# **Electronic Supporting Information**

#### Facile construction of organometallic rotaxane-terminated

## dendrimers using neutral platinum-acetylides as main scaffold

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## **Table of Contents (38 Pages)**

Section A. Materials/General Methods/Instrumentation S2			
Section B. Synthetic Protocols and Characterization S			
1.	Synthesis of key building block platinum-acetylide complex 2	S4	
2.	Synthesis of the dendrimers Gn-T and Gn-H	<b>S</b> 7	
3.	Synthesis of the rotaxane-terminated dendrimers Gn	S22	
Section C. Additional Characterizations of Rotaxane-terminated Dendrimers S			
1.	DLS experiments of the rotaxane-terminated dendrimers G2-G4	S41	
2.	TEM images for the rotaxne-terminated dendrimers Gn	S41	
3.	UV-vis spectra of the rotaxane-terminated dendrimers Gn	S42	

#### Section A. Materials/General Methods/Instrumentation

All reagents were commercially available and used as supplied without further purification, organometallic [2]rotaxane **3**<sup>S1</sup> were prepared according to the published procedures. Deuterated solvents were purchased from Cambridge Isotope Laboratory (Andover, MA).

NMR spectra were recorded on a Bruker DRX 400 (400 MHz) spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts are reported relative to residual solvent signals, and <sup>31</sup>P{<sup>1</sup>H} NMR chemical shifts are referenced to an external unlocked sample of 85% H<sub>3</sub>PO<sub>4</sub> ( $\delta$  0.0). The twodimensional diffusion-ordered NMR spectroscopy (2D-DOSY) was recorded on a Bruker DRX500 spectrometer. The AFM samples were prepared by dropping the solutions onto a mica sheet. AFM images were obtained on a Dimension FastScan (Bruker) by using ScanAsyst mode under the ambient condition. The DLS samples were prepared in DCM at a concentration of 10<sup>-6</sup> mol/L. DLS measurements were performed under a Malvern Zetasizer Nano-ZS light scattering apparatus (Malvern Instruments, U.K.) with a He-Ne laser (633 nm, 4 mW). TEM images were obtained using a Philips TECNAI-12 instrument with an accelerating voltage of 120 kV. UV–vis spectra were recorded in a quartz cell (light path 10 mm) on a Cary 50Bio UV-Visible spectrophotometer. Steady-state fluorescence spectra were recorded in a conventional quartz cell (light path 10 mm) on a Cary Eclipse fluorescence

The MALDI MS experiments were carried out on a Bruker UltrafleXtreme MALDI TOF/TOF Mass Spectrometer (Bruker Daltonics, Billerica, MA), equipped with smartbeam-II laser. All spectra were measured in positive reflectron or linear mode. The instrument was calibrated externally with BSA, myoglobin, cytochrom C and insulin standards prior to each measurement. *Trans*-2-[3-(4-tert-butylphenyl)-2-methyl-2-propenylidene]malononitrile(DCTB) dissolved in CHCl<sub>3</sub> at 5 mg/mL was used as matrix. The samples were dissolved in CHCl<sub>3</sub> at 5-10 mg/mL. Sample preparation involved depositing 0.5  $\mu$ L of matrix on the wells of a 384-well ground-steel plate, allowing the spots to dry, depositing 0.5  $\mu$ L of each sample on top of a dry matrix spot, and adding another 0.5  $\mu$ L of matrix on top of the dry sample (sandwich method). Compared to the commonly used dry droplet method, in which a mixture of matrix and sample is deposited onto the target plate, the sandwich method minimizes the disintegration of large molecules and allows for facile variation of the matrix to maximize the signal intensity.

#### Section B. Synthetic Protocols and Characterization

1. Synthesis of key building block platinum-acetylide complex 2.

Scheme S1: The synthesis route of platinum-acetylide complex 2.



**Synthesis of 2:** A Schlenk flask was charge with 1-ethynyl-3,5-bis-[(triisopropylsilanyl)-ethynyl]benzene(**A**) (800 mg, 1.73 mmol) and Pt(PEt<sub>3</sub>)<sub>2</sub>I<sub>2</sub> (4.74 g, 6.92 mmol). The Schlenk flask was then evacuated via reduced pressure and backfilled with N<sub>2</sub> for three times. Next, THF (50 mL) and Et<sub>2</sub>NH (25 mL) was added via syringe. The resultant solution was stirred for 5 min. Then CuI (33 mg, 0.173 mmol) was added to the mixture under N<sub>2</sub> atmosphere, and the mixture was allowed to stir overnight. The solvent was then removed by reduced pressure, and the compound was purified by column chromatography on silica gel (eluent: petroleum ether/dichloromethane, v/v, 50:1 to 10:1). A yellow solid was obtained (1.21 g, 68% yield). M.p. 94 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.34 (s, 1H), 7.28 (br, 2H), 2.21 (m, 12H), 1.16 (m, 18H), 1.12 (s, 42H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  9.18; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  134.1, 131.9, 128.9, 123.7, 106.3, 99.0, 92.3, 91.2, 18.8, 16.7, 11.4, 8.40; MS: simulated value for C<sub>42</sub>H<sub>75</sub>IP<sub>2</sub>PtSi<sub>2</sub> (1019.36). (MALDI-TOF-MS) 894.5 ([M - I]<sup>+</sup>).







Figure S2. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of 2.



Figure S3. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 100 MHz) of 2.



Figure S4. MALDI-TOF-MS spectrum of 2.

2. Synthesis of dendrimers Gn-T and Gn-H.

Scheme S2: The synthesis route of Gn-T and Gn-H.







Synthesis of G2-T: Schlenk flask was charge with 1,3,5-triethynylbenzene (44.4 mg 0.296 mmol) and 2 (950 mg, 0.932 mmol). The Schlenk flask was then evacuated via reduced pressure and backfilled with N<sub>2</sub> for three times. Next, Et<sub>2</sub>NH (2 mL) was added via syringe. The solution was stirred for 5 min. Then CuI (9.0 mg) was added into the mixture under N<sub>2</sub> atmosphere, and the mixture was allowed to stir overnight. The solvent was then removed by reduced pressure, and the compound was purified by alumina column chromatography (eluent: ether/dichloromethane (2:1)). A slight yellow solid was obtained (620 mg, 67% yield). M.p. 115 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.30 (br, 3H), 7.28 (br, 6H), 7.00 (s, 3H), 2.15 (m, 36H), 1.20 (m, 54H), 1.12 (s, 126H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$ 11.60; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  134.3, 131.4, 130.6, 129.1, 127.9, 123.4, 110.7, 109.7, 108.0, 106.4, 105.9, 90.7, 18.8, 16.4, 11.4, 8.4; MS: simulated value for C<sub>138</sub>H<sub>228</sub>P<sub>6</sub>Pt<sub>3</sub>Si<sub>6</sub> (2824.38). (MALDI-TOF-MS) 2824.4 ([M]<sup>+</sup>).

Synthesis of G2-H: A solution of the G2-T (150 mg, 0.053 mmol) in THF was cooled in ice bath, and a solution of  $Bu_4NF$  (0.48 mmol) in THF (1.0 M) was added dropwise. The reaction mixture was allowed to warm to room temperature and stirred for 10 min. The solvent was removed under reduced pressure and the residue was purified by alumina column chromatography with dichloromethane: 85 mg (85% yield) of a slight yellow solid was afforded. The related characterization was consistent with the literature.<sup>82</sup>

Synthesis of G3-T: G2-H (60 mg 0.0318 mmol) and 2 (204 mg, 0.200 mmol) were reacted in a similar manner to that of the preparation of G2-T. Column chromatography on alumina was performed with dichloromethane as eluent: 78 mg (34% yield) of a slight yellow solid was afforded. M.p. 167 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.30 (br, 6H), 7.28 (br, 12H), 7.00 (s, 12H), 2.15 (m, 108H), 1.20 (m, 162H), 1.12 (s, 252H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  11.40, 11.14; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  134.2, 131.4, 130.6, 129.1, 128.0, 127.9, 123.4, 109.7, 109.6, 108.8, 106.4, 106.2, 105.8, 91.0, 90.7, 18.7, 16.4, 11.3, 8.3; MS: simulated value for C<sub>336</sub>H<sub>552</sub>P<sub>18</sub>Pt<sub>9</sub>Si<sub>12</sub> (7236.25). (MALDI-TOF-MS) 7359.5 ([M + Na]<sup>+</sup>).

Synthesis of G3-H: A solution of the G3-T (130 mg, 0.018 mmol) in THF was cooled in ice bath, and a solution of  $Bu_4NF$  (0.32 mmol) in THF (1.0 M) was added dropwise. The reaction mixture was allowed to warm to room temperature and stirred for 10 min. The solvent was removed under reduced pressure and the residue was purified by alumina column chromatography with dichloromethane: 87 mg (91% yield) of a slight yellow solid was afforded. M.p. >250 °C (decomp). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.36 (br, 6H), 7.33 (br, 12H), 6.99 (br, 12H), 3.05 (s, 12H), 2.15 (m, 108H), 1.20 (m, 162H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  11.68, 11.39; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  134.8, 132.0, 130.7, 129.5, 128.0, 122.2, 110.0, 107.6, 82.9, 29.8, 18.0, 16.5, 8.5; MS: simulated value for C<sub>228</sub>H<sub>312</sub>P<sub>18</sub>Pt<sub>9</sub> (5362.65). (MALDI-TOF-MS) 5386.3 ([M + Na]<sup>+</sup>).

**Synthesis of G4-T: G3-H** (50 mg 0.0093 mmol) and 2 (119 mg, 0.117 mmol) were reacted in a similar manner to that of the preparation of **G2-T**. Alumina column chromatography was performed with dichloromethane as eluent: 79 mg (53% yield) of a slight yellow solid was afforded. M.p. 190 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.29 (br,12H), 7.28 (br, 24H), 6.99 (s, 30H), 2.15 (m, 252H), 1.20 (m, 378H), 1.12 (s, 504H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  11.64, 11.38; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  134.2, 131.5, 130.5, 129.1, 128.0, 123.4, 106.5, 90.6, 37.1, 29.7, 27.1, 18.7, 16.5, 16.3, 12.8, 11.3, 8.3. MS: simulated value for C<sub>732</sub>H<sub>1200</sub>P<sub>42</sub>Pt<sub>21</sub>Si<sub>24</sub> (16059.99). (GPC) *M*<sub>n</sub> 11463.

**Synthesis of G4-H: G4-T** (128 mg, 0.0080 mmol) was treated with a solution of Bu<sub>4</sub>NF (0.29 mL) in THF (1.0 M) in a similar manner to that of preparation of the **G2-H**. The crude product was purified by alumina column chromatography with dichloromethane as eluent: 81 mg (82% yield) of a slight yellow solid was afforded. M.p. >250 °C (decomp). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.36 (br, 12H), 7.33 (br, 24H), 6.99 (br, 30H), 3.05 (s, 24H), 2.15 (m, 252H) 1.20 (m, 378H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  11.65, 11.36; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  130.9, 130.6, 129.4, 128.8, 127.8, 112.1, 122.1, 82.8, 68.2, 38.7, 30.4, 28.9, 23.0, 17.9, 16.4, 14.0, 11.0, 8.4. MS: simulated value for C<sub>516</sub>H<sub>720</sub>P<sub>42</sub>Pt<sub>21</sub> (12312.79). (GPC) *M*<sub>n</sub> 9328.



Figure S4. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G2-T.



Figure S5. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G2-T.



Figure S6. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G2-T.



Figure S7. MALDI-TOF-MS spectrum of G2-T.



Figure S8. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G3-T.



Figure S9. <sup>13</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G3-T.



Figure S10. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G3-T.



Figure S11. MALDI-TOF-MS spectrum of G3-T.



Figure S12. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G3-H.



Figure S13. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G3-H.



Figure S14. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G3-H.



Figure S15. MALDI-TOF-MS spectrum of G3-H.



Figure S16. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G4-T.



Figure S17. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G4-T.



Figure S18. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G4-T.



Figure S19. GPC data of G4-T.



Figure S21.<sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G4-H.



Figure S22. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G4-H.



Figure S23. GPC data of G4-H.

3. Synthesis of rotaxane-terminated dendrimers Gn.

Scheme S3: The synthesis route of Gn.









Synthesis of G1: A Schlenk flask was charge with 1,3,5-triethynylbenzene (3.60 mg 0.024 mmol) and 3 (153 mg, 0.0756 mmol). The Schlenk flask was then evacuated via reduced pressure and backfilled with N2 for three times. Next, Et2NH (2 mL) was added via syringe. The solution was stirred for 5 min. Then CuI (9 mg) was added to the mixture under N<sub>2</sub> atmosphere, and the mixture was allowed to stir overnight. The solvent was then removed by reduced pressure, and the compound was purified by alumina column chromatography (eluent: dichloromethane): 80 mg (58% yield) of a slight yellow solid was afforded. M.p. 102 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.18 (d, J = 8.4 Hz, 6H), 7.02 (m, 6H), 6.93 (s, 15H), 6.87 (s, 15H), 6.65 (d, J = 8.4 Hz, 6H), 6.62 (d,  $J = 10^{-1}$ 8.4 Hz, 6H), 3.64-4.00 (m, 114H), 2.47 (m, 6H), 2.23 (m, 36H), 1.67-1.91 (m, 66H), 1.27 (m, 60H), 1.12 (m, 51H), 0.97 (m, 45H), 0.63 (m, 6H), -0.25 (m, 6H), -0.72 (m, 6H), -1.54 (m, 6H), -2.23 (m, 6H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz): δ 11.26; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 157.8 (Ar Cipso-OCH<sub>3</sub>), 153.8 (Ar Cipso-OCH<sub>2</sub>), 149.8 (Pillararene, Ar Cipso-OPr), 149.7 (Pillararene, Ar Cipso-OPr), 137.6, (Ar C<sub>ipso</sub>-OCH<sub>3</sub>), 131.7 (Ar C<sub>meta</sub>), 130.6 (Ar C<sub>meta</sub>), 128.3 (Pillararene, Ar C<sub>meta</sub>), 128.1 (Pillararene, Ar Cmeta), 124.8 (Ar Cmeta), 123.6 (Ar Cmeta), 120.2 (Ar Cpara), 114.5 (Pillararene, Ar Cortho), 114.0 (Pillararene, Ar Cortho), 113.7 (Ar Cortho), 109.7 (C-Pt), 109.5 (C-Pt), 105.4 (Ar Cortho), 73.8 (ArO-CH2), 70.0 (Pillararene, O-CH2), 69.6 (Pillararene, O-CH2), 68.5 (ArO-CH<sub>3</sub>), 56.2 (Pillararene, Ar-CH<sub>2</sub>-Ar), 31.6 (O-C<sub>n</sub>H<sub>2n</sub>), 31.3 (O-C<sub>n</sub>H<sub>2n</sub>), 30.9 (O-C<sub>n</sub>H<sub>2n</sub>), 30.5 (O-C<sub>n</sub>H<sub>2n</sub>), 29.2 (O-C<sub>n</sub>H<sub>2n</sub>), 28.6 (O-C<sub>n</sub>H<sub>2n</sub>), 28.1 (O-C<sub>n</sub>H<sub>2n</sub>), 27.1 (O-C<sub>n</sub>H<sub>2n</sub>), 23.4 (Pillararene, OCH2-CH2), 23.3 (Pillararene, OCH2-CH2), 22.4 (O-CnH2n), 16.5 (PCH2), 10.9 (Pillararene, OCH<sub>2</sub>CH<sub>2</sub>-CH<sub>3</sub>), 10.7 (Pillararene, OCH<sub>2</sub>CH<sub>2</sub>-CH<sub>3</sub>), 8.5 (PCH<sub>2</sub> CH<sub>3</sub>).; MS: simulated value for

 $C_{321}H_{462}O_{42}P_6Pt_3$  (5,760.14). (MALDI-TOF-MS) 5,880.1 ([M + Na]<sup>+</sup>).

Synthesis of G2: G2-H (23 mg 0.012 mmol) and 3 (153 mg, 0.076 mmol) were reacted in a similar manner to that of the preparation of G1. Alumina column chromatography was performed with dichloromethane as eluent: 94 mg (60% yield) of a slight yellow solid was afforded. M.p. 140 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.17 (d, *J* = 8.4 Hz, 12H), 7.02 (br, 18H), 6.92 (s, 30H), 6.86 (s, 30H), 6.65 (d, *J* = 8.8 Hz, 12H), 6.61 (d, *J* = 8.0 Hz, 12H), 3.65-3.97 (m, 228H), 2.47 (m, 12H), 2.23 (m, 108H), 1.67-1.91 (m, 132H), 1.27 (m, 174H), 1.11 (m, 102H), 0.94 (m, 90H), 0.61 (m, 12H), -0.26 (m, 12H), -0.73 (m, 12H), -1.55 (m, 12H), -2.24 (m, 12H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$  11.17, 11.05; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  157.8, 153.9, 149.8, 149.7, 137.6, 131.8, 129.2, 123.6, 114.5, 114.1, 113.7, 105.5, 73.8, 70.0, 69.6, 68.5, 56.2, 31.0, 30.5, 29.2, 28.6, 27.1, 23.3, 16.3, 10.8, 8.5, 8.2; MS: simulated value for C<sub>702</sub>H<sub>1020</sub>O<sub>84</sub>P<sub>18</sub>Pt<sub>9</sub> (13,107.77). (MALDI-TOF-MS) 13,106.1 (M<sup>+</sup>).

Synthesis of G3: G3-H (37 mg 0.007 mmol) and 3 (178 mg, 0.088 mmol) were reacted in a similar manner to that of the preparation of G1. Alumina column chromatography was performed with dichloromethane as eluent: 95 mg (49% yield) of a slight yellow solid was afforded. M.p. 151 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.18 (d, *J* = 8.4 Hz, 24H), 7.02 (br, 42H), 6.92 (s, 60H), 6.86 (s, 60H), 6.65 (d, *J* = 8.4 Hz, 24H), 6.61 (d, *J* = 8.4 Hz, 24H), 3.60-3.79 (m, 456H), 2.47 (m, 24H), 2.23 (m, 252H), 1.67-1.91 (m, 264H), 1.27 (m, 402H), 1.11 (m, 204H), 0.96 (m, 180H), 0.61 (m, 24H), -0.26 (m, 24H), -0.73 (m, 24H), -1.54 (m, 24H), -2.23 (m, 24H); <sup>31</sup>P NMR (CDCl<sub>3</sub>, 161.9 MHz):  $\delta$ 11.12, 11.01; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$ 157.8, 153.9, 149.8, 149.7, 137.6, 131.8, 128.3, 120.3, 114.5, 114.1, 113.7, 105.5, 73.8, 70.0, 69.6, 68.5, 56.2, 31.0, 30.5, 29.2, 28.6, 28.1, 27.1, 22.4, 16.3, 10.8, 8.4, 8.2; MS: simulated value for C<sub>1464</sub>H<sub>2136</sub>O<sub>168</sub>P<sub>42</sub>Pt<sub>21</sub> (27,803.02). (MALDI-TOF-MS) 27,839.4 ( [M + K]<sup>+</sup>).

**Synthesis of G4: G4-H** (42 mg 0.0034 mmol) and **3** (173 mg, 0.086 mmol) were reacted in a similar manner to that of the preparation of **G1**. Alumina column chromatography was performed with dichloromethane as eluent: 80 mg (41% yield) of a slight yellow solid was afforded. M.p. 175 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.18 (d, *J* = 8.0 Hz, 48H), 7.01 (br, 114H), 6.92 (s, 120H), 6.86 (s, 120H), 6.64 (d, *J* = 8.4 Hz, 48H), 6.60 (d, *J* = 8.4 Hz, 48H), 3.65-3.97 (m, 912H), 2.44 (m, 48H), 2.23 (m, 540H), 1.67-1.91 (m, 528H), 1.27 (m, 858H), 1.11 (m, 408H), 0.96 (m, 360H), 0.61 (m, 48H), -0.26 (m, 48H), -0.73 (m, 48H), -1.55 (m, 48H), -2.23 (m, 48H); <sup>31</sup>P NMR (CDCl<sub>3</sub>,

161.9 MHz):  $\delta$  11.36, 11.23; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  157.6, 153.8, 149.7, 149.6, 137.5, 131.8, 128.3, 123.5, 114.4, 114.1, 113.7, 105.4, 73.7, 69.8, 69.5, 68.4, 56.1, 30.8, 30.5, 29.2, 28.6, 28.1, 27.0, 23.2, 16.3, 10.8, 8.4, 8.2. MS: simulated value for C<sub>2988</sub>H<sub>4368</sub>O<sub>336</sub>P<sub>90</sub>Pt<sub>4</sub> (57,193.52). (GPC)  $M_n$  52,117.



Figure S24. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G1.











Figure S30. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G2.



Figure S31.<sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G2.



Figure S32. MALDI-TOF-MS spectrum of G2.















Figure S39. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, room temperature, 400 MHz) of G4.



Figure S40. <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>, room temperature, 161.9 MHz) of G4.



Figure S41. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>, room temperature, 125 MHz) of G4.



Figure S42. GPC data of G4.







Figure S44. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of Gn-H.



Figure S45. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of Gn-T.



Figure S46. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of 1, G1 and 3.



Figure S47. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of G2-H, G2 and 3.



Figure S48. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of G3-H, G3 and 3.



Figure S49. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of G4-H, G4 and 3.



Figure S50. Partial <sup>1</sup>H and <sup>31</sup>P NMR spectra (CDCl<sub>3</sub>, room temperature) of **3** and **Gn**.



Figure S51. MALDI-TOF-MS spectra of G1, G2 and G3.

Section C. Additional Characterization of Rotaxane-terminated Dendrimers



**Figure S52.** Dynamic light scattering (DLS) measurements of the hydrodynamic radius ( $R_h$ ) of rotaxane-terminated dendrimers **G2-G4**. (It was not possible to obtain the hydrodynamic radius ( $R_h$ ) of G1 by DLS because its size is below the limit for this technique.)



Figure S53. TEM image of the rotaxane-terminated dendrimers Gn.



**Figure S54.** (a) UV-vis absorption spectra of rotaxane-terminated dendrimers **Gn** (n = 1, 2, 3, and4) in CHCl<sub>3</sub>; (b) emission spectra of rotaxane-terminated dendrimers **Gn** (n = 1, 2, 3, and 4) in THF.

	absorption	emission
compound	$\lambda_{abs}/nm (\epsilon/dm^3 mol^{-1} cm^{-1})$	$\lambda_{\rm em}/{\rm nm}$
G1	269 (30715), 292 (31695), 347 (23555)	380
G2	269 (30570), 292 (38600), 347 (40365)	380
G3	269 (90270), 292 (105765), 347 (98450)	380
G4	269 (2150350), 292 (2317150), 347 (1938800)	380

Table S1. Photophysical Properties of rotaxane-terminated dendrimers Gn.

#### References

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