 8.83 (dd, $J = 4.1, 1.4$ Hz, 1H), 8.89 (d, $J = 8.7$ Hz, 1H), 9.92 (s, 1H).

4-bromo-2-chloro-N-(5-chloroquinolin-8-yl)-6-iodobenzamide (2n).



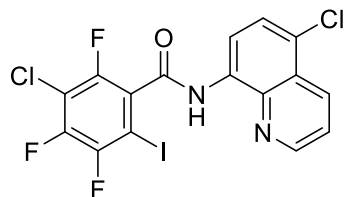
$^1\text{H-NMR}$ (400 MHz, Chloroform-D) δ : 7.59 (dd, $J = 8.7, 4.1$ Hz, 1H), 7.64 (d, $J = 1.8$ Hz, 1H), 7.68 (d, $J = 8.2$ Hz, 1H), 7.98 (d, $J = 1.4$ Hz, 1H), 8.60 (dd, $J = 8.5, 1.6$ Hz, 1H), 8.83 (dd, $J = 4.1, 1.8$ Hz, 1H), 8.86 (d, $J = 8.7$ Hz, 1H), 9.94 (s, 1H).

N-(5-chloroquinolin-8-yl)-3-fluoro-2-iodo-6-methylbenzamide (2o).



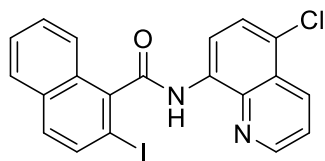
$^1\text{H-NMR}$ (400 MHz, Chloroform-D) δ : 2.45 (s, 3H), 7.05 (t, $J = 8.0$ Hz, 1H), 7.22 (dd, $J = 7.8, 5.0$ Hz, 1H), 7.58 (dd, $J = 8.2, 4.1$ Hz, 1H), 7.69 (d, $J = 8.2$ Hz, 1H), 8.60 (dd, $J = 8.7, 1.4$ Hz, 1H), 8.82 (dd, $J = 4.4, 1.6$ Hz, 1H), 8.90 (d, $J = 8.2$ Hz, 1H), 9.89 (s, 1H).

3-chloro-N-(5-chloroquinolin-8-yl)-2,4,5-trifluoro-6-iodobenzamide (2p).



$^1\text{H-NMR}$ (400 MHz, Chloroform-D) δ : 7.59-7.62 (m, 1H), 7.68 (dd, $J = 8.2, 1.4$ Hz, 1H), 8.61 (d, $J = 8.7$ Hz, 1H), 8.82-8.84 (m, 2H), 10.07 (s, 1H).

***N*-(5-chloroquinolin-8-yl)-2-iodo-1-naphthamide (2q)**



R_f 0.48 (hexane/EtOAc = 4/1). Off white colour solid. MP = 256 °C. ¹H-NMR (400 MHz, Chloroform-D) δ: 7.50-7.58 (m, 3H), 7.66 (d, *J* = 8.7 Hz, 1H), 7.74 (d, *J* = 8.2 Hz, 1H), 7.89 (m, 2H), 7.97 (d, *J* = 8.7 Hz, 1H), 8.60 (dd, *J* = 8.5, 1.6 Hz, 1H), 8.75 (dd, *J* = 4.1, 1.4 Hz, 1H), 9.04 (d, *J* = 8.2 Hz, 1H), 10.08 (s, 1H). ¹³C-NMR (101 MHz, Chloroform-D) δ: 91.35, 117.06, 122.46, 125.09, 125.28, 126.08, 127.02, 127.25, 127.91, 128.21, 130.63, 131.25, 132.54, 133.43, 133.52, 135.28, 139.10, 140.92, 148.86, 167.79. IR (neat) 3334 w, 2987 w, 1746 m, 1588 w, 1517 s, 1466 s, 1374 w, 1242 s, 1047 m. MS *m/z* (relative intensity, %) 458 (M⁺, 19), 331 (14), 295 (15), 282 (12), 281 (100), 166 (45), 155 (64), 126 (31). HRMS calcd for C₂₀H₁₂ClIN₂O: 457.9683; found: 457.9682.

VIII. Deuterium Labeling Experiments (Scheme 3a)

To an oven-dried 5 mL screw-capped vial, **1a-d₇** (46 mg, 0.15 mmol), iodine (76 mg, 0.3 mmol), Co(OAc)₂·4H₂O (7.6 mg, 0.03 mmol), Ag₂CO₃ (42 mg, 0.15 mmol), KOTf (57 mg, 0.3 mmol) and DCE (0.5 mL) were added. The mixture was stirred at 120 °C followed by cooling. The resulting mixture was filtered through a celite pad and concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (eluent: hexane/EtOAc= 10/1). NMR yields were determined using 1,1,2,2-tetrachloroethane as the reference. The deuterium content was also determined by ¹H-NMR.

IX. Reactions with Radical Scavengers (Scheme 3b)

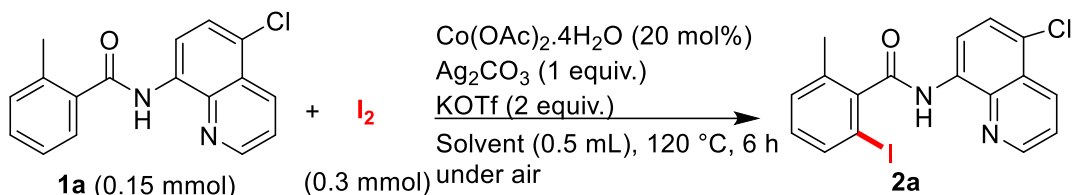
To an oven-dried 5 mL screw-capped vial, **1a** (45 mg, 0.15 mmol), iodine (76 mg, 0.3 mmol), radical scavenger (0.3 mmol), Co(OAc)₂·4H₂O (7.6 mg, 0.03 mmol), Ag₂CO₃ (42 mg, 0.15 mmol), KOTf (57 mg, 0.3 mmol), radical scavenger (TEMPO/BHT (0.18 mmol)) and DCE (0.5 mL) were added. The mixture was stirred at 120 °C followed by cooling. The resulting mixture was filtered through a celite pad and concentrated *in vacuo*. The yield of the iodinated product **2a** and the recovery of **1a** were determined by ¹H-NMR.

X. Competition Experiment for Kinetic Isotope Effect (Scheme 4)

To an oven-dried 5 mL screw-capped vial, **1a-d₇** or **1a** (45 mg, 0.15 mmol), iodine (76 mg, 0.3 mmol), Co(OAc)₂·4H₂O (7.6 mg, 0.03 mmol), Ag₂CO₃ (42 mg, 0.15 mmol), KOTf (57 mg, 0.3 mmol) and DCE (0.5 mL) were added. The mixture was stirred at 120 °C followed by cooling. The resulting mixture was filtered through a celite pad and concentrated *in vacuo*. The yields of iodinated product and SM was determined by ¹H-NMR in CDCl₃ as NMR solvent.

XI. Optimization Study

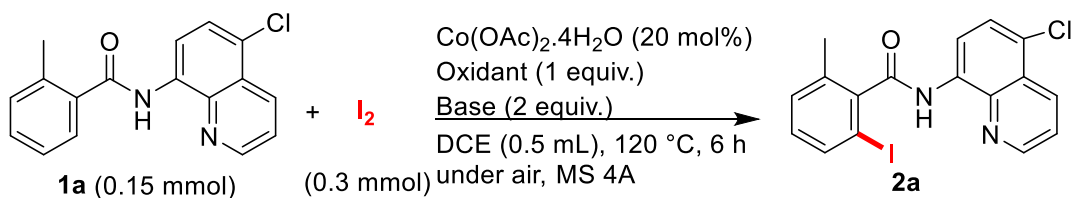
Table S1: Solvent Screening.^a



Entry	Solvent	Yields (2a/1a)(%) ^{b, c}	note
1	DCE	85/trace (80)	
2	PhCl	68/14 (62)	
3	CF ₃ CH ₂ OH	19/35	
4	DMF	0/100	no reaction
5	DMSO	0/72	no reaction
6	1,4-dioxane	18/45	

^a All reactions involved treating **1a** (0.15 mmol), I₂ (0.3 mmol) in a solvent (0.5 mL) in the presence of the Co(OAc)₂·4H₂O (0.03 mmol) and the Ag₂CO₃ (0.15 mmol), KOTf (0.3 mmol) at 120 °C for 6 h under air, unless otherwise stated. ^b Yields were determined by ¹H NMR analysis of the crude reaction mixture using 1,1,2,2-tetrachloroethane as an internal standard. ^c Values given in the parentheses are isolated product yields.

Table S2: Optimization.^a



Entry	Base	Oxidant	Yields (2a/1a)(%) ^{b, c}
1	CsOAc	Ag ₂ CO ₃	0/83
2 ^d	KOTf	PhI(OAc) ₂	17/69
3 ^e	KOTf	Ag ₂ CO ₃	83/trace (79)
4 ^f	KOTf	Ag ₂ CO ₃	0/99
5	KOTf	-	0/99

^a All reactions involved treating **1a** (0.15 mmol), I₂ (0.3 mmol) in a solvent (0.5 mL) in the presence of the Co(OAc)₂·4H₂O (0.03 mmol) and the Ag₂CO₃ (0.15 mmol), Base (0.3 mmol) at 120 °C for 6 h under air, unless otherwise stated. ^b Yields were determined by ¹H NMR analysis of the crude reaction mixture using 1,1,2,2-tetrachloroethane as an internal standard. ^c Values given in the parentheses are isolated product yields. ^d NaI was used as an iodinating reagent in place of I₂. ^e The reaction was carried out under N₂ atmosphere and Co(OAc)₂ (20 mol%) was used. ^f The reaction was carried out in the absence of Co(OAc)₂·4H₂O.

Density Functional Theory Calculations

XII. Transition-State Geometry of the CMD Pathway

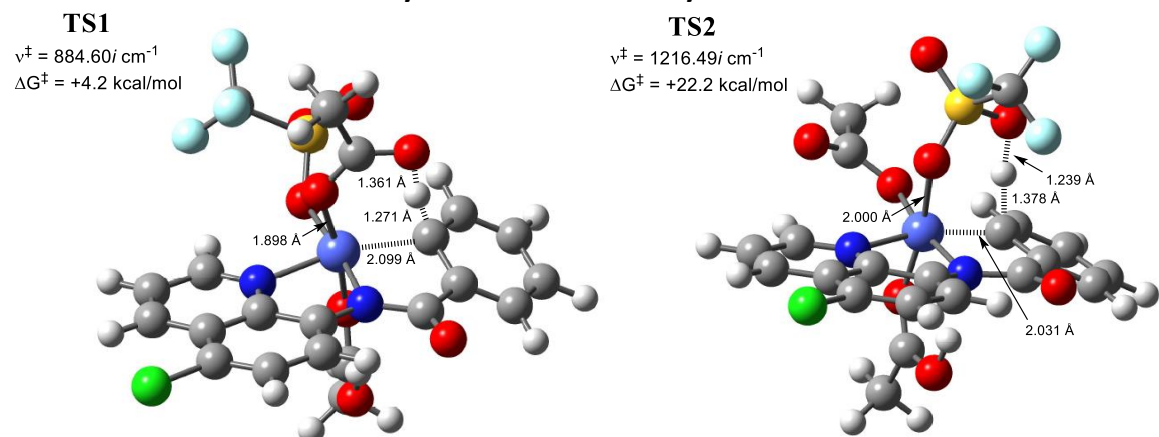


Figure S1. Transition-state geometry of the CMD pathway accelerated by acetate (TS1) and triflate (TS2).

XIII. Energy Surface for CMD Pathway

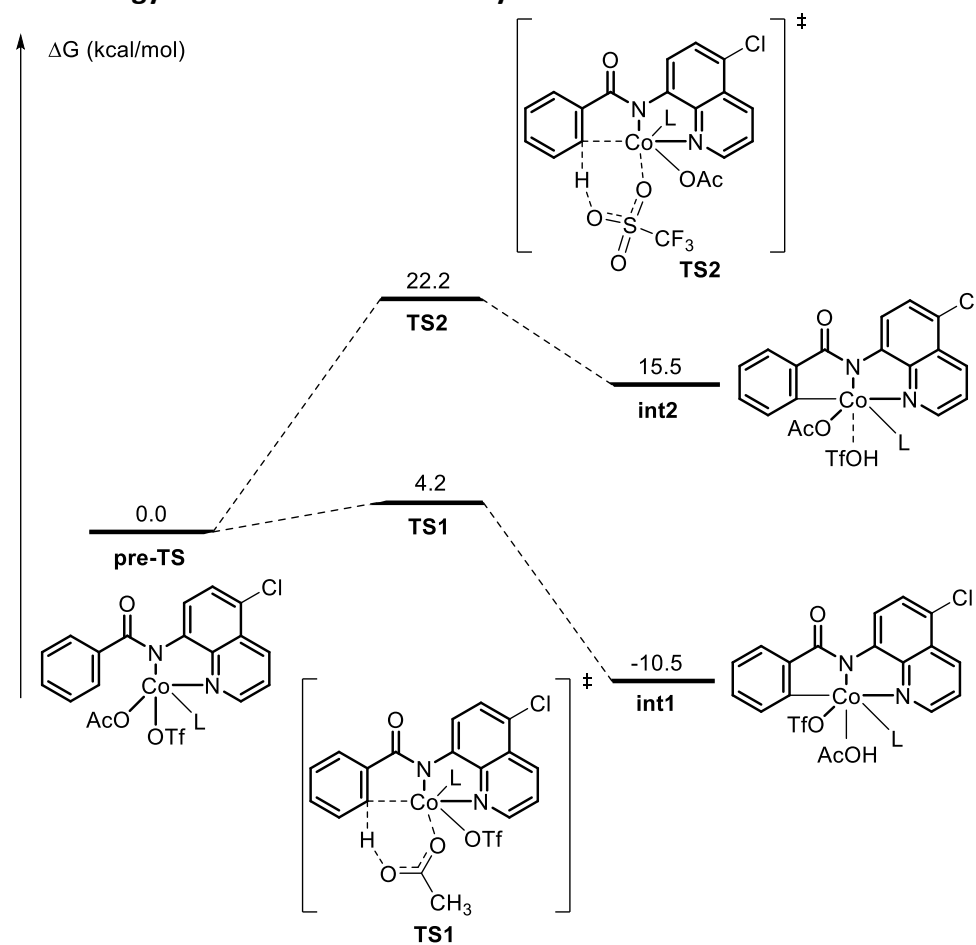


Figure S2. Energy surface for the CMD mechanism accelerated by acetate (TS1) and triflate (TS2). Energies are represented as ΔG (kcal/mol). L = AcOH.

In order to investigate the CMD pathways, we first hypothesized $\text{Co}(\text{OAc})_2(\text{OTf})$ generated from the oxidation of $\text{Co}(\text{OAc})_2$ by $\text{Ag}(\text{I})$ species functions as the actual catalyst. In the CMD pathway, either OAc or OTf can promote the reaction (Figure S2). The energy barrier for the acetate-promoted deprotonation is surprisingly low at 4.2 kcal/mol and is exothermic by 10.5 kcal/mol. On the other hand, the triflate-promoted deprotonation has a high energy barrier of 22.2 kcal/mol. These findings explain the experimental KIE value of 1.08. In addition, we found that the base does not accelerate the CMD process as expected.

XIV. Spin Densities of Proposed Intermediates in SET Pathway

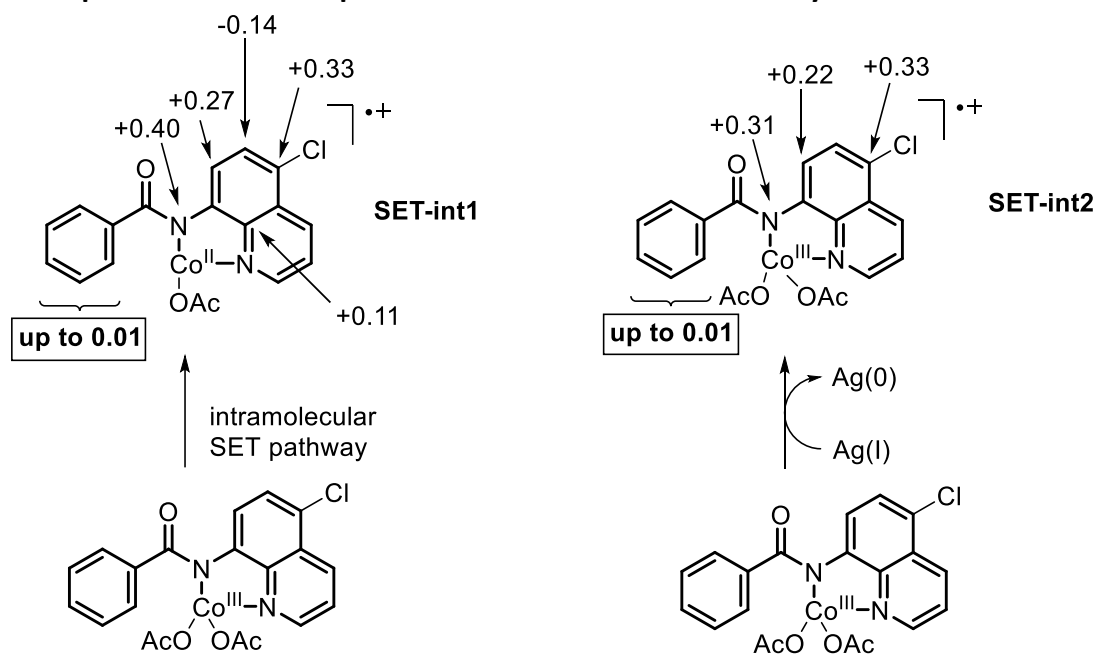


Figure S3. Spin densities of proposed radical cation intermediates in the intra and inter molecular SET pathways. The numbers represent the spin densities.

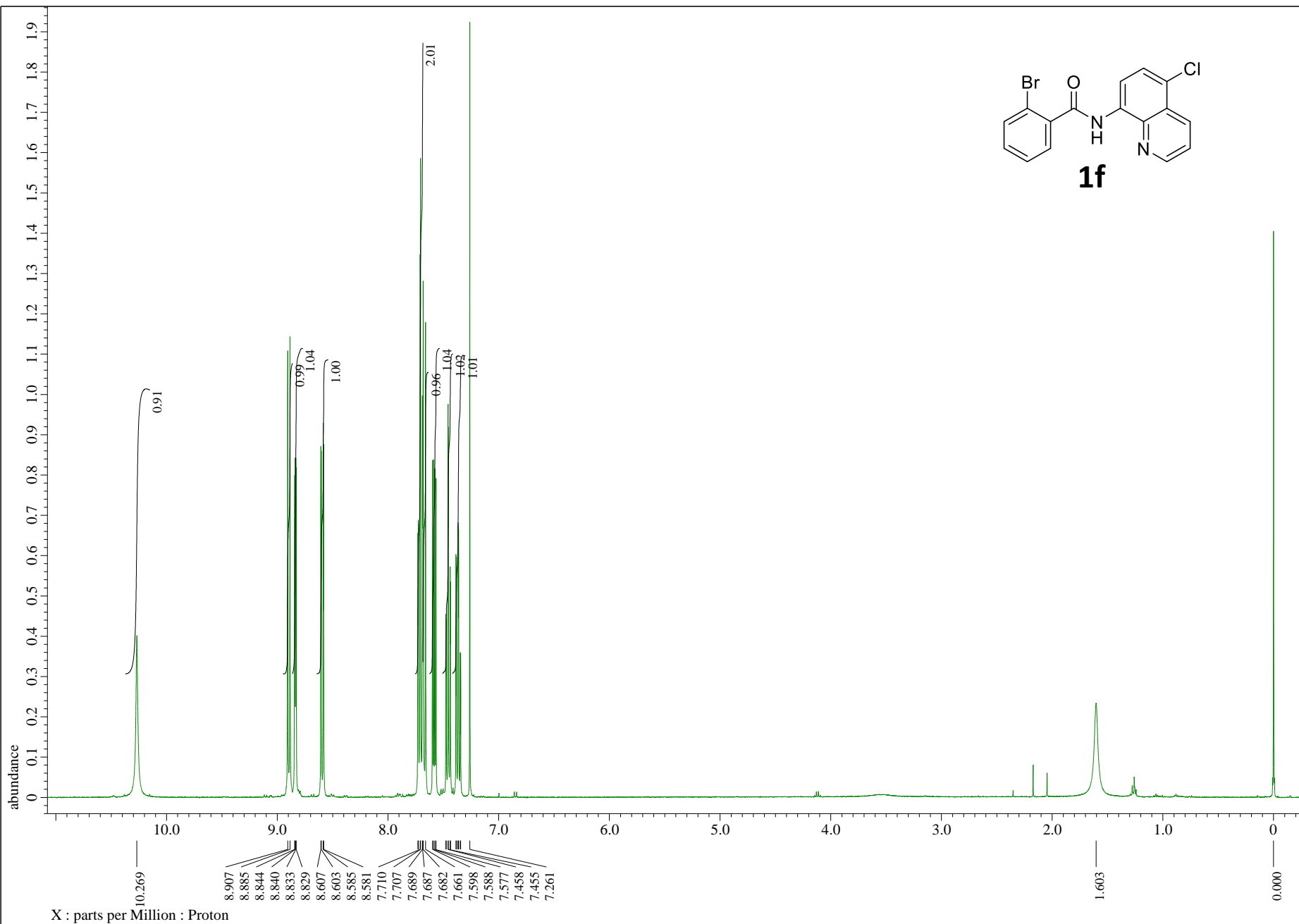
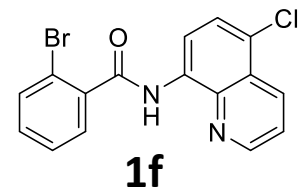
As demonstrated in Scheme 5, there are two proposed paths for the generation of the key intermediate **E** from **A**. Path a involves a SET process and path b involves a CMD pathway. To gain more insights into the mechanism, the spin densities of the radical cation species were calculated. The calculations show that electron spins exist on the quinoline ring (especially on the C5 and C7 position) rather than on the benzene ring. This result indicates that the SET process is not involved in our catalytic system. This result also supports the experimental results shown in Scheme 2, in which the iodination took place at the C5 position of the quinoline ring in **1b** leading to the production of **1b'** and **2b'** as the major products of the reaction.

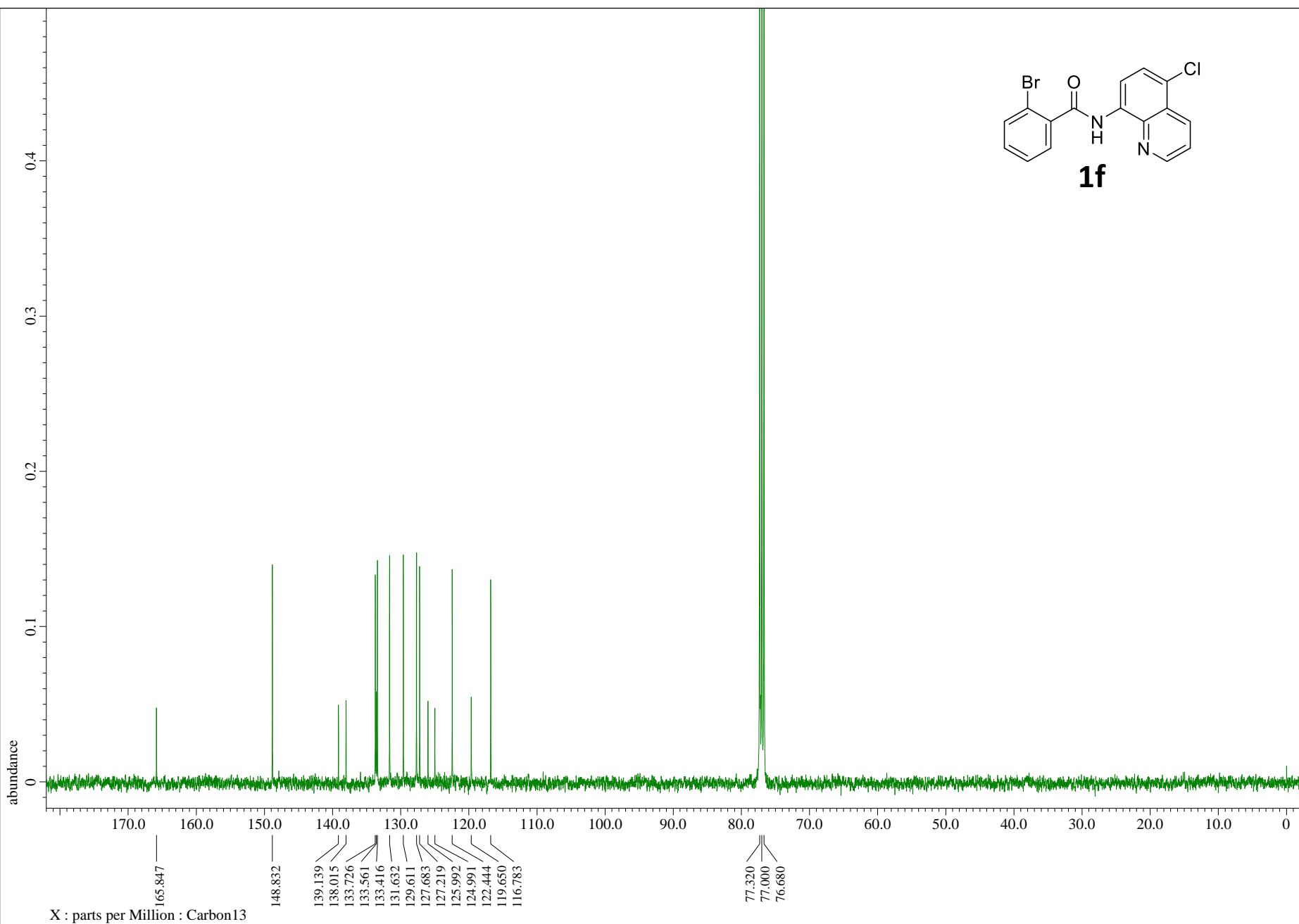
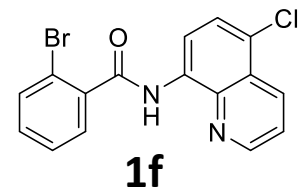
XV. Computational Details

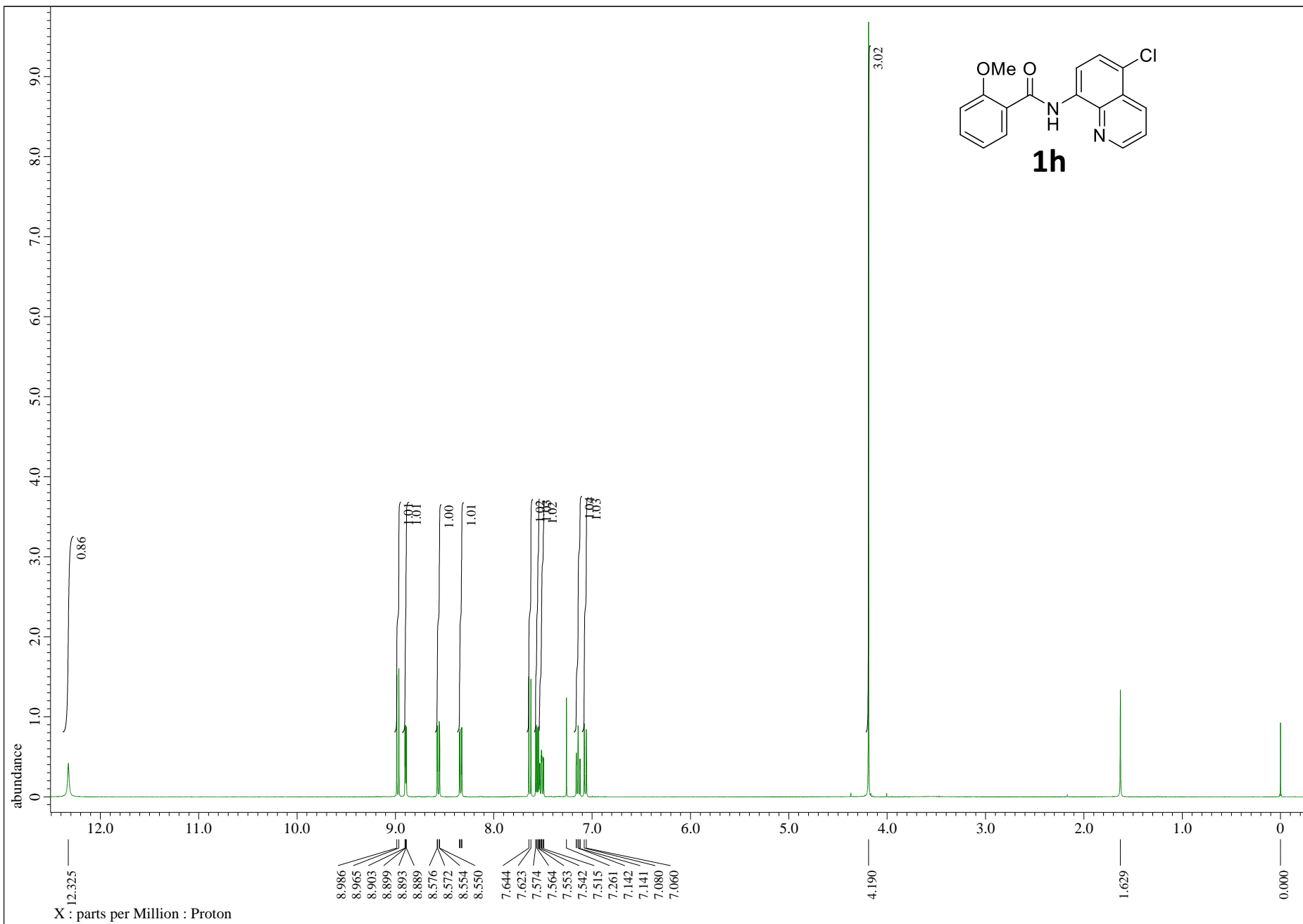
Calculations were performed with the Gaussian 09 (G09) program.⁴ Geometry optimizations for all reported structures were performed at the B3LYP/[[6-31G(d)](for C, H, N, O, F, S, Cl) + LanI2dz (for Co) SCRF = (PCM, solvent = dichloroethane)} level of theory with the corresponding effective core potential for Co.⁵⁻⁷ Single point energy calculations are performed at the B3LYP/[[6-311+G(d,p)] (for C, H, N, O, F, S, Cl) + SDD (for Co) SCRF = (PCM, solvent = dichloroethane)} level of theory with the corresponding effective core potentials for Co.⁸⁻¹⁰ PCM¹¹⁻¹³ solvent effects are incorporated for all calculations with dichloroethane as the solvent. TS geometry was characterized by vibrational analysis, which checked whether the obtained geometry has single imaginary frequencies (ν^\ddagger). From TSs, reaction paths were traced by the intrinsic reaction coordinate (IRC) method^{14, 15} to obtain the energy-minimum geometries. Energy changes were shown by the use of Gibbs free energies (T = 298.15 K and P = 1 atm).

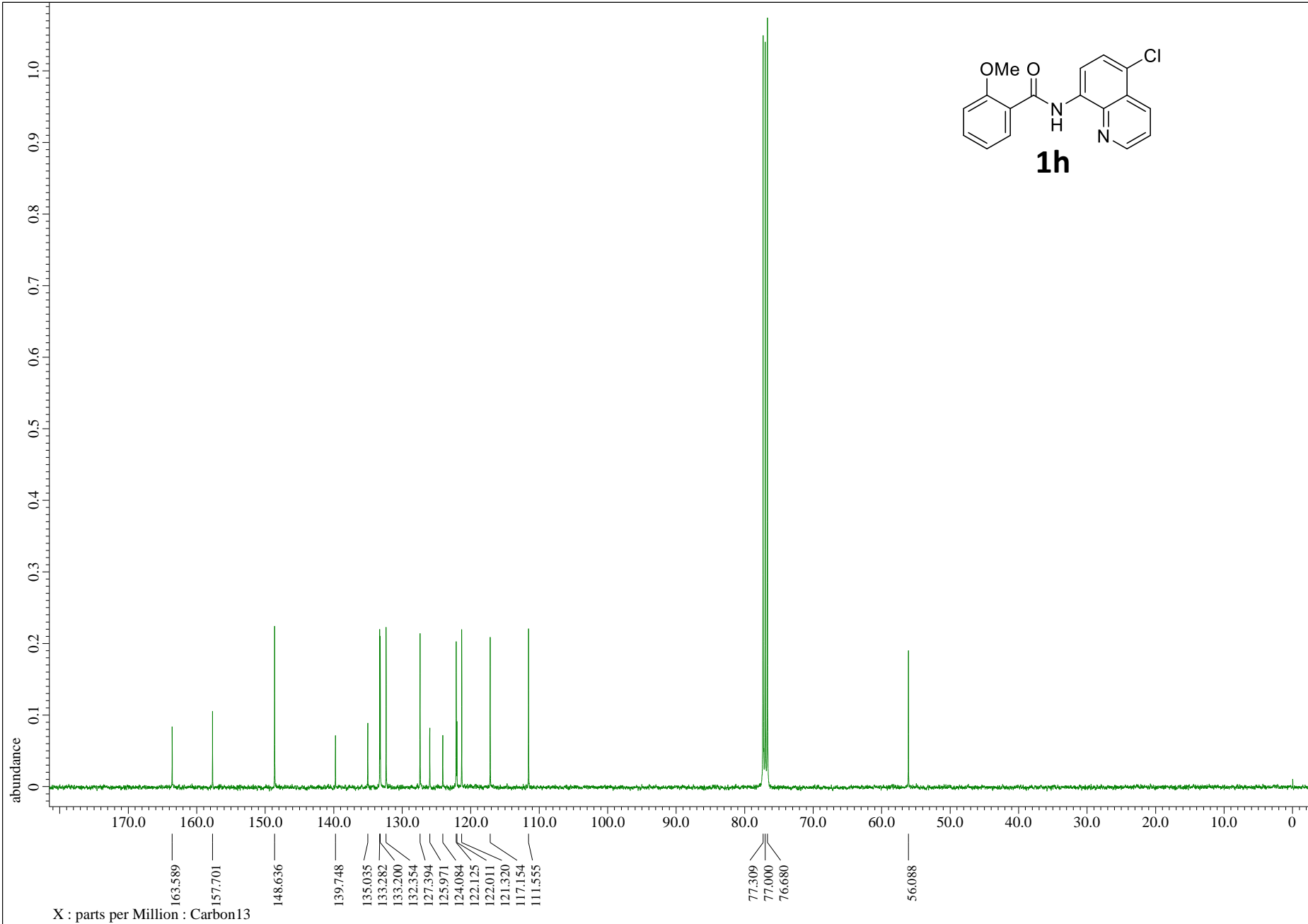
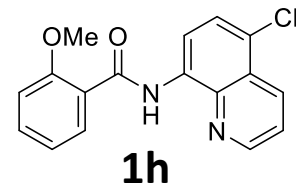
XVI. References:

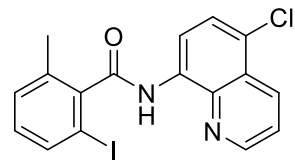
1. G. He, S.-Y. Zhang, W. A. Nack, Q. Li and G. Chen, *Angew. Chem. Int. Ed.*, 2013, **52**, 11124.
2. (a) Y. Aihara, M. Tobisu, Y. Fukumoto and N. Chatani, *J. Am. Chem. Soc.*, 2014, **136**, 15509; (b) Y. Aihara and N. Chatani, *ACS Catal.*, 2016, **6**, 4323.
3. (a) Y. Ano, M. Tobisu and N. Chatani, *Org. Lett.*, 2012, **14**, 354; (b) Y. Aihara and N. Chatani, *Chem. Sci.*, 2012, **4**, 664; (c) K. Shibata and N. Chatani, *Org. Lett.*, 2014, **16**, 5148.
4. Gaussian 09, Revision **D.01**, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, M. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford, CT, 2009.
5. P. J. Hay and W. R. Wadt, *J. Chem. Phys.*, 1985, **82**, 270-283.
6. P. J. Hay and W. R. Wadt, *J. Chem. Phys.*, 1985, **82**, 299-310.
7. W. R. Wadt and P. J. Hay, *J. Chem. Phys.*, 1985, **82**, 284-298.
8. D. Andrae, U. Haussermann, M. Dolg, H. Stoll and H. Preuss, *Theor Chim Acta*, 1990, **77**, 123-141.
9. G. Igelmann, H. Stoll and H. Preuss, *Mol. Phys.*, 1988, **65**, 1321-1328.
10. L. Vonszentpaly, P. Fuentealba, H. Preuss and H. Stoll, *Chem. Phys. Lett.*, 1982, **93**, 555-559.
11. E. Cancès, B. Mennucci and J. Tomasi, *J. Chem. Phys.*, 1997, **107**, 3032-3041.
12. M. Cossi, V. Barone, B. Mennucci and J. Tomasi, *Chem. Phys. Lett.*, 1998, **286**, 253-260.
13. B. Mennucci and J. Tomasi, *J. Chem. Phys.*, 1997, **106**, 5151-5158.
14. K. Fukui, *J. Phys. Chem.*, 1970, **74**, 4161-4163.
15. C. Gonzalez and H. B. Schlegel, *J. Chem. Phys.*, 1989, **90**, 2154-2161.



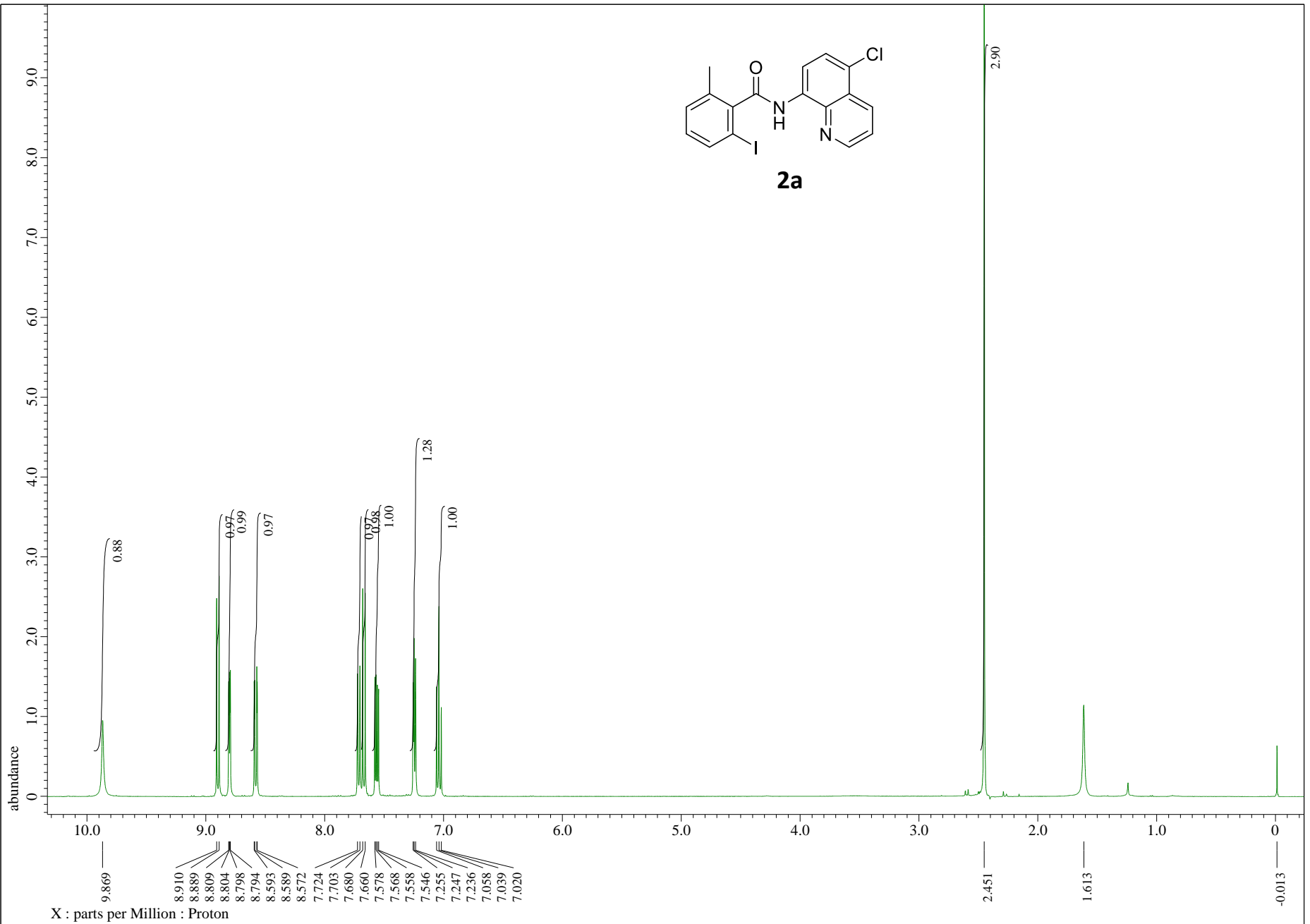


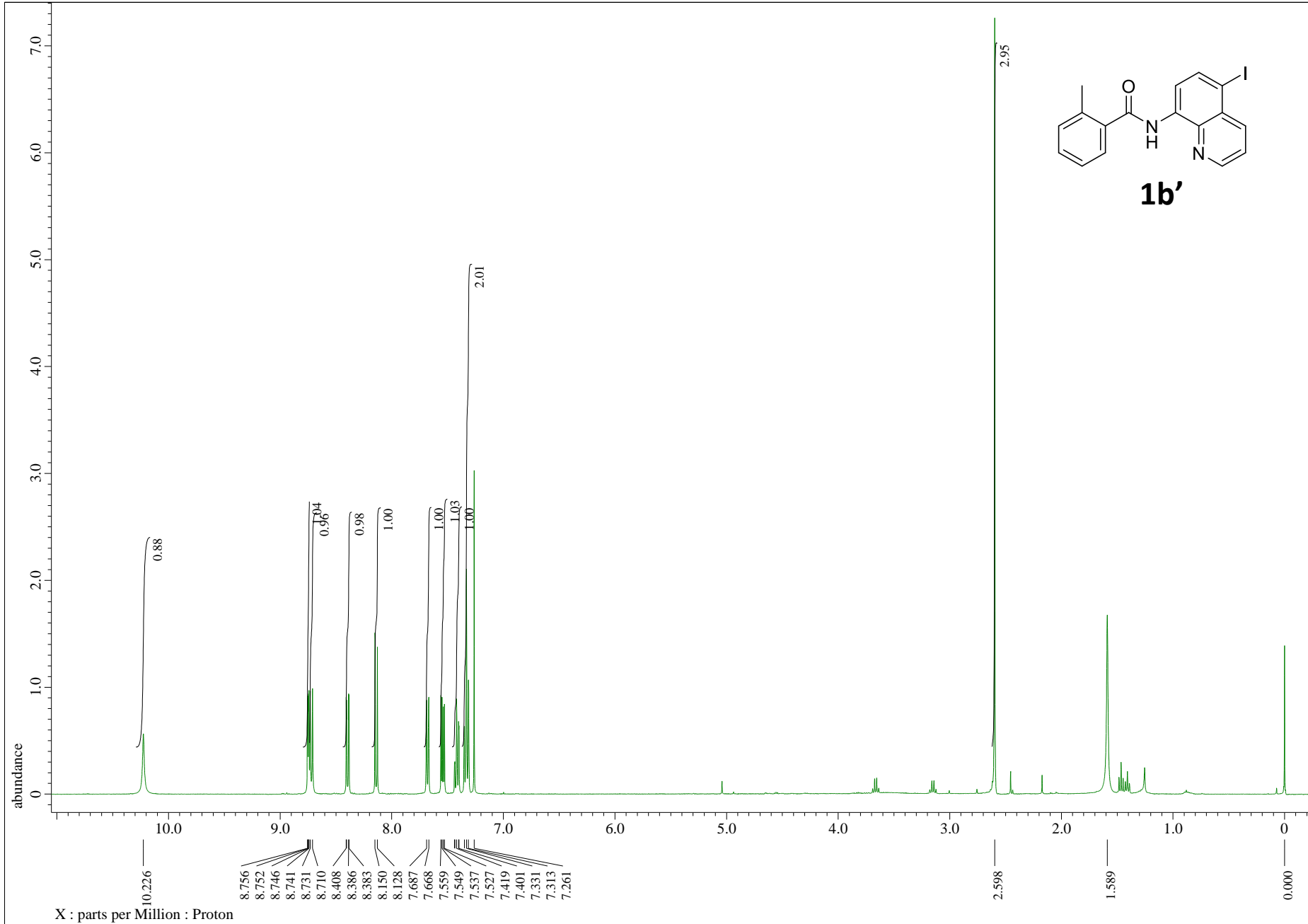


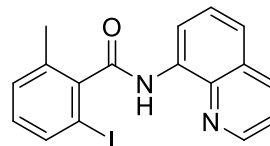




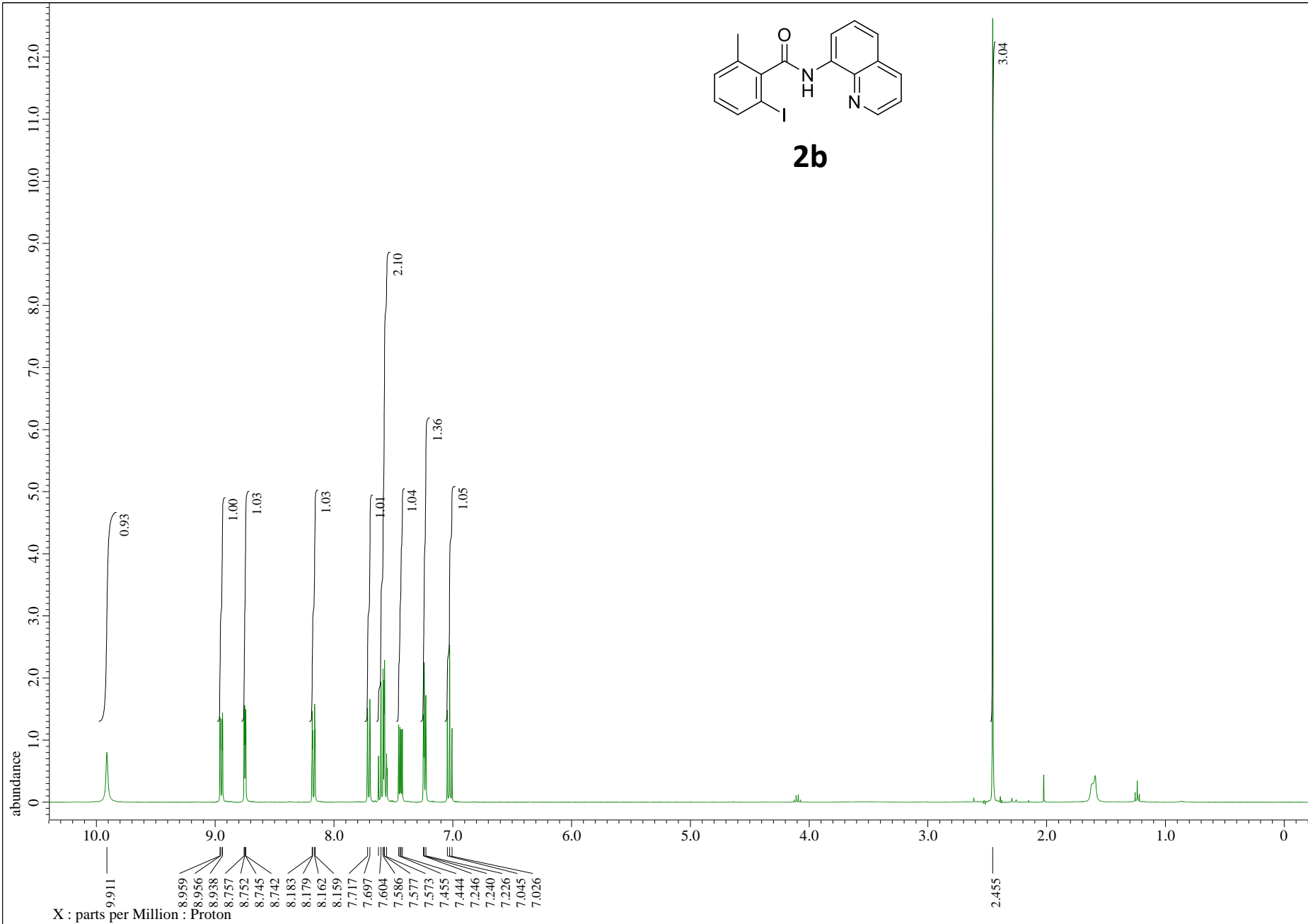
2a

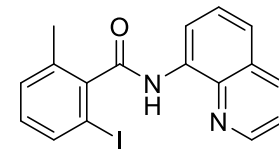




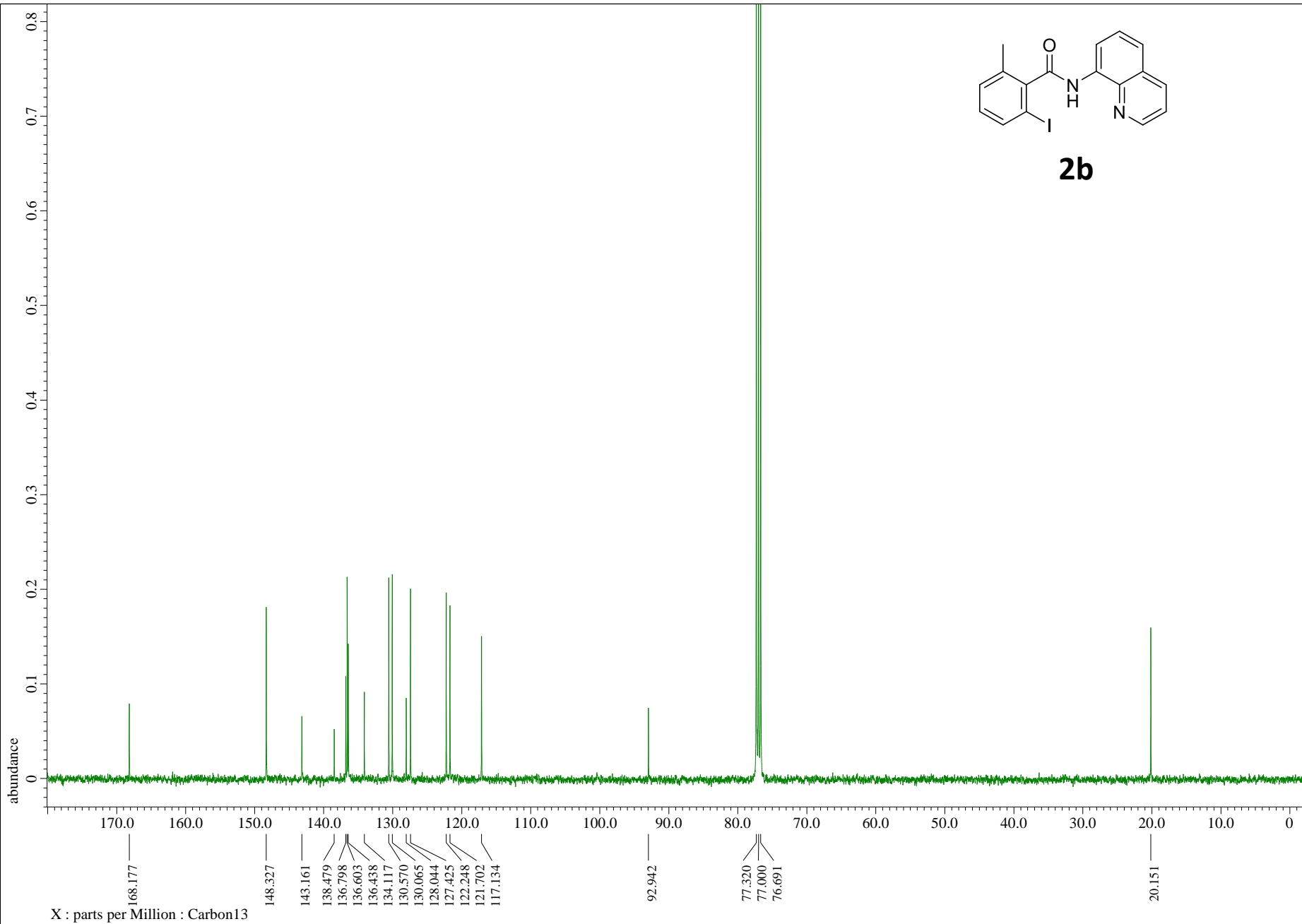


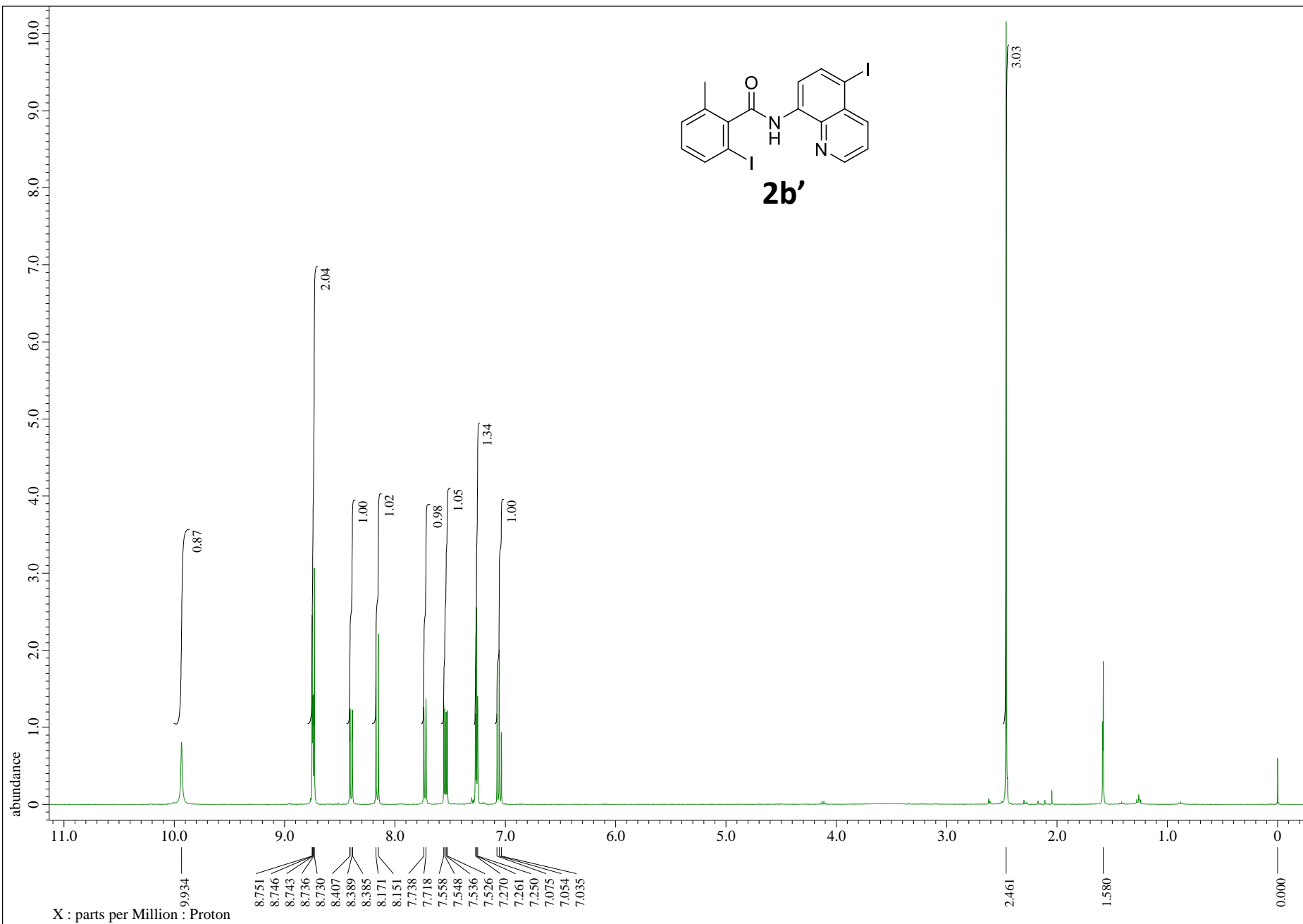
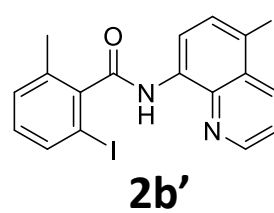
2b

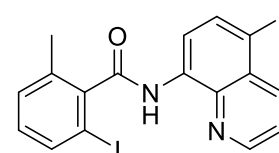




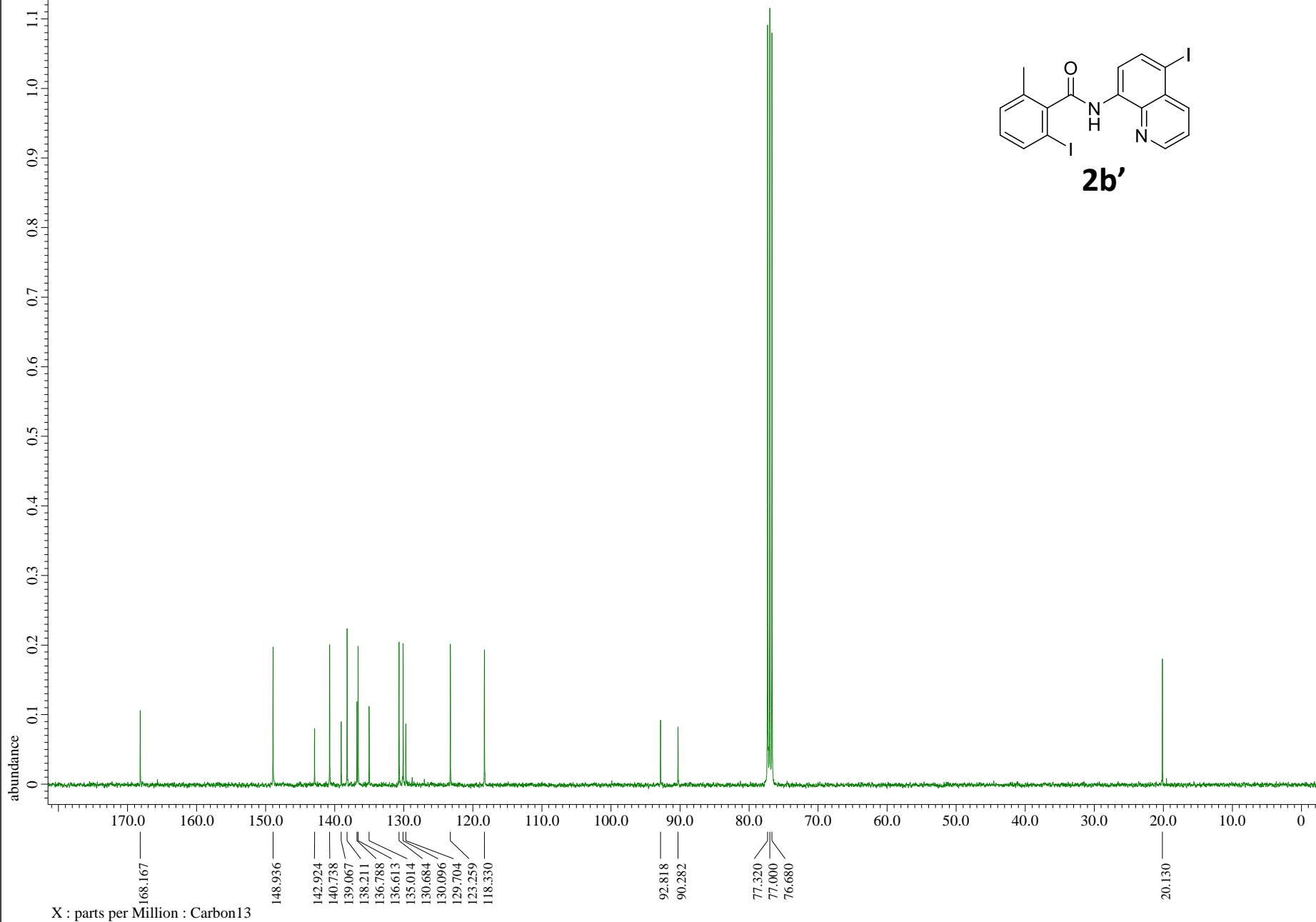
2b

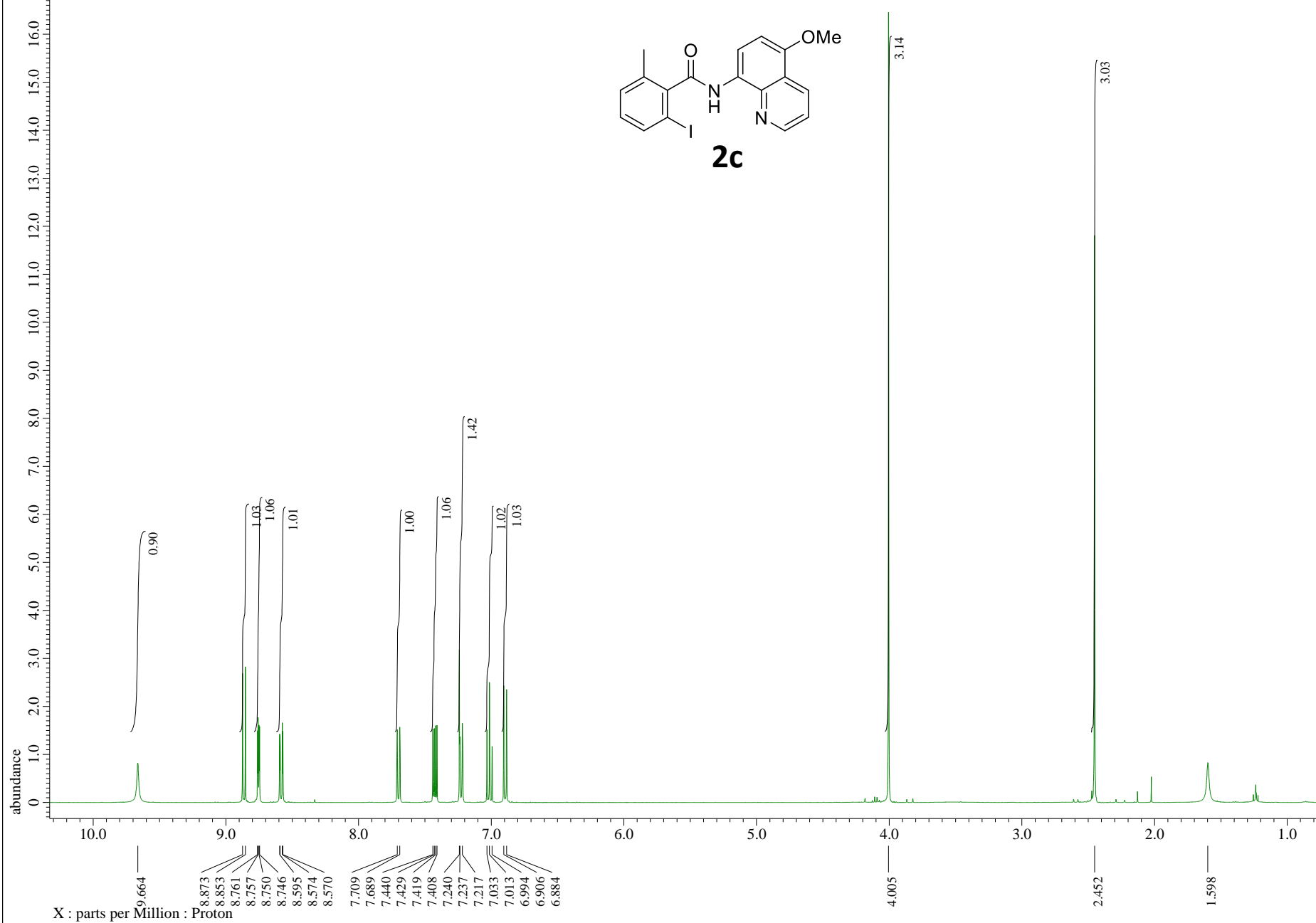
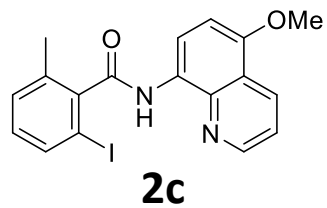


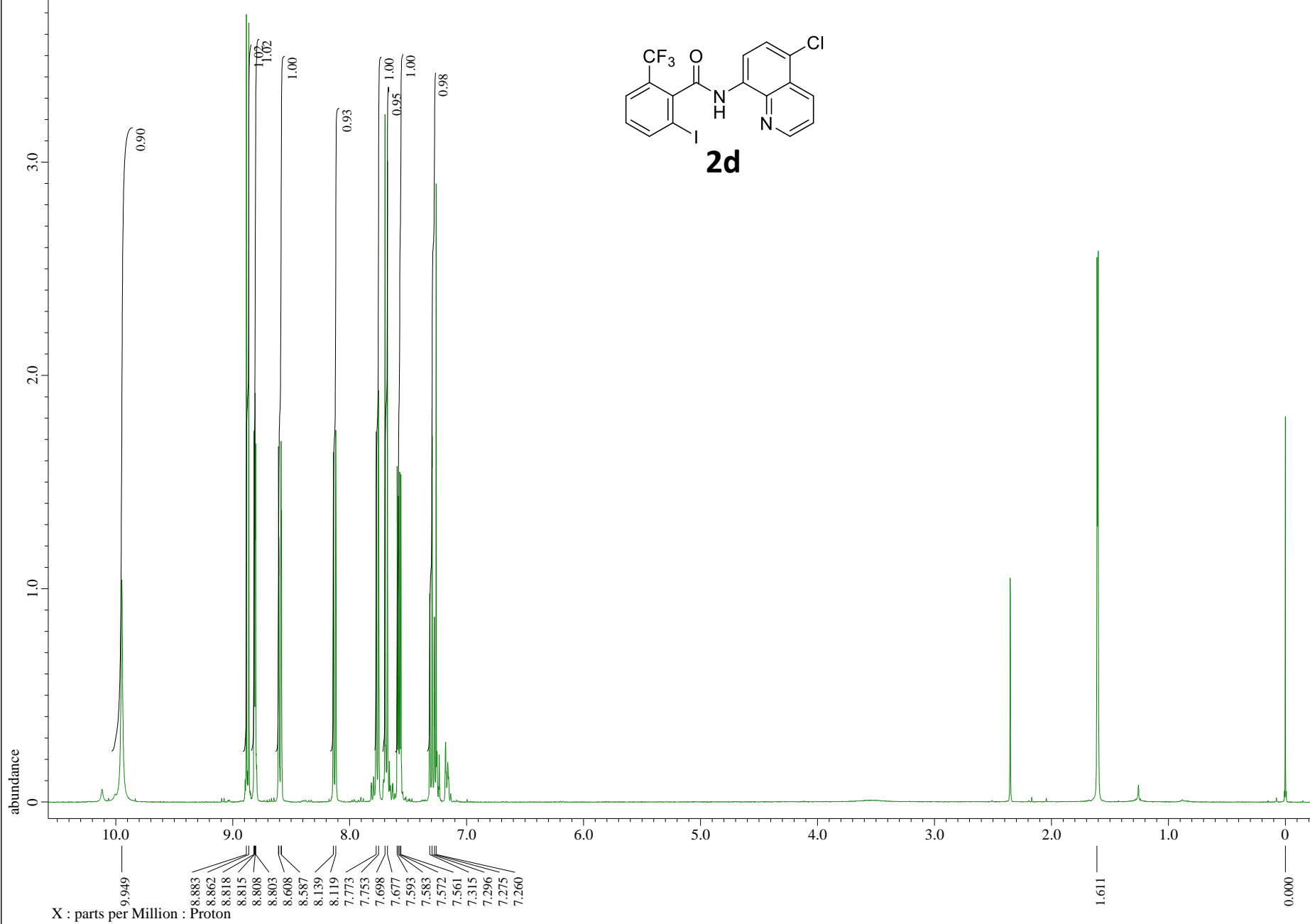
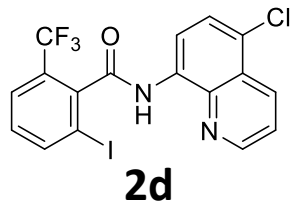


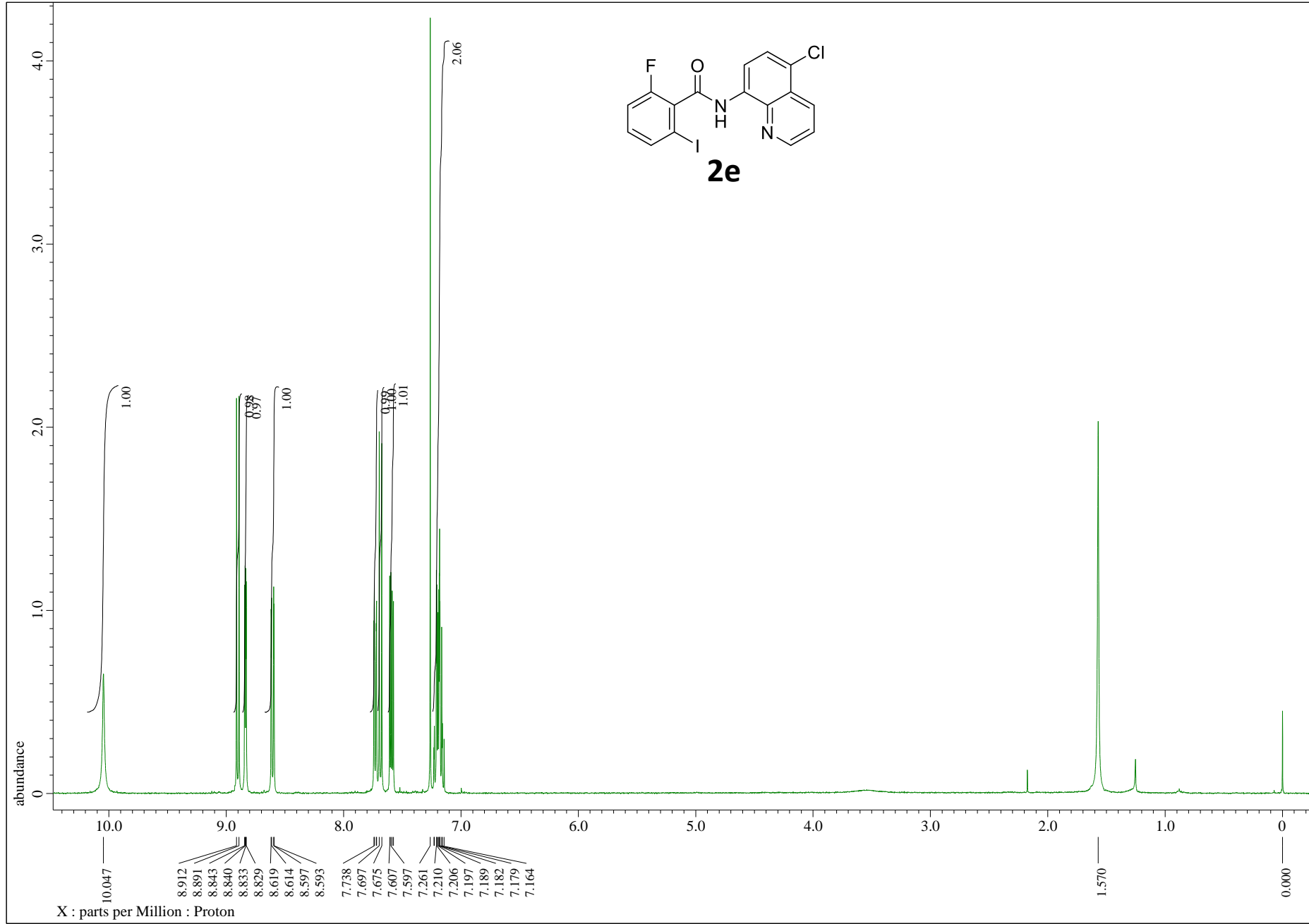
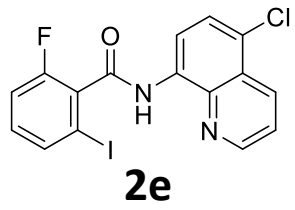


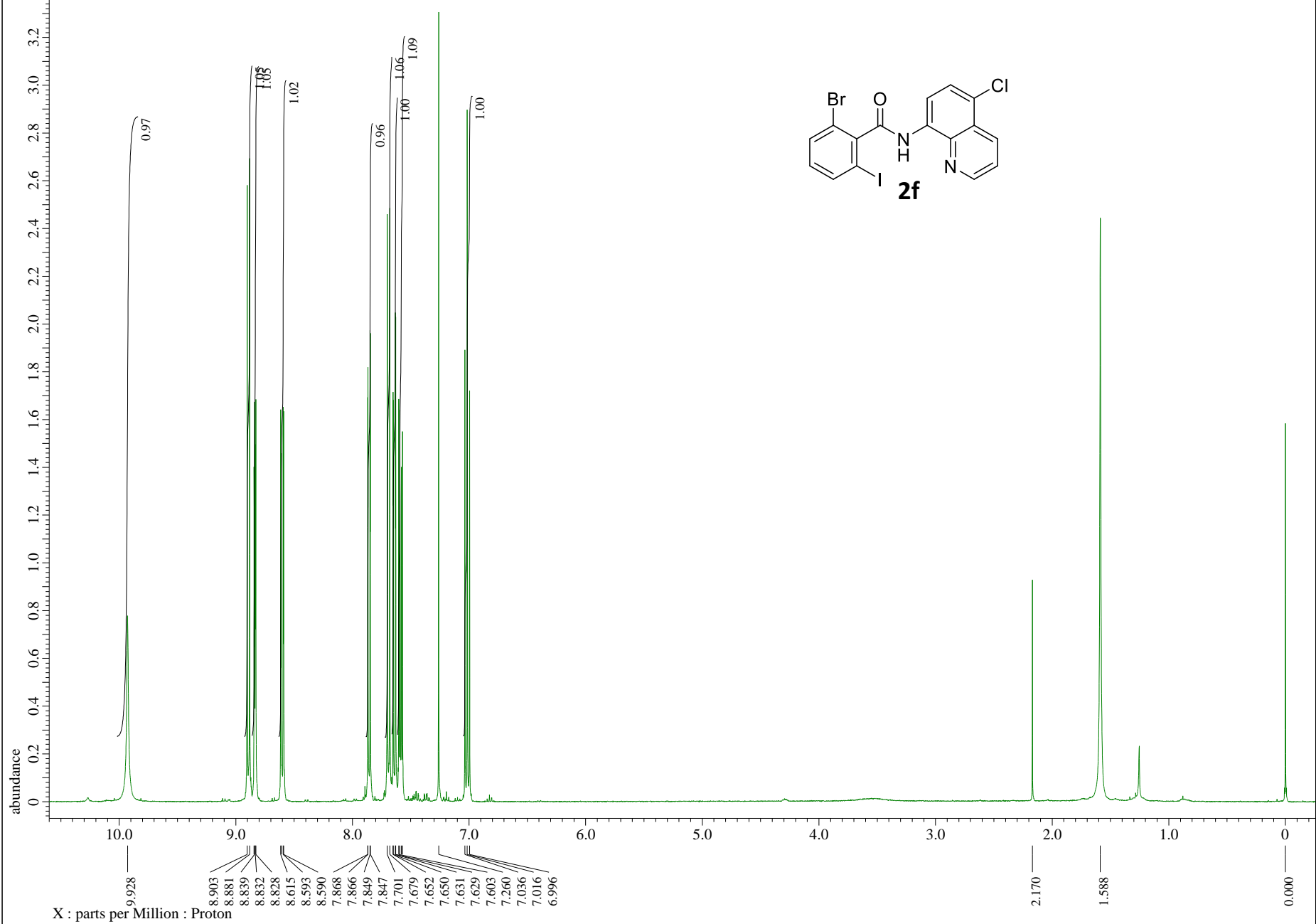
2b'

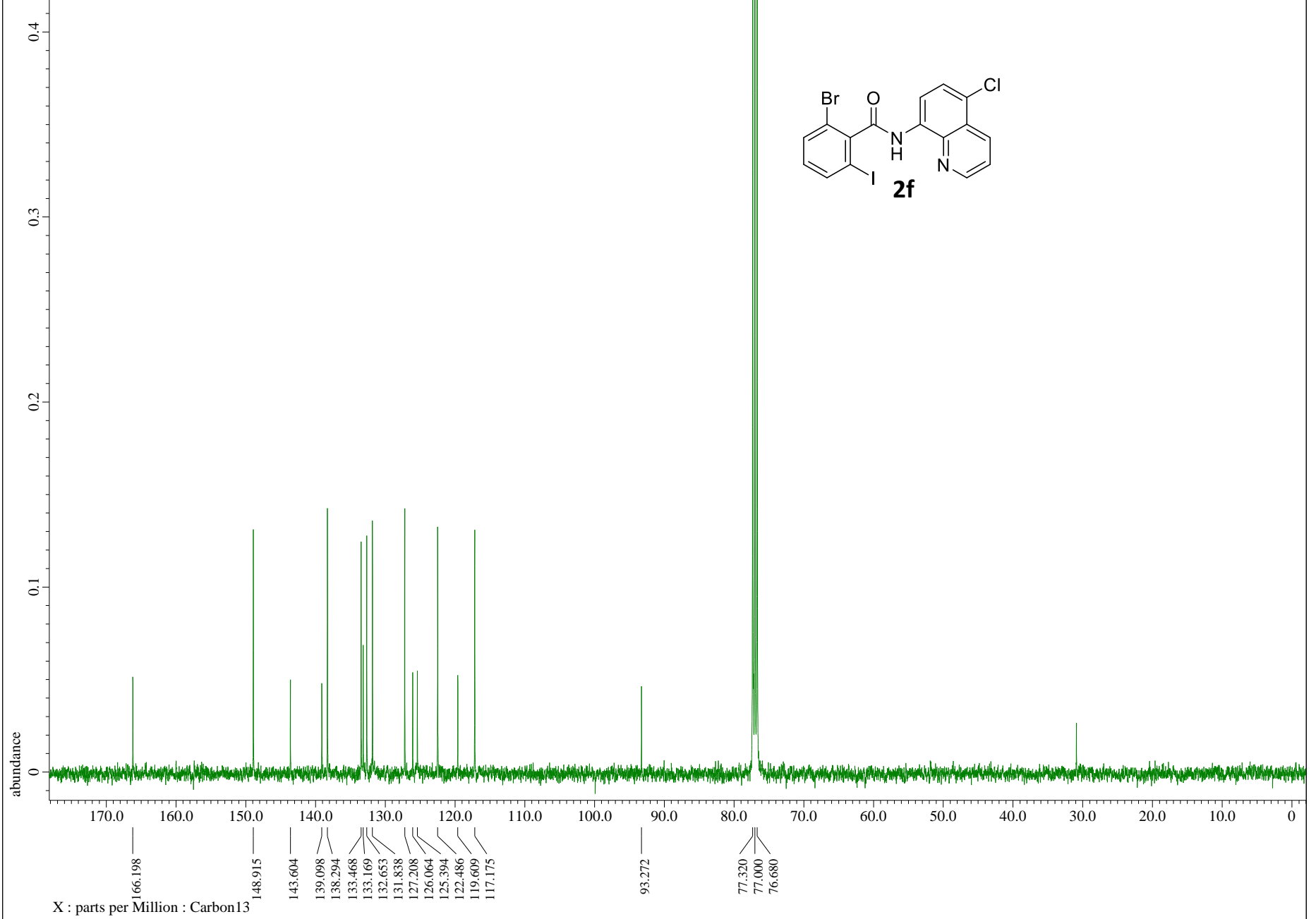
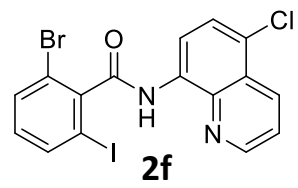


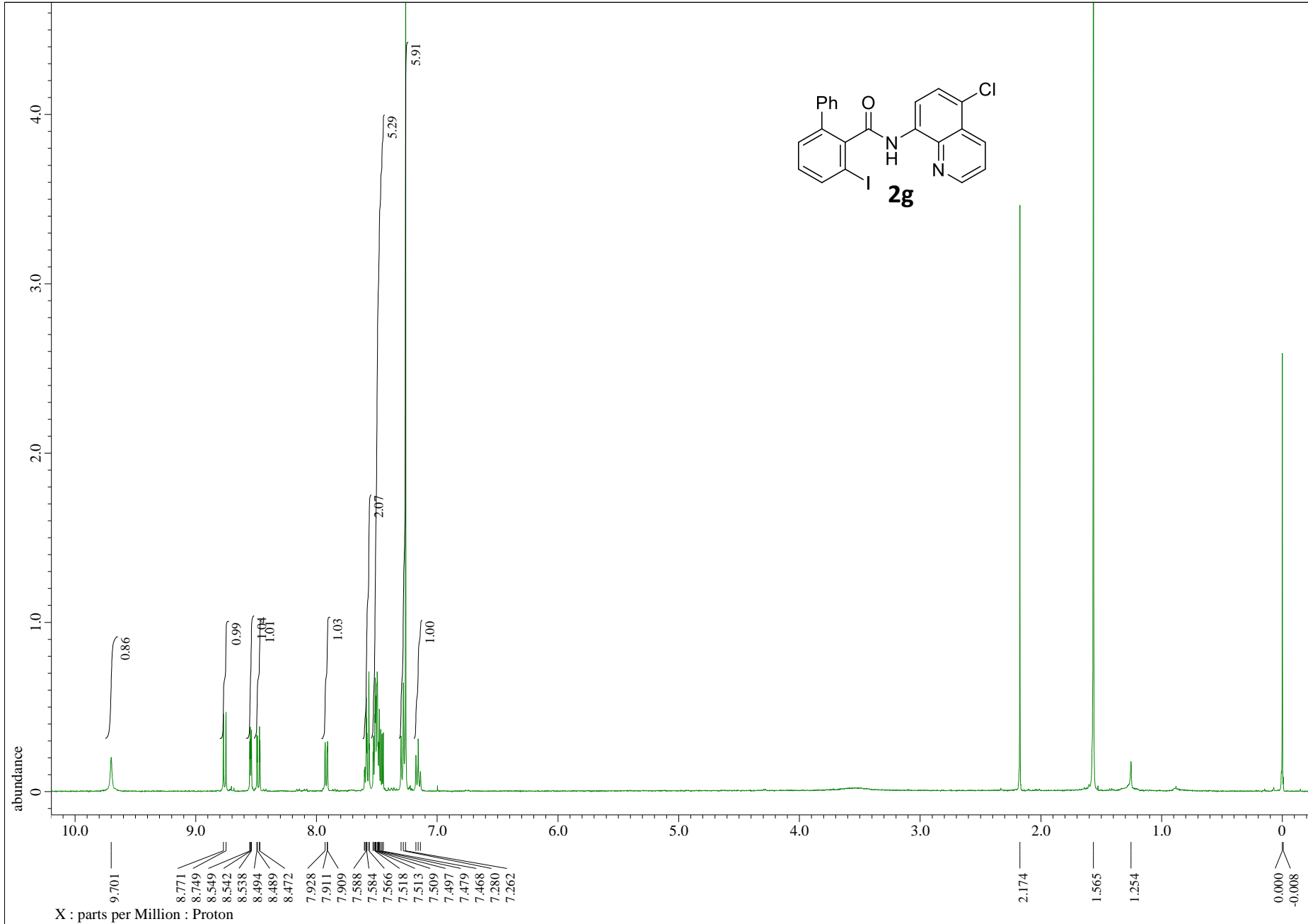


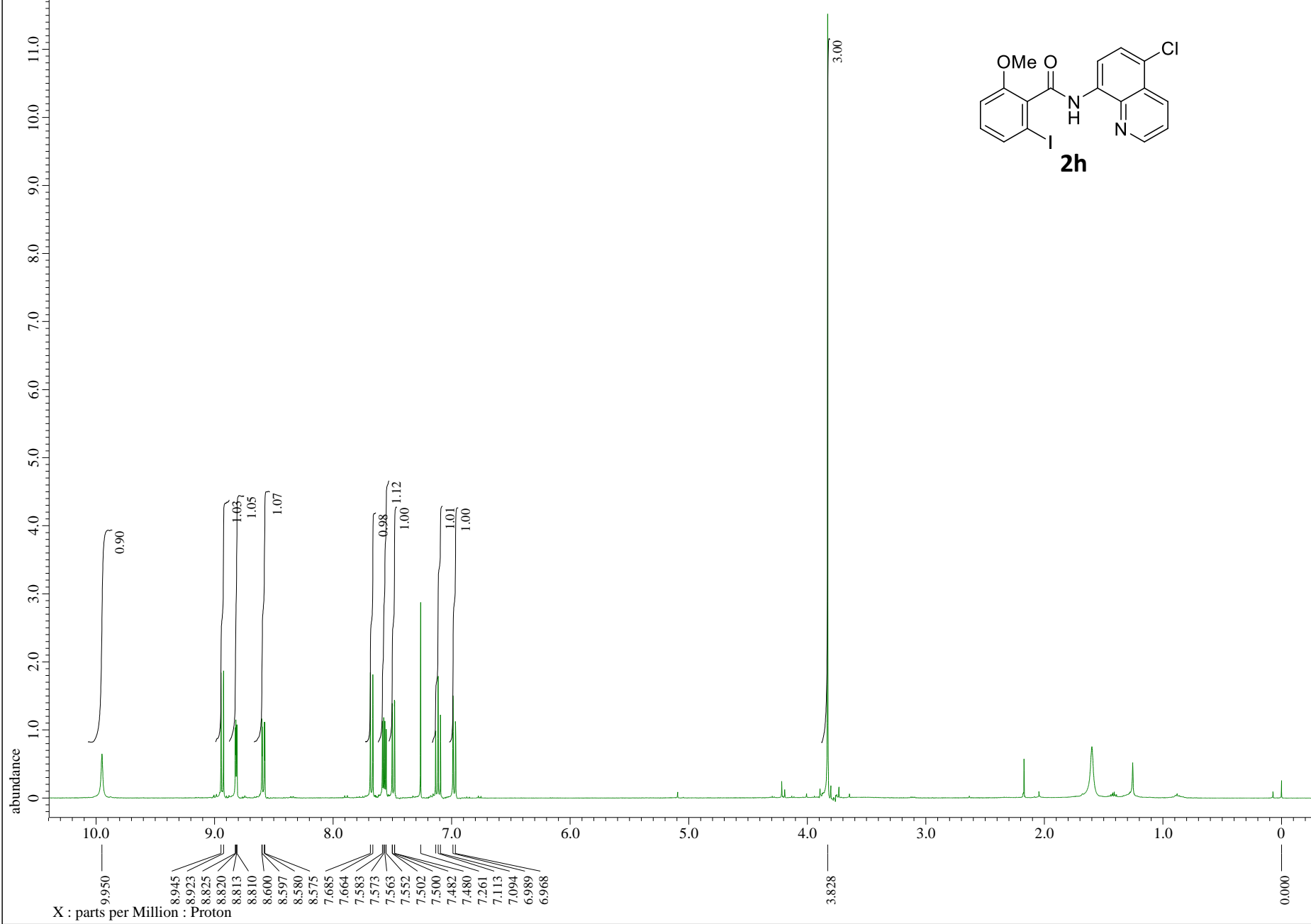


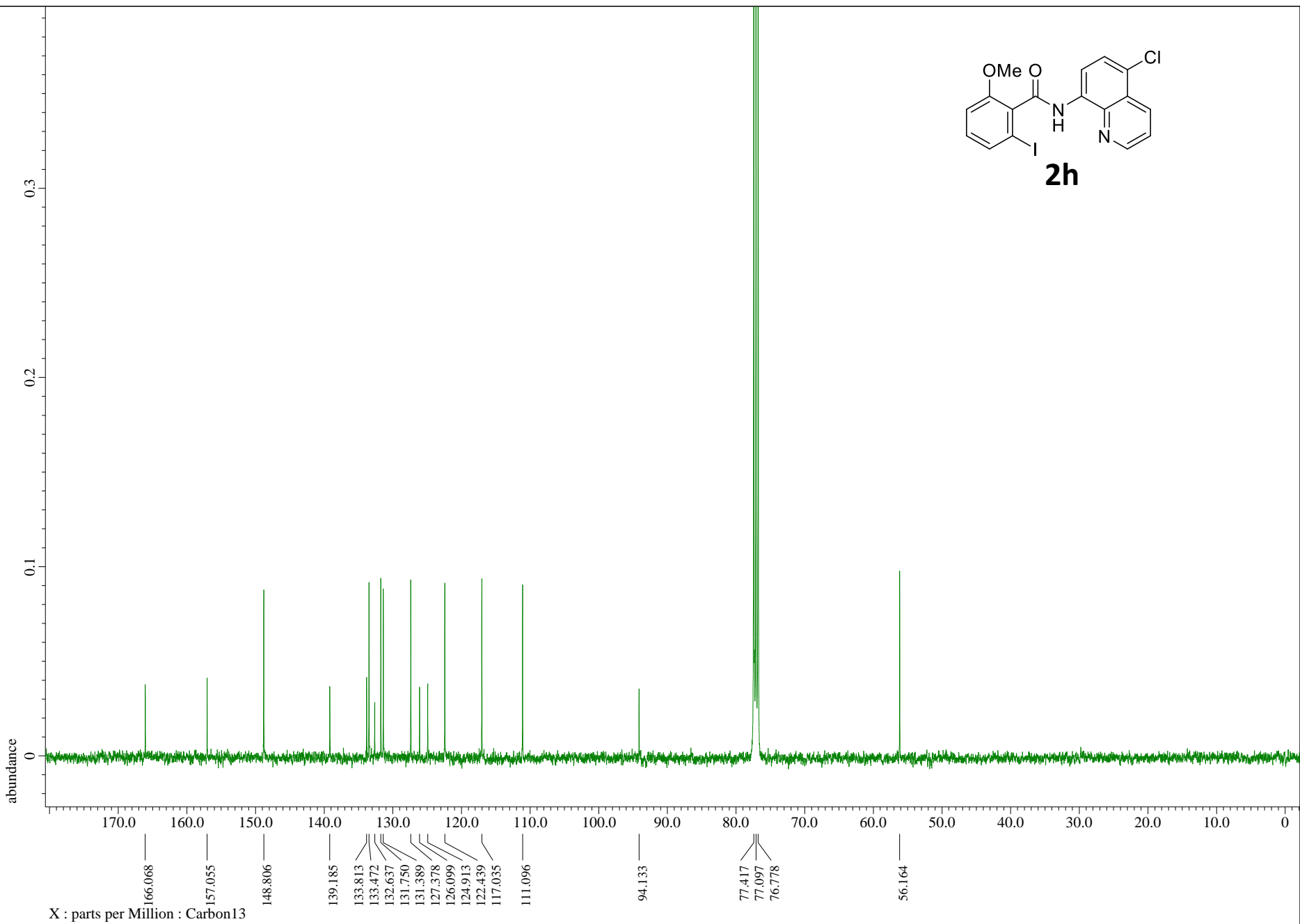
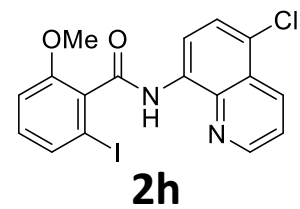


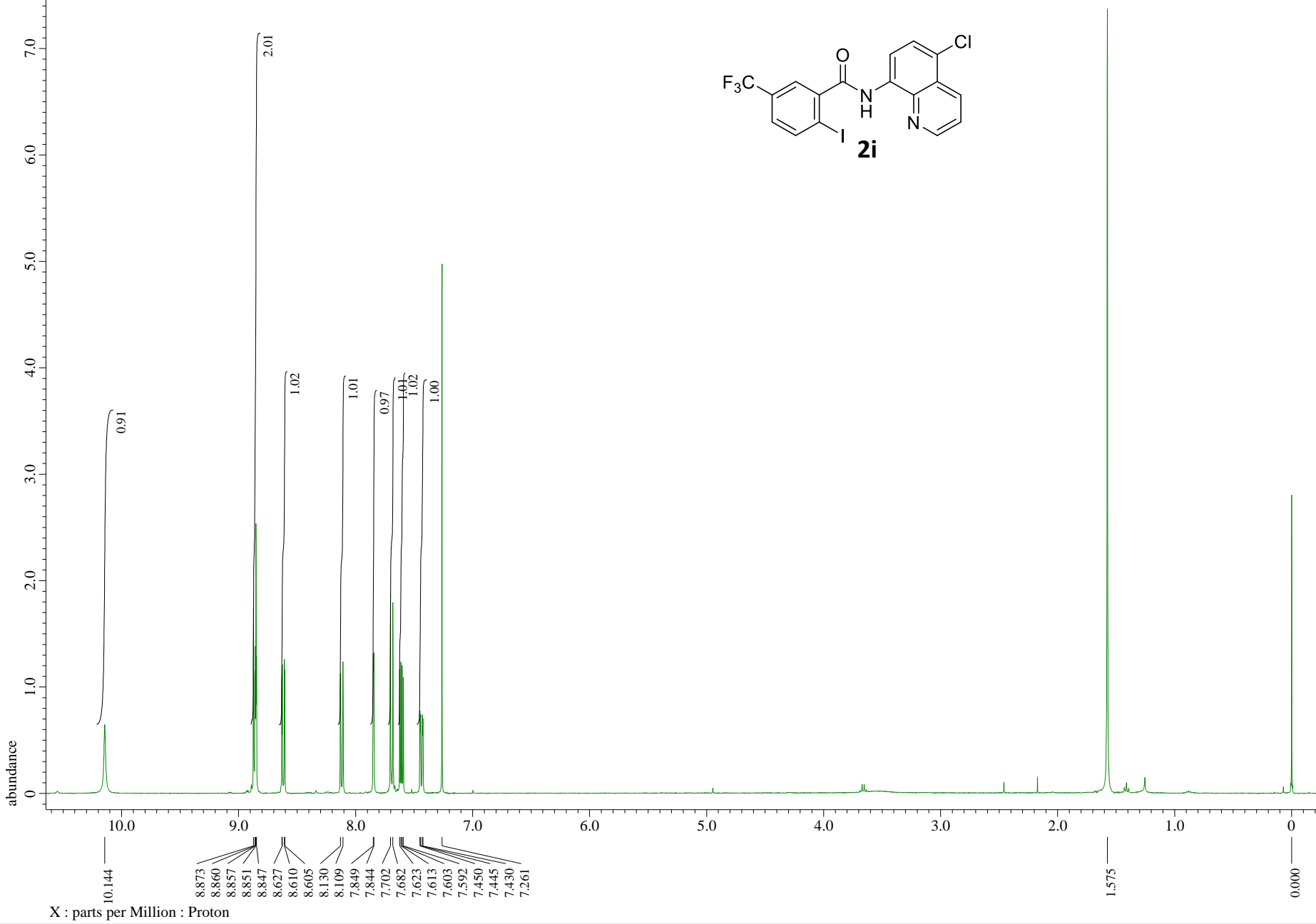
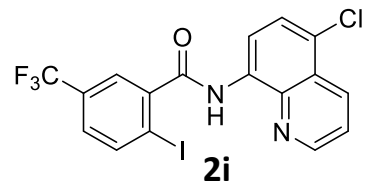


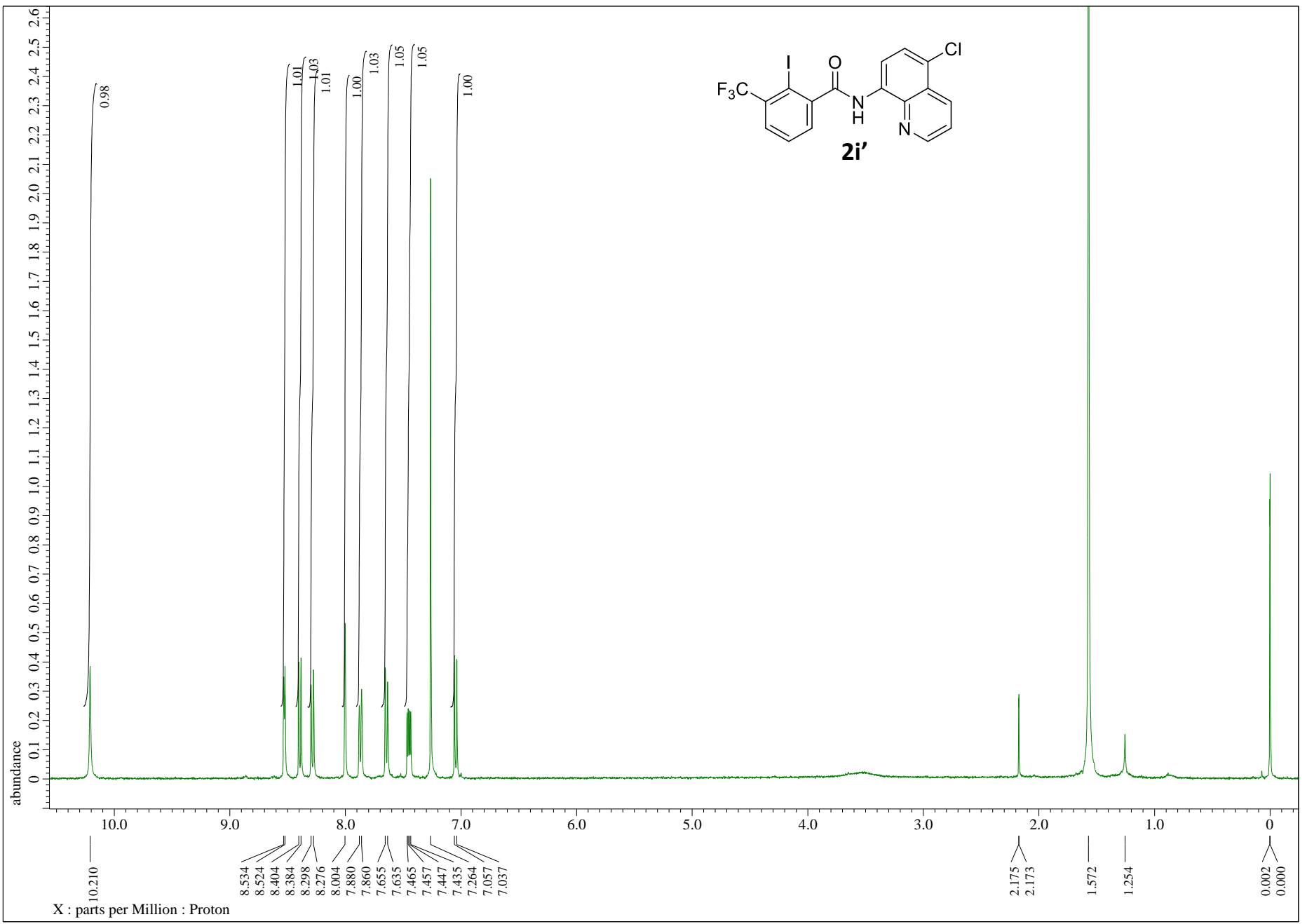


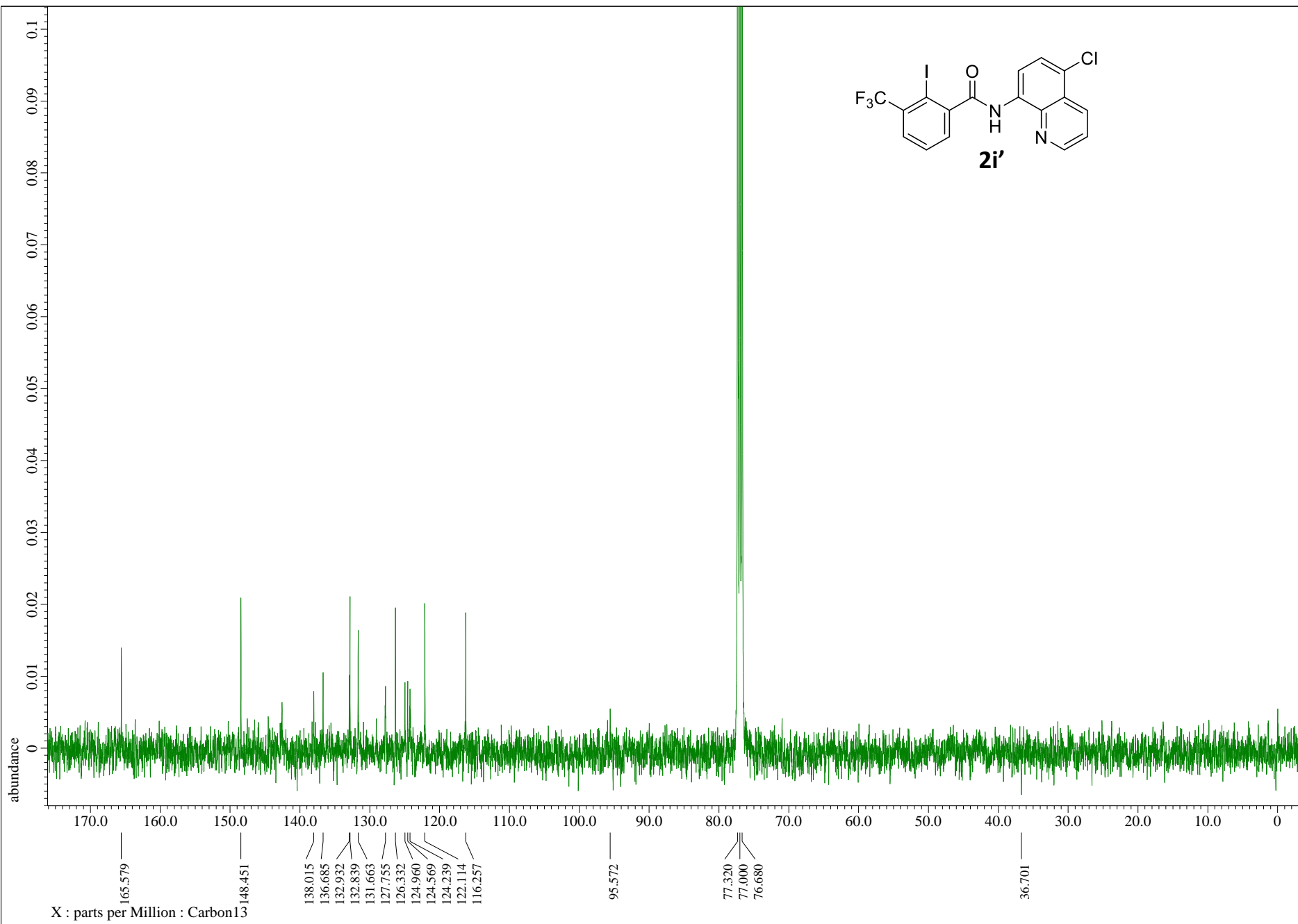
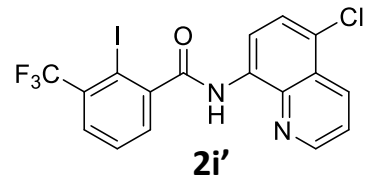


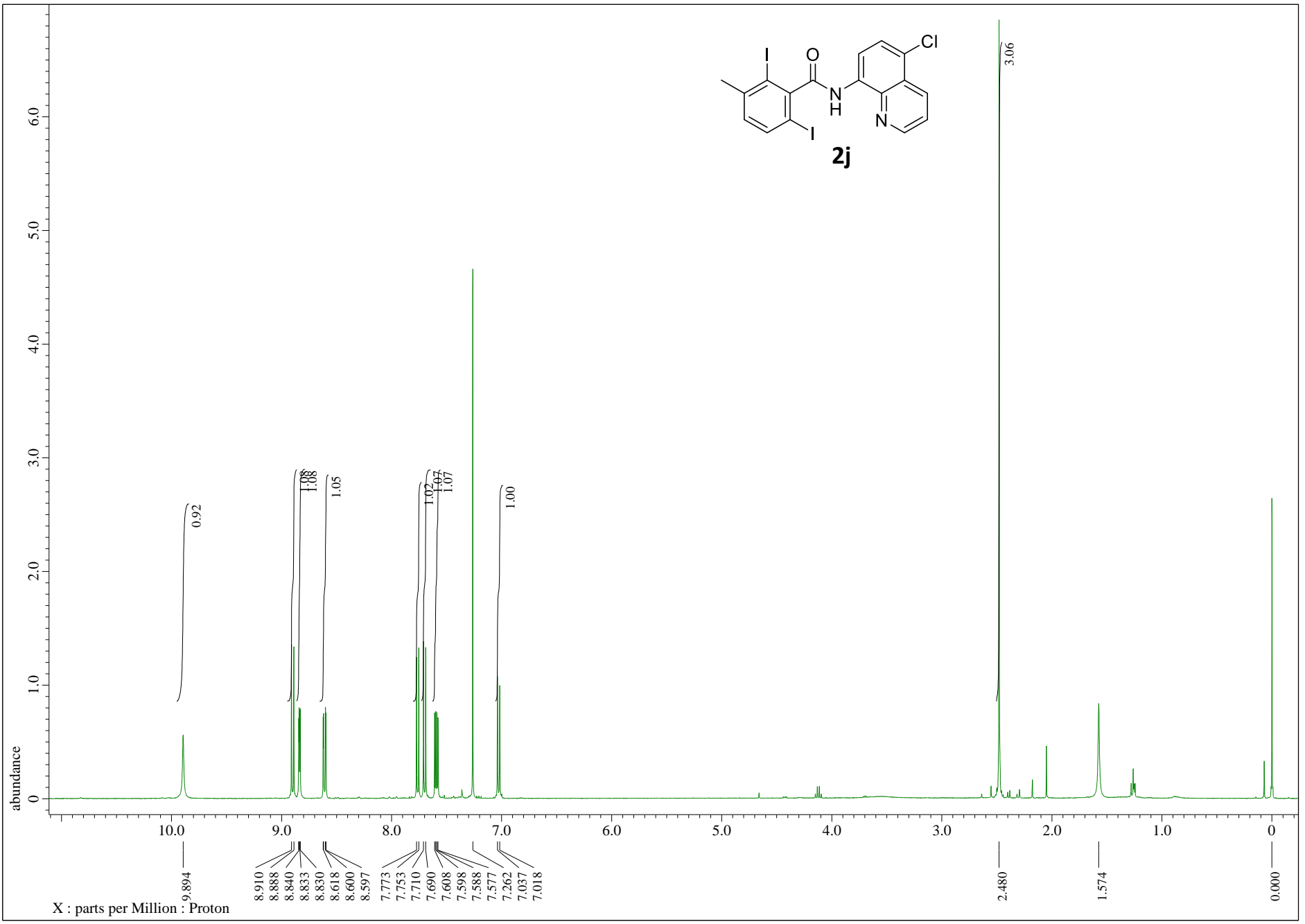
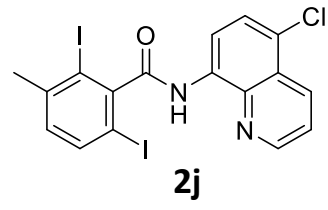


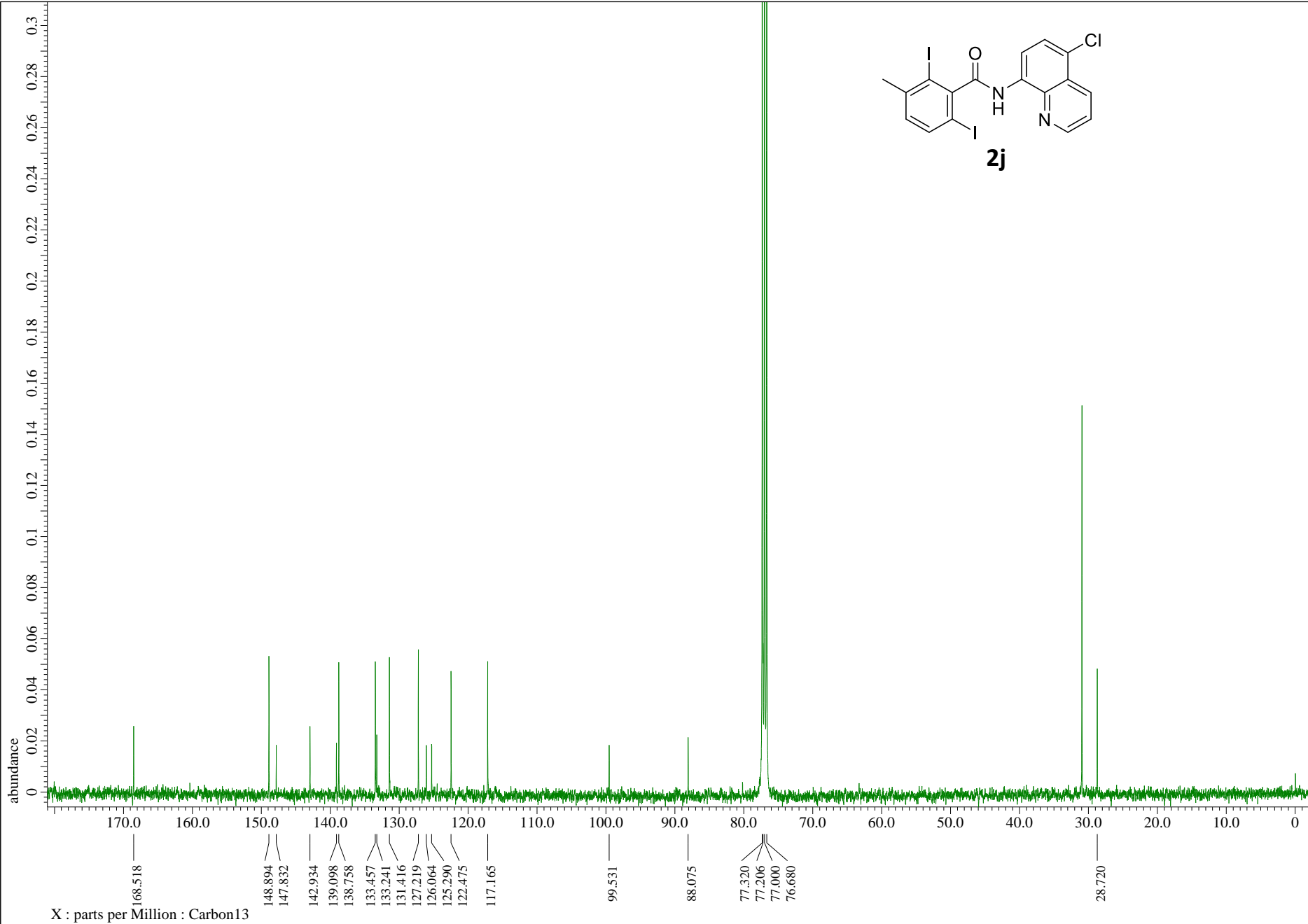
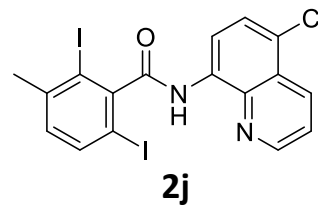


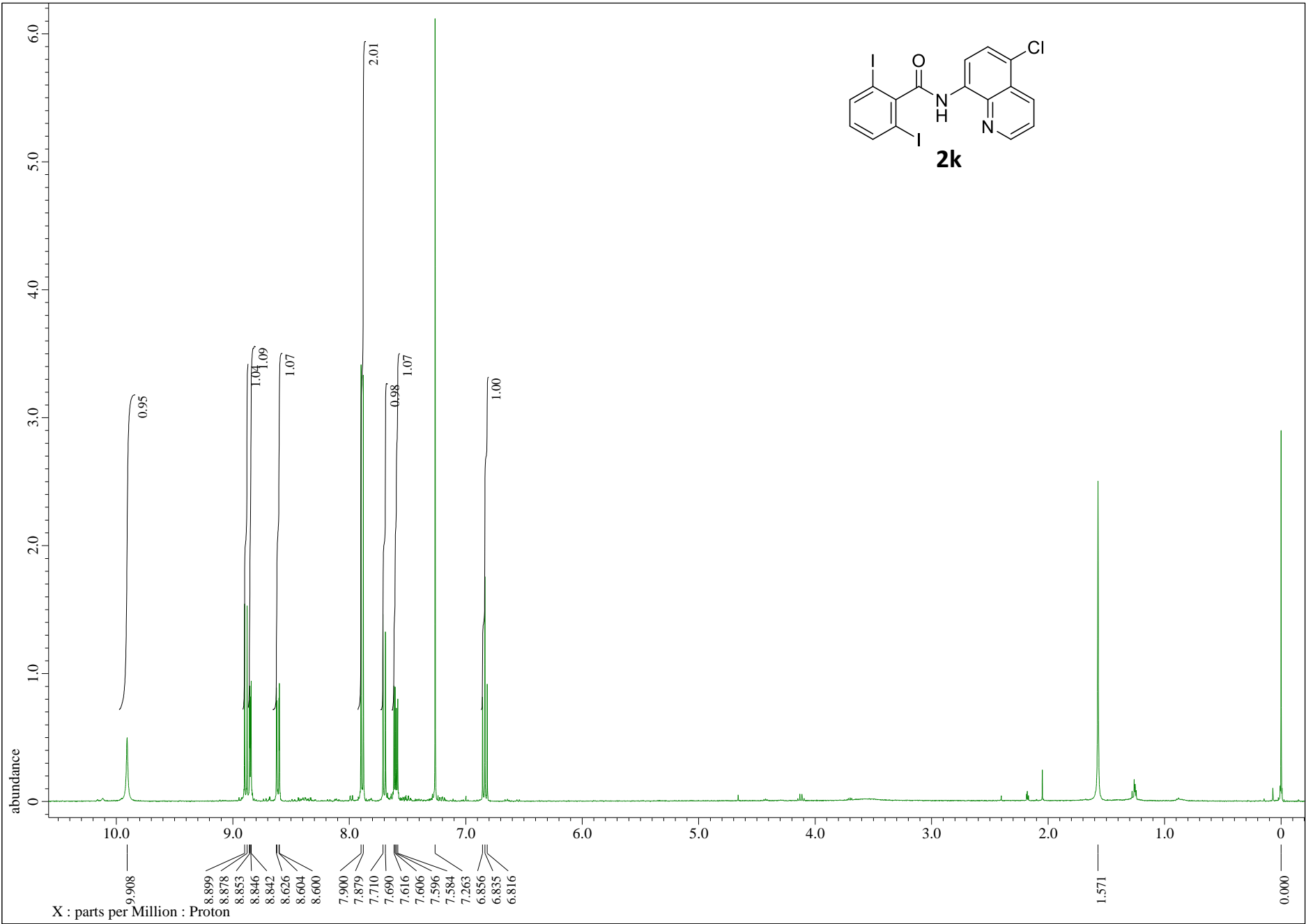
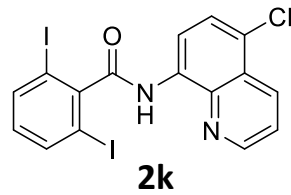


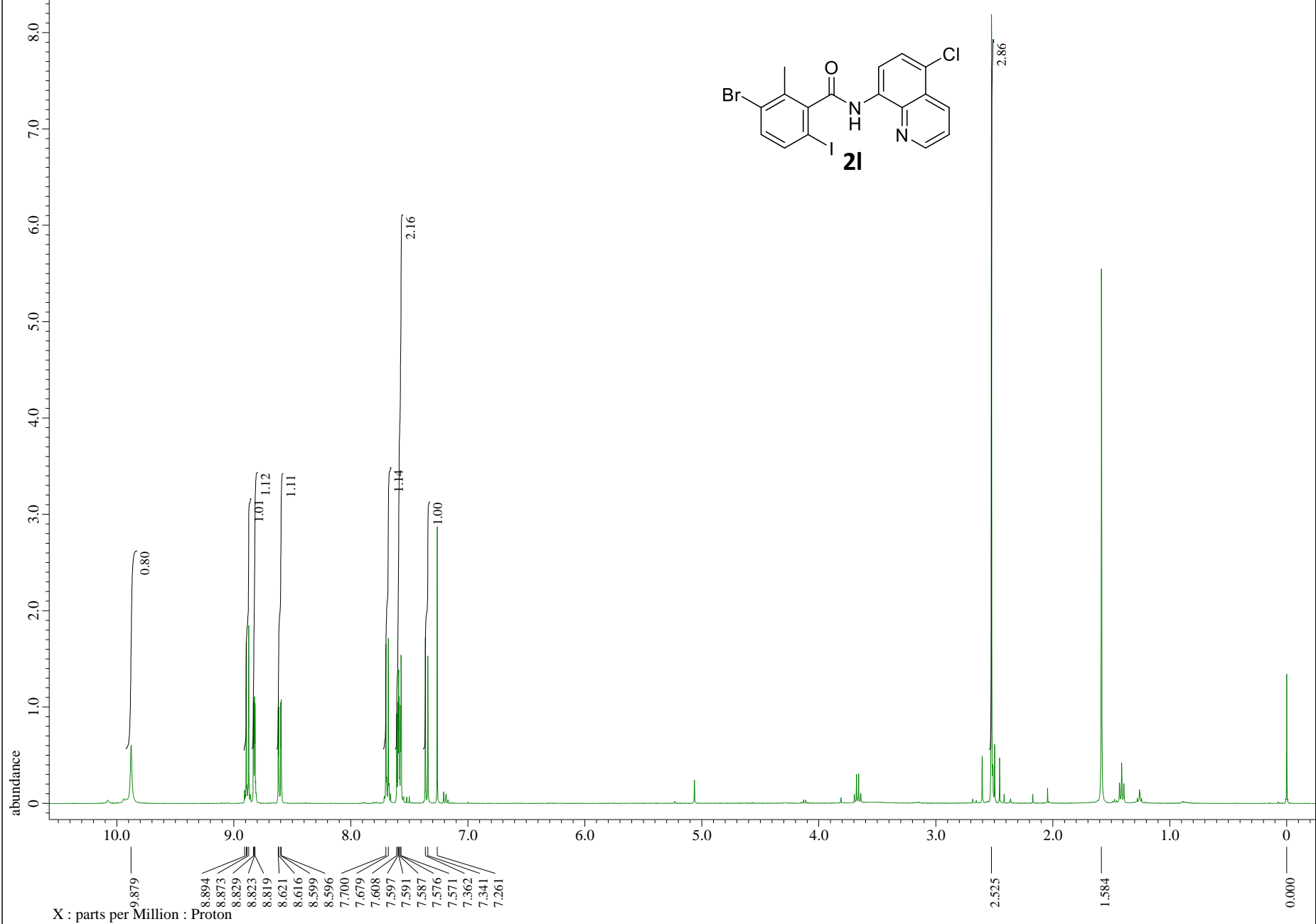
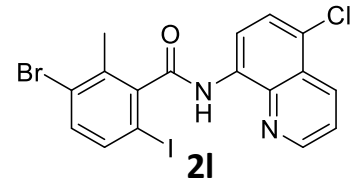


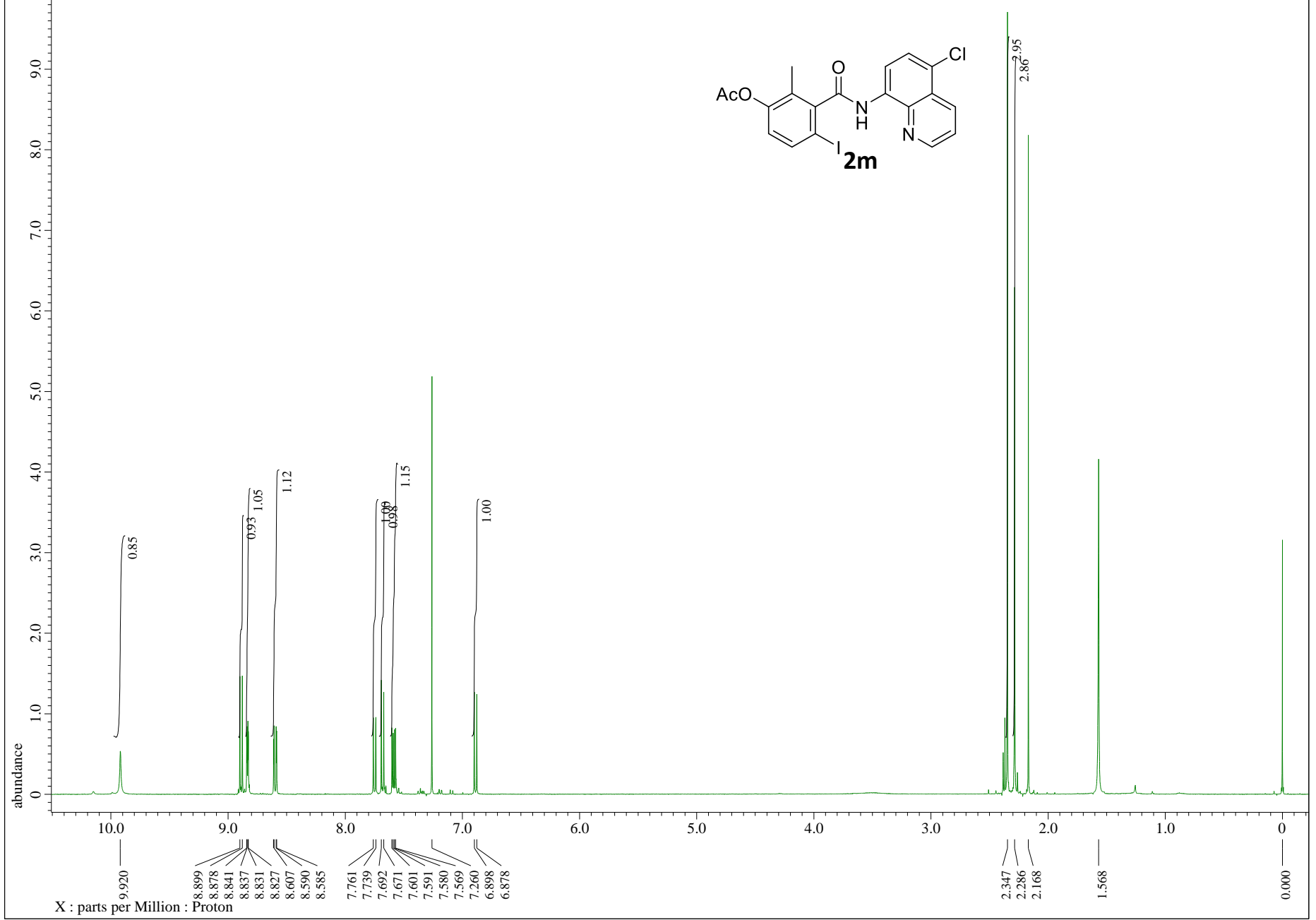
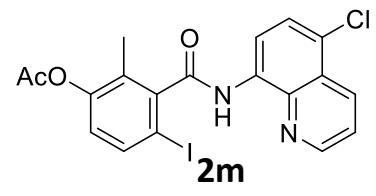


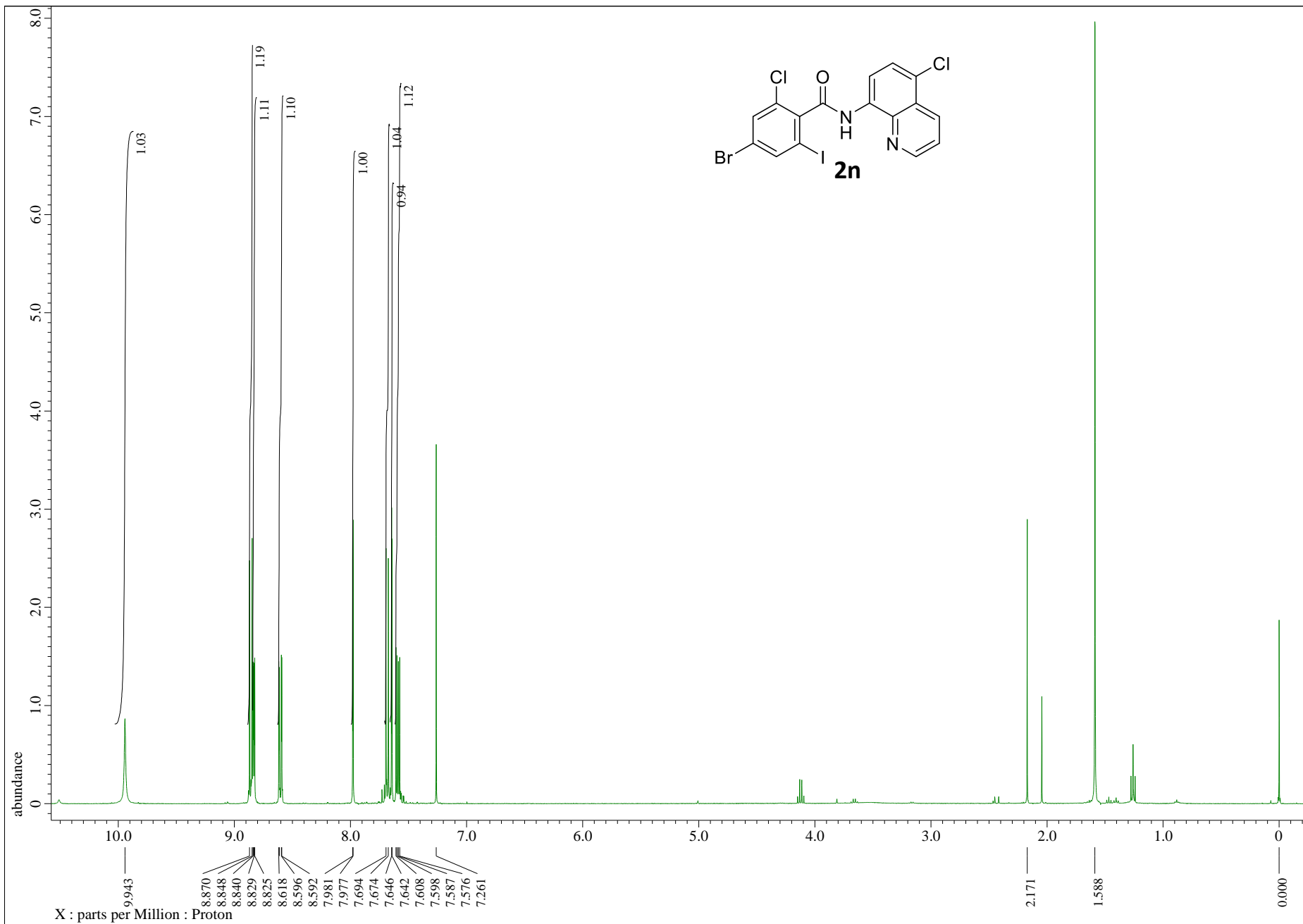


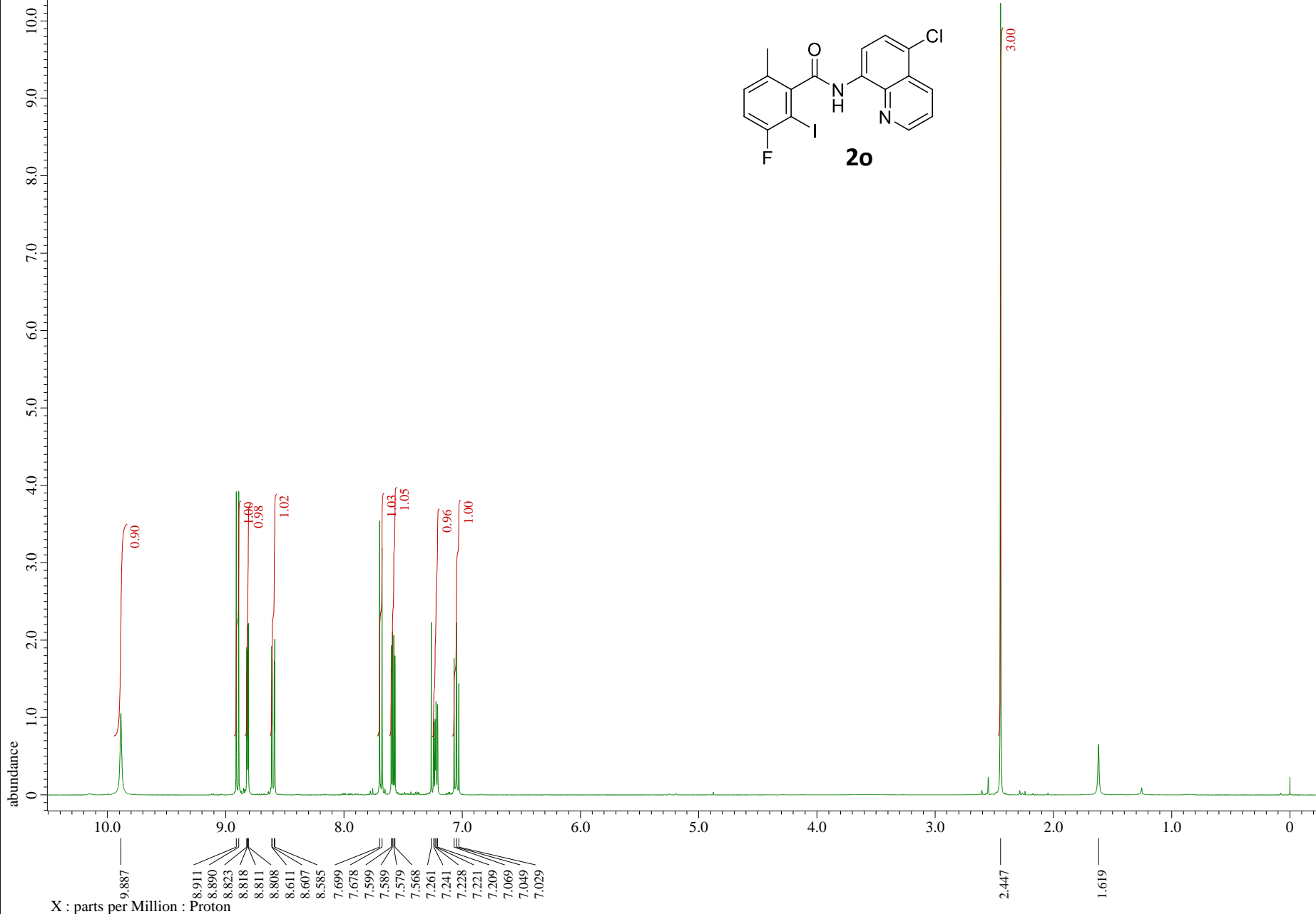
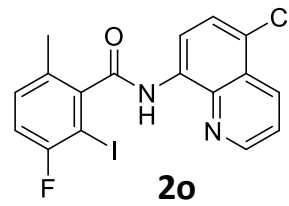


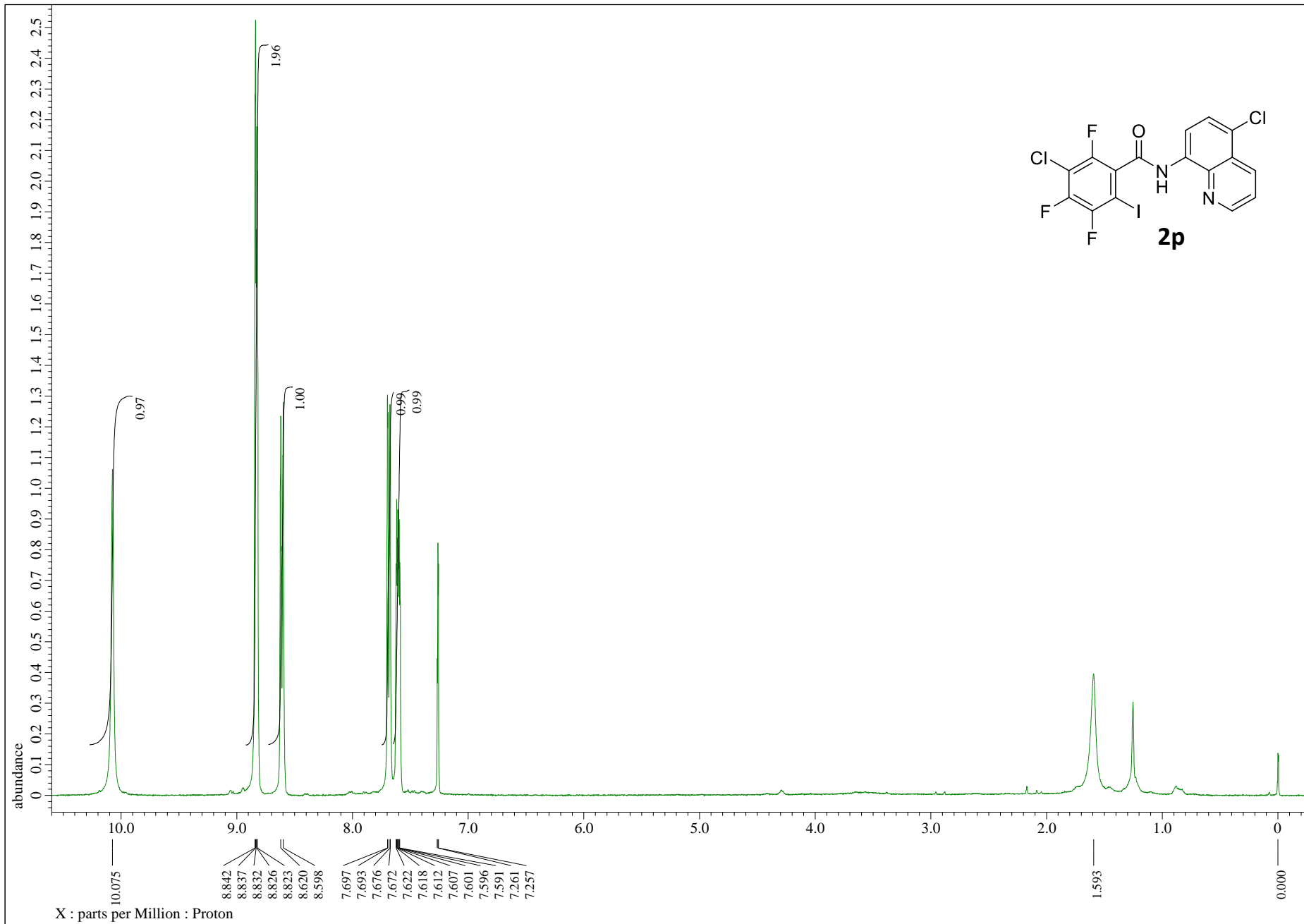


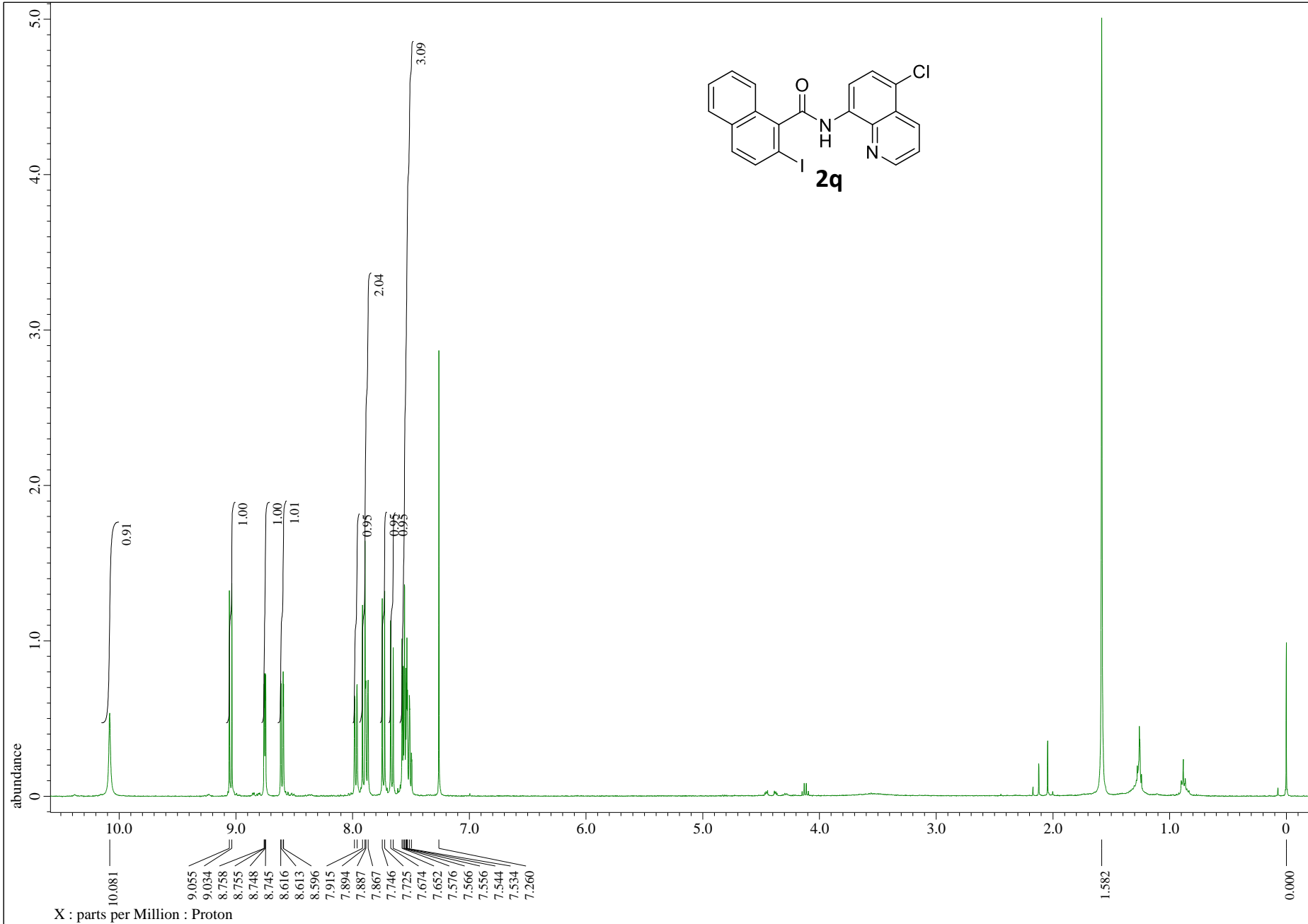


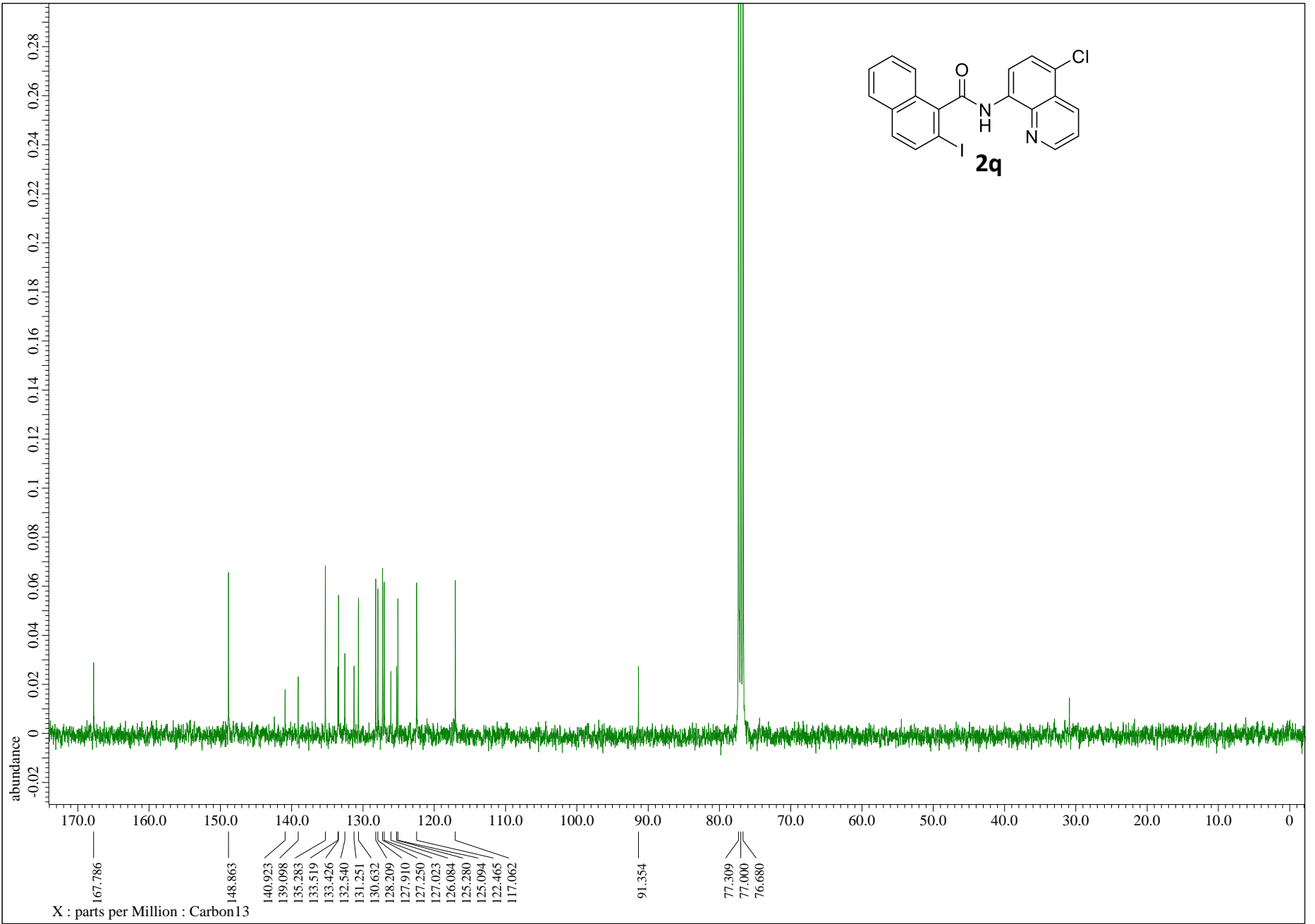
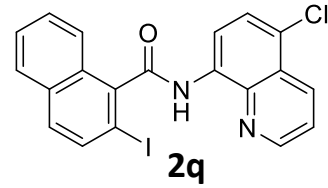


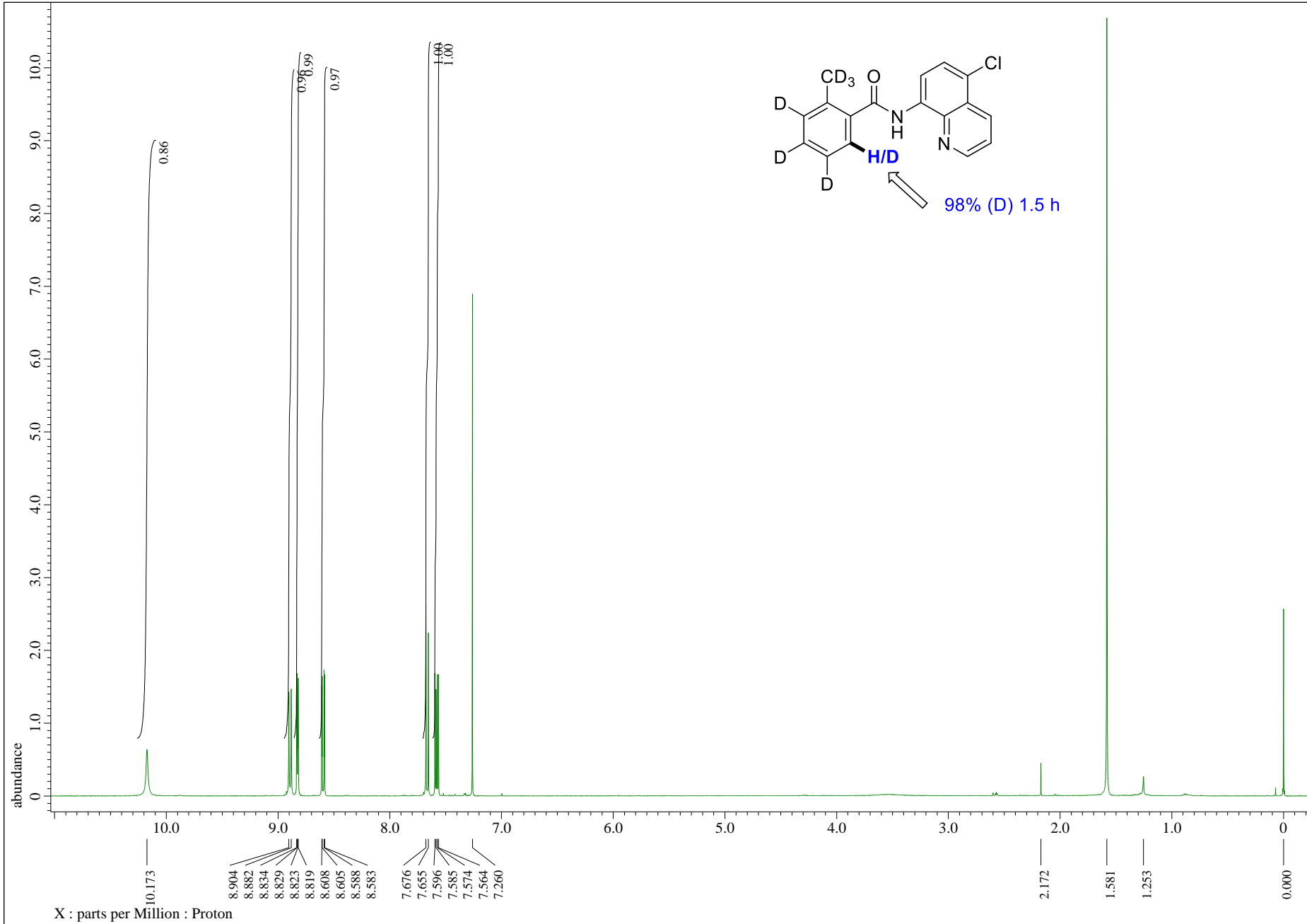


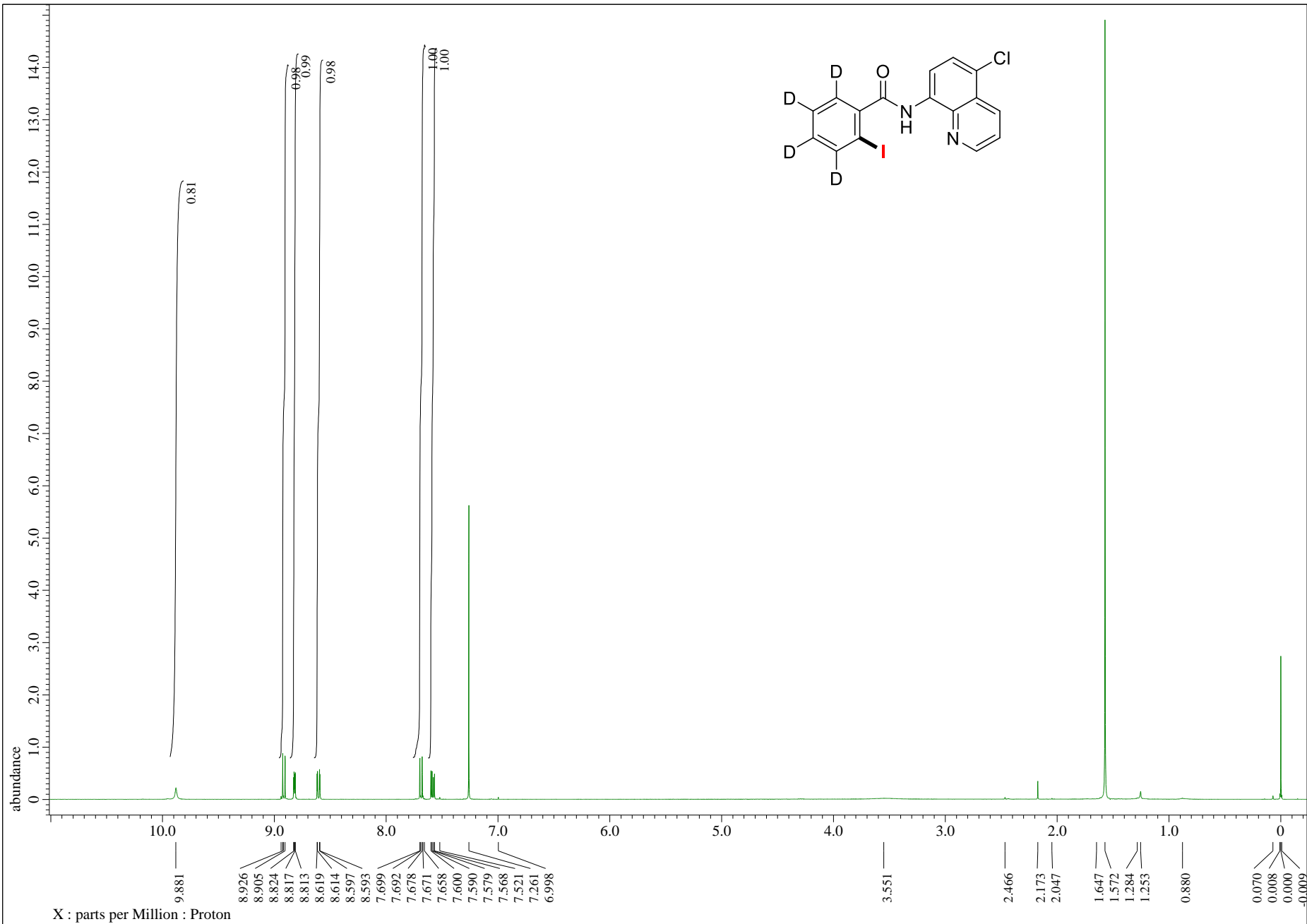


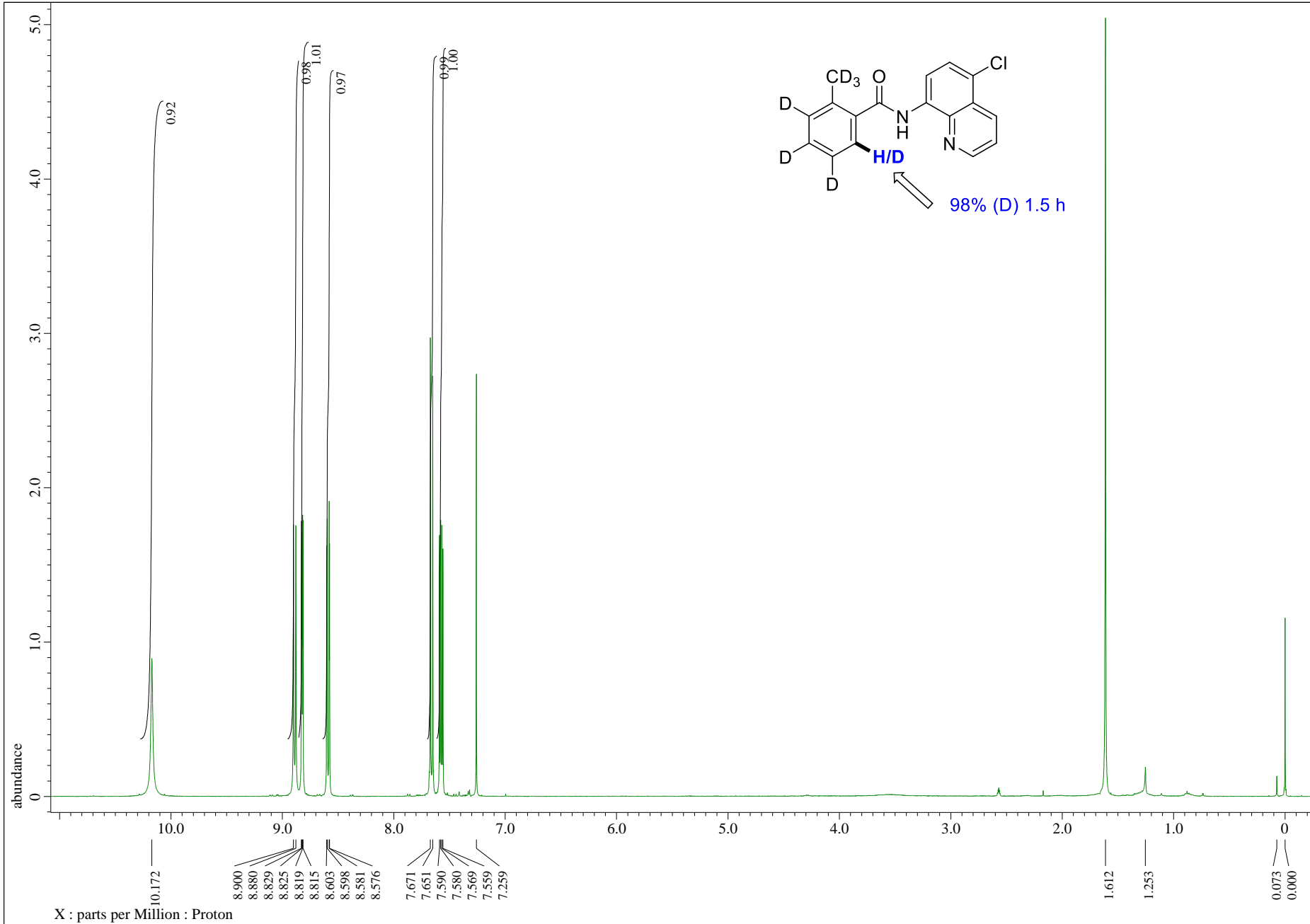












Calculated data of Cartesian coordinates and energies

Figure S1

TS1

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.001517	2.145892	-0.188157
2	6	0	1.660445	1.924450	-0.557956
3	6	0	0.816037	3.052285	-0.727682
4	6	0	1.296723	4.349834	-0.583937
5	6	0	2.634424	4.540869	-0.221835
6	6	0	3.482139	3.445314	-0.020113
7	6	0	-0.627322	2.775985	-0.998842
8	8	0	-1.401597	3.561338	-1.543731
9	7	0	-0.969866	1.549789	-0.458915
10	6	0	-2.227871	0.936674	-0.529547
11	6	0	-2.288183	-0.293650	0.190103
12	6	0	-3.382084	1.402770	-1.141506
13	6	0	-3.493752	-1.035700	0.322753
14	6	0	-4.583672	0.665713	-1.038371
15	6	0	-1.088215	-1.838685	1.467978
16	6	0	-3.436729	-2.225125	1.092272
17	6	0	-4.647028	-0.511078	-0.325481
18	6	0	-2.247501	-2.616088	1.667839
19	7	0	-1.119753	-0.724033	0.757192
20	17	0	-6.175276	-1.372301	-0.213059
21	1	0	3.676130	1.301770	-0.089292
22	1	0	1.601782	1.170818	-1.579892
23	1	0	0.629276	5.193453	-0.731864
24	1	0	3.018370	5.549339	-0.096537
25	1	0	4.520172	3.609517	0.254262
26	1	0	-3.364371	2.336077	-1.686231
27	1	0	-5.475272	1.045203	-1.525628
28	1	0	-0.124450	-2.132132	1.866591
29	1	0	-4.333520	-2.820626	1.222299
30	1	0	-2.182700	-3.520596	2.261720
31	27	0	0.416454	0.380180	0.130424
32	6	0	0.795088	-0.435117	-2.568151
33	8	0	0.234073	-0.669394	-1.440277
34	8	0	1.521711	0.565660	-2.796494
35	6	0	0.559553	-1.461196	-3.649129
36	1	0	0.915002	-1.091883	-4.611563
37	1	0	1.106281	-2.370834	-3.380045
38	1	0	-0.502884	-1.711889	-3.704411
39	8	0	1.751818	-0.899211	0.884543
40	16	0	2.958827	-1.635693	0.321393

41	8	0	3.993665	-1.816942	1.348048
42	8	0	3.381364	-1.166782	-1.006153
43	6	0	2.286257	-3.352893	0.044173
44	9	0	1.751129	-3.826837	1.180169
45	9	0	1.339711	-3.355168	-0.903021
46	9	0	3.276651	-4.166272	-0.336992
47	6	0	-0.052893	1.921088	2.653761
48	8	0	0.651306	1.180816	1.951378
49	8	0	-1.169617	2.474919	2.226262
50	1	0	-1.283755	2.227578	1.272861
51	6	0	0.314867	2.258691	4.064480
52	1	0	0.492518	3.336412	4.142379
53	1	0	-0.519692	2.012562	4.728192
54	1	0	1.208677	1.711371	4.361542

SCF Done: E(RB3LYP) = -2826.20867693 A.U. after 23 cycles

Zero-point correction= 0.365512 (Hartree/Particle)
Thermal correction to Energy= 0.400525
Thermal correction to Enthalpy= 0.401470
Thermal correction to Gibbs Free Energy= 0.297462
Sum of electronic and zero-point Energies= -2824.462024
Sum of electronic and thermal Energies= -2824.427010
Sum of electronic and thermal Enthalpies= -2824.426066
Sum of electronic and thermal Free Energies= -2824.530074

TS2

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-3.460533	-1.350256	-0.396851
2	6	0	-2.225477	-0.760260	-0.733304
3	6	0	-1.860477	-0.704301	-2.102755
4	6	0	-2.732920	-1.117352	-3.104141
5	6	0	-3.975791	-1.651995	-2.744665
6	6	0	-4.335231	-1.778278	-1.398661
7	6	0	-0.459938	-0.274366	-2.404892
8	8	0	-0.090628	0.153991	-3.496796
9	7	0	0.343171	-0.521514	-1.302327
10	6	0	1.694824	-0.155054	-1.206535
11	6	0	2.245114	-0.351138	0.095868
12	6	0	2.529395	0.294598	-2.219693
13	6	0	3.632393	-0.158300	0.356823
14	6	0	3.897624	0.532855	-1.963542
15	6	0	1.841301	-1.033311	2.289332
16	6	0	4.094389	-0.456242	1.663321
17	6	0	4.439824	0.302680	-0.719977

18	6	0	3.208797	-0.909836	2.612941
19	7	0	1.378504	-0.750225	1.081901
20	17	0	6.159736	0.576972	-0.465611
21	8	0	-1.706464	-0.731865	1.969905
22	6	0	-1.512353	-0.383959	3.213249
23	8	0	-0.462979	0.027012	3.710809
24	6	0	-2.777710	-0.540037	4.055669
25	1	0	-3.741386	-1.413513	0.649686
26	1	0	-2.472003	0.502504	-0.238789
27	1	0	-2.430944	-1.048352	-4.144979
28	1	0	-4.661612	-1.984331	-3.518834
29	1	0	-5.296611	-2.208997	-1.133633
30	1	0	2.131230	0.452773	-3.211508
31	1	0	4.530931	0.887275	-2.769624
32	1	0	1.108952	-1.313655	3.032864
33	1	0	5.144689	-0.329152	1.901227
34	1	0	3.537996	-1.153831	3.616738
35	1	0	-3.576860	0.088406	3.648428
36	1	0	-3.128176	-1.577230	4.021777
37	1	0	-2.579940	-0.253676	5.090564
38	8	0	-0.355881	1.295786	0.587637
39	16	0	-1.490286	2.274194	0.726624
40	8	0	-1.675588	2.929378	2.011998
41	8	0	-2.755283	1.650881	0.130835
42	6	0	-1.032908	3.598558	-0.510702
43	9	0	0.116630	4.156563	-0.140854
44	9	0	-0.899394	3.046103	-1.715416
45	9	0	-1.994252	4.517744	-0.541227
46	8	0	-0.642664	-2.629385	0.544623
47	6	0	-0.156200	-3.510593	-0.190174
48	27	0	-0.558163	-0.687205	0.424522
49	8	0	0.486864	-3.251911	-1.300255
50	1	0	0.494866	-2.252979	-1.430680
51	6	0	-0.283778	-4.961060	0.156188
52	1	0	-0.816306	-5.478269	-0.648439
53	1	0	0.714277	-5.404309	0.230650
54	1	0	-0.819287	-5.080477	1.097282

SCF Done: E(RB3LYP) = -2826.17759399 A.U. after 28 cycles

Zero-point correction=	0.363705 (Hartree/Particle)
Thermal correction to Energy=	0.398836
Thermal correction to Enthalpy=	0.399780
Thermal correction to Gibbs Free Energy=	0.295014
Sum of electronic and zero-point Energies=	-2824.430573
Sum of electronic and thermal Energies=	-2824.395443
Sum of electronic and thermal Enthalpies=	-2824.394498
Sum of electronic and thermal Free Energies=	-2824.499264

Figure S2

pre-TS

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.145821	2.321332	-0.347423
2	6	0	1.851665	1.992978	-0.768453
3	6	0	0.859181	2.986979	-0.866843
4	6	0	1.182321	4.313965	-0.575280
5	6	0	2.476114	4.638593	-0.160725
6	6	0	3.454227	3.645475	-0.036787
7	6	0	-0.566482	2.622542	-1.151989
8	8	0	-1.335971	3.340897	-1.781990
9	7	0	-0.937083	1.475157	-0.469028
10	6	0	-2.208378	0.884437	-0.506210
11	6	0	-2.273175	-0.303161	0.272914
12	6	0	-3.369657	1.330492	-1.119184
13	6	0	-3.478699	-1.023516	0.482650
14	6	0	-4.576944	0.614990	-0.945708
15	6	0	-1.041051	-1.801925	1.595832
16	6	0	-3.408915	-2.167484	1.316293
17	6	0	-4.641392	-0.518177	-0.164434
18	6	0	-2.205941	-2.542966	1.876157
19	7	0	-1.095668	-0.726216	0.826381
20	17	0	-6.175793	-1.346365	0.038307
21	1	0	3.905248	1.547903	-0.307074
22	1	0	1.705585	1.025053	-1.263682
23	1	0	0.415625	5.078102	-0.655075
24	1	0	2.721595	5.670916	0.071203
25	1	0	4.458085	3.909292	0.281999
26	1	0	-3.353693	2.229499	-1.718225
27	1	0	-5.474712	0.978949	-1.433361
28	1	0	-0.064693	-2.088969	1.966028
29	1	0	-4.306870	-2.744417	1.507331
30	1	0	-2.132247	-3.413424	2.517488
31	27	0	0.390693	0.257050	0.156142
32	6	0	0.250976	-0.494434	-2.610419
33	8	0	0.226618	-0.842003	-1.340522
34	8	0	0.411698	0.640990	-3.046755
35	6	0	0.067831	-1.701819	-3.519006
36	1	0	0.133878	-1.390127	-4.562723
37	1	0	0.836572	-2.448261	-3.300691
38	1	0	-0.906293	-2.165814	-3.332826
39	8	0	1.723364	-0.997375	0.907702
40	16	0	2.996898	-1.559271	0.275286
41	8	0	4.074735	-1.650911	1.267839
42	8	0	3.314754	-0.974829	-1.035812
43	6	0	2.523461	-3.335060	-0.047603

44	9	0	2.050826	-3.885704	1.079885
45	9	0	1.586171	-3.430746	-0.994420
46	9	0	3.605918	-4.014265	-0.441808
47	6	0	0.085095	2.002577	2.576837
48	8	0	0.715980	1.153728	1.925722
49	8	0	-0.991669	2.605093	2.119728
50	1	0	-1.141373	2.291664	1.187686
51	6	0	0.508942	2.422191	3.948769
52	1	0	0.789464	3.480601	3.929225
53	1	0	-0.332731	2.316303	4.640020
54	1	0	1.352911	1.819995	4.283055

SCF Done: E(RB3LYP) = -2826.21712751 A.U. after 26 cycles

Zero-point correction= 0.369715 (Hartree/Particle)
Thermal correction to Energy= 0.405546
Thermal correction to Enthalpy= 0.406491
Thermal correction to Gibbs Free Energy= 0.299209
Sum of electronic and zero-point Energies= -2824.466401
Sum of electronic and thermal Energies= -2824.430570
Sum of electronic and thermal Enthalpies= -2824.429625
Sum of electronic and thermal Free Energies= -2824.536907

Int1

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.983333	2.015579	0.257632
2	6	0	1.619747	1.894119	-0.020724
3	6	0	0.944784	2.979113	-0.609809
4	6	0	1.614214	4.160324	-0.944507
5	6	0	2.977055	4.270470	-0.669272
6	6	0	3.654655	3.204935	-0.067905
7	6	0	-0.511998	2.794902	-0.847754
8	8	0	-1.219326	3.555589	-1.505009
9	7	0	-0.955393	1.654036	-0.175757
10	6	0	-2.218183	1.063765	-0.364078
11	6	0	-2.334039	-0.260976	0.161531
12	6	0	-3.340006	1.644053	-0.939472
13	6	0	-3.567930	-0.974324	0.136526
14	6	0	-4.562724	0.938536	-0.992429
15	6	0	-1.239787	-2.020906	1.225271
16	6	0	-3.577658	-2.269403	0.714200
17	6	0	-4.681902	-0.327315	-0.465618
18	6	0	-2.425001	-2.785041	1.262865
19	7	0	-1.200540	-0.812990	0.690472
20	17	0	-6.236070	-1.150985	-0.540147

21	1	0	3.532982	1.204866	0.724401
22	1	0	2.201104	1.111688	-1.887095
23	1	0	1.062056	4.972720	-1.408696
24	1	0	3.511888	5.182088	-0.919411
25	1	0	4.715530	3.295077	0.150828
26	1	0	-3.278565	2.644091	-1.343422
27	1	0	-5.424273	1.414020	-1.448874
28	1	0	-0.306612	-2.404228	1.621830
29	1	0	-4.497115	-2.844223	0.719333
30	1	0	-2.410282	-3.771531	1.712560
31	27	0	0.424316	0.424770	0.348160
32	6	0	0.994589	-0.182874	-2.537496
33	8	0	0.374374	-0.312548	-1.467966
34	8	0	1.971664	0.668320	-2.737511
35	6	0	0.659157	-1.045562	-3.714058
36	1	0	1.055195	-0.622336	-4.637630
37	1	0	1.115297	-2.027612	-3.543873
38	1	0	-0.422131	-1.180665	-3.779396
39	8	0	1.749523	-0.960903	0.942215
40	16	0	2.899575	-1.719119	0.303315
41	8	0	3.952721	-2.021584	1.281102
42	8	0	3.314823	-1.195980	-1.008200
43	6	0	2.123722	-3.369801	-0.076676
44	9	0	1.622886	-3.921658	1.039095
45	9	0	1.127439	-3.230530	-0.964795
46	9	0	3.042097	-4.194563	-0.588984
47	6	0	-0.147216	1.711600	2.923036
48	8	0	0.487858	0.881625	2.246741
49	8	0	-1.051530	2.513580	2.417587
50	1	0	-1.115498	2.321044	1.431704
51	6	0	0.089225	1.857429	4.393891
52	1	0	0.478582	2.860927	4.595767
53	1	0	-0.860174	1.757209	4.928471
54	1	0	0.800145	1.107635	4.739155

SCF Done: E(RB3LYP) = -2826.23563497 A.U. after 28 cycles

Zero-point correction=	0.370536 (Hartree/Particle)
Thermal correction to Energy=	0.406080
Thermal correction to Enthalpy=	0.407025
Thermal correction to Gibbs Free Energy=	0.300963
Sum of electronic and zero-point Energies=	-2824.482139
Sum of electronic and thermal Energies=	-2824.446594
Sum of electronic and thermal Enthalpies=	-2824.445650
Sum of electronic and thermal Free Energies=	-2824.551711

Int2

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.141083	1.638452	-0.480152
2	6	0	1.901296	1.086555	-0.812573
3	6	0	1.712265	0.538667	-2.092322
4	6	0	2.750055	0.514499	-3.027993
5	6	0	3.991987	1.054245	-2.687531
6	6	0	4.184646	1.615412	-1.422378
7	6	0	0.358834	-0.012434	-2.382439
8	8	0	0.080081	-0.683287	-3.374767
9	7	0	-0.530765	0.372565	-1.378637
10	6	0	-1.814853	-0.175935	-1.221379
11	6	0	-2.390079	0.029086	0.072361
12	6	0	-2.571914	-0.826344	-2.185922
13	6	0	-3.725468	-0.378574	0.366857
14	6	0	-3.883572	-1.261493	-1.896371
15	6	0	-2.096186	0.892592	2.214313
16	6	0	-4.221223	-0.090638	1.663203
17	6	0	-4.451140	-1.037532	-0.663366
18	6	0	-3.416218	0.551519	2.575407
19	7	0	-1.601138	0.635964	1.014837
20	17	0	-6.101539	-1.576442	-0.363991
21	8	0	1.337296	1.476124	1.832631
22	6	0	1.274979	1.003147	3.046378
23	8	0	0.587037	0.049638	3.420744
24	6	0	2.181307	1.755957	4.017536
25	1	0	3.291510	2.095182	0.494041
26	1	0	3.017209	-0.219272	0.699633
27	1	0	2.574771	0.075576	-4.006451
28	1	0	4.805893	1.042953	-3.406720
29	1	0	5.149320	2.044476	-1.163112
30	1	0	-2.153568	-0.997267	-3.167274
31	1	0	-4.453420	-1.768861	-2.667457
32	1	0	-1.426210	1.355199	2.926751
33	1	0	-5.233556	-0.378716	1.924178
34	1	0	-3.772846	0.785892	3.572222
35	1	0	3.225617	1.665087	3.698933
36	1	0	1.932993	2.822562	4.018423
37	1	0	2.074174	1.352075	5.026197
38	8	0	0.653192	-1.093638	0.563696
39	16	0	1.748733	-1.914272	1.126187
40	8	0	1.569633	-2.704483	2.323257
41	8	0	3.079390	-1.043823	1.255624
42	6	0	2.201257	-3.097609	-0.259661
43	9	0	1.115186	-3.790818	-0.578183
44	9	0	2.624472	-2.399751	-1.308504
45	9	0	3.159744	-3.910425	0.165459
46	8	0	-0.004593	2.834840	0.130206
47	6	0	-0.540232	3.497404	-0.781199
48	27	0	0.325827	0.935552	0.281766

49	8	0	-1.002624	2.971141	-1.883608
50	1	0	-0.857671	1.967460	-1.832591
51	6	0	-0.687318	4.982850	-0.661823
52	1	0	-0.158312	5.464076	-1.490768
53	1	0	-1.745400	5.250498	-0.744630
54	1	0	-0.285030	5.329222	0.289496

SCF Done: E(RB3LYP) = -2826.19175886 A.U. after 27 cycles

Zero-point correction= 0.368382 (Hartree/Particle)
Thermal correction to Energy= 0.404242
Thermal correction to Enthalpy= 0.405186
Thermal correction to Gibbs Free Energy= 0.298530
Sum of electronic and zero-point Energies= -2824.439950
Sum of electronic and thermal Energies= -2824.404090
Sum of electronic and thermal Enthalpies= -2824.403146
Sum of electronic and thermal Free Energies= -2824.509802

Figure S3

SET-int1

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-3.579019	-1.680906	-2.047486
2	6	0	-2.468368	-1.361770	-1.269384
3	6	0	-2.420883	-1.753907	0.078934
4	6	0	-3.492729	-2.473986	0.635036
5	6	0	-4.603136	-2.781358	-0.143516
6	6	0	-4.647026	-2.385305	-1.484915
7	6	0	-1.251405	-1.483027	0.938398
8	8	0	-1.068917	-1.982254	2.035686
9	7	0	-0.286953	-0.533916	0.437171
10	6	0	1.014709	-0.851287	0.350921
11	6	0	1.874025	0.249612	-0.021516
12	6	0	1.608692	-2.123517	0.542174
13	6	0	3.264257	0.094048	-0.205644
14	6	0	2.979851	-2.288330	0.366996
15	6	0	1.958051	2.518983	-0.530217
16	6	0	4.001217	1.243707	-0.575200
17	6	0	3.800371	-1.220117	0.002085
18	6	0	3.345381	2.449246	-0.738142
19	7	0	1.251693	1.444671	-0.181866
20	17	0	5.496568	-1.496763	-0.201888
21	8	0	-1.217647	3.202615	0.022709
22	6	0	-2.412740	2.843401	0.330715
23	8	0	-2.568544	1.592094	0.548076

24	6	0	-3.528941	3.826599	0.455969
25	1	0	-3.611650	-1.382411	-3.090781
26	1	0	-1.637259	-0.821212	-1.712077
27	1	0	-3.439508	-2.777964	1.675009
28	1	0	-5.433821	-3.329518	0.290348
29	1	0	-5.513817	-2.628754	-2.092527
30	1	0	0.999545	-2.970278	0.832354
31	1	0	3.422945	-3.266249	0.515589
32	1	0	1.406592	3.445986	-0.641896
33	1	0	5.072101	1.174507	-0.728003
34	1	0	3.885870	3.344503	-1.021892
35	1	0	-3.510278	4.256023	1.464793
36	1	0	-3.397474	4.639315	-0.262055
37	1	0	-4.490036	3.330923	0.306756
38	27	0	-0.657816	1.356394	0.190605

SCF Done: E(UB3LYP) = -1635.18696290 A.U. after 65 cycles

Zero-point correction= 0.276838 (Hartree/Particle)
Thermal correction to Energy= 0.298619
Thermal correction to Enthalpy= 0.299563
Thermal correction to Gibbs Free Energy= 0.222748
Sum of electronic and zero-point Energies= -1633.870077
Sum of electronic and thermal Energies= -1633.848296
Sum of electronic and thermal Enthalpies= -1633.847351
Sum of electronic and thermal Free Energies= -1633.924166

SET-int2

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-4.086367	0.249501	0.059804
2	6	0	-2.787394	0.620148	0.421475
3	6	0	-2.318632	1.920956	0.143690
4	6	0	-3.168004	2.848006	-0.468424
5	6	0	-4.463488	2.470789	-0.824773
6	6	0	-4.921077	1.172772	-0.571293
7	6	0	-0.904757	2.298966	0.384091
8	8	0	-0.508221	3.437343	0.561179
9	7	0	-0.037047	1.178757	0.272150
10	6	0	1.299623	1.237306	0.187280
11	6	0	1.902407	-0.023160	-0.187946
12	6	0	2.155700	2.352574	0.397117
13	6	0	3.295707	-0.182303	-0.335326
14	6	0	3.529403	2.205592	0.260552
15	6	0	1.497430	-2.252656	-0.755312
16	6	0	3.771951	-1.457476	-0.714194

17	6	0	4.105024	0.979564	-0.093312
18	6	0	2.871295	-2.485999	-0.920473
19	7	0	1.045475	-1.049025	-0.412309
20	17	0	5.816065	0.875251	-0.245625
21	8	0	-1.540483	-2.206129	-0.711842
22	6	0	-1.679239	-1.638201	-1.855086
23	8	0	-1.241381	-0.437350	-1.912384
24	6	0	-2.284680	-2.335402	-3.025619
25	1	0	-4.441490	-0.745146	0.308408
26	1	0	-2.312177	-0.052309	1.158400
27	1	0	-2.799472	3.845418	-0.683594
28	1	0	-5.117087	3.190159	-1.308104
29	1	0	-5.931017	0.889748	-0.850935
30	1	0	1.728958	3.305723	0.672705
31	1	0	4.176205	3.057972	0.433887
32	1	0	0.757438	-3.031657	-0.890999
33	1	0	4.835513	-1.624874	-0.838493
34	1	0	3.207727	-3.475831	-1.204820
35	1	0	-1.551531	-3.037574	-3.439480
36	1	0	-3.157028	-2.910470	-2.704515
37	1	0	-2.562374	-1.614221	-3.795831
38	8	0	-0.305596	-1.055408	1.746091
39	6	0	-1.179807	-1.644270	2.536327
40	8	0	-2.386367	-1.719392	2.309678
41	27	0	-0.754196	-0.569018	-0.017403
42	6	0	-0.539486	-2.218939	3.787858
43	1	0	-1.311057	-2.618139	4.448060
44	1	0	0.158575	-3.016396	3.512402
45	1	0	0.032728	-1.445441	4.309125

SCF Done: E(UB3LYP) = -1863.74504969 A.U. after 42 cycles

Zero-point correction= 0.328860 (Hartree/Particle)
Thermal correction to Energy= 0.355534
Thermal correction to Enthalpy= 0.356478
Thermal correction to Gibbs Free Energy= 0.269376
Sum of electronic and zero-point Energies= -1862.266166
Sum of electronic and thermal Energies= -1862.239492
Sum of electronic and thermal Enthalpies= -1862.238548
Sum of electronic and thermal Free Energies= -1862.325650

Calculated data of spin densities

SET-int1

Mulliken charges and spin densities:

	1	2
1 C	-0.143441	0.001903
2 C	-0.169837	0.000251
3 C	0.090015	0.005666

4	C	-0.156239	-0.002494
5	C	-0.141285	0.002143
6	C	-0.119675	-0.002862
7	C	0.546792	-0.029498
8	O	-0.464606	0.047317
9	N	-0.708402	0.403125
10	C	0.429157	-0.074084
11	C	0.318155	0.110322
12	C	-0.126098	0.273143
13	C	0.190776	-0.081297
14	C	-0.127768	-0.136505
15	C	0.111712	0.038083
16	C	-0.123813	0.055754
17	C	-0.134483	0.328056
18	C	-0.149929	-0.034994
19	N	-0.664538	-0.033411
20	Cl	0.086650	0.049567
21	O	-0.545827	-0.023296
22	C	0.718584	0.004903
23	O	-0.530360	-0.023323
24	C	-0.553281	-0.000543
25	H	0.165368	-0.000077
26	H	0.181081	-0.001222
27	H	0.173757	0.000044
28	H	0.166199	-0.000001
29	H	0.167734	0.000171
30	H	0.215846	-0.011899
31	H	0.213253	0.004355
32	H	0.233817	-0.002327
33	H	0.217288	-0.002341
34	H	0.208883	0.001246
35	H	0.220421	0.000226
36	H	0.206928	-0.000246
37	H	0.206564	-0.000313
38	Co	0.790603	1.134458

Sum of Mulliken charges = 1.00000 2.00000

SET-int2

Mulliken charges and spin densities:

	1	2	
1	C	-0.137223	-0.003557
2	C	-0.271812	0.006931
3	C	0.092878	0.001638
4	C	-0.140557	0.009191
5	C	-0.125413	-0.005337
6	C	-0.112432	0.014399
7	C	0.596068	-0.027673
8	O	-0.469858	0.079114
9	N	-0.707588	0.306693

10	C	0.444455	0.021990
11	C	0.346340	0.087825
12	C	-0.106381	0.223968
13	C	0.195816	-0.071400
14	C	-0.119107	-0.106070
15	C	0.110204	0.026825
16	C	-0.116526	0.054931
17	C	-0.124872	0.333149
18	C	-0.143836	-0.017917
19	N	-0.622677	-0.005501
20	Cl	0.123756	0.066315
21	O	-0.507879	-0.002580
22	C	0.711419	0.000111
23	O	-0.491984	0.007912
24	C	-0.552256	0.000023
25	H	0.195892	0.000116
26	H	0.299136	0.000012
27	H	0.187178	-0.000490
28	H	0.184143	0.000129
29	H	0.182289	-0.000603
30	H	0.238864	-0.010065
31	H	0.223240	0.003050
32	H	0.243422	-0.001442
33	H	0.226782	-0.002322
34	H	0.219527	0.000634
35	H	0.222678	0.000100
36	H	0.218918	0.000041
37	H	0.211678	-0.000020
38	O	-0.554996	-0.000589
39	C	0.663185	0.000224
40	O	-0.526901	-0.000532
41	Co	0.657514	0.010489
42	C	-0.529067	0.000226
43	H	0.182116	0.000073
44	H	0.192648	-0.000010
45	H	0.191217	0.000001
Sum of Mulliken charges = 1.00000 1.00000			