

## SUPPORTING INFORMATION

# Two-component molecular crystals: evaluation of formation thermodynamics based on melting points and sublimation data

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**Table 1SI.** Experimental and calculated data for the co-crystals of Acridine

API	CF <sup>a</sup>	Stoich	$T_{fus}^{API}$ / °C	$T_{fus}^{CF}$ / °C	$T_{fus}^{CC}$ / °C	$\Delta G_{sub}^{API}$ / kJ·mol <sup>-1</sup>	$\Delta H_{sub}^{API}$ / kJ·mol <sup>-1</sup>	$\Delta G_{sub}^{CF}$ / kJ·mol <sup>-1</sup>	$\Delta H_{sub}^{CF}$ / kJ·mol <sup>-1</sup>	$\Delta G_f^{CC}$ / kJ·mol <sup>-1</sup>	$\Delta H_f^{CC}$ / kJ·mol <sup>-1</sup>	$T\Delta S_f^{CC}$ / kJ·mol <sup>-1</sup>
Acridine	4-F-BA	1:1	100.8	185.0	381.5	40.5	97.9	36.8	93.1	-10.5	-13.6	-3.1
Acridine	BA	1:1	100.8	123.3	94.7	40.5	97.9	34.4	90.5	4.7	5.7	1.0
Acridine	2-F-BA	1:1	100.8	125.5	99.0	40.5	97.9	35.5	94.4	2.9	2.0	-0.9
Acridine	2-Cl-BA	1:1	100.8	140.8	95.5	40.5	97.9	41.0	101.5	-0.3	-2.3	-2.0
Acridine	2-Br-BA	1:1	100.8	149.6	93.5	40.5	97.9	44.1	108.5	-2.3	-6.9	-4.5
Acridine	2-I-BA	1:1	100.8	163.3	99.3	40.5	97.9	47.8	112.8	-3.8	-7.8	-4.0
Acridine	3-F-BA	1:1	100.8	124.6	122.9	40.5	97.9	34.4	93.6	-2.6	-1.8	0.8
Acridine	3-Cl-BA	1:1	100.8	154.8	129.5	40.5	97.9	41.0	100.6	0.0	0.1	0.1
Acridine	3-Br-BA	1:1	100.8	158.5	137.3	40.5	97.9	43.5	105.9	0.4	1.1	0.7
Acridine	3-I-BA	1:1	100.8	188.2	140.2	40.5	97.9	48.4	111.1	-0.4	-0.6	-0.3
Acridine	Phloroglucinol	2:1	100.8	216.0	222.0	40.5	97.9	69.4	135.5	20.8	27.0	6.3

<sup>a</sup> BA – Benzoic Acid;