Supporting Information

Switchable dielectric phase transition originating from disorder-order transformation and distortion in $\{[(C_4H_4N_2)C_0(H_2O)_4]SO_4 \cdot 2H_2O\}_n$

Table S1 Selected bond lengths (Å) and bond angles (°) for compound 1 at 296 K and 100 K.

Table S2 Hydrogen bonds for compound 1 at 296 K and 100 K.

Figure S1 IR spectrum of compound 1.

Figure S2 PXRD pattern of compound 1.

Figure S3 The TG curves of compound **1** in flowing air with a heating rate of 10 K/min.

Figure S4 The packing diagrams of compound 1 at (a) 296 K and (b) 100K.

100 18.			
296K			
Co1–O1W	2.099(2)	Co1–O2W	2.038(2)
Co1–O2WA	2.201(2)	Co1–O1WA	2.099(2)
Co1-N1	2.038(2)	Co1–N1A	2.201(2)
S1-O4B	1.385(7)	S1-O1B	1.430(8)
S1-O1	1.430(8)	S1-O2	1.528(5)
S1-O2B	1.528(5)	S1-O3	1.471(4)
S1-O4	1.385(7)	N1-C1	1.329(4)
N1-C2	1.332(4)	C1-C2C	1.384(4)
O1W-Co1-O2W	91.31(9)	O1W-Co1-N1	87.37(8)
O1W-Co1-O1WA	180	O1-Co1-O2WA	88.69(9)
O1W-Co1-N1A	92.63(8)	O2W-Co1-N1	87.02(8)
O1WA-Co1-O2W	88.69(9)	O2W-Co1-O2WA	180
O2W-Co1-N1A	92.98(8)	O1W_b-Co1-N1	92.63(8)
O2WA-Co1-N1	92.98(8)	N1-Co1-N1A	180
O1WA-Co1-O2WA	91.31(9)	01-S1-01B	152.6(5)
O1-S1-O2	105.2(4)	O1-S1-O2B	62.9(4)
O1A-S1-O2	62.9(4)	O1B-S1-O2B	105.2(4)
O1-S1-O3	103.7(4)	O1B-S1-O3	103.7(4)
01-S1-O4A	51.6(5)	O1B-S1-O4	51.6(5)
01-S1-O4	118.4(5)	O1B-S1-O4B	118.4(5)
O2-S1-O2A	131.8(3)	O2B-S1-O3	114.1(2)
O2-S1-O3	114.1(2)	O2-S1-O4	107.9(4)
O2B-S1-O4	56.2(4)	O2-S1-O4B	56.2(4)
O2B-S1-O4A	107.9(4)	O3-S1-O4B	107.8(3)
O4-S1-O4A	144.4(5)	Co1-N1-C1	124.4(2)
Co1-N1-C1	124.4(2)	C1-N1-C2	116.3(3)
C1-N1-C2	116.3(3)	Co1-N1-C2	119.3(2)
Co1-N1-C2	119.3(2)	N1-C1-C2C	121.7(3)
N1-C1-C2C	121.7(3)	N1-C2-C1C	122.0(3)
N1-C2-C1C	122.0(3)		
100K			
Co1–O1W	2.1120(9)	Co1–O2W	2.0314(10)
Co1-O3W	2.0915(9)	Co1-O4W	2.0468(9)
Co1-N1	2.1898(12)	Co1-N2	2.1966(12)
S1-O1	1.4689(10)	S1-O2	1.4751(10)
S1-O3	1.4787(11)	S1-O4	1.4726(11)
N1-C1	1.3432(17)	N1-C2	1.3405(17)
N2-C3	1.3409(17)	N2-C4	1.3440(17)
C1-C2A	1.3850(19)	C3-C4B	1.3877(19)
O1W-Co1-O2W	92.37(4)	O1W-Co1-O3W	178.84(4)
O1W-Co1-O4W	88.96(4)	O1W-Co1-N1	92.09(4)
O1W-Co1-N2	86.22(4)	O2W-Co1-O3W	88.67(4)
O2W-Co1-O4W	178.65(4)	O2W-Co1-N1	92.98(4)
O2W-Co1-N2	85.71(4)	O3W-Co1-O4W	90.00(4)

Table S1 Selected bond lengths (Å) and bond angles (°) for compound 1 at 296 K and 100 K.

O3W-Co1-N1	87.33(4)	O3W-Co1-N2	94.38(4)
O4W-Co1-N1	86.73(4)	O3W-Co1-N2	94.38(4)
O4W-Co1-N1	86.73(4)	O4W-Co1-N2	94.62(4)
N1-Co1-N2	177.82(4)	O1-S1-O3	108.11(6)
O1-S1-O4	110.43(6)	O2-S1-O3	109.38(6)
O2-S1-O4	108.92(6)	O3-S1-O4	108.78(6)
O1-S1-O2	111.18(6)	Co1-N1-C1	119.19(9)
Co1-N1-C2	124.21(9)	C1-N1-C2	116.60(11)
C3-N2-C4	116.52(11)	Co1-N2-C4	117.92(9)
Co1-N2-C3	125.56(9)	N1-C1-C2A	121.34(12)
N1-C2-C1A	122.06(12)	N2-C3-C4B	121.80(12)
N2-C4-C3B	121.68(12)		

Symmetry transformations of RTP used to generate equivalent atoms: A = -x, 1-y, 1-z; B = -1-x, y, 1/2-z; C = 1/2-x, 3/2-y, 1-z.

Symmetry transformations of LTP used to generate equivalent atoms: A = -x, 1-y, -z; B = 1-x, -y, -z.

296 K				
D−H···A	d(D-H)	$d(H \cdots A)$	$d(D \cdots A)$	∠(DHA)
O1W-H1WB…O2	0.86	1.93	2.709(6)	151
O1W-H1WB…O4 ^a	0.86	2	2.709(7)	139
O1W-H1WA…O3W ^b	0.86	1.93	2.780(3)	170
O2W−H2WA…O3W	0.86	1.96	2.717(4)	145
O2W-H2WB····O3c	0.87	1.75	2.593(4)	162
O3W-H3WB…O1d	0.79	1.86	2.630(9)	164
O3W-H3WB…O2e	0.79	2.05	2.759(7)	149
$O3W\text{-}H3WA\cdots O1^{\rm f}$	0.95	1.86	2.812(8)	177
O3W−H3WA…O4 ^g	0.95	2.11	2.949(8)	147
100 K				
O1W-H1WB…O2 ^a	0.82	1.9	2.7013(13)	165
O1W-H1WA…O5W ^b	0.79	1.98	2.7641(13)	175
O2W-H2WB…O6W ^c	0.82	1.93	2.7063(14)	158
O2W−H2WA…O3 ^c	0.8	1.82	2.6156(14)	168
O3W-H3WB…O6W	0.82	1.96	2.7797(13)	174
O3W-H3WA…O4	0.8	1.92	2.7149(14)	175
O4W−H4WB…O5W ^d	0.82	1.91	2.7178(14)	170
O4W H4WA…O3 ^e	0.74	1.9	2.6271(13)	173
$O5W\text{-}H5WB\cdots O1^{\rm f}$	0.83	1.82	2.6532(14)	173
O5W−H5WA…O2 ^g	0.82	2.02	2.8302(14)	168
O6W-H6WB…O1°	0.82	1.99	2.8120(14)	177
$O6WH6WA\cdots O4^h$	0.84	1.9	2.7162(14)	161
С3-Н3А…ОЗе	0.93	2.58	3.1953(17)	124

Table S2 Hydrogen bonds for compound 1 at 296 K and 100 K.

Symmetry transformations(296 K): a = -1-x, y, 1/2-z; b = -1/2+x, 1/2+y, z; c = 1/2+x, -1/2+y, z; d = 1+x, y, z; e = -x, y, 1/2-z; f = -x, 1-y, 1-z; g = 1+x, 1-y, 1/2+z.

Symmetry transformations(100 K): a = -1+x, 1/2-y, -1/2+z; b =x, 1/2-y, -1/2+z; c = 1-x, 1-y, -z; d = -x, -1/2+y, 1/2-z; e = 1-x, -1/2+y, 1/2-z; f = 1-x, 1/2+y, 1/2-z; g = 1-x, 1-y, 1-z; h = 1-x, 1/2+y, 1/2-z.



Figure S1 IR spectrum of compound 1.



Figure S2 PXRD pattern of compound 1.



Figure S3 The TG curves of compound 1 in flowing air with a heating rate of 10 K/min.



Figure S4 The packing diagrams of compound **1** at (a) 296 K and (b) 100K.