Electronic Supplementary Material (ESI) for CrystEngComm. This journal is © The Royal Society of Chemistry 2017

Electronic Supplementary Information

Acid-base directed supramolecular isomers of isophthalate based MOFs for CO₂

adsorption and transformation

Jason A. Perman,^a Meng Chen,^a Antony A. Mikhail,^a Zheng Niu,^a and Shengqian Ma^a*

^aDepartment of Chemistry, University of South Florida, 4202 East Fowler Avenue, Tampa,

Florida, 33620

Email: sqma@usf.edu

Experiments and Instruments.

Ligand Synthesis: 20 mmol of 5-aminoisophthalic acid, 10 mmol of 1,4,5,8naphthalenetetracarboxylate dianhydride and five drops of N,N-dimethylformamide were added to a mortar and ground together with a pestle before a purple powder was heated at 180 °C for four hours to achieve a similar yield of H₄L.¹

MOF Synthesis: Cu(NO₃)₂·2.5H₂O (11.6 mg, 0.05 mmol) and H₄L (14.8 mg, 0.025 mmol) heated to 80 °C for up to 48 h. TFA was added from 5 to 40 μ L to perform the acid controlled syntheses and pyridine (0.05 mmol) was added for the base controlled synthesis.

Measurements from 4000-400 cm⁻¹ were taken on a Perkin Elmer FT-IR Spectrometer Spectrum Two (UATR Two) with 4 cm⁻¹ resolution. TA Instruments TGA Q50 was used to record thermal gravimetric analysis (TGA) data from room temperature to 800 °C at a 10 °C/min rate. Gas adsorption measurements were performed using a Micromeritics ASAP 2020 surface area analyzer to collect N₂ (surface area measurement at 77 K) and CO₂ isotherms at 273 K and 298 K, using high purity gas. Powder X-ray diffraction (PXRD) data was collected at room temperature using a Bruker D8 Advance theta-2theta diffractometer with copper radiation (Cu K α , $\lambda = 1.5406$ Å) operating at 40 kV and 40 mA and a secondary monochromator; whereby samples were measured between 3° and 25° at 0.5 second/step and step size of 0.01°.



Fig. S1. Infrared spectra of $[Cu_2L]_n$ **nbo** and **lvt**.



Fig. S2. TGA plots of $[Cu_2L]_n$ **nbo** and **lvt**.



Fig. S3. CO₂ sorption isotherms for $[Cu_2L]_n$ **nbo** and **lvt**.



Fig. S4. N₂ sorption isotherms for for $[Cu_2L]_n$ **nbo** and **lvt**.



Fig. S5. PXRD patterns after catalyzing propylene oxide into propylene carbonate. High

background due to low loading (<5 mg) of MOF along with pipetted reagents.

Reference:

1. J. A. Perman, A. J. Cairns, L. Wojtas, M. Eddaoudi and M. J. Zaworotko,

CrystEngComm, 2011, 13, 3130