Structural landscape of multicomponent solids based on sulfa drugs

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Table S1: Synthons considered for the calculation of energies as listed in Table 6.



Table S2. Occurrences of various SD–coformer synthons in the previously reported multicomponent solid forms of SDs with distinct organic coformers.

SD-Carboxylic acid	SD-amide	SD-pyridine	SD-2-aminopyridine	SD-amine	SD-banzamidine
$H_{2N} \xrightarrow{P} H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} $	$H_2N \longrightarrow H_2N \longrightarrow H_1 H_2N \longrightarrow H_2N \longrightarrow H_1 H_2N \longrightarrow H_2N \longrightarrow H_1 H_2N \longrightarrow H_2N$	$H_2N \xrightarrow{P} H_2N \xrightarrow{R_1} H_2N \xrightarrow{R_2} H_1$	$H_2N \longrightarrow H_1 H_1 H_1 H_1 H_1 H_1 H_1 H_1 H_1 H_1$	$H_2N \qquad \qquad$	$H_2N \xrightarrow{P}_{R_1} H_2N \xrightarrow{R_1}_{R_2} H_2N \xrightarrow{R_1}_{R_3} H_2N \xrightarrow{R_1}_$
$H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} H_{2N} \xrightarrow{R_{1}} H_{2N} \xrightarrow{R_{2}} H_{2N} R_{$	H_2N H_2N H_2N H_2N H_1 H_2N H_1 H_2N H_1 H_2N H_1 H_2N H_1 H_1 H_2N H_1 H	$H_2N \longrightarrow S^{R_1} H_1$		$H_2N \xrightarrow{P} H_2N \xrightarrow{R_1} H_1$	



Table S3. Occurrences of various SD-coformer synthons in the previously reported SD based solids and those reported in the present study(red colored), and the type of SD-SD synthons present in the input SDs. Various possible SD–coformer synthons are shown in Table 1 and Scheme 1.

	SD–SD synthon present in	SD-coformer synthon type (see Scheme 1)						
Sulfa drug (SD)	the most stable form of the parent SD	SD-acid	SD-amide	SD-pyridine	SD-amino pyridine	SD-amine	SD-bemza midine	
Sulfadiazine (sdz) H ₂ N- $(-)$ - $(-)$ -NH N- $(-)$ - $(-)$	$ \begin{array}{c} $			5		1		
Sulfamerazine (smr) $H_2N \longrightarrow H_2 N + H_3 + CH_3$	$\begin{array}{c} & & R_{3} \\ & & & \\ R_{1} \\ & & \\ R_{2} \\ & & \\ & & \\ R_{3} \end{array}$			1		1	1	
Sulfapyridine	$\begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$					1		





Sulfamethazine (smz) $H_2N \longrightarrow H_3C$	RZ N H. RZ RZ	19 2	3	3 1	2	1 1	
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Table S4. Classification of the previously reported SD based solids with various coformers that do not fit in Table S1 and those with various solvents based on whether or not the coformer/solvent disrupt the key SD–SD synthon present in the input SD. The reported salts with strong organic and inorganic acids shown in the last column are not considered in the analysis.

SD–SD synthon present		Coformer		S		
Sulfa drug	in the most stable form of the parent SD	SD-SD disrupted	SD-SD undisrupted	SD-SD disrupted	SD-SD undisrupted	Salt former (anion)
Sdz H ₂ N- N- N- N- N- N- N- N- N- N- N- N- N- N	$\begin{array}{c} 0 \\ R1 \\ R2 \\ R3 \end{array}$		(E)-1,2-bis(4- pyridyl)ethylene & 1,2-bis(4- pyridyl)acetylene			I ⁻ , BF ₄ ⁻ , NO ₃ ⁻ , CH ₃ -CH ₂ -SO ₃ ⁻ , HO-C ₆ H ₆ -SO ₃ ⁻ , Cl ⁻
Smr H ₂ N - SNH N - CH ₃	$\begin{array}{c} 0 \\ R_1 \\ R_2 \\ R_3 \end{array}$		2-methyl-1,4- nephthoquinone		1,4-dioxane (Form I & II), DMF, DMAc, Cyclopentanone, 3- methylpyridine, CH3CN + Crown Ether and Tetrahydrofuran	
Sulfapyridine	$\begin{array}{c} 0 \\ R_1 \\ R_2 \\ R_3 \end{array}$				Nitromethane, oxalic acid dibutyl ester, DMSO	





I ADIE 55. ΔD Ka values of 5D based solids forms reported in the present study	Table S5. /	Δp Ka values	of SD based	solids forms	reported in the	present study.
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Solid	<i>p</i> Ka of SD	<i>p</i> Ka of coformer	$\Delta p \text{Ka} = p \text{Ka}(\text{base})$ $- p \text{Ka}(\text{acid})$	solid form
1	6.99	8.95	1.96	Salt
2	6.99	9.56	2.57	Salt
3	6.16	8.95	2.79	Salt
4	6.16	9.56	3.4	Salt
5	-1.97	1.88	-3.85	Cocrystal
6	-1.97	3.88	-5.85	Cocrystal
7	6.99	8.95	1.96	Salt
8	6.99	9.56	2.57	Salt

pKa of SD's and coformers were calculated using ChemAxon calculator, Marvin 15.2.16.0 (http://www.chemaxon.com).



Solid 1





Solid 4







Solid 6







Figure S1. DSC scans of Sulfa drugs based solids 1–8 showing a single endothermic peak, except for the solid 3.

















Solid 5



Solid 6



Solid 7



Figure S2. Rietveld refinement plots for **1–8**, indicating the homogeneity of the bulk samples. The plot shows the experimental powder XRD profile (blue line), the calculated (red line), and blue tick lines shows Bragg positions in the bottom.

Footnote: *Peak(s) marked with star correspond to an unidentified phase.