Electronic Supplementary Material (ESI) for CrystEngComm. This journal is © The Royal Society of Chemistry 2017

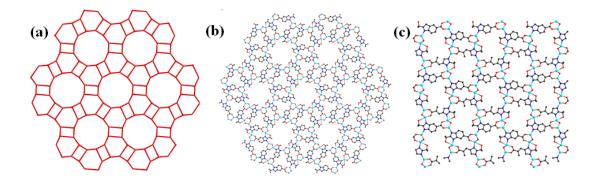
Supporting information

## Sythesis of Zeolite-like Metal-organic Frameworks from a Dual-ligand Strategy

Yu-Huan Tang, ab Xin Wu, Fei Wang\*a and Jian Zhang\*a

<sup>a</sup>State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, the Chinese Academy of Sciences, Fuzhou, Fujian 350002, P. R. China. E-mail: wangfei04@fjirsm.ac.cn, zhj@fjirsm.ac.cn

<sup>&</sup>lt;sup>b</sup>University of Chinese Academy of Sciences, 100049 Beijing, P. R. China



**Figure S1** (a) AlPO<sub>4</sub>-5 layer, (b) the layer generated by ebimc ligands and dizinc units within compound **1**, (c) the layer generated by ebimc ligands and dizinc units within compound **2**.

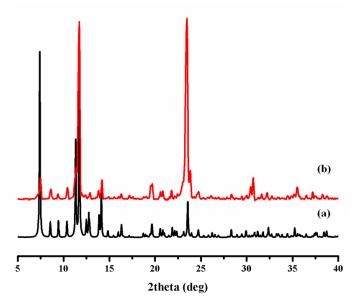


Figure S2 The powder XRD patterns of (1). (a) simulated (1); (b) as-synthesized (1)

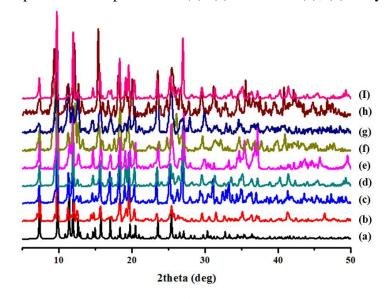
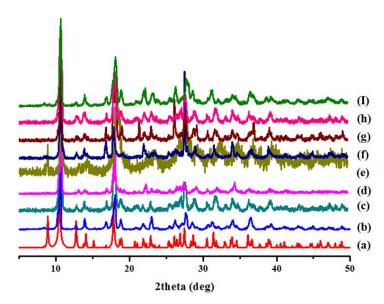


Figure S3 The powder XRD patterns of (2) under different conditions. (a) simulated

(2); (b) as-synthesized (2); (c) (2) soaked in DMF; (d) (2) soaked in ethanol; (e) (2) soaked in  $H_2O$ ; (f) (2) soaked in methanol; (g) (2) soaked in acetone; (h) (2) soaked in  $CH_2Cl_2$ ; (I) activated (2).



**Figure S4** The powder XRD patterns of (3) under different conditions. (a) simulated (3); (b) as-synthesized (3); (c) (3) soaked in DMF; (d) (3) soaked in ethanol; (e) (3) soaked in H<sub>2</sub>O; (f) (3) soaked in methanol; (g) (3) soaked in acetone; (h) (3) soaked in CH<sub>2</sub>Cl<sub>2</sub>; (I) activated (3).

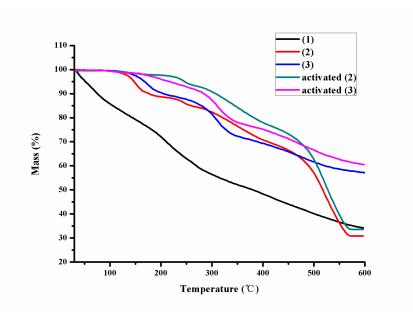
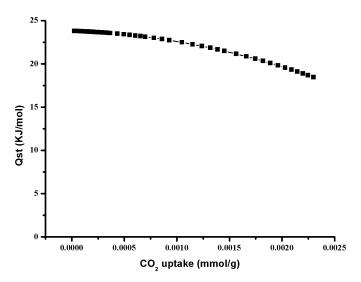


Figure S5 The TG plots of compounds (1), (2) and (3).

In compound 2, the number of residual electrons after SQUEEZE is 12.5, corresponding to 5/4 water molecules. However, from the TG plot, the initial gradual weight-loss step of 13% in the

TGA, which indicates one guest DMF molecule and one water molecule per  $Zn_2(\mu_2\text{-OH})(\text{ebimc})(5\text{-mtz})$  unit for compound 2. As for 1, we identify one DMF per  $Zn_2(\mu_2\text{-OH})(\text{ebimc})(5\text{-mtz})$  unit. Considering its poor stability and no obvious platform on the TG plot, so it's meaningless to use this method to determine the remaining disordered solvent.



**Figure S6**. The isosteric heat of CO<sub>2</sub> adsorption for **2** estimated by Viral equation.

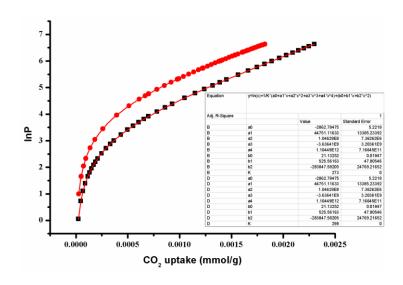
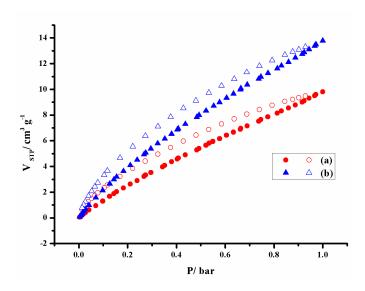


Figure S7 Virial analysis of the CO<sub>2</sub> sorption data for compound 2.



**Figure S8** The sorption isotherms of (3). The  $CO_2$  adsorption and desorption (a) at 298K; the  $CO_2$  adsorption and desorption (b) at 273K.

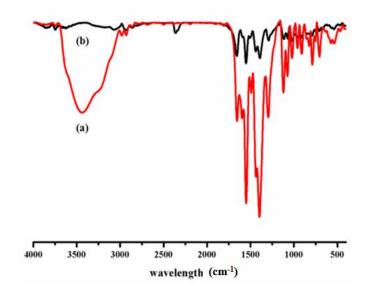


Figure S9 The IR plots of (2); (b) the IR plots of activated (2).

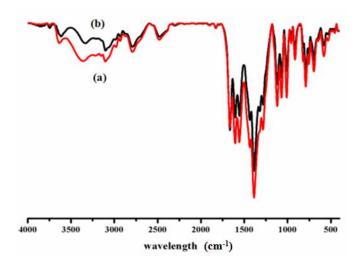


Figure S10 The IR plots of (3); (b) the IR plots of activated (3).

The characteristic peak of DMF molecules (~1600cm<sup>-1</sup>) are covered by the peak of crystals, which can not provide evidence for complete activation.