

Electronic Supplementary Information (ESI)

Generation of cocrystals of Tavaborole (AN2690): opportunities for boron-containing APIs

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SI. Experimental information

All reagents were obtained from Sigma Aldrich and used without further purification. **AN2690** (Tavaborole) was synthesized and purified using a reported literature procedure.¹

Preparation of cocrystals: Cocrystals of composition **2(AN2690)·(bpy)**, **2(AN2690)·(bpe)**, **2(AN2690)·(bpeta)**, and **(AN2690)·(ina)** were obtained by dissolving **AN2690** (0.132 mmol) and the appropriate pyridine (0.066 mmol for **bpy**, **bpe** and **bpeta**, and 0.132 mmol for **ina**) in methanol. Slow evaporation over 2 days yielded single crystals suitable for X-ray diffraction.

S2. Single-crystal X-ray diffraction measurements

Single-crystal X-ray diffraction measurements. Single crystal XRD for the cocrystals was measured on a Nonius Kappa CCD single-crystal X-ray diffractometer using MoK_α radiation ($\lambda=0.71073 \text{ \AA}$). Structure solution and refinement were accomplished using SHELXS and SHELXL, respectively² within the Olex2³ graphical user interface. All non-hydrogen atoms were refined anisotropically.

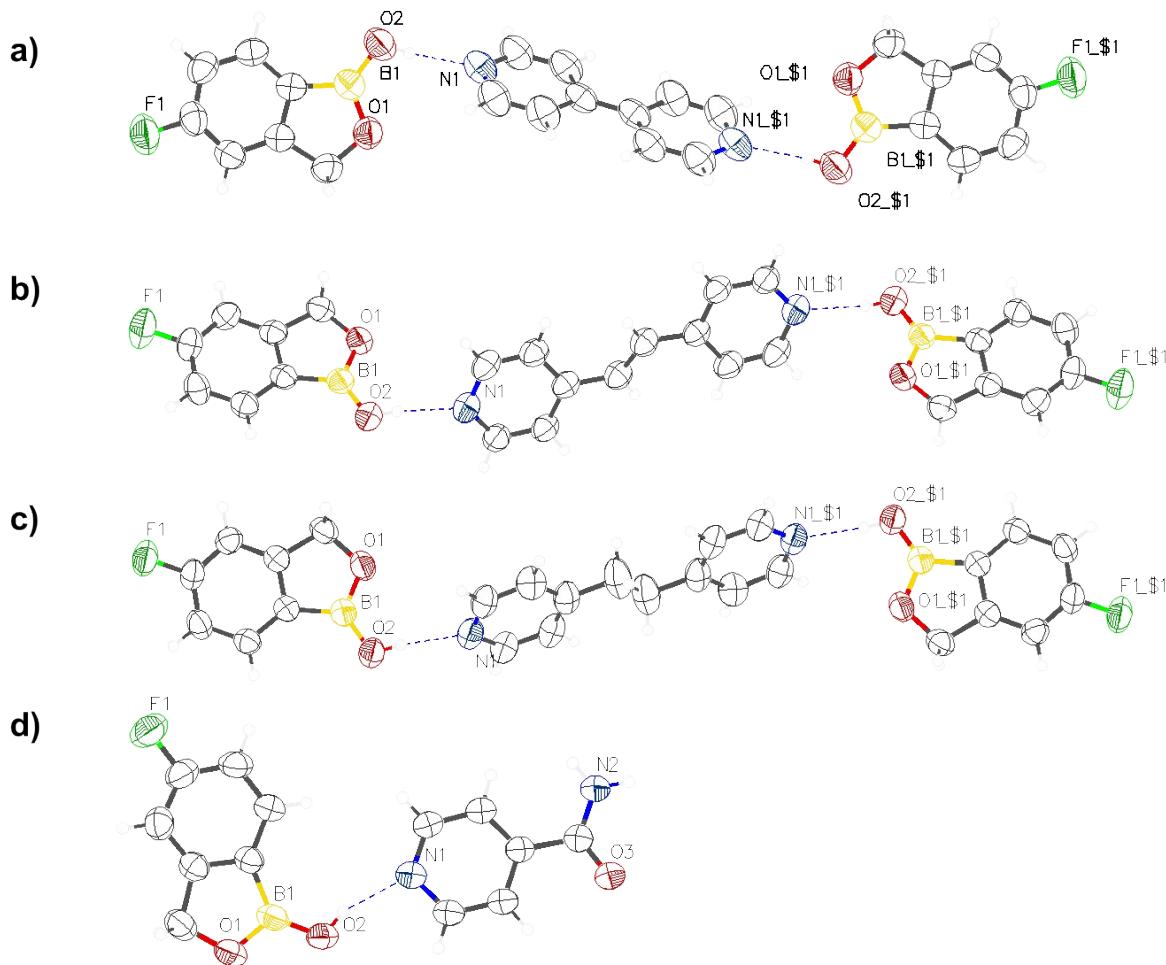


Fig. S1. ORTEP representation of a) 2(AN2690)·(bpy), b) 2(AN2690)·(bpe), c) 2(AN2690)·(bpeta) and (AN2690)·(ina) with ellipsoids drawn at the 50% probability level. [Symmetry codes (_\$1): 2(AN2690)·(bpe): -x + 1, -y, -z ; 2(AN2690)·(bpe): -x, -y + 2, -z; 2(AN2690)·(bpeta): -x, -y + 2, -z].

Table S1. Crystallographic parameters for 2(**AN2690**)·(**bpy**).

Compound name	2(AN2690)·(bpy)
Empirical formula	C ₁₂ H ₁₀ BFNO ₂
Formula weight	230.02
Temperature/K	298.15
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.5378(15)
b/Å	6.0517(6)
c/Å	14.4276(14)
α/°	90
β/°	118.366(5)
γ/°	90
Volume/Å ³	1116.9(2)
Z	4
ρ _{calc} g/cm ³	1.368
μ/mm ⁻¹	0.103
F(000)	476.0
Crystal size/mm ³	0.26 × 0.24 × 0.18
Radiation	MoKα ($\lambda = 0.71073$)
2θ range for data collection/°	6.37 to 50.75
Index ranges	-17 ≤ h ≤ 17, -7 ≤ k ≤ 7, -17 ≤ l ≤ 17
Reflections collected	6425
Independent reflections	2036 [R _{int} = 0.0279, R _{sigma} = 0.0245]
Data/restraints/parameters	2036/0/155
Goodness-of-fit on F ²	1.088
Final R indexes [I>=2σ (I)]	R ₁ = 0.0413, wR ₂ = 0.1033
Final R indexes [all data]	R ₁ = 0.0595, wR ₂ = 0.1153
Largest diff. peak/hole / e Å ⁻³	0.11/-0.14
CCDC deposition number	1531822

Table S2. Crystallographic parameters for 2(**AN2690**)·(**bpe**).

Compound name	2(AN2690)·(bpe)
Empirical formula	C ₁₃ H ₁₁ BFNO ₂
Formula weight	243.04
Temperature/K	298.15
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.8742(15)
b/Å	5.9305(6)
c/Å	14.3285(14)
α/°	90
β/°	109.171(5)
γ/°	90
Volume/Å ³	1193.8(2)
Z	4
ρ _{calcd} g/cm ³	1.352
μ/mm ⁻¹	0.100
F(000)	504.0
Crystal size/mm ³	0.39 × 0.39 × 0.11
Radiation	MoKα ($\lambda = 0.71073$)
2θ range for data collection/°	5.8 to 50.558
Index ranges	-16 ≤ h ≤ 17, -7 ≤ k ≤ 7, -17 ≤ l ≤ 17
Reflections collected	6833
Independent reflections	2162 [R _{int} = 0.0225, R _{sigma} = 0.0210]
Data/restraints/parameters	2162/0/164
Goodness-of-fit on F ²	0.999
Final R indexes [I>=2σ (I)]	R ₁ = 0.0374, wR ₂ = 0.1067
Final R indexes [all data]	R ₁ = 0.0484, wR ₂ = 0.1186
Largest diff. peak/hole / e Å ⁻³	0.14/-0.14
CCDC deposition number	1531823

Table S3. Crystallographic parameters for 2(**AN2690**)·(**bpet**a).

Compound name	2(AN2690)·(bpet a)
Empirical formula	C ₁₃ H ₁₂ BFNO ₂

Formula weight	244.05
Temperature/K	298.15
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	15.0294(15)
b/Å	5.9476(6)
c/Å	14.5384(15)
$\alpha/^\circ$	90
$\beta/^\circ$	110.690(5)
$\gamma/^\circ$	90
Volume/Å ³	1215.8(2)
Z	4
$\rho_{\text{calc}} \text{g/cm}^3$	1.333
μ/mm^{-1}	0.099
F(000)	508.0
Crystal size/mm ³	0.34 × 0.23 × 0.14
Radiation	MoKα ($\lambda = 0.71073$)
2 Θ range for data collection/°	5.658 to 50.734
Index ranges	-18 ≤ h ≤ 18, -7 ≤ k ≤ 7, -17 ≤ l ≤ 17
Reflections collected	6955
Independent reflections	2208 [$R_{\text{int}} = 0.0200$, $R_{\text{sigma}} = 0.0229$]
Data/restraints/parameters	2208/0/164
Goodness-of-fit on F ²	1.020
Final R indexes [I>=2σ (I)]	$R_1 = 0.0467$, $wR_2 = 0.1319$
Final R indexes [all data]	$R_1 = 0.0599$, $wR_2 = 0.1443$
Largest diff. peak/hole / e Å ⁻³	0.30/-0.27
CCDC deposition number	1531824

Table S4. Crystallographic parameters for **(AN2690)·(ina)**.

Compound name	(AN2690)·(ina)
Empirical formula	C ₁₃ H ₁₂ BFN ₂ O ₃
Formula weight	274.06
Temperature/K	298.15

Crystal system	monoclinic					
Space group	P2 ₁ /c					
cocrystal	D-H···A	<i>d</i> (D-H) (Å)	<i>d</i> (H···A) (Å)	<i>d</i> (D···A) (Å)	θ (D-H···A) (deg)	symmetry code
2(AN2690)·(bpy)	O2-H2···N1	0.82	1.93	2.743(2)	174	
	C7-H7A···O2	0.97	2.49	3.422(2)	161	x, y + 1, z
	C5-H5···F1	0.93	2.63	3.434(2)	145	x - 2, -y + 1, -z + 2
2(AN2690)·(bpe)	O2-H2···N1	0.82	1.92	2.740(2)	173	
	C7-H7B···O2	0.97	2.51	3.432(2)	158	x, y - 1, z
	C9-H9···O2	0.93	2.61	3.492(2)	158	x, -y + 5/2, z - 1/2
	C5-H5···F1	0.93	2.63	3.424(2)	144	-x + 1, -y + 1, -z + 2
2(AN2690)·(bpeta)	O2-H2···N1	0.82	1.92	2.737(2)	173	
	C7-H7B···O2	0.97	2.51	3.435(2)	158	x, y - 1, z
(AN2690)·(ina)	O2-H2···N1	0.82	1.93	2.707(2)	158.5	
a/Å	15.8870(16)					
b/Å	5.9848(6)					
c/Å	13.8225(14)					
α /°	90					
β /°	93.469(5)					
γ /°	90					
Volume/Å ³	1311.8(2)					
Z	4					
ρ_{calc} g/cm ³	1.388					
μ/mm^{-1}	0.107					
F(000)	568.0					
Crystal size/mm ³	0.27 × 0.26 × 0.05					
Radiation	MoK α ($\lambda = 0.71073$)					
2 Θ range for data collection/°	5.138 to 50.726					
Index ranges	$-19 \leq h \leq 17, -7 \leq k \leq 6, -16 \leq l \leq 16$					
Reflections collected	7571					
Independent reflections	2398 [$R_{\text{int}} = 0.0371, R_{\text{sigma}} = 0.0342$]					
Data/restraints/parameters	2398/0/182					
Goodness-of-fit on F ²	1.069					
Final R indexes [I>=2σ(I)]	$R_1 = 0.0468, wR_2 = 0.1170$					
Final R indexes [all data]	$R_1 = 0.0794, wR_2 = 0.1346$					
Largest diff. peak/hole / e Å ⁻³	0.25/-0.15					
CCDC deposition number	1531825					

	N2-H2A···O3	0.86	2.05	2.902(2)	171	-x + 1, -y + 3, -z + 1	
	N2-H2B···O2	0.86	2.28	3.110(2)	162	x, -y + 3/2, z - 1/2	
	C5-H5···F1	0.93	2.59	3.404(3)	147	-x + 2, -y, -z + 1	
	C7-H7B···F1	0.97	2.52	3.412(3)	152	x, -y - 1/2, z + 1/2	
	C8-H8···O3	0.93	2.46	3.173(2)	133	-x + 1, y - 1/2, -z + 3/2	

S4. Hydrogen bond table

¹ calculated from pyridine ring plane and plane generated by atoms O2, B1, O1, C7, C2 and C1.

S4. FTIR Data

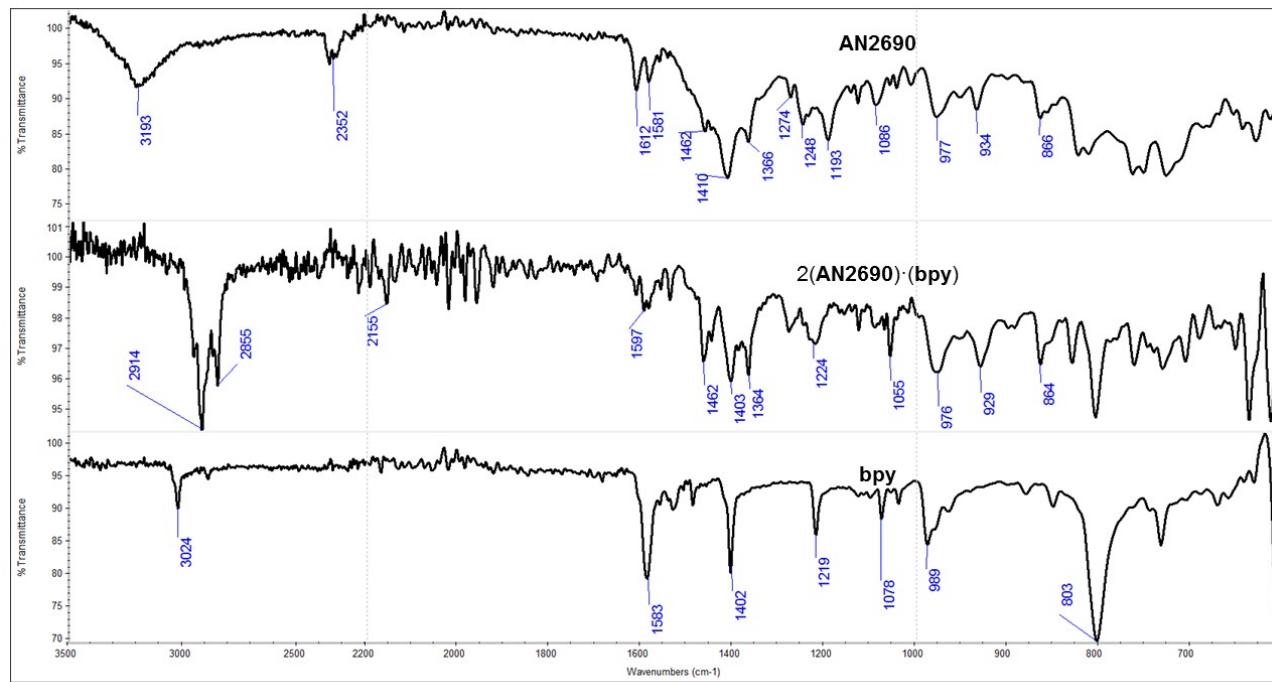


Fig. S2. FTIR spectra of cocrystal 2(AN2690)·(bpy), and starting materials.

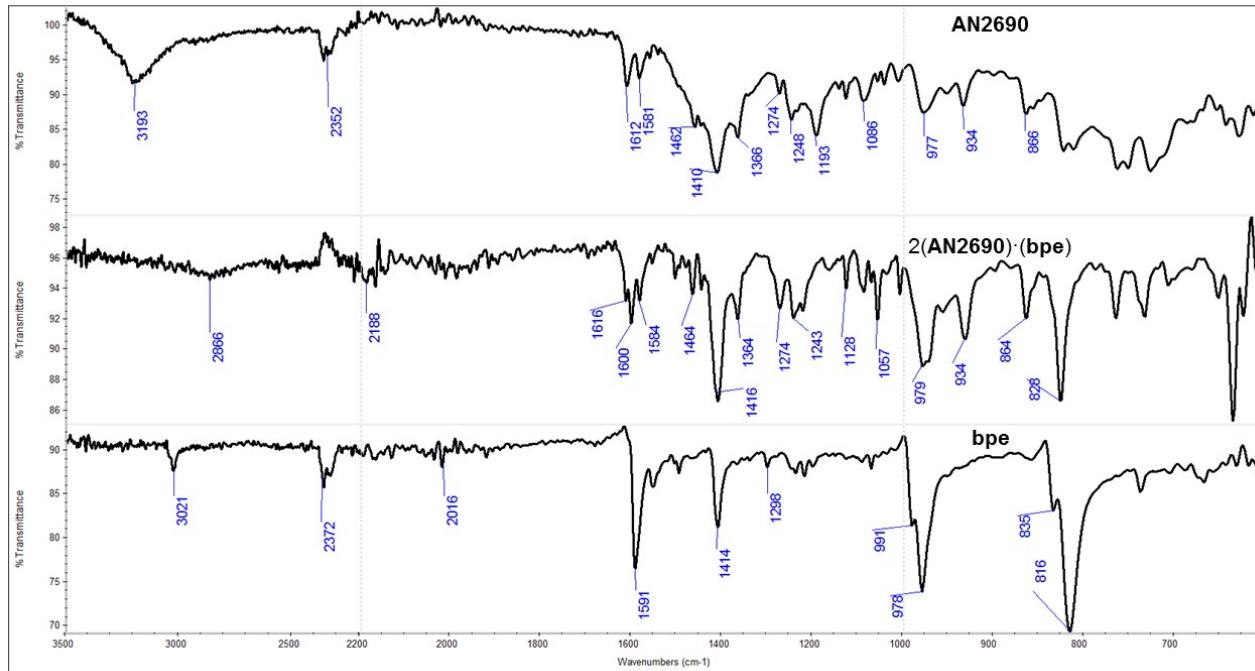


Fig. S3. FTIR spectra of cocrystal 2(AN2690)·(bpe), and starting materials.

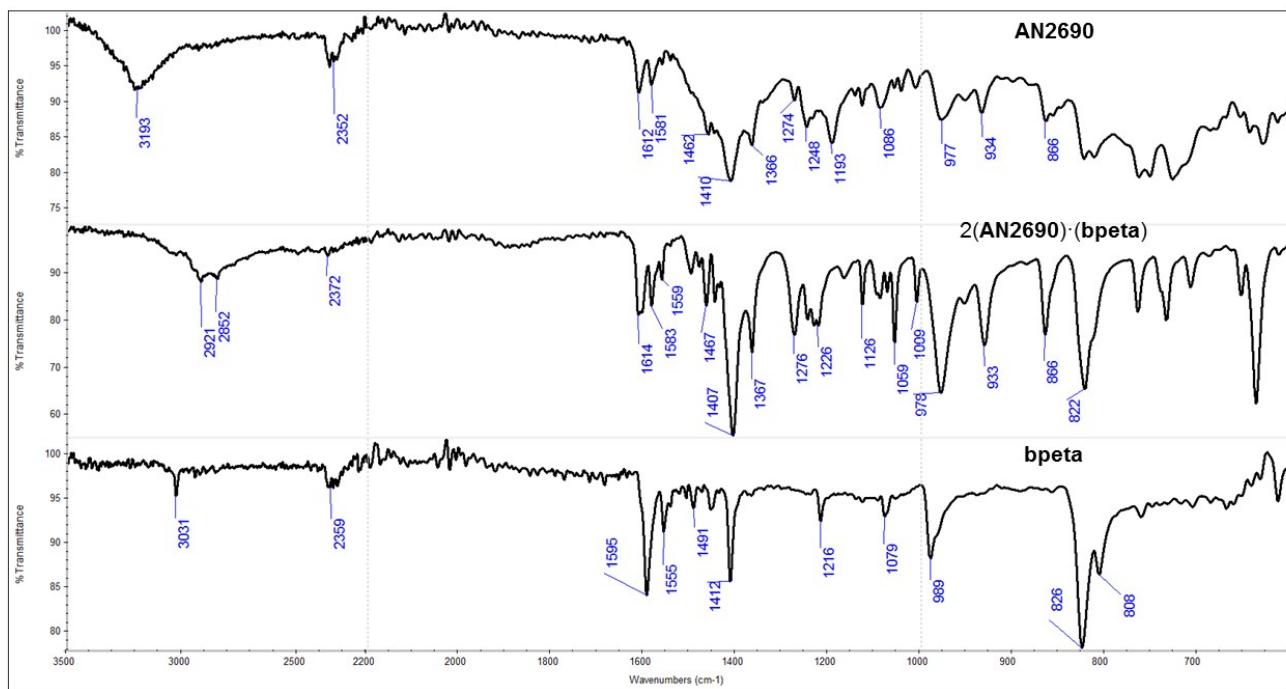


Fig. S4. FTIR spectra of cocrystal 2(AN2690)·(bpeta), and starting materials.

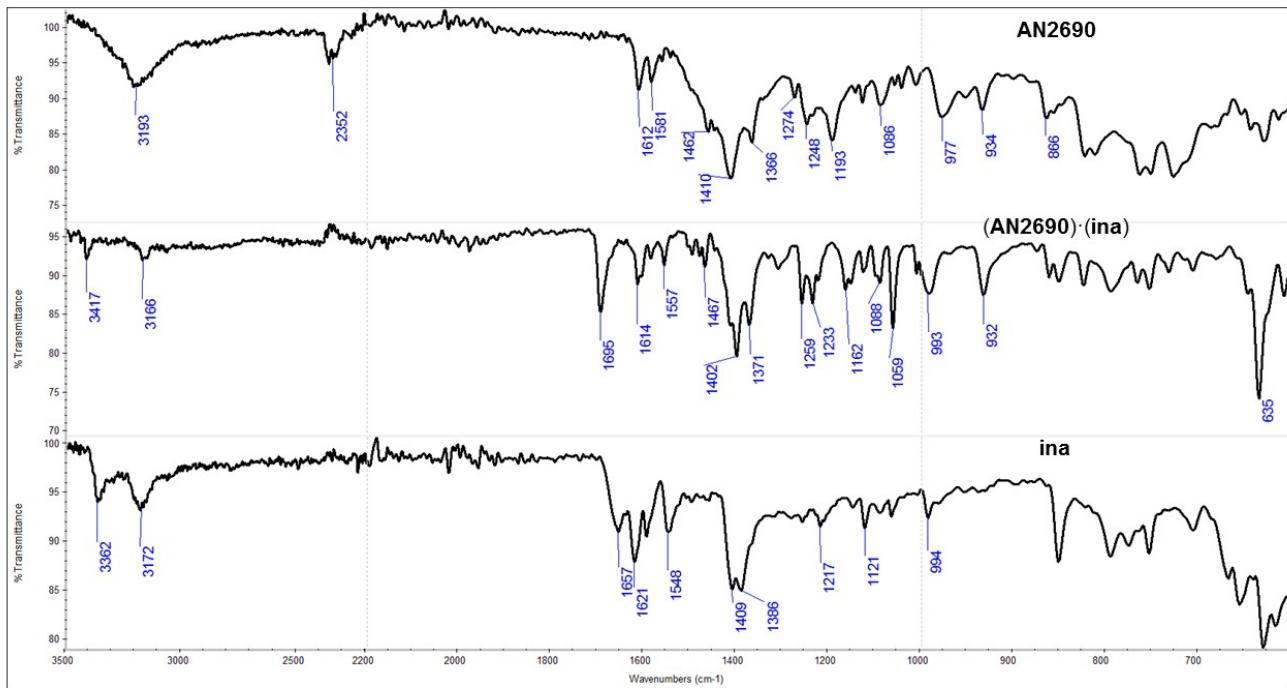


Fig. S5. FTIR spectra of cocrystal (AN2690)·(ina), and starting materials.

S5. Powder X-ray diffraction data

Powder X-ray diffraction data was collected from samples mounted on glass slides by a Siemens D5000 X-ray diffractometer using Cu K_α1 radiation ($\lambda = 1.54056 \text{ \AA}$) between 3°- 35° two-theta (scan type: locked coupled; scan mode: continuous; step size: 0.02°).

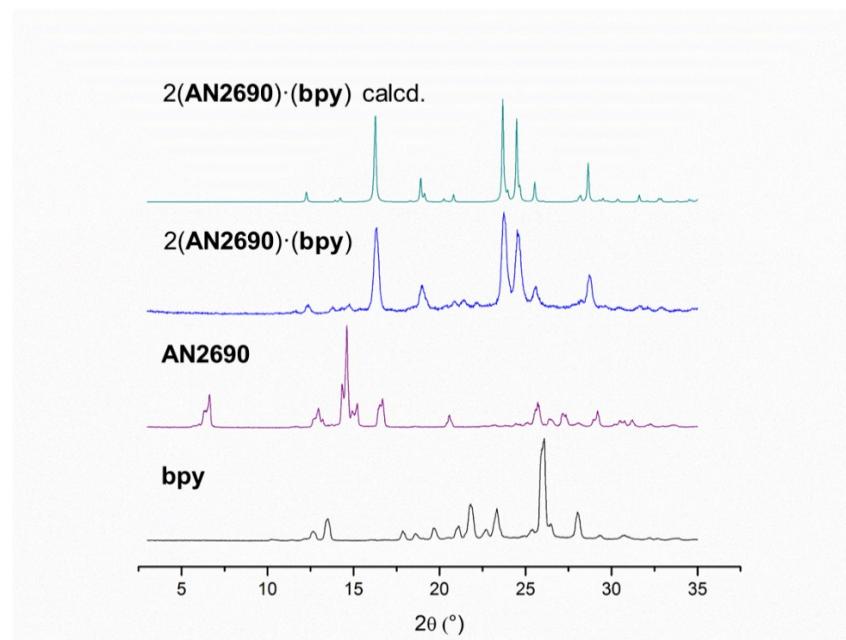


Fig. S6. PXRD analysis of the solids of cocrystal 2(AN2690)·(bpy), and starting materials.

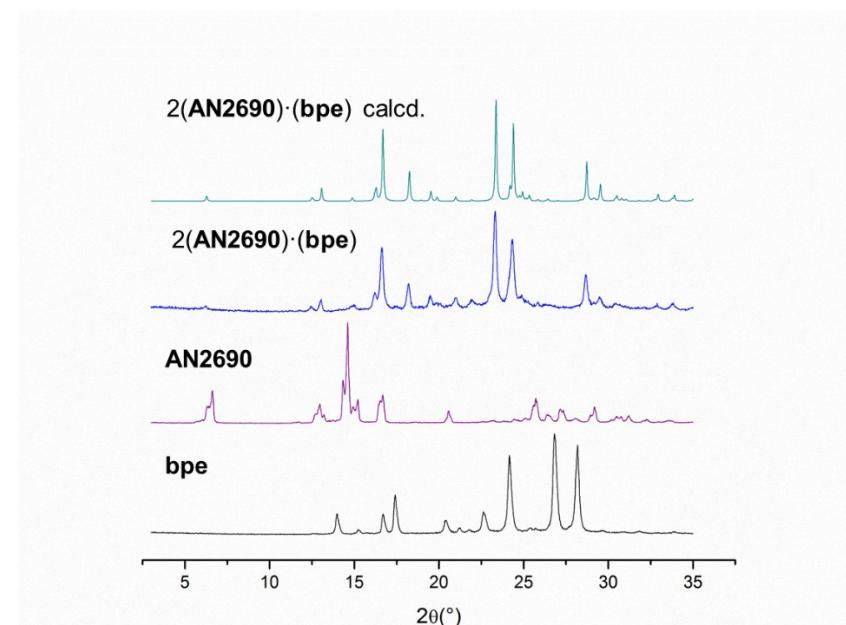


Fig. S7. PXRD analysis of the solids of cocrystal 2(AN2690)·(bpe), and starting materials.

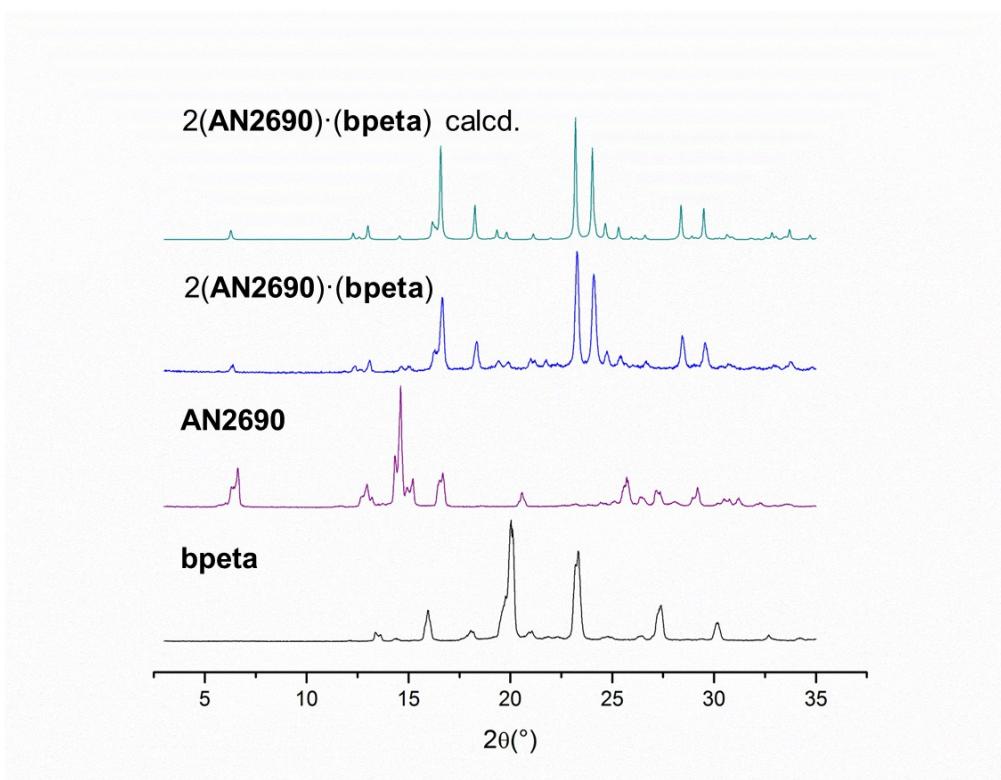


Fig. S8. PXRD analysis of the solids of cocrystal **2(AN2690)·(bpeta)**, and starting materials.

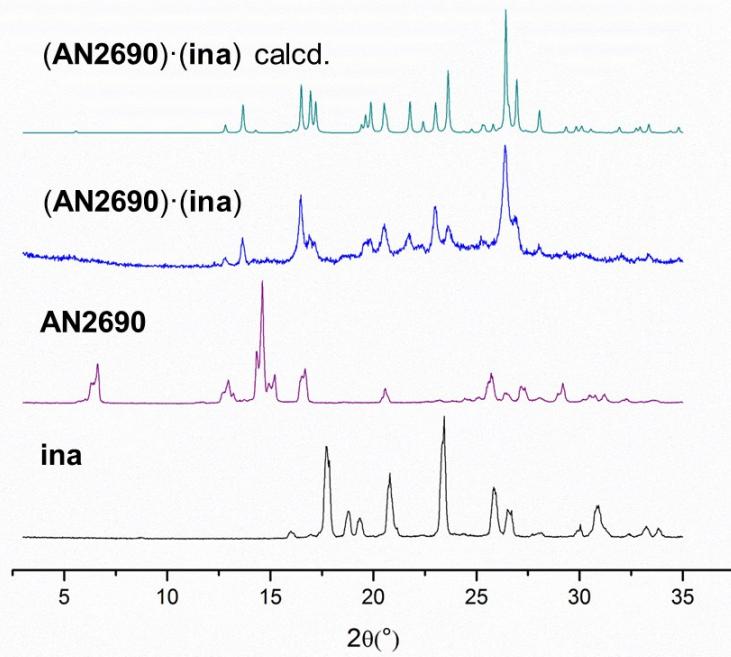


Fig. S9. PXRD analysis of the solids of cocrystal **(AN2690)·(ina)**, and starting materials.

References

1. S. Sene, D. Berthomieu, B. Donnadieu, S. Richeter, J. Vezzani, D. Granier, S. Bégu, H. Mutin, C. Gervais and D. Laurencin, *CrystEngComm*, 2014, **16**, 4999-5011.
2. G. M. Sheldrick, *Acta Crystallogr. Sect. A: Found. Crystallogr.*, 2008, **64**, 112-122.
3. O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. Howard and H. Puschmann, *J. Appl. Crystallogr.*, 2009, **42**, 339-341.