

A New Approach to Facilely Synthesize Crystalline $\text{Co}_2(\text{OH})_3\text{Cl}$ Microstructures in Eggshell Reactor Systems

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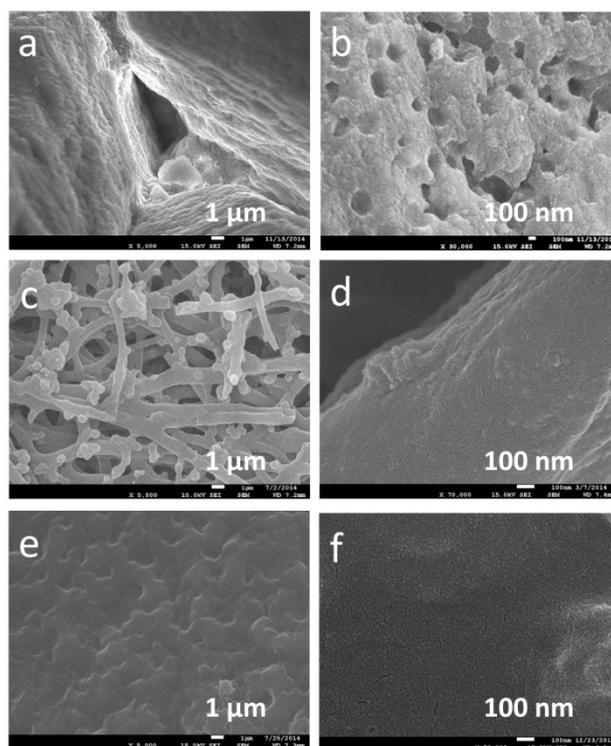


Figure S1. Eggshell characterization: FESEM images of (a, b) porous eggshell with pores visible; (c, d) outer layer eggshell membrane; (e, f) inner layer eggshell membrane at (a, c, e) high and (b, d, f) low magnification views, as previously observed.¹

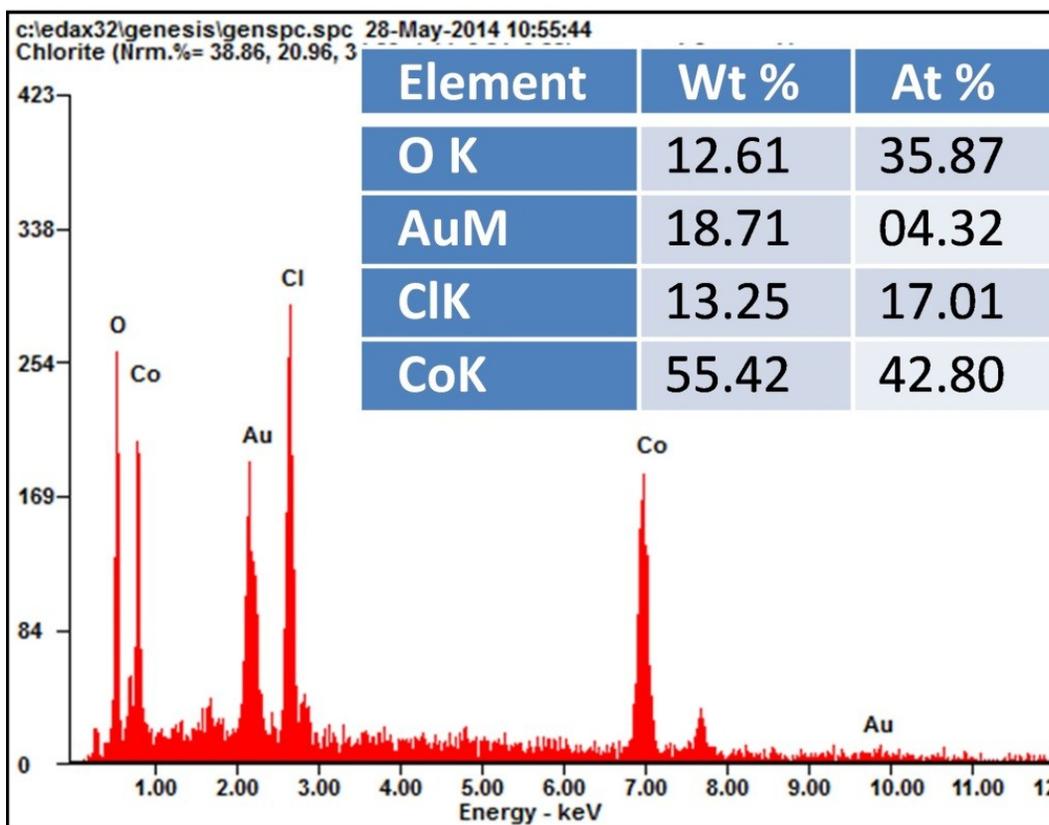


Figure S2. EDS analysis of typical $\text{Co}_2(\text{OH})_3\text{Cl}$ microparticles obtained from the eggshell reactor system. Reaction time was 3 days; other experimental conditions unchanged.

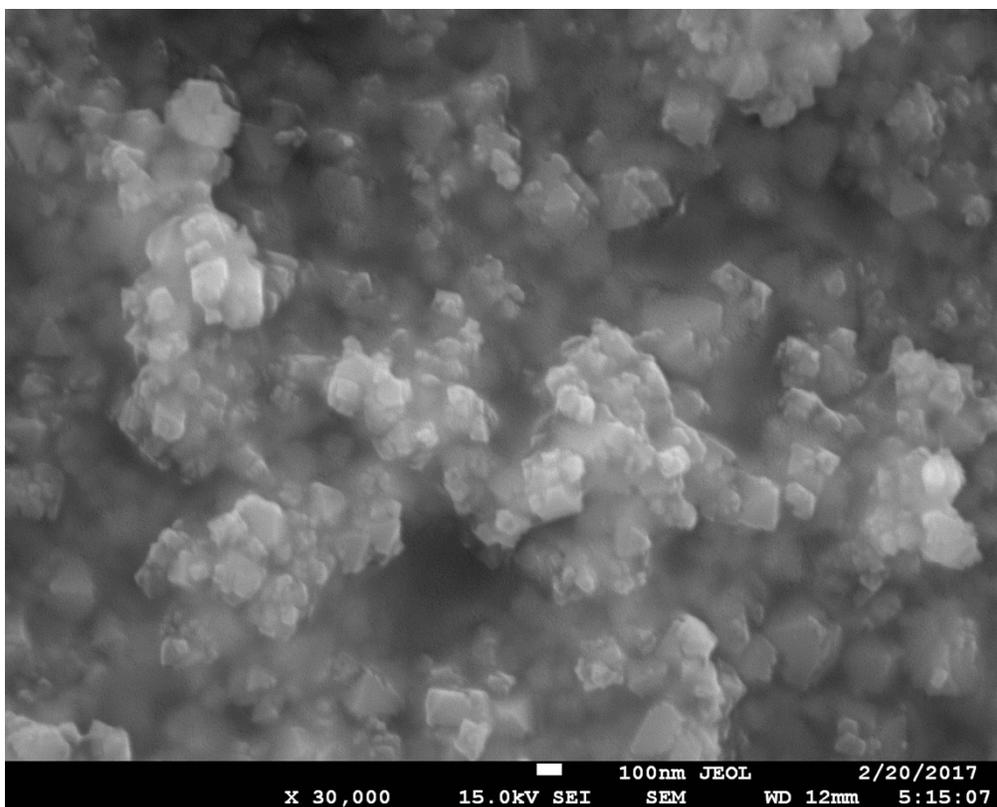


Figure S3. FESEM image of the particles obtained from a control experiment where the aqueous solutions of 30 ml 1 M CoCl_2 with 10 mM surfactant CTAB and 30 ml 1 M NaOH were directly mixed and stirred in a beaker and then kept at 50 °C for 5 days without stirring. Note, the $\text{Co}_2(\text{OH})_3\text{Cl}$ nanoparticles obtained in the control experiment are significantly smaller as compared to those crystalline $\text{Co}_2(\text{OH})_3\text{Cl}$ microparticles obtained in our eggshell reactor system, indicating the important role played by eggshell reactor system in facilitating the formation and growth of crystalline $\text{Co}_2(\text{OH})_3\text{Cl}$ microstructures.

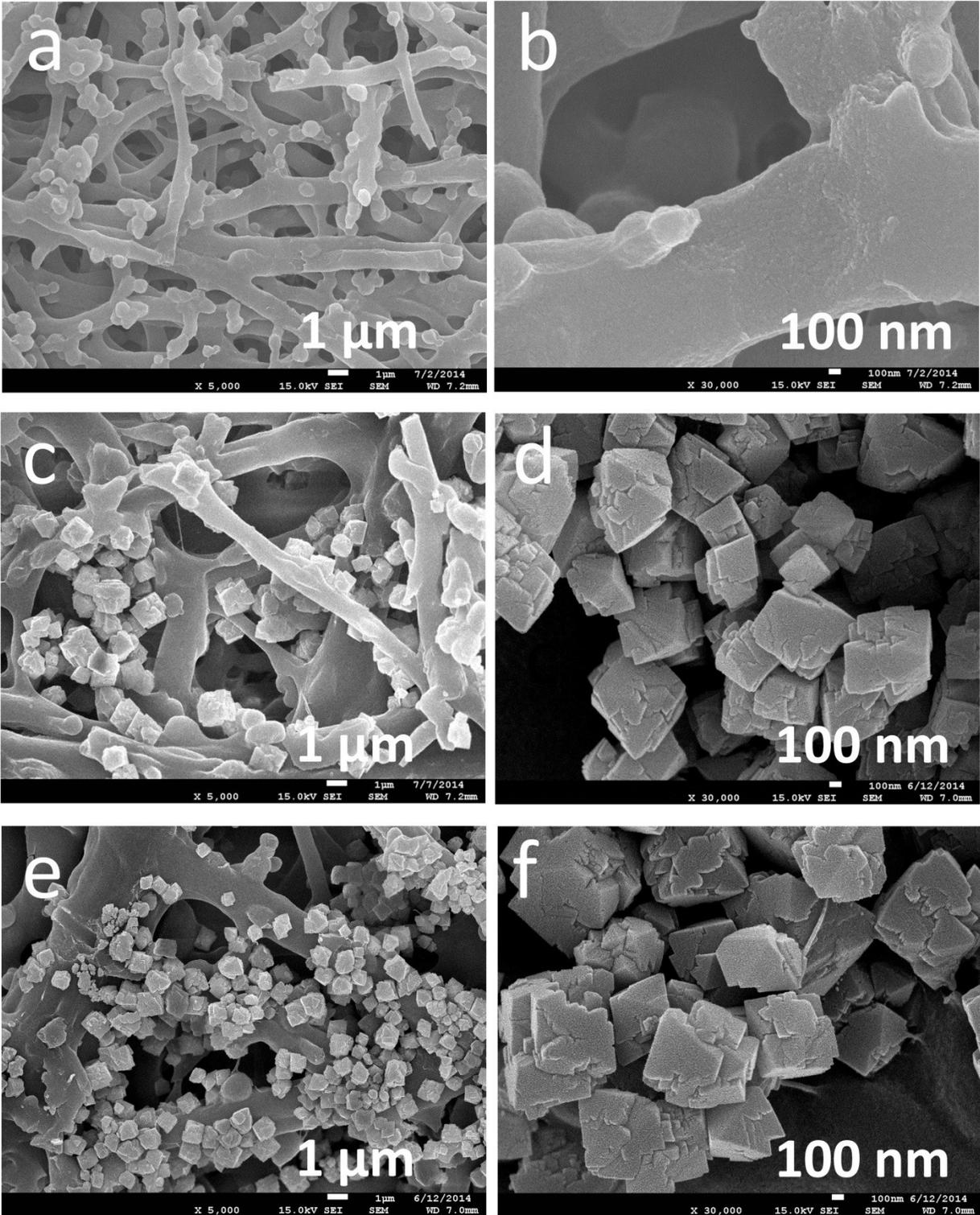


Figure S4. Time-course experiments to monitor the changes in forming $\text{Co}_2(\text{OH})_3\text{Cl}$ on outer layer of the eggshell membrane after reaction time of (a,b) 6 h, (c,d) 1 day and (e,f) 3 days at low (a,c,e) and high (b,d,f) magnification views.

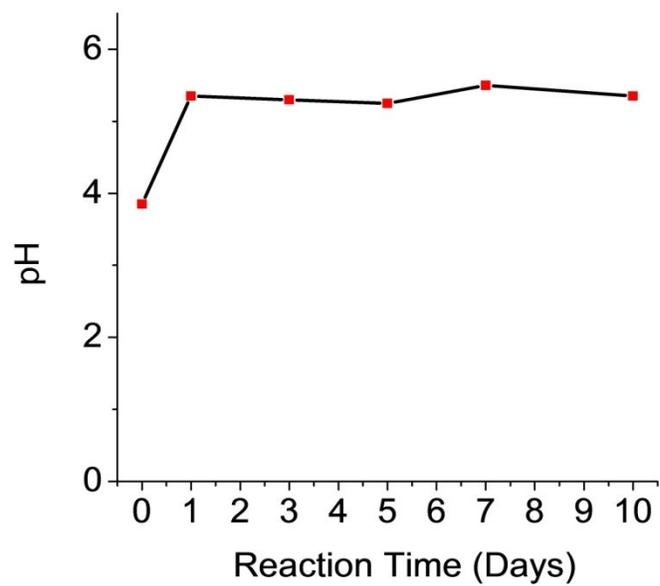


Figure S5. The pH measured along with time inside of the eggshell reactor.

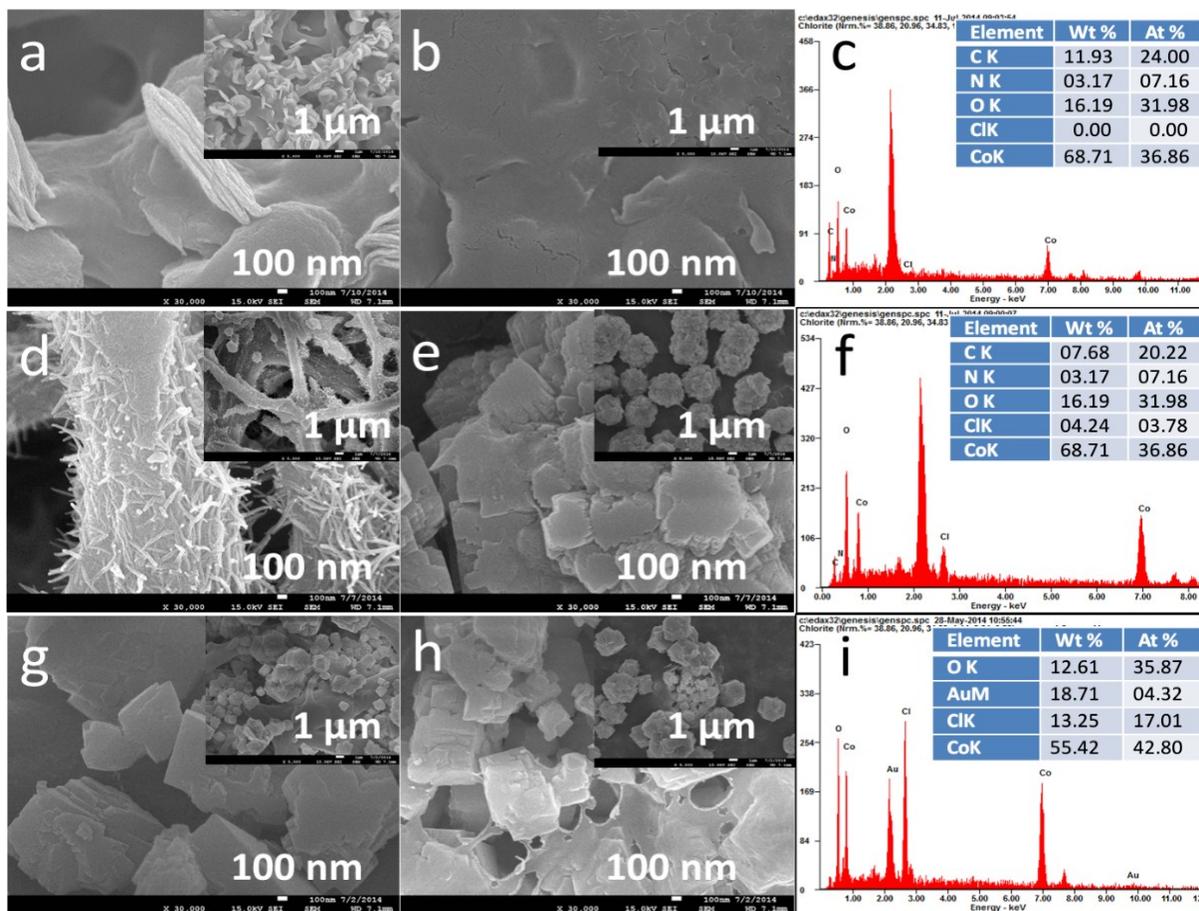


Figure S6. FESEM images and EDS results of solid products obtained from the eggshell reactor systems after 3 days of reaction where the concentration of CoCl_2 solution was changed while kept other experimental parameter unchanged: (a, b, c) 0.01 M CoCl_2 , (d, e, f) 0.25 M CoCl_2 , and (g, h, i) 1 M CoCl_2 solution. To be noted here, the concentration of CoCl_2 inside the eggshell reactor is critically important which could determine the morphology and composition of the products obtained.

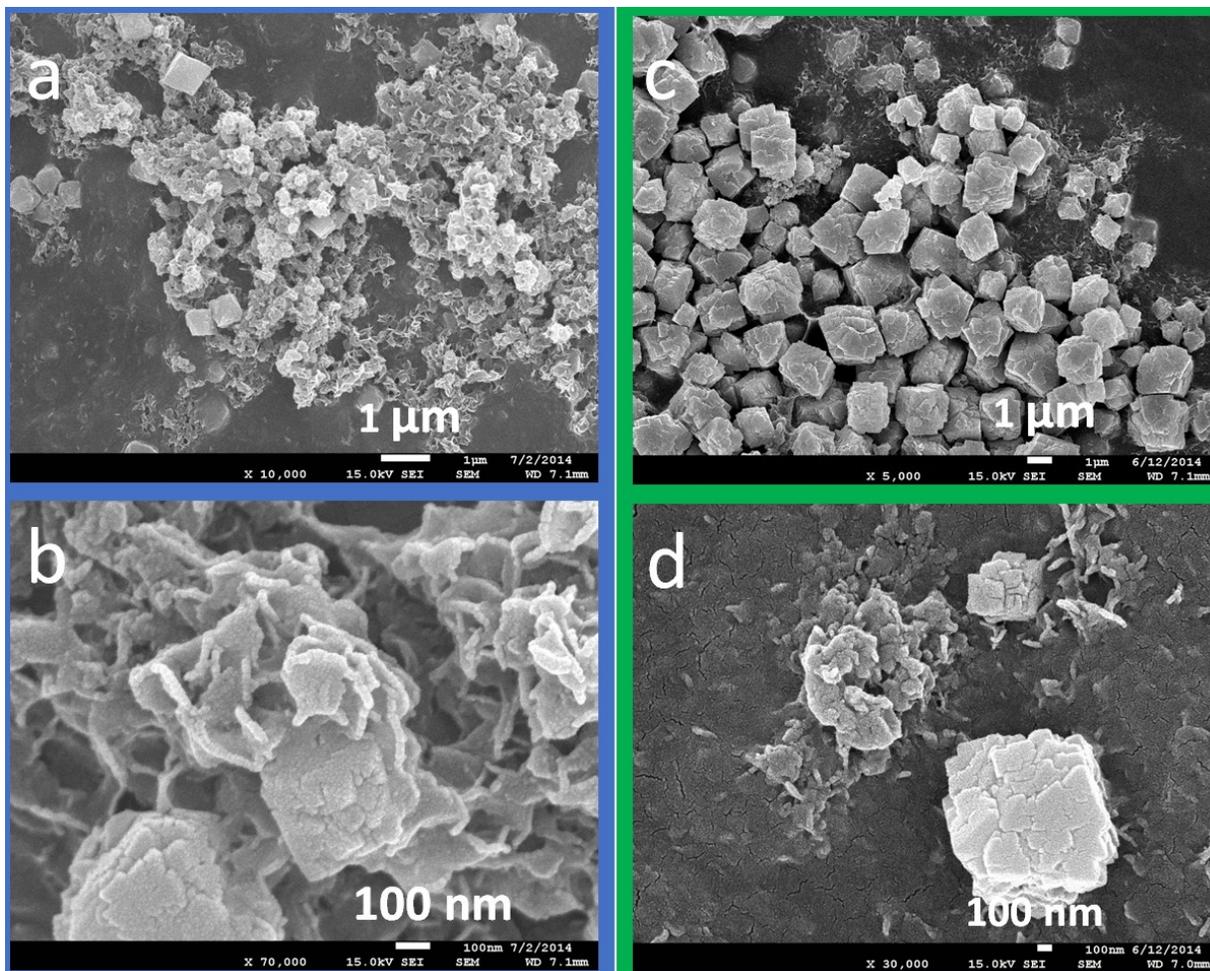


Figure S7. Additional FESEM images for the samples obtained after different reaction time inside the eggshell reactor system at various degrees of magnification: (a,b) after 12 hours; (c,d) after 24 hours at low (a,c) and high (b,d) magnification views.

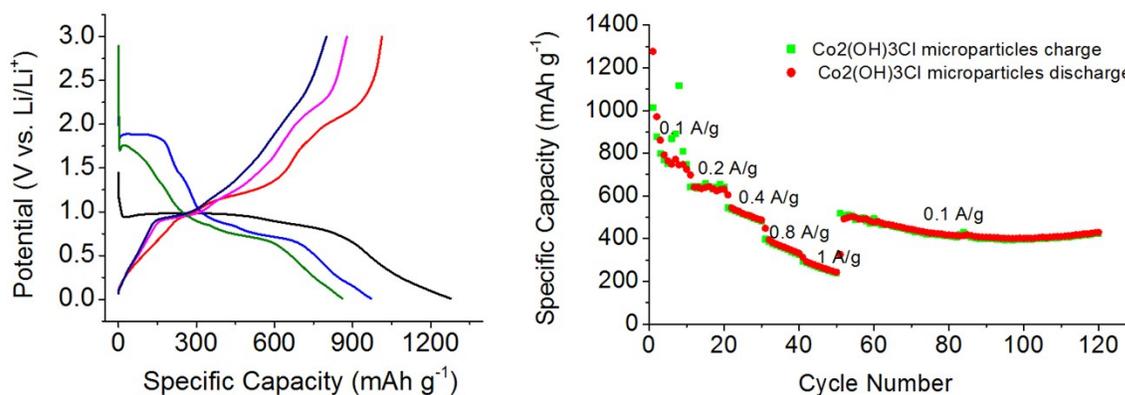


Figure S8. Electrochemical performance of the as-prepared $\text{Co}_2(\text{OH})_3\text{Cl}$ microstructures collected from the eggshell reactor system: (left) first three cycle charge-discharge profiles; (right) capacity vs cycle number plots tested at various current conditions.

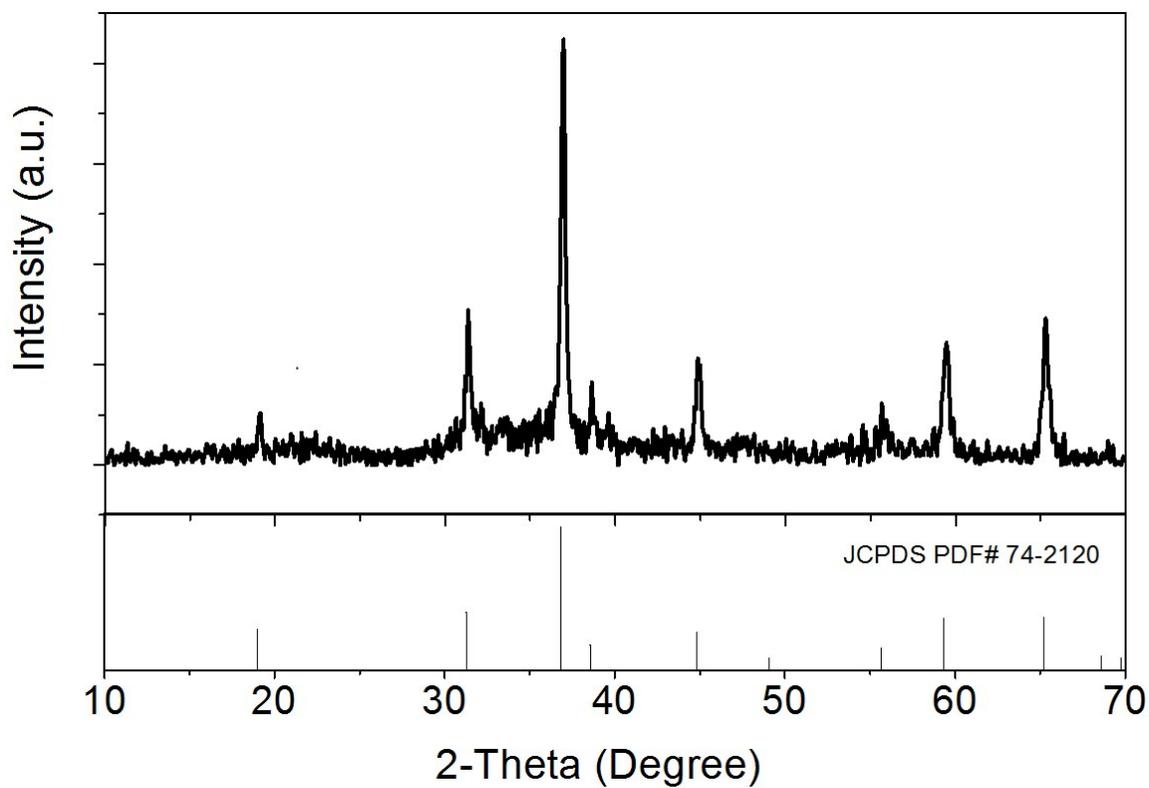


Figure S9. XRD pattern of Co₃O₄ derived from Co₂(OH)₃Cl microstructures after calcination at 450 °C in air for 2 h.

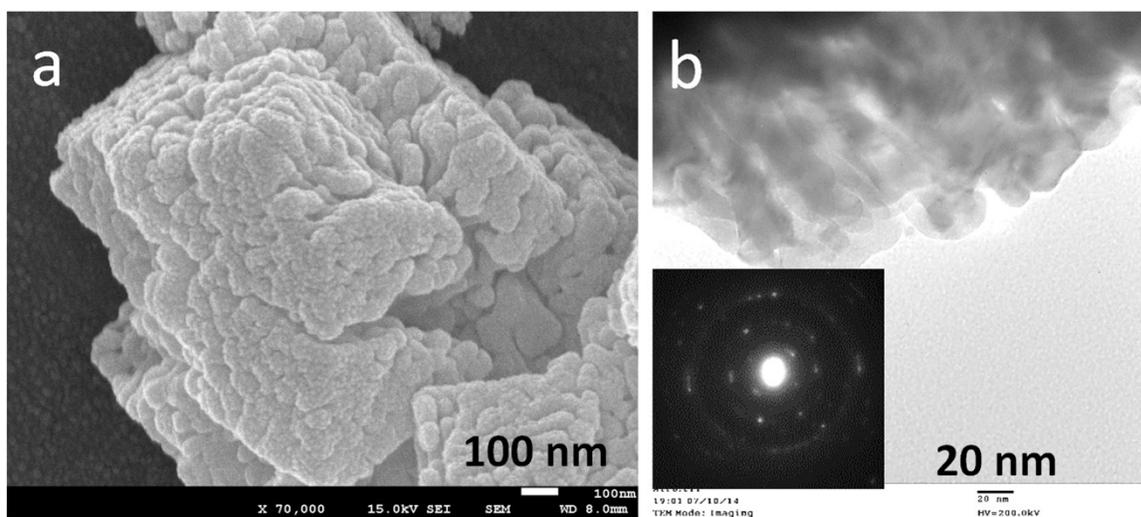


Figure S10. (a) SEM and (b) TEM characterization of the Co₃O₄ microstructures derived from Co₂(OH)₃Cl microparticles after calcination at 450 °C in air for 2 h. Inset of (b) is SAED which agrees with XRD analysis.

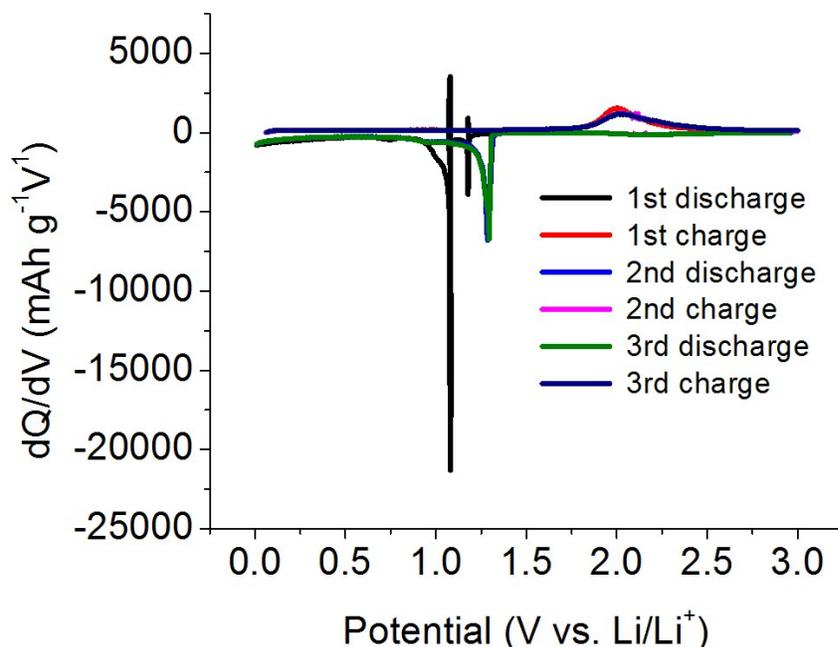


Figure S11. The dQ/dV vs V plots for the first three cycles of the as-derived nanoporous Co_3O_4 microstructures with nanopores.

Table S1. Summaries of reported battery performances of Co_3O_4 reported in the literature.

Morphologies	Methods	Specific capacities	Current densities	Cycles	Ref
nanofibers	Hydrothermal	937 mAh/g	100 mA/g	150	2
mesoporous cubes	Hydrothermal	1010 mAh/g	0.1 C	60	3
nanoparticles with C	Hydrothermal	888.8 mAh/g	0.2 C	80	4
mesoporous nanoflakes	Microwave assisted	806 mAh/g	0.1 C	300	5
mesoporous flakes	Hydrothermal	1115 mAh/g	0.05 C	100	6
mesoporous particles	Hydrothermal	913 mAh/g	200 mA/g	60	7
nanowire arrays	Hydrothermal	1031 mAh/g	100 mA/g	100	8
hierarchical star-like	Hydrothermal	1200 mA h/g	50 mA/g	100	9
<i>High-order microstructured</i>	<i>Eggshell reactor and calcination</i>	<i>900 mAh/g</i>	<i>100 mA/g</i>	<i>110</i>	<i>This work</i>

References:

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