

Electronic Supplementary Information

Facile synthesis and bridgehead-functionalization of bicyclo[3.3.3]pentasiloxanes

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1. Experimental Details

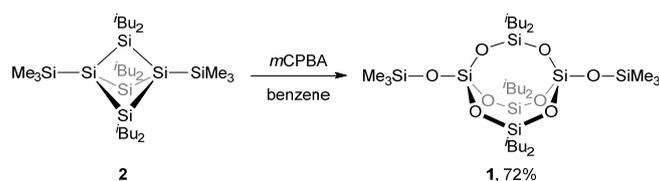
General Procedure

All reactions treating air-sensitive compounds were carried out under an inert atmosphere (N₂ or Ar) using a high-vacuum line and standard Schlenk techniques, or a glove box, as well as dry and oxygen-free solvents. NMR spectra were recorded on a Bruker Avance III 500 FT NMR spectrometer. The ¹H NMR chemical shifts were referenced to residual ¹H of the solvents; C₆D₆ (¹H δ 7.16).^{S1} The ¹³C and ²⁹Si NMR chemical shifts were relative to Me₄Si in ppm (δ 0.00). Sampling of air-sensitive compounds was carried out using a VAC NEXUS 100027 type glove box. Mass spectra were recorded on a Bruker Daltonics SolariX 9.4T. UV-vis spectra were recorded on a JASCO V-660 spectrometer. Recycling preparative HPLC using ethanol (EtOH) as an eluent was performed at 5.0 mL/min rate using a Japan Analytical Industry Co., LC9201 equipped with an ODS column (FUJI SILYSIA CHEMICAL LTD., Chromatorex[®] C18, 20φ mm × 250 mm). Thermogravimetric analysis (TGA) was undertaken using a Perkin-Elmer Pyris 1 TGA apparatus under a nitrogen atmosphere (100 mL/min). Differential scanning calorimetry (DSC) was measured using Pyris Diamond DSC (PerkinElmer) under a nitrogen atmosphere (100 mL/min).

Materials

Benzene, toluene, and THF were dried with VAC-103991 type solvent purifiers. Benzene-*d*₆ (C₆D₆) was dried with molecular sieves (3A). Potassium *tert*-butoxide (^tBuOK), 18-crown-6-ether (18-c-6), *m*-chloroperoxybenzoic acid (*m*CPBA) in 60~70% purity, iodomethane (MeI), *tert*-butylchlorodimethylsilane (^tBuMe₂SiCl), chlorodimethylvinylsilane, triethylamine (NEt₃), and standard organic solvents were commercially available and used without further purification. 1,3-bis(trimethylsilyl)hexaisobutylbicyclo[1.1.1]pentasilane (**2**) was prepared according to the published procedures.^{S2}

Synthesis of 1



In a round-bottom flask (200 mL) equipped with a magnetic stir bar, **2** (795 mg, 1.26 mmol), *m*CPBA (3.10 g, ca. 11.5 mmol) and benzene (40 mL) were placed. The reaction mixture was stirred for 2 d at 50 °C and then the volatiles were removed under reduced pressure. The resulting residue was dissolved in hexane and then the mixture was subjected to silica-gel column chromatography to remove insoluble materials, the resulting benzoic acid, and unidentified yellow materials. Removal of the solvent in *vacuo* afforded analytically pure **1** (696 mg, 0.919 mmol, 72%) as a colorless powder.

1: a colorless powder; mp 68-69 °C; ¹H NMR (500 MHz, C₆D₆, 296 K) δ 0.23 (s, 18H, Si(CH₃)₃), 0.90 (d, *J* = 7.1 Hz, 12H, CH₂CH(CH₃)₂), 1.13 (d, *J* = 6.6 Hz, 36H, CH₂CH(CH₃)₂), 2.02 – 2.11 (m, 6H, CH₂CH(CH₃)₂); ¹³C{¹H} NMR (126 MHz, C₆D₆, 297 K) δ 1.6 (Si(CH₃)₃), 24.4 (CH₂CH(CH₃)₂), 26.5 (CH₂CH(CH₃)₂), 27.4 (CH₂CH(CH₃)₂); ²⁹Si NMR (99.4 MHz, C₆D₆, 296 K) δ -101.7 (*S*^Q), -15.2 (*S*^D), 11.5 (*S*^M); Elemental Anal. calcd for C₃₀H₇₂O₆Si₇: C, 47.57; H, 9.58%; found: C, 47.70; H, 9.66%; MS (EI, 70 eV) *m/z* (%) 741 (29) [M⁺-CH₃], 699 (100) [M⁺-C₄H₉].

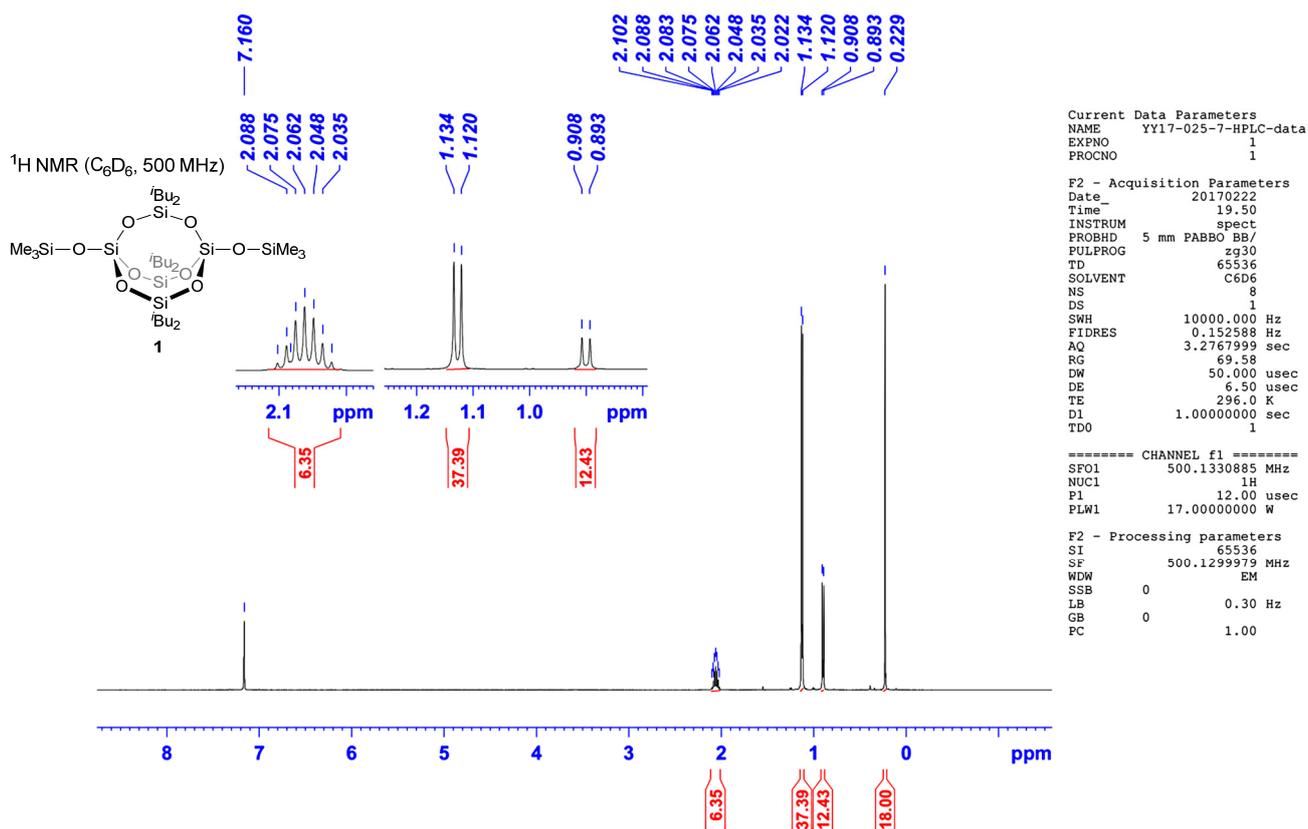


Figure S1. ¹H NMR spectrum of **1** in C₆D₆ (126 MHz, 296 K).

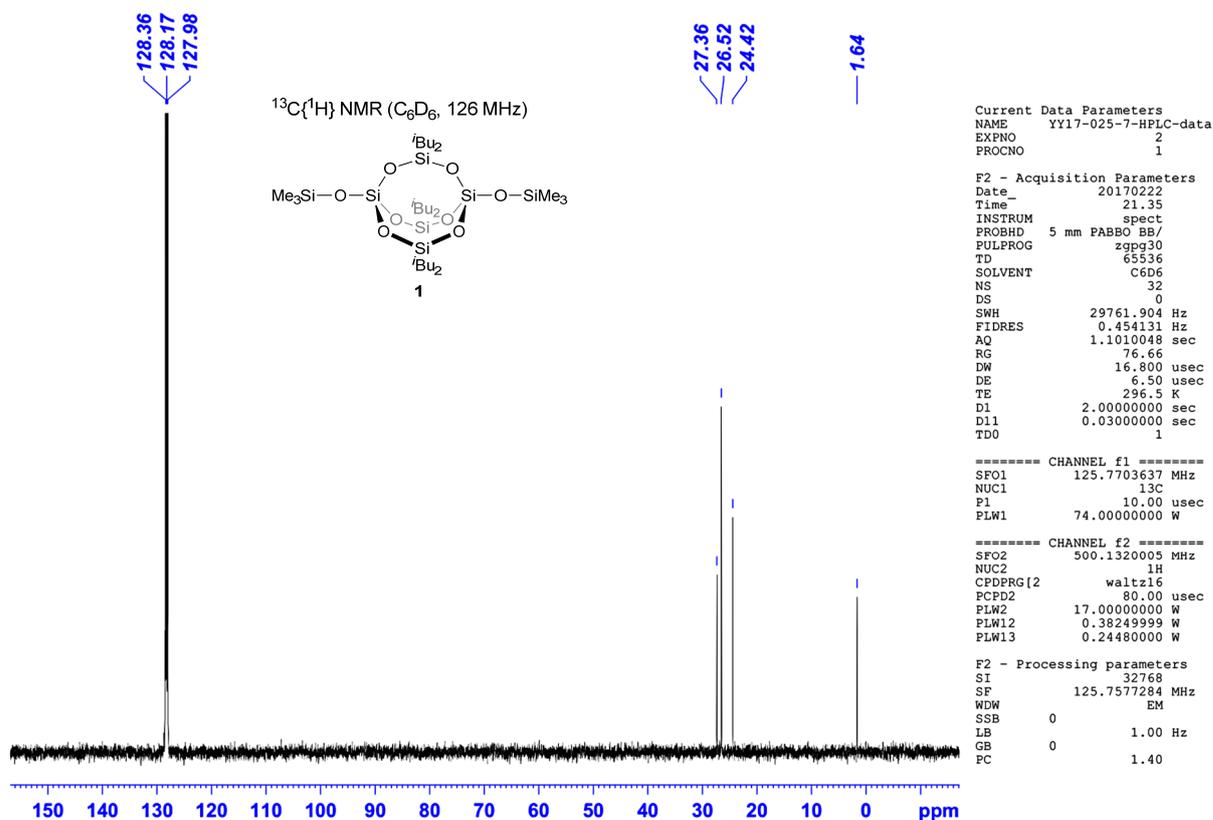


Figure S2. ¹³C{¹H} NMR spectrum of **1** in C₆D₆ (126 MHz, 297 K).

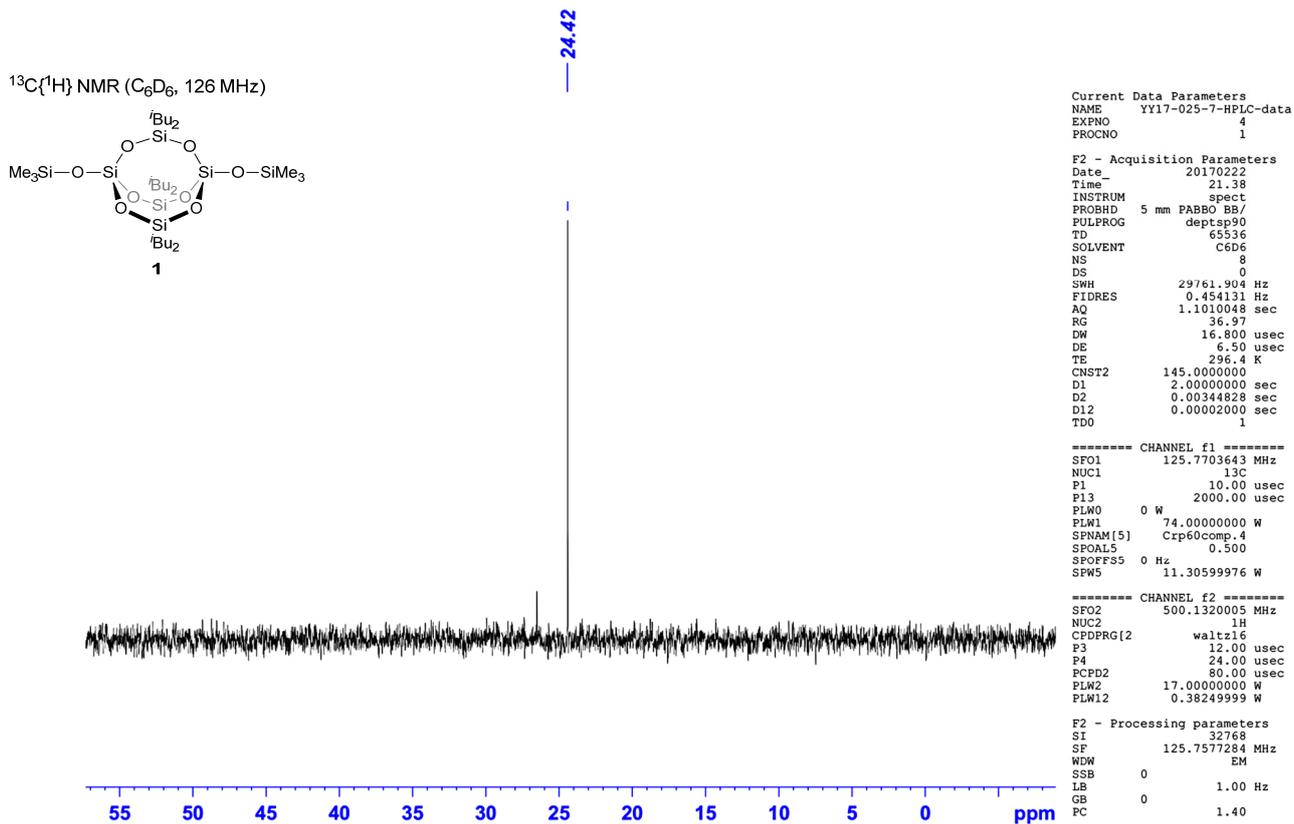


Figure S3. $^{13}\text{C}\{^1\text{H}\}$ NMR (dept90) spectrum of **1** in C_6D_6 (126 MHz, 296 K).

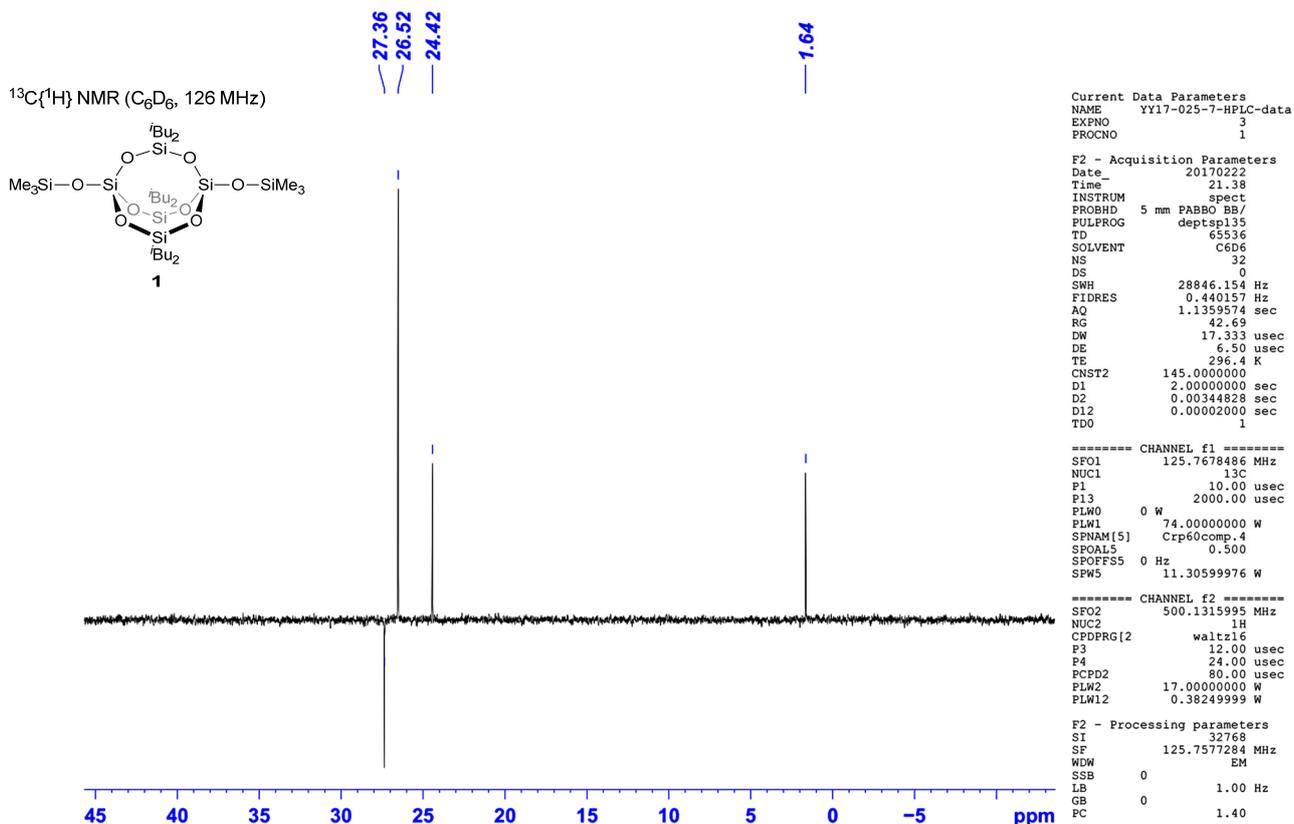


Figure S4. $^{13}\text{C}\{^1\text{H}\}$ NMR (dept135) spectrum of **1** in C_6D_6 (126 MHz, 296 K).

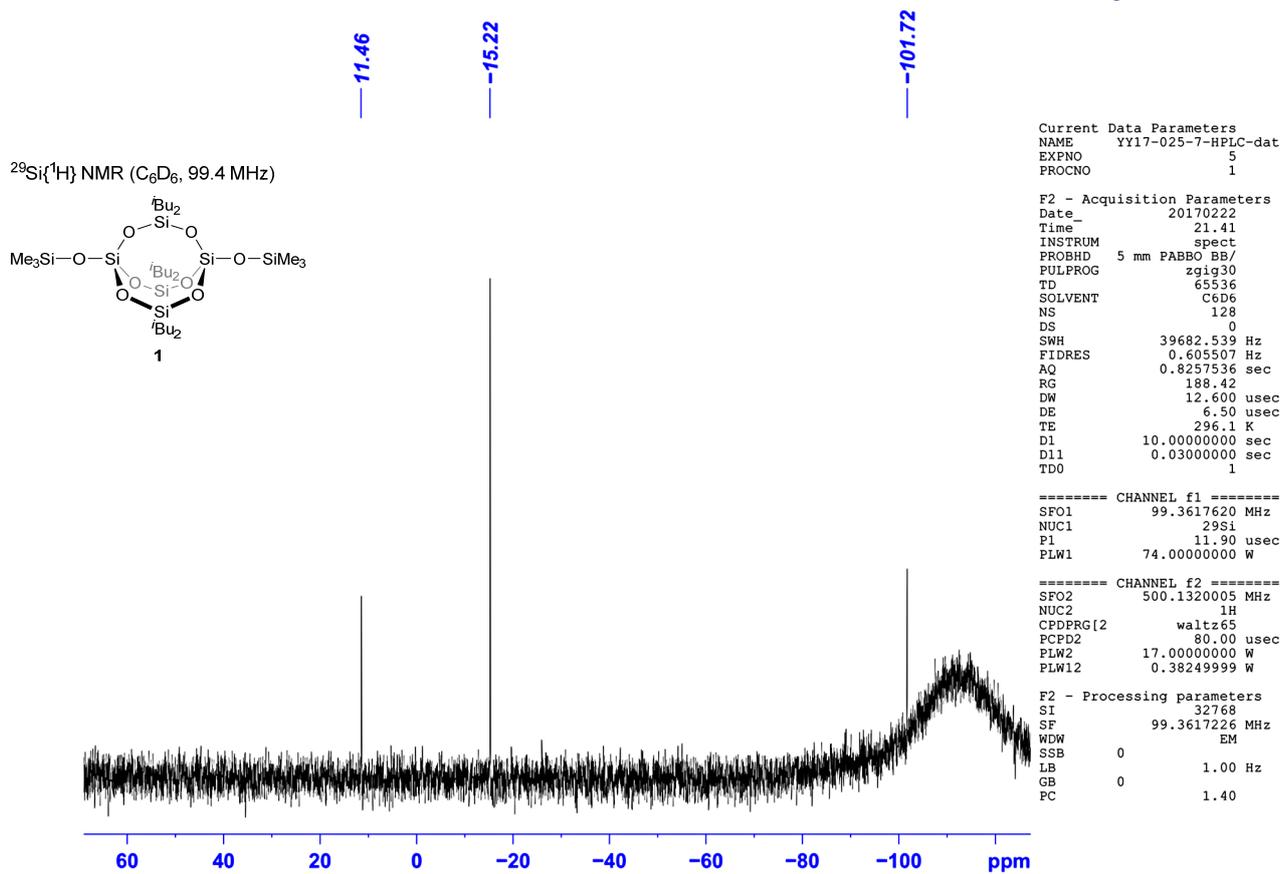


Figure S5. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **1** in C_6D_6 (99.4 MHz, 296 K).

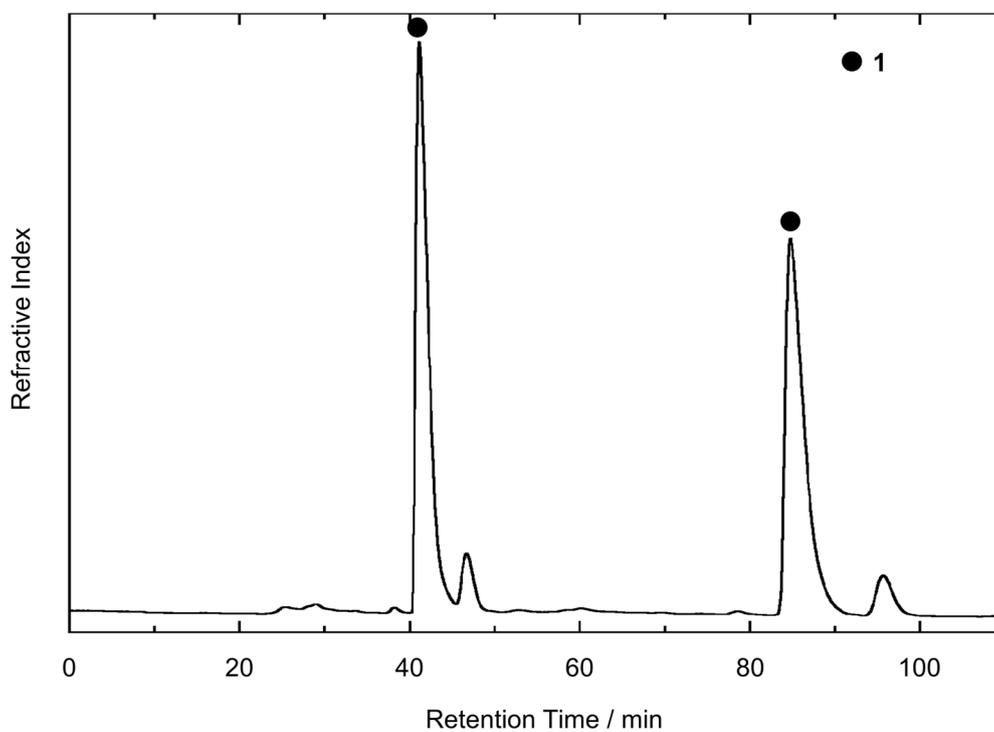
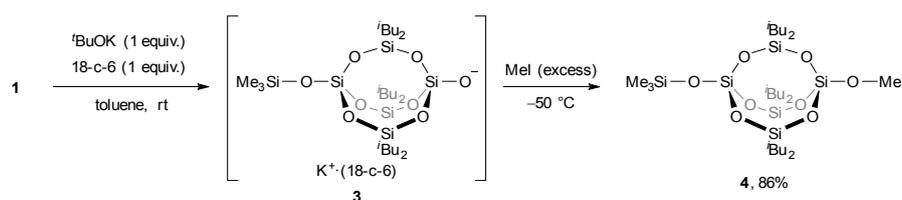


Figure S6. HPLC chart recorded during purification of **1**.

Synthesis of 4



In a Schlenk tube (50 mL) equipped with a magnetic stir bar, **1** (100 mg, 0.132 mmol), 18-c-6 (35.3 mg, 0.134 mmol), ^tBuOK (15.4 mg, 0.137 mmol) and toluene (3 mL) were charged. The reaction mixture was stirred for 1.0 h at room temperature and then the mixture was cooled to -50 °C. To the cooled solution, MeI (316 mg, 2.11 mmol) was added with a syringe. After the reaction mixture was allowed to warm to ambient temperature, the resulting insoluble materials were filtered off through a silica gel. The resulting colorless oil was subjected to HPLC (ODS, EtOH as an eluent, room temperature) to afford **4** (79.1 mg, 0.113 mmol, 86%) as a colorless oil.

4: a colorless oil; ¹H NMR (500 MHz, C₆D₆, 300 K) δ 0.23 (s, 9H, Si(CH₃)₃), 0.89 (d, *J* = 7.0 Hz, 12H, CH₂CH(CH₃)₂), 1.13 (d, *J* = 6.5 Hz, 36H, CH₂CH(CH₃)₂), 2.00 – 2.09 (m, 6H, CH₂CH(CH₃)₂), 3.54 (s, 3H, OCH₃); ¹³C{¹H} NMR (126 MHz, C₆D₆, 300 K) δ 1.6 (Si(CH₃)₃), 24.4 (CH₂CH(CH₃)₂), 26.46 (CH₂CH(CH₃)₂), 26.49 (CH₂CH(CH₃)₂), 27.3 (CH₂CH(CH₃)₂), 51.0 (OCH₃); ²⁹Si{¹H} NMR (99.4 MHz, C₆D₆, 300 K) δ -101.9 (*S*ⁱ_Q), -94.7 (*S*ⁱ_T), -14.6 (*S*ⁱ_D), 11.5 (*S*ⁱ_M); MS (EI, 70 eV) *m/z* (%) 684 (11) [M⁺+H-CH₃], 642 (100) [M⁺+H-(ⁱBu)], 585 (27) [M⁺+H-2(ⁱBu)]; Elemental analysis calcd for C₂₈H₆₆O₈Si₆: C, 48.09; H, 9.51; found: C, 48.03; H, 9.49.

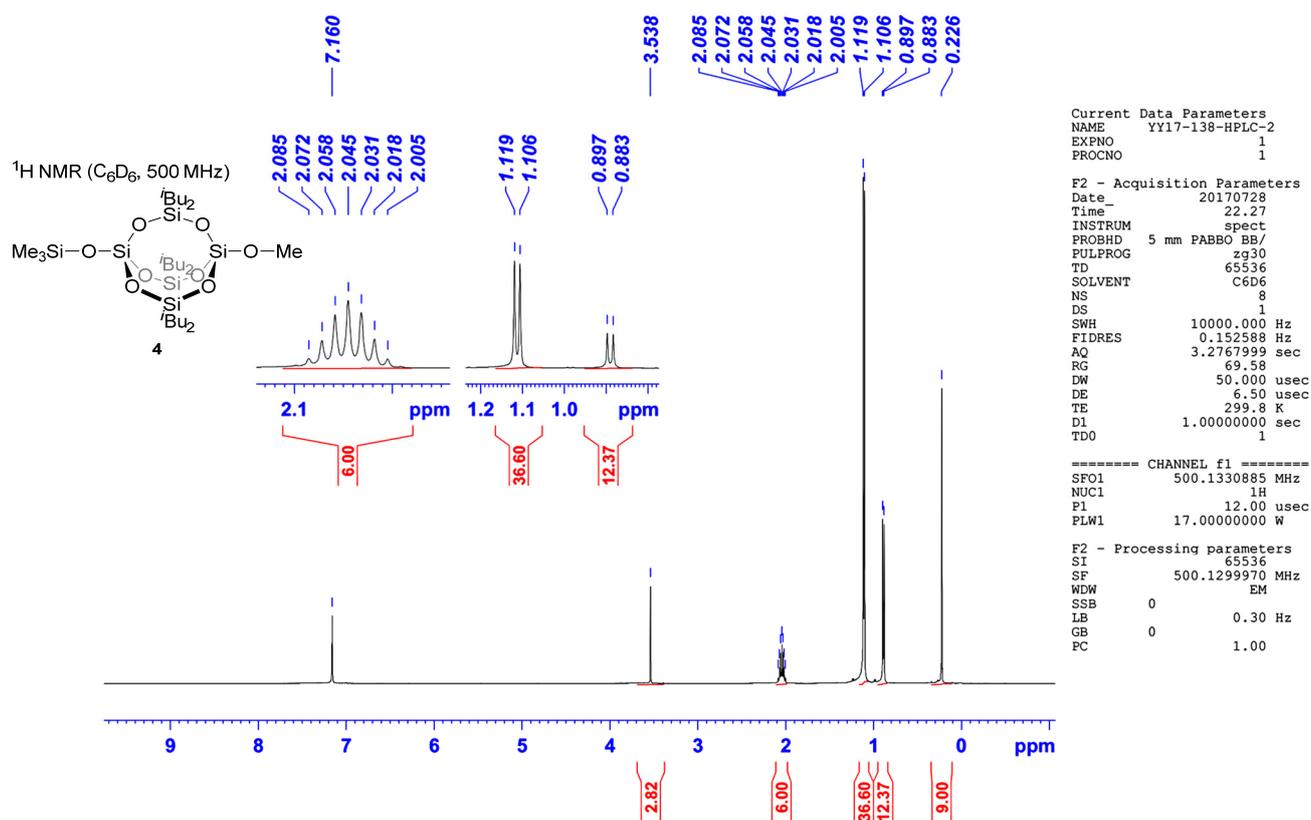


Figure S7. ¹H NMR spectrum of **4** in C₆D₆ (126 MHz, 300 K).

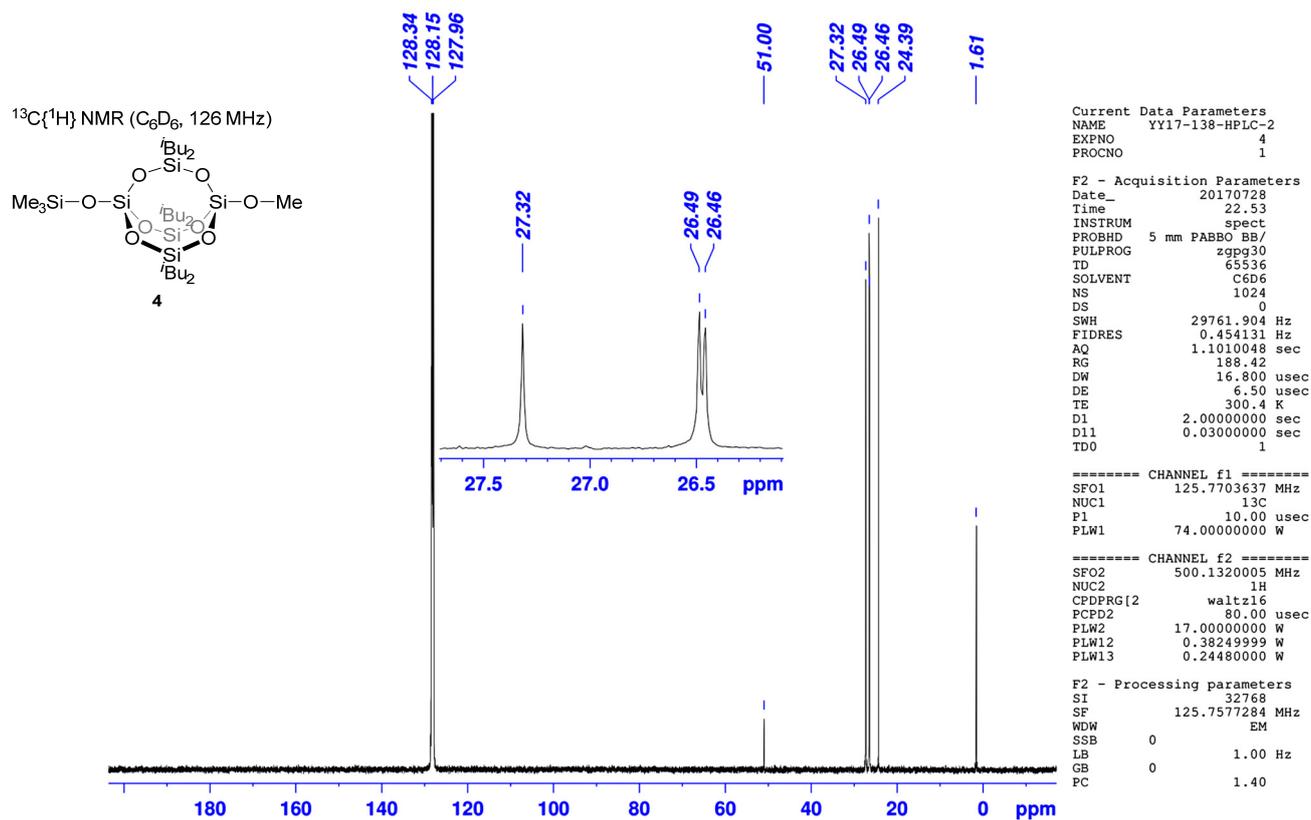


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **4** in C_6D_6 (126 MHz, 300 K).

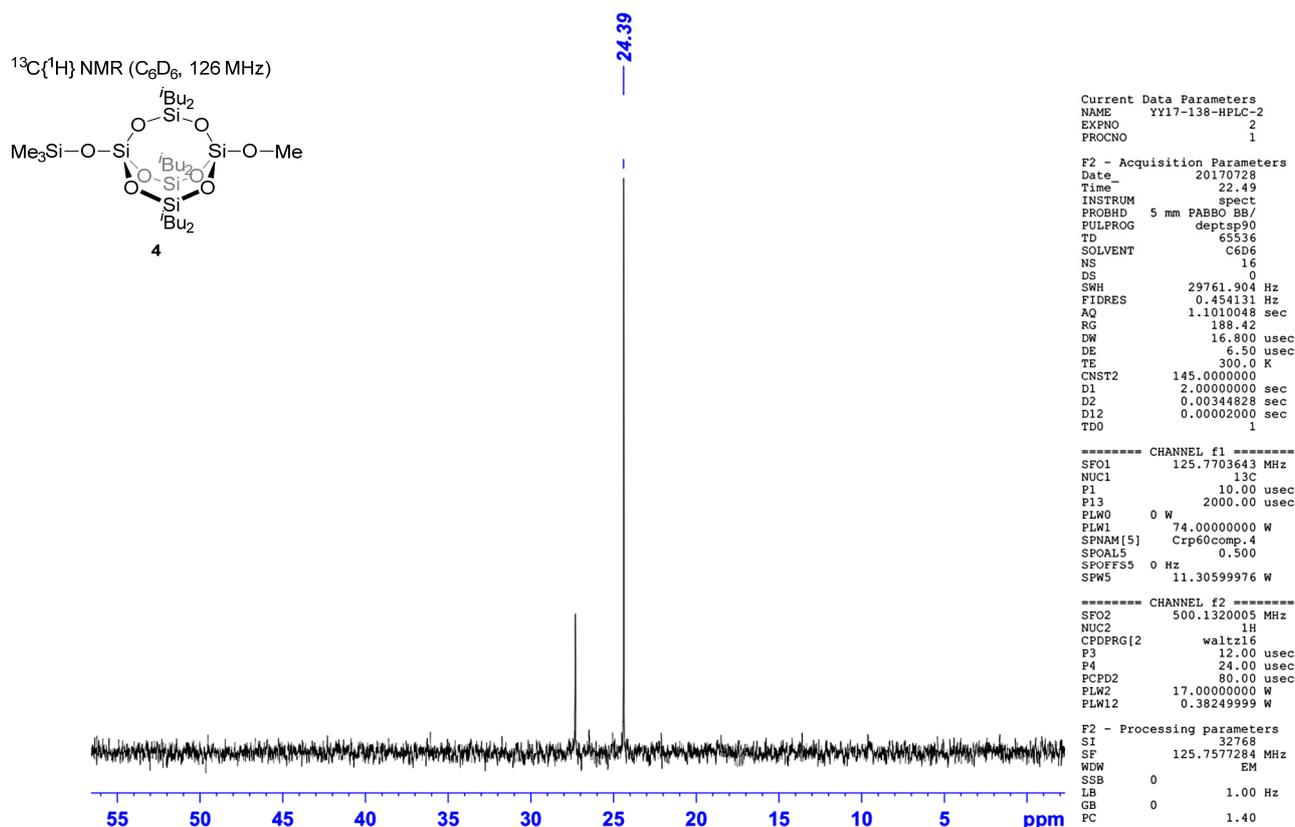


Figure S9. $^{13}\text{C}\{^1\text{H}\}$ NMR (dept90) spectrum of **4** in C_6D_6 (126 MHz, 300 K).

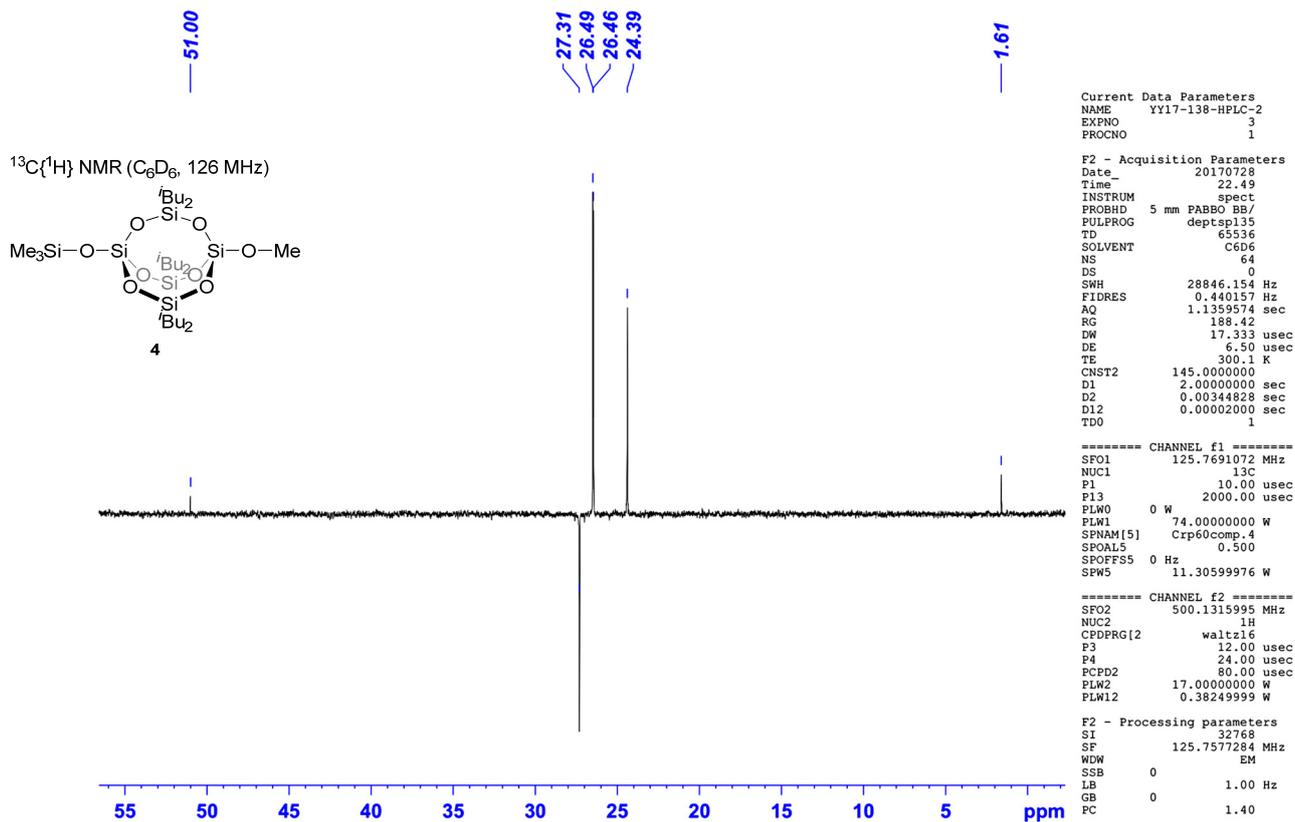


Figure S10. ¹³C{¹H} NMR (dept135) spectrum of **4** in C₆D₆ (126 MHz, 300 K).

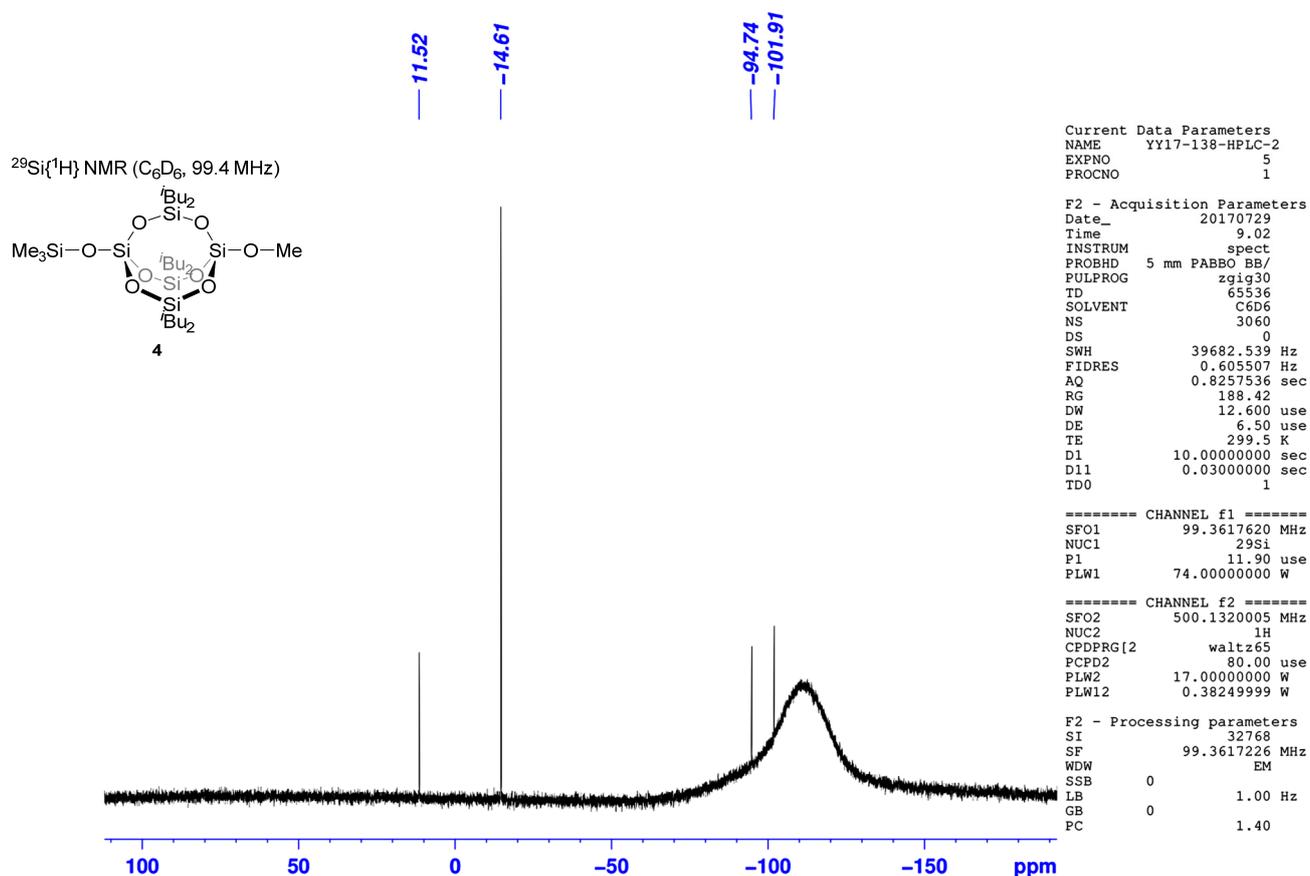


Figure S11. ²⁹Si{¹H} NMR spectrum of **4** in C₆D₆ (99.4 MHz, 300 K).

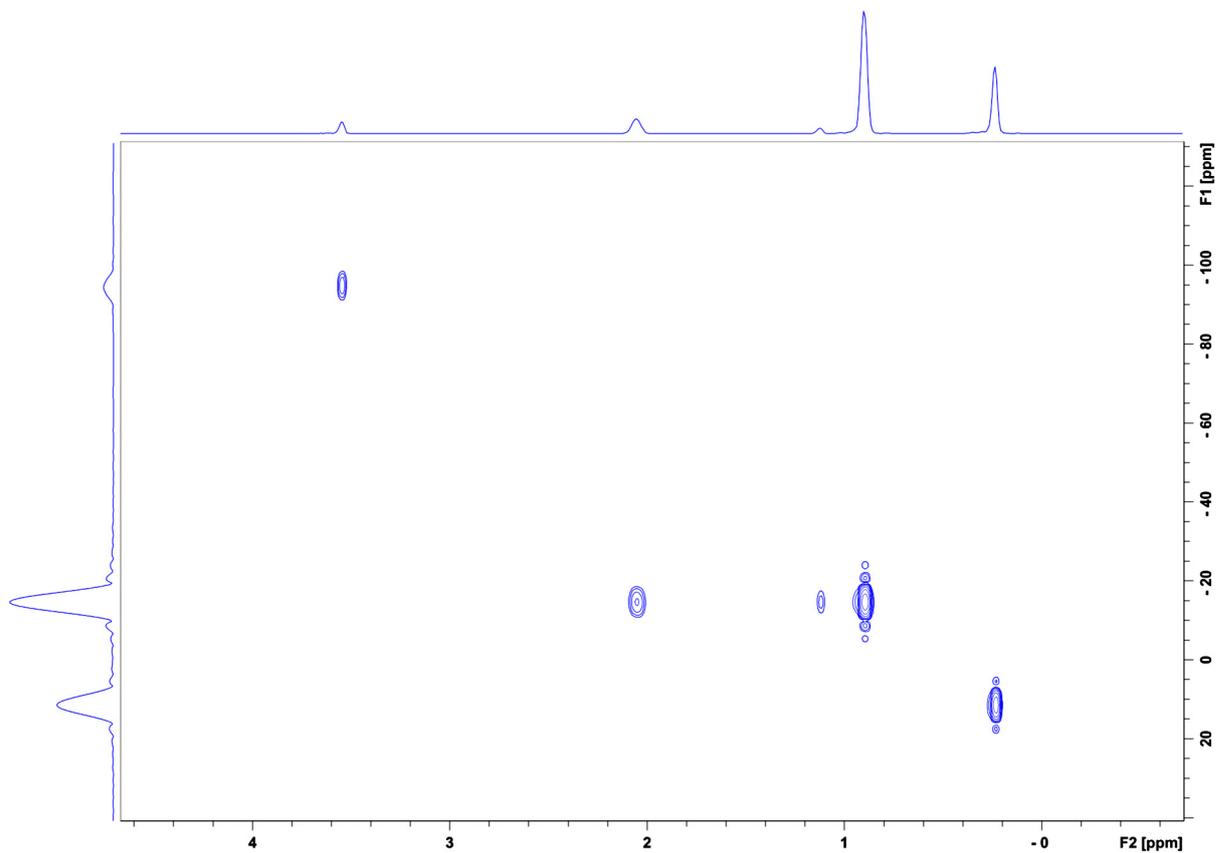


Figure S12. ^1H - ^{29}Si HMBC 2D NMR spectrum of **4** in C_6D_6 (300 K).

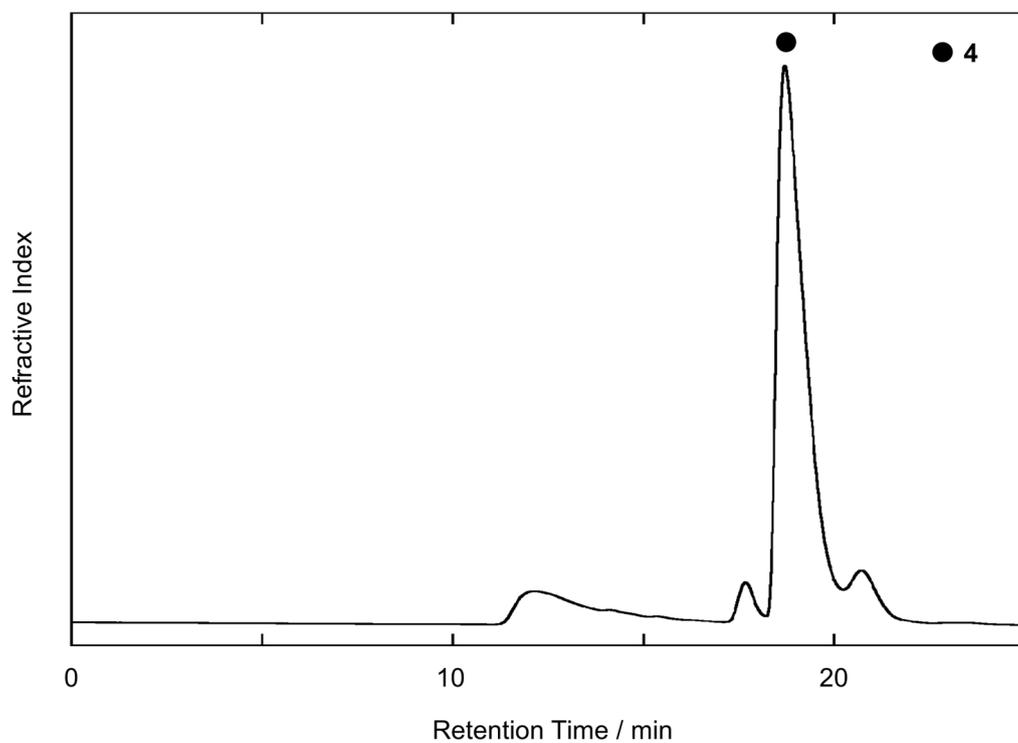
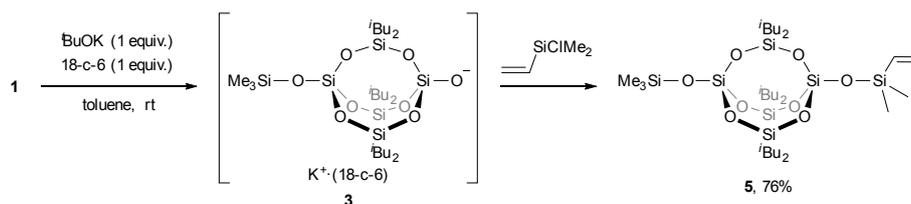


Figure S13. HPLC chart recorded during purification of **4**.

Synthesis of 5



In a Schlenk tube (50 mL) equipped with a magnetic stir bar, compound **1** (100.0 mg, 0.132 mmol), 18-crown-6-ether (35.2 mg, 0.133 mmol), $t\text{BuOK}$ (15.3 mg, 0.136 mmol) and toluene (3 mL) were charged. The reaction mixture was stirred for 1.0 h at room temperature and then chlorodimethylvinylsilane (18 mg, 0.149 mmol) was added to the mixture. The resulting insoluble materials were filtered off through a silica gel. After removal of the solvent from the filtrate in *vacuo*, the resulting oil was subjected to HPLC (ODS, EtOH as an eluent, room temperature) to afford **5** (77.0 mg, 0.100 mmol, 76%) as a colorless oil.

5: a colorless oil; ^1H NMR (500 MHz, C_6D_6 , 300 K) δ 0.23 (s, 9H, $\text{Si}(\text{CH}_3)_3$), 0.30 (s, 6H, $\text{Si}(\text{CH}_3)_2(\text{CH}=\text{CH}_2)$), 0.90 (d, $J = 7.1$ Hz, 12H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 1.13 (d, $J = 6.6$ Hz, 36H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 2.02 – 2.10 (m, 6H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 5.88 (dd, $J = 3.8, 20.4$ Hz, 1H, $\text{SiMe}_2(\text{CH}=\text{CH}_2)$), 5.96 (dd, $J = 3.8, 14.9$ Hz, 1H, $\text{SiMe}_2(\text{CH}=\text{CH}_2)$), 6.24 (dd, $J = 14.9, 20.4$ Hz, 1H, $\text{SiMe}_2(\text{CH}=\text{CH}_2)$); $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, C_6D_6 , 301 K) δ 0.2 ($\text{Si}(\text{CH}_3)_2(\text{CH}=\text{CH}_2)$), 1.6 ($\text{Si}(\text{CH}_3)_3$), 24.4 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 26.51 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 26.52 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 27.4 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 132.5 ($\text{SiMe}_2(\text{CH}=\text{CH}_2)$), 138.7 ($\text{SiMe}_2(\text{CH}=\text{CH}_2)$); $^{29}\text{Si}\{^1\text{H}\}$ NMR (99.4 MHz, C_6D_6 , 300 K) δ -101.78 (Si^{O}), -101.76 (Si^{O}), -15.2 (Si^{D}), -0.4 (Si^{M} , $\text{SiMe}_2(\text{HC}=\text{CH}_2)$), 11.5 (Si^{M} , SiMe_3); MS (EI, 70 eV) m/z (%) 754 (11) [$\text{M}^+\text{H}-\text{Me}$], 712 (78) [$\text{M}^+\text{H}-(i\text{-Bu})$], 655 (59) [$\text{M}^+\text{H}-2(i\text{-Bu})$], 73 (100) [SiMe_3^+]. Elemental analysis calcd for $\text{C}_{31}\text{H}_{72}\text{O}_8\text{Si}_7$: C, 48.39; H, 9.43; found: C, 48.42; H, 9.34.

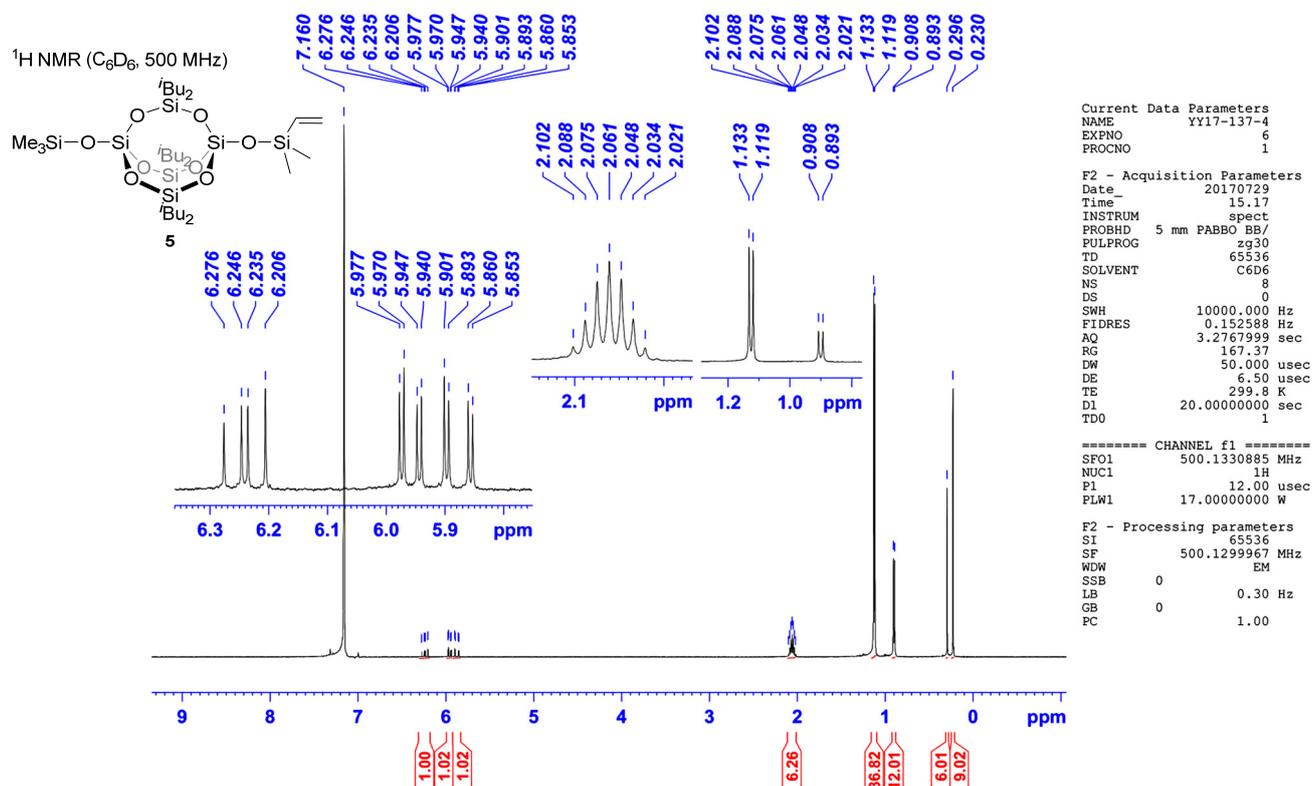


Figure S14. ^1H NMR spectrum of **5** in C_6D_6 (126 MHz, 300 K).

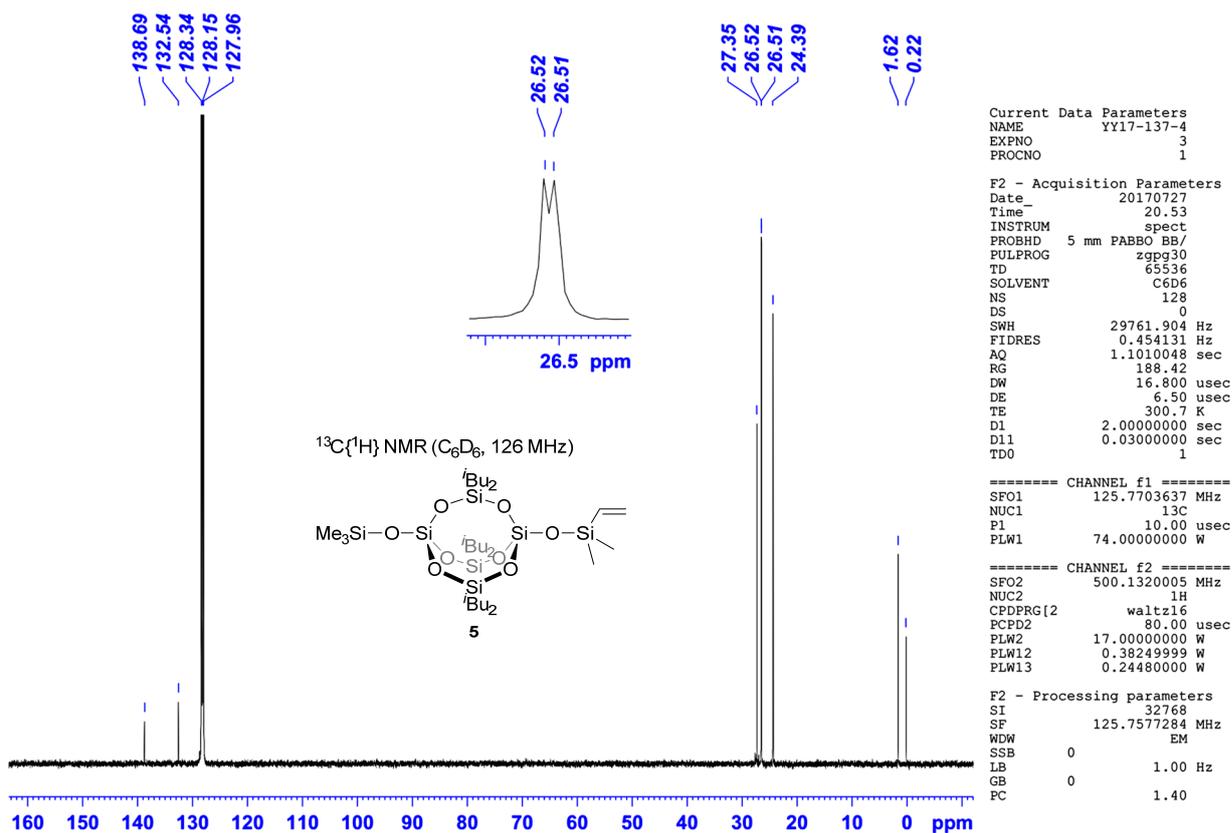


Figure S15. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **5** in C_6D_6 (126 MHz, 301 K).

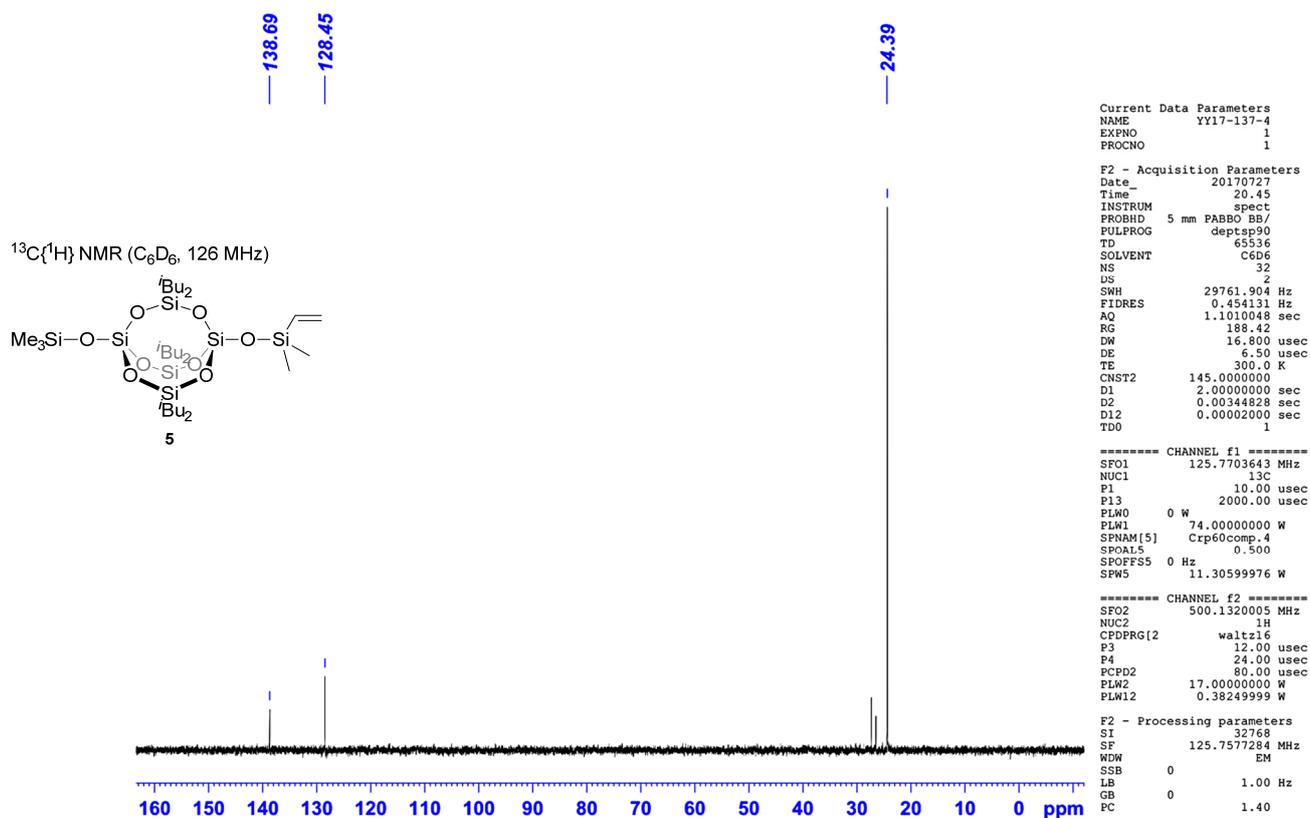


Figure S16. $^{13}\text{C}\{^1\text{H}\}$ NMR (dept90) spectrum of **5** in C_6D_6 (126 MHz, 300 K).

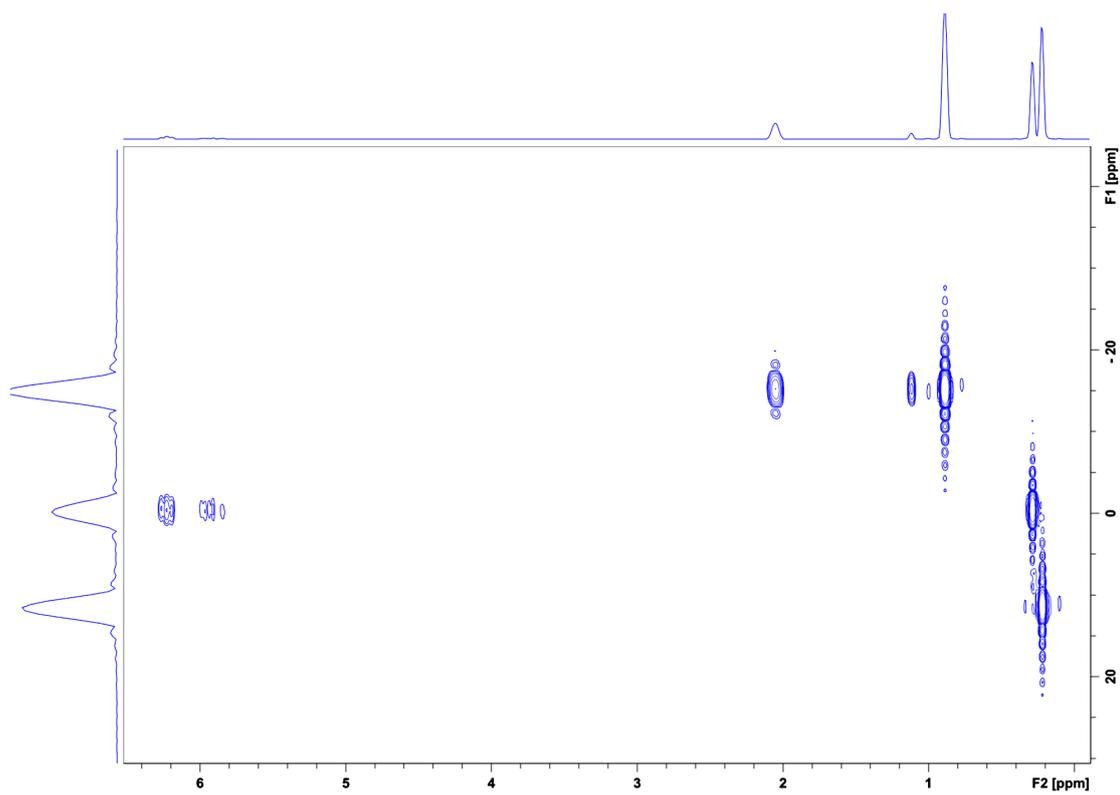


Figure S19. ^1H - ^{29}Si HMBC 2D NMR spectrum of **5** in C_6D_6 (300 K).

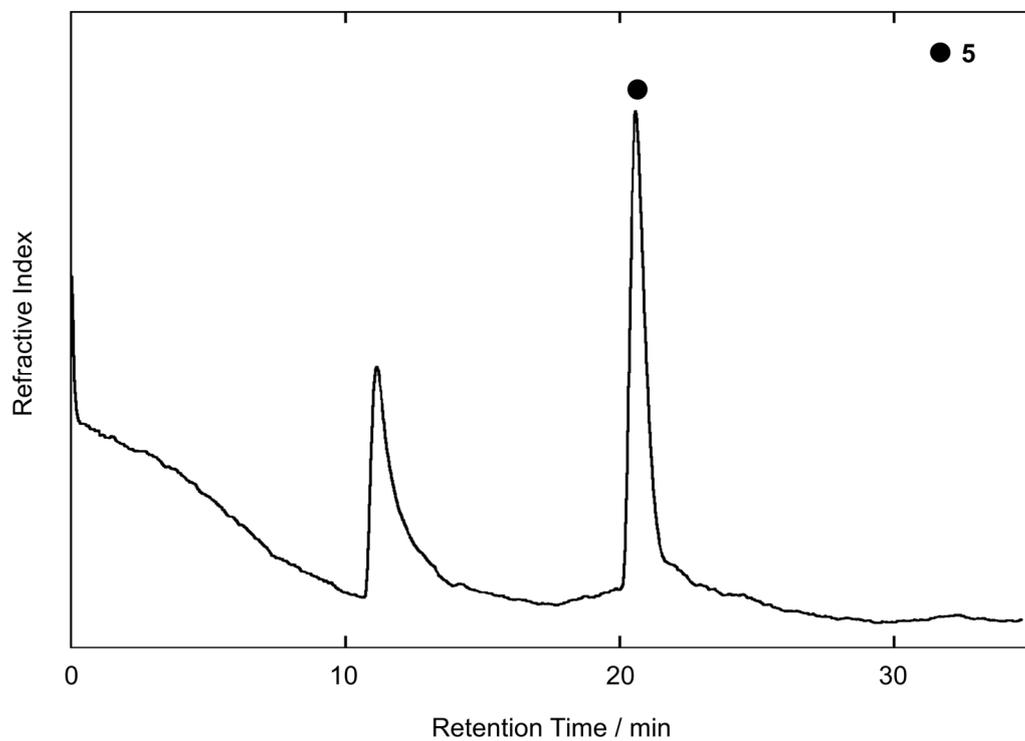
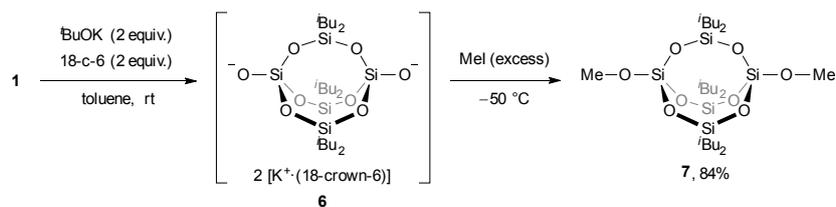


Figure S20. HPLC chart recorded during purification of **5**.

Synthesis of 7



In a Schlenk tube (50 mL) equipped with a magnetic stir bar, **1** (100 mg, 0.132 mmol), $18\text{-c-}6$ (73.5 mg, 0.278 mmol), tBuOK (31.6 mg, 0.282 mmol), and toluene (6 mL) were charged. The reaction mixture was stirred for 1.0 h at room temperature and then toluene (4 mL) was added to mixture. The mixture was cooled to $-50\text{ }^\circ\text{C}$ before MeI (406 mg, 2.71 mmol) was added to the cooled solution through a syringe. After the resulting suspension was allowed to warm to ambient temperature, the resulting insoluble materials were filtered off through a silica gel. Removal of the volatiles under reduced pressure afforded **7** as a colorless waxy solid (70.8 mg, 0.110 mmol, 84%).

7: a colorless waxy solid; ^1H NMR (500 MHz, C_6D_6 , 298 K) δ 0.88 (d, $J = 7.0$ Hz, 12H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 1.10 (d, $J = 6.5$ Hz, 36H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 1.99 – 2.07 (m, 6H, $\text{CH}_2\text{CH}(\text{CH}_3)_2$), 3.53 (s, 6H, OCH_3); $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, C_6D_6 , 299 K) δ 24.4 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 26.5 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 27.3 ($\text{CH}_2\text{CH}(\text{CH}_3)_2$), 51.0 (OCH_3); $^{29}\text{Si}\{^1\text{H}\}$ NMR (99.4 MHz, C_6D_6 , 298 K) δ -94.9 (Si^{T}), -14.0 (Si^{D}); MS (EI, 70 eV) m/z (%): 583 (100) [$\text{M}^+ - (i\text{-Bu})$], 527 (65.1) [$\text{M}^+ + \text{H} - 2(i\text{-Bu})$], 471 (14) [$\text{M}^+ + \text{H} - 3(i\text{-Bu})$]. Elemental analysis calcd for $\text{C}_{26}\text{H}_{60}\text{O}_8\text{Si}_7$: C, 48.70; H, 9.43; found: C, 48.85; H, 9.61.

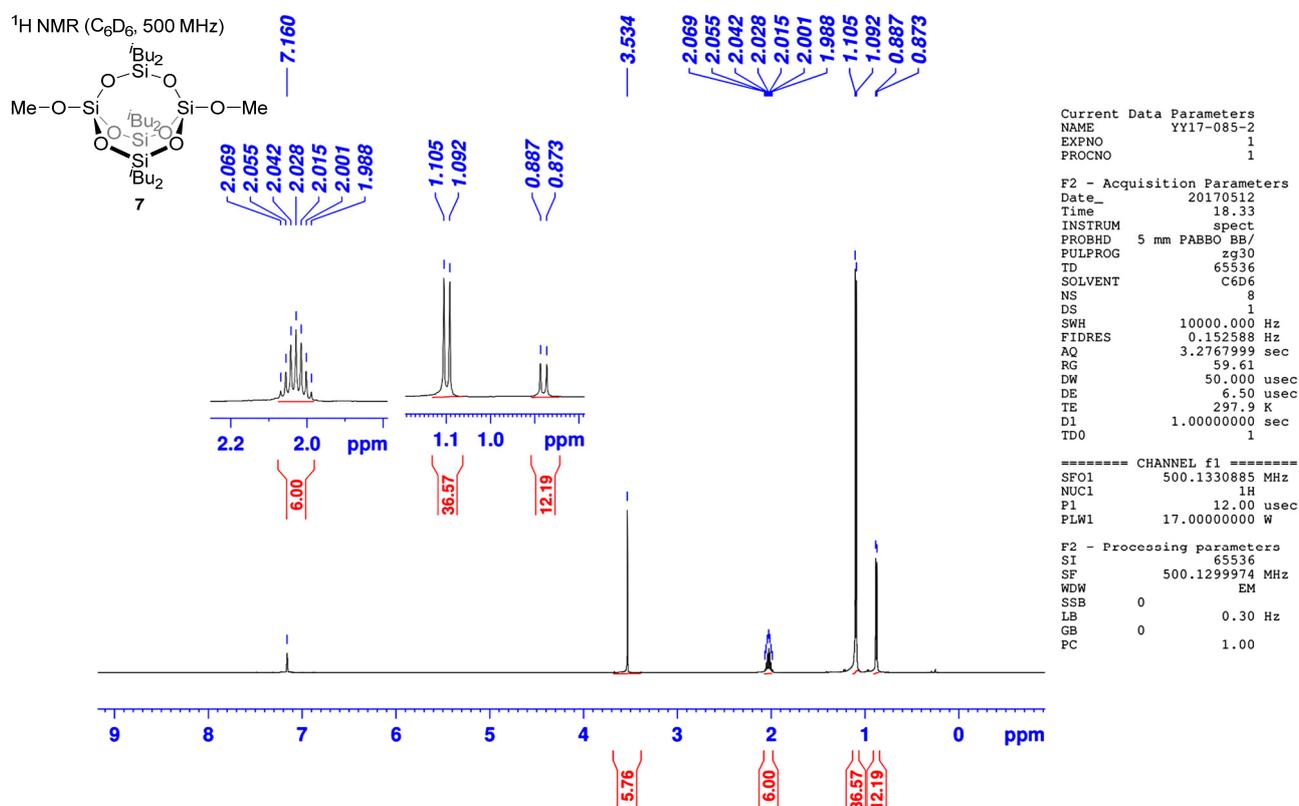


Figure S21. ^1H NMR spectrum of **7** in C_6D_6 (126 MHz, 298 K).

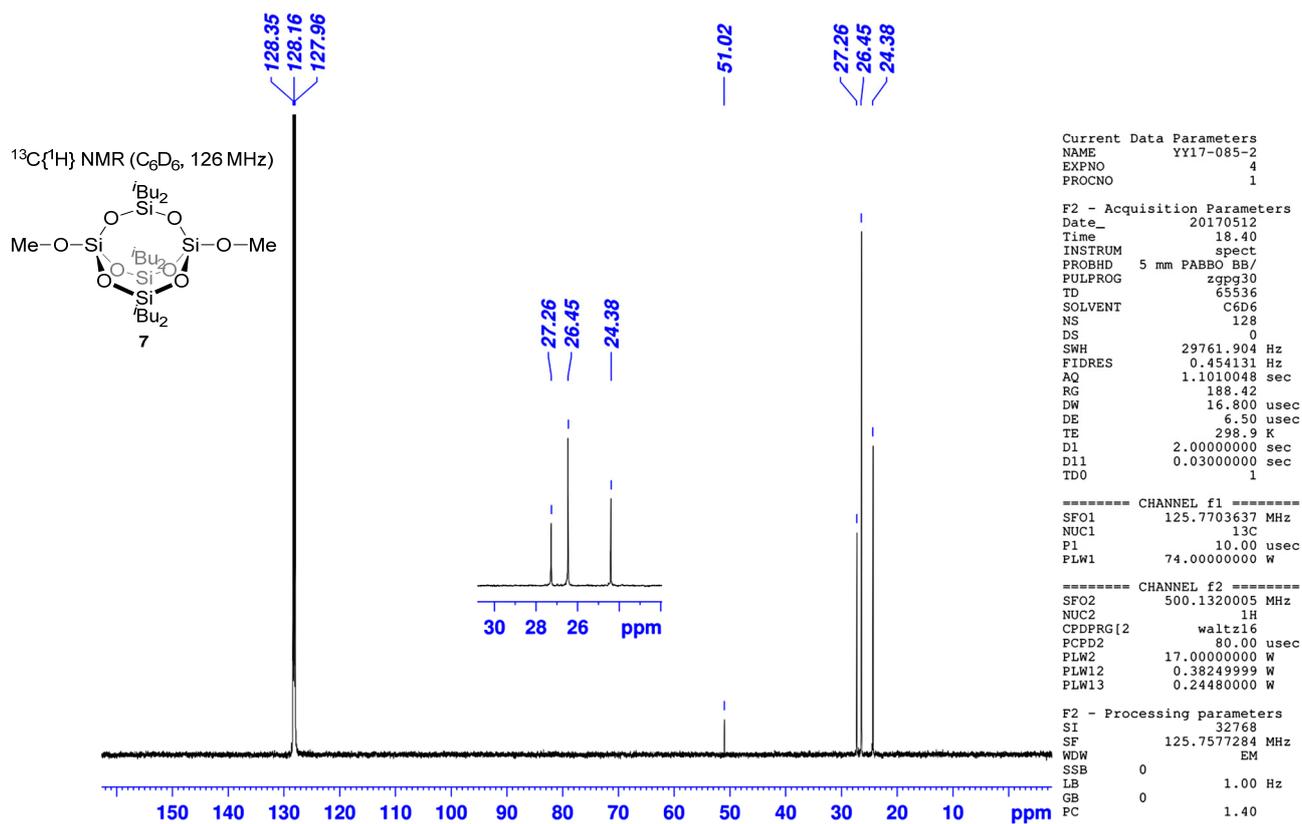


Figure S22. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **7** in C_6D_6 (126 MHz, 299 K).

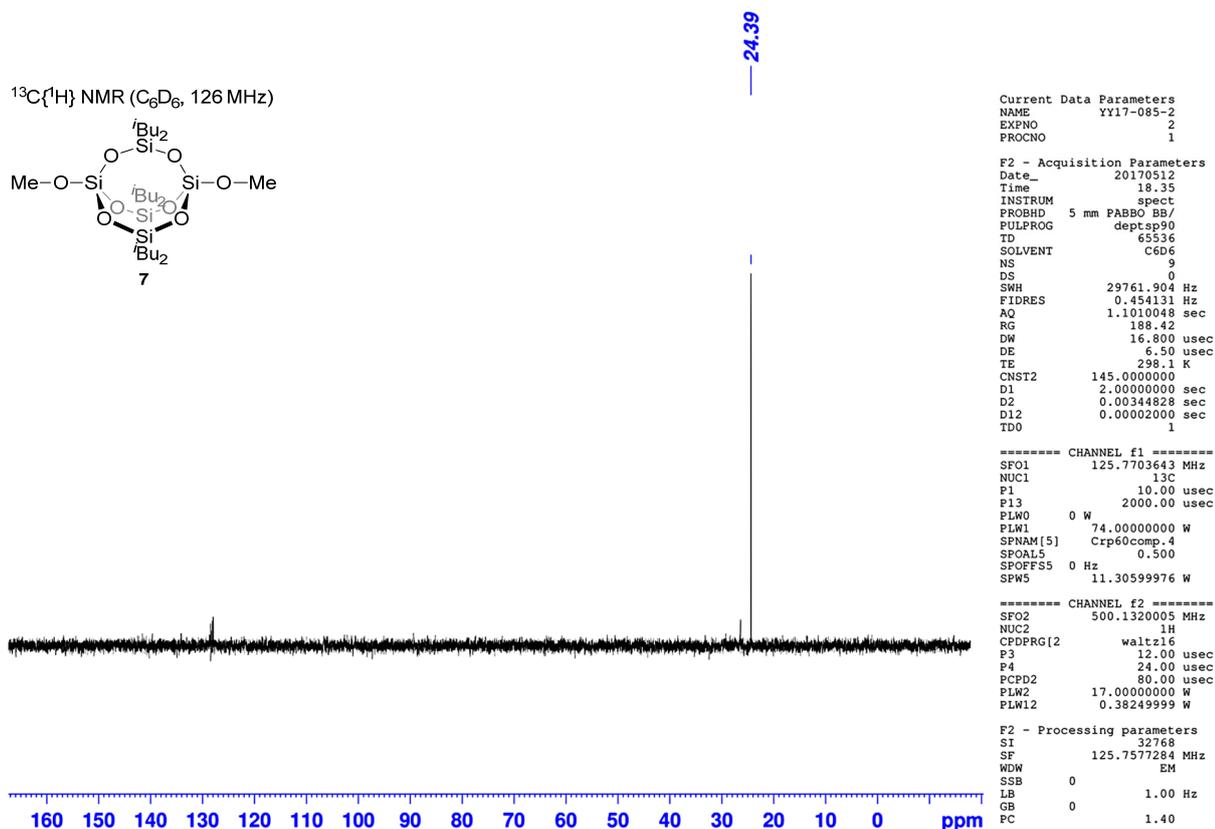


Figure S23. $^{13}\text{C}\{^1\text{H}\}$ NMR (dept90) spectrum of **7** in C_6D_6 (126 MHz, 298 K).

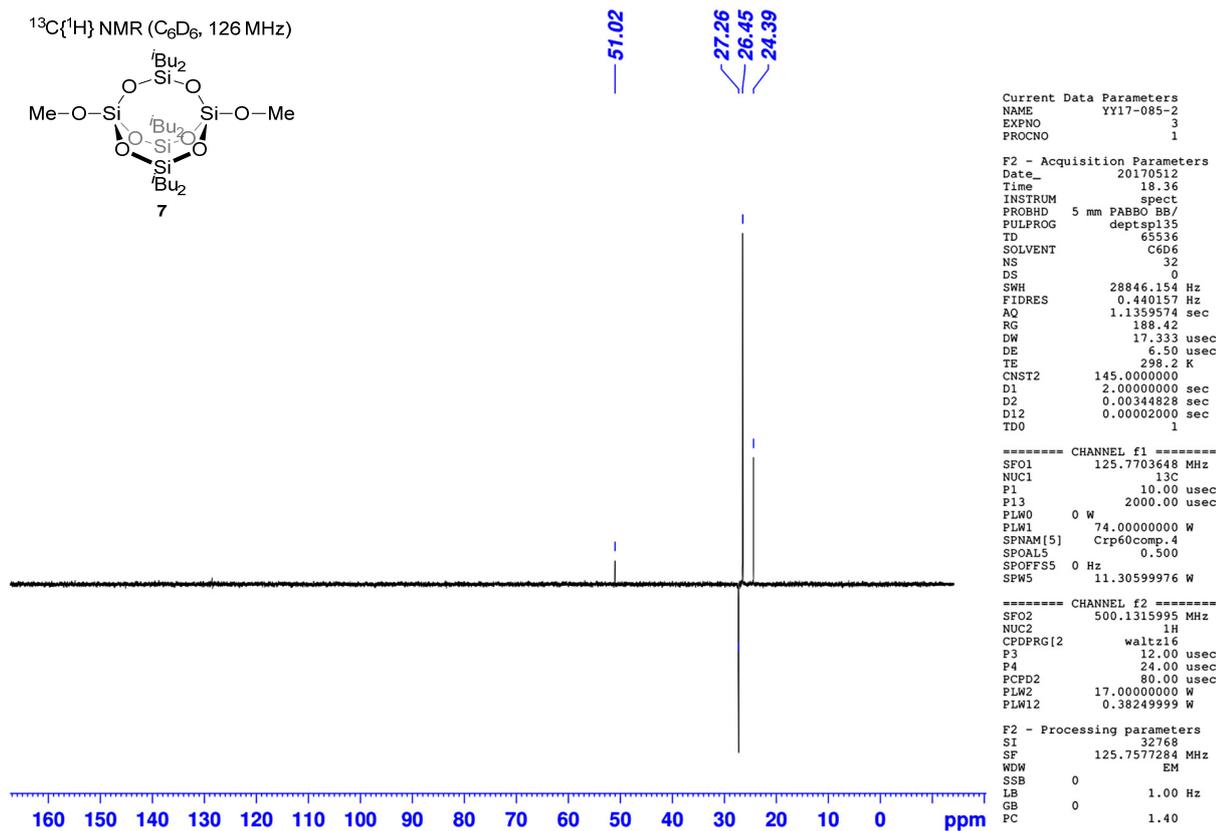


Figure S24. ¹³C{¹H} NMR (dept135) spectrum of **7** in C₆D₆ (126 MHz, 298 K).

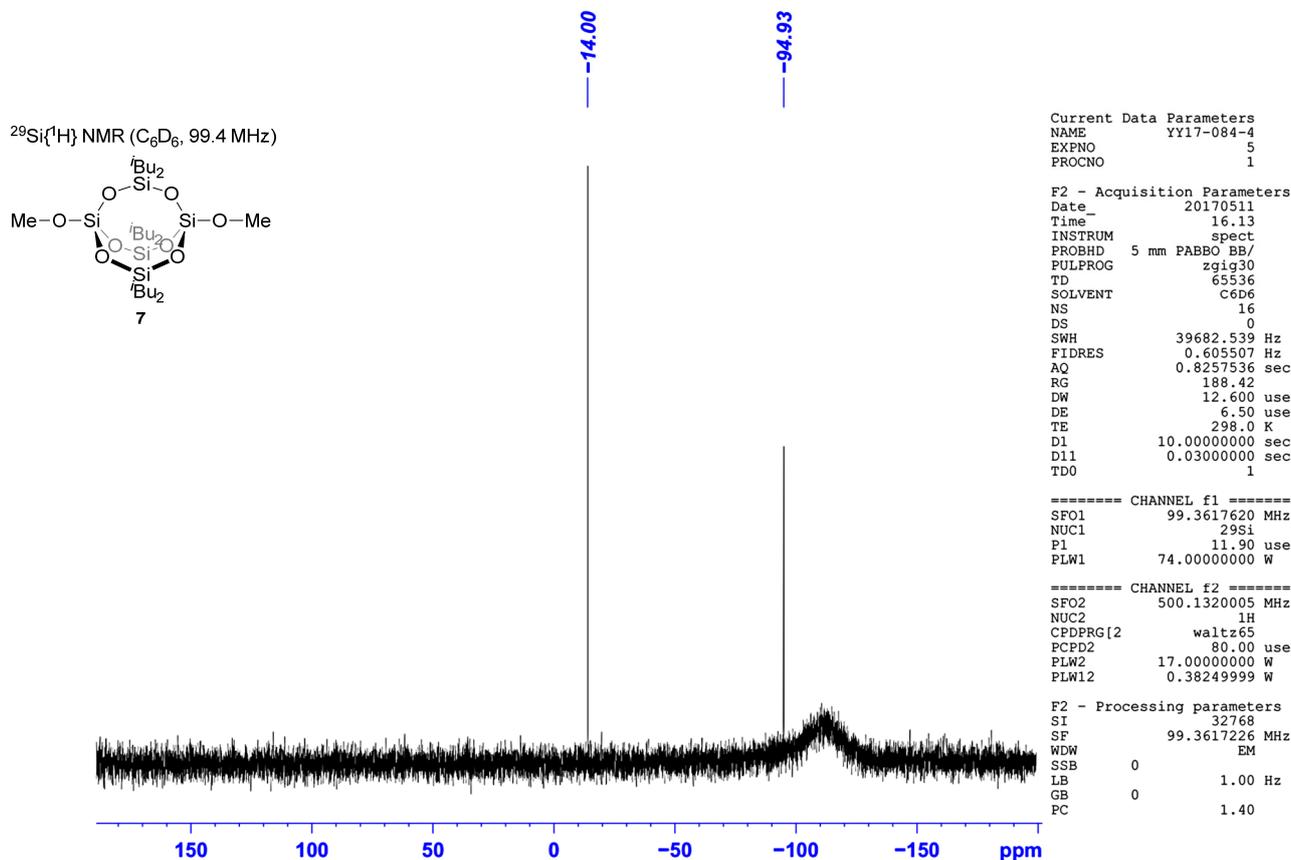
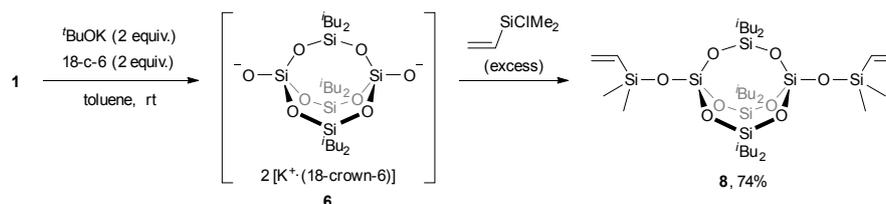


Figure S25. ²⁹Si{¹H} NMR spectrum of **7** in C₆D₆ (99.4 MHz, 298 K).

Synthesis of **8**



In a Schlenk tube (50 mL) equipped with a magnetic stir bar, compound **1** (50.0 mg, 0.0660 mmol), 18-crown-6-ether (38.4 mg, 0.145 mmol), ^tBuOK (16.3 mg, 0.145 mmol) and toluene (3 mL) were charged. The reaction mixture was stirred for 1.0 h at room temperature and then chlorodimethylvinylsilane (39.1 mg, 0.324 mmol) was added to the mixture. The resulting insoluble materials were filtered off through a silica gel. After removal of the solvent from the filtrate in *vacuo*, the resulting oil was subjected to HPLC (ODS, EtOH as an eluent, room temperature) to afford **8** (38.2 mg, 0.0489 mmol, 74%) as a colorless oil.

8: a colorless oil; ¹H NMR (500 MHz, C₆D₆, 298 K) δ 0.30 (s, 12H, Si(CH₃)₂(CH=CH₂)), 0.90 (d, *J* = 7.0 Hz, 12H, CH₂CH(CH₃)₂), 1.12 (d, *J* = 6.5 Hz, 36H, CH₂CH(CH₃)₂), 2.02 – 2.10 (m, 6H, CH₂CH(CH₃)₂), 5.88 (dd, *J* = 4.0, 20.5 Hz, 2H, SiMe₂(CH=CH₂)), 5.96 (dd, *J* = 4.0, 15.0 Hz, 2H, SiMe₂(CH=CH₂)), 6.24 (dd, *J* = 15.0, 20.5 Hz, 2H, SiMe₂(CH=CH₂)); ¹³C{¹H} NMR (126 MHz, C₆D₆, 298 K) δ 0.2 (Si(CH₃)₂(HC=CH₂)), 24.4 (CH₂CH(CH₃)₂), 26.5 (CH₂CH(CH₃)₂), 27.3 (CH₂CH(CH₃)₂), 132.6 (SiMe₂(CH=CH₂)), 138.7 (SiMe₂(CH=CH₂)); ²⁹Si{¹H} NMR (99.4 MHz, C₆D₆, 298 K) δ -101.8 (Si^Q), -15.0 (Si^D), -0.4 (Si^M); MS (EI, 70 eV) *m/z* (%): 766 (12) [M⁺+H-Me], 724 (100) [M⁺+H-(*i*-Bu)], 667 (67) [M⁺+H-2(*i*-Bu)]. Elemental analysis calcd for C₃₂H₇₂O₈Si₇: C, 49.18; H, 9.29; found: C, 49.24; H, 9.55.

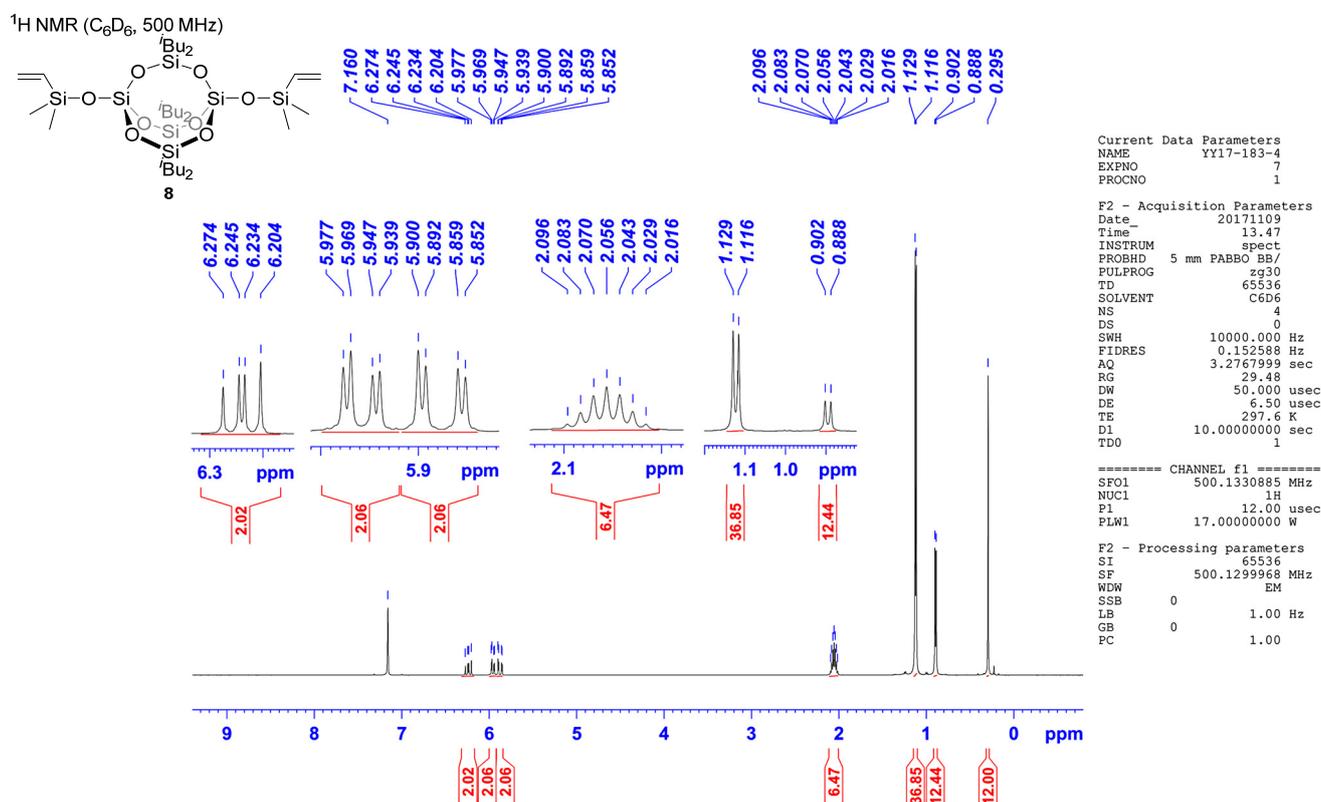


Figure S26. ¹H NMR spectrum of **8** in C₆D₆ (126 MHz, 298 K).

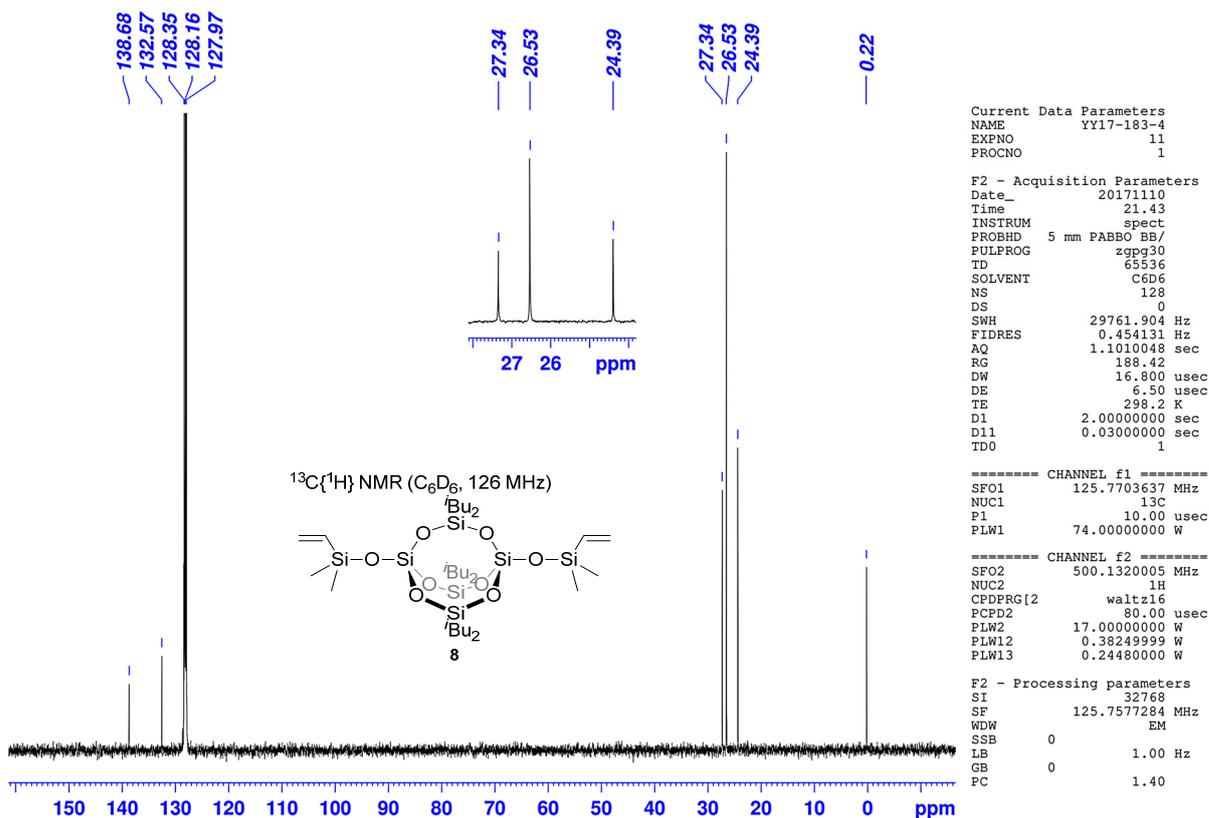


Figure S27. ¹³C{¹H} NMR spectrum of **8** in C₆D₆ (126 MHz, 298 K).

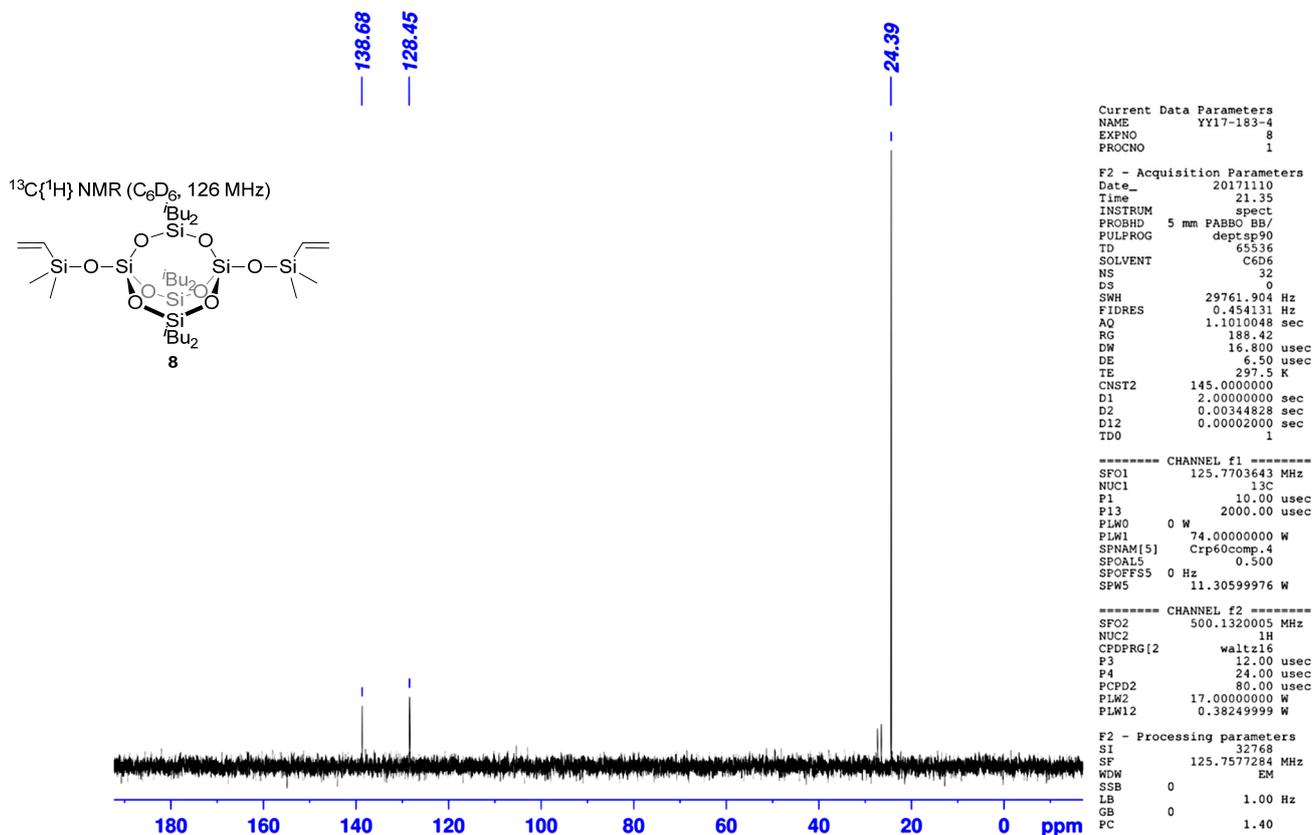


Figure S28. ¹³C{¹H} NMR (dept90) spectrum of **8** in C₆D₆ (126 MHz, 298 K).

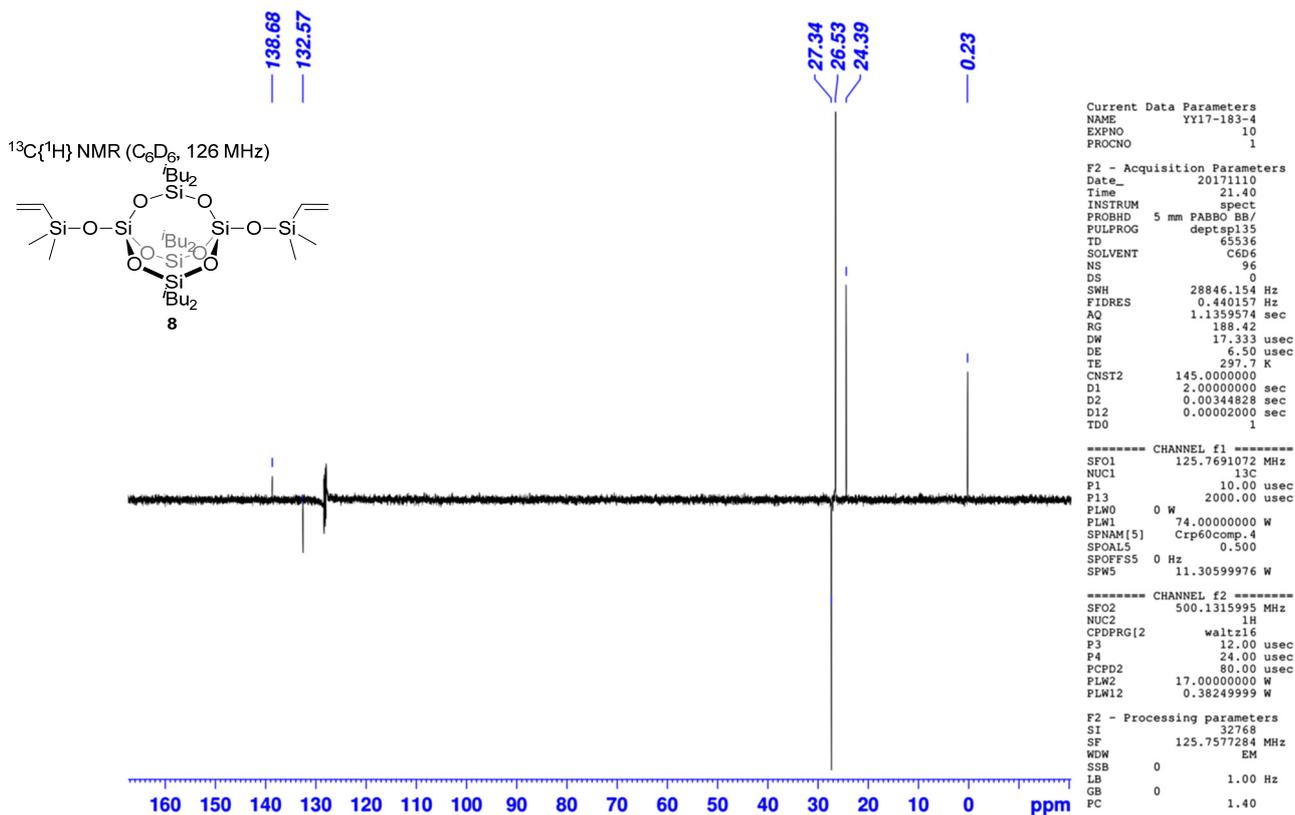


Figure S29. ¹³C{¹H} NMR (dept135) spectrum of **8** in C₆D₆ (126 MHz, 298 K).

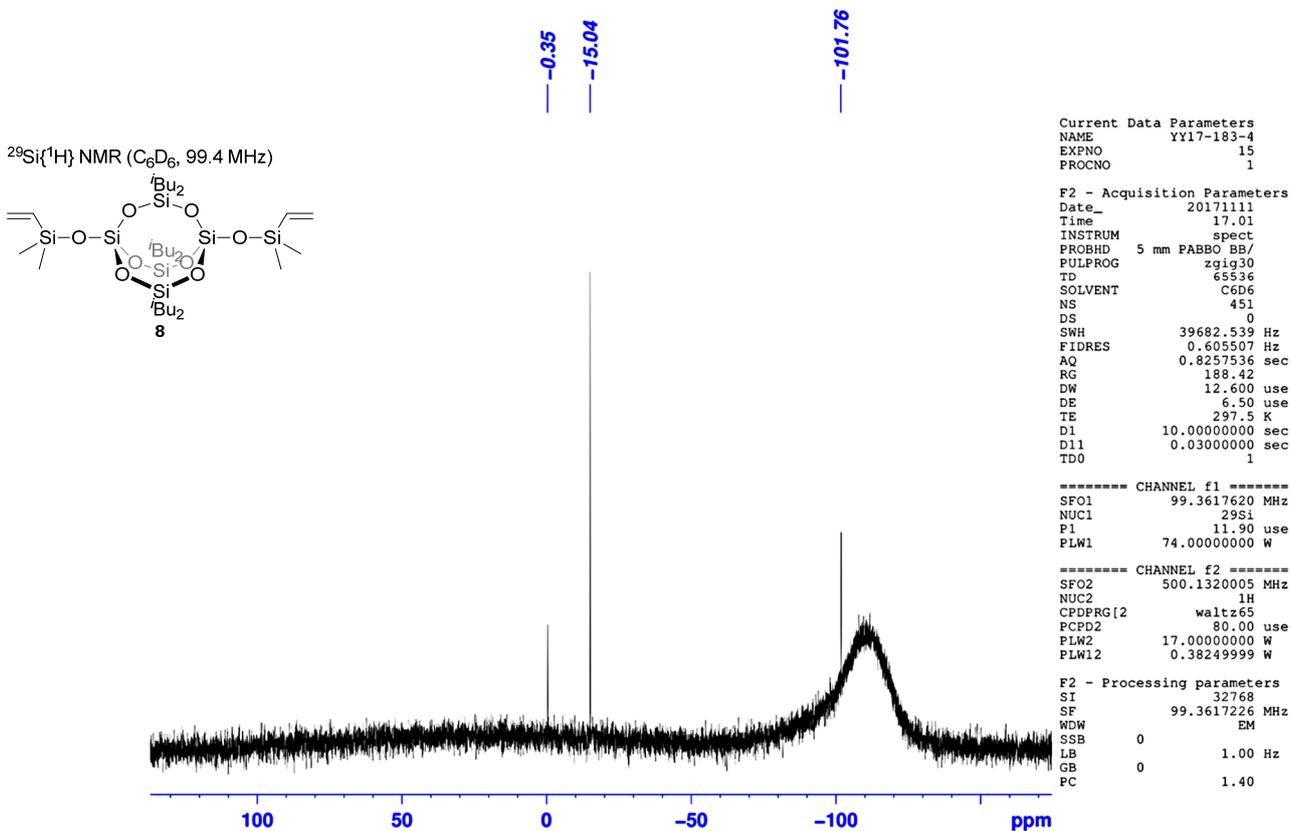


Figure S30. ²⁹Si{¹H} NMR spectrum of **8** in C₆D₆ (99.4 MHz, 298 K).

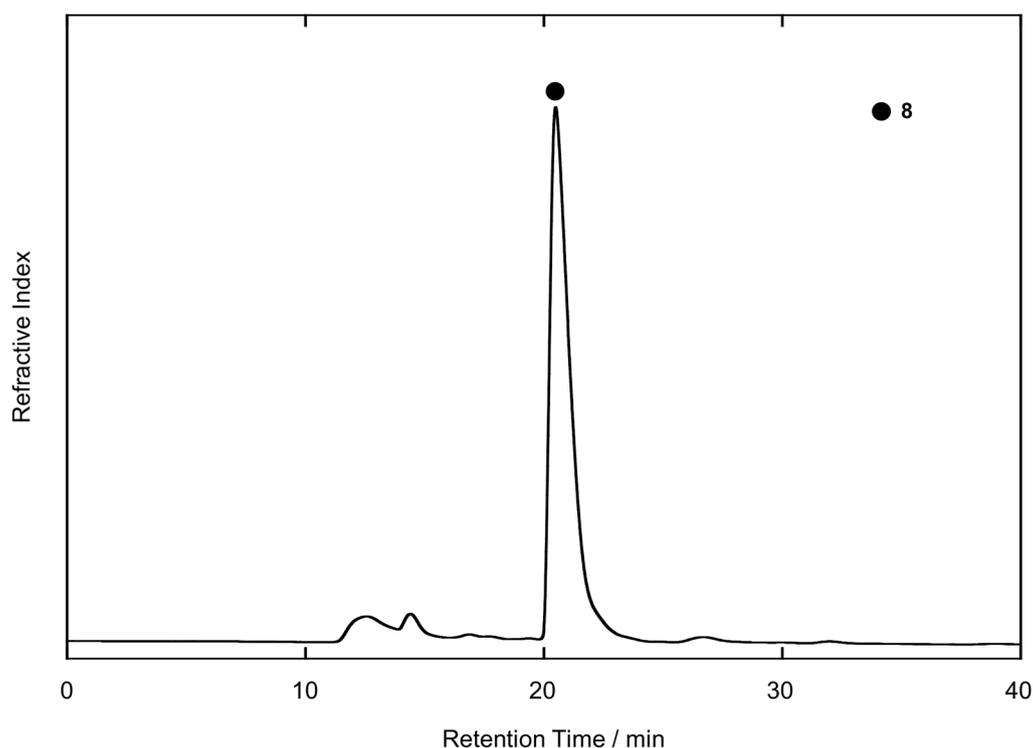
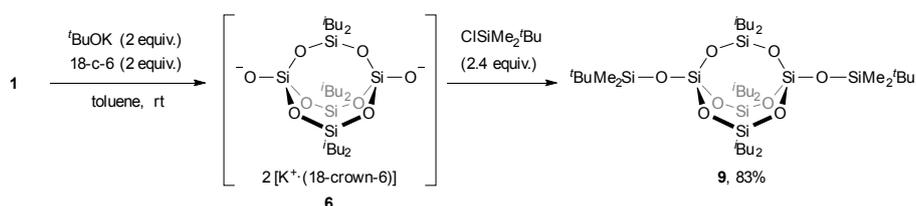


Figure S31. HPLC chart recorded during purification of **8**.

Synthesis of **9**



In a Schlenk tube (50 mL) equipped with a magnetic stir bar, **1** (50.0 mg, 0.0660 mmol), 18-c-6 (36.7 mg, 0.139 mmol), ^tBuOK (15.6 mg, 0.139 mmol), and toluene (3 mL) were charged. The reaction mixture was stirred for 1.5 h at room temperature and then *tert*-butylchlorodimethylsilane (24.1 mg, 0.161 mmol) was added to the mixture. After the volatiles were removed in *vacuo*, hexane (10 mL) was added to the resulting solid. The resulting insoluble materials were filtered off through a silica gel. The resulting colorless powder was subjected to HPLC (ODS, EtOH as an eluent, room temperature) to afford **9** (42.1 mg, 0.0500 mmol, 83%) as a white powder.

9: a white powder; mp 120-121 °C; ¹H NMR (500 MHz, C₆D₆, 298 K) δ 0.21 (s, 12H, Si(CH₃)₂^tBu), 0.90 (d, *J* = 7.5 Hz, 12H, CH₂CH(CH₃)₂), 1.02 (s, 18H, C(CH₃)₃), 1.14 (d, *J* = 6.5 Hz, 36H, CH₂CH(CH₃)₂), 2.03 – 2.11 (m, 6H, CH₂CH(CH₃)₂); ¹³C{¹H} NMR (126 MHz, C₆D₆, 297 K) δ -3.1 (Si(CH₃)₂^tBu), 18.4 (C(CH₃)₃), 24.4 (CH₂CH(CH₃)₂), 25.8 (C(CH₃)₃), 26.5 (CH₂CH(CH₃)₂), 27.3 (CH₂CH(CH₃)₂); ²⁹Si{¹H} NMR (99.4 MHz, C₆D₆, 296 K) δ -102.2 (*Si*^Q), -15.3 (*Si*^D), 13.8 (*Si*^M); MS (EI, 70 eV) *m/z* (%) 826 (2) [^M+H-CH₃], 784 (100) [^M+H-C₄H₉]; Elemental analysis calcd for C₃₆H₈₄O₈Si₇: C, 51.37; H, 10.06; found: C, 51.58; H, 10.19.

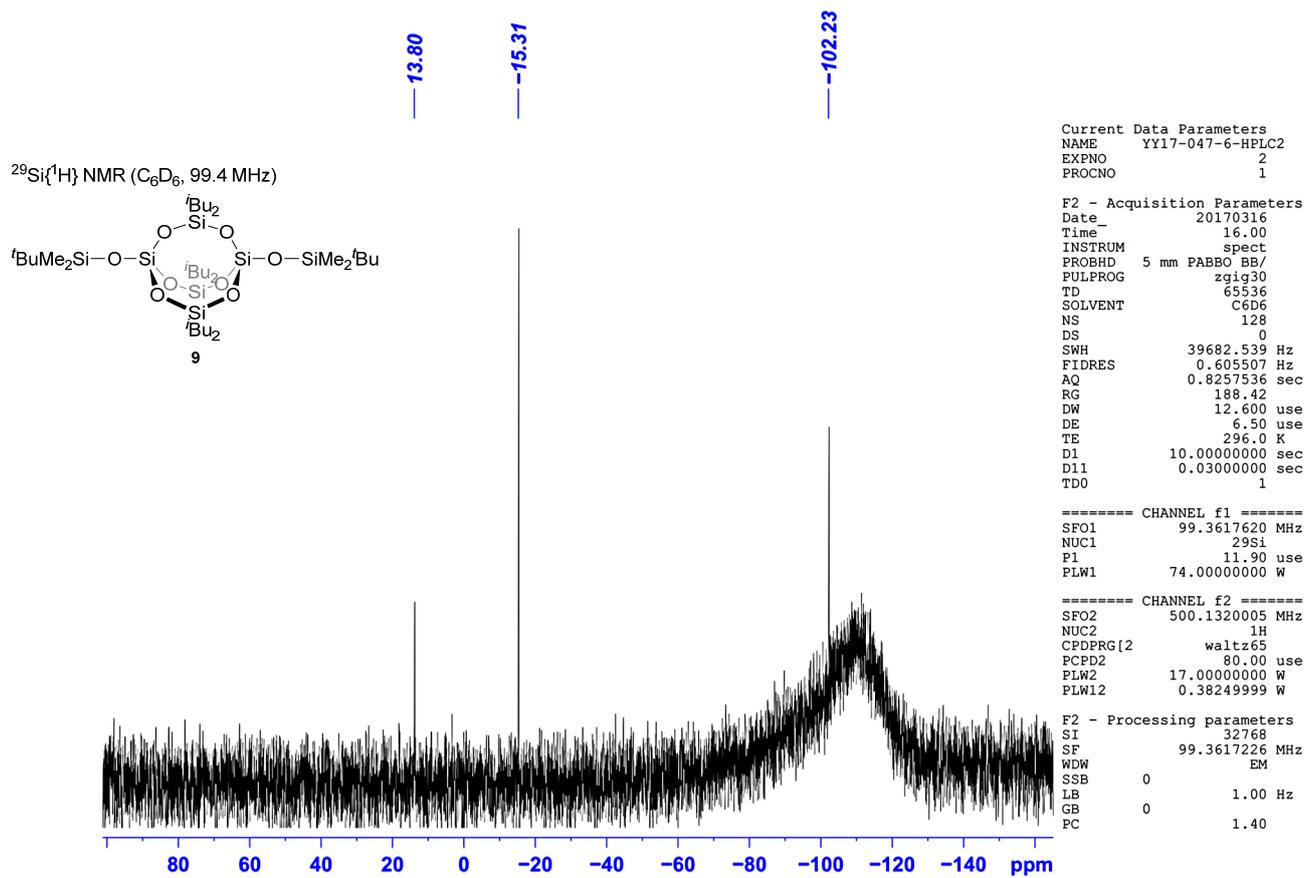


Figure S36. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **9** in C_6D_6 (99.4 MHz, 296 K).

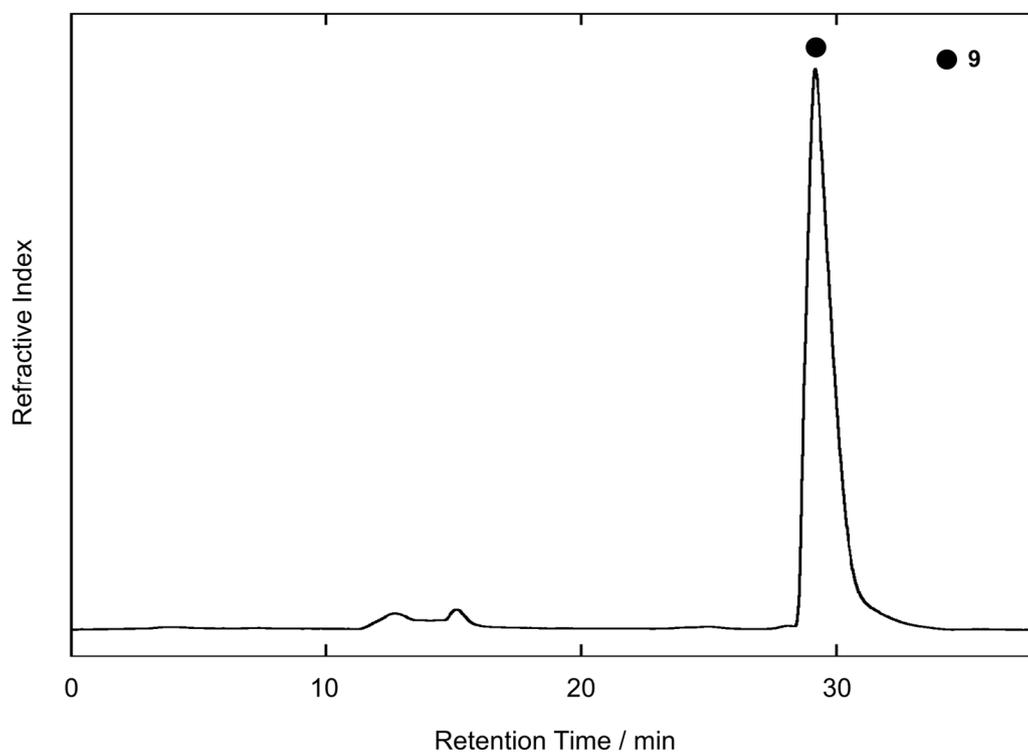


Figure S37. HPLC chart of **9**.

2. Single crystal X-ray diffraction analyses of **1** and **9**

Single crystals suitable for X-ray diffraction study were obtained by recrystallization from EtOH at room temperature for **1** and **9**. The single crystals for data collection coated by Apiezon® grease was mounted on a glass fiber and then transferred to the cold nitrogen gas stream of the diffractometer. X-ray diffraction data were collected on a Bruker AXS APEX II CCD diffractometer with graphite monochromated Mo-K α radiation. An empirical absorption correction based on the multiple measurement of equivalent reflections was applied using the program SADABS^{S3} and the structures were solved by direct methods and refined by full-matrix least squares against F^2 using all data (SHELEX-2014).^{S4} Molecular structures were analyzed by Yadokari-XG^{S5} software.

Compound **9** was measured at different temperatures (100 K, 173 K, 223 K) using the same single crystal in the order 100 K \rightarrow 223 K \rightarrow 173 K.

Crystal data for **1** (CCDC-1576828) (100 K): 0.20 mm \times 0.20 mm \times 0.10 mm; C₃₀H₇₂O₈Si₇; Formula weight 757.50; tetragonal; space group $I4_1/a$; $a = 29.5020(5)$ Å, $b = 29.5020(5)$ Å, $c = 10.4338(2)$ Å, $V = 9081.2(4)$ Å³, $Z = 8$, $D_{\text{calcd}} = 1.108$ Mg m⁻³, 58962 reflections measured, 4457 unique ($R_{\text{int}} = 0.0306$), which were used in all calculations; $R1 = 0.0810$ ($I > 2\sigma(I)$), $wR2 = 0.2110$ (all data), GOF = 1.077, max/min residual electron densities 0.573/−0.308 e/Å³.

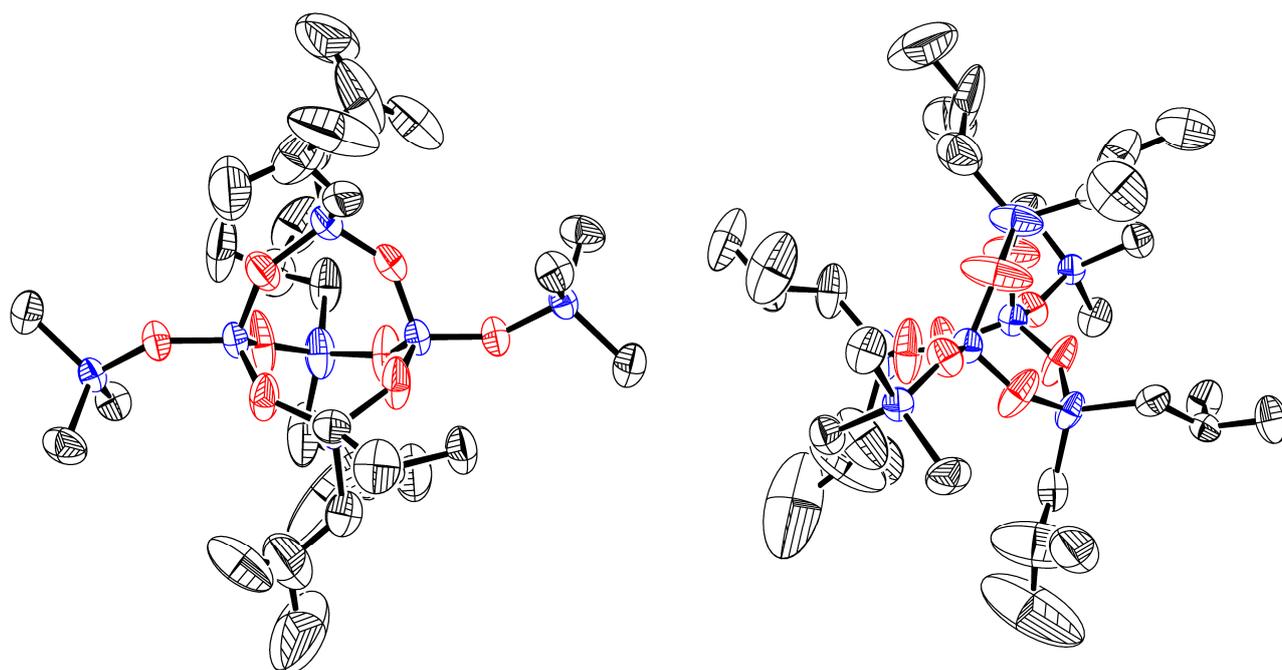


Figure S38. ORTEP drawings of **1** (atomic displacement parameters are set at the 50% possibility; hydrogen atoms are omitted for clarity).

Crystal data for **9** (CCDC-1576829) (100 K): 0.30 mm \times 0.30 mm \times 0.30 mm; C₃₆H₈₄O₈Si₇; Formula weight 841.66; monoclinic; space group $C2/c$; $a = 18.2557(12)$ Å, $b = 13.4289(9)$ Å, $c = 21.0576(13)$ Å, $\beta = 104.5550(10)^\circ$, $V = 4996.7(6)$ Å³, $Z = 4$, $D_{\text{calcd}} = 1.119$ Mg m⁻³, 36402 reflections measured, 5751 unique ($R_{\text{int}} = 0.0205$), which were used in all calculations; $R1 = 0.0297$ ($I > 2\sigma(I)$), $wR2 = 0.0790$ (all data), GOF = 1.064, max/min residual electron densities 0.514/−0.437 e/Å³.

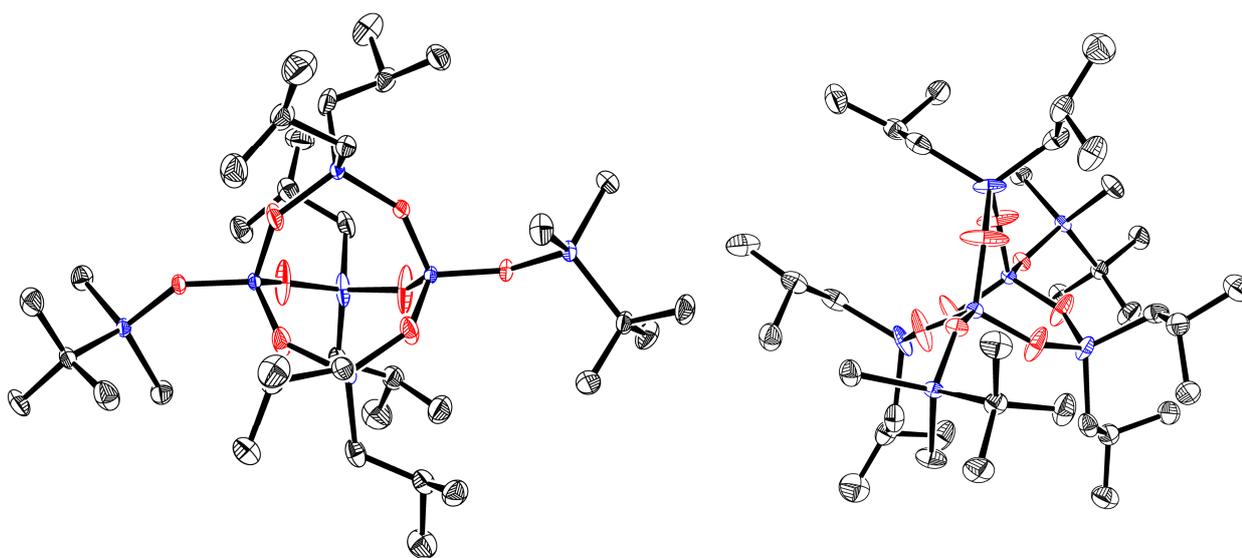


Figure S39. ORTEP drawings of **9** (atomic displacement parameters are set at the 50% possibility; hydrogen atoms are omitted for clarity). The sample was measured at 100 K.

Crystal data for **9** (CCDC: 1576899) (173 K): 0.30 mm × 0.30 mm × 0.30 mm; C₃₆H₈₄O₈Si₇; Formula weight 841.66; monoclinic; space group *C2/c*; *a* = 18.4236(4) Å, *b* = 13.5947(3) Å, *c* = 21.1068(5) Å, β = 104.2450(10)°, *V* = 5123.9(2) Å³, *Z* = 4, *D*_{calcd} = 1.091 Mg m⁻³, 30476 reflections measured, 4518 unique (*R*_{int} = 0.0221), which were used in all calculations; *R*1 = 0.0537 (*I* > 2σ(*I*)), *wR*2 = 0.1513 (all data), GOF = 1.039, max/min residual electron densities 1.446/− 0.516 e/Å³. The observed large residual density (Q peak, 1.45 e/Å³) found near silicon atoms of the bridgehead siloxy group (alert B after CIFcheck) should be due to a disordered silicon atom as the similar disordered silicon atom was found at the similar positions in the same single crystal measured at 223 K. While the silicon atoms were modeled at two positions with the ratio of 92:8, the corresponding carbon atoms with a minor site occupancy factor were not able to be modeled properly.

Crystal data for **9** (CCDC-1576830) (223 K): 0.30 mm × 0.30 mm × 0.30 mm; C₃₆H₈₄O₈Si₇; Formula weight 841.66; monoclinic; space group *C2/c*; *a* = 19.0349(5) Å, *b* = 13.7423(3) Å, *c* = 21.0422(5) Å, β = 103.8890(10)°, *V* = 5343.4(2) Å³, *Z* = 8, *D*_{calcd} = 1.046 Mg m⁻³, 32950 reflections measured, 4920 unique (*R*_{int} = 0.0308), which were used in all calculations; *R*1 = 0.0703 (*I* > 2σ(*I*)), *wR*2 = 0.2335 (all data), GOF = 1.061, max/min residual electron densities 0.453/− 0.390 e/Å³.

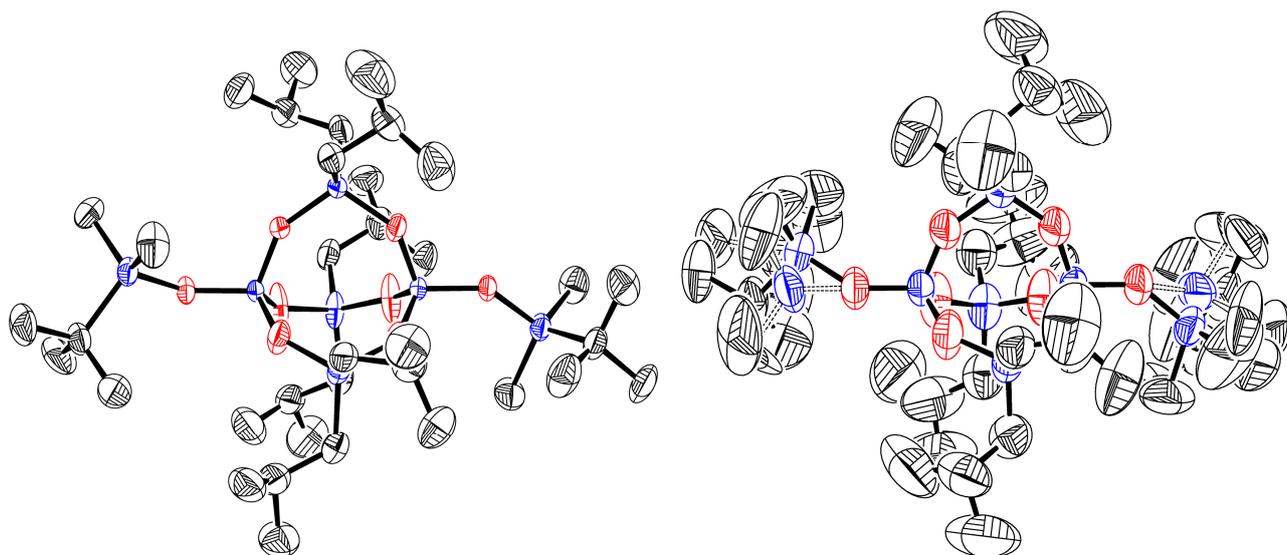


Figure S40. ORTEP drawings of **9** measured at 173 K (left) and at 223 K (right). Atomic displacement parameters are set at the 50% possibility. Hydrogen atoms are omitted for clarity.

3. Thermogravimetric Analyses of 1

Thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC) of **1** were examined under dry nitrogen gas flow (100 mL min^{-1}) at the temperature range of $25\text{--}400 \text{ }^\circ\text{C}$ with heating rates of 0.1 K min^{-1} , 1.0 K min^{-1} or 10 K min^{-1} for **1**. The samples ($2\text{--}4 \text{ mg}$) were loaded in an open aluminum pan for each measurement.

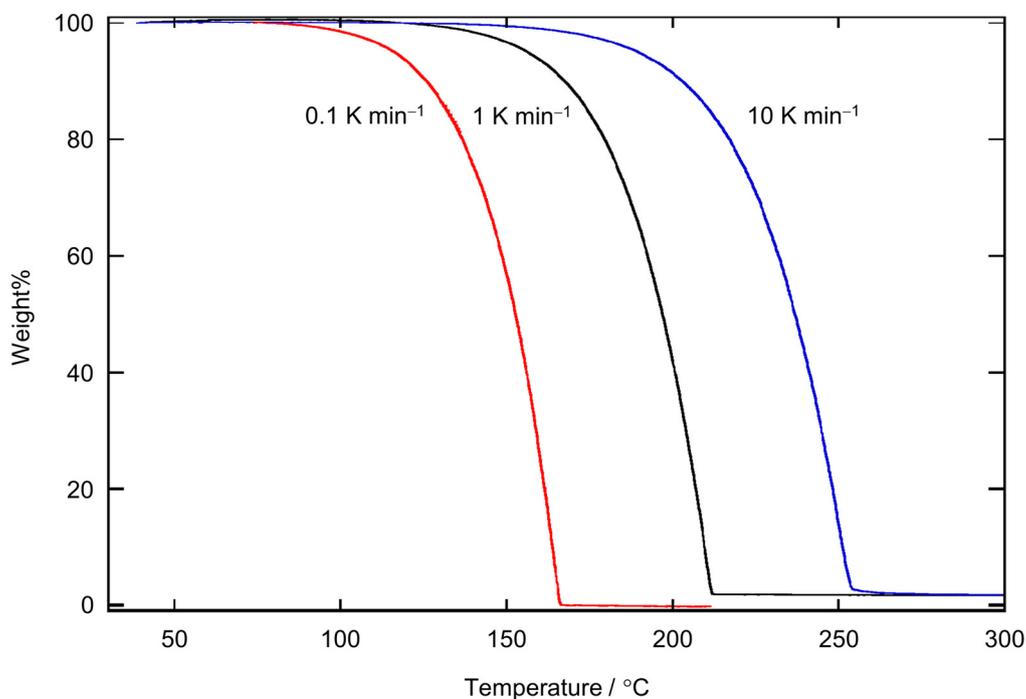


Figure S41. TGA curves of **1** with variable heating rates.

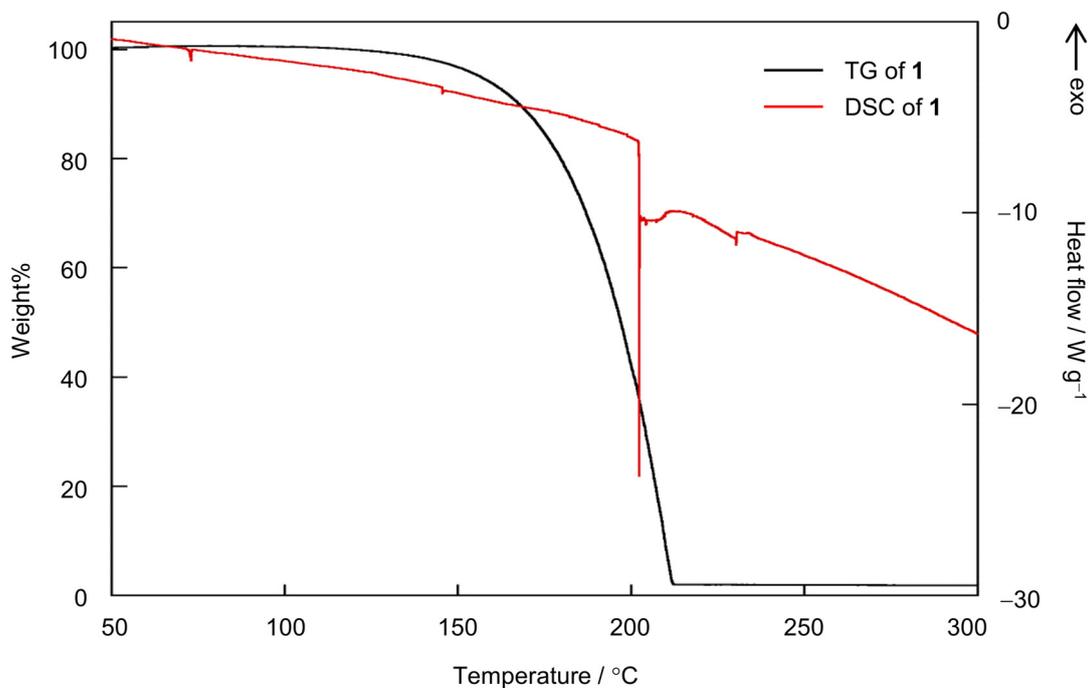
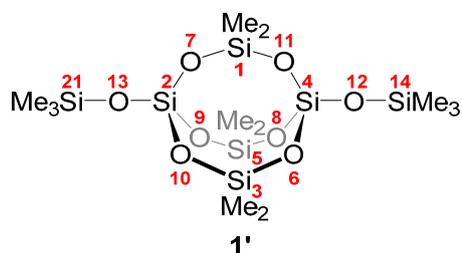


Figure S42. TGA and DSC profiles of **1** with heating rate of 1.0 K min^{-1} .

4. Theoretical calculation

All theoretical calculations were performed using Gaussian 09^{S6} and NBO programs.^{S7} Geometry optimization and NBO analysis of **1'** was performed at the B3PW91/6-31G(d) level of theory. Atomic coordinates and energies of **1'** were summarized in Tables S1. Natural population analysis was summarized in Table S2.



TableS1. Atomic Coordinates of **1'** Optimized at the B3PW91/6-31G(d) Level of Theory

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	14	0	-0.276557	-1.017511	2.123009
2	14	0	1.851598	0.216733	0.218741
3	14	0	0.000650	2.681145	0.001860
4	14	0	-1.852346	0.217655	-0.217979
5	14	0	0.276595	-1.014545	-2.123921
6	8	0	-1.340374	1.716778	0.207274
7	8	0	1.094774	-0.237053	1.602373
8	8	0	-1.096168	-0.238188	-1.601241
9	8	0	1.477391	-0.845811	-0.980881
10	8	0	1.340335	1.715730	-0.208072
11	8	0	-1.479042	-0.844045	0.982600
12	8	0	-3.449425	0.244823	-0.465009
13	8	0	3.448662	0.242954	0.465985
14	14	0	-4.966613	-0.358092	-0.108214
15	6	0	0.809480	-0.208172	-3.722365
16	6	0	-0.065274	-2.841402	-2.316039
17	6	0	0.067167	-2.844979	2.305948
18	6	0	-0.807922	-0.219171	3.725953
19	6	0	-0.195188	3.717332	-1.539596
20	6	0	0.197986	3.710466	1.547741
21	14	0	4.965932	-0.358615	0.107151
22	1	0	0.011570	-0.255122	-4.472143
23	1	0	1.059124	0.845961	-3.559944
24	1	0	1.694580	-0.702843	-4.138598
25	1	0	-0.833136	-3.018321	-3.077822
26	1	0	0.838583	-3.384819	-2.614187
27	1	0	-0.422886	-3.271278	-1.374043
28	1	0	-0.835963	-3.390690	2.602102
29	1	0	0.424438	-3.269819	1.361543
30	1	0	0.835790	-3.024957	3.066248
31	1	0	-1.693015	-0.715519	4.140187
32	1	0	-0.009539	-0.270458	4.474943
33	1	0	-1.057081	0.835940	3.569232
34	1	0	0.697986	4.325212	-1.723388
35	1	0	-0.358809	3.081440	-2.416412
36	1	0	-1.053725	4.392774	-1.451137
37	1	0	1.056992	4.385649	1.461873
38	1	0	-0.694654	4.318197	1.734592
39	1	0	0.361518	3.070577	2.421659
40	6	0	-5.404656	0.102665	1.661728
41	1	0	-4.697824	-0.343216	2.370840
42	1	0	-5.380441	1.189066	1.804862

43	1	0	-6.409811	-0.247829	1.926459
44	6	0	-4.950639	-2.227342	-0.315818
45	1	0	-4.691267	-2.510292	-1.342533
46	1	0	-4.215302	-2.688308	0.353577
47	1	0	-5.931612	-2.661526	-0.086961
48	6	0	-6.153972	0.444005	-1.322590
49	1	0	-7.183258	0.103110	-1.157451
50	1	0	-6.141499	1.535241	-1.221874
51	1	0	-5.881811	0.202039	-2.356261
52	6	0	5.402502	0.104866	-1.662445
53	1	0	4.695554	-0.340515	-2.371756
54	1	0	5.377517	1.191442	-1.804098
55	1	0	6.407720	-0.244634	-1.928254
56	6	0	6.153730	0.442398	1.321807
57	1	0	7.183062	0.102202	1.155515
58	1	0	6.140689	1.533762	1.222578
59	1	0	5.882378	0.198905	2.355332
60	6	0	4.951012	-2.228153	0.312138
61	1	0	4.692684	-2.512635	1.338694
62	1	0	4.215263	-2.688507	-0.357224
63	1	0	5.931967	-2.661592	0.081807

E(RB3PW91) = -3107.41908612

Zero-point correction =

0.492760 (Hartree/Particle)

Thermal correction to Gibbs Free Energy (298.15 K) =

0.409430

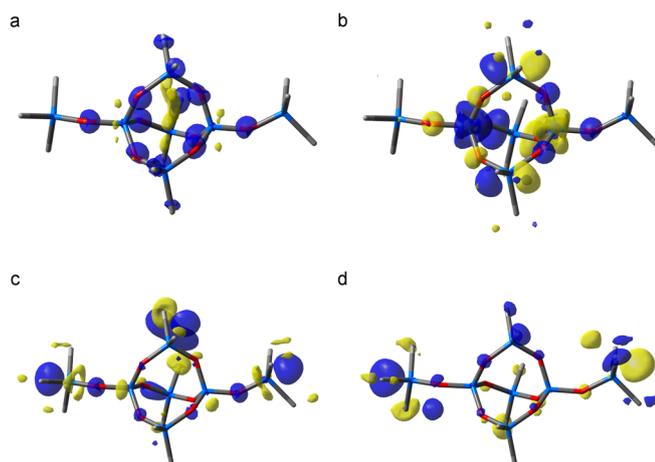


Figure S43. Frontier Kohn-Sham orbitals of **1'**; LUMO (a), LUMO+1 (b), LUMO+2 (c), LUMO+3 (d).

TableS2. Summary of Natural Population Analysis of **1'** Optimized at the B3PW91/6-31G(d) Level of Theory

Atom	No	Natural Charge	Core	Valence	Rydberg	Total
Si	1	2.20587	9.99757	1.74261	0.05395	11.79413
Si	2	2.50346	9.99657	1.43041	0.06956	11.49654
Si	3	2.20713	9.99756	1.74172	0.05359	11.79287
Si	4	2.50358	9.99657	1.43031	0.06954	11.49642
Si	5	2.20587	9.99757	1.74262	0.05395	11.79413
O	6	-1.26561	1.99982	7.25464	0.01116	9.26561
O	7	-1.26740	1.99981	7.25656	0.01103	9.26740
O	8	-1.26748	1.99981	7.25663	0.01103	9.26748
O	9	-1.26762	1.99982	7.25690	0.01090	9.26762
O	10	-1.26558	1.99982	7.25460	0.01116	9.26558
O	11	-1.26768	1.99982	7.25696	0.01090	9.26768

O	12	-1.26017	1.99981	7.25007	0.01030	9.26017
O	13	-1.26018	1.99981	7.25007	0.01030	9.26018
Si	14	2.02797	9.99776	1.93094	0.04334	11.97203
C	15	-1.26381	1.99947	5.25545	0.00890	7.26381
C	16	-1.26318	1.99947	5.25485	0.00886	7.26318
C	17	-1.26322	1.99947	5.25489	0.00887	7.26322
C	18	-1.26376	1.99947	5.25540	0.00889	7.26376
C	19	-1.26356	1.99947	5.25525	0.00884	7.26356
C	20	-1.26360	1.99947	5.25529	0.00884	7.26360
Si	21	2.02795	9.99776	1.93095	0.04334	11.97205
H	22	0.26571	0.00000	0.73372	0.00057	0.73429
H	23	0.26942	0.00000	0.72974	0.00083	0.73058
H	24	0.26243	0.00000	0.73699	0.00059	0.73757
H	25	0.26429	0.00000	0.73513	0.00058	0.73571
H	26	0.26387	0.00000	0.73554	0.00059	0.73613
H	27	0.26762	0.00000	0.73155	0.00083	0.73238
H	28	0.26388	0.00000	0.73554	0.00059	0.73612
H	29	0.26765	0.00000	0.73152	0.00083	0.73235
H	30	0.26427	0.00000	0.73515	0.00058	0.73573
H	31	0.26245	0.00000	0.73696	0.00059	0.73755
H	32	0.26570	0.00000	0.73373	0.00057	0.73430
H	33	0.26939	0.00000	0.72978	0.00083	0.73061
H	34	0.26461	0.00000	0.73481	0.00058	0.73539
H	35	0.26864	0.00000	0.73053	0.00082	0.73136
H	36	0.26462	0.00000	0.73480	0.00058	0.73538
H	37	0.26460	0.00000	0.73482	0.00058	0.73540
H	38	0.26462	0.00000	0.73480	0.00058	0.73538
H	39	0.26868	0.00000	0.73049	0.00083	0.73132
C	40	-1.24702	1.99950	5.23894	0.00859	7.24702
H	41	0.26315	0.00000	0.73598	0.00087	0.73685
H	42	0.26381	0.00000	0.73544	0.00075	0.73619
H	43	0.25848	0.00000	0.74094	0.00058	0.74152
C	44	-1.24807	1.99950	5.24000	0.00857	7.24807
H	45	0.26250	0.00000	0.73674	0.00076	0.73750
H	46	0.26532	0.00000	0.73378	0.00090	0.73468
H	47	0.25864	0.00000	0.74078	0.00057	0.74136
C	48	-1.24379	1.99949	5.23593	0.00837	7.24379
H	49	0.25783	0.00000	0.74158	0.00058	0.74217
H	50	0.26428	0.00000	0.73497	0.00076	0.73572
H	51	0.26411	0.00000	0.73515	0.00075	0.73589
C	52	-1.24702	1.99950	5.23894	0.00859	7.24702
H	53	0.26313	0.00000	0.73600	0.00087	0.73687
H	54	0.26382	0.00000	0.73542	0.00075	0.73618
H	55	0.25850	0.00000	0.74092	0.00058	0.74150
C	56	-1.24380	1.99949	5.23593	0.00837	7.24380
H	57	0.25784	0.00000	0.74157	0.00058	0.74216
H	58	0.26428	0.00000	0.73496	0.00076	0.73572
H	59	0.26412	0.00000	0.73514	0.00075	0.73588
C	60	-1.24807	1.99950	5.24000	0.00857	7.24807
H	61	0.26250	0.00000	0.73674	0.00076	0.73750
H	62	0.26534	0.00000	0.73376	0.00090	0.73466
H	63	0.25865	0.00000	0.74078	0.00057	0.74135

* Total * 0.00000 109.97365 159.42308 0.60327 270.00000

5. References

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