

**Supporting Information for:**

Nickel(II) Metal-Organic Frameworks with N,N'-di(4-pyridyl)-1,4,5,8-naphthalenetetracarboxydiimide Ligands: Influence of Secondary Building Unit Geometry on Dimensionality and Framework Dimensions.

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**Table S1.** Bond lengths and angles for **1**.

Bond	Oxygen ligand	Bond Length (Å)	Angle	Oxygen ligand	Angle (°)	Angle	Oxygen ligand	Angle (°)
Ni1–N1 <sup>b</sup>	N/A	2.058(7)	O3–Ni1–O3 <sup>1</sup>	DMF	178.8(3)	O4A–Ni1–O3	Nitrate/DMF	82.4(4)
Ni1–N4 <sup>2</sup>	N/A	2.078(7)	O3–Ni1–N4 <sup>2</sup>	DMF	90.59(14)	O4A <sup>1</sup> –Ni1–N1	Nitrate	89.2(4)
Ni1–O3	DMF	2.075(6)	O3 <sup>1</sup> –Ni1–N4 <sup>2</sup>	DMF	90.59(14)	O4A–Ni1–N1	Nitrate	89.2(4)
Ni1–O3 <sup>1</sup>	DMF	2.075(6)	O3–Ni1–N1	DMF	89.41(14)	O4A <sup>1</sup> –Ni1–N4 <sup>2</sup>	Nitrate	90.8(4)
Ni1–O4A	Nitrate	2.046(9)	O3 <sup>1</sup> –Ni1–N1	DMF	89.41(14)	O4A–Ni1–N4 <sup>2</sup>	Nitrate	90.8(4)
Ni1–O4A <sup>1</sup>	Nitrate	2.046(9)	O4A–Ni1–O3 <sup>1</sup>	Nitrate/DMF	97.6(4)	O4A <sup>1</sup> –Ni1–O4A	Nitrate	178.4(8)
			O4A <sup>1</sup> –Ni1–O3 <sup>1</sup>	Nitrate/DMF	82.4(4)	N1–Ni1–N4 <sup>2</sup>	N/A	180
			O4A <sup>1</sup> –Ni1–O3	Nitrate/DMF	97.6(4)			

<sup>a</sup>For the disordered ligands (DMF and nitrate molecules) the highest occupancy molecule was used. <sup>b</sup>N designates the DPNDI ligand. <sup>1</sup> and <sup>2</sup> designate different ligands.

**Table S2.** Bond lengths for **2** and **3**.

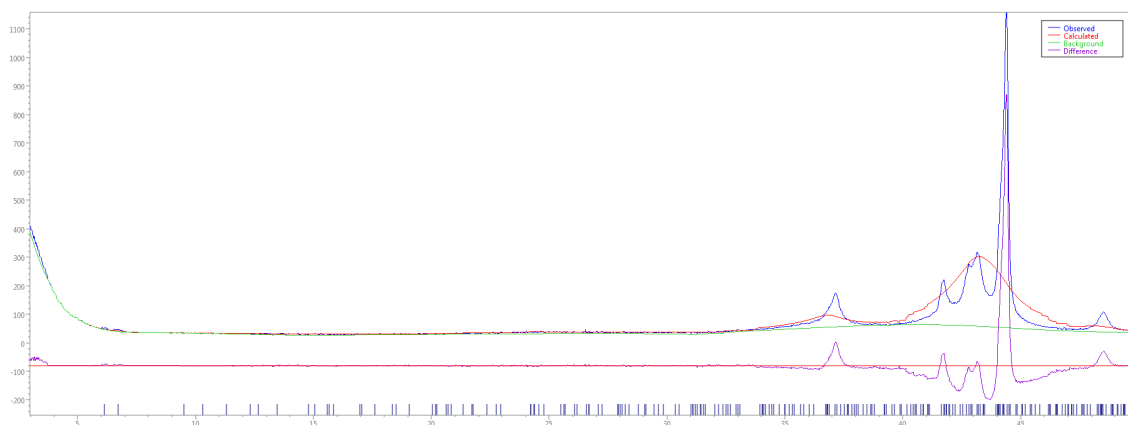
Bond	<b>2</b>	<b>3</b>
	Bond Length (Å)	Bond Length (Å)
Ni–N <sup>i</sup> (DPNDI pyridine)	2.084 (2)	2.081 (2)
Ni–N <sup>ii</sup> (DPNDI pyridine)	2.087 (2)	2.091 (2)
Ni–O <sup>i</sup> (carboxylic acid)	1.9836 (19)	1.993 (2)
Ni–O <sup>ii</sup> (carboxylic acid)	2.0087 (18)	1.9945 (18)
Ni–O <sup>i</sup> (nitrate)	2.1299 (19)	2.153 (2)
Ni–O <sup>ii</sup> (nitrate)	2.159 (2)	2.1738 (19)

\*<sup>i</sup> and <sup>ii</sup> designate atoms from different ligands

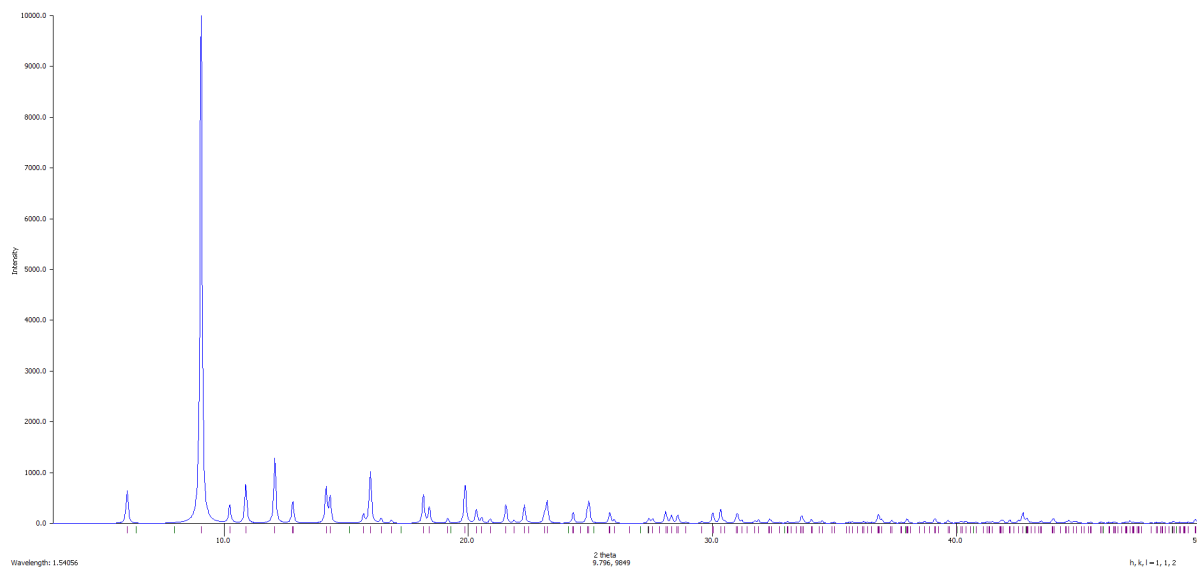
**Table S3.** Bond Angles for **2** and **3**.

Bonds	<b>2</b>	<b>3</b>
	Angle (°)	Angle (°)
N <sup>i</sup> –Ni–O <sup>i</sup> (carboxylic acid)	86.90 (8)	88.47 (9)
N <sup>i</sup> –Ni–O <sup>ii</sup> (carboxylic acid)	89.19 (9)	96.57 (8)
N <sup>i</sup> –Ni–O <sup>i</sup> (nitrate)	88.72 (8)	87.32 (8)
N <sup>i</sup> –Ni–O <sup>ii</sup> (nitrate)	89.30 (8)	90.79 (9)
N <sup>ii</sup> –Ni–O <sup>i</sup> (carboxylic acid)	89.99 (9)	88.62 (8)
N <sup>ii</sup> –Ni–O <sup>i</sup> (carboxylic acid)	95.70 (8)	89.00 (9)
N <sup>ii</sup> –Ni–O <sup>i</sup> (nitrate)	88.33 (8)	88.37 (8)
N <sup>ii</sup> –Ni–O <sup>ii</sup> (nitrate)	91.03 (9)	89.57 (9)
N <sup>i</sup> –Ni–N <sup>ii</sup>	177.40 (9)	174.81 (9)

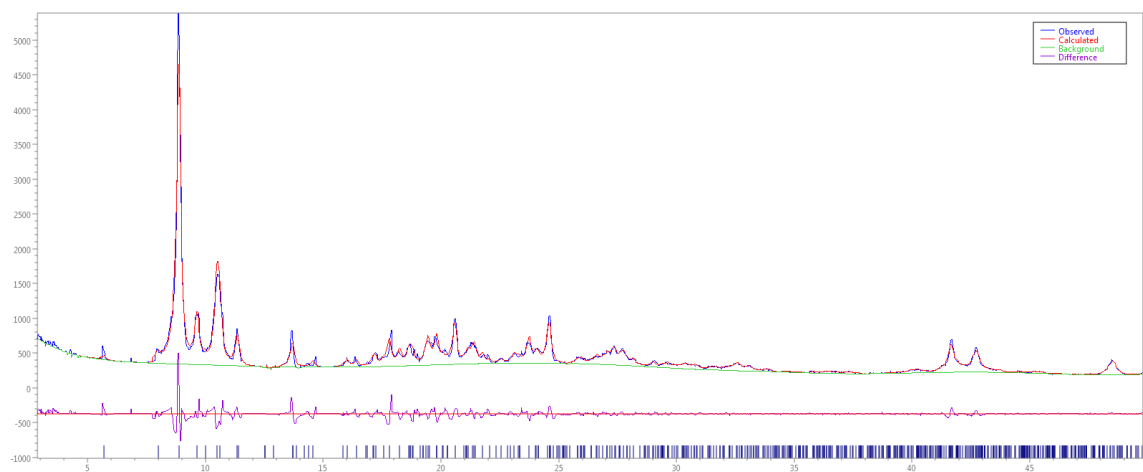
\*N designates a nitrogen atom on the pyridine group of the DPNDI molecule. <sup>i</sup> and <sup>ii</sup> designate atoms from different ligands.



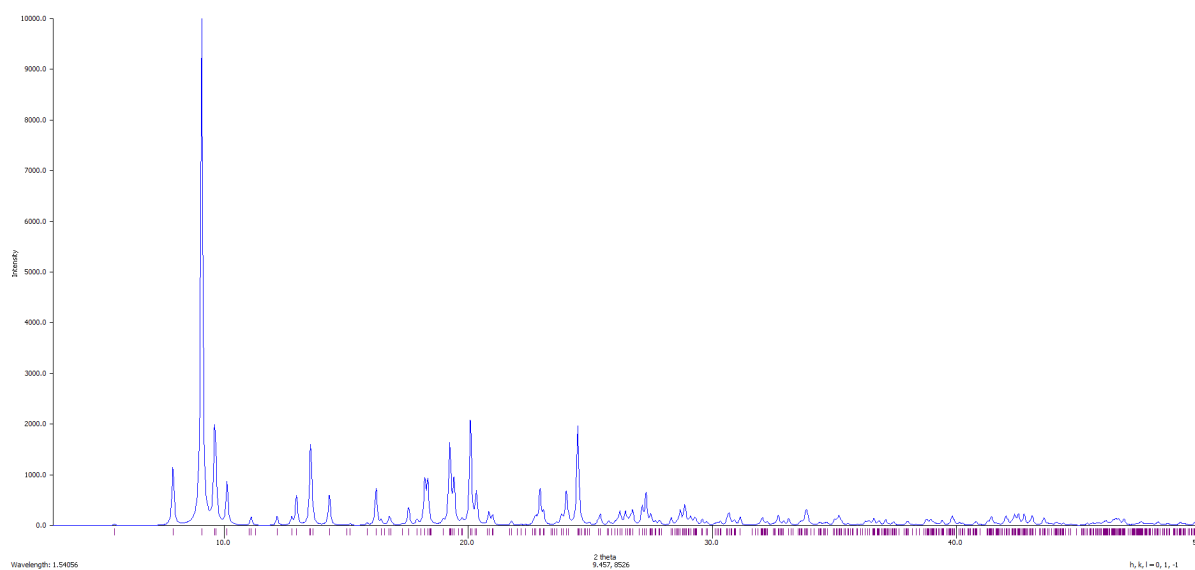
**Fig. S1.** PXRD pattern for **1**. Observed experimental diffractogram – blue trace. Pattern indicates significant loss of crystallinity upon drying.



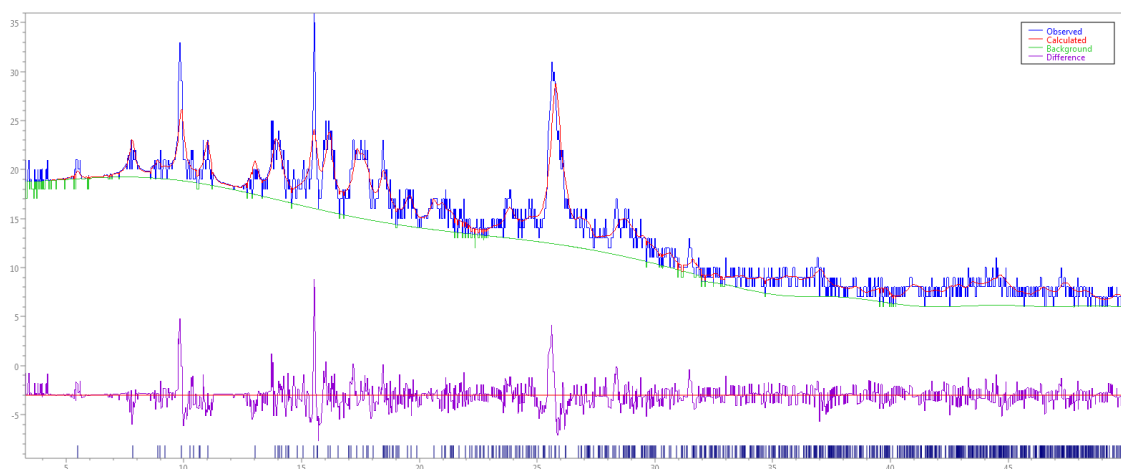
**Fig. S2.** Calculated PXRD pattern for **1**. Calculated using Mercury.<sup>S1</sup>



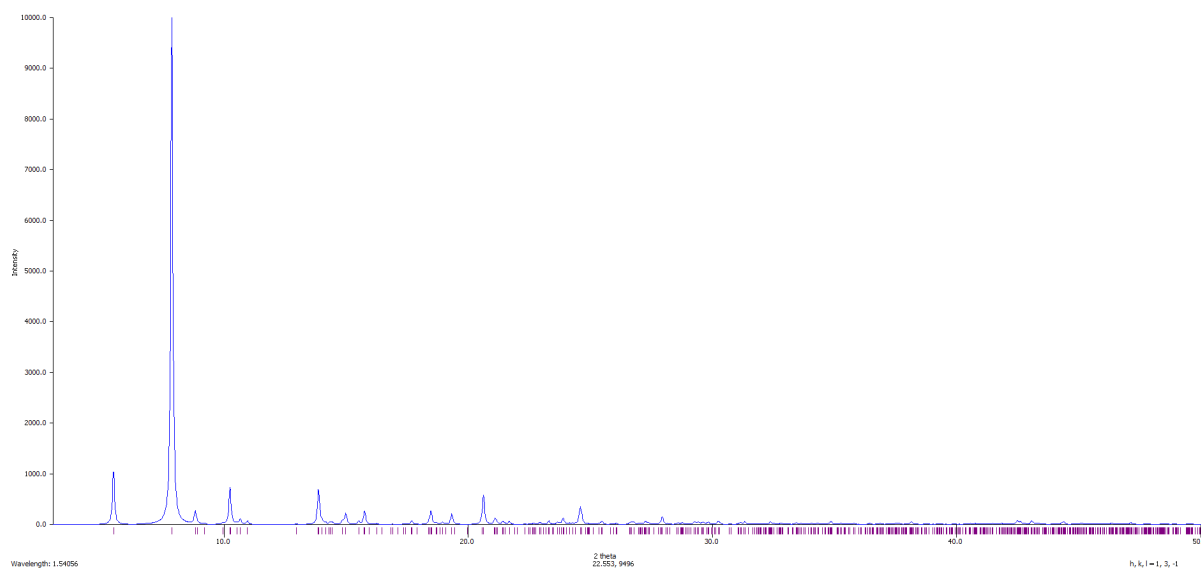
**Fig. S3.** PXRD pattern for **2**. Observed experimental diffractogram – blue trace; calculated pattern in red.



**Fig. S4.** Calculated PXRD pattern for **2**. Calculated using Mercury.<sup>S1</sup>



**Fig. S5.** PXRD pattern for **3**. Observed experimental diffractogram – blue trace; calculated pattern in red. Pattern indicates low crystallinity of sample upon drying.



**Fig. S6.** Calculated PXRD pattern for **3**. Calculated using Mercury.<sup>S1</sup>

## References

S1. C. F. Macrae, I. J. Bruno, J. A. Chisholm, P. R. Edgington, P. McCabe, E. Pidcock, L. Rodriguez-Monge, R. Taylor, J. van de Streek and P. A. Wood, *J. Appl. Cryst.*, 2008, **41**, 466-470.