

Electronic Supplementary Information (ESI)

Nucleophilic Fluorination Facilitated by a CsF-CaF₂ Packed Bed Reactor in Continuous Flow

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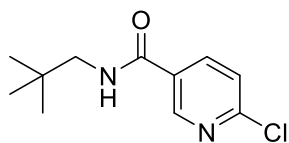
Table of Contents

General considerations	SI2
Synthesis of starting material	SI2
Preparation of CsF supported on CaF ₂ ⁸	SI2
Preparation of KF supported on CaF ₂ ⁸	SI3
Packed bed reactor	SI3
Determination of internal volume	SI3
General experimental Setup for Continuous Flow	SI4
In-line Drying of CsF-CaF ₂ Reactor	SI4
Influence of CsF Loading on CaF ₂ in Nucleophilic Fluoride Substitution of 1-bromomethyl-4- tertbutylbenzene in Continuous Flow.....	SI5
Procedure for Nucleophilic Fluorination of Benzyl Bromides: Substrate Scope and Isolated Yields in Flow	SI8
Procedure for Nucleophilic (hetero)aromatic Substitutions: Substrate Scope and Isolated Yields in Flow.	SI10
Procedure for Deprotection of Silyl Ethers: Substrate Scope and Isolated Yields in Flow.	SI12
Procedure for CsF-mediated Trifluoromethylation: Substrate Scope and Isolated Yields in Flow.....	SI13
Trifluoromethylation of 4-acetylbenzonitrile in Continuous Flow - Scale-out experiment.....	SI15
References	SI16
NMR Spectra	SI18

General considerations

All purchased chemicals were used as received without further purification. Solvents used in flow experiments were kept over 3 Å molecular sieves. CsF (98%) was purchased from Combi Blocks and synthetic CaF₂ was acquired from Solvay. Ethyl 3 α -hydroxy-5 β -cholan-24-oate was synthesized from lithocholic acid¹, propan-2-yl 2-chloro-5-nitrobenzoate was synthesized from 2-chloro-5-nitrobenzoic acid², and silyl ethers were synthesized from ethyl 3 α -hydroxy-5 β -cholan-24-oate³, vanillin⁴⁻⁵ and methyl 2-((4-chlorophenyl)(hydroxy)methyl)acrylate⁶, according to literature procedures. Flash column chromatography was carried out on silica gel 60 (230-400 mesh). The chemical shifts in ¹H NMR and ¹³C NMR are reported in ppm relative to the solvent residual peak⁷. All ¹⁹F NMR chemicals shifts are unlocked. ¹H NMR spectra were recorded at 400 MHz, ¹³C NMR spectra were recorded at 101 MHz, and ¹⁹F NMR spectra were recorded at 376 MHz on a 400 MHz spectrometer, all at room temperature on a Bruker 400 spectrometer. HRMS spectra were recorded on an LC TOF (ES) apparatus. Melting points (mp) were obtained on a capillary melting point apparatus and are uncorrected. The flow system consists of a HPLC pump (Knauer Azura P 4.1S), stainless stain tubing (1/16" OD x 0.75 mm ID, Vici), an injection valve (2 position: load and inject, 6-port, 1/16", Vici), a sample loop (2.0 mL, 453 cm length) made from stainless stain tubing (1/16" OD x 0.75 mm ID, Vici), a spring loaded back pressure regulator (BPR) cartridge (500 psi, IDEX Health & Science), an empty stainless steel HPLC column (25 cm length, 0.46 cm ID) heated by a PID regulated oven, and all connections were made using HPLC fittings (IDEX Health & Science).

Synthesis of starting material



6-Chloro-N-neopentylnicotinamide. A 1.0 M solution of 6-chloronicotinoyl chloride (4.39 g, 24.5 mmol, 1.0 equiv.) in CH₂Cl₂ was added dropwise to a suspension of neopentylamine (5.8 mL, 49.6 mmol, 2.0 equiv.) and K₂CO₃ (3.24 g, 30.6 mmol, 1.2 equiv.) in 10 mL CH₂Cl₂ at 0°C. The reaction mixture was

allowed to warm to room temperature and stirred continuously for 4 h. The crude reaction mixture was washed with sat. NaHCO₃ (3 x 30 mL), H₂O (3 x 30 mL) and brine (3 x 30 mL), dried above Na₂SO₄ and concentrated under reduced pressure. The crude product was purified using silica-plug filtration eluting with CH₂Cl₂/Et₂O 1:1, affording title product (5.23 g, 93%) as a white solid. mp 102-103°C; ¹H NMR (400 MHz, CDCl₃) δ 8.72 (d, *J* = 2.2 Hz, 1H), 8.05 (dd, *J* = 8.3 Hz, *J* = 2.4 Hz, 1H), 7.39 (d, *J* = 8.3 Hz, 1H), 6.30 (br. s, 1H), 3.26 (d, *J* = 6.3 Hz, 2H), 0.97 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 164.9, 154.2, 147.9, 138.0, 129.7, 124.5, 51.3, 32.3, 27.4; HRMS *m/z* calculated for C₁₁H₁₆ClN₂O [M + H]⁺ 227.0946, found 227.0946.

Preparation of CsF supported on CaF₂⁸

The CaF₂-supported CsF (CsF on CaF₂) was prepared by slowly evaporating to dryness a suspension of a 0.3 M solution of CsF (15.03 g, 98.9 mmol) in methanol with added CaF₂ (13.23 g) for ca. 45 min at 55°C under reduced pressure using rotary evaporator. The resulting solid was grounded gently for ca. 1 min in a mortar and dried for 2 h at 100°C under vacuum affording a white solid [3.5 mmol CsF/(g CsF on CaF₂)].

CsF was also supported on CaF₂ with a loading of 2.2 mmol CsF/(g CsF on CaF₂) using the same procedure but with less CsF.

Preparation of KF supported on CaF₂⁸

The CaF₂-supported KF (KF on CaF₂) was prepared by slowly evaporating to dryness a suspension of a 0.75 M solution of KF (6.51 g, 112 mmol) in methanol with added CaF₂ (25.00 g) for ca. 45 min at 55°C under reduced pressure using rotary evaporator. The resulting solid was grounded gently for ca. 1 min in a mortar and dried for 2 h at 100°C under vacuum affording a white solid [3.5 mmol KF/(g KF on CaF₂)].

KF was also supported on CaF₂ with a loading of 4.0 mmol KF/(g KF on CaF₂) using the same procedure but with more KF.

Packed bed reactor

An empty HPLC column (25 cm length, 0.46 cm ID) was fitted with a small ball of cotton (20 mg) in 1 of the ends. Then the reactor was filled with the supported metal fluoride inside a fume hood. For CsF on CaF₂ with a loading of 3.5 mmol CsF/(g CsF on CaF₂), 21-22 mmol of CsF was packed into the column. For CsF on CaF₂ with a loading of 2.2 mmol CsF/(g CsF on CaF₂), 13-14 mmol of CsF was packed into the column. For KF on CaF₂ with a loading of 3.5 mmol CsF/(g KF on CaF₂), 18-19 mmol of KF was packed into the column. For KF on CaF₂ with a loading of 4.0 mmol CsF/(g KF on CaF₂), 22 mmol of KF was packed into the column. *Unless otherwise stated, the packed bed reactor was packed with CsF on CaF₂ with a loading of 3.5 mmol CsF/(g CsF on CaF₂).*

Determination of internal volume

The internal volume of the packed bed reactor was determined by weighing the freshly packed bed reactor, m_1 . Then toluene or acetonitrile was pumped through the reactor to fill the reactor void with solvent and the packed bed reactor was weighed again, m_2 . The reactor void volume, V_{void} , was determined by:

$$V_{\text{void}} = \frac{m_2 + m_1}{\rho_{\text{solvent}}}$$

General experimental Setup for Continuous Flow

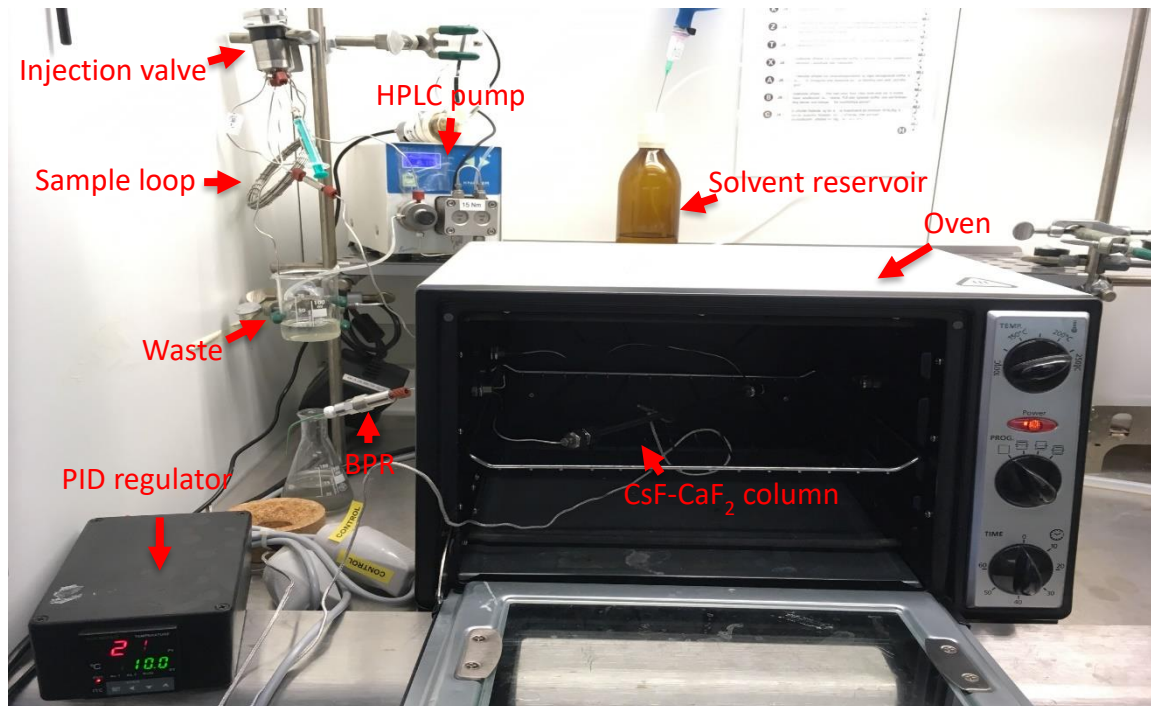


Figure S1: Picture of Continuous flow setup.

A HPLC pump (Knauer Azura P 4.1S) was used to pump solvent through the reactor. A sample loop (2 mL) was filled by syringe with a solution of substrate and additive in the chosen solvent. The sample loop was made of stainless steel tubing (1/16" OD x 0.75 mm ID) and had a volume of 2 mL (453 cm length). An injection valve (2 position: load and inject, 6-port, 1/16", Vici) was used to introduce the reagents from the sample loop. A backpressure regulator (BPR) was used to prevent solvent boiling. All units were connected using stainless steel tubing (1/16" OD x 0.75 mm ID) and HPLC fittings (T-pieces, unions etc. all with 1/16" ID). The packed bed was heated in an oven and was dried by passing dry acetonitrile or toluene through the system at 1.50 mL/min at 180°C for 30 min after which the column was ready for use.

In-line Drying of CsF-CaF₂ Reactor

A freshly packed CsF-CaF₂ reactor was heated to 230°C and flushed with acetonitrile with an initial water content of 9.2 ppm with a flow rate of 1.00 mL/min. The exiting flow was collected in fractions of 24 mL in septum-sealed round-bottomed flasks for a total of 240 mL. The water content of each fraction was determined by Karl-Fischer measurements and depicted in Figure S2.

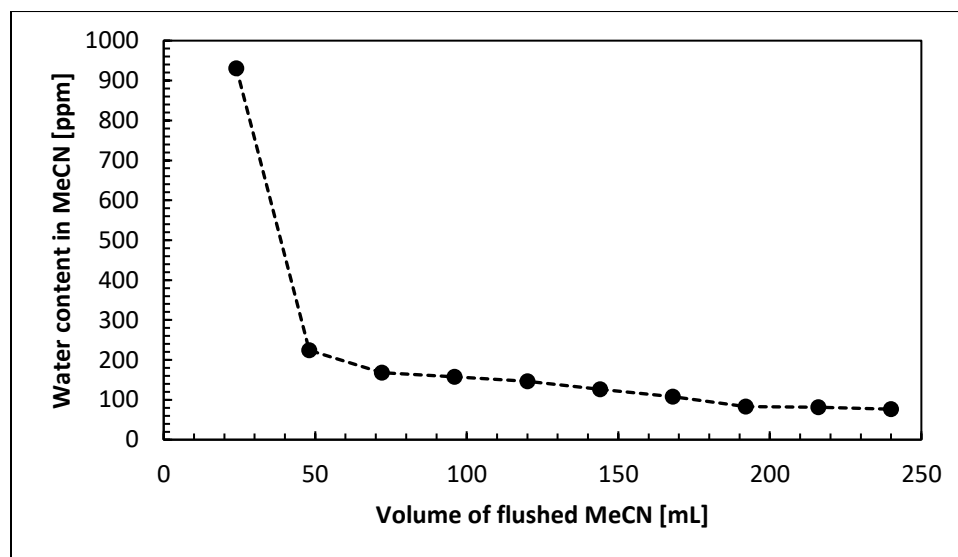


Figure S2: Water content of exiting solvent as a function of flushed acetonitrile (MeCN)

Influence of CsF Loading on CaF_2 in Nucleophilic Fluoride Substitution of 1-bromomethyl-4-*tert*butylbenzene in Continuous Flow

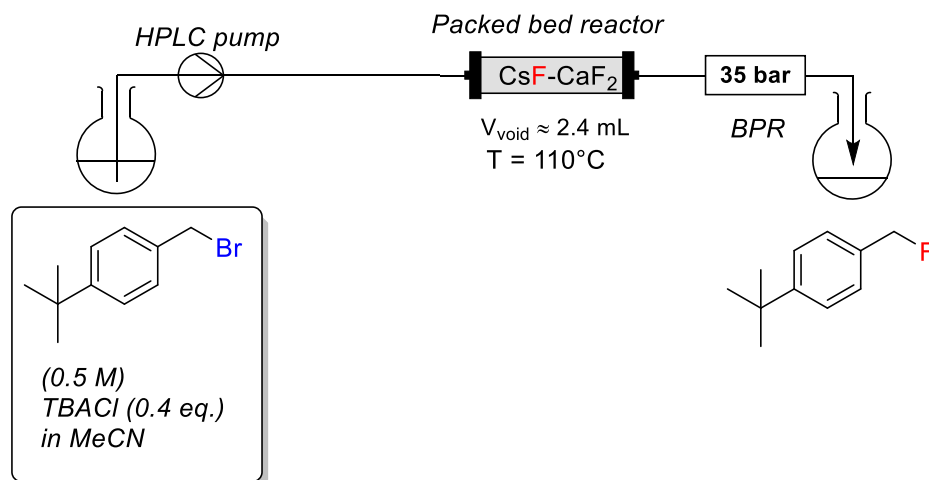


Figure S3: Schematic presentation of the scale-out experiment of the fluorination of 1-bromomethyl-4-*tert*butylbenzene.

1-bromomethyl-4-*tert*butylbenzene (5.68 g, 25 mmol, 1 equiv.) and tetrabutylammonium chloride (2.74 g, 9.9 mmol, 0.4 equiv.) was dissolved in acetonitrile (0.5 M). The reaction mixture was pumped through the packed bed reactor, containing 21 mmol CsF, with a flow rate of 0.097 mL/min corresponding to a residence time of 25 min at 110°C. The flow setup was run for 1.5 times the residence time, to reach steady state, before sample collection was initiated. A batch was collected every 20 minutes, concentrated under reduced pressure and conversion and purity was determined by ^1H -NMR analysis using dimethyl terephthalate as internal standard. Full conversion was observed in all fractions. ^1H NMR yields are depicted in the Figure S4. Fraction 1-10 were combined and purified by flash column chromatography eluting with

pentane/CH₂Cl₂ 20:1 affording 1-fluoromethyl-4-*tert*butylbenzene (**2**) (1.07 g, 68%) as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.44 (d, *J* = 8.0 Hz, 2H), 7.34 (d, *J* = 8.2 Hz, 2H), 5.36 (d, *J* = 48.1 Hz, 2H), 1.34 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 152.1 (d, *J* = 3.4 Hz), 133.4 (d, *J* = 17.1 Hz), 127.8 (d, *J* = 5.4 Hz), 125.7 (d, *J* = 1.6 Hz), 84.7 (d, *J* = 165.0 Hz), 34.8, 31.5; ¹⁹F NMR (376 MHz, CDCl₃) δ -204.4 (t, *J* = 48.1 Hz, 1F); HRMS *m/z* calculated for C₁₁H₁₆ [M - F]⁺ 147.1168, found 147.1166.

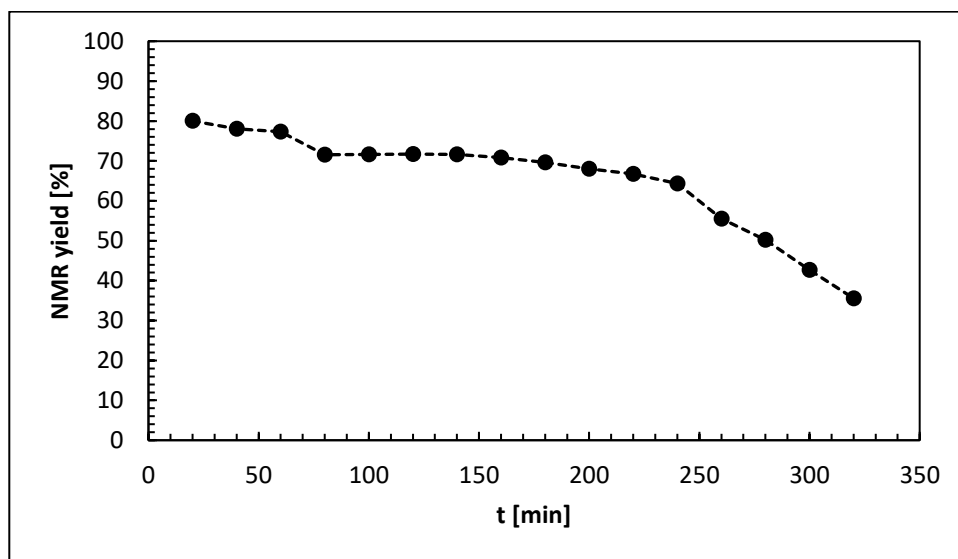


Figure S4: NMR yield of 1-fluoromethyl-4-*tert*butylbenzene as a function time using 3.5 mmol CsF/(g CsF-CaF₂).

1-bromomethyl-4-*tert*butylbenzene (5.68 g, 25 mmol, 1 equiv.) and tetrabutylammonium chloride (2.74 g, 9.9 mmol, 0.4 equiv.) was dissolved in acetonitrile (0.5 M). The reaction mixture was pumped through the packed bed reactor, containing 14 mmol CsF [2.2 mmol CsF/(g CsF on CaF₂)], with a flow rate of 0.089 mL/min corresponding to a residence time of 25 min at 110°C. The flow setup was run for 1.5 times the residence time, to reach steady state, before sample collection was initiated. A batch was collected every 20 minutes, concentrated under reduced pressure and yield was determined by ¹H-NMR analysis using dimethyl terephthalate as internal standard. ¹H NMR yields are depicted in the Figure S5.

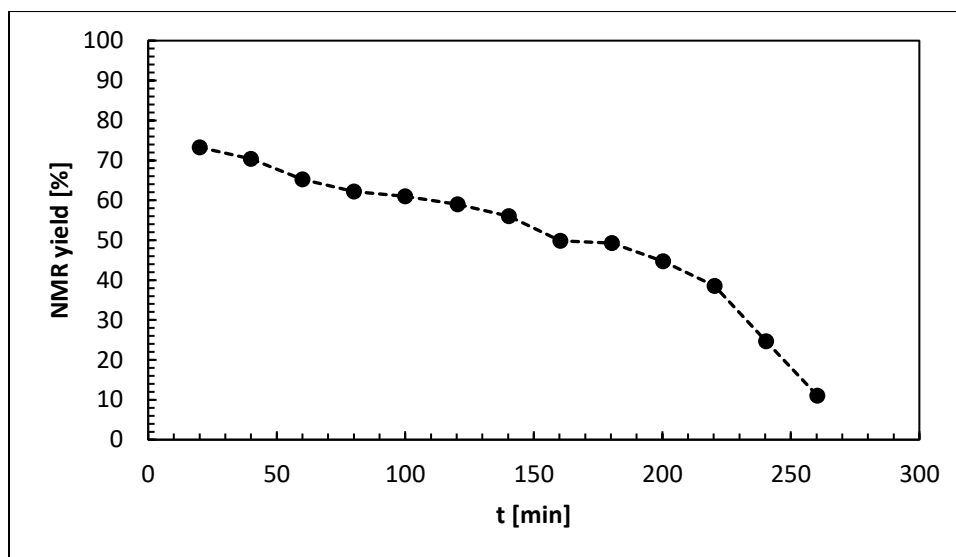


Figure S5: NMR yield of 1-fluoromethyl-4-*tert*butylbenzene as a function time using 2.2 mmol CsF/(g CsF-CaF₂).

1-bromomethyl-4-*tert*butylbenzene (5.68 g, 25 mmol, 1 equiv.) and tetrabutylammonium chloride (2.74 g, 9.9 mmol, 0.4 equiv.) was dissolved in acetonitrile (0.5 M). The reaction mixture was pumped through the packed bed reactor, containing 19 mmol KF [3.5 mmol KF/(g KF on CaF₂)], with a flow rate of 0.089 mL/min corresponding to a residence time of 25 min at 110°C. The flow setup was run for 1.5 times the residence time, to reach steady state, before sample collection was initiated. A batch was collected every 20 minutes, concentrated under reduced pressure and yield was determined by ¹H-NMR analysis using dimethyl terephthalate as internal standard. Full conversion was observed in all fractions. ¹H NMR yields are depicted in the Figure S6.

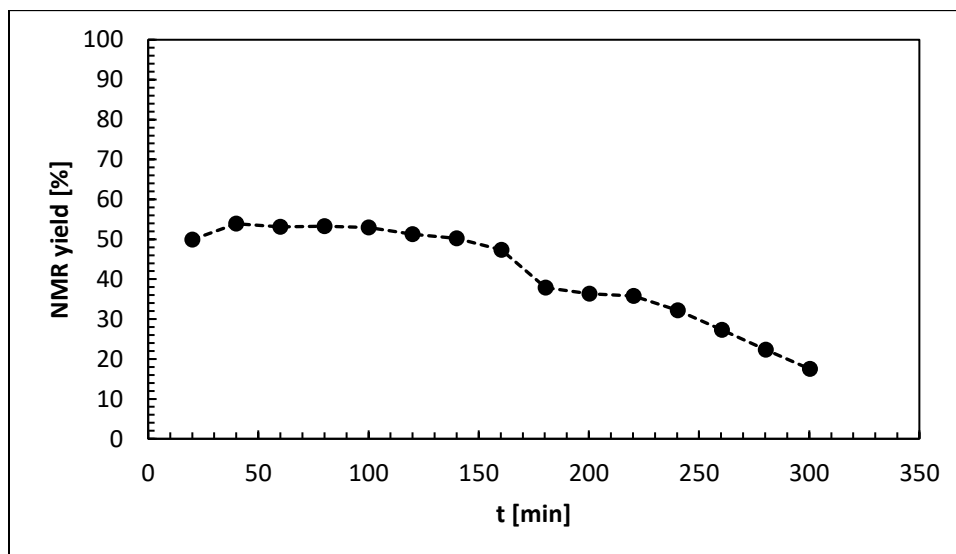


Figure S6: NMR yield of 1-fluoromethyl-4-*tert*butylbenzene as a function time using 3.5 mmol KF/(g KF-CaF₂).

Procedure for Nucleophilic Fluorination of Benzyl Bromides: Substrate Scope and Isolated Yields in Flow

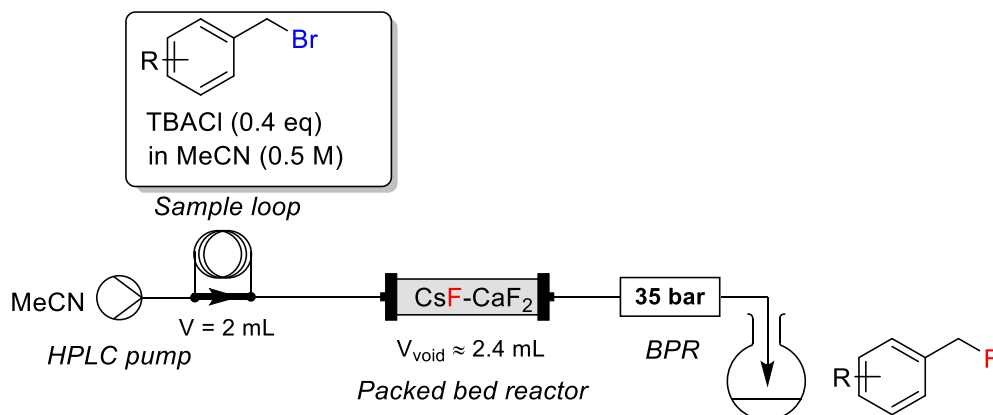
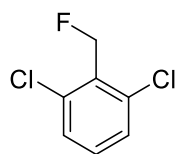


Figure S7: Schematic presentation of fluorination of benzyl bromides.

General Method A: The benzyl bromide (1 equiv.) and tetrabutylammonium chloride (TBACl, 0.40 equiv.) were dissolved in acetonitrile (0.50 M). 2.0 mL of the stock solution was loaded into the sample loop and pumped through the packed bed reactor, containing 21 mmol CsF, with a flow rate of 0.098 mL/min corresponding to a residence time of 25 min at 110°C, Figure S7. Product was collected for 105 min, concentrated under reduced pressure and purified by flash column chromatography.



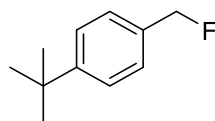
1-Fluoromethyl-2,6-dichlorobenzene (1). The crude product was obtained by using General Method A and purified using flash column chromatography eluting with pentane/CH₂Cl₂ 10:1 affording the title product (129 mg, 70%) as a white solid. mp 63–64°C; ¹H NMR (400 MHz, CDCl₃) δ 7.36 (d, *J* = 8.0 Hz, 2H), 7.32–7.20 (m, 1H), 5.69 (d, *J* = 47.4 Hz, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 137.8 (d, *J* = 3.8 Hz), 131.4 (d, *J* = 4.0 Hz), 131.4 (d, *J* = 14.6 Hz), 128.7 (d, *J* = 2.9 Hz), 78.8 (d, *J* = 166.7 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -211.6 (t, *J* = 46.4 Hz, 1F); HRMS *m/z* calculated for C₇H₆Cl₂F [M+H]⁺ 178.9825, found 178.9810.

The title product (**1**) was also produced from 1-chloromethyl-2,6-dichlorobenzene otherwise following General Method A and purified using flash column chromatography eluting with pentane/CH₂Cl₂ 20:1, affording an inseparable mixture of the title product (116 mg, 65%) and 1-chloromethyl-2,6-dichlorobenzene (26 mg, 13%) as a white solid.

1-Fluoromethyl-2,6-dichlorobenzene: ¹H NMR (400 MHz, CDCl₃) δ *inter alia* 5.72 (d, *J* = 47.4 Hz, 2H); ¹⁹F NMR (376 MHz, CDCl₃) δ -211.6 (t, *J* = 47.4 Hz, 1F);

1-Chloromethyl-2,6-dichlorobenzene: ¹H NMR (400 MHz, CDCl₃) δ *inter alia* 4.91 (s, 2H);

1-Fluoromethyl-2,6-dichlorobenzene and 1-chloromethyl-2,6-dichlorobenzene: ¹³C NMR (101 MHz, CDCl₃) δ *inter alia* 137.4 (d, *J* = 3.9 Hz), 136.3, 133.6, 131.4 (d, *J* = 4.0 Hz), 131.4 (d, *J* = 14.6 Hz), 130.4, 128.7, 128.7 (d, *J* = 3.1 Hz), 78.9 (d, *J* = 166.7 Hz), 40.8.



1-Fluoromethyl-4-tertbutylbenzene (2).⁹ The crude product was obtained by using General Method A and purified using flash column chromatography eluting with pentane/CH₂Cl₂ 10:1 affording the title product (125 mg, 74%) as a colorless oil. ¹H

NMR (400 MHz, CDCl_3) δ 7.44 (d, J = 8.0 Hz, 2H), 7.34 (dd, J = 8.2 Hz, J = 2.0 Hz, 2H), 5.36 (d, J = 48.1 Hz, 2H), 1.35 (s, 9H); ^{13}C NMR (101 MHz, CDCl_3) δ 152.1 (d, J = 3.4 Hz), 133.4 (d, J = 17.2 Hz), 127.8 (d, J = 5.5 Hz), 125.7 (d, J = 1.7 Hz), 84.7 (d, J = 165.0 Hz), 34.8, 31.5; ^{19}F NMR (376 MHz, CDCl_3) δ -204.4 (t, J = 48.2 Hz, 1F); HRMS m/z calculated for $\text{C}_{11}\text{H}_{15}$ $[\text{M} - \text{F}]^+$ 147.1168, found 147.1166.

The title product (**1**) was also produced from 1-chloromethyl-4-tertbutylbenzene otherwise following General Method A and purified using flash column chromatography eluting with pentane/ CH_2Cl_2 20:1, affording the title product (118 mg, 72%) as a colorless oil. ^1H NMR (400 MHz, CDCl_3) 7.43 (d, J = 7.8 Hz, 2H), 7.34 (d, J = 8.1 Hz, 2H), 5.36 (d, J = 48.1 Hz, 2H), 1.34 (s, 9H); ^{13}C NMR (101 MHz, CDCl_3) δ 152.0 (d, J = 3.4 Hz), 133.2 (d, J = 17.1 Hz), 127.6 (d, J = 5.5 Hz), 125.6 (d, J = 1.6 Hz), 84.6 (d, J = 165.0 Hz), 34.7, 31.3; ^{19}F NMR (376 MHz, CDCl_3) δ -204.4 (t, J = 48.1 Hz, 1F); HRMS m/z calculated for $\text{C}_{11}\text{H}_{15}$ $[\text{M} - \text{F}]^+$ 147.1168, found 147.1160.

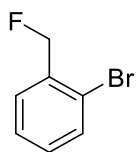
The title product (**1**) was also produced using a KF- CaF_2 column containing 22 mmol KF [4.0 mmol KF/(g KF on CaF_2)] otherwise following General Method A and purified using flash column chromatography eluting with pentane/ CH_2Cl_2 20:1, affording an inseparable mixture of the title product (100 mg, 60%), 1-bromomethyl-4-tertbutylbenzene (19 mg, 8%) and 1-chloromethyl-4-tertbutylbenzene (48 mg, 26%) as a colorless oil.

1-Fluoromethyl-4-tertbutylbenzene: ^1H NMR (400 MHz, CDCl_3) δ *inter alia* 5.36 (d, J = 48.1 Hz, 2H); ^{19}F NMR (376 MHz, CDCl_3) δ -204.4 (t, J = 48.1 Hz, 1F);

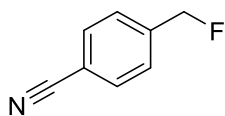
1-Bromomethyl-4-tertbutylbenzene: ^1H NMR (400 MHz, CDCl_3) δ *inter alia* 4.50 (s, 2H).

1-Chloromethyl-4-tertbutylbenzene: ^1H NMR (400 MHz, CDCl_3) δ *inter alia* 4.59 (s, 2H);

1-Fluoromethyl-4-tertbutylbenzene, 1-bromomethyl-4-tertbutylbenzene and 1-chloromethyl-4-tertbutylbenzene: ^{13}C NMR (101 MHz, CDCl_3) δ *inter alia* 152.10 (d, J = 3.4 Hz), 151.72, 151.67, 134.90, 134.65, 133.35 (d, J = 17.1 Hz), 128.91, 128.49, 127.77 (d, J = 5.5 Hz), 125.92, 125.85, 125.69 (d, J = 1.6 Hz), 84.71 (d, J = 165.0 Hz), 46.31, 34.82, 34.82, 34.78, 33.77, 31.44, 31.43, 31.41.



1-Fluoromethyl-2-bromobenzene (**3**):¹⁰ The crude product was obtained by using method A and purified using flash column chromatography eluting with pentane/ CH_2Cl_2 10:1 affording the title product (132 mg, 68%) as a colorless oil. ^1H NMR (400 MHz, CDCl_3) δ 7.57 (d, J = 8.0 Hz, 1H), 7.48 (d, J = 7.6 Hz, 1H), 7.37 (t, J = 7.5 Hz, 1H), 7.22 (t, J = 7.6 Hz, 1H), 5.48 (d, J = 47.2 Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 136.0 (d, J = 18.0 Hz), 132.8, 130.0 (d, J = 2.3 Hz), 128.6 (d, J = 9.6 Hz), 127.7, 121.8 (d, J = 5.3 Hz), 83.8 (d, J = 169.7 Hz); ^{19}F NMR (376 MHz, CDCl_3) δ -216.4 (t, J = 47.2 Hz, 1F); HRMS m/z calculated for $\text{C}_7\text{H}_6\text{BrF}$ $[\text{M}]^+$ 187.9637, found 187.9644.



1-Fluoromethyl-4-cyanobenzene (**4**):⁹ The crude product was obtained by using method A and purified using flash column chromatography eluting with pentane/ EtOAc 20:1 affording the title product (96 mg, 68%) as a white solid. mp 29-30°C; ^1H NMR (400 MHz, CDCl_3) δ 7.69 (d, J = 8.0 Hz, 2H), 7.47 (d, J = 7.8 Hz, 2H), 5.45 (d, J = 46.9 Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 136.0 (d, J = 18.0 Hz), 132.8, 130.0 (d, J = 2.3 Hz), 128.6 (d, J = 9.6 Hz), 127.7, 121.8 (d, J = 5.3 Hz), 83.8 (d, J = 169.7 Hz); ^{19}F NMR (376 MHz, CDCl_3) δ -216.4 (t, J = 47.2 Hz, 1F); HRMS m/z calculated for $\text{C}_8\text{H}_7\text{FN}$ $[\text{M} + \text{H}]^+$ 136.0557, found 136.0554.

Procedure for Nucleophilic (hetero)aromatic Substitutions: Substrate Scope and Isolated Yields in Flow.

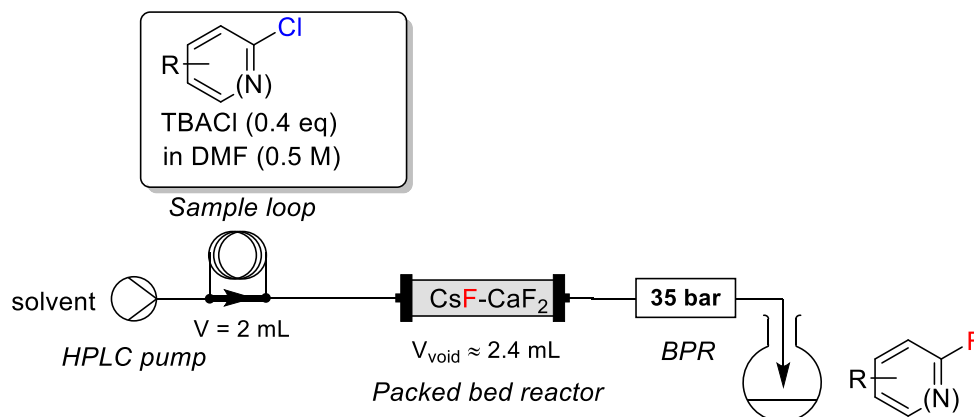
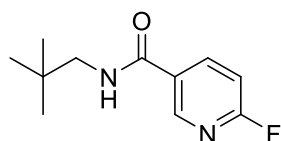
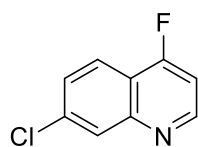


Figure S8: Schematic presentation of fluorination of chlorinated (hetero)aromatics.

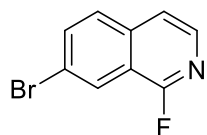
General Method B: The (hetero)aromatic chloride (1 equiv.) and tetrabutylammonium chloride (TBACl, 0.40 equiv.) were dissolved in DMF (0.50 M) unless otherwise stated. 2.0 mL of the stock solution was loaded into the sample loop and pumped through the packed bed reactor, containing 21 mmol CsF, with MeCN as background solvent unless otherwise stated, Figure S8. The residence time and temperature are stated individually below. The sample was collected for 4 times the residence time and extracted with Et₂O (4 x 20 mL). The combined ethereal phases was washed with water (4 x 20 mL), brine (2 x 20 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified using flash column chromatography.



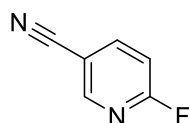
6-Fluoro-N-neopentylnicotinamide (6). The reaction was run in DMSO at 150°C with a residence time of 25 min and DMSO was used as background solvent, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 3:1 affording the title product (145 mg, 68%) as a white solid. mp 110-112°C; ¹H NMR (400 MHz, CDCl₃) δ 8.59 (d, *J* = 1.8 Hz, 1H), 8.23 (td, *J* = 8.2 Hz, *J* = 2.4 Hz, 1H), 7.00 (dd, *J* = 8.5 Hz, *J* = 2.8 Hz, 1H), 6.21 (br. s, 1H), 3.28 (d, *J* = 6.3 Hz, 2H), 0.98 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 165.1 (d, *J* = 244.1 Hz), 164.8, 126.5 (d, *J* = 15.9 Hz), 141.0 (d, *J* = 8.8 Hz), 129.2 (d, *J* = 4.5 Hz), 109.9 (d, *J* = 39.4 Hz), 51.3, 32.3, 27.4; ¹⁹F NMR (376 MHz, CDCl₃) δ -63.6 (s, 1F); HRMS *m/z* calculated for C₁₁H₁₆FN₂O [M + H]⁺ 211.1241, found 211.1243.



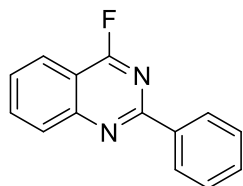
7-Chloro-4-fluoroquinoline (7).¹¹ The reaction was run at 130°C with a residence time of 30 min, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/Et₂O 8:1 affording the title product (106 mg, 58%) as a white solid. mp 53-55°C; ¹H NMR (400 MHz, CDCl₃) δ 8.86 (dd, *J* = 8.2 Hz, *J* = 5.0 Hz, 1H), 8.13 (s, 1H), 8.04 (d, *J* = 8.9 Hz, 1H), 7.65-7.49 (m, 1H), 7.10 (dd, *J* = 9.5 Hz, *J* = 5.0 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 165.4 (d, *J* = 269.5 Hz), 152.8 (d, *J* = 8.1 Hz), 151.0 (d, *J* = 4.2 Hz), 136.9, 128.4 (d, *J* = 3.5 Hz), 128.2, 122.1 (d, *J* = 4.9 Hz), 106.1 (d, *J* = 14.5 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -111.2-112.7 (m, 1F); HRMS *m/z* calculated for C₉H₆ClFN [M + H]⁺ 182.0167, found 182.0182.



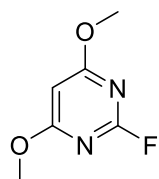
7-Bromo-1-fluoroisoquinoline (8).¹² The reaction was run at 120°C with a residence time of 30 min with a substrate concentration of 0.28 M, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/CH₂Cl₂ 1:1 affording the title product (83 mg, 64%) as a pale orange solid. mp 81-82°C; ¹H NMR (400 MHz, CDCl₃) δ 8.33 (s, 1H), 8.09 (d, *J* = 5.6 Hz, 1H), 7.85 (dd, *J* = 8.8 Hz, *J* = 1.8 Hz, 1H), 7.75 (d, *J* = 8.7 Hz, 1H), 7.51 (d, *J* = 5.5 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 159.0 (d, *J* = 246.9 Hz), 139.9 (d, *J* = 15.8 Hz), 138.2 (d, *J* = 5.3 Hz), 135.2, 128.2 (d, *J* = 3.9 Hz), 125.7, 121.9, 119.2 (d, *J* = 5.0 Hz), 118.8 (d, *J* = 32.8 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -70.6 (s, 1F); HRMS *m/z* calculated for C₉H₆BrFN [M + H]⁺ 225.9662, found 225.9685.



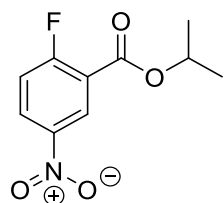
6-Fluoronicotinonitrile (9). The reaction was run at 50°C with a residence time of 40 min, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 20:1 affording the title product (73 mg, 59%) as a white solid. mp 52-53°C; ¹H NMR (400 MHz, CDCl₃) δ 8.63-8.53 (m, 1H), 8.20-7.97 (m, 1H), 7.10 (dd, *J* = 8.5 Hz, *J* = 2.9 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 165.1 (d, *J* = 249.2 Hz), 152.4 (d, *J* = 16.8 Hz), 144.5 (d, *J* = 9.3 Hz), 115.7, 111.0 (d, *J* = 38.0 Hz), 108.2 (d, *J* = 5.0 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -57.4 (s, 1F); HRMS *m/z* calculated for C₆H₄FN₂ [M + H]⁺ 123.0353, found 123.0353.



4-Fluoro-2-phenylquinazoline (10). The reaction was run at 60°C with a residence time of 40 min with a substrate concentration of 0.25 M, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 40:1 affording the title product (75 mg, 66%) as a white solid. mp 86-88°C; ¹H NMR (400 MHz, CDCl₃) δ 8.66-8.48 (m, 2H), 8.12 (d, *J* = 8.6 Hz, 2H), 8.01-7.86 (m, 1H), 7.64 (t, *J* = 7.5 Hz, 1H), 7.59-7.51 (m, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 167.3 (d, *J* = 259.9 Hz), 160.6 (d, *J* = 14.5 Hz), 154.5 (d, *J* = 5.4 Hz), 136.8, 135.2, 131.3, 128.8 (2C), 128.4 (d, *J* = 5.0 Hz), 127.9, 122.8, 113.2 (d, *J* = 27.1 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -64.4 (s, 1F); HRMS *m/z* calculated for C₁₄H₁₀FN₂ [M + H]⁺ 225.0823, found 225.0825.



2-Fluoro-4,6-dimethoxypyrimidine (11). The reaction was run at 90°C with a residence time of 25 min, otherwise following General Method B. No further purification was needed and the title product was isolated (104 mg, 65%) as a white solid. mp 85-87°C; ¹H NMR (400 MHz, CDCl₃) δ 5.93 (d, *J* = 2.4 Hz, 1H), 3.94 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 173.7 (d, *J* = 15.7 Hz), 161.9 (d, *J* = 216.0 Hz), 87.7 (d, *J* = 6.8 Hz), 54.8; ¹⁹F NMR (376 MHz, CDCl₃) δ -44.4 (s, 1F); HRMS *m/z* calculated for C₆H₈FN₂O₂ [M + H]⁺ 159.0564, found 159.0563.



Isopropyl 2-fluoro-5-nitrobenzoate (12). The reaction was run at 130°C with a residence time of 10 min with a substrate concentration of 0.25 M, otherwise following General Method B. The crude product was purified using flash column chromatography eluting with pentane/Et₂O 8:1 affording the title product (86 mg, 75%) as a yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 8.80 (dd, *J* = 6.2 Hz, *J* = 2.9 Hz, 1H), 8.39 (ddd, *J* = 9.0 Hz, *J* = 3.9 Hz, *J* = 2.9 Hz, 1H), 7.31 (d, *J* = 9.2 Hz, 1H), 5.30 (hept, *J* = 6.3 Hz, 1H), 1.40 (d, *J* = 6.3 Hz, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 165.2 (d, *J* = 270.9 Hz), 161.8 (d, *J* = 4.1 Hz), 144.0 (d, *J* = 3.0 Hz), 129.3 (d, *J* = 11.0 Hz), 128.2 (d, *J* = 3.0 Hz), 120.8 (d, *J* = 11.8 Hz), 118.5 (d, *J* = 25.0 Hz).

Hz), 70.4, 22.0; ^{19}F NMR (376 MHz, CDCl_3) δ -98.7 (s, 1F); HRMS m/z calculated for $\text{C}_{10}\text{H}_{10}\text{FNNaO}_4$ $[\text{M} + \text{Na}]^+$ 250.0486, found 250.0487.

Procedure for Deprotection of Silyl Ethers: Substrate Scope and Isolated Yields in Flow.

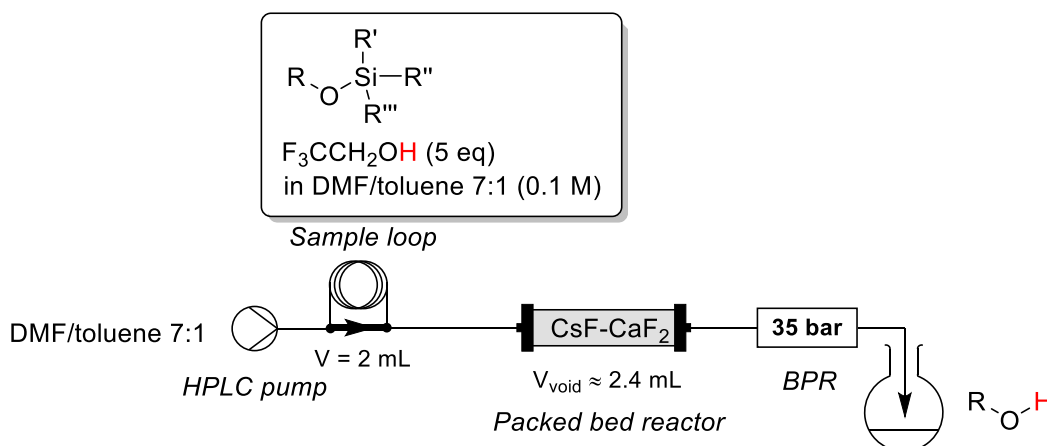
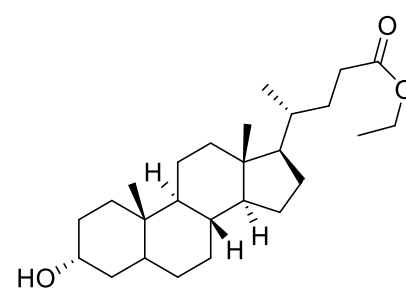
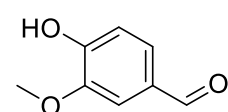


Figure S9: Schematic presentation of the deprotection of silyl ethers.

General Method C: Silyl ether (1 equiv.) and 2,2,2-trifluoroethanol (5.0 equiv.) was dissolved in DMF/toluene 7:1 (V/V) (0.10 M). 2.0 mL of the stock solution was loaded into the sample loop and pumped through the packed bed reactor, containing 22 mmol CsF, with a flow rate of 0.475 mL/min corresponding to a residence time of 5 min at 120°C with DMF/toluene 7:1 (V/V) as background solvent, Figure S9. The sample was collected for 20 min, diluted with water (10 mL), and extracted with Et_2O (4 x 10 mL). The combined ethereal phases was washed with water (4 x 10 mL), brine (2 x 10 mL), dried over Na_2SO_4 and concentrated under reduced pressure. The crude product was purified using flash column chromatography.

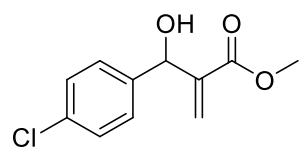
 **Ethyl 3α-hydroxy-5β-cholan-24-oate (13).**¹ Ethyl 3α-((triisopropylsilyl)oxy)-5β-cholan-24-oate was deprotected following General Method C. The crude product was purified using flash column chromatography eluting with toluene/ EtOAc 10:1 affording the title product (79 mg, 94%) as a colorless wax. ^1H NMR (400 MHz, CDCl_3) δ 4.11 (q, $J = 7.1$ Hz, 2H), 3.62 (tt, $J = 10.6, 4.7$ Hz, 1H), 2.33 (ddd, $J = 15.3, 10.1, 5.2$ Hz, 1H), 2.20 (ddd, $J = 15.6, 9.6, 6.6$ Hz, 1H), 2.02 – 0.82 (m, 36H), 0.63 (s, 3H); ^{13}C NMR (101 MHz, CDCl_3) δ 174.5, 72.0, 60.3, 56.6, 56.1, 42.9, 42.2, 40.6, 40.3, 36.6, 36.0, 35.5 (2C), 34.7, 31.5, 31.1, 30.7, 28.3, 27.3, 26.6, 24.4, 23.5, 21.0, 18.4, 14.4, 12.2; HRMS m/z calculated for $\text{C}_{26}\text{H}_{44}\text{NaO}_3$ $[\text{M} + \text{Na}]^+$ 427.3183, found 427.3184.

 **Vanillin (14).**¹³ 3-methoxy-4-((triethylsilyl)oxy)benzaldehyde was deprotected following General Method C. The crude product was purified using flash column chromatography eluting with toluene/ EtOAc 10:1 affording the title product (29 mg, 93%) as a white solid. mp 81–82°C; ^1H NMR (400 MHz, CDCl_3) δ 9.83 (s, 1H), 7.51–7.39 (m, 2H), 7.04 (d, $J =$

8.5 Hz, 1H), 6.17 (s, 1H), 3.97 (s, 3H); ^{13}C NMR (101 MHz, CDCl_3) δ 191.0, 151.8, 147.3, 130.1, 127.7, 114.5, 108.9, 56.3; HRMS m/z calculated for $\text{C}_8\text{H}_9\text{O}_3$ $[\text{M} + \text{H}]^+$ 153.0546, found 153.0545.

The title product (**14**) was also produced running with a silyl ether concentration on 0.5 M, affording the title product (146 mg, 93%) as an off-white solid.

3-Methoxy-4-((tert-butyldimethylsilyl)oxy)benzaldehyde was deprotected following General Method C. The crude product was purified using flash column chromatography eluting with toluene/EtOAc 10:1 affording the title product (28 mg, 90%) as a white solid.

 *Methyl 2-((4-chlorophenyl)(hydroxy)methyl)acrylate (15).*¹⁴ Methyl 2-(((tert-butyldimethylsilyl)oxy)(4-chlorophenyl)methyl)acrylate was deprotected at room temperature with a residence time of 20 min, otherwise following General Method C. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 8:1 affording the title product (39 mg, 86%) as a white solid. mp 39-40°C; ^1H NMR (400 MHz, CDCl_3) δ 7.31 (s, 4H), 6.34 (s, 1H), 5.82 (s, 1H), 5.52 (d, J = 5.1 Hz, 1H), 3.73 (s, 3H), 3.10 (d, J = 5.5 Hz, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 166.6, 141.6, 139.8, 133.6, 128.6, 128.0, 126.4, 72.7, 52.1; HRMS m/z calculated for $\text{C}_{11}\text{H}_{10}\text{ClNaO}_3$ $[\text{M} + \text{Na}]^+$ 249.0289, found 249.0287.

Procedure for CsF-mediated Trifluoromethylation: Substrate Scope and Isolated Yields in Flow

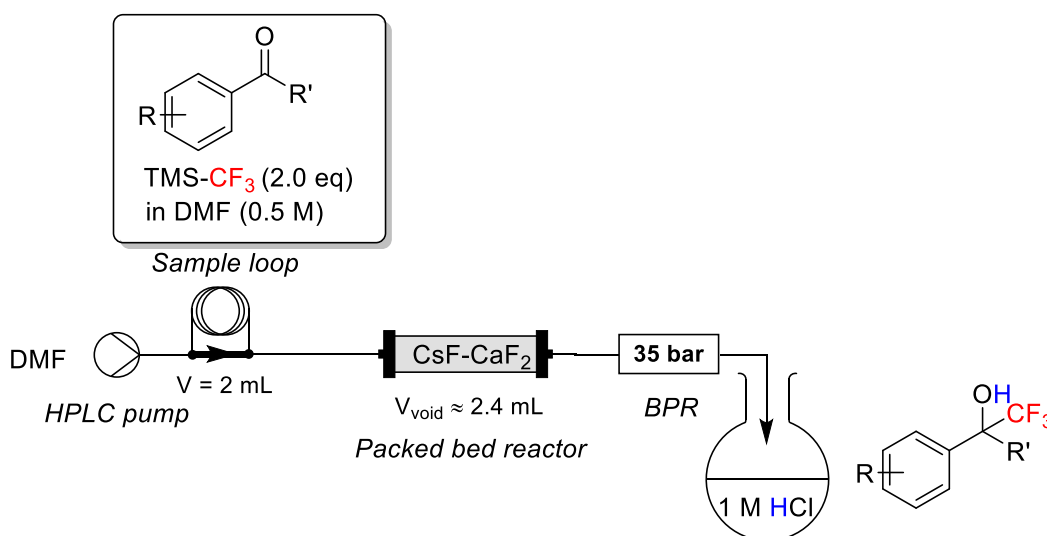
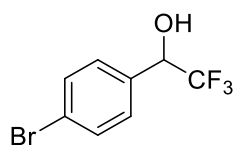


Figure S10: Schematic presentation of fluoride mediated trifluoromethylation.

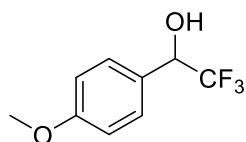
General Method D: Aldehyde/ketone (1 equiv.) and trimethyl(trifluoromethyl)silane (2 equiv.) was dissolved in DMF (0.50 M). 2.0 mL of the stock solution was loaded into the sample loop and pumped through the packed bed reactor, containing 21 mmol CsF, with a flow rate of 0.466 mL/min corresponding to a residence time of 5 min at room temperature, Figure S10. Product was collected in round-bottomed flask containing 20 mL 1 M HCl over 30 min and extracted with Et_2O (4 x 20 mL). The combined ethereal

phases was washed with water (4 x 20 mL), brine (2 x 20 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified using flash column chromatography.

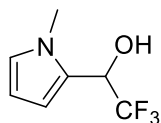
Derivatized MS Procedure¹⁵. Phosphorylation of 2,2,2-trifluoromethylcarbinols: Obtaining HRMS data proved difficult for 2 entries and these were therefore derivatized prior to HRMS analysis as their corresponding phosphorylated products. In a vial (4 mL) were added the 2,2,2-trifluoromethylcarbinol (10 mg), K₂CO₃ (35 mg, 0.25 mmol), diethyl chlorophosphate (43 mg, 0.25 mmol), and acetone (1 mL). The reaction mixture was stirred for 2 h at room temperature. The crude reaction mixture was filtered, and an aliquot was sampled for MS analysis detecting the corresponding diethyl (2,2,2-trifluoro-1-arylethyl)phosphate.



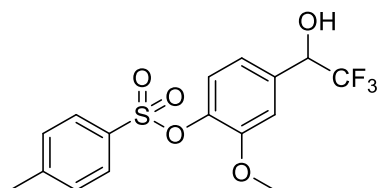
1-(4-Bromophenyl)-2,2,2-trifluoroethan-1-ol (16).¹⁶ The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 10:1 affording the title product (213 mg, 83%) as a yellow solid. mp 49-51°C; ¹H NMR (400 MHz, CDCl₃) δ 7.55 (d, *J* = 8.4 Hz, 2H), 7.36 (d, *J* = 8.3 Hz, 2H), 5.00 (q, *J* = 6.4 Hz, 1H), 2.61 (s, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 132.9, 132.0, 129.2, 124.1 (q, *J* = 282.2 Hz), 123.9, 72.4 (q, *J* = 32.2 Hz); ¹⁹F NMR (376 MHz, CDCl₃) δ -78.5 (d, *J* = 6.6 Hz, 3F); HRMS was determined using the derivatized MS procedure: HRMS *m/z* calculated for C₁₂H₁₆BrF₃O₄P [M + H]⁺ 390.9916, found 390.9917.



1-(4-Methoxyphenyl)-2,2,2-trifluoroethan-1-ol (17).¹⁷ The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 10:1 affording the title product (161 mg, 77%) as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.40 (d, *J* = 8.6 Hz, 2H), 6.93 (d, *J* = 8.7 Hz, 2H), 4.97 (q, *J* = 6.7 Hz, 1H), 3.83 (s, 3H), 2.52 (s, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 160.6, 128.9, 126.2, 124.5 (q, *J* = 281.9 Hz), 114.2, 72.7 (q, *J* = 32.1 Hz), 55.5; ¹⁹F NMR (376 MHz, CDCl₃) δ -78.6 (d, *J* = 6.7 Hz, 3F); HRMS was determined using the derivatized MS procedure: HRMS *m/z* calculated for C₁₃H₁₉F₃O₅P [M + H]⁺ 343.0917, found 343.0916.

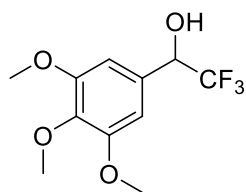


1-(N-methyl-1H-pyrrole-2-yl)-2,2,2-trifluoroethan-1-ol (18). The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 6:1 affording the title product (154 mg, 84%) as a brown solid. mp 45-47°C; ¹H NMR (400 MHz, CDCl₃) δ 7.40 (d, *J* = 8.6 Hz, 2H), 6.93 (d, *J* = 8.7 Hz, 2H), 4.97 (q, *J* = 6.7 Hz, 1H), 3.83 (s, 3H), 2.52 (s, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 125.3, 124.9, 124.3 (q, *J* = 281.6 Hz), 109.4, 107.7, 66.5 (q, *J* = 33.7 Hz), 34.4; ¹⁹F NMR (376 MHz, CDCl₃) δ -77.0 (d, *J* = 6.8 Hz, 3F); HRMS *m/z* calculated for C₇H₉F₃NO [M + H]⁺ 180.0631, found 180.0628.

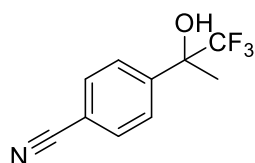


1-(3-Trimethoxy-4-tosyl-phenyl)-2,2,2-trifluoroethan-1-ol (19). The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 4:1 affording the title product (269 mg, 70%) as a viscous, yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, *J* = 8.3 Hz, 2H), 7.31 (d, *J* = 8.3 Hz, 2H), 7.16 (d, *J* = 8.2 Hz, 1H), 7.03-6.92 (m 2H), 4.99 (q, *J* = 6.6 Hz, 1H), 3.59 (s, 3H), 2.45

(s, 3H), 2.22 (br s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 152.1, 145.4, 139.2, 134.1, 133.2, 129.6, 128.7, 124.1 (q, $J = 282.1\text{ Hz}$), 124.1, 120.0, 111.8, 72.3 (q, $J = 32.2\text{ Hz}$), 55.8, 21.8; ^{19}F NMR (376 MHz, CDCl_3) δ -78.3 (d, $J = 6.7\text{ Hz}$, 3F); HRMS m/z calculated for $\text{C}_{16}\text{H}_{16}\text{F}_3\text{O}_5\text{S}$ [$\text{M} + \text{H}$] $^+$ 377.0665, found 377.0657.



1-(3,4,5-Trimethoxyphenyl)-2,2,2-trifluoroethan-1-ol (20).¹⁸ The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 4:1 affording the title product (187 mg, 68%) as a white solid. mp 95-97°C; ^1H NMR (400 MHz, CDCl_3) δ 6.67 (s, 2H), 4.95 (q, $J = 6.7\text{ Hz}$, 1H), 3.86 (s, 6H), 3.85 (s, 3H), 2.81 (s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 153.4, 138.8, 129.7, 124.3 (q, $J = 282.1\text{ Hz}$), 104.7, 73.0 (q, $J = 32.0\text{ Hz}$), 61.0, 56.3; ^{19}F NMR (376 MHz, CDCl_3) δ -78.2 (d, $J = 6.7\text{ Hz}$, 3F); HRMS m/z calculated for $\text{C}_{11}\text{H}_{14}\text{F}_3\text{O}_4$ [$\text{M} + \text{H}$] $^+$ 267.0839, found 267.0833.



4-(1,1,1-Trifluoro-2-hydroxypropan-2-yl)benzonitrile (21).¹⁹ The crude product was obtained by using General Method D. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 10:1 affording the title product (184 mg, 94%) as a yellow solid. mp 117-119°C; ^1H NMR (400 MHz, CDCl_3) δ 7.73 (d, $J = 8.7\text{ Hz}$, 2H), 7.70 (d, $J = 8.7\text{ Hz}$, 2H), 2.55 (s, 1H), 1.80 (s, 3H); ^{13}C NMR (101 MHz, CDCl_3) δ 143.5, 132.2, 127.2, 125.2 (q, $J = 285.5\text{ Hz}$), 118.5, 112.8, 74.8 (q, $J = 29.6\text{ Hz}$), 24.1; ^{19}F NMR (376 MHz, CDCl_3) δ -80.9 (s, 3F); HRMS m/z calculated for $\text{C}_{10}\text{H}_9\text{F}_3\text{NO}$ [$\text{M} + \text{H}$] $^+$ 216.0631, found 216.0626.

Trifluoromethylation of 4-acetylbzenonitrile in Continuous Flow - Scale-out experiment

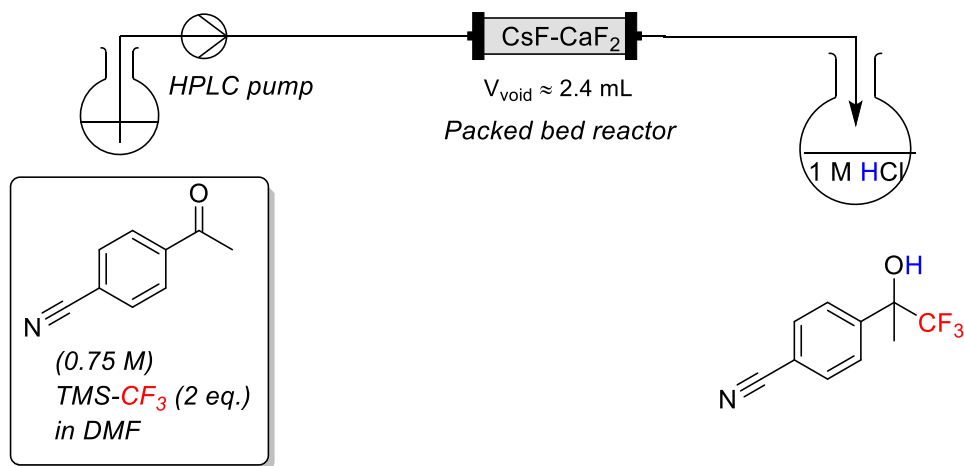


Figure S11: Schematic presentation of scale-out experiment of the trifluoromethylation of 4-acetylbzenonitrile.

4-Acetylbzenonitrile (10.88 g, 75 mmol, 1 equiv.) and trimethyl(trifluoromethyl)silane (21.34 g, 150 mmol, 2 equiv.) was dissolved in DMF (0.75 M). A packed bed reactor, containing 22 mmol CsF, was cooled using a water bath. The stock solution was pumped through the packed bed reactor with a residence time of 2.0 min at room temperature. The flow setup was run for 1.5 times the residence time, to reach steady state, before sample collection was initiated. A batch was collected every 15 minutes in round-bottomed

flasks containing 1 M HCl (20 mL). During the experiment the pressure drop over the packed bed reactor rose slowly and the reactor was shut down at 200 bar (42 minutes total collection time after reaching steady-state corresponding to a throughput of 36.4 mmol). TLC analysis was used to determine full conversion of the three batches, which was combined and extracted with Et₂O (4 x 60 mL). The combined ethereal phases was washed with water (4 x 60 mL), brine (2 x 60 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified using flash column chromatography eluting with pentane/EtOAc 10:1 affording 4-(1,1,1-trifluoro-2-hydroxypropan-2-yl)benzonitrile (**21**) (7.37 g, 94%) as a yellow solid. mp 118-119°C; ¹H NMR (400 MHz, CDCl₃) δ 7.73 (d, *J* = 8.7 Hz, 2H), 7.70 (d, *J* = 8.8 Hz, 2H), 2.52 (s, 1H), 1.80 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 143.7, 132.2, 127.2, 125.2 (q, *J* = 285.5 Hz), 118.5, 112.8, 74.6 (q, *J* = 29.6 Hz), 24.0; ¹⁹F NMR (376 MHz, CDCl₃) δ -80.9 (s, 3F); HRMS *m/z* calculated for C₁₀H₉F₃NO [M + H]⁺ 216.0631, found 216.0629.

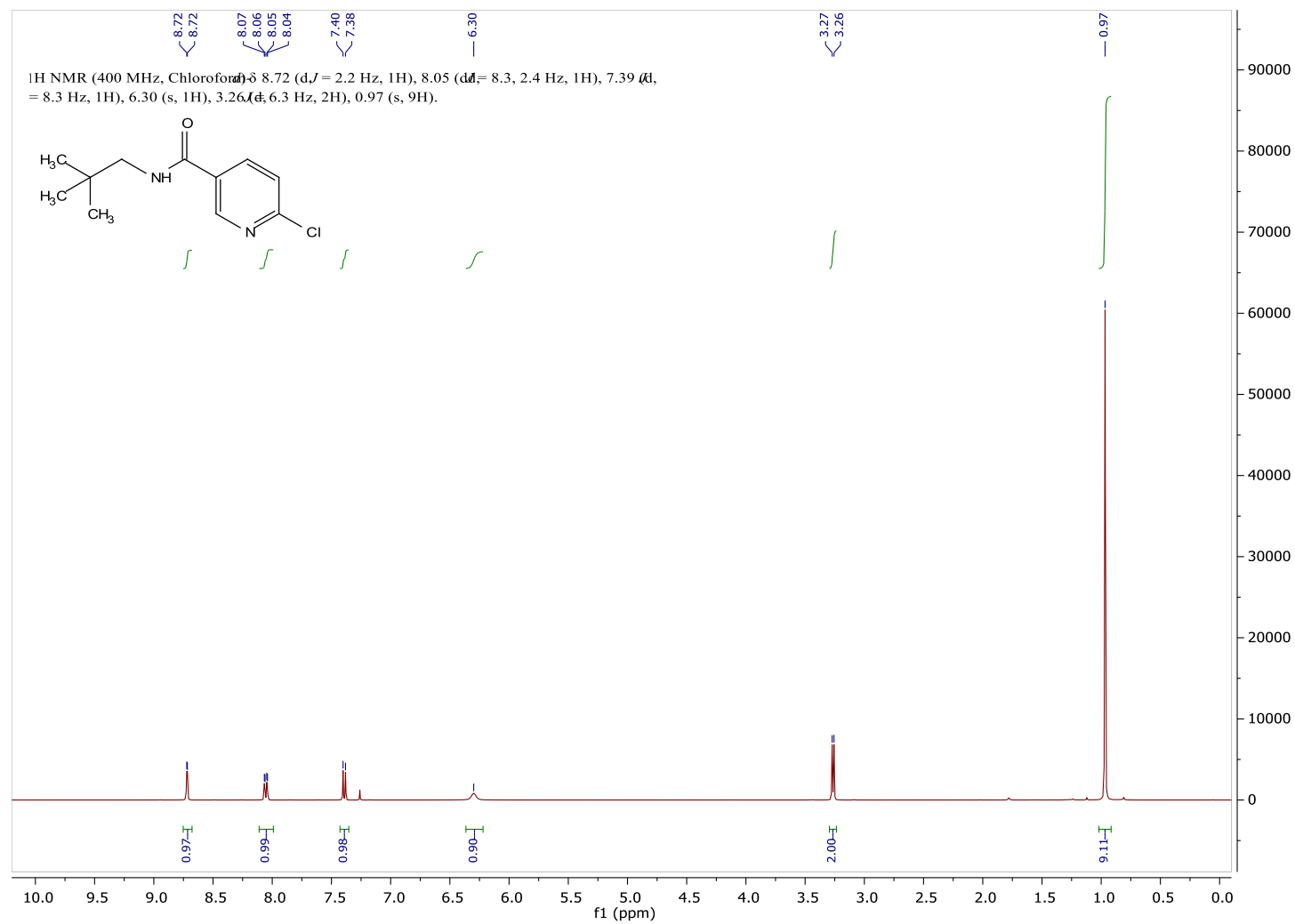
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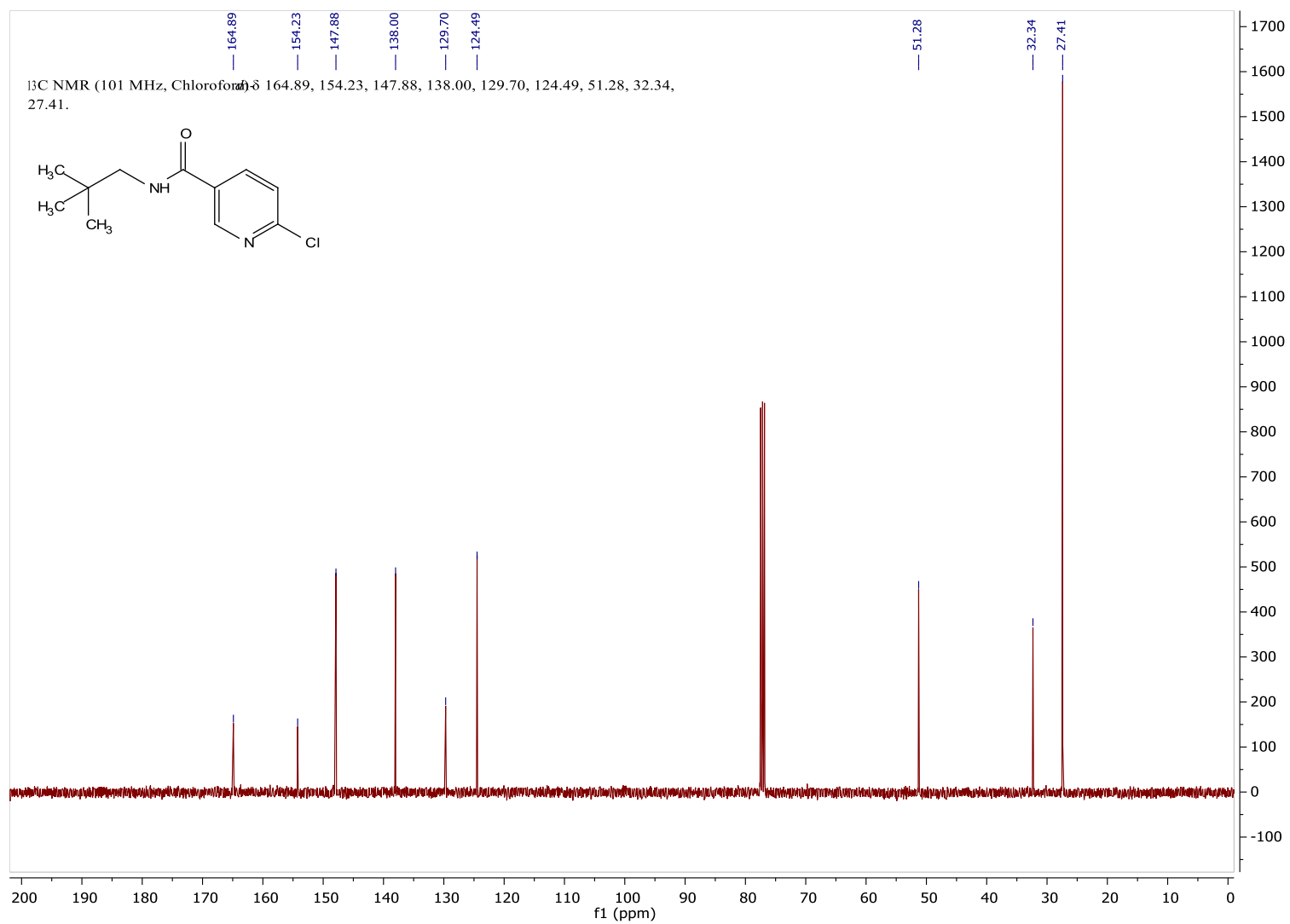
- do Nascimento, P. G.; Lemos, T. L.; Almeida, M. C.; de Souza, J. M.; Bizerra, A. M.; Santiago, G. M.; da Costa, J. G.; Coutinho, H. D., Lithocholic acid and derivatives: Antibacterial activity. *Steroids* **2015**, *104*, 8-15.
- Alam, A.; Takaguchi, Y.; Ito, H.; Yoshida, T.; Tsuboi, S., Multi-functionalization of gallic acid towards improved synthesis of alpha- and beta-DDB. *Tetrahedron* **2005**, *61* (7), 1909-1918.
- Visintin, C.; Aliev, A. E.; Riddall, D.; Baker, D.; Okuyama, M.; Hoi, P. M.; Hiley, R.; Selwood, D. L., Membrane receptor probes: solid-phase synthesis of biotin-Asp-PEG-arvanil derivatives. *Org Lett* **2005**, *7* (9), 1699-702.
- Arlt, A.; Benson, S.; Schulthoff, S.; Gabor, B.; Furstner, A., A total synthesis of spirastrellolide A methyl ester. *Chemistry* **2013**, *19* (11), 3596-608.
- Miles-Barrett, D. M.; Neal, A. R.; Hand, C.; Montgomery, J. R.; Panovic, I.; Ojo, O. S.; Lancefield, C. S.; Cordes, D. B.; Slawin, A. M.; Lebl, T.; Westwood, N. J., The synthesis and analysis of lignin-bound Hibbert ketone structures in technical lignins. *Org Biomol Chem* **2016**, *14* (42), 10023-10030.
- Mateus, C. R.; Almeida, W. P.; Coelho, F., Diastereoselective heterogeneous catalytic hydrogenation of Baylis-Hillman adducts. *Tetrahedron Lett* **2000**, *41* (15), 2533-2536.
- Gottlieb, H. E.; Kotlyar, V.; Nudelman, A., NMR Chemical Shifts of Common Laboratory Solvents as Trace Impurities. *J Org Chem* **1997**, *62* (21), 7512-7515.
- Clark, J. H.; Hyde, A. J.; Smith, D. K., Calcium fluoride-supported alkali metal fluorides. New reagents for nucleophilic fluorine transfer reactions. *Journal of the Chemical Society, Chemical Communications* **1986**, (10), 791-793.
- An, L.; Xiao, Y. L.; Min, Q. Q.; Zhang, X., Facile Access to Fluoromethylated Arenes by Nickel-Catalyzed Cross-Coupling between Arylboronic Acids and Fluoromethyl Bromide. *Angew Chem Int Ed Engl* **2015**, *54* (31), 9079-83.
- Champagne, P. A.; Benhassine, Y.; Desroches, J.; Paquin, J. F., Friedel-Crafts reaction of benzyl fluorides: selective activation of C-F bonds as enabled by hydrogen bonding. *Angew Chem Int Ed Engl* **2014**, *53* (50), 13835-9.
- Papadopoulou, M. V.; Bloomer, W. D.; Rosenzweig, H. S.; Kaiser, M., The antitrypanosomal and antitubercular activity of some nitro(triazole/imidazole)-based aromatic amines. *Eur J Med Chem* **2017**, *138*, 1106-1113.
- Xiong, H.; Hoyer, A. T.; Fan, K. H.; Li, X. M.; Clemens, J.; Horchler, C. L.; Lim, N. C.; Attardo, G., Facile Route to 2-Fluoropyridines via 2-Pyridyltrialkylammonium Salts Prepared from Pyridine N-Oxides and Application to F-18-Labeling. *Org Lett* **2015**, *17* (15), 3726-3729.

13. Jiang, J.-A.; Du, J.-L.; Liao, D.-H.; Wang, Z.-G.; Ji, Y.-F., Efficient Co(OAc)₂-catalyzed aerobic oxidation of EWG-substituted 4-cresols to access 4-hydroxybenzaldehydes. *Tetrahedron Letters* **2014**, *55* (8), 1406-1411.
14. Dordonne, S.; Crousse, B.; Bonnet-Delpon, D.; Legros, J., Fluorous tagging of DABCO through halogen bonding: recyclable catalyst for the Morita-Baylis-Hillman reaction (vol 47, pg 5855, 2011). *Chem Commun* **2011**, *47* (48), 12890-12890.
15. Jensen, A. B.; Lindhardt, A. T., Decarboxylative Trichloromethylation of Aromatic Aldehydes and Its Applications in Continuous Flow. *Journal of Organic Chemistry* **2014**, *79* (3), 1174-1183.
16. Castro, L. C. M.; Bézier, D.; Sortais, J.-B.; Darcel, C., Iron Dihydride Complex as the Pre-catalyst for Efficient Hydrosilylation of Aldehydes and Ketones Under Visible Light Activation. *Adv Synth Catal* **2011**, *353* (8), 1279-1284.
17. Pauff, S. M.; Miller, S. C., A trifluoroacetic acid-labile sulfonate protecting group and its use in the synthesis of a near-IR fluorophore. *J Org Chem* **2013**, *78* (2), 711-6.
18. Folléas, B. t.; Marek, I.; Normant, J.-F.; Saint-Jalmes, L., Fluoroform: an Efficient Precursor for the Trifluoromethylation of Aldehydes. *Tetrahedron* **2000**, *56* (2), 275-283.
19. Okusu, S.; Hirano, K.; Yasuda, Y.; Tokunaga, E.; Shibata, N., Flow trifluoromethylation of carbonyl compounds by Ruppert-Prakash reagent and its application for pharmaceuticals, efavirenz and HSD-016. *RSC Advances* **2016**, *6* (86), 82716-82720.

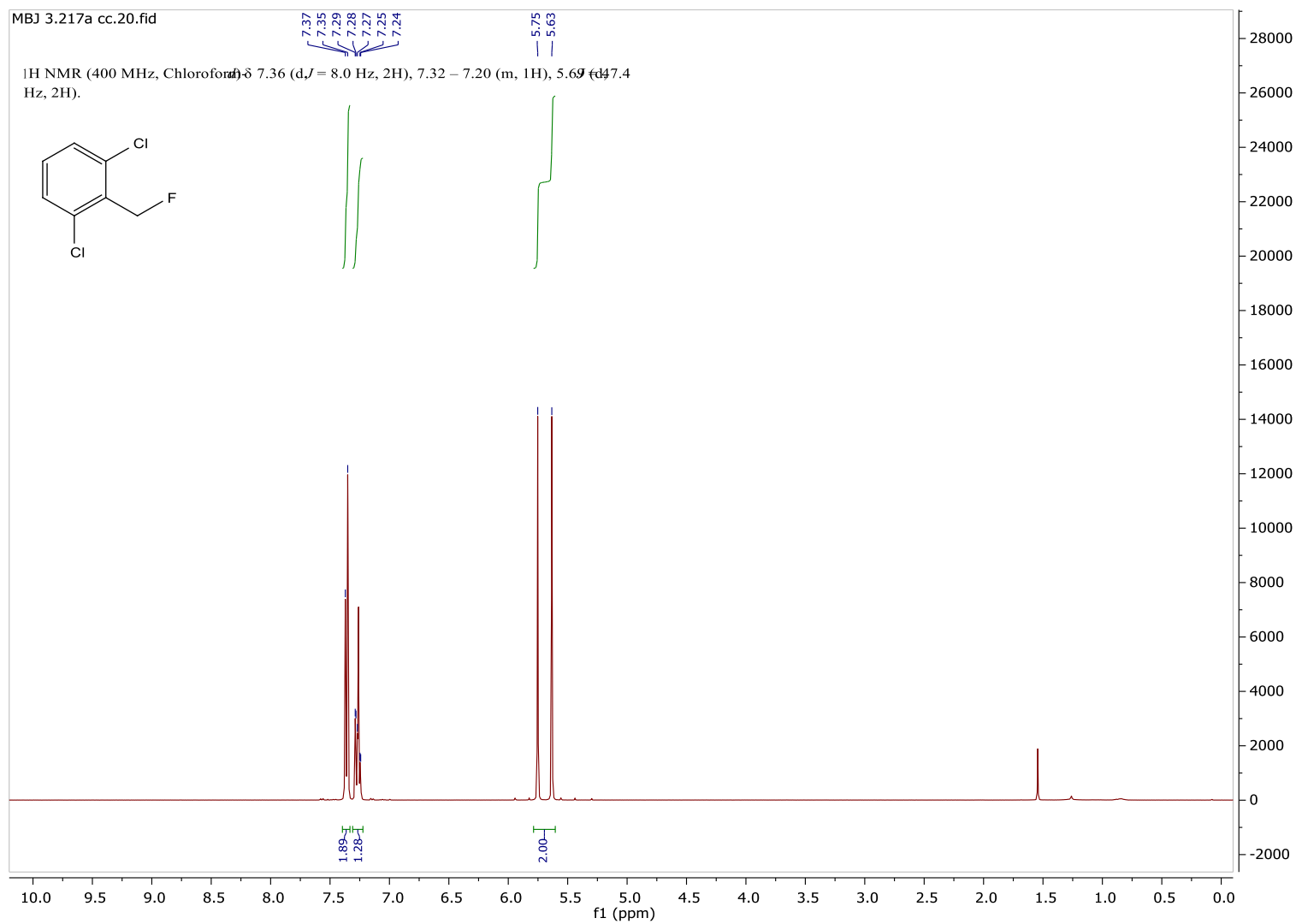
NMR Spectra

6-Chloro-N-neopentylnicotinamide



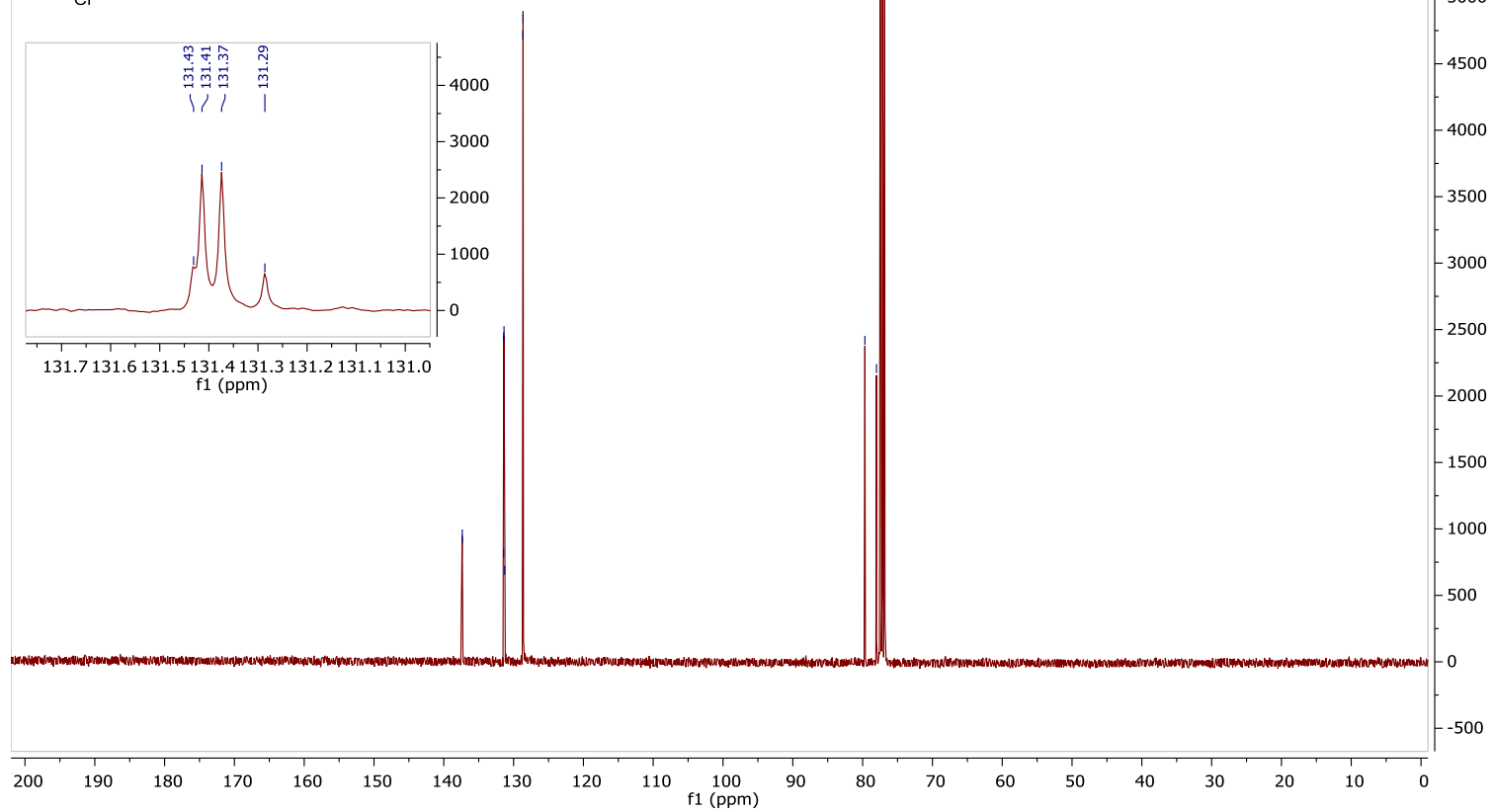
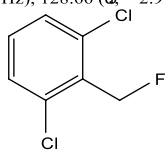


1-Fluoromethyl-2,6-dichlorobenzene (1) from 1-bromomethyl-2,6-dichlorobenzene using CsF-CaF₂.

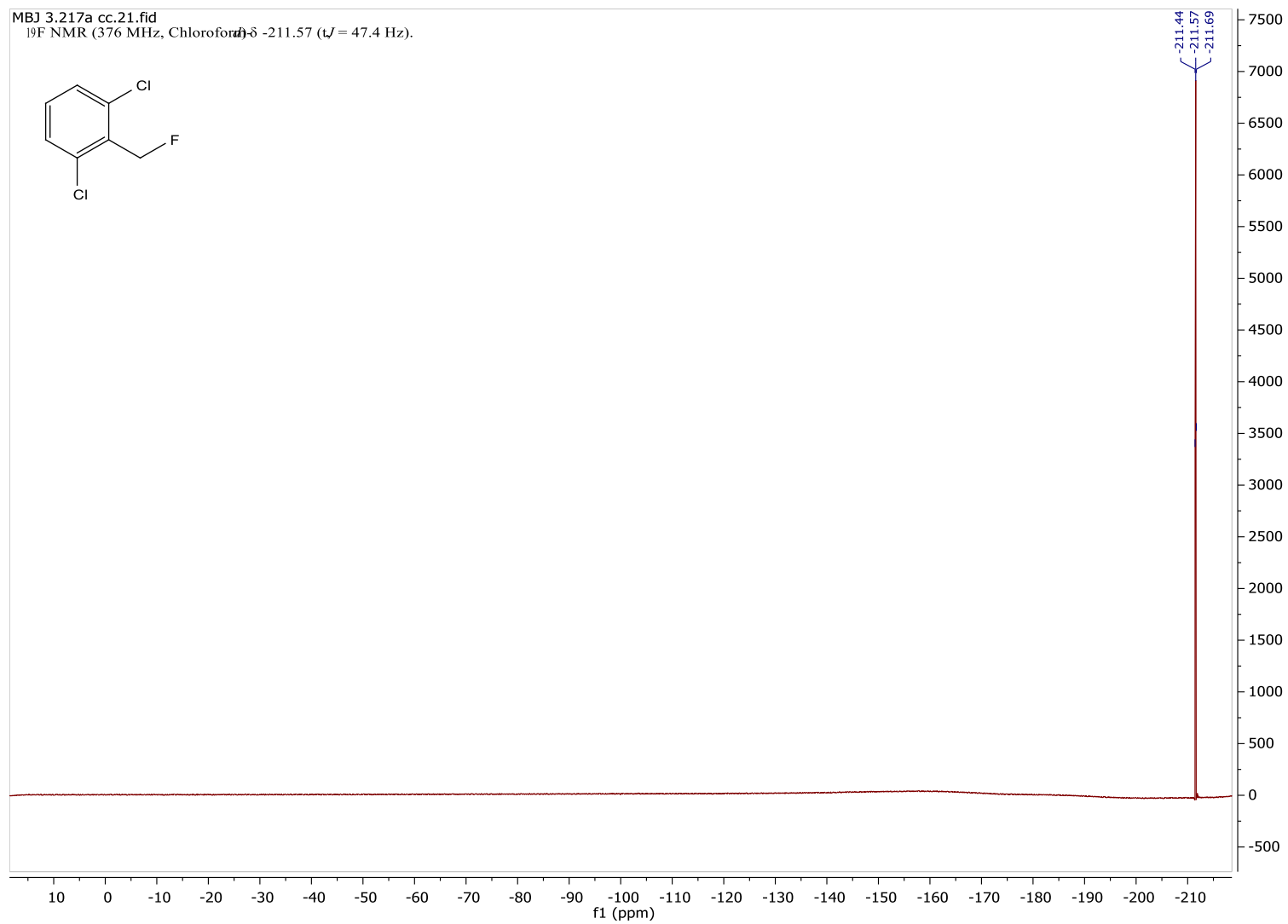
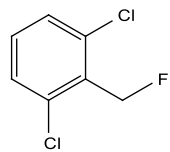


MBJ 3.217a cc.22.fid

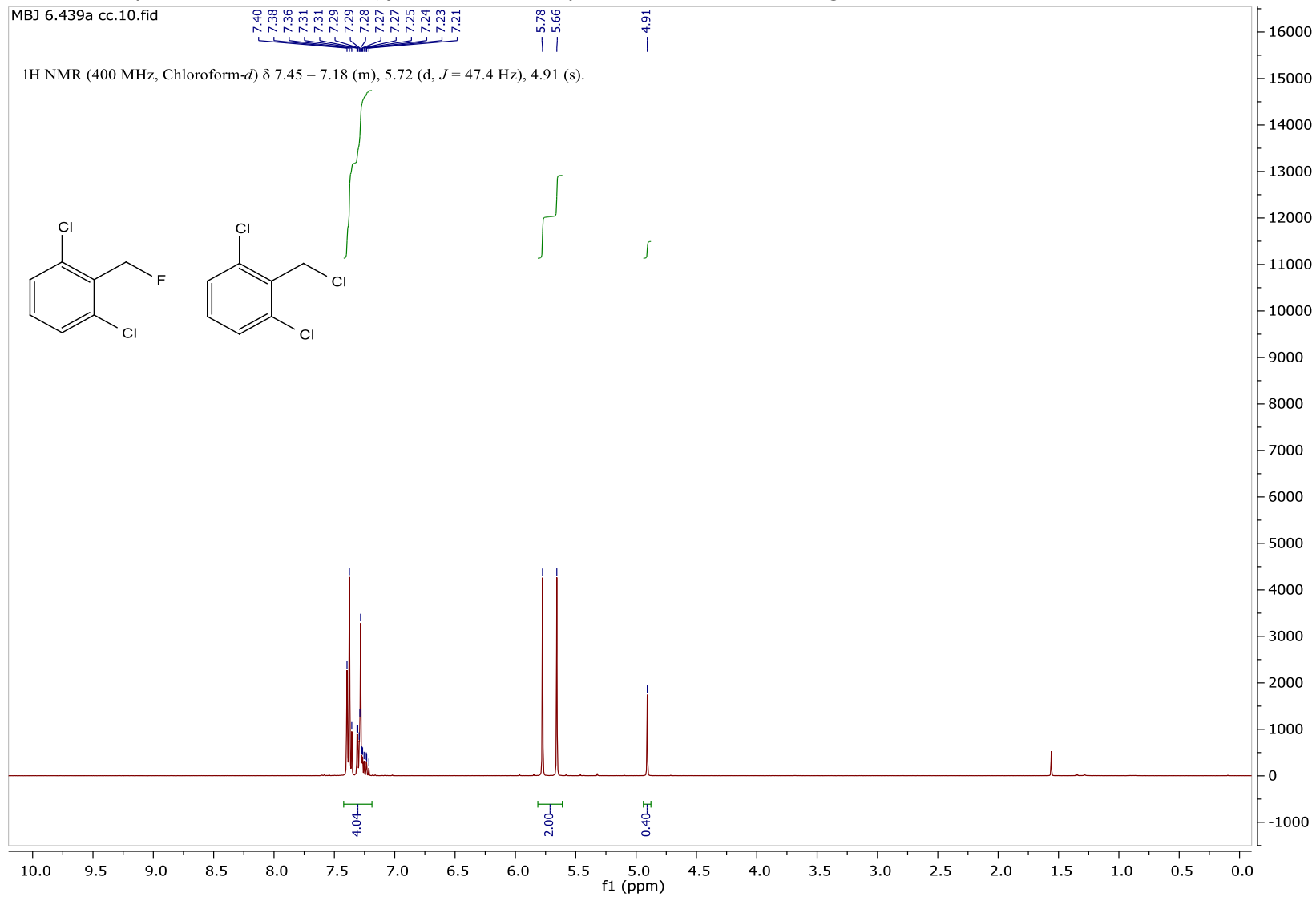
^{13}C NMR (101 MHz, Chloroform- d_3) δ 137.37 ($d_f = 3.8$ Hz), 131.37 ($d_f = 4.0$ Hz), 131.36 ($d_f = 14.6$ Hz), 128.66 ($d_f = 2.9$ Hz), 78.84 ($d_f = 166.7$ Hz).



MBJ 3.217a cc.21.fid

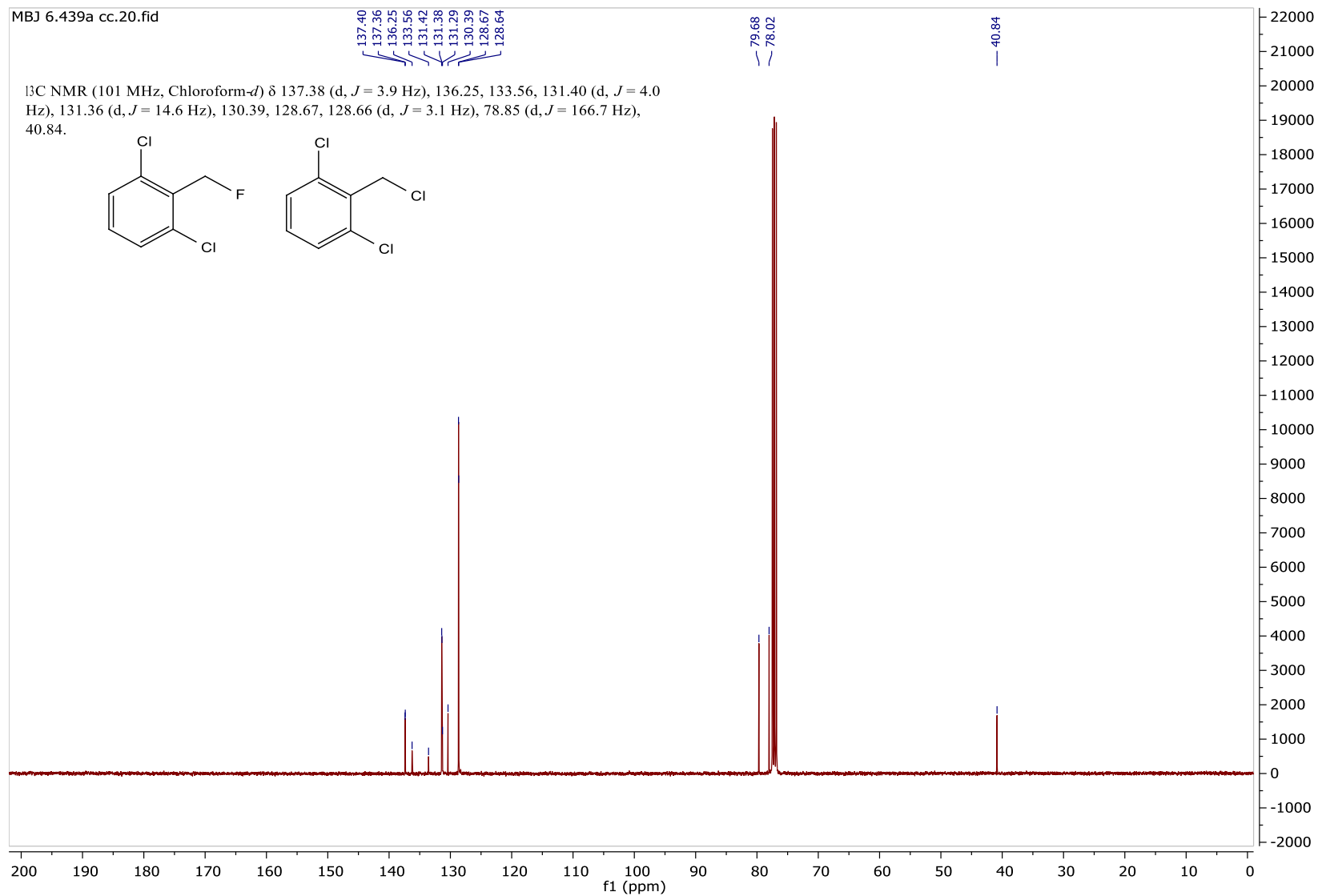
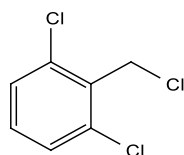
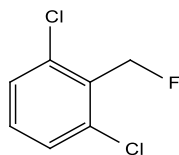
 ^{19}F NMR (376 MHz, Chloroform- d_3) δ -211.57 (t , J = 47.4 Hz).

1-Fluoromethyl-2,6-dichlorobenzene (**1**) from 1-chloromethyl-2,6-dichlorobenzene using CsF-CaF₂.

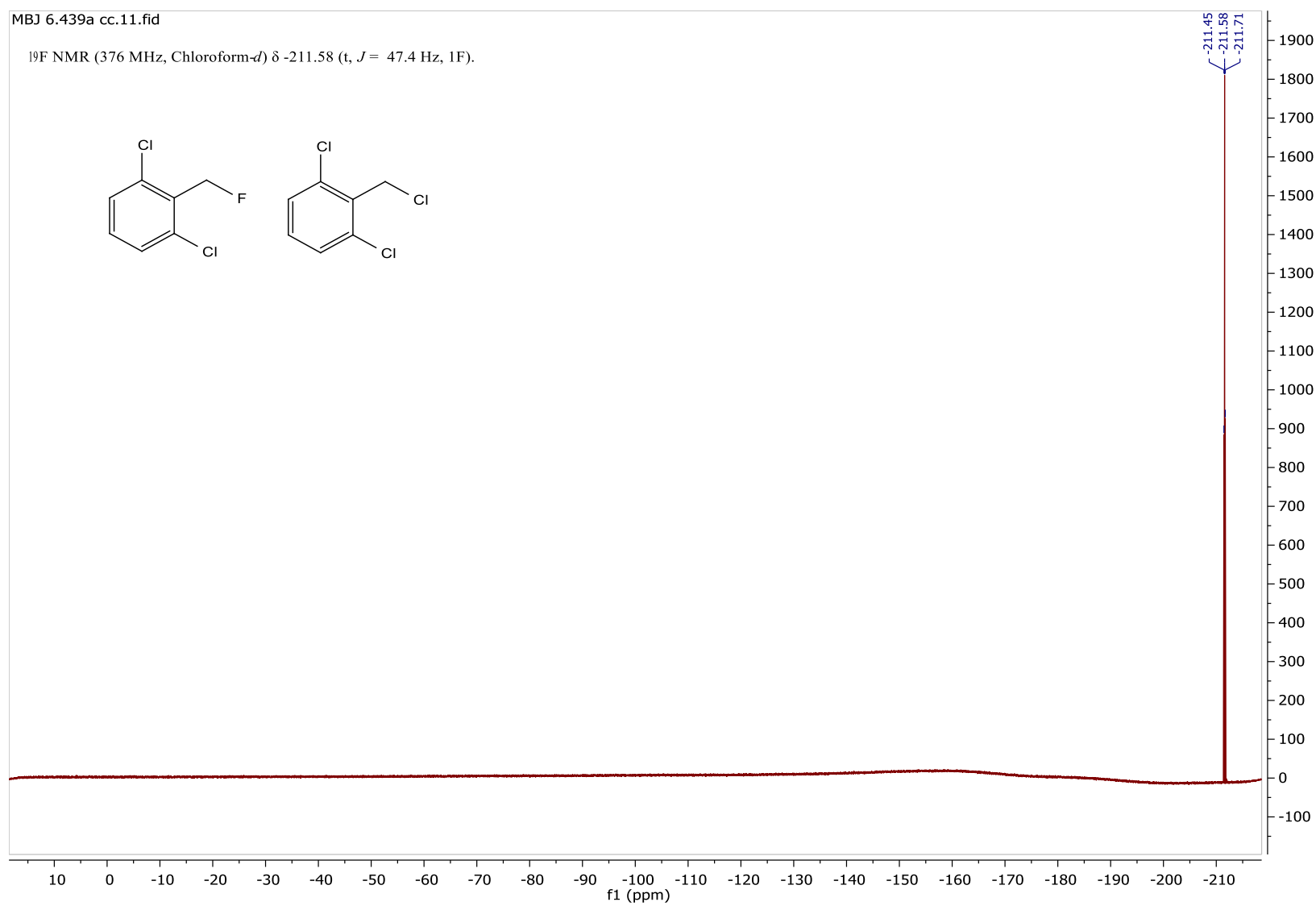
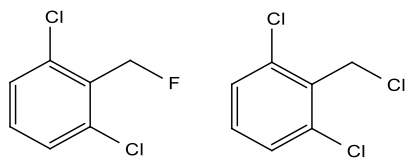


MBJ 6.439a cc.20.fid

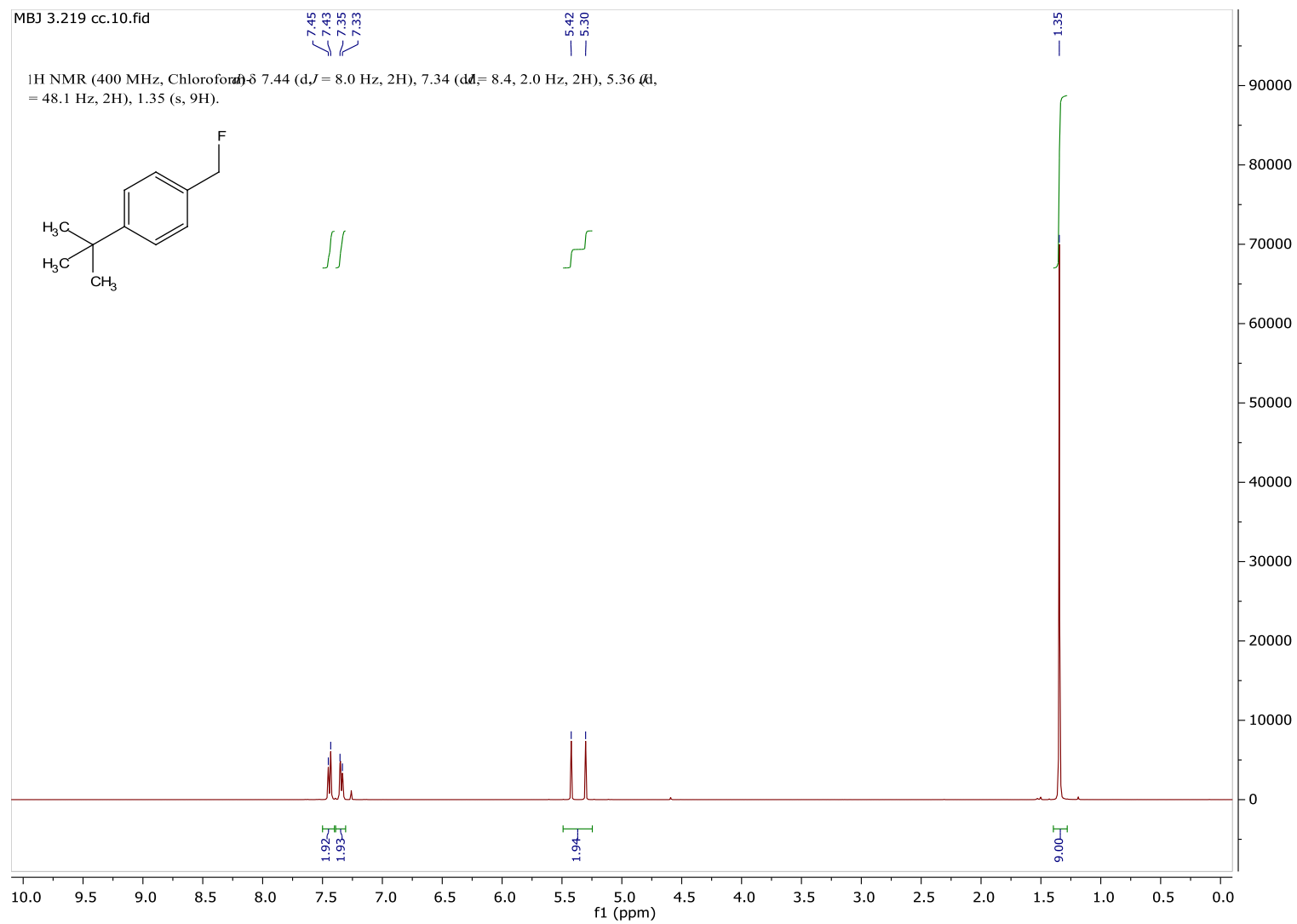
^{13}C NMR (101 MHz, Chloroform- d) δ 137.38 (d, $J = 3.9$ Hz), 136.25, 133.56, 131.40 (d, $J = 4.0$ Hz), 131.36 (d, $J = 14.6$ Hz), 130.39, 128.67, 128.66 (d, $J = 3.1$ Hz), 78.85 (d, $J = 166.7$ Hz), 40.84.



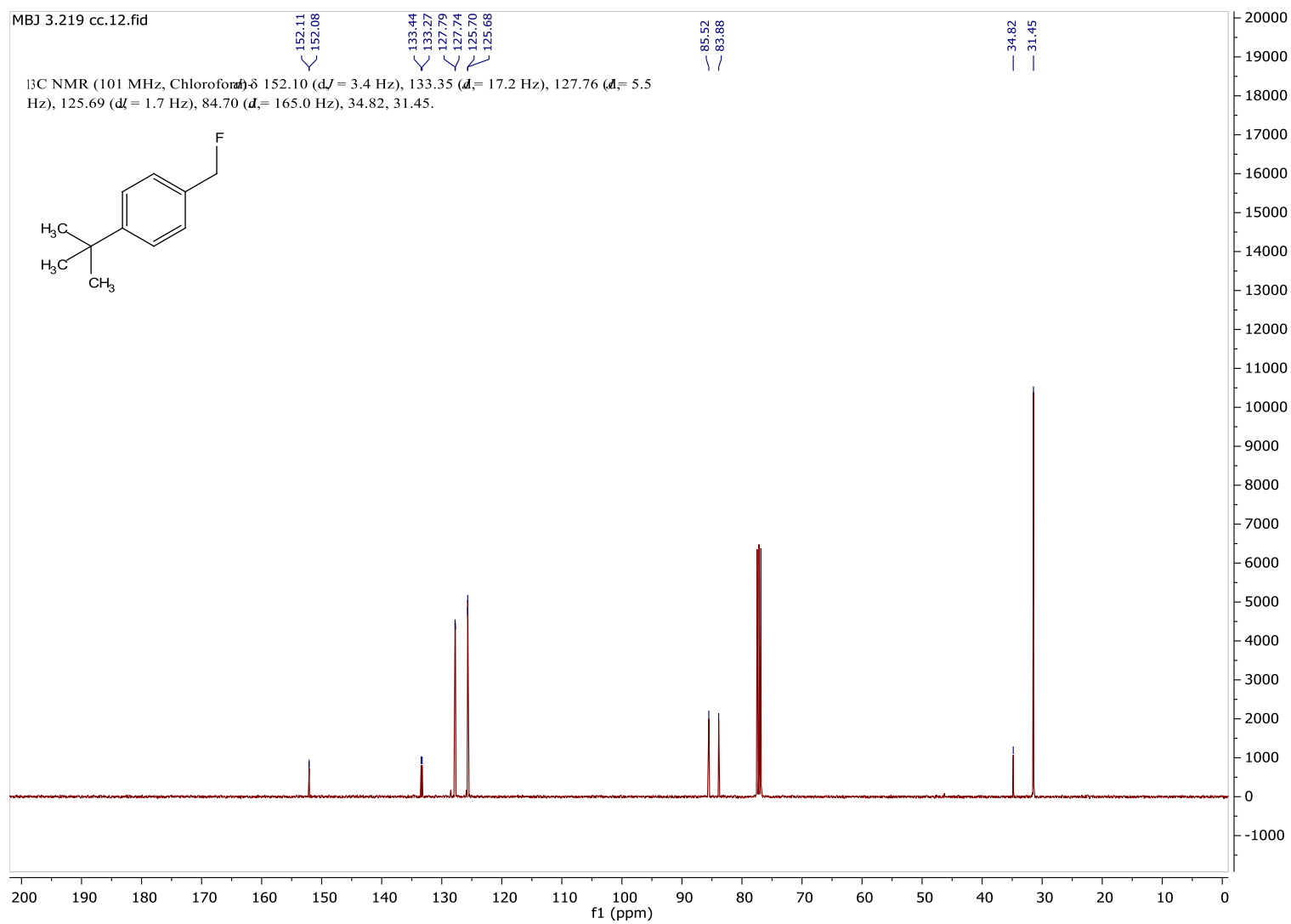
MBJ 6.439a cc.11.fid

 ^{19}F NMR (376 MHz, Chloroform- d) δ -211.58 (t, J = 47.4 Hz, 1F).

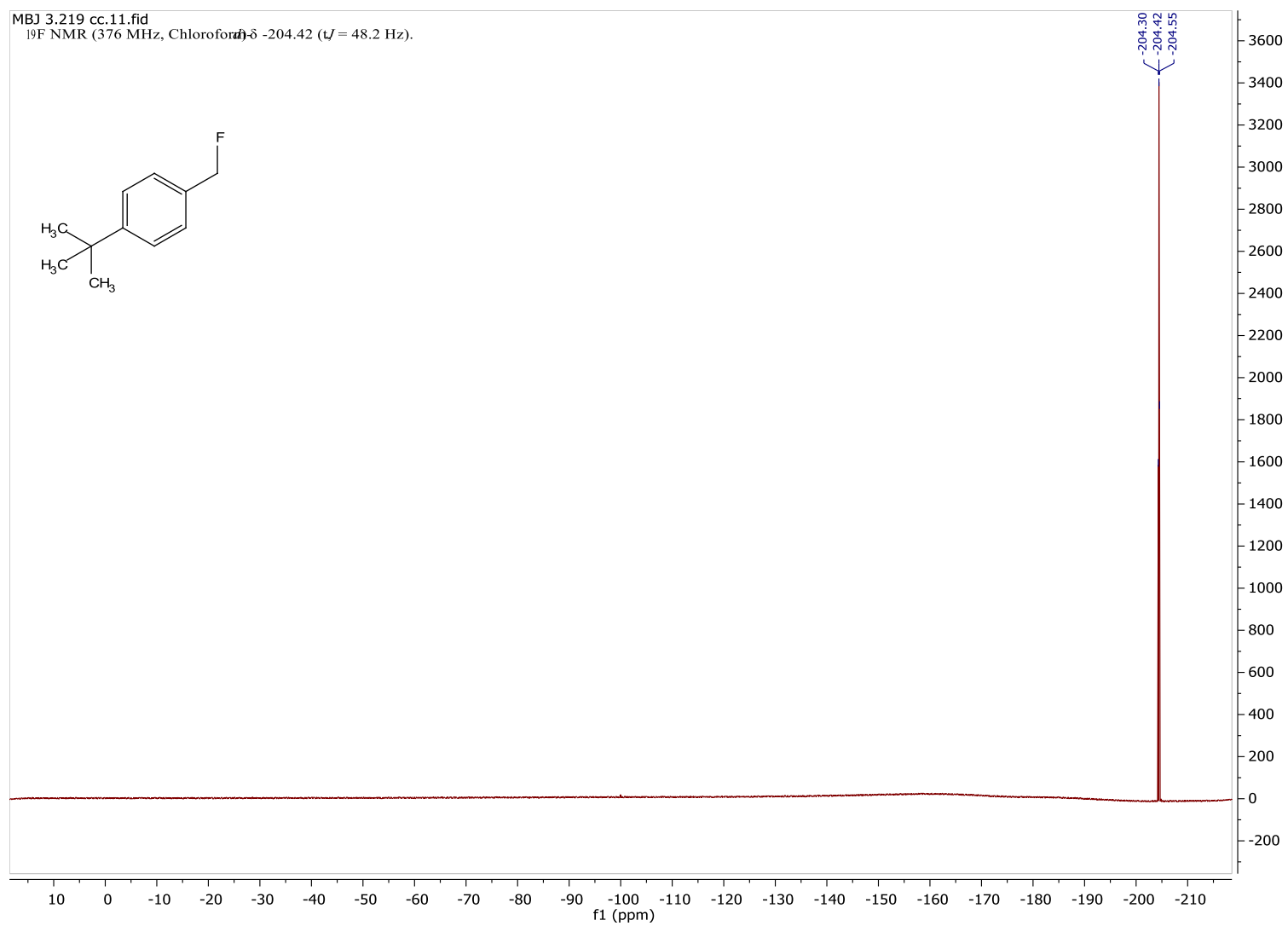
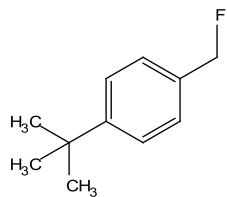
1-Fluoromethyl-4-tertbutylbenzene (2) from 1-bromomethyl-4-tertbutylbenzene using CsF-CaF₂.



MBJ 3.219 cc.12.fid



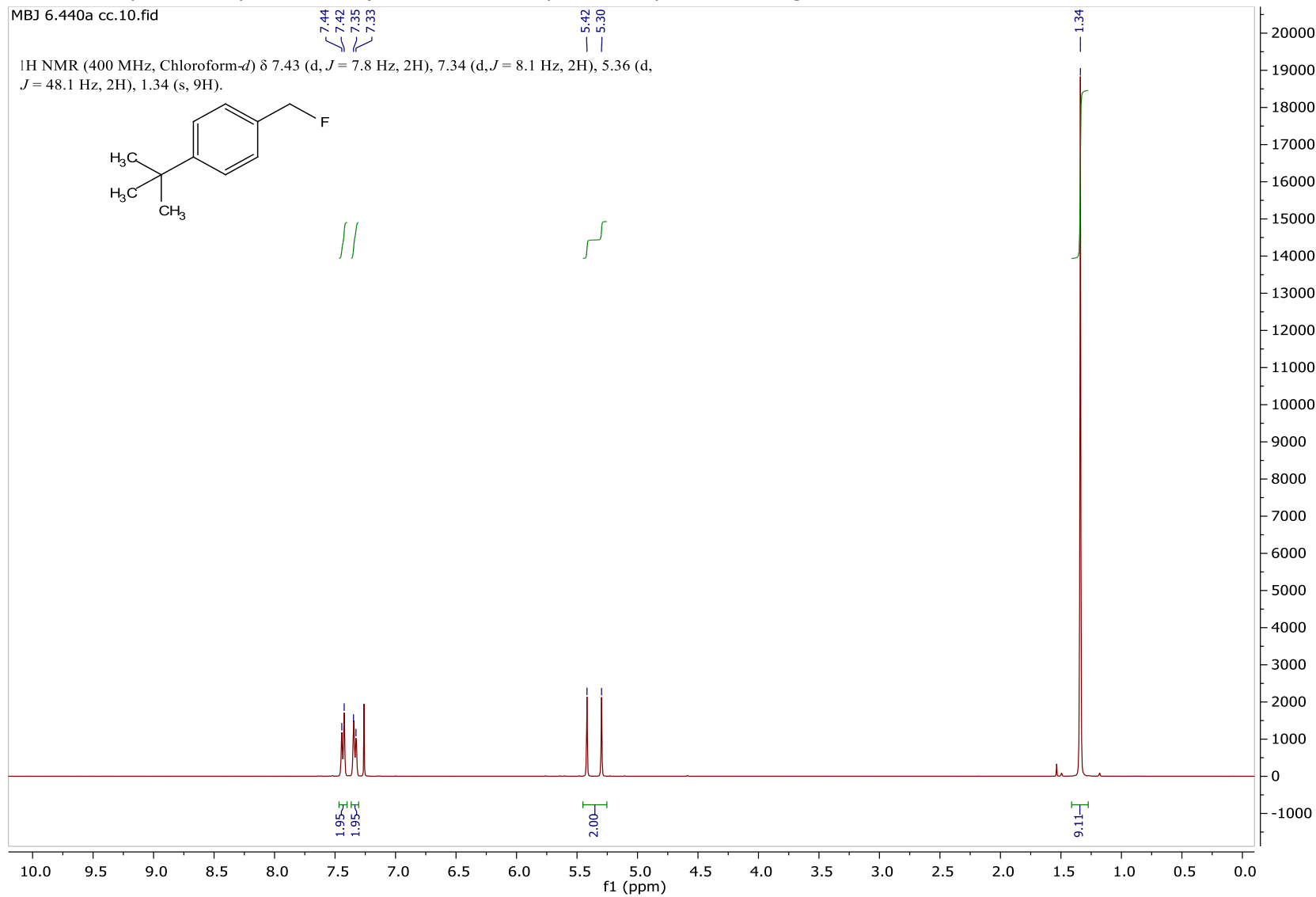
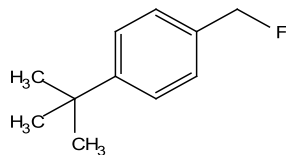
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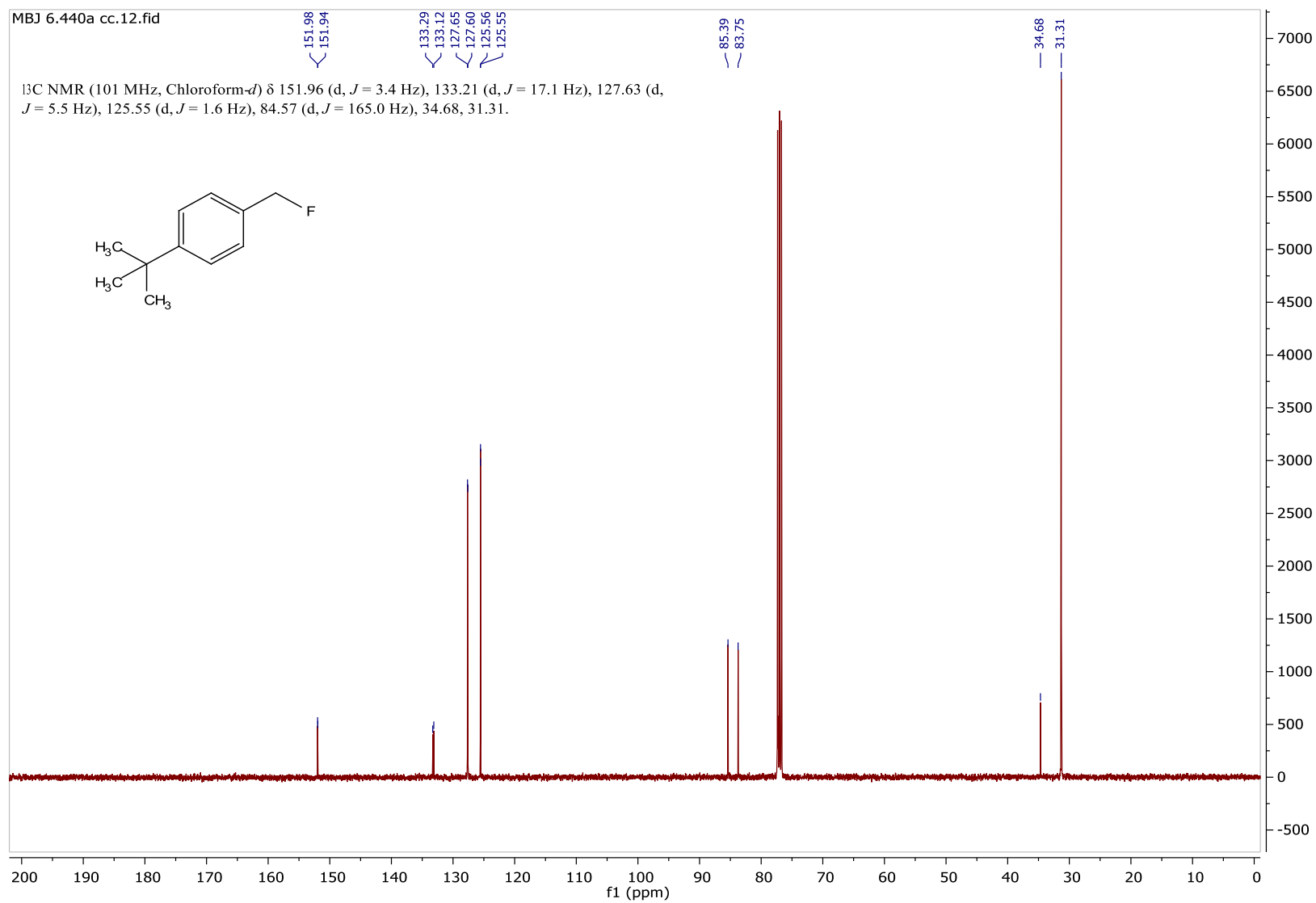
 ^{19}F NMR (376 MHz, Chloroform- d_3) δ -204.42 (t = 48.2 Hz).

1-Fluoromethyl-4-tertbutylbenzene (2) from 1-chloromethyl-4-tertbutylbenzene using CsF-CaF₂.

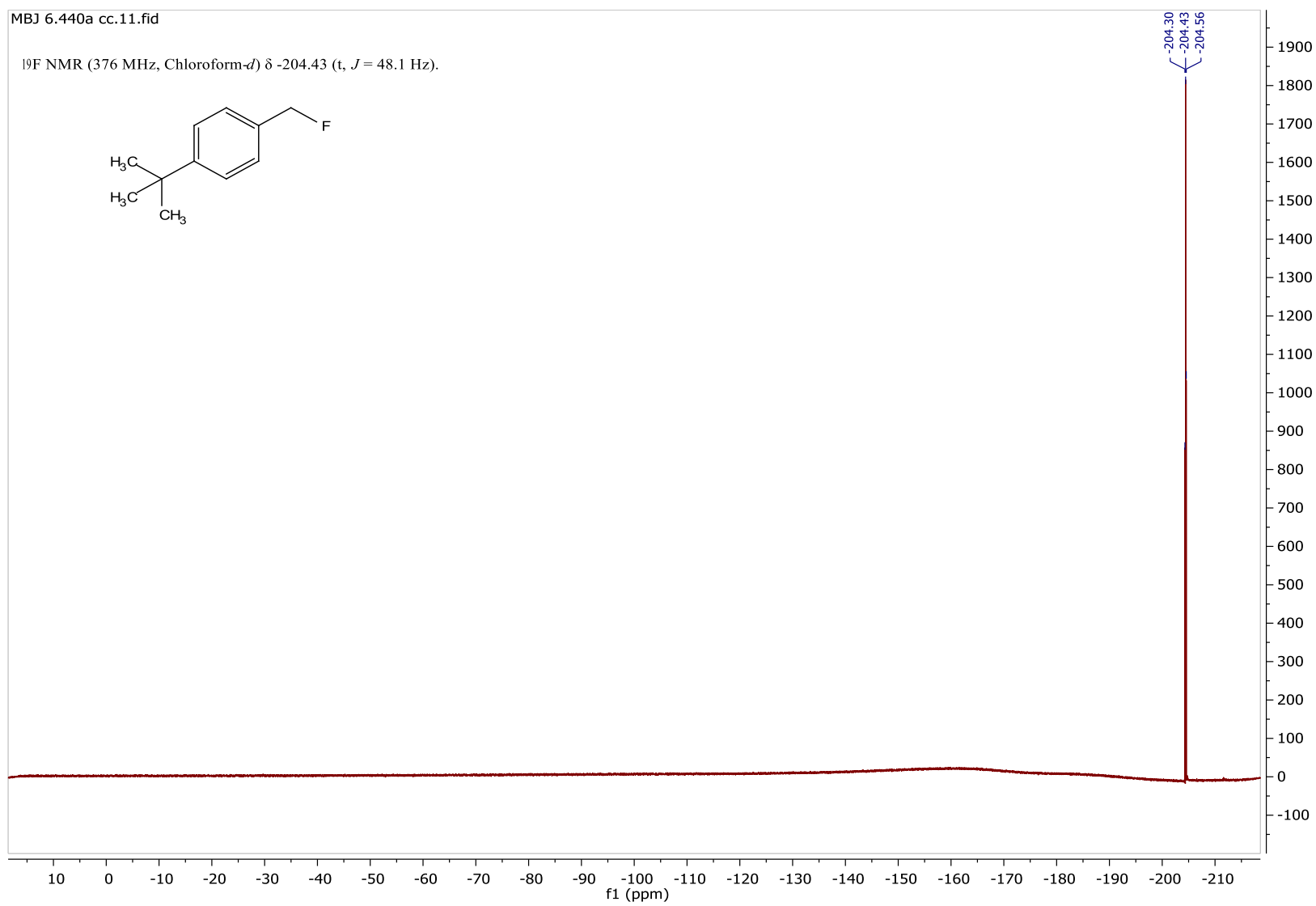
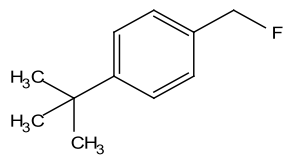
MBJ 6.440a cc.10.fid

¹H NMR (400 MHz, Chloroform-*d*) δ 7.43 (d, J = 7.8 Hz, 2H), 7.34 (d, J = 8.1 Hz, 2H), 5.36 (d, J = 48.1 Hz, 2H), 1.34 (s, 9H).



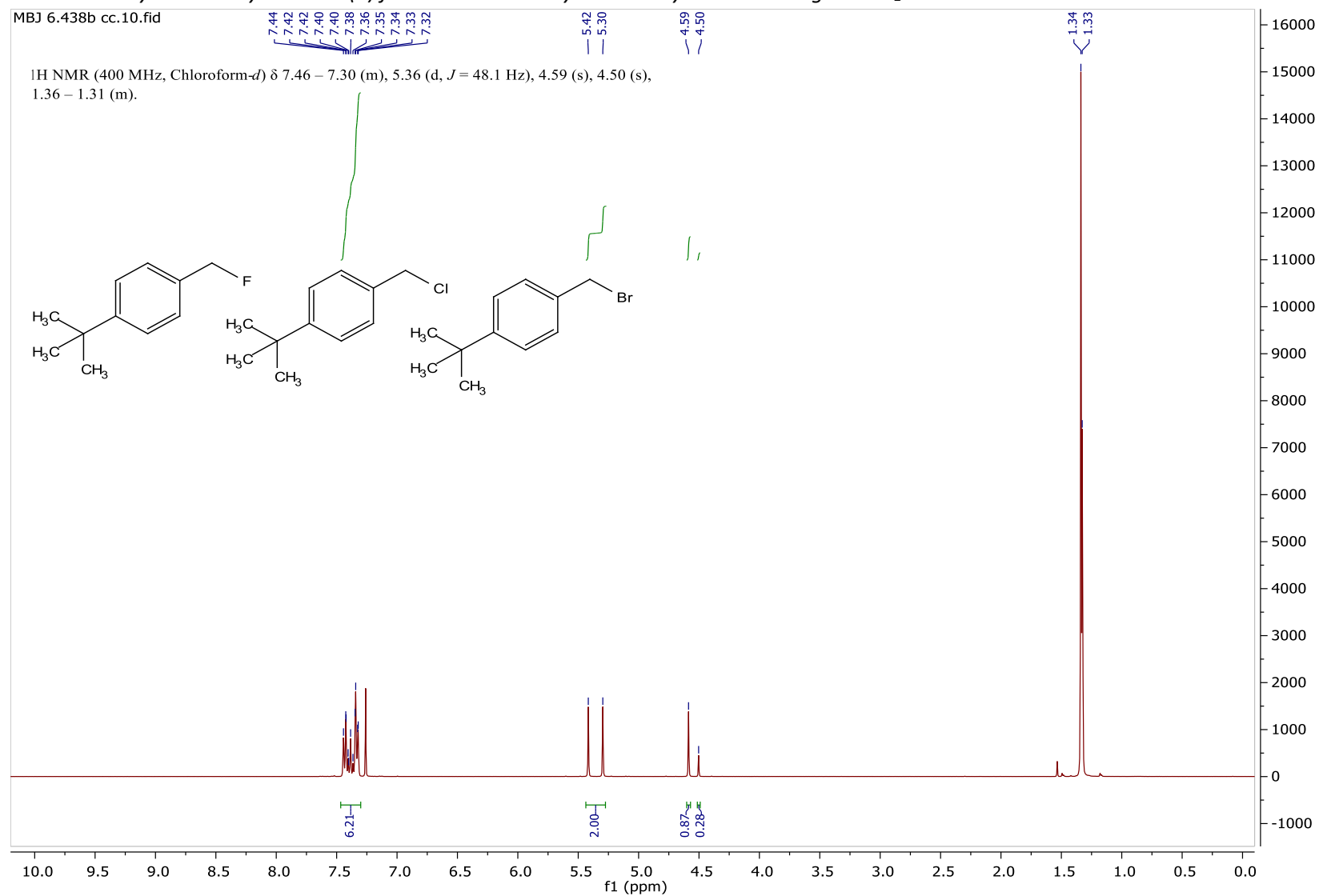


MBJ 6.440a cc.11.fid

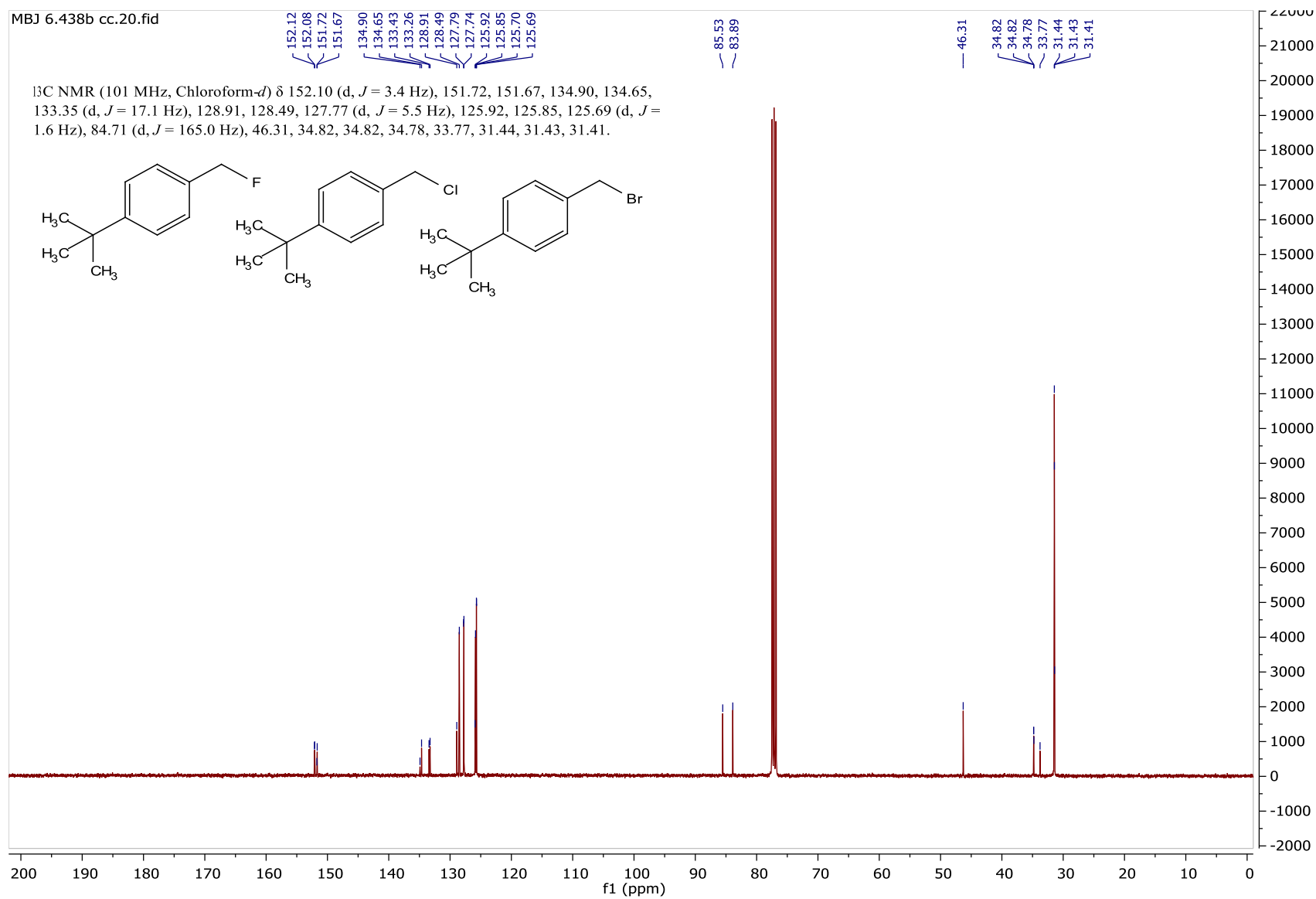
 ^{19}F NMR (376 MHz, Chloroform- d) δ -204.43 (t, $J = 48.1$ Hz).

1-Fluoromethyl-4-tertbutylbenzene (2) from 1-bromomethyl-4-tertbutylbenzene using KF-CaF₂.

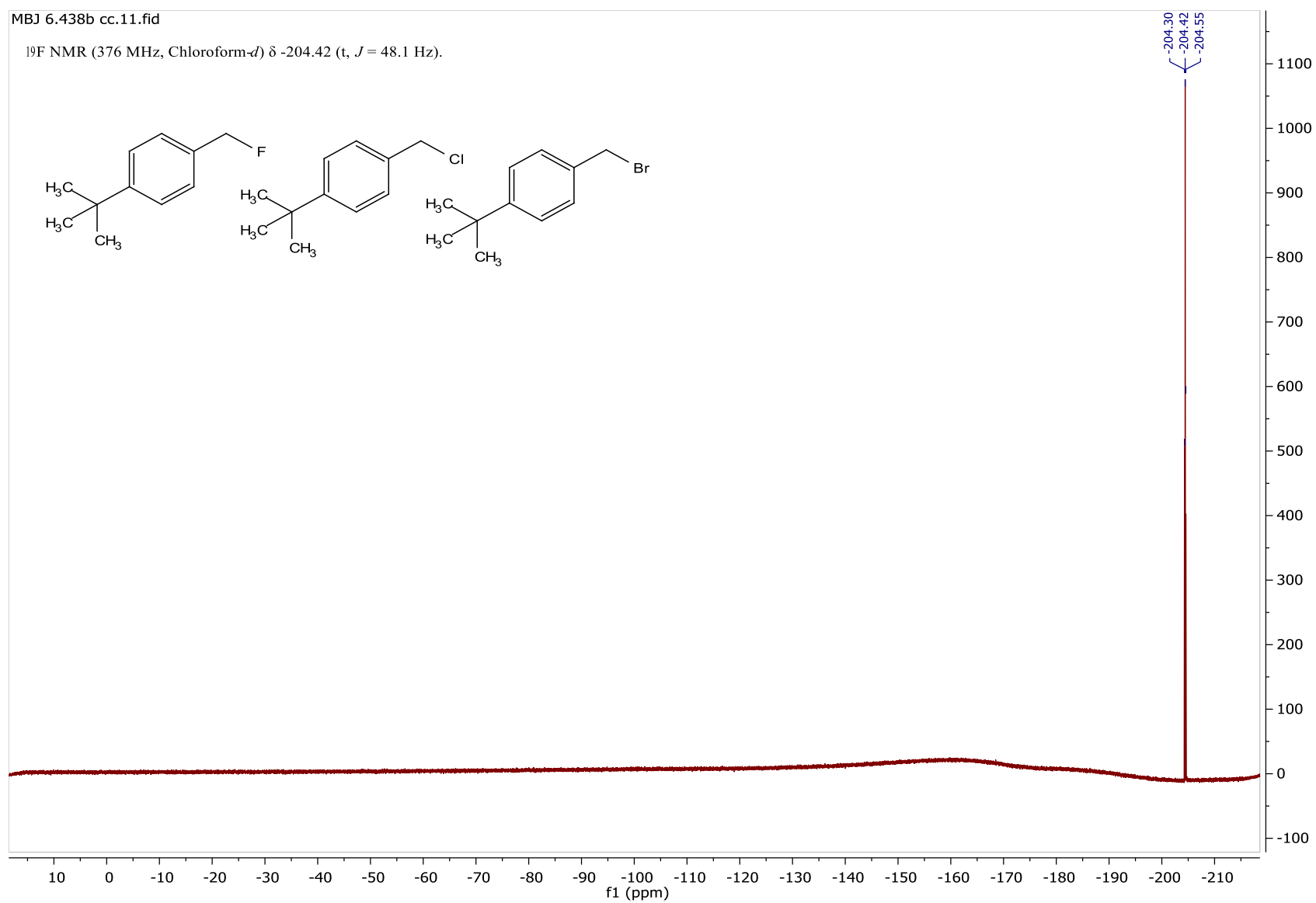
MBJ 6.438b cc.10.fid



MBJ 6.438b cc.20.fid



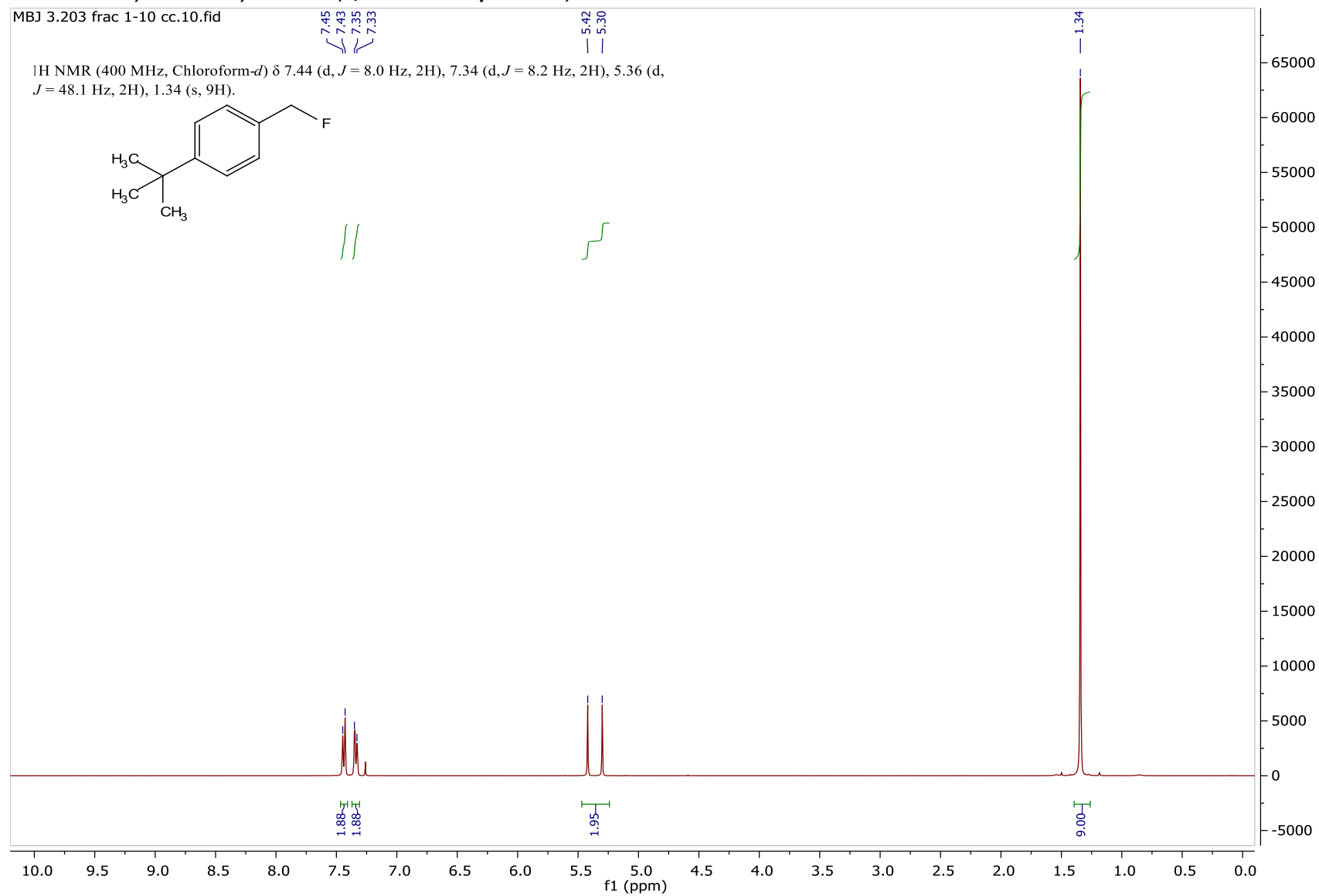
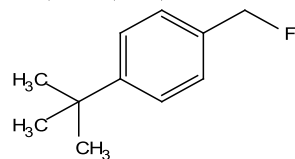
MBJ 6.438b cc.11.fid

 ^{19}F NMR (376 MHz, Chloroform-*d*) δ -204.42 (t, $J = 48.1$ Hz).

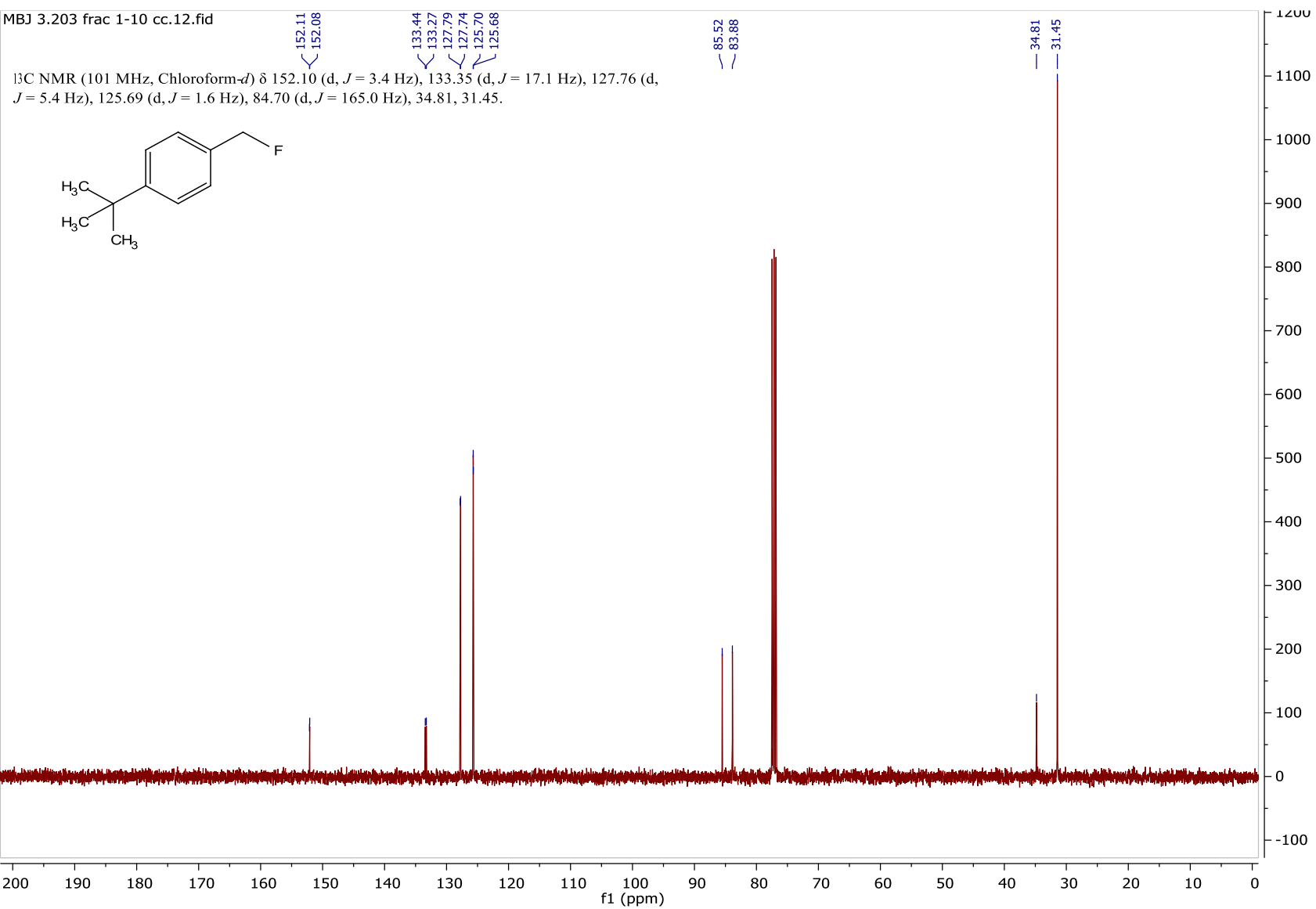
1-Fluoromethyl-4-tertbutylbenzene (2, Scale-out experiment).

MBJ 3.203 frac 1-10 cc.10.fid

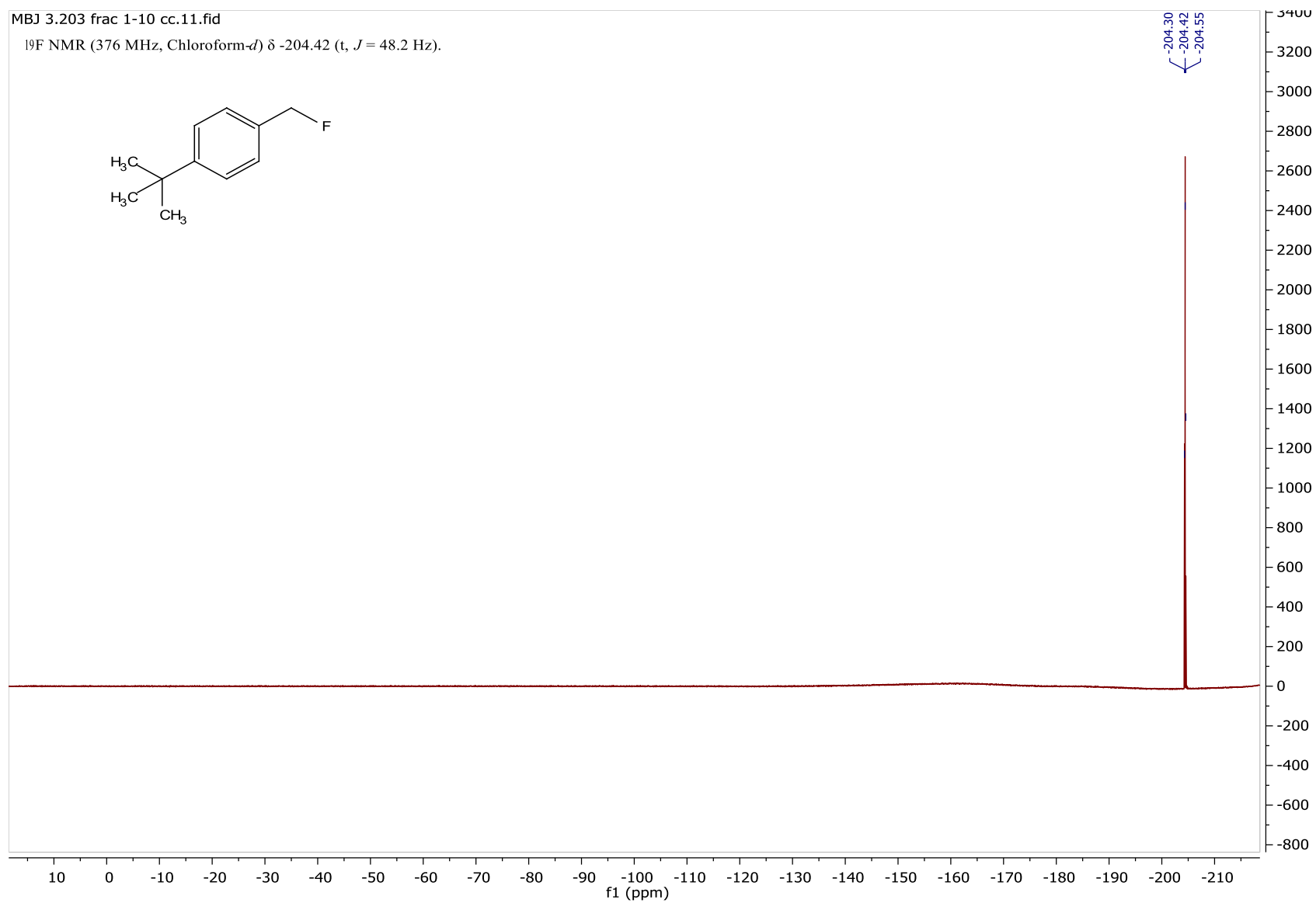
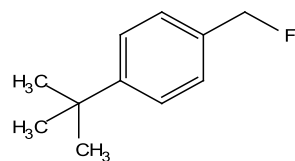
¹H NMR (400 MHz, Chloroform-*d*) δ 7.44 (d, $J = 8.0$ Hz, 2H), 7.34 (d, $J = 8.2$ Hz, 2H), 5.36 (d, $J = 48.1$ Hz, 2H), 1.34 (s, 9H).

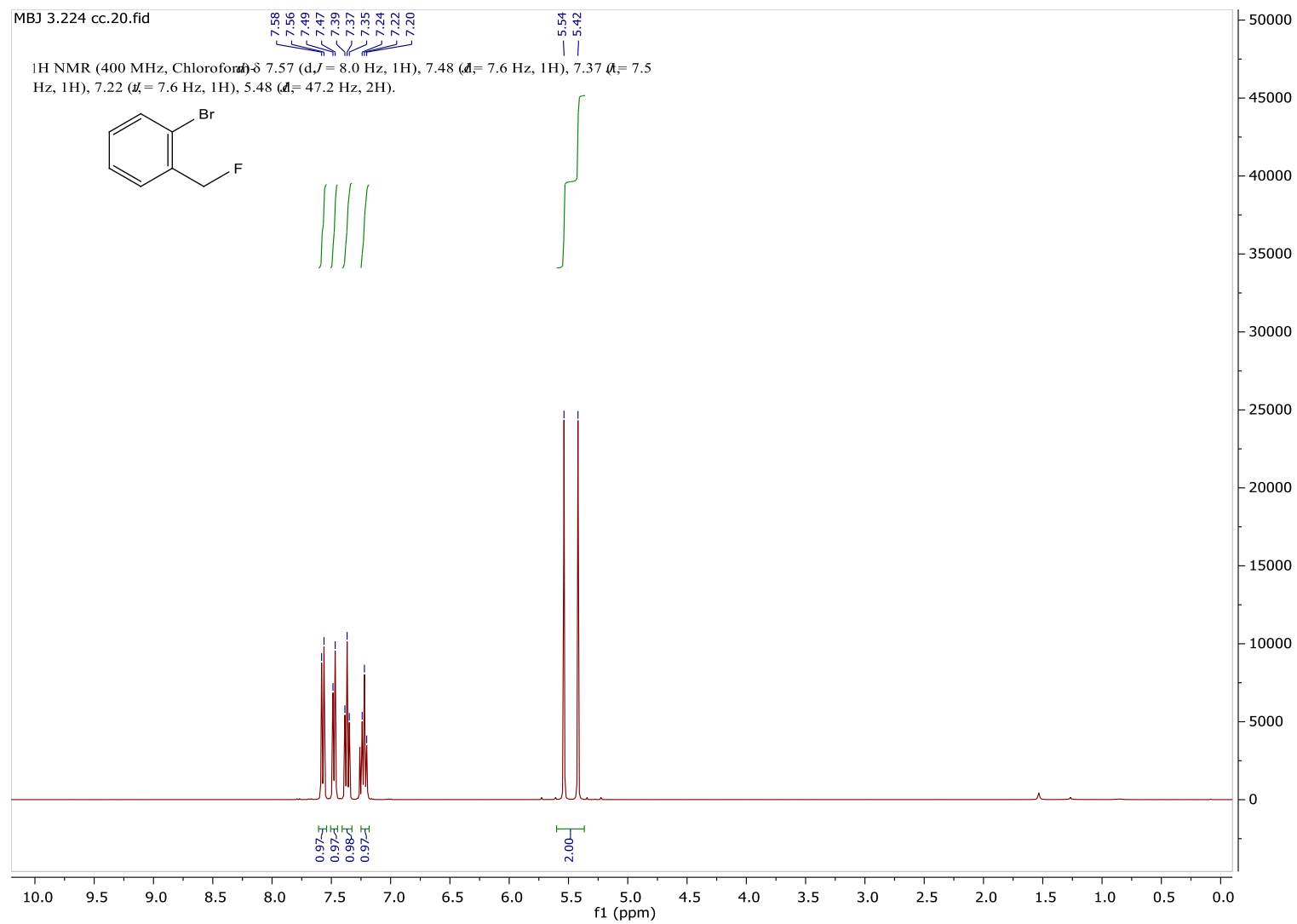


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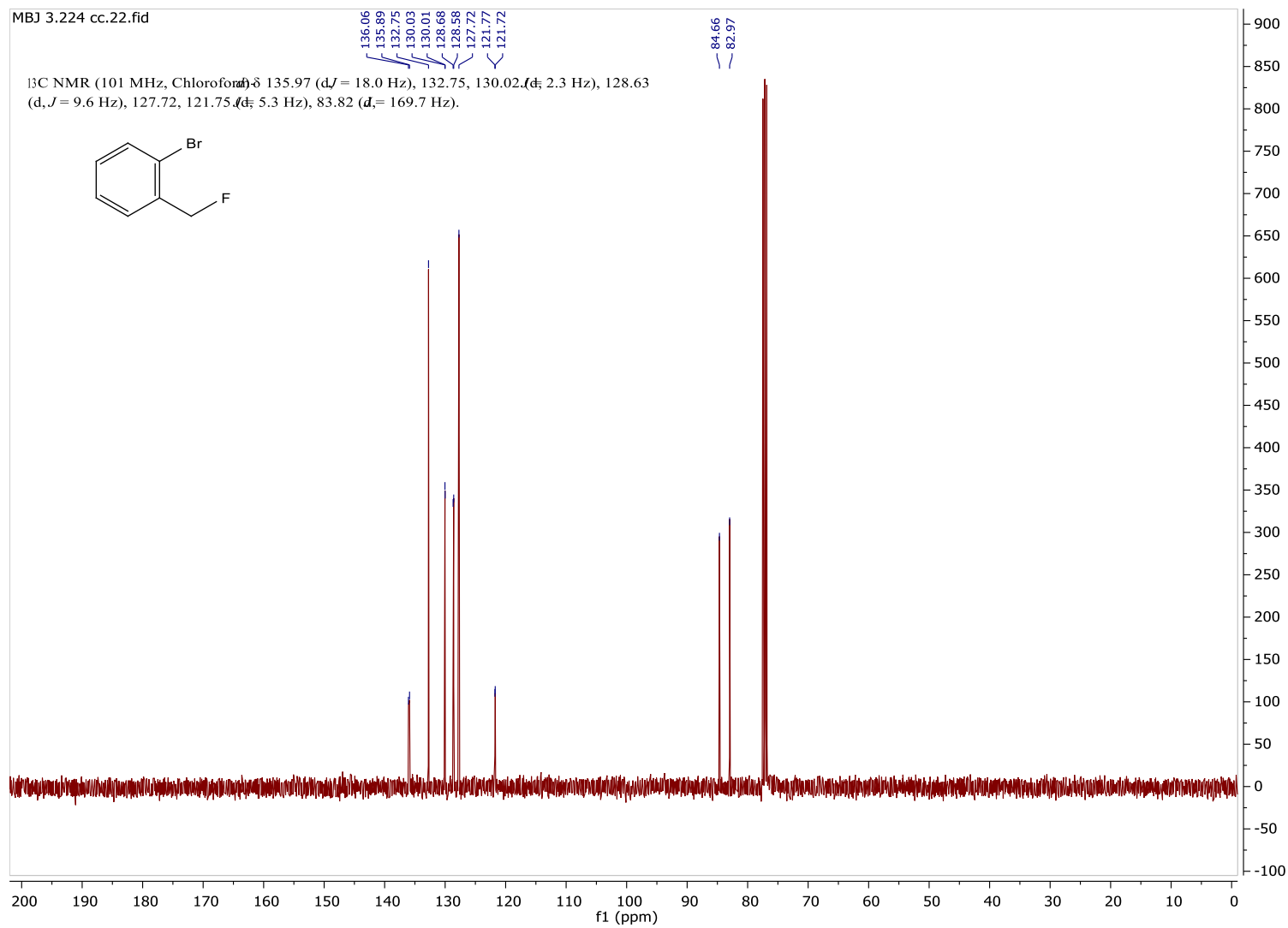
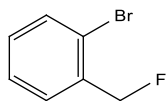
MBJ 3.203 frac 1-10 cc.11.fid

 ^{19}F NMR (376 MHz, Chloroform-*d*) δ -204.42 (t, $J = 48.2$ Hz).

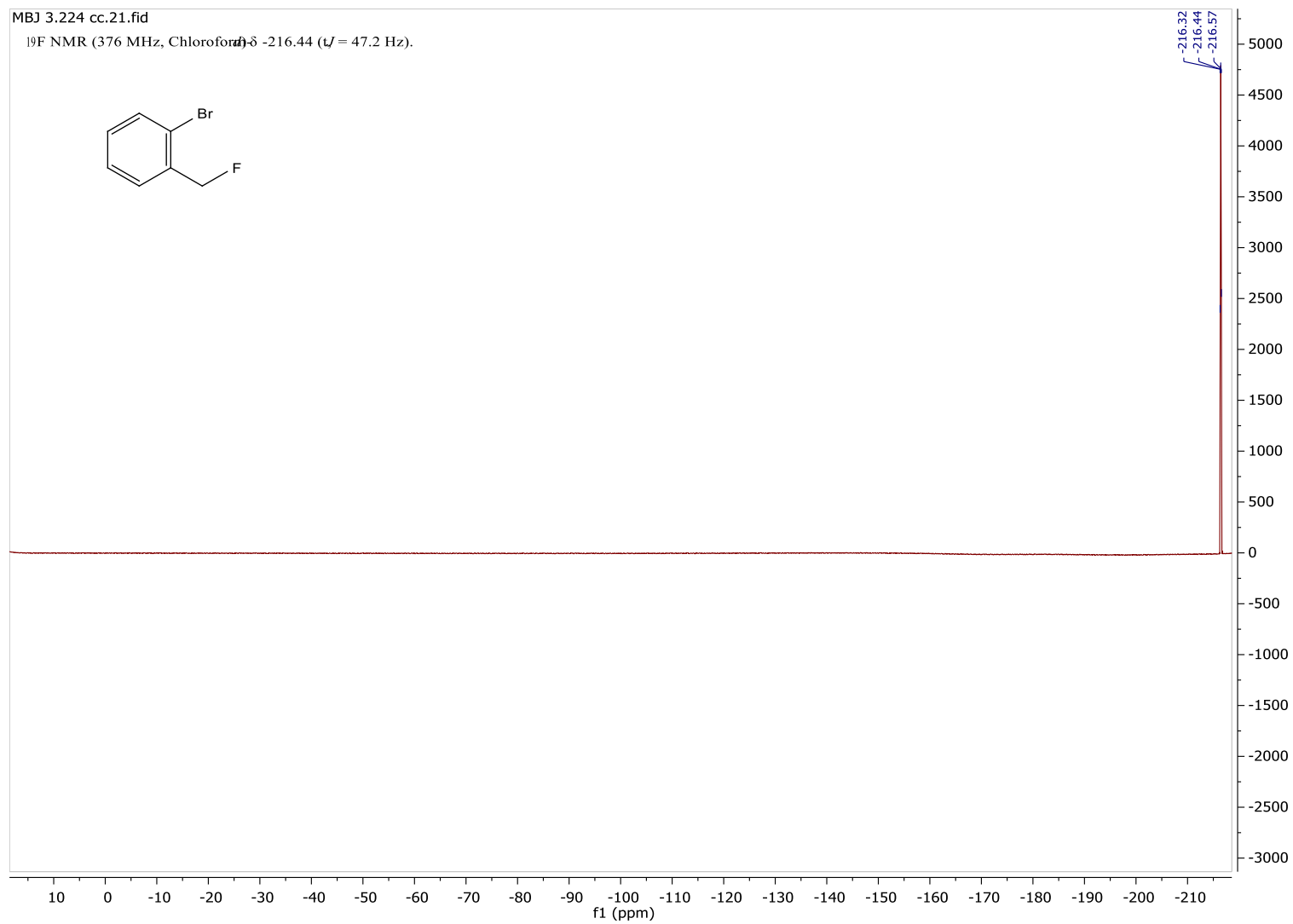
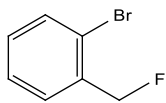
1-Fluoromethyl-2-bromobenzene (3)

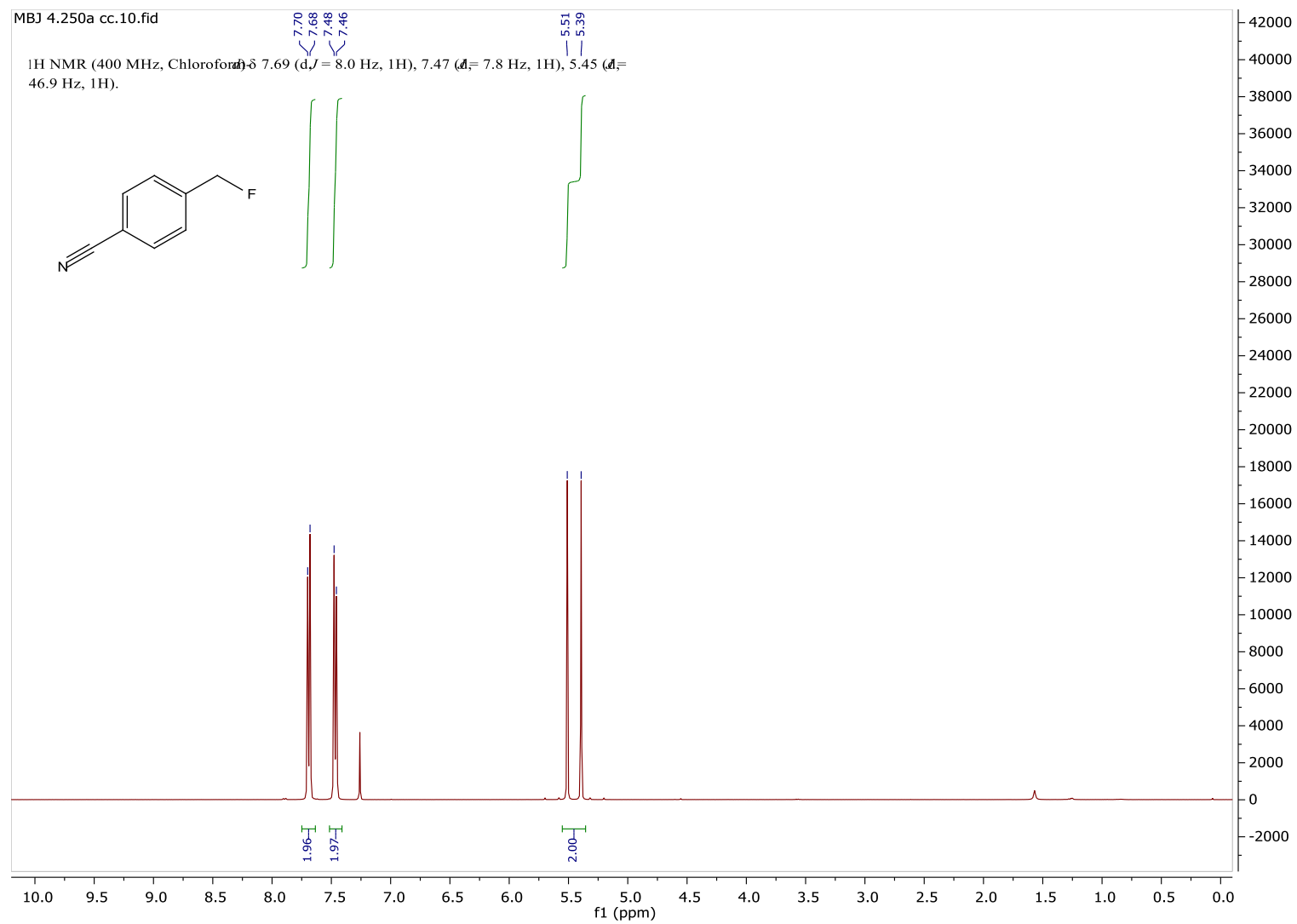
MBJ 3.224 cc.22.fid

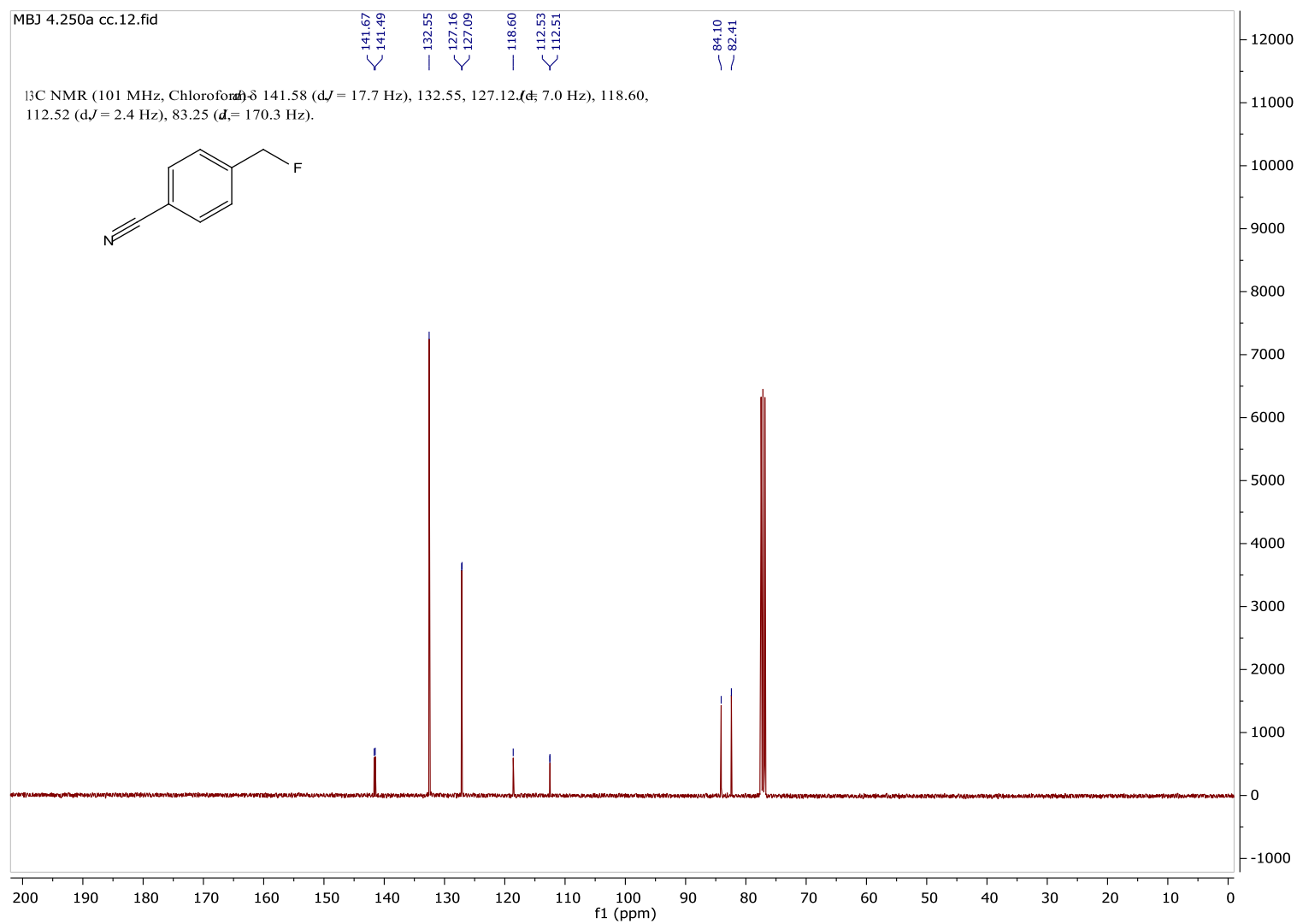
^{13}C NMR (101 MHz, Chloroform- d_3) δ 135.97 (d, $J = 18.0$ Hz), 132.75, 130.02, 128.63 (d, $J = 9.6$ Hz), 127.72, 121.75 (d, $J = 5.3$ Hz), 83.82 (d, $J = 169.7$ Hz).



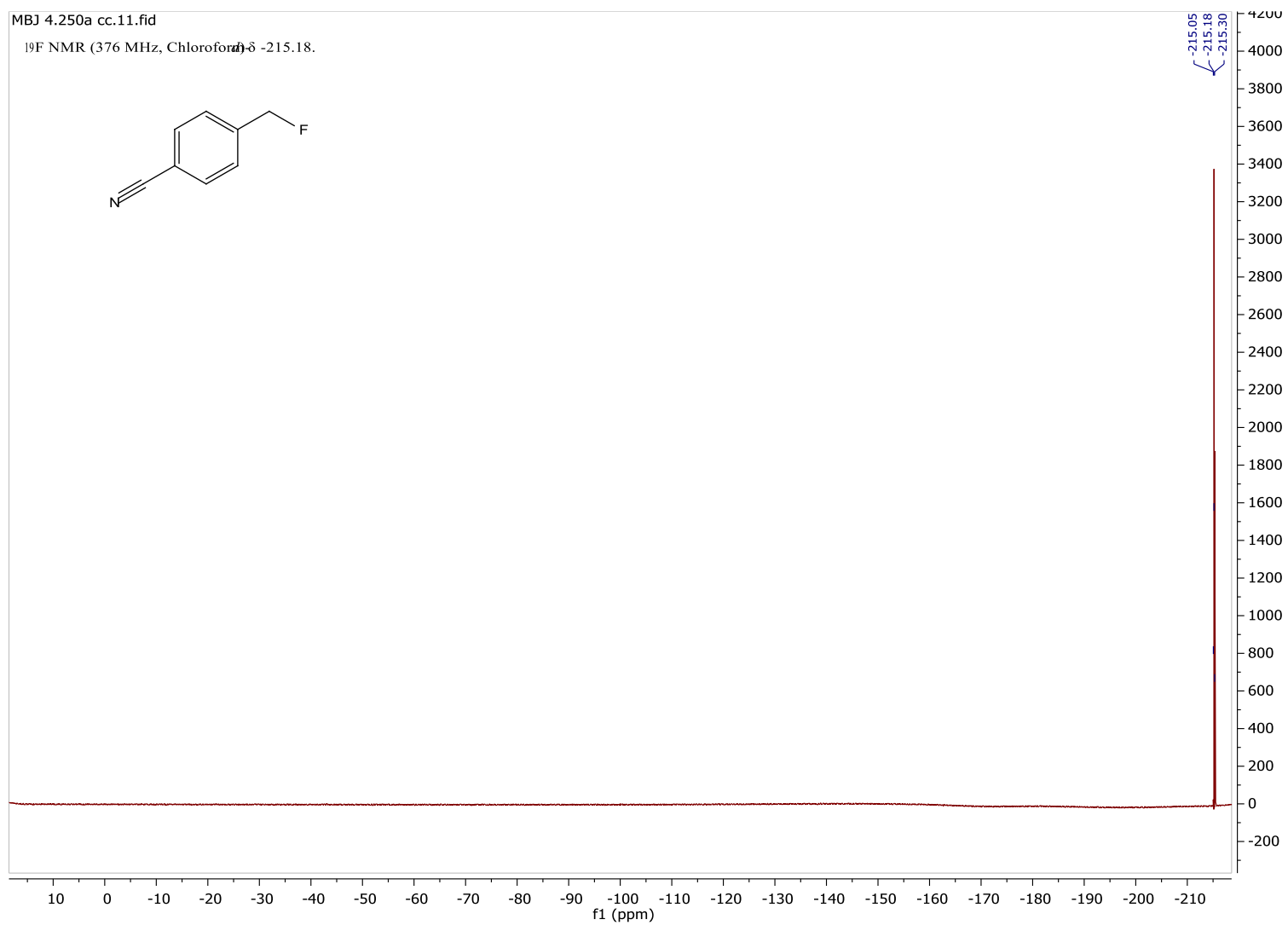
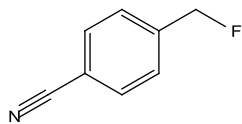
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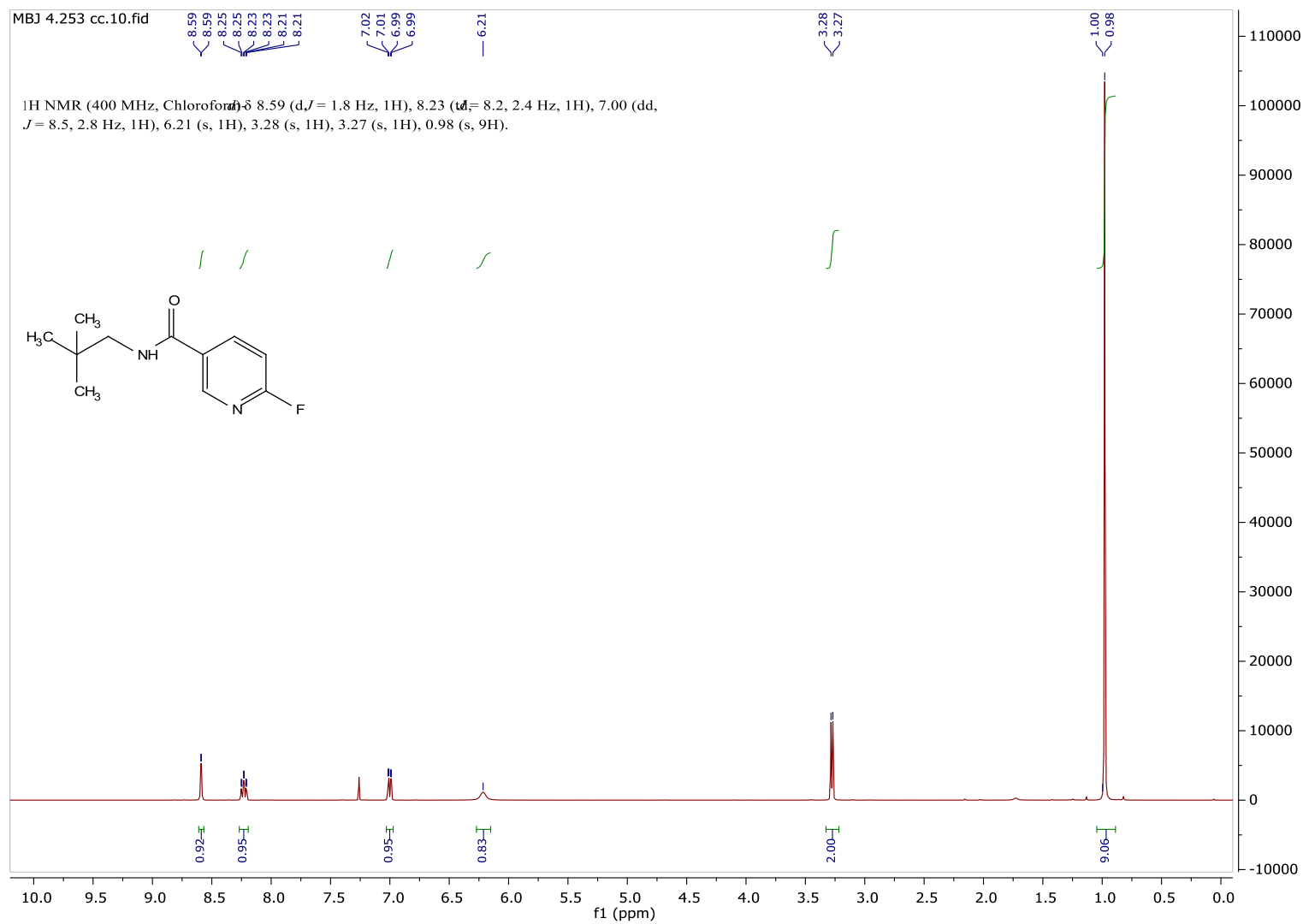
 ^{19}F NMR (376 MHz, Chloroform- d_3) δ -216.44 (J = 47.2 Hz).

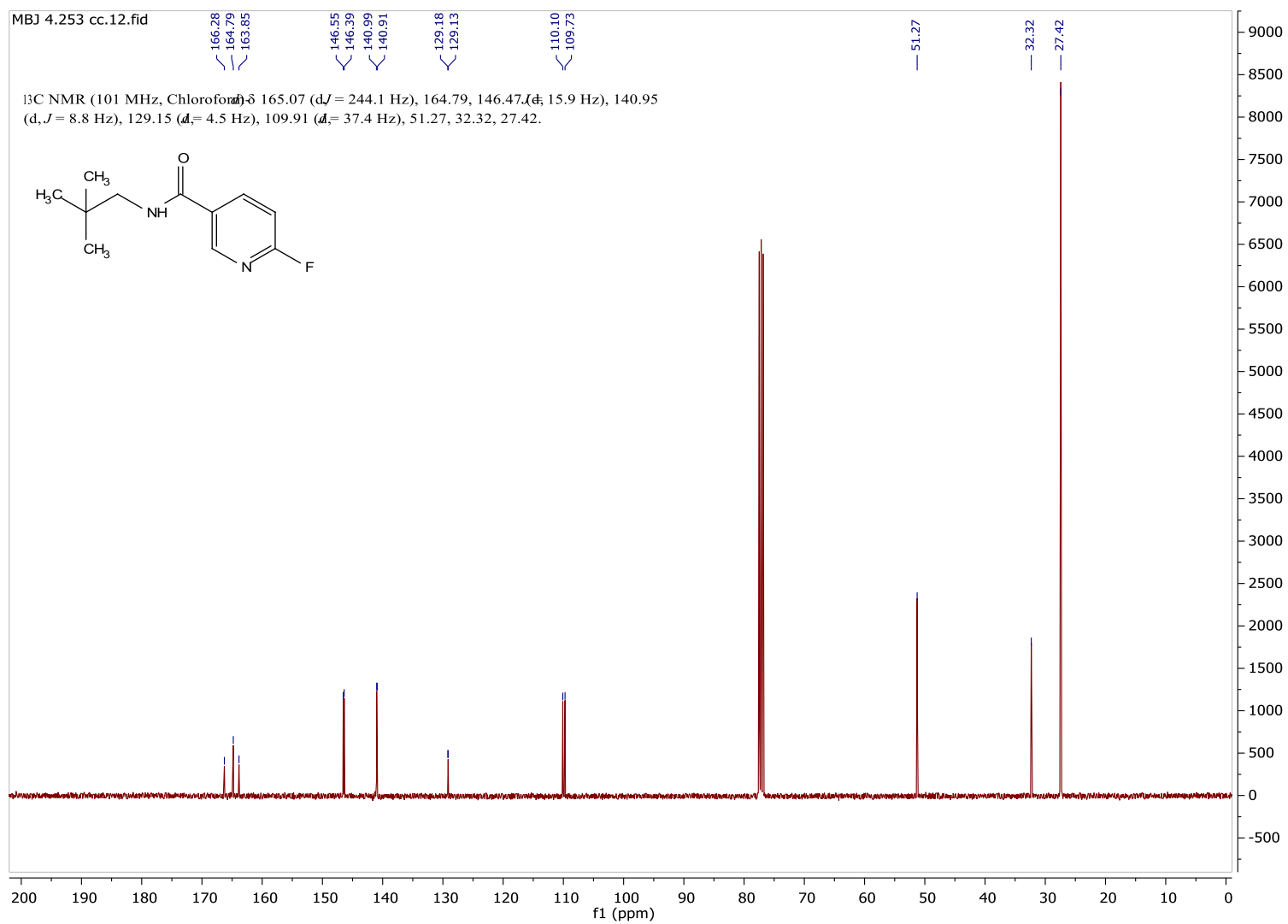
1-Fluoromethyl-4-Cyanobenzene (4).



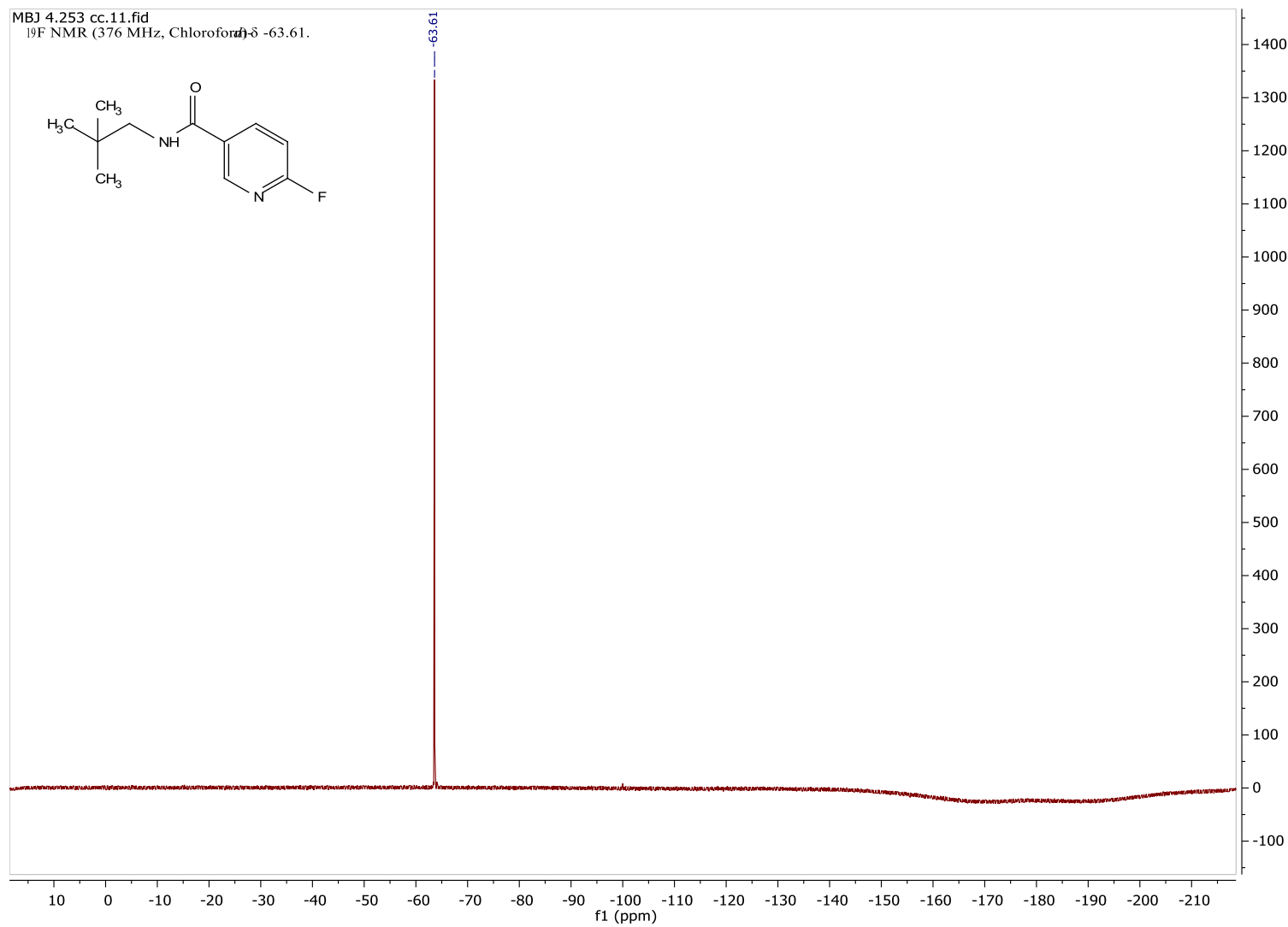
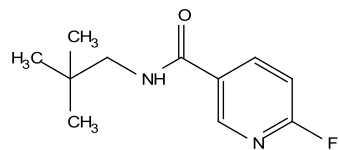
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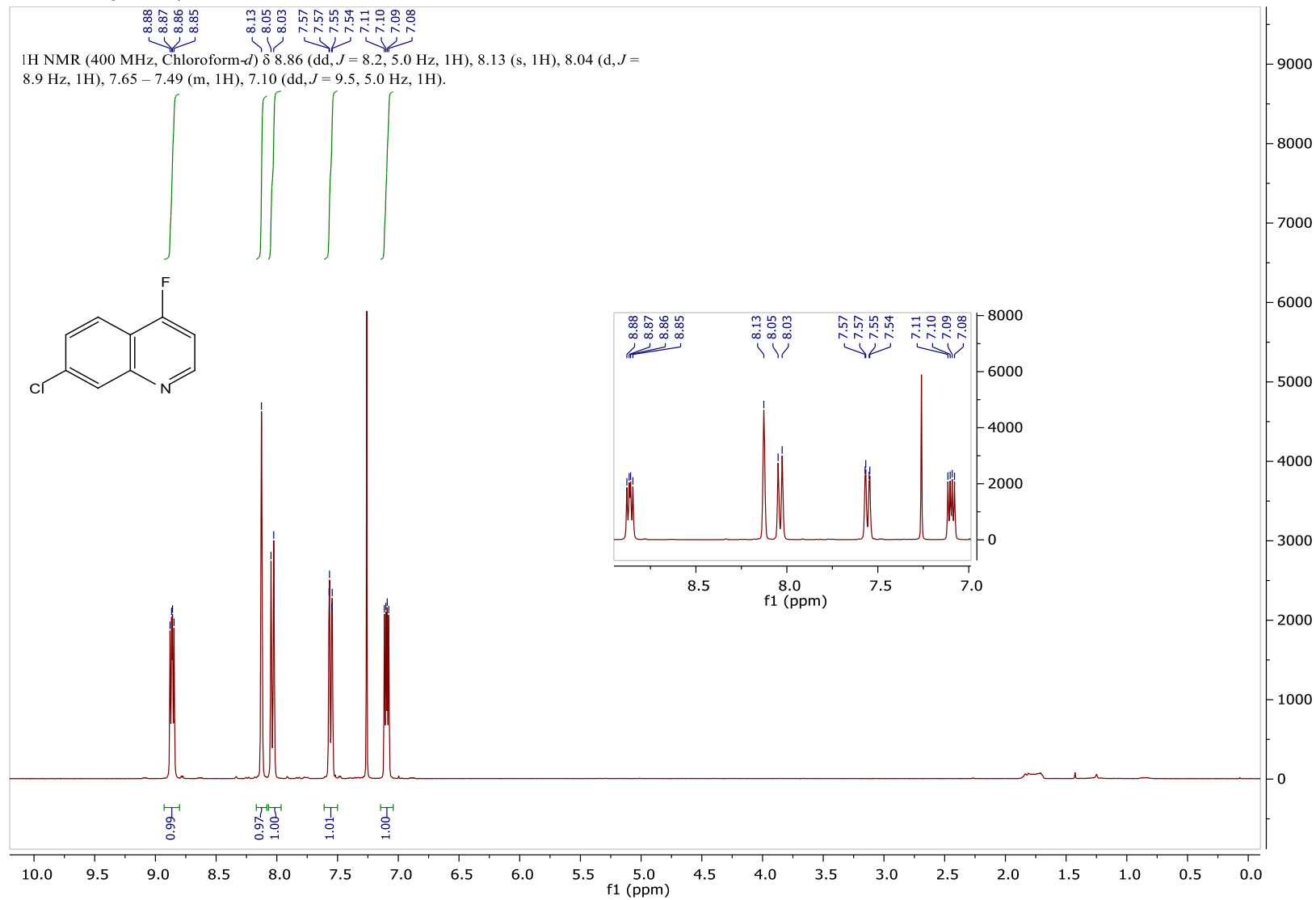
 ^{19}F NMR (376 MHz, Chloroform- d_3) δ -215.18.

6-Fluoro-N-neopentylnicotinamide (6).

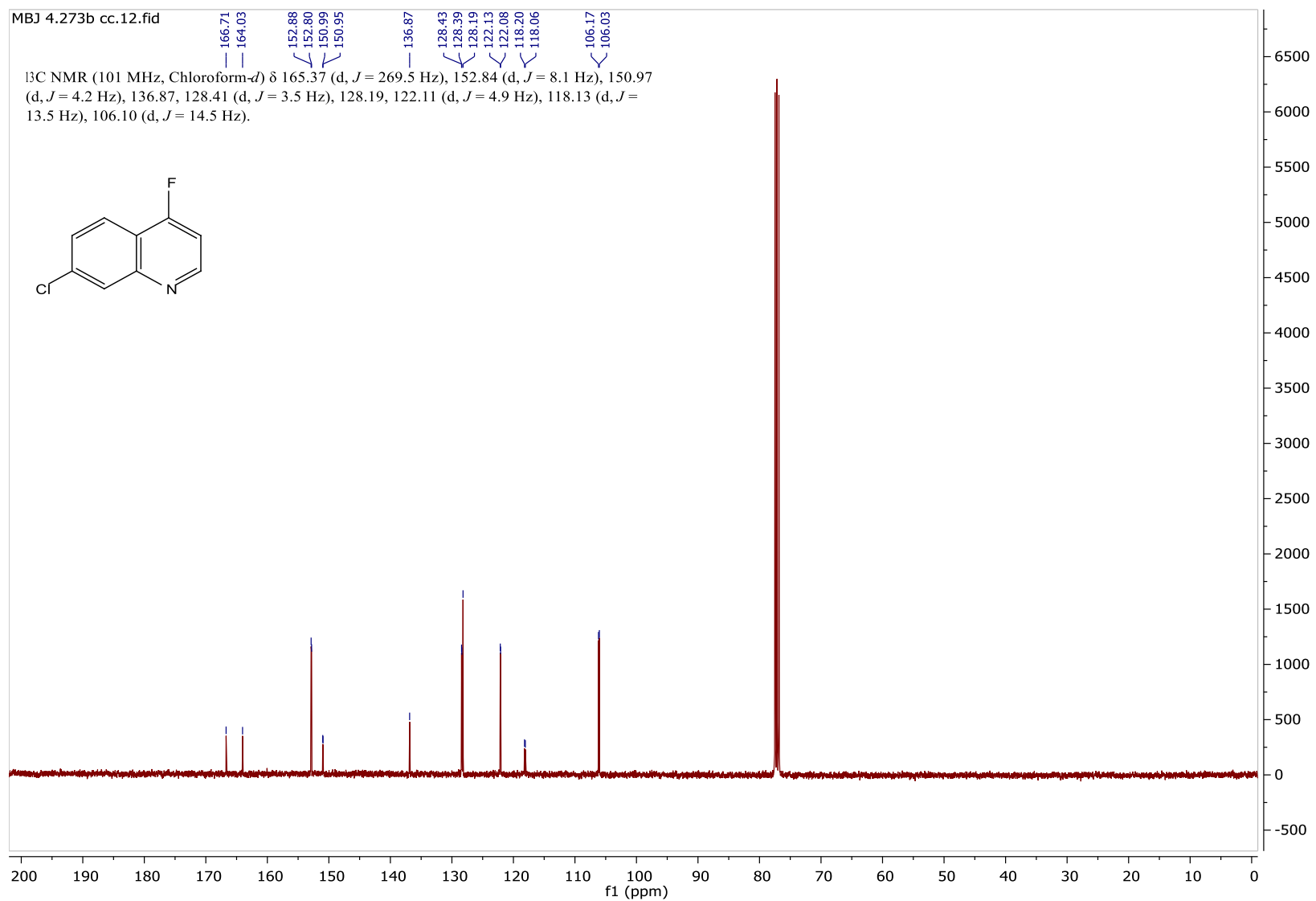


MBJ 4.253 cc.11.fid
19F NMR (376 MHz, Chloroform-d) δ -63.61.

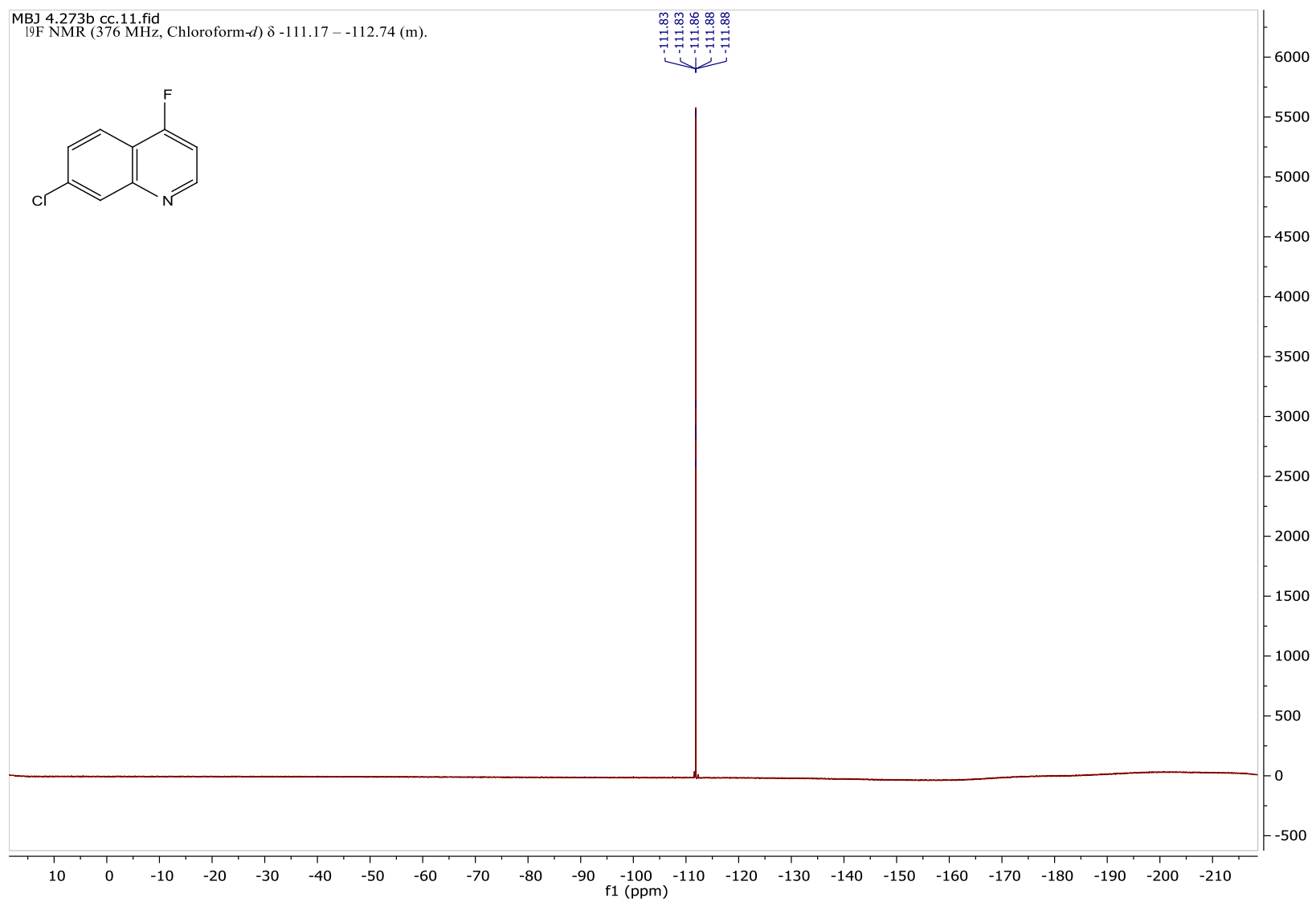
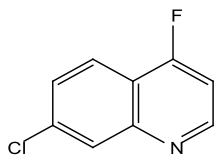


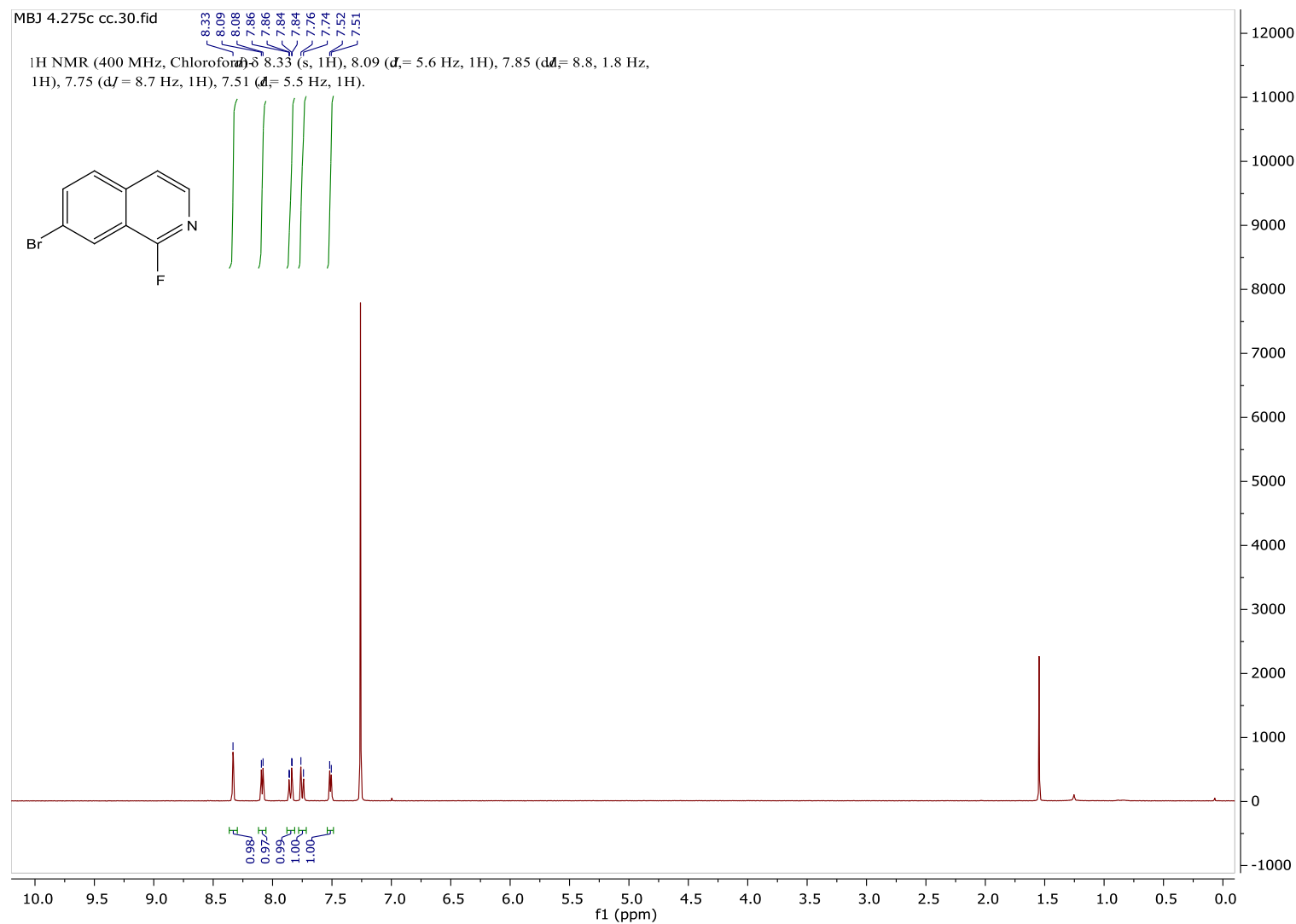
7-Chloro-4-fluoroquinoline (7).

MBJ 4.273b cc.12.fid

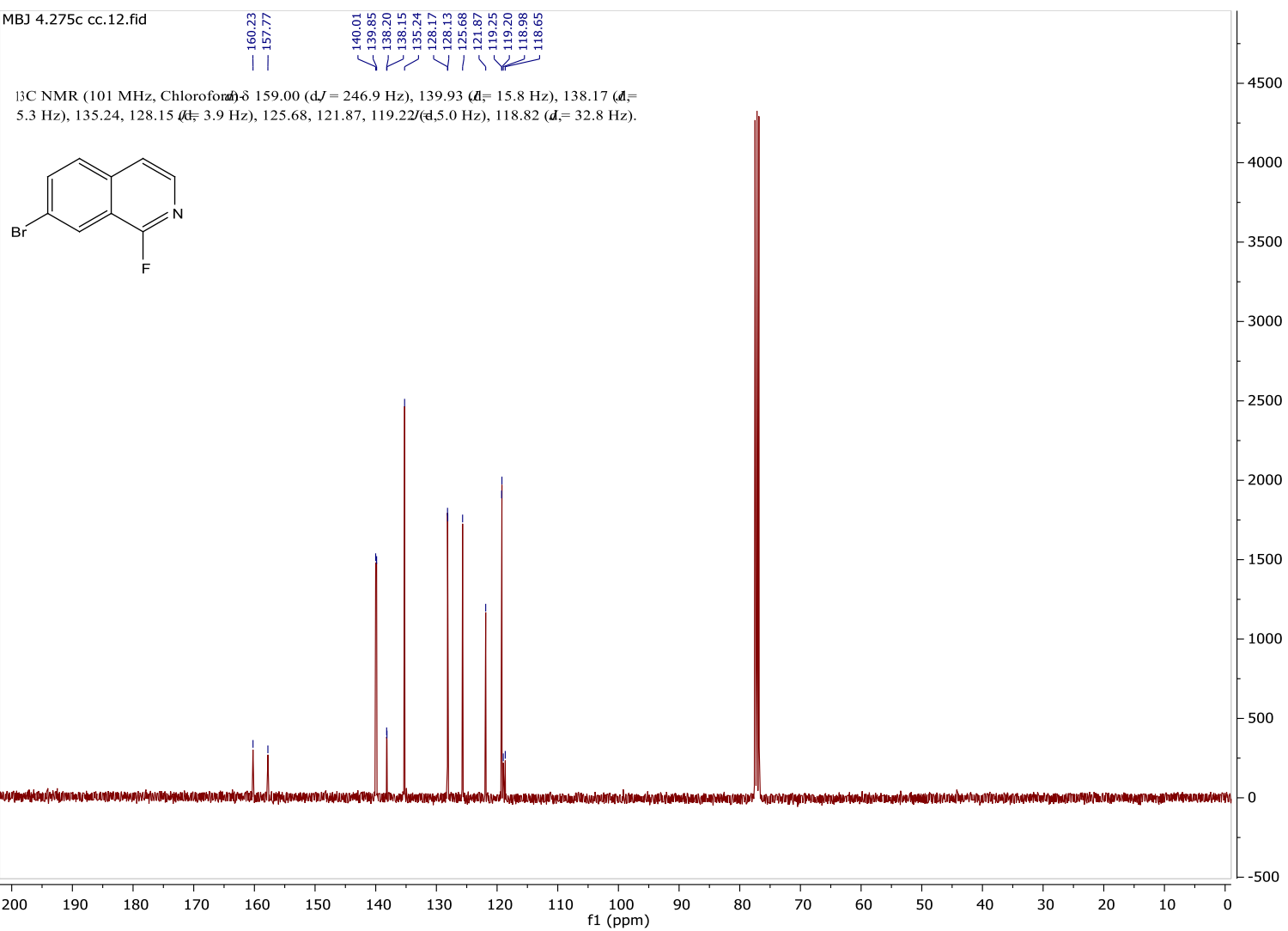


MBJ 4.273b cc.11.fid

 ^{19}F NMR (376 MHz, Chloroform-*d*) δ -111.17 – -112.74 (m).

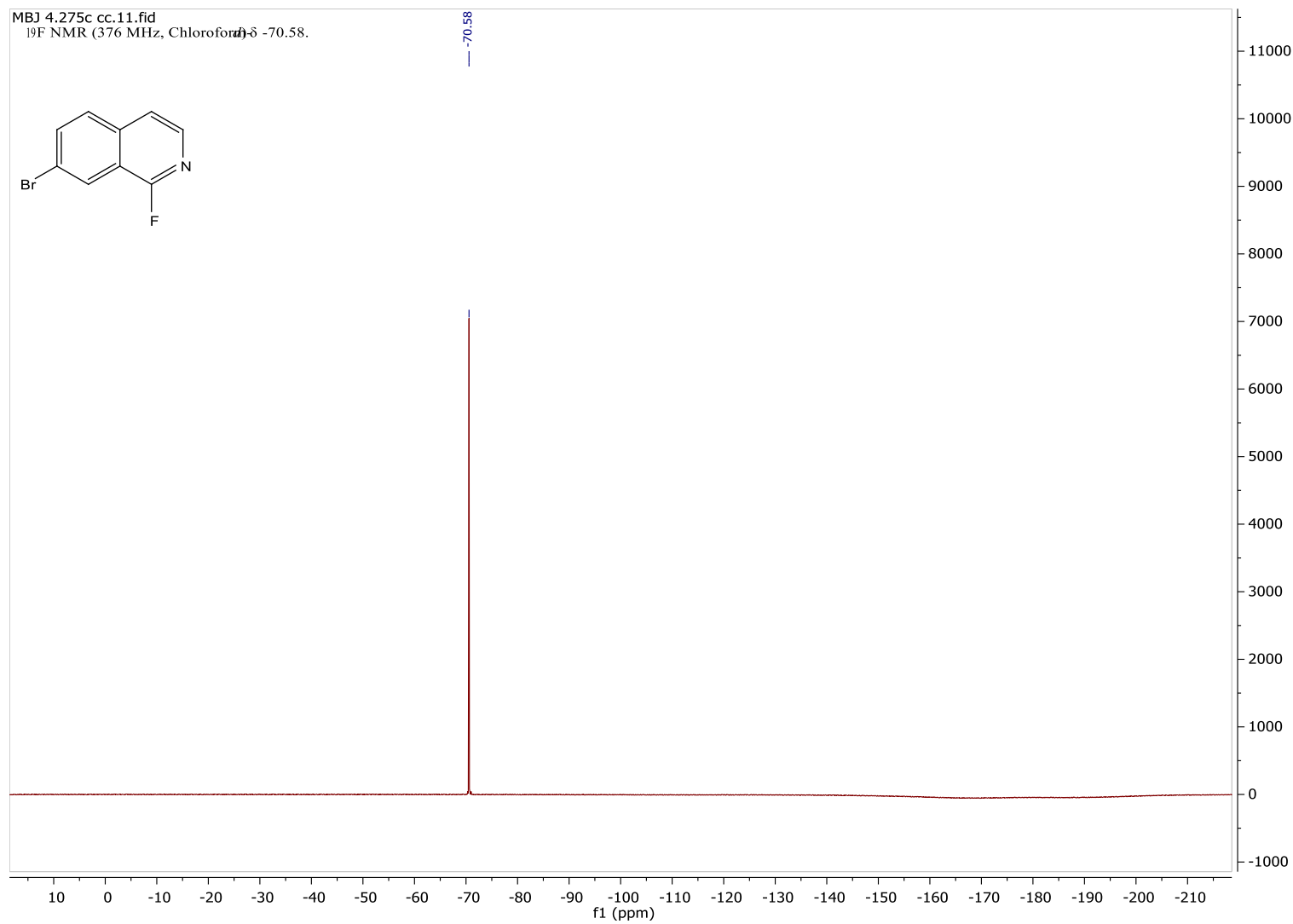
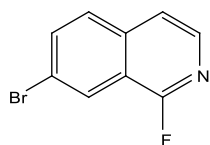
7-Bromo-1-fluoroisoquinoline (8).

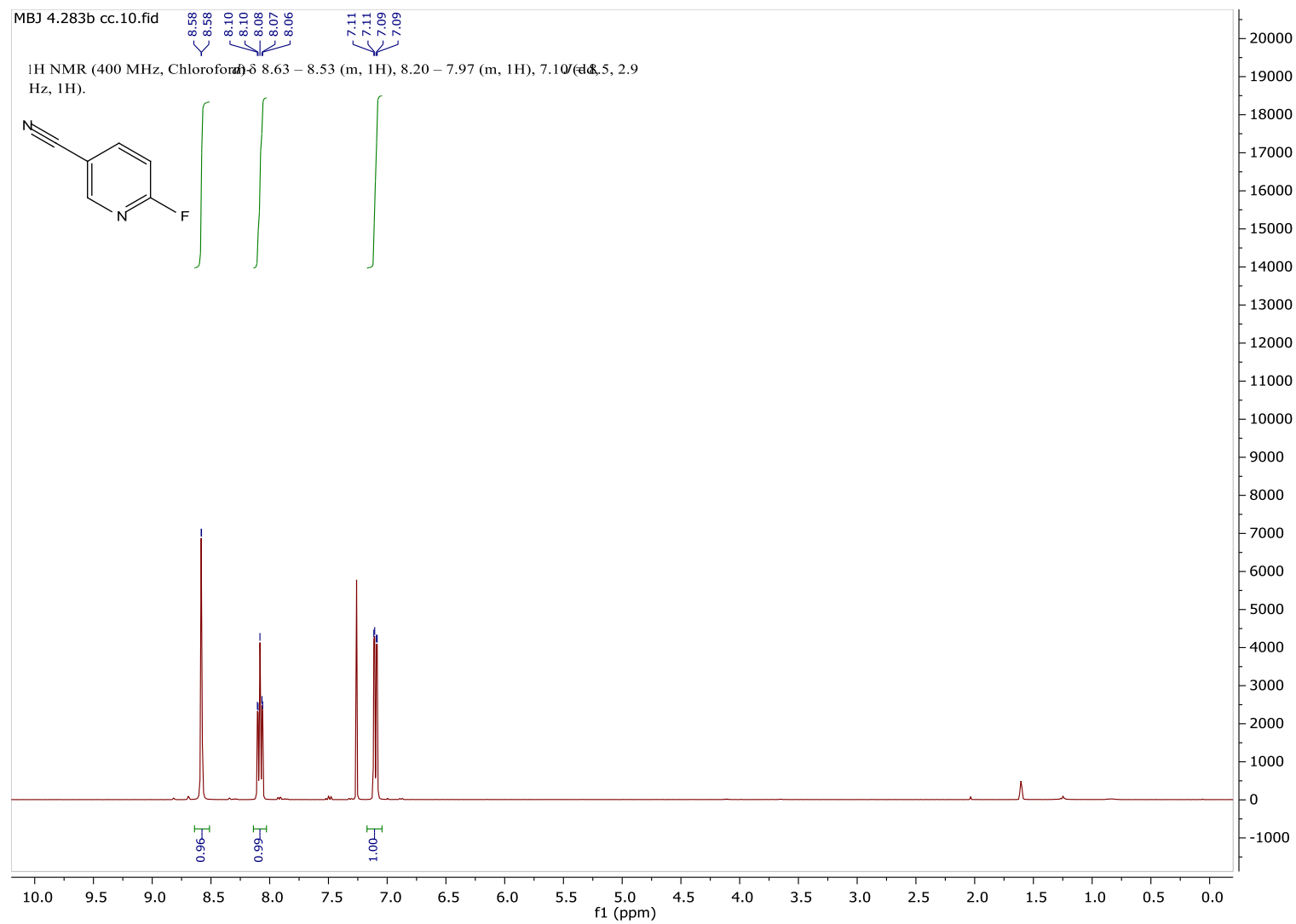
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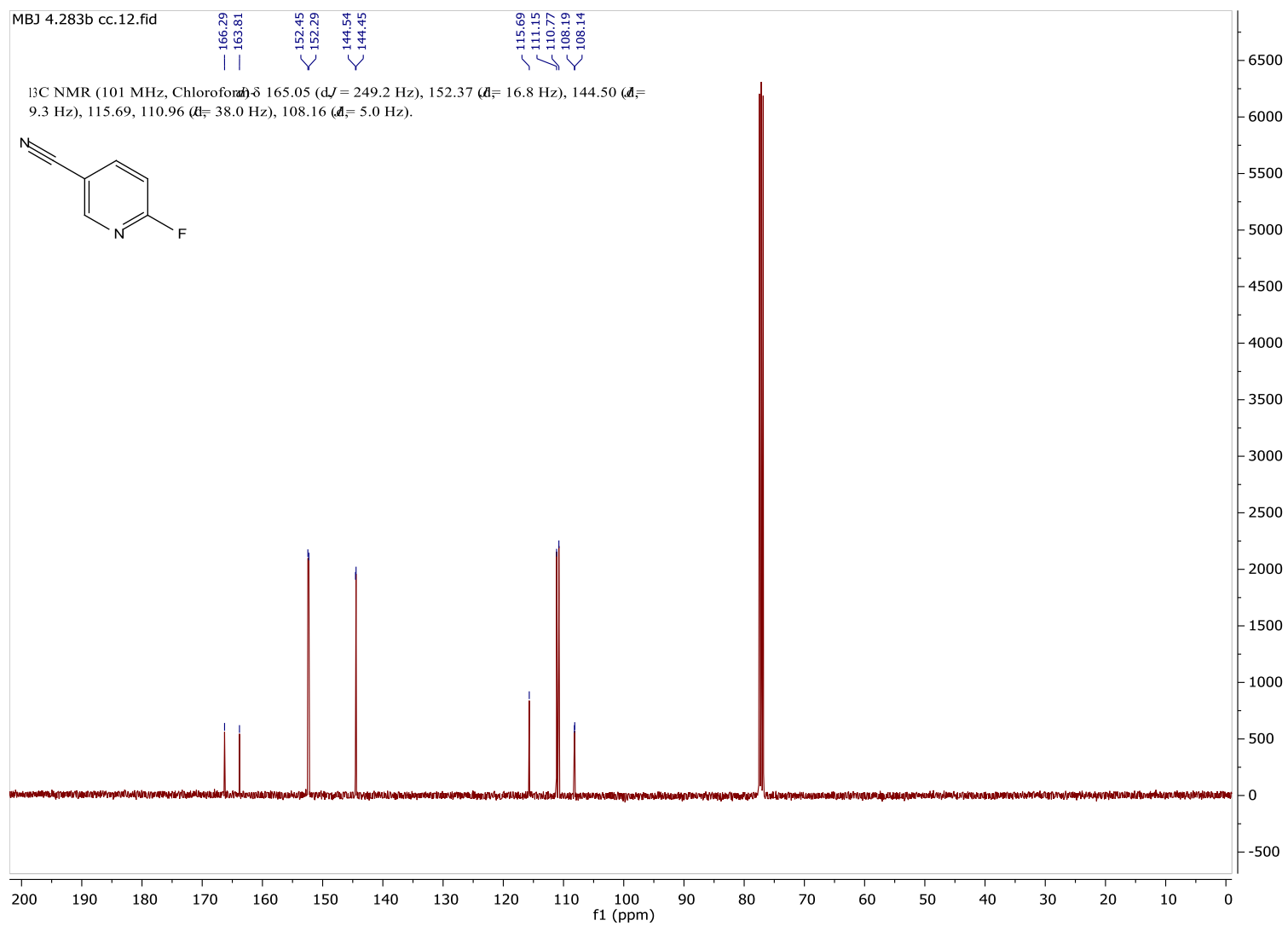


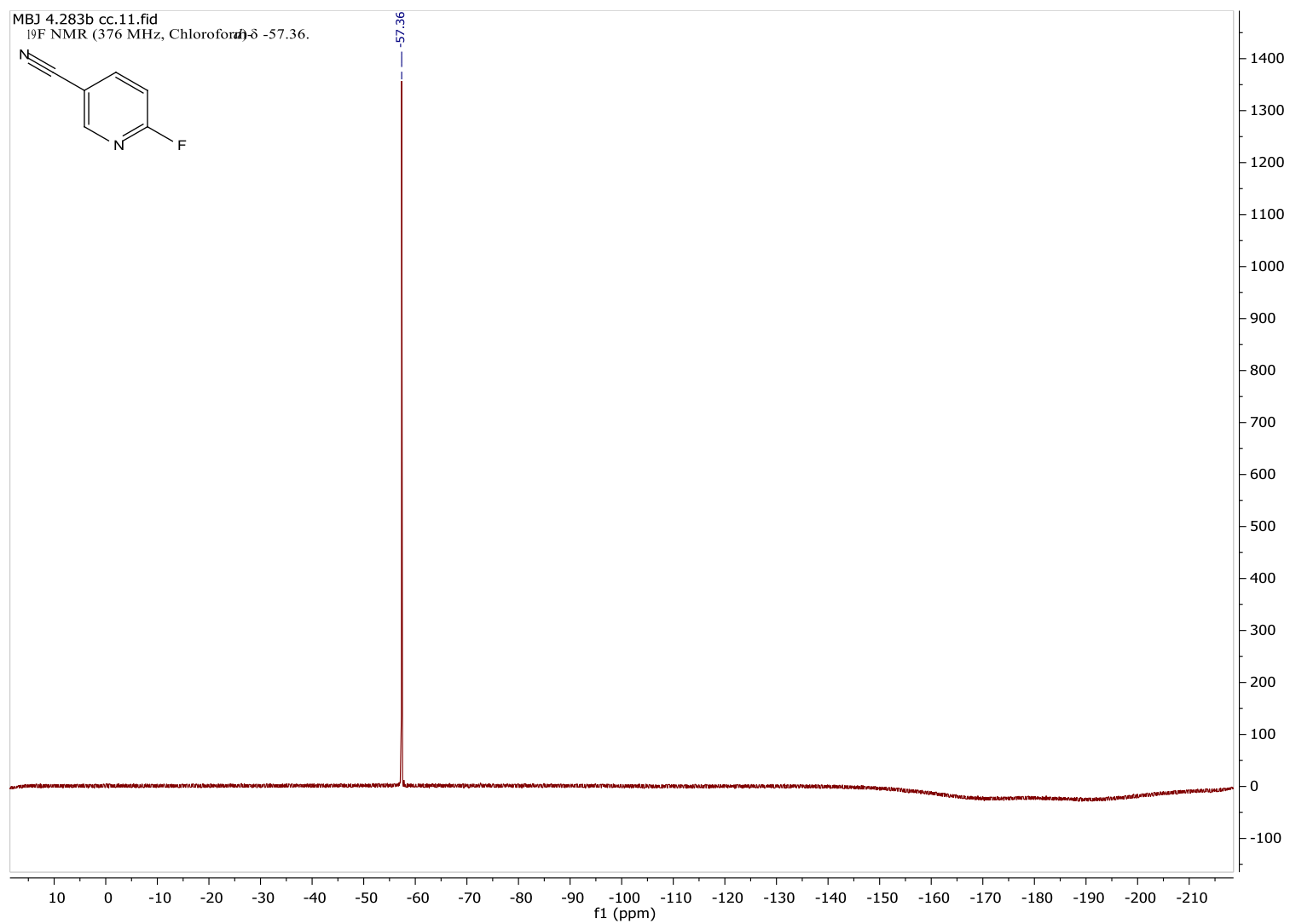
SI-52

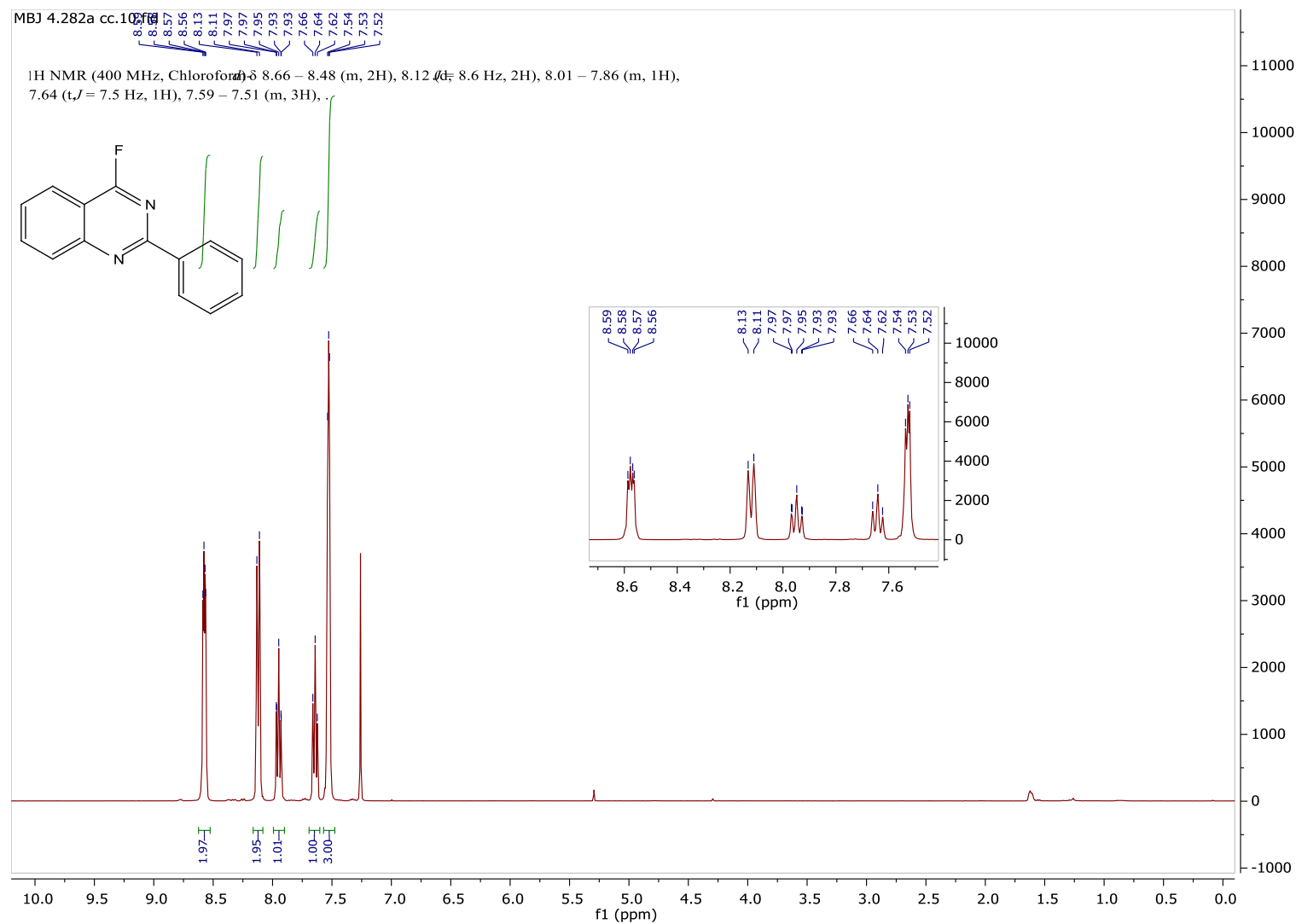
MBJ 4.275c cc.11.fid
¹⁹F NMR (376 MHz, Chloroform-d)
-70.58.

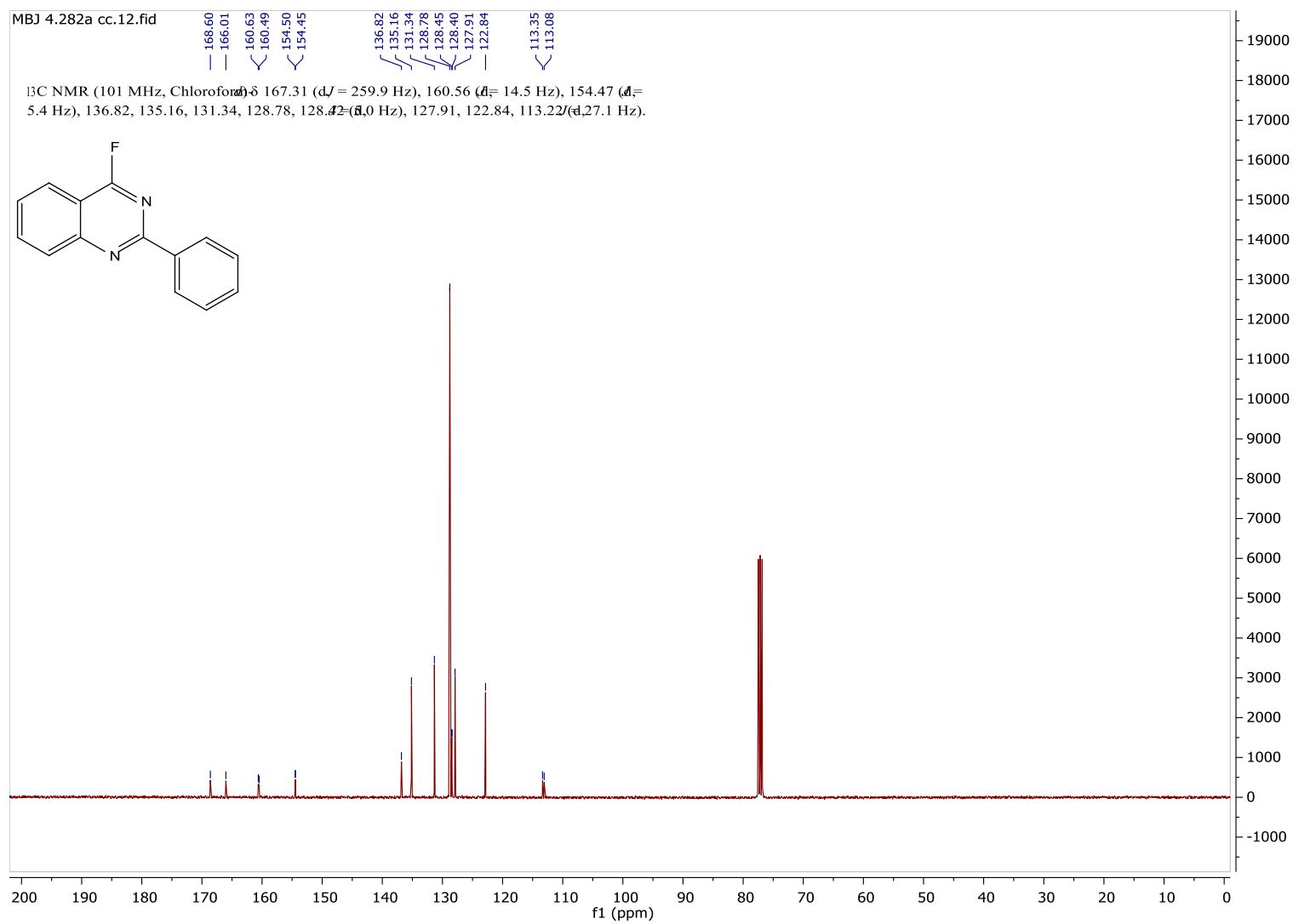


6-Fluoronicotinonitrile (9).



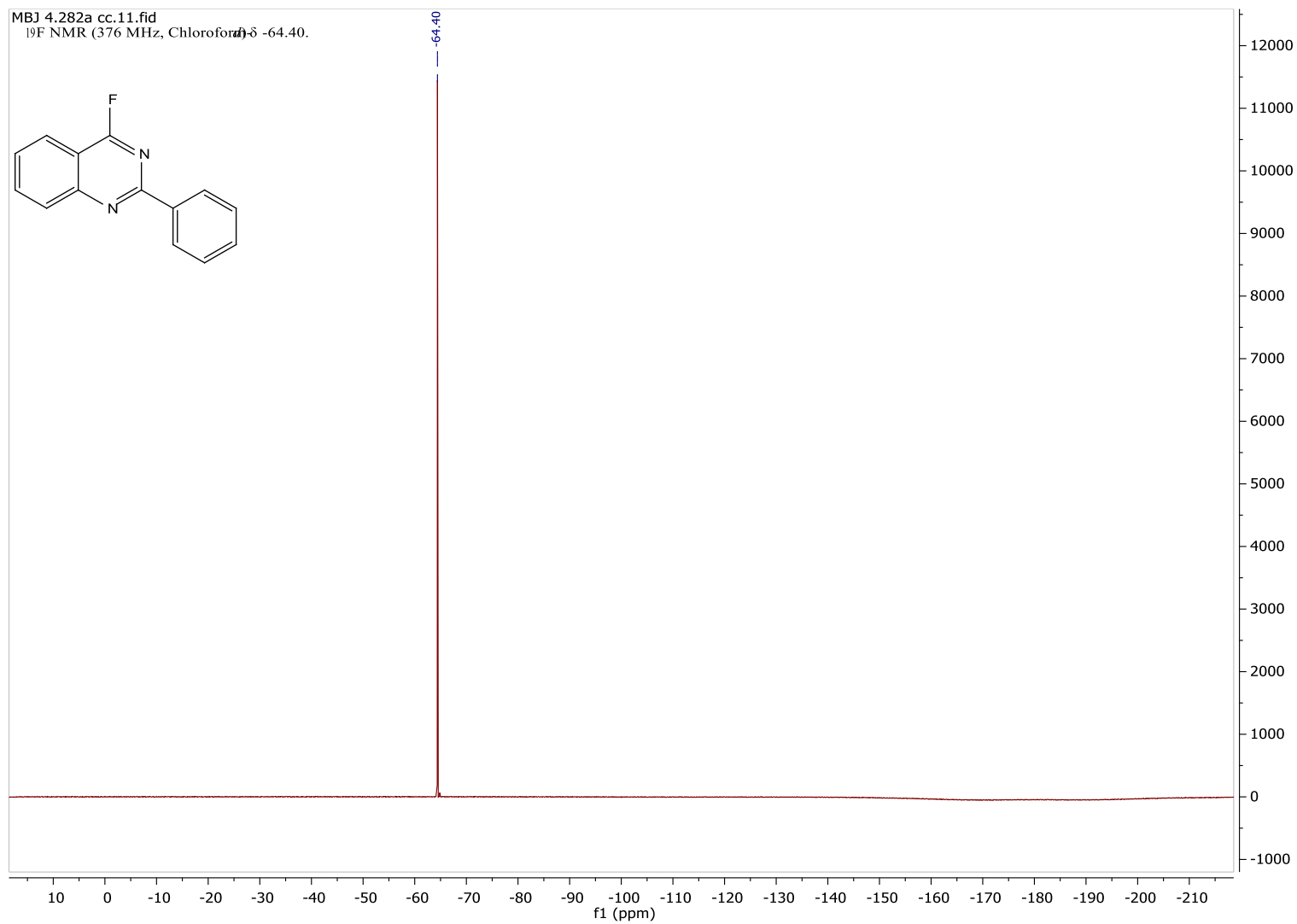
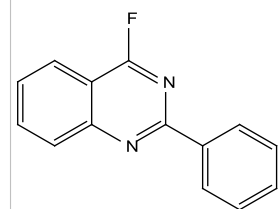


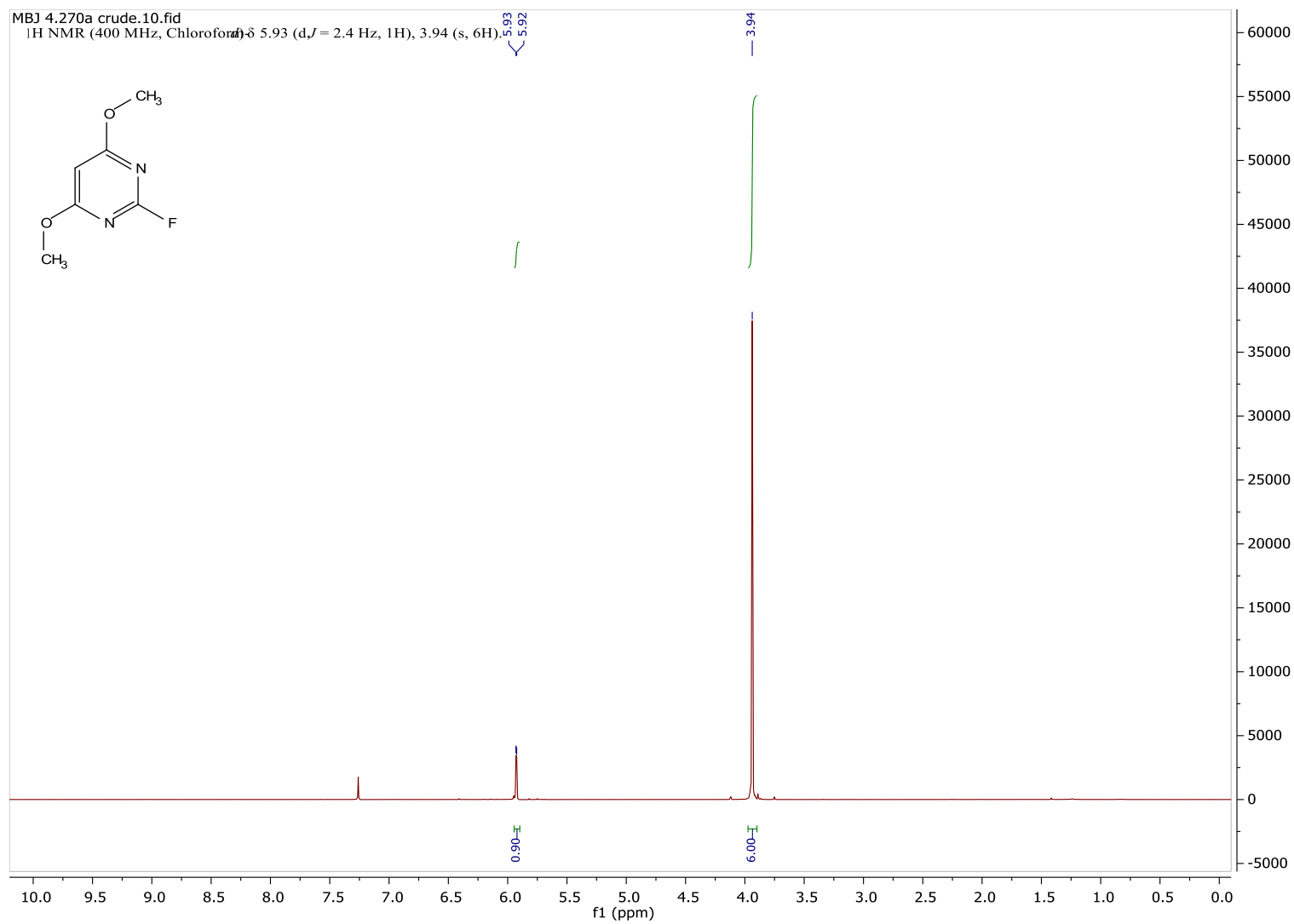
4-Fluoro-2-phenylquinazoline (10).

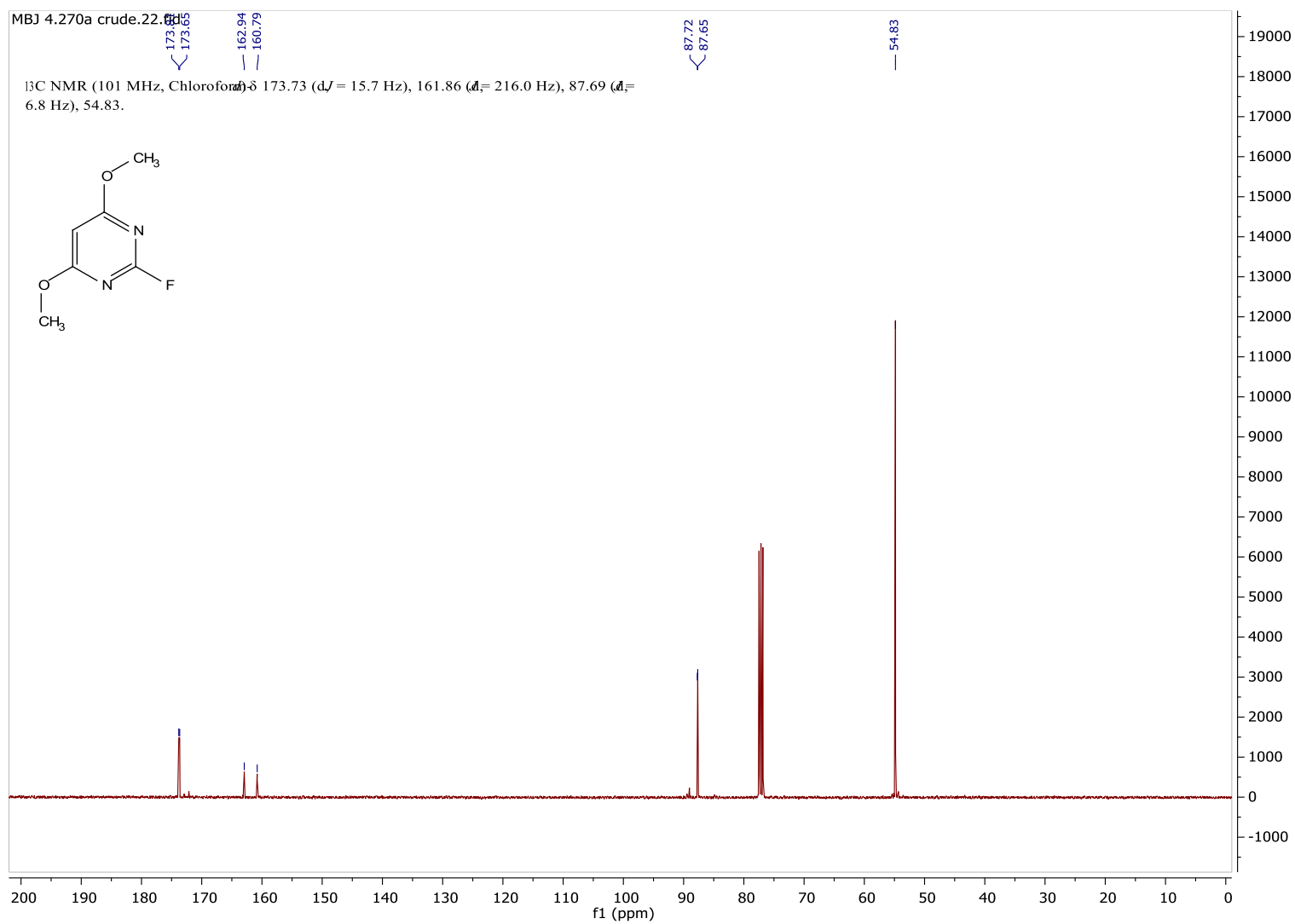


SI-58

MBJ 4.282a cc.11.fid
19F NMR (376 MHz, Chloroform-d)

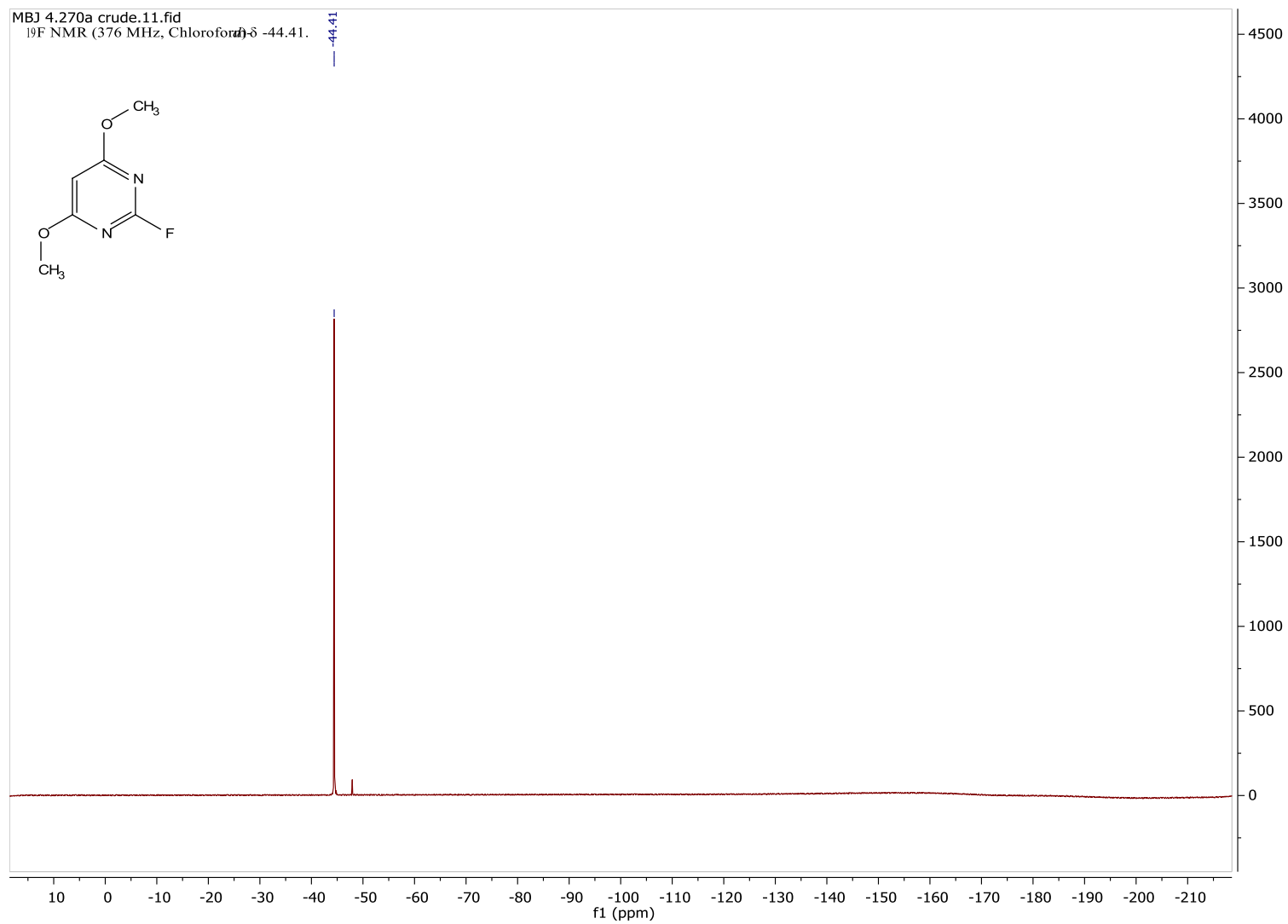
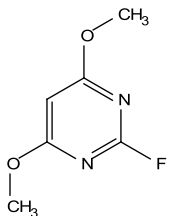


2-Fluoro-4,6-dimethoxypyrimidine (11).



SI-61

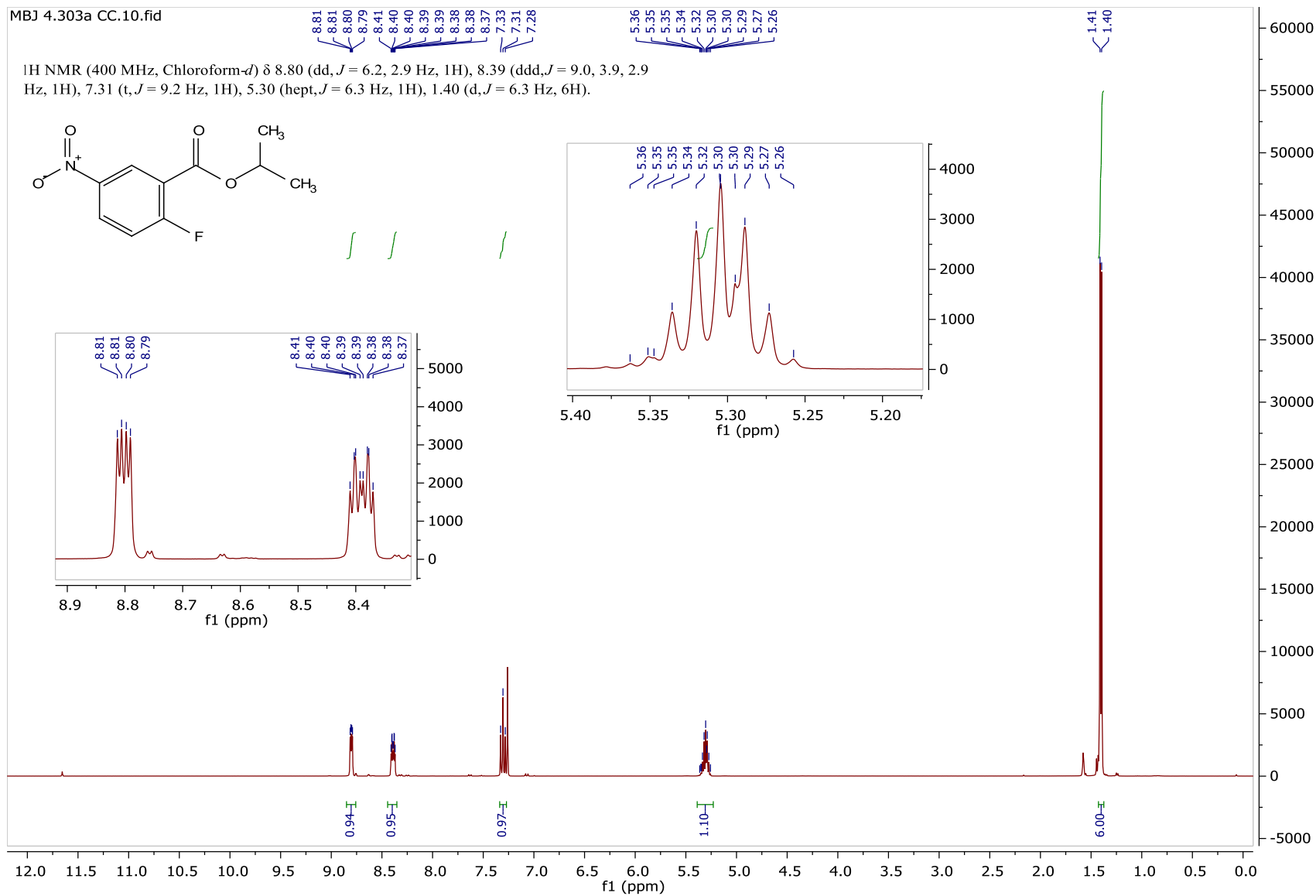
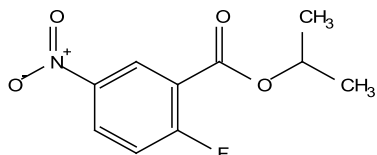
MBJ 4.270a crude.11.fid
19F NMR (376 MHz, Chloroform-d)
-44.41.



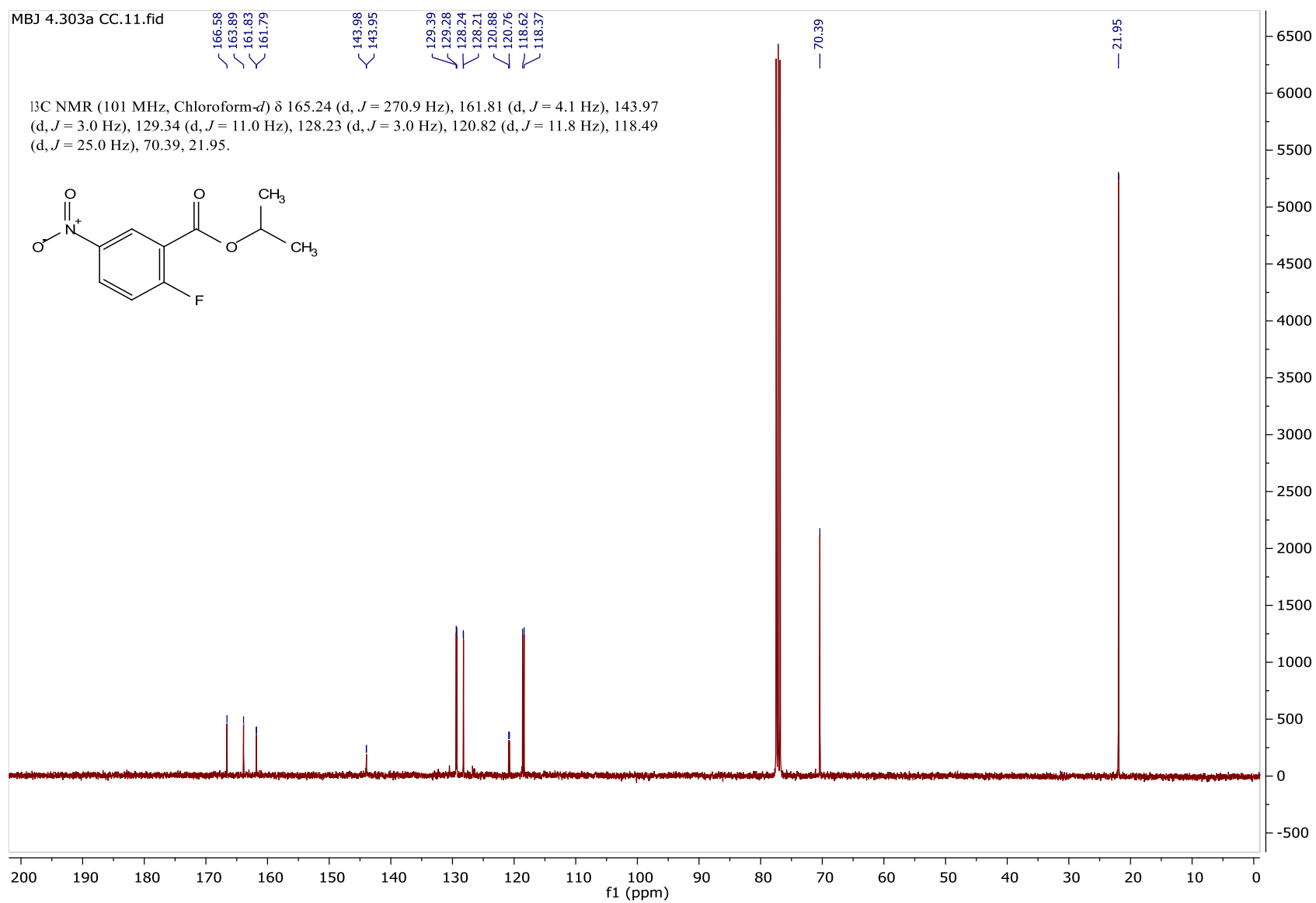
Isopropyl 2-fluoro-5-nitrobenzoate (**12**).

MBJ 4.303a CC.10.fid

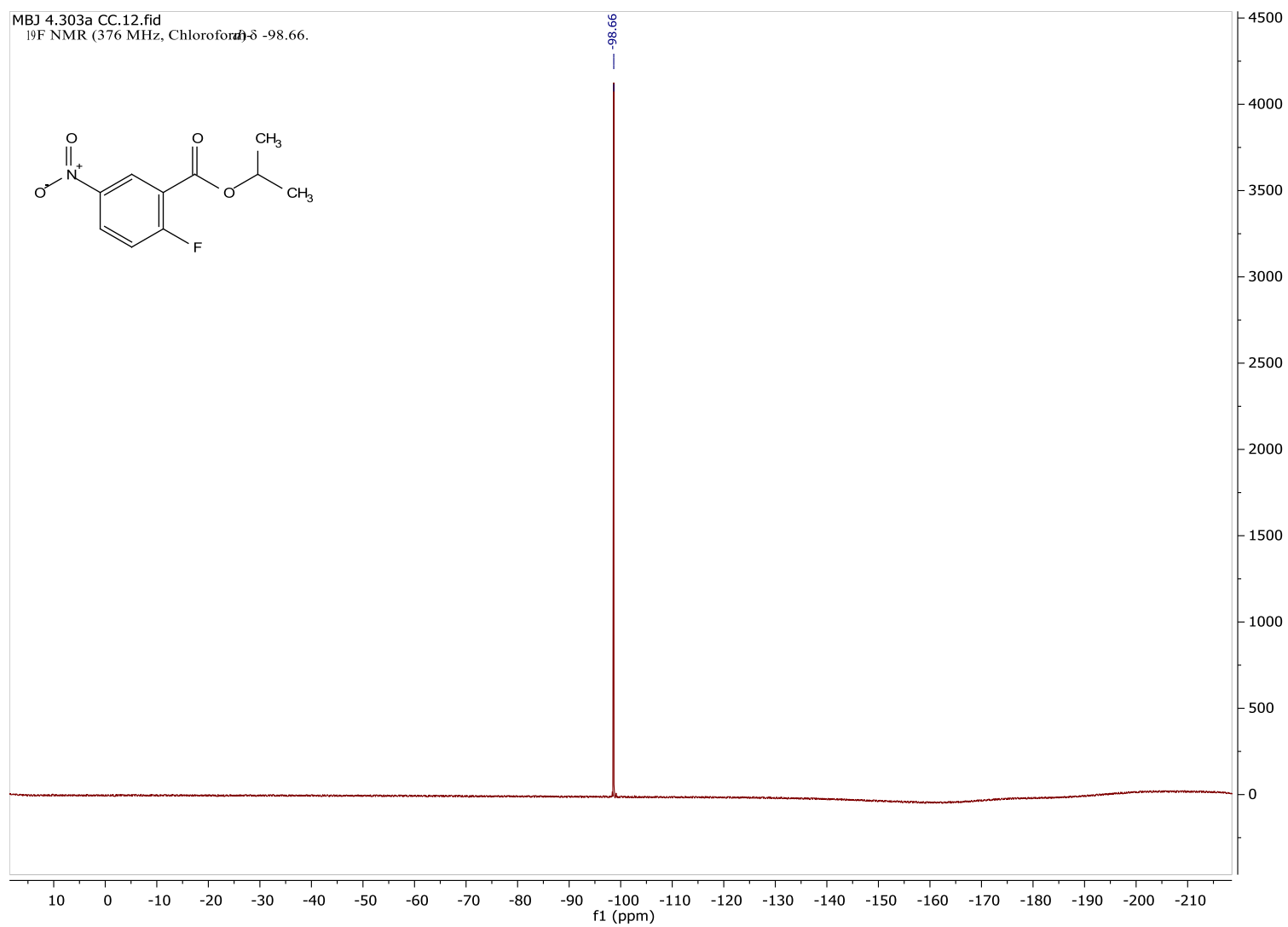
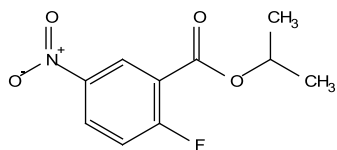
^1H NMR (400 MHz, Chloroform- d) δ 8.80 (dd, $J = 6.2, 2.9$ Hz, 1H), 8.39 (ddd, $J = 9.0, 3.9, 2.9$ Hz, 1H), 7.31 (t, $J = 9.2$ Hz, 1H), 5.30 (hept, $J = 6.3$ Hz, 1H), 1.40 (d, $J = 6.3$ Hz, 6H).



MBJ 4.303a CC.11.fid



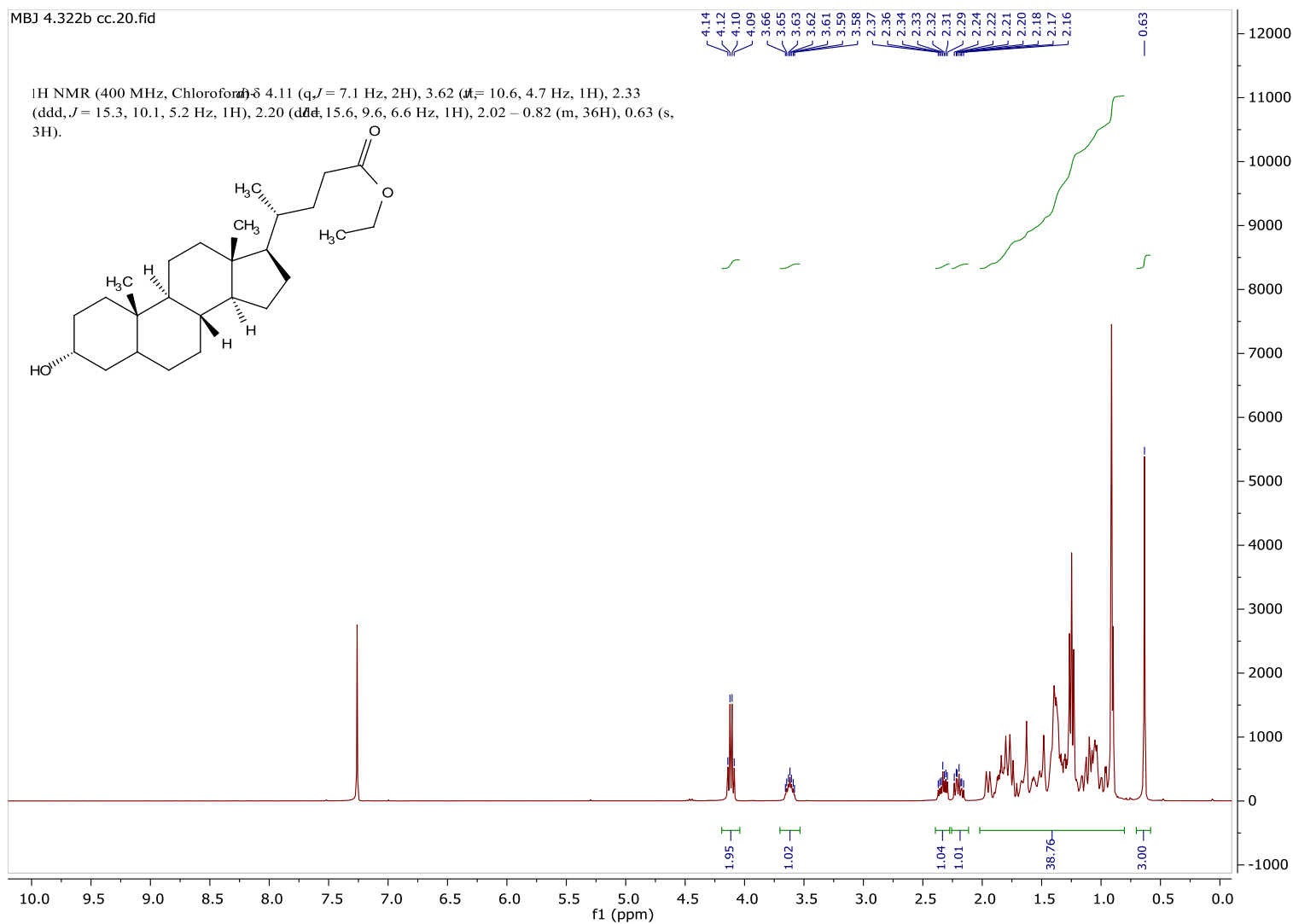
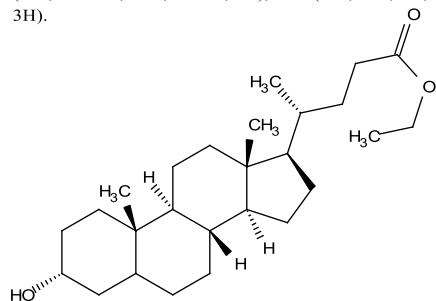
MBJ 4.303a CC.12.fid
¹⁹F NMR (376 MHz, Chloroform-d₃) -98.66.

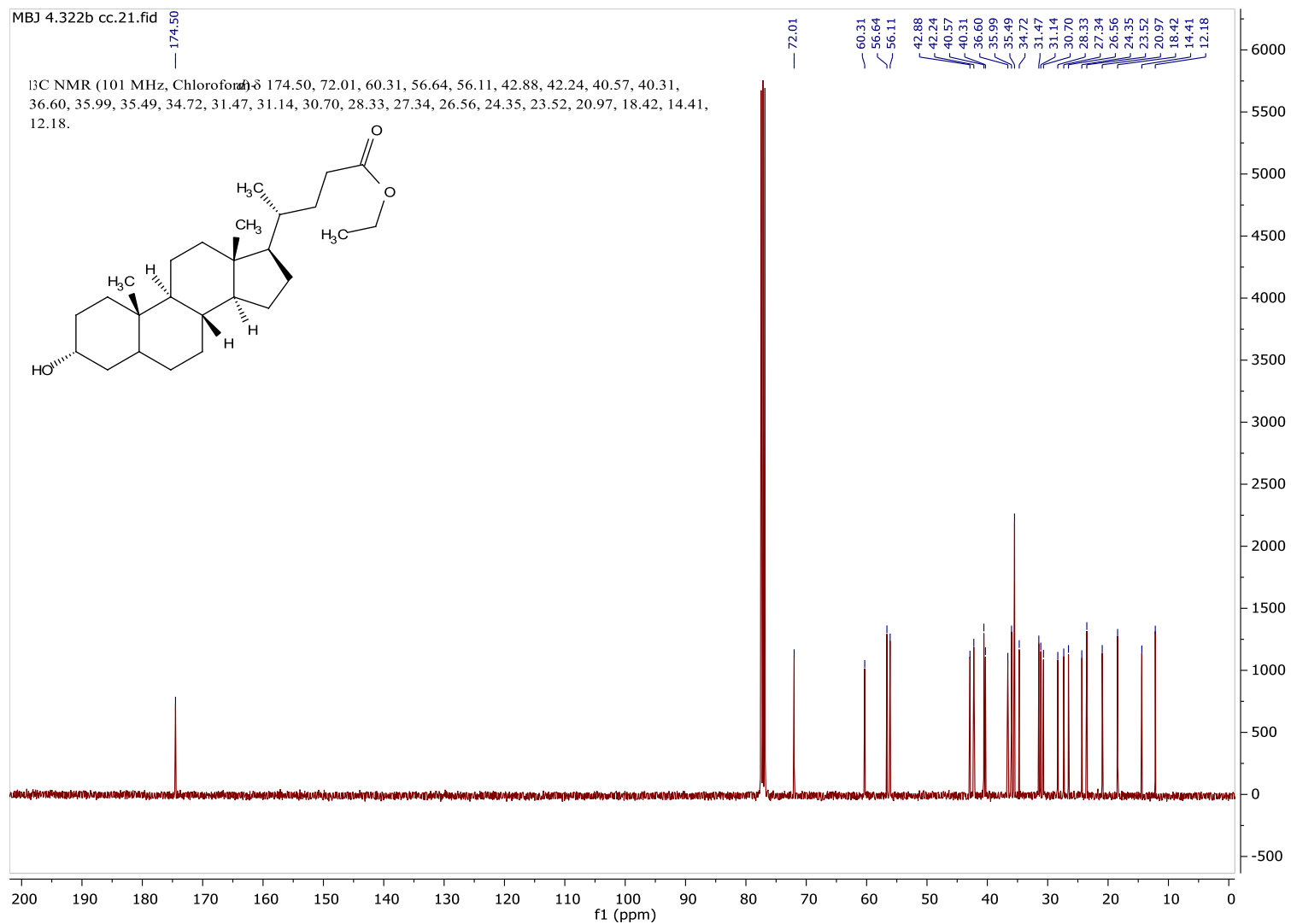


Ethyl 3 α -hydroxy-5 β -cholan-24-oate (**13**)

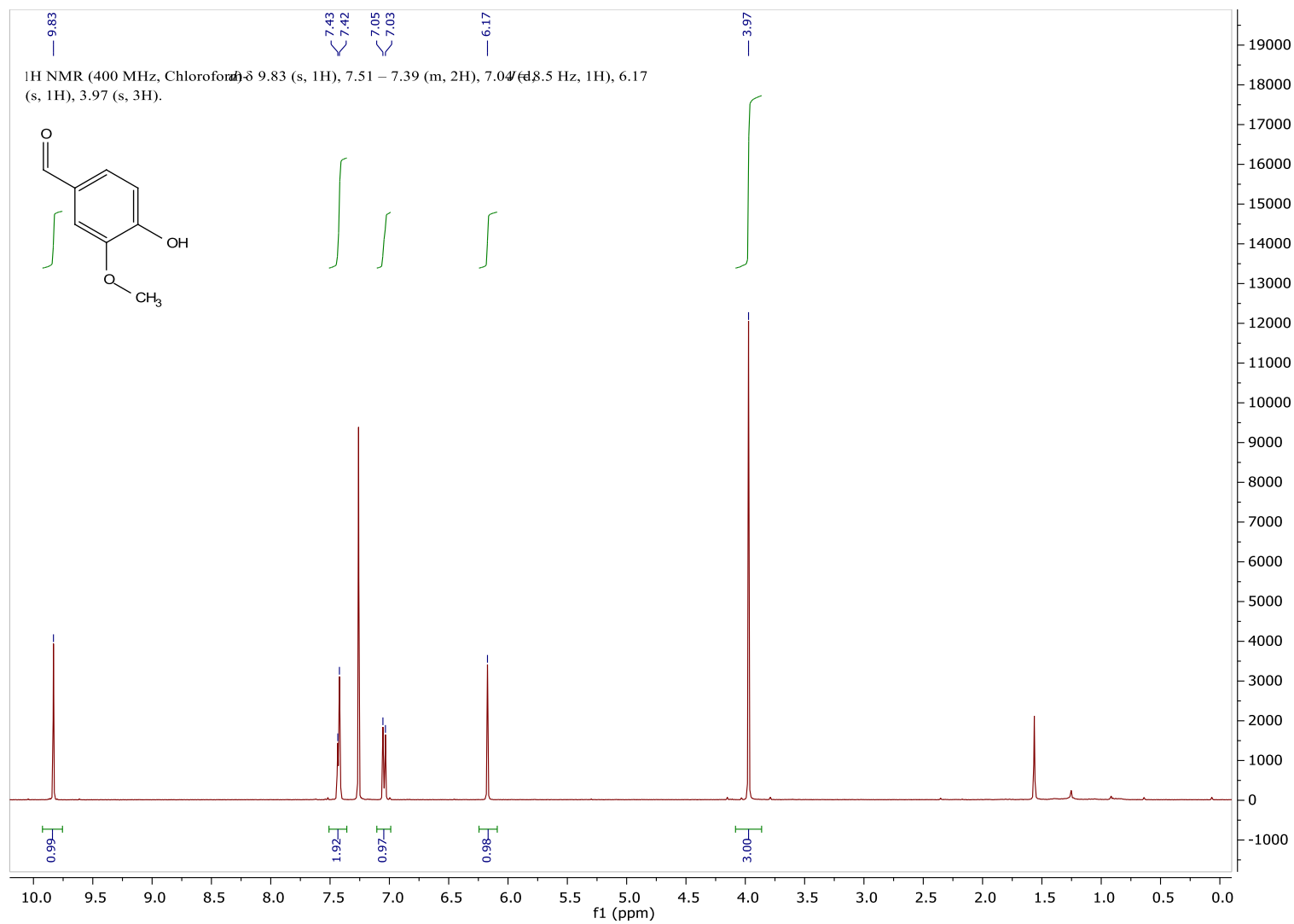
MBJ 4.322b cc.20.fid

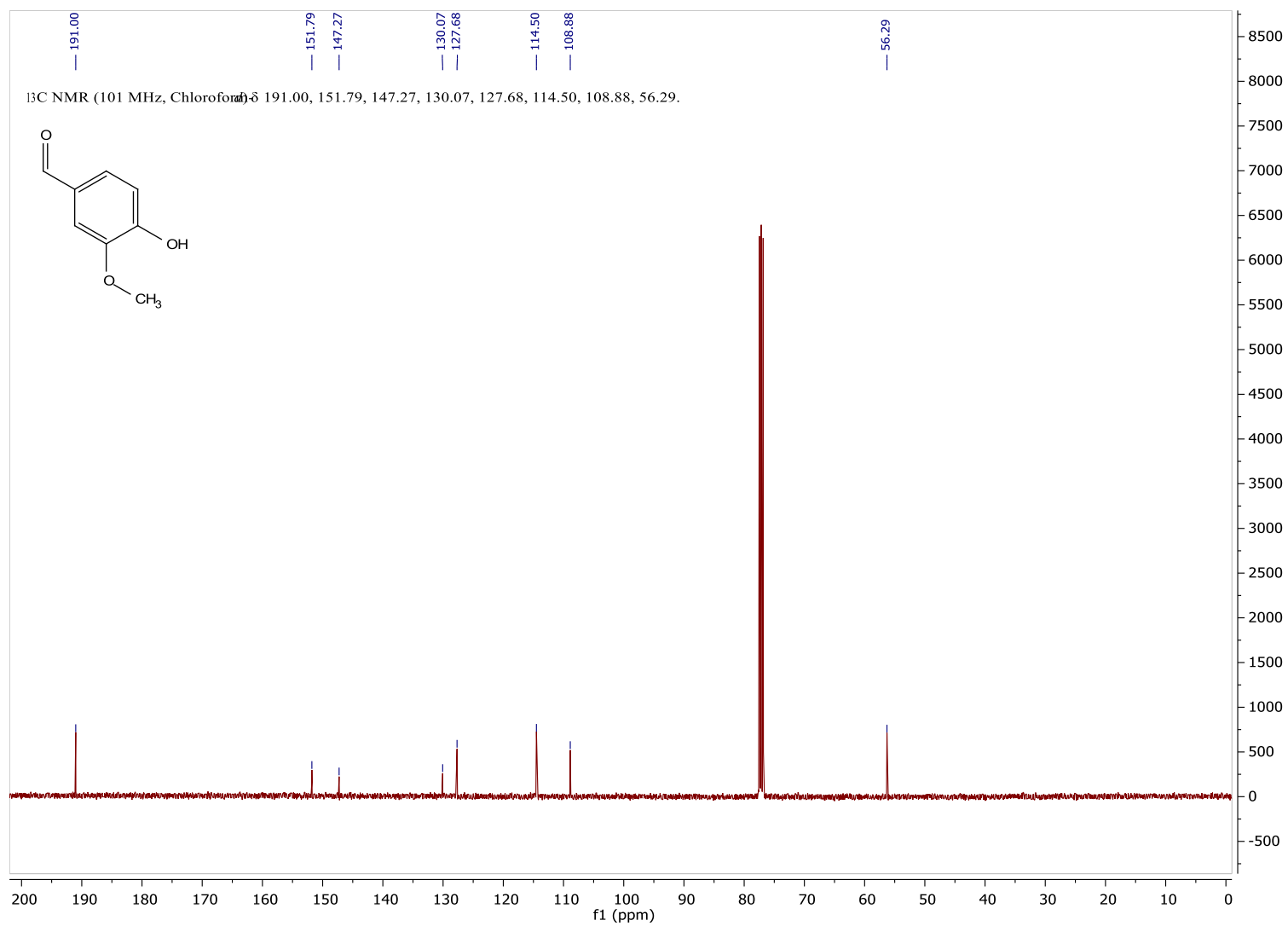
¹H NMR (400 MHz, Chloroform-d) δ 4.11 (q, J = 7.1 Hz, 2H), 3.62 (d, J = 10.6, 4.7 Hz, 1H), 2.33 (ddd, J = 15.3, 10.1, 5.2 Hz, 1H), 2.20 (ddd, J = 15.6, 9.6, 6.6 Hz, 1H), 2.02 – 0.82 (m, 36H), 0.63 (s, 3H).





Vanillin (14)

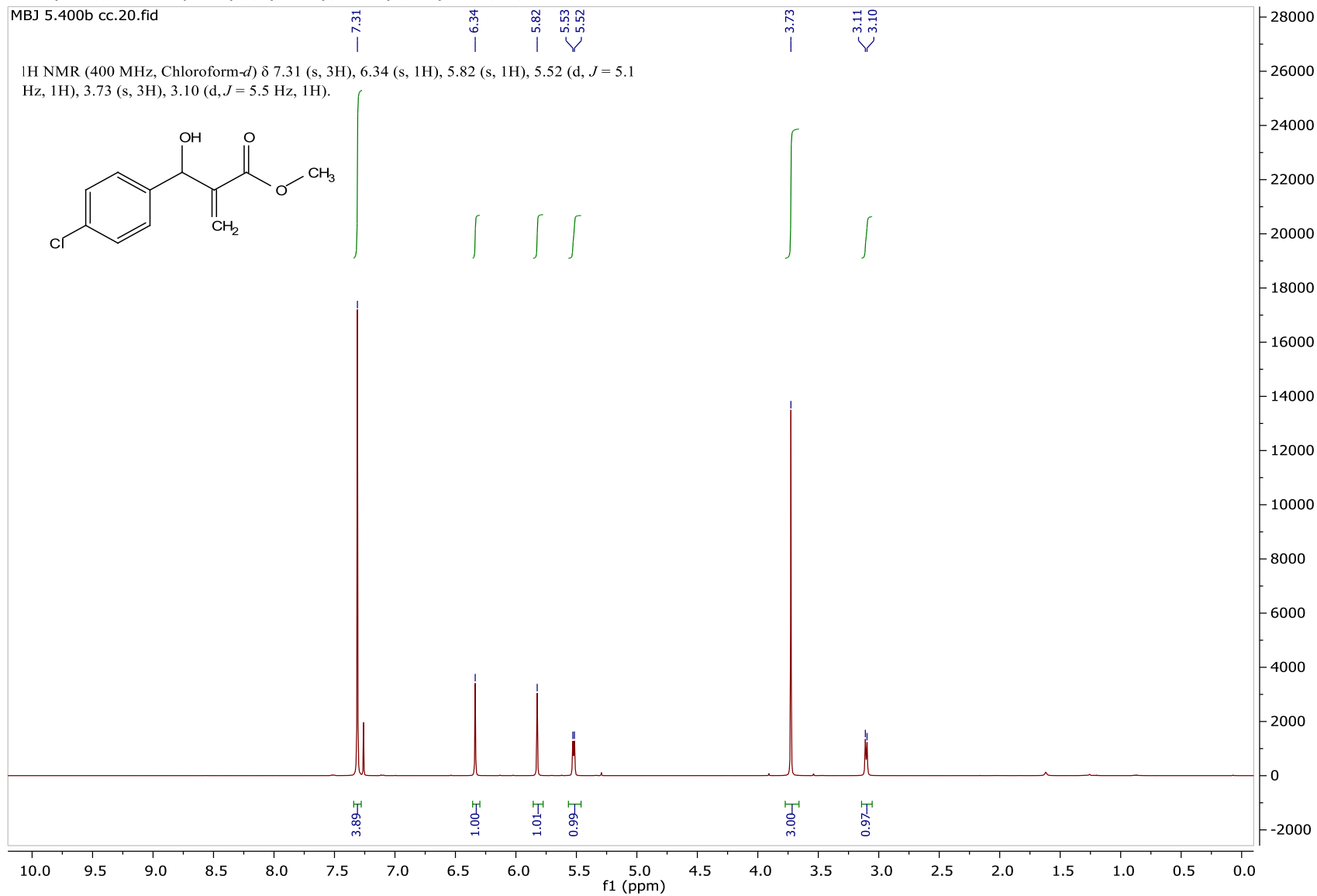
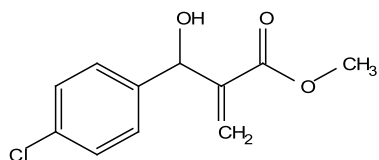


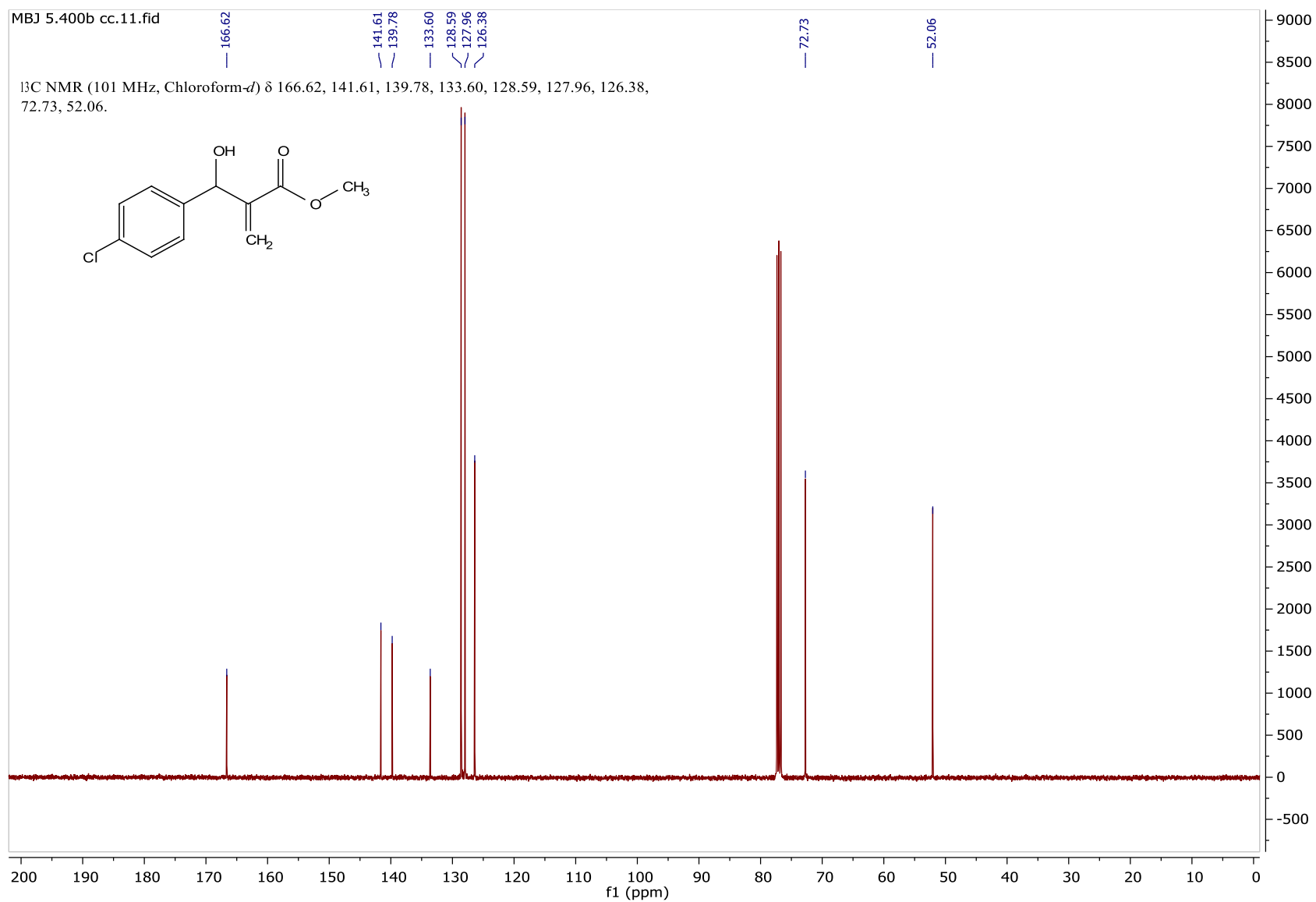


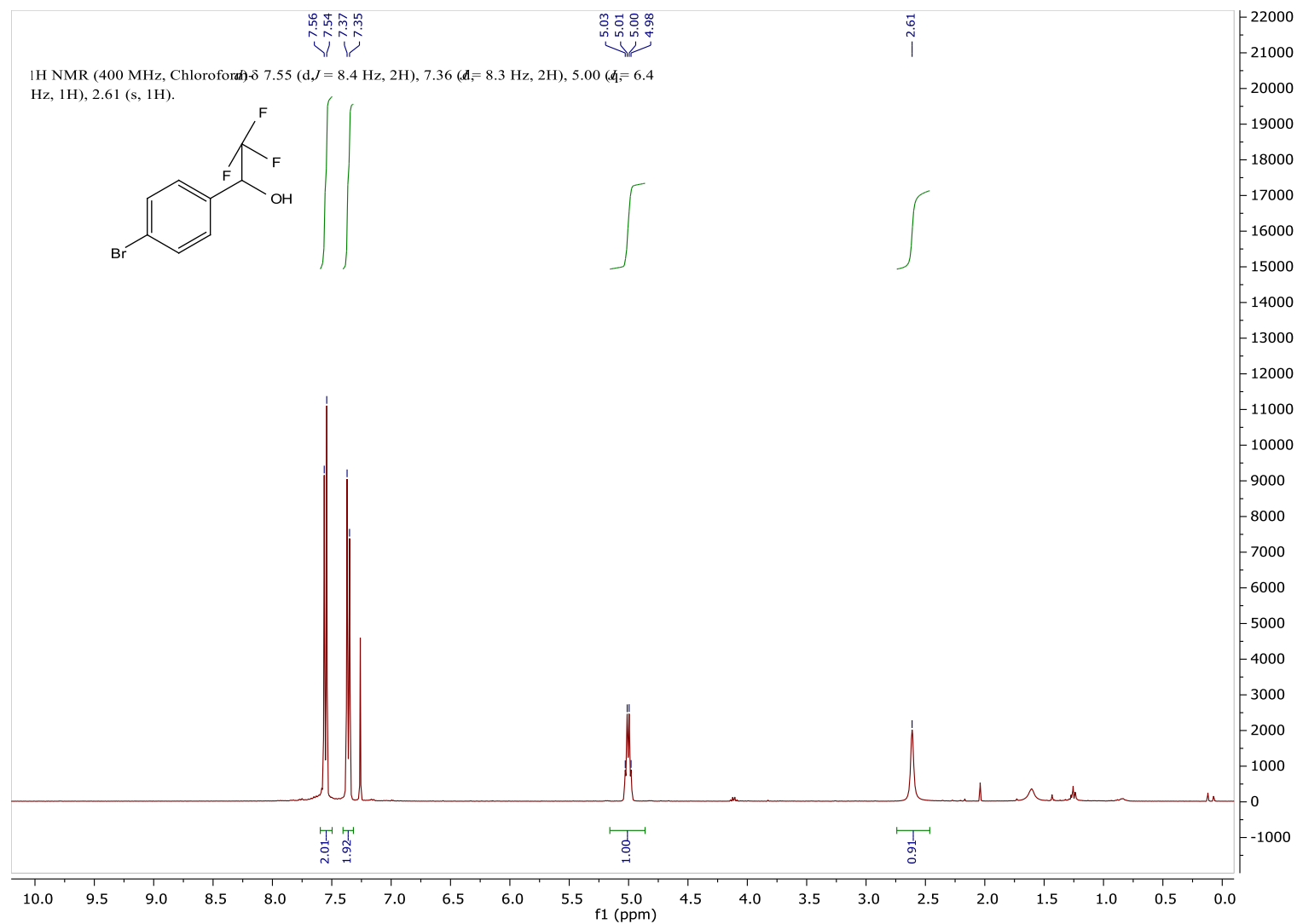
Methyl 2-((4-chlorophenyl)(hydroxy)methyl)acrylate (15**)**

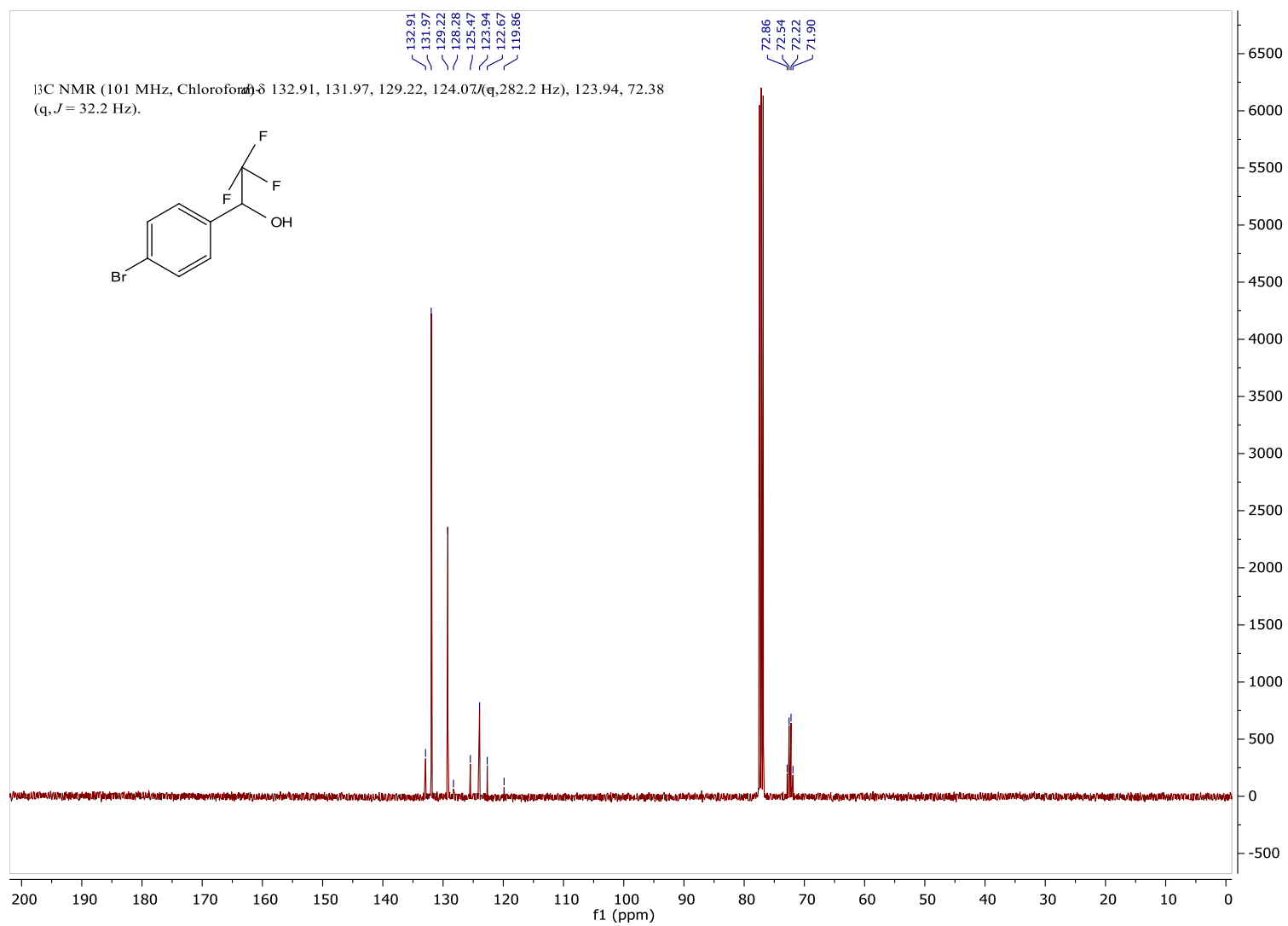
MBJ 5.400b cc.20.fid

¹H NMR (400 MHz, Chloroform-*d*) δ 7.31 (s, 3H), 6.34 (s, 1H), 5.82 (s, 1H), 5.52 (d, $J = 5.1$ Hz, 1H), 3.73 (s, 3H), 3.10 (d, $J = 5.5$ Hz, 1H).

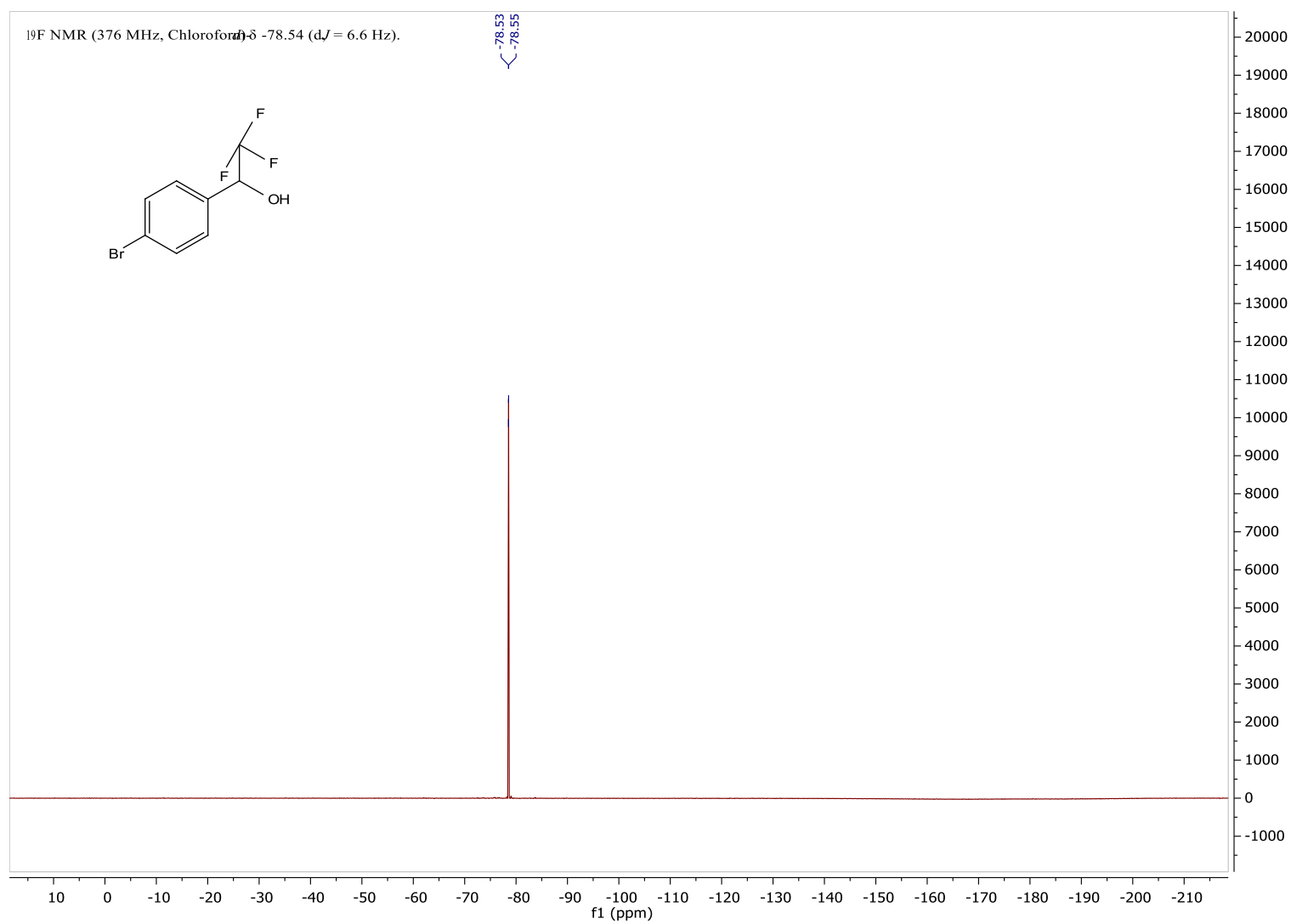


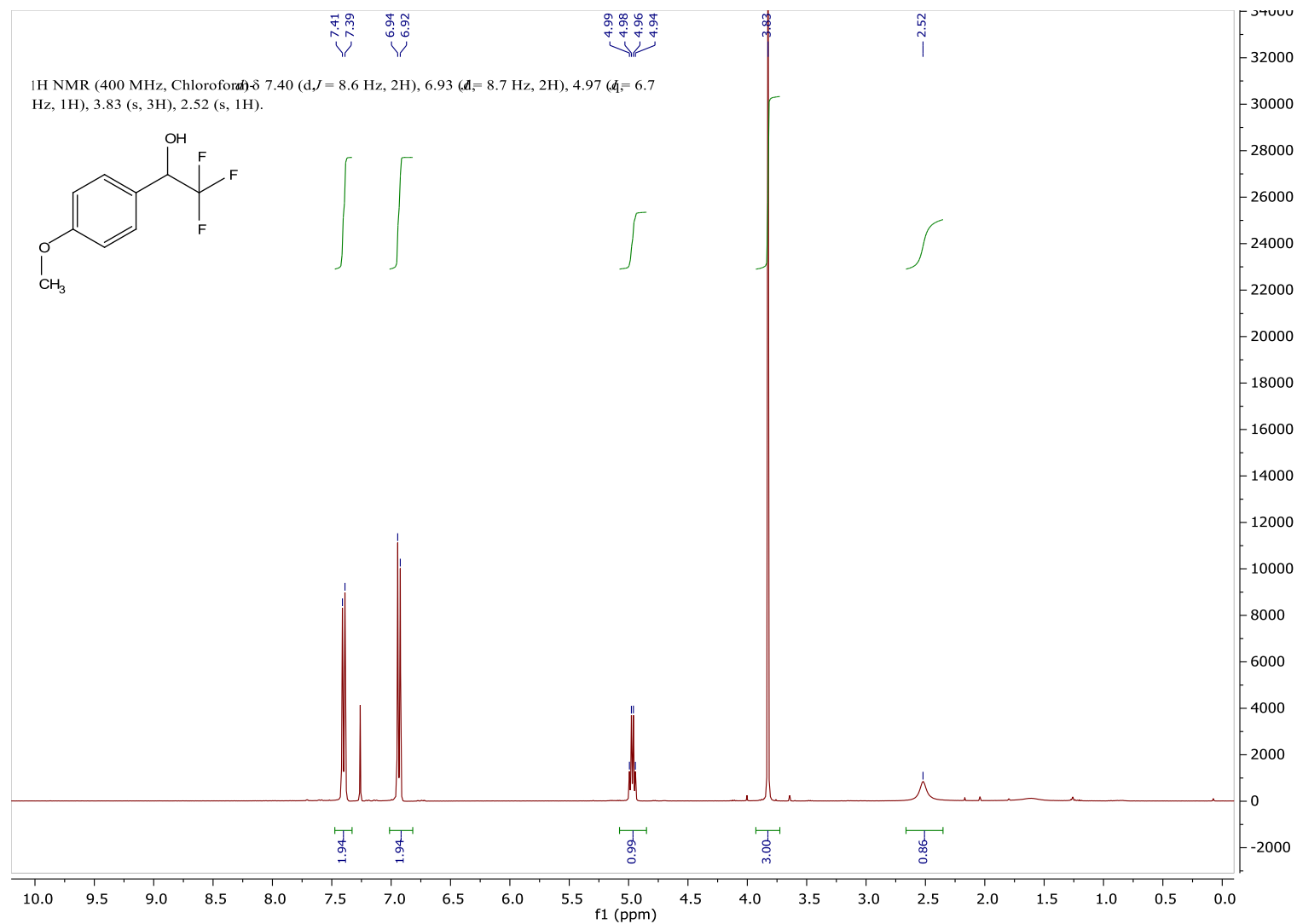


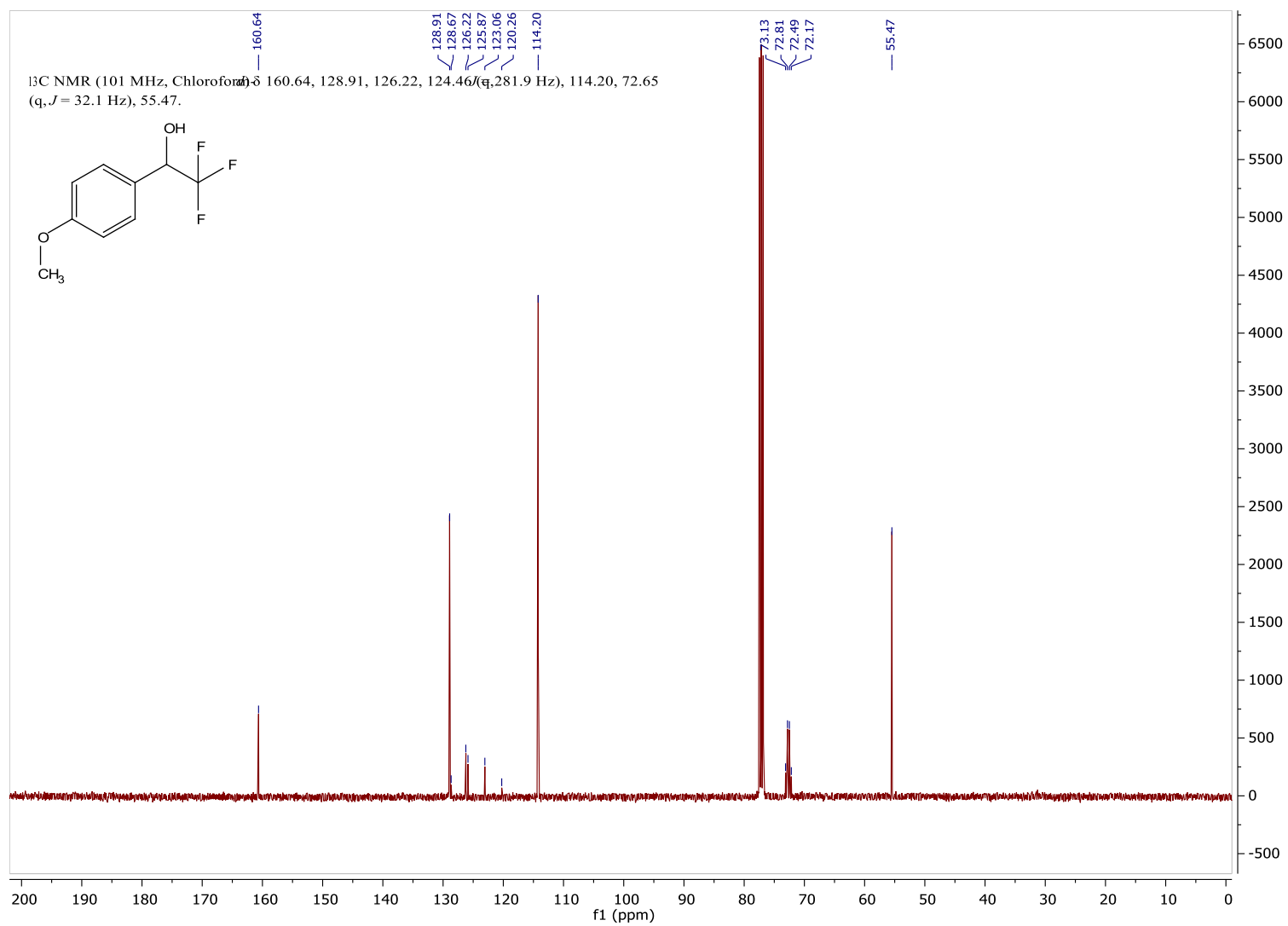
1-(4-Bromophenyl)-2,2,2-trifluoroethan-1-ol (16).



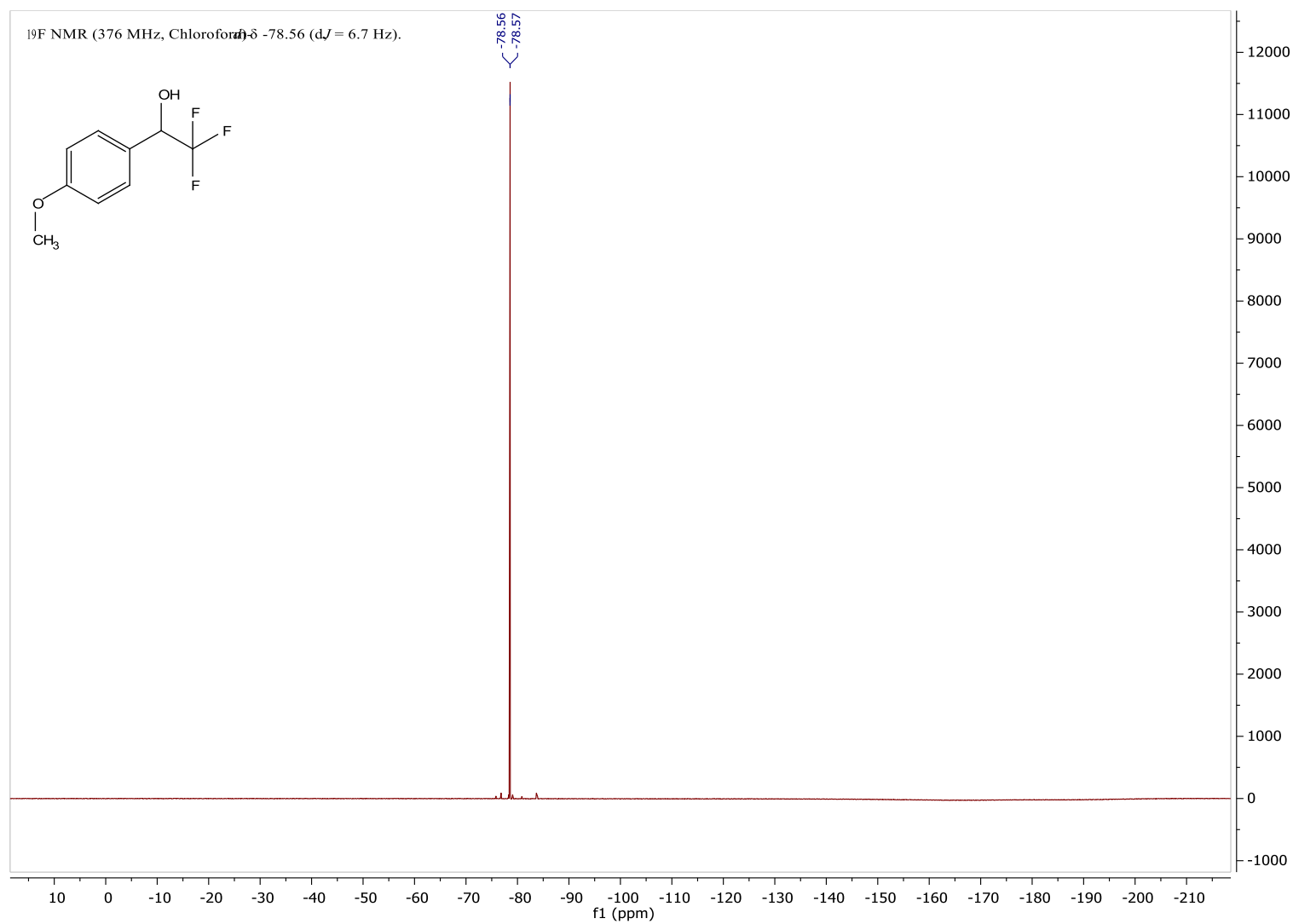
SI-73

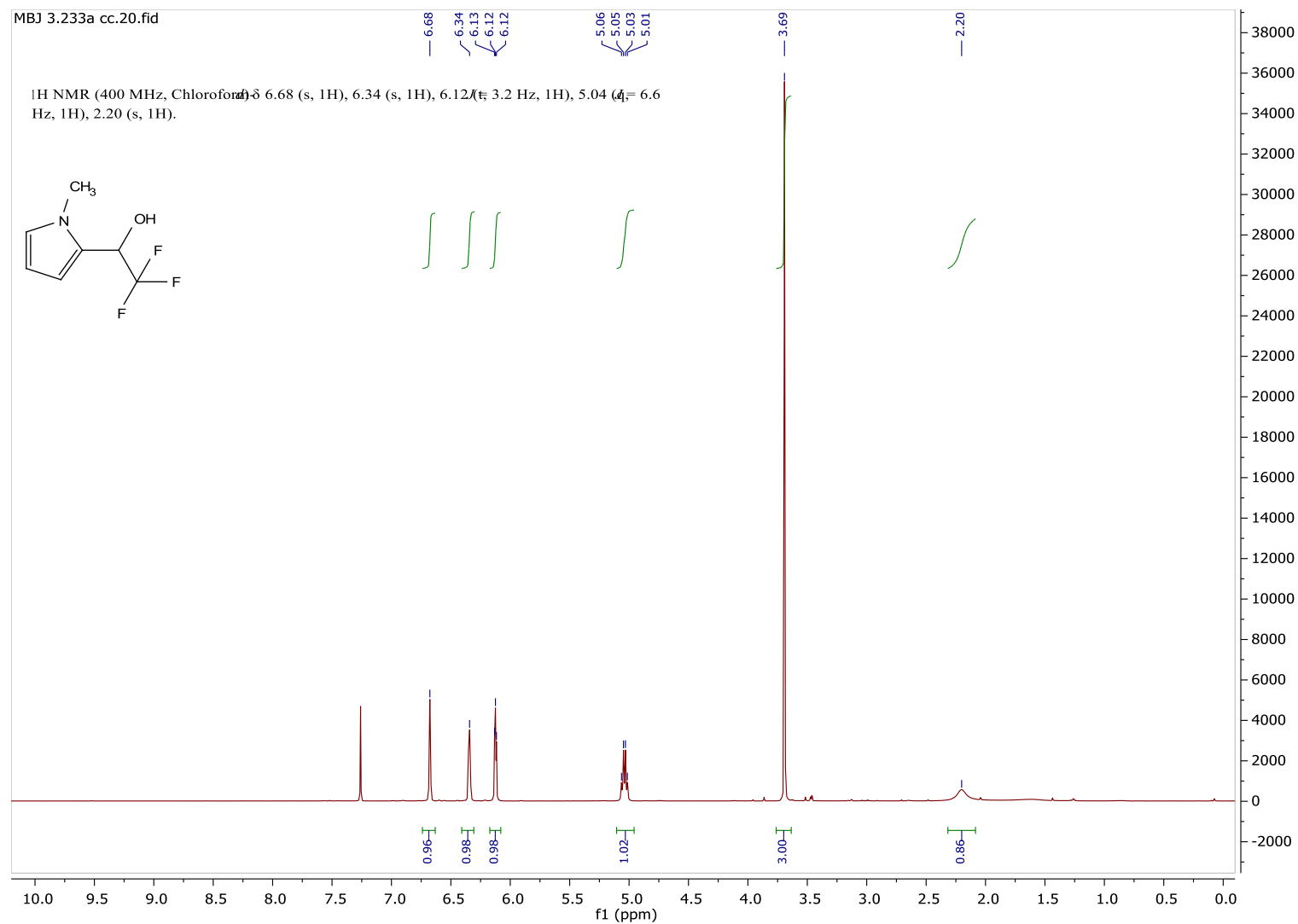


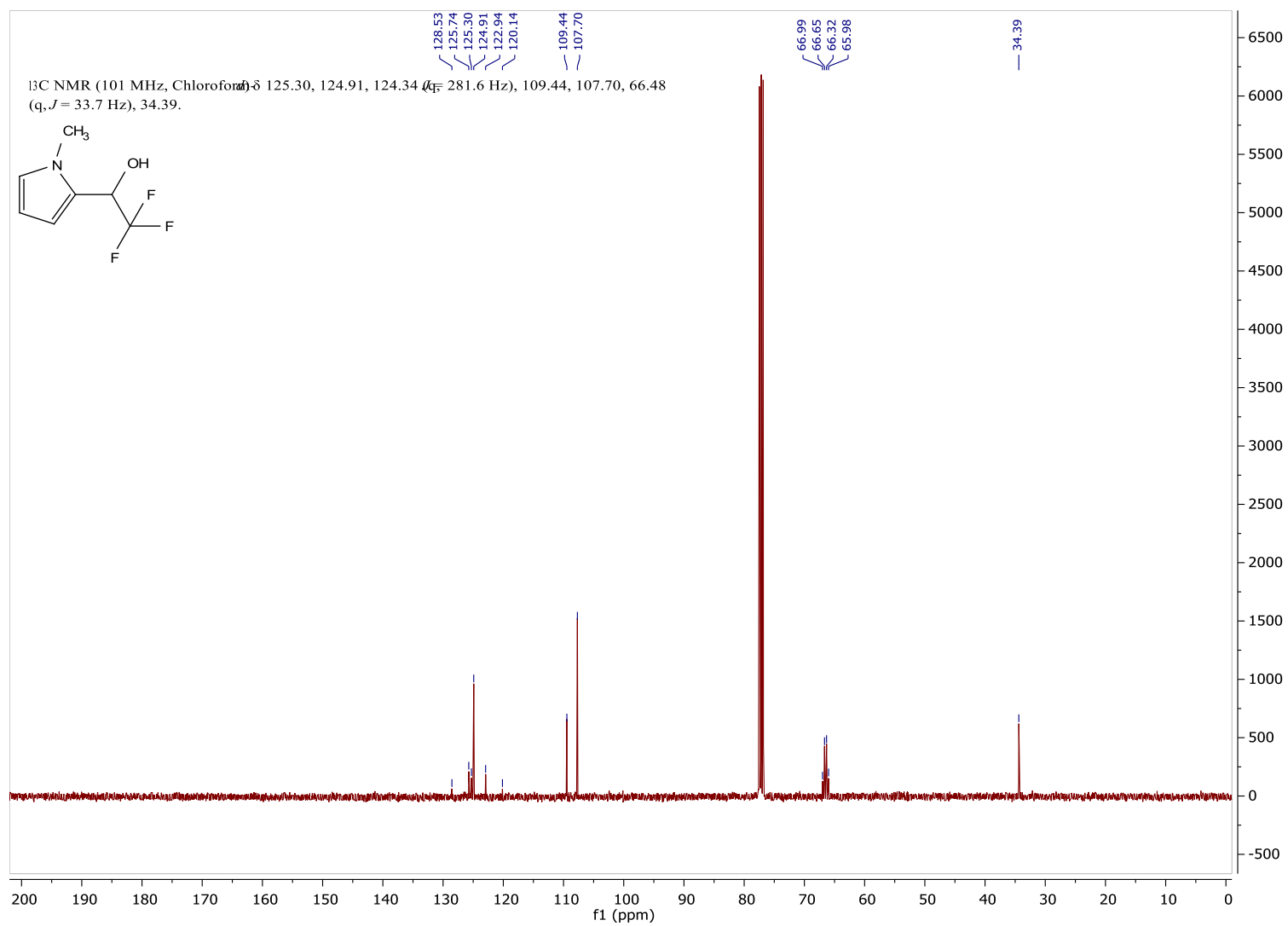
1-(4-Methoxyphenyl)-2,2,2-trifluoroethan-1-ol (17).

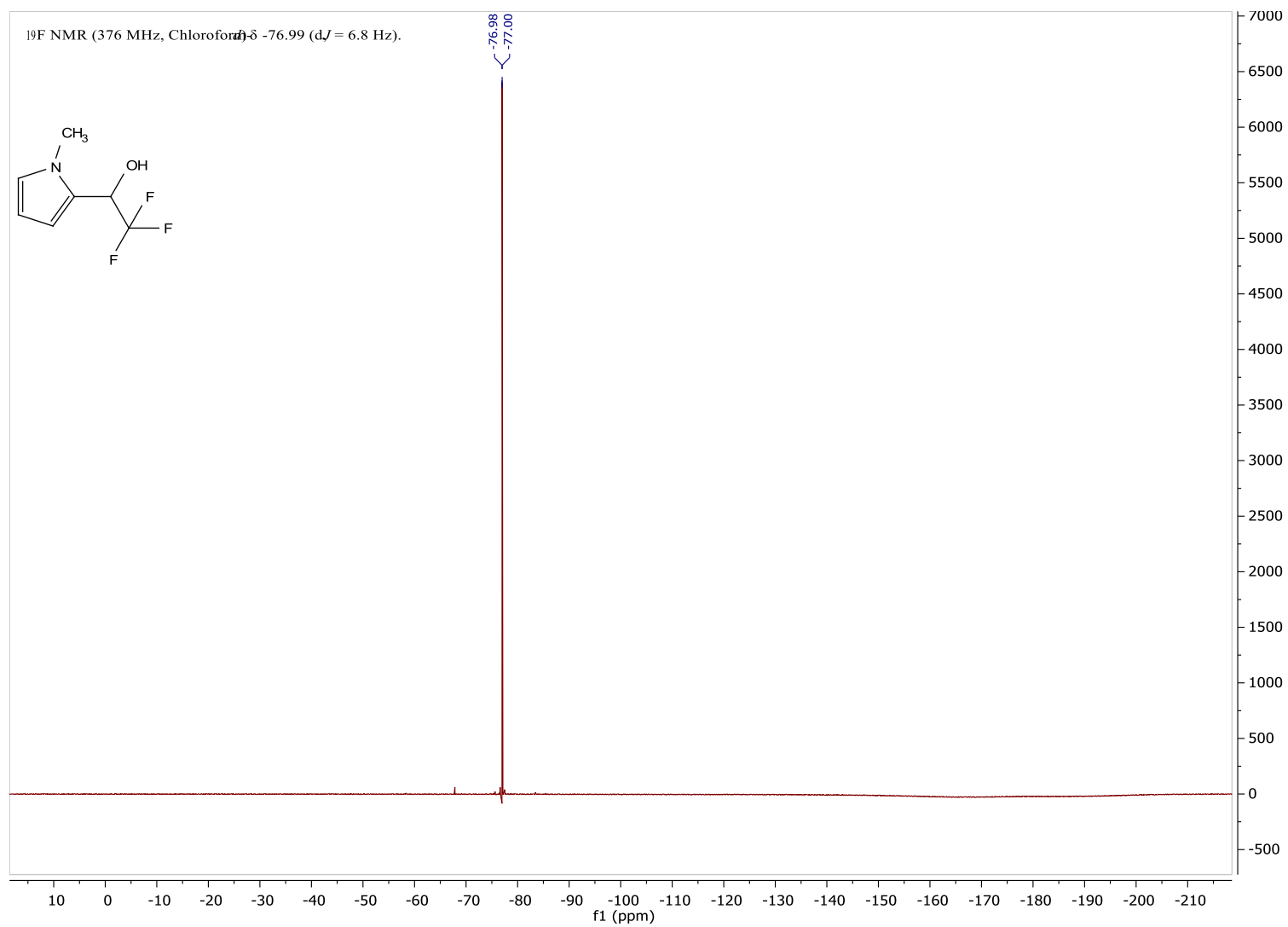


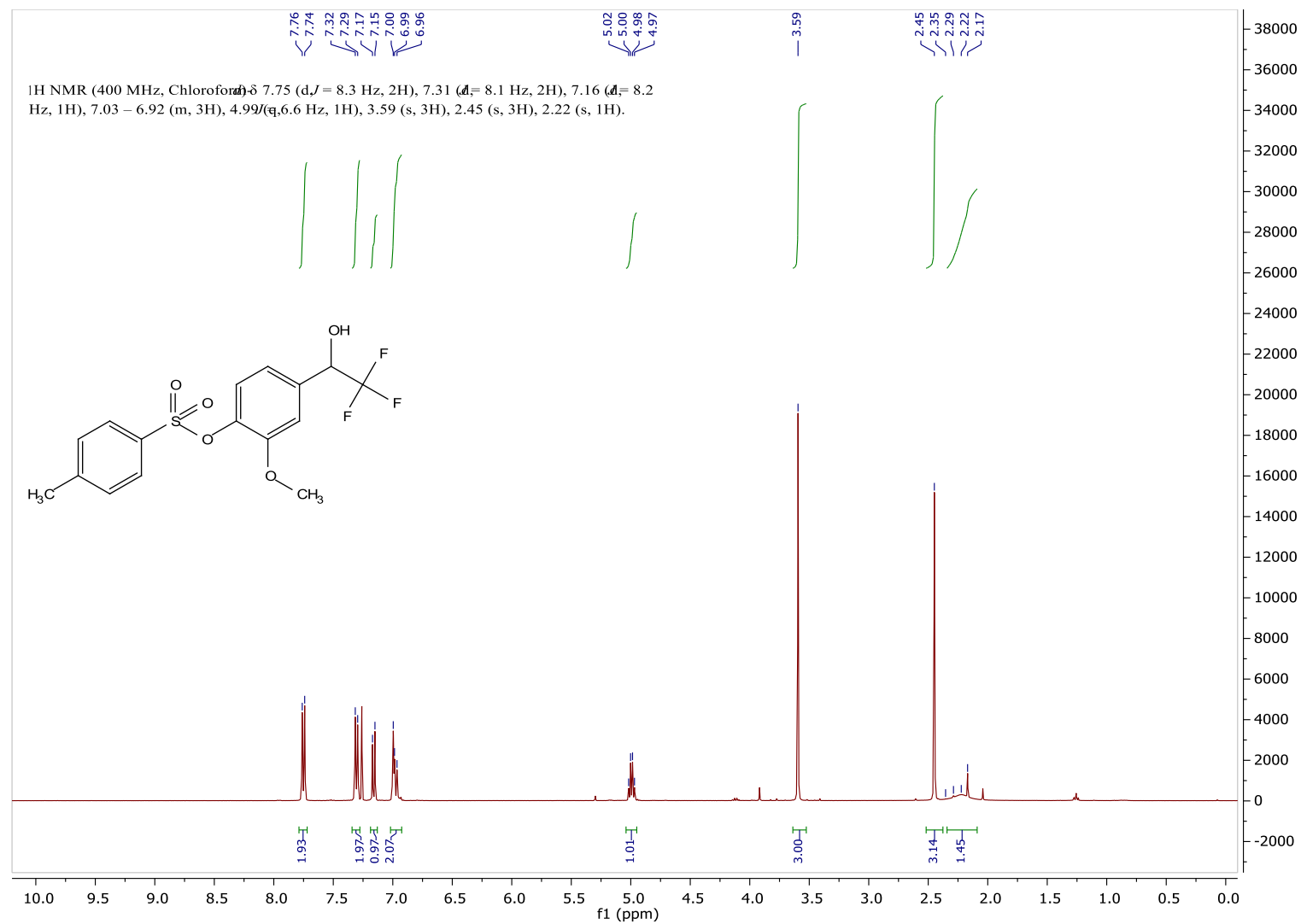
SI-76

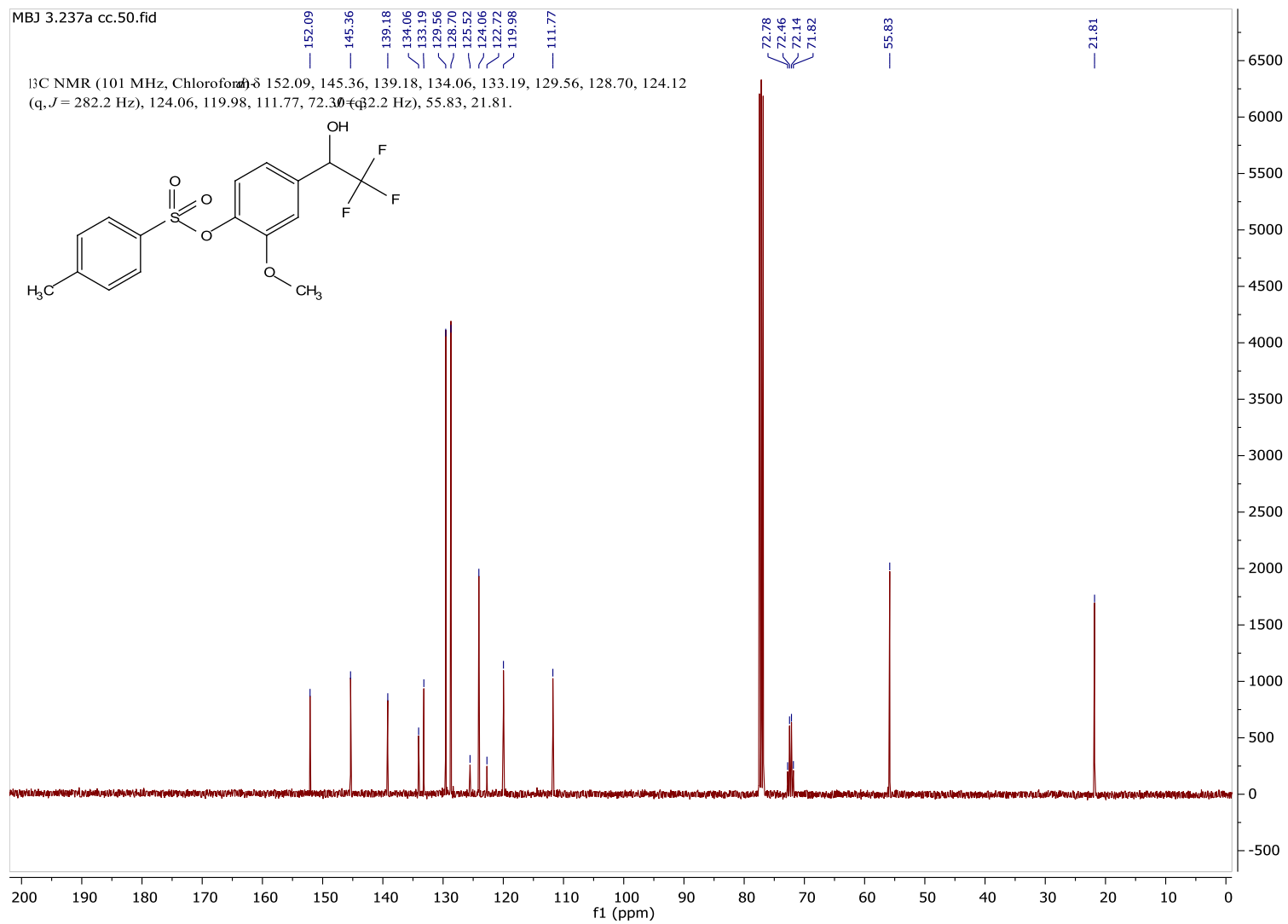


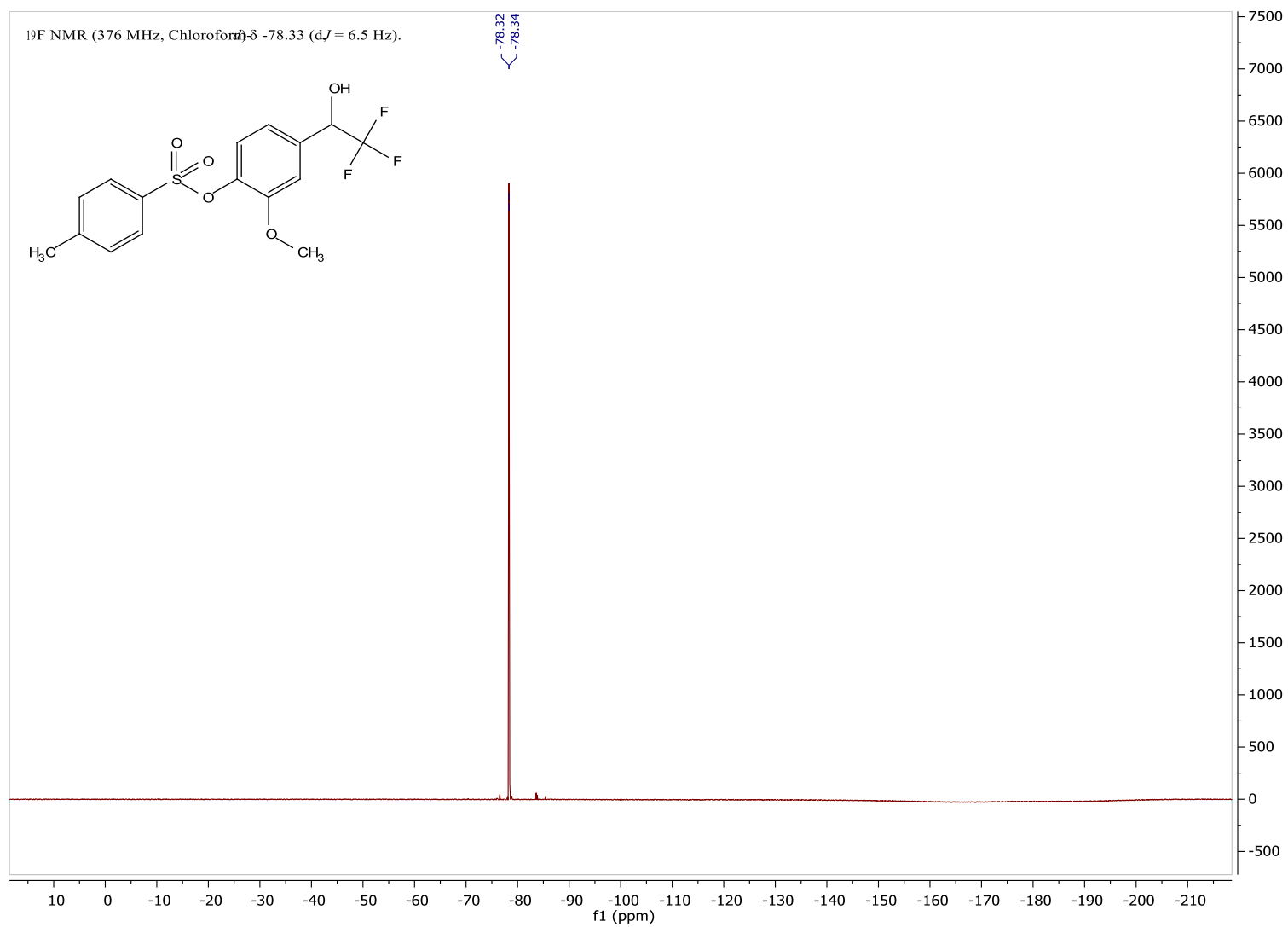
1-(N-methyl-1H-pyrrole-2-yl)-2,2,2-trifluoroethan-1-ol (18).

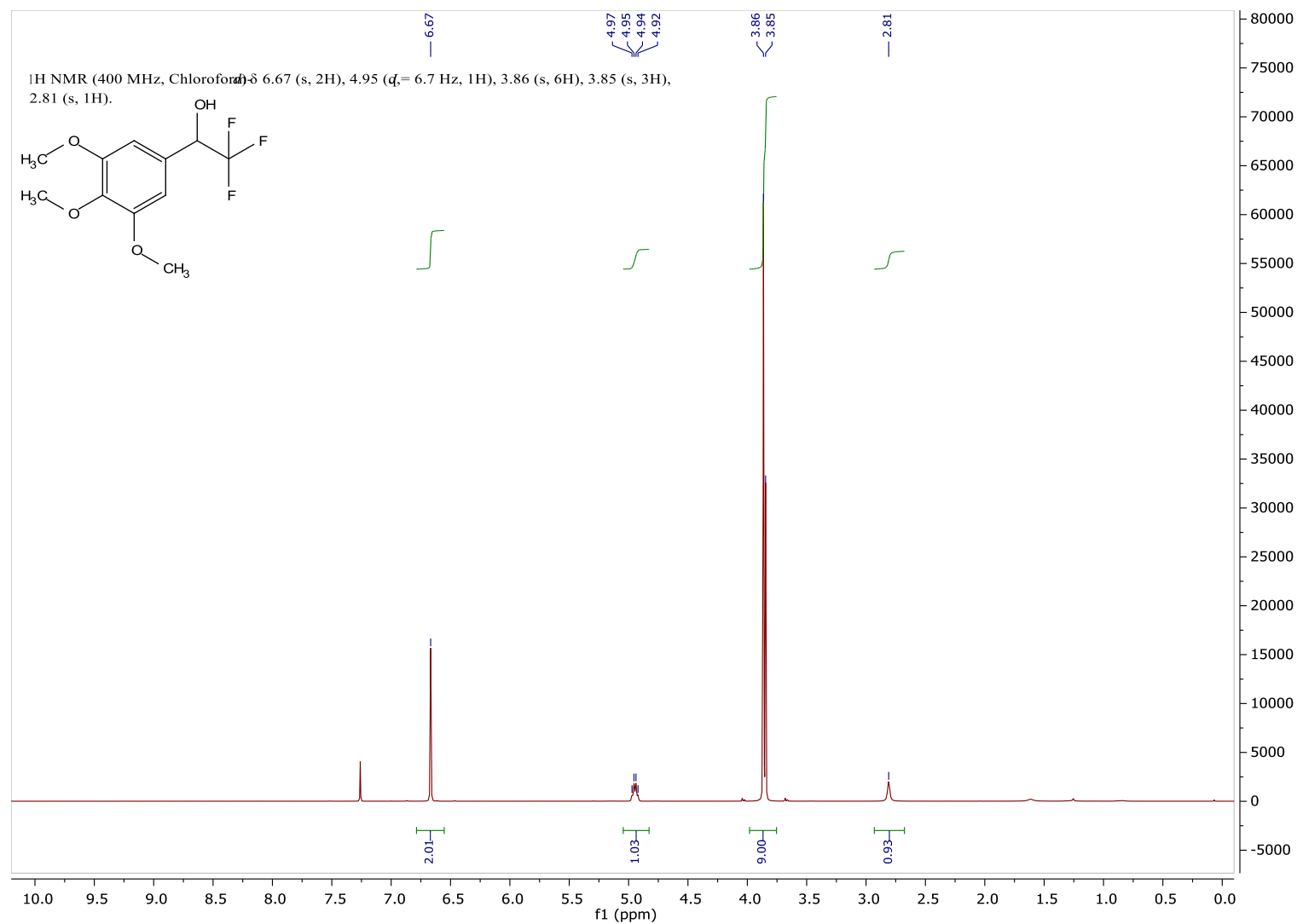


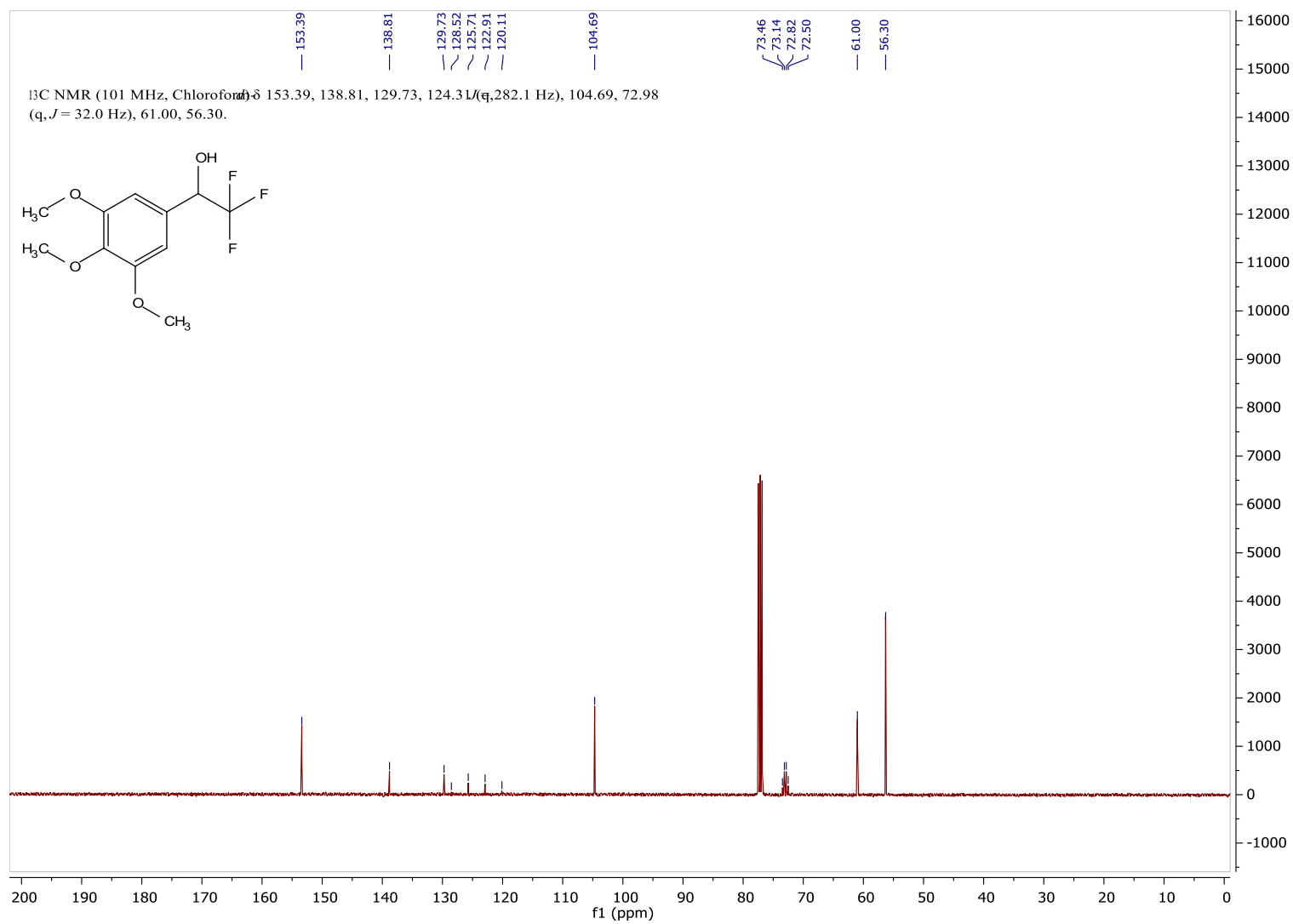


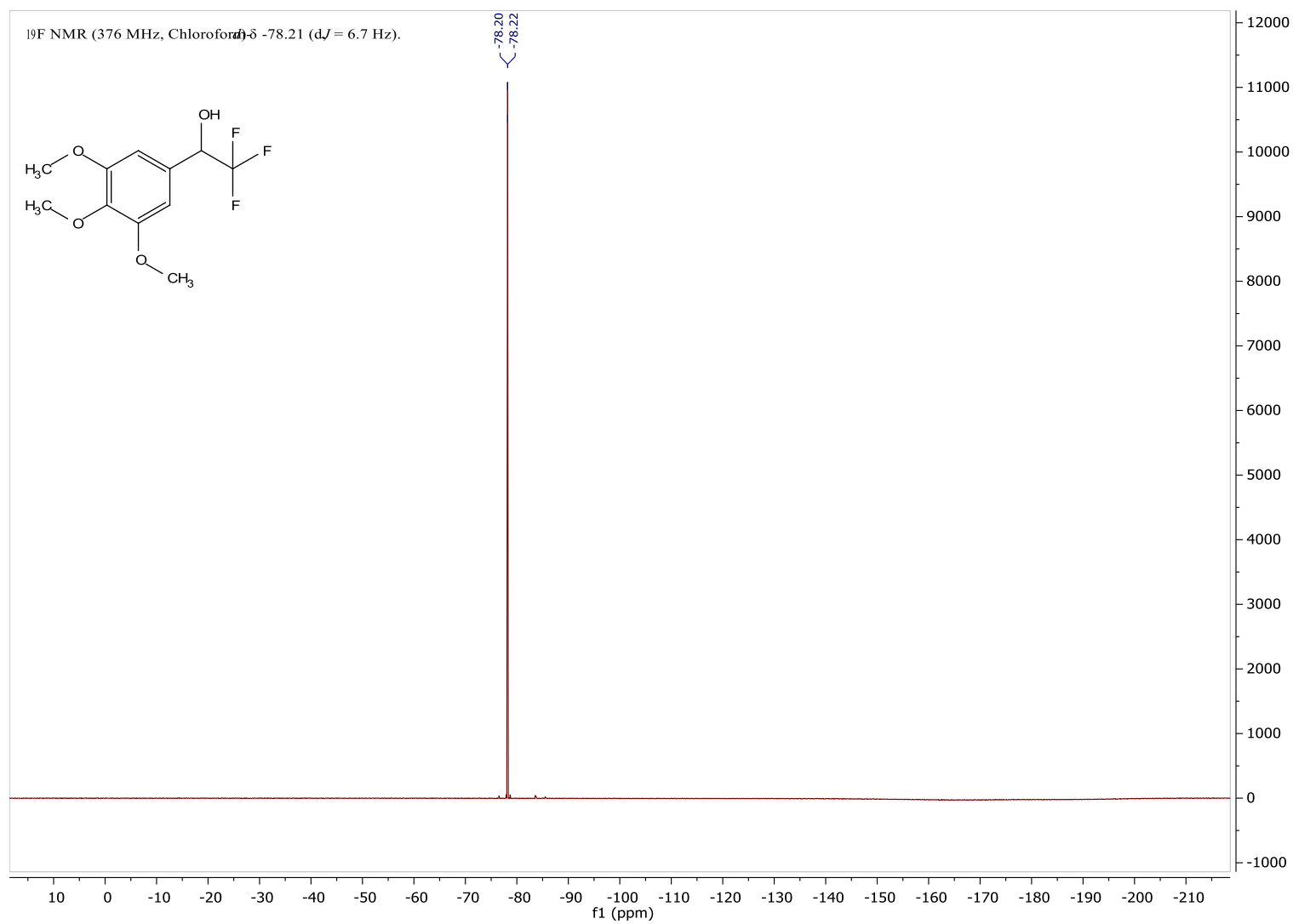
1-(3-Trimethoxy-4-tosyl-phenyl)-2,2,2-trifluoroethan-1-ol (19).

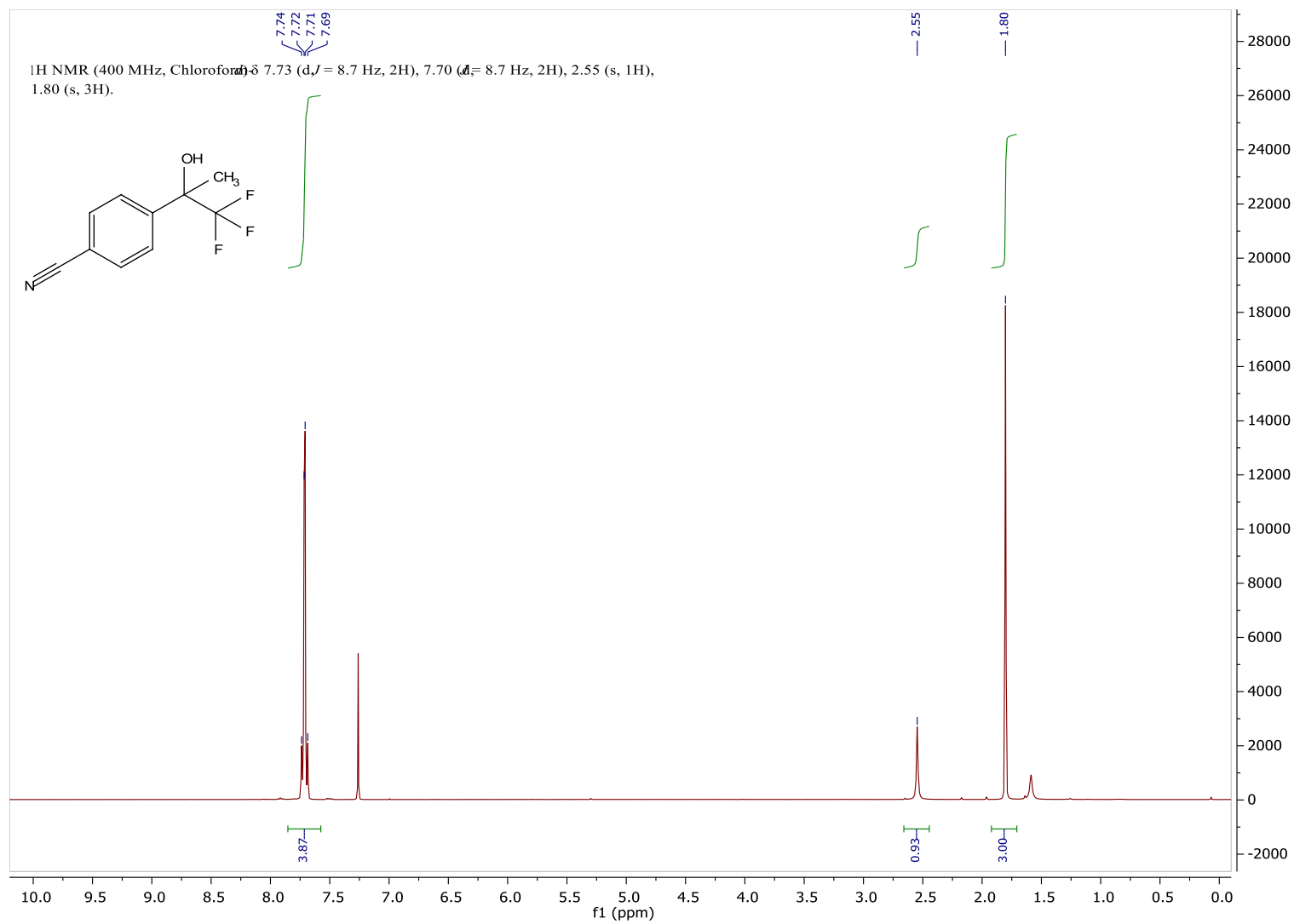


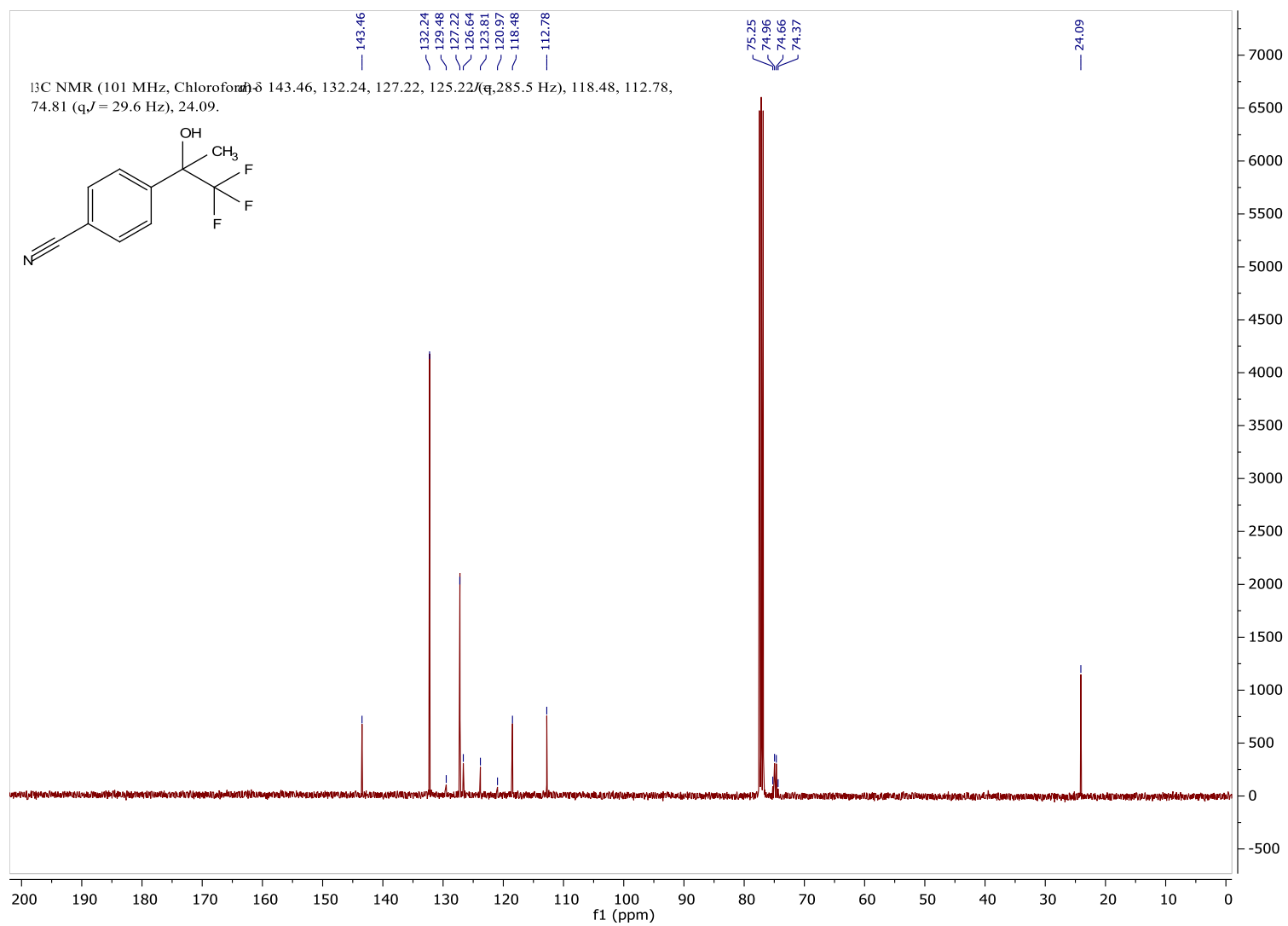


1-(3,4,5-Trimethoxyphenyl)-2,2,2-trifluoroethan-1-ol (20).

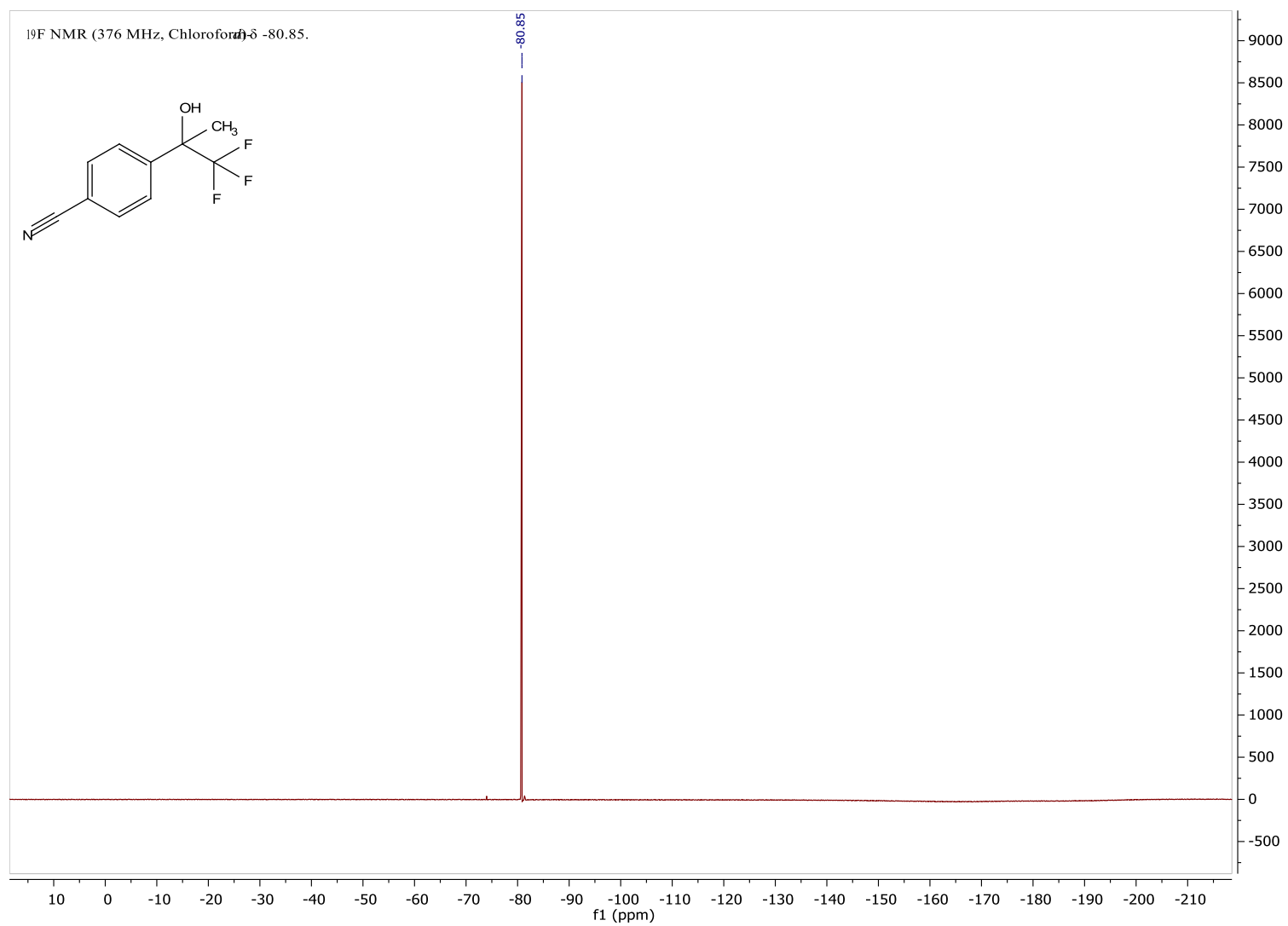




4-(1,1,1-Trifluoro-2-hydroxypropan-2-yl)benzonitrile (21).



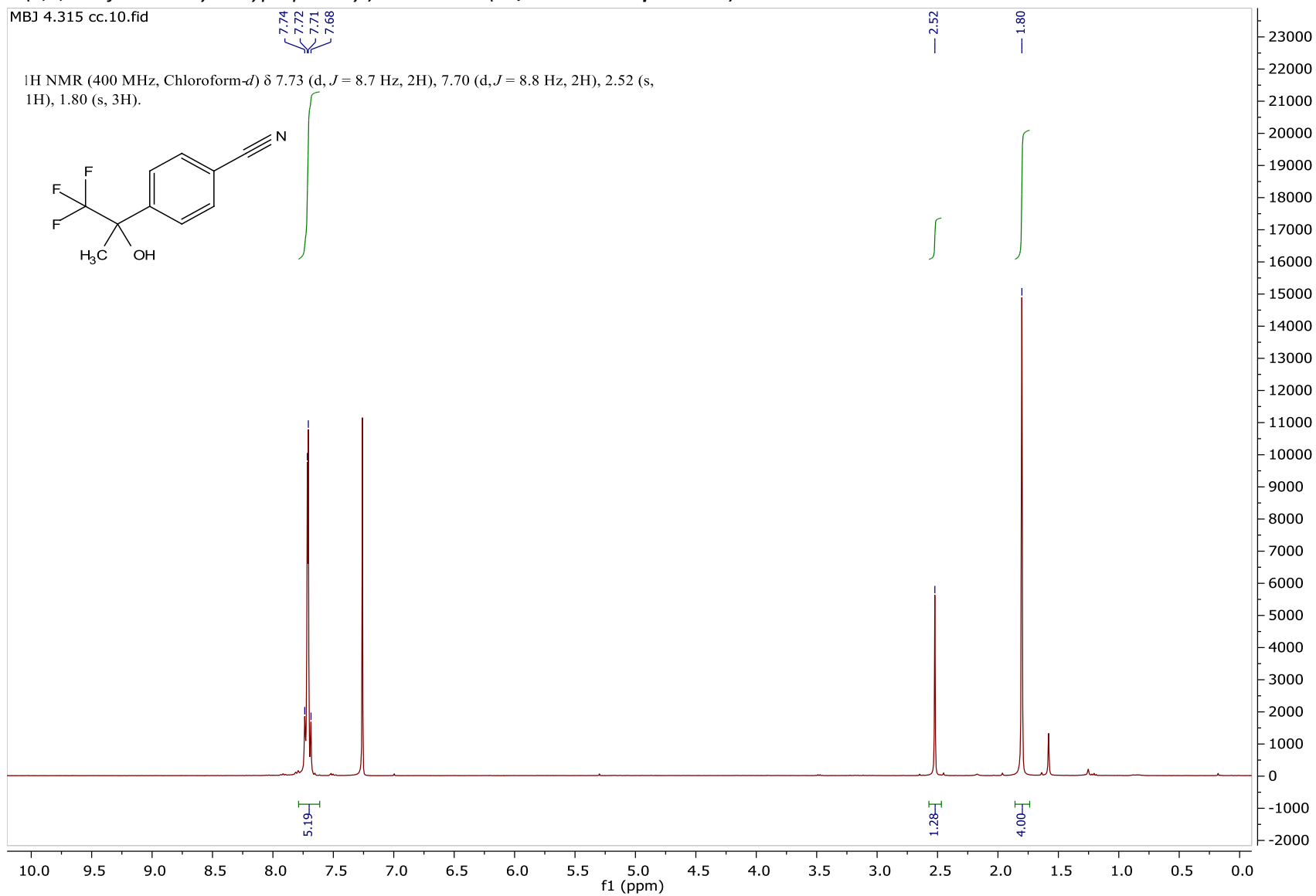
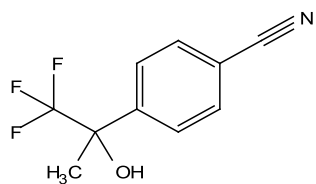
SI-88



4-(1,1,1-Trifluoro-2-hydroxypropan-2-yl)benzonitrile (21, Scale-out experiment)

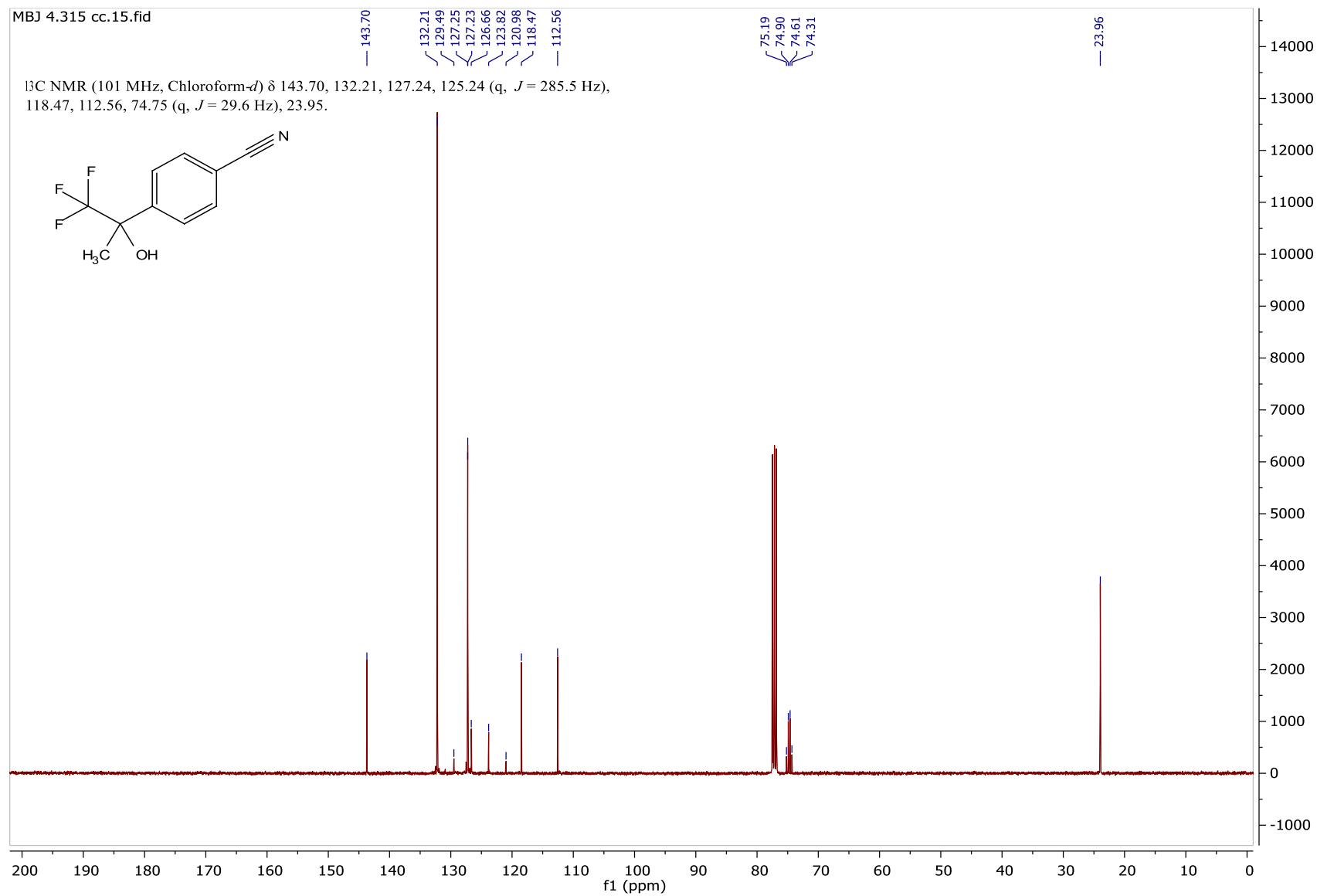
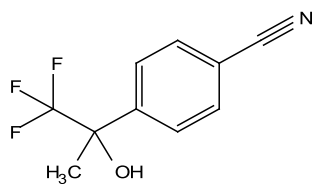
MBJ 4.315 cc.10.fid

¹H NMR (400 MHz, Chloroform-*d*) δ 7.73 (d, $J = 8.7$ Hz, 2H), 7.70 (d, $J = 8.8$ Hz, 2H), 2.52 (s, 1H), 1.80 (s, 3H).



MBJ 4.315 cc.15.fid

^{13}C NMR (101 MHz, Chloroform- d) δ 143.70, 132.21, 127.24, 125.24 (q, $J = 285.5$ Hz), 118.47, 112.56, 74.75 (q, $J = 29.6$ Hz), 23.95.



MBJ 4.315 cc.11.fid

¹⁹F NMR (376 MHz, Chloroform-*d*) δ -80.85.