

Supporting Materials

Synthesis of New Framework Cobalt Aluminophosphate by Opening Cobalt Methylphosphonate Layered Material

Moussa Zaarour,^a Olivier Perez,^{b,*} Philippe Boullay,^b Jörn Martens,^c Boriana Mihailova,^d Konstantin Karaghiosoff,^c Lukáš Palatinus,^e Svetlana Mintova^{a,*}

^aLaboratoire Catalyse et Spectrochimie, Normandie Université, CNRS, ENSICAEN, 6, boulevard Maréchal Juin, 14000 Caen, France

^bCRISMAT, ENSICAEN, 6 Boulevard Maréchal Juin, 14050 Caen, France

^cDepartment of Chemistry, Ludwig-Maximilians-Universität München (LMU), Butenandtstr. 5-13, HausMunich, Germany

^dMineralogisch-Petrographisches Institut, Grindelallee 48, University of Hamburg, Germany

^eInstitute of Physics of the AS CR, Na Slovance 2, Prague, Czech Republic

Table S1. Positional parameters of $\text{Co}_{1.16}\text{Al}_2\text{P}_4\text{O}_{20}\text{H}_{12}$ in the space group $\text{P2}_1/\text{n}$ with $a=8.402(4)\text{\AA}$, $b=16.569(7)\text{\AA}$, $c=5.11(1)\text{\AA}$ and $\beta=90.6^\circ$.

Atom	Wyckoff position	Occu-pancy	X	y	z	$u_{\text{eq}} (\text{\AA}^2)$
Co1	2d	1	0.5	0	0	0.0292(3)
P4	4e	1	0.80996(14)	-0.08670(6)	-0.2388(3)	0.0156(3)
Al1	4e	1	0.63583(15)	0.29195(7)	0.2465(3)	0.0132(3)
P2	4e	1	0.48742(13)	-0.19170(6)	0.2474(2)	0.0132(3)
Co2	2d	0.161(4)	0.5	0	-0.5	0.0107(15)
O1	4e	1	0.5124(4)	-0.09836(16)	0.2747(6)	0.0211(9)
O2	4e	1	0.6905(5)	0.05492(14)	0.2081(7)	0.0357(13)
O3	4e	1	0.9261(6)	-0.0112(2)	-0.2600(9)	0.0419(14)
O4	4e	1	0.4066(4)	-0.20174(16)	-0.0198(6)	0.0164(9)
O5	4e	1	0.3836(4)	-0.21853(16)	0.4712(6)	0.0151(9)
O6	4e	1	0.6403(5)	-0.0560(2)	-0.2723(7)	0.0313(11)
O7	4e	1	0.8533(4)	-0.14201(18)	-0.4570(6)	0.0200(10)
O8	4e	1	0.6472(4)	-0.23489(17)	0.2624(7)	0.0169(8)
O9	4e	1	0.0897(3)	0.81494(16)	0.2375(7)	0.0226(10)
O10	4e	1	0.8342(4)	-0.12060(18)	0.0310(6)	0.0207(10)
H1-O9	4e	1	0.1481	0.7871	0.146	0.027178
H2-O9	4e	1	0.1268	0.8525	0.3223	0.027178
H1-O2	4e	1	0.7779	0.033	0.2131	0.04282
H2-O2	4e	1	0.6766	0.0981	0.2833	0.04282
H1-O1	4e	1	0.5255	-0.0495	0.289	0.025354
H1-O3	4e	1	0.9859	0.0277	-0.2708	0.050317

Table S2. Details of the single crystal XRD data collection

Chemical formula	Co _{1.16} Al ₂ P ₄ O ₂₀ H ₁₂
Molecularweight (g.mol ⁻¹)	578.36
Space group	P2 ₁ /n
<i>a</i> (Å)	8.402(4)
<i>b</i> (Å)	16.569(7)
<i>c</i> (Å)	5.11(1)
α (°)	90
β (°)	90.6(2)
γ (°)	90
Cell volume (Å ³)	711.2(16)
<i>Z</i>	2
Density (g.cm ⁻³)	2.7009
μ (mm ⁻¹)	2.08
Data collection temperature (K)	293
wavelength (Å)	0.71069
scan strategy / Dx (mm)	ω/ϕ scan / 35
θ_{max}	32.89
Reflections index limit	-12 ≤ <i>h</i> ≤ 11 -24 ≤ <i>h</i> ≤ 24 -7 ≤ <i>h</i> ≤ 7
unique reflections with $I \geq 3\sigma(I)$	1721
Absorption correction	multi-scan / SADABS
Internal R value after correction (%)	6.3
number of refinement parameters parameters	132
<i>Number of restraints</i>	8
$\rho_{min}/\rho_{max}(e/\text{Å}^3)$	-1.18/0.86
F(000)	579
Reliability factors	0.0581

Table S3. ADP harmonic parameters.

atoms	$u_{11}(\text{\AA}^2)$	$u_{22}(\text{\AA}^2)$	$u_{33}(\text{\AA}^2)$	$u_{12}(\text{\AA}^2)$	$u_{13}(\text{\AA}^2)$	$u_{23}(\text{\AA}^2)$
Co1	0.0221(6)	0.0196(4)	0.0457(7)	0.0053(5)	0.0109(7)	0.0087(4)
P4	0.0164(6)	0.0135(5)	0.0167(5)	0.0000(4)	0.0037(7)	0.0005(5)
Al1	0.0108(6)	0.0150(6)	0.0137(5)	0.0000(5)	0.0031(8)	0.0007(6)
P2	0.0107(5)	0.0154(5)	0.0134(4)	-0.0018(4)	0.0028(7)	-0.0013(5)
Co2	0.012(3)	0.006(2)	0.014(3)	0.002(2)	-0.002(3)	- 0.0059(18)
O1	0.0215(17)	0.0118(13)	0.0301(16)	- 0.0051(13)	0.0025(19)	- 0.0047(13)
O2	0.035(2)	0.0193(17)	0.053(3)	0.0075(17)	-0.006(2)	- 0.0037(19)
O3	0.050(3)	0.0301(19)	0.046(2)	- 0.0254(19)	0.000(3)	-0.001(2)
O4	0.0179(19)	0.0168(14)	0.0144(15)	- 0.0014(14)	- 0.0001(14)	0.0016(12)
O5	0.0131(18)	0.0171(14)	0.0152(15)	0.0006(13)	0.0037(13)	0.0013(12)
O6	0.0230(19)	0.0336(18)	0.038(2)	0.0136(16)	0.001(2)	-0.001(2)
O7	0.021(2)	0.0194(15)	0.0192(15)	0.0033(15)	- 0.0008(15)	- 0.0023(13)
O8	0.0115(15)	0.0159(13)	0.0233(14)	- 0.0010(12)	0.0019(18)	0.0015(14)
O9	0.0156(16)	0.0277(17)	0.0246(17)	- 0.0001(14)	- 0.0021(18)	- 0.0100(17)
O10	0.022(2)	0.0208(15)	0.0194(16)	0.0048(15)	0.0009(17)	- 0.0007(13)

Table S4: Interatomic distances. Symmetry codes: (i) $-x+1,-y,-z$; (ii) $-x+1,-y,-z+1$; (iii) $-x+3/2,y+1/2,-z-1/2$; (iv) $-x+3/2,y+1/2,-z+1/2$; (v) $-x+1/2,y-1/2,-z+1/2$; (vi) $x,y,z-1$; (vii) $-x+1,-y,-z-1$.

	d (Å)		d (Å)		d (Å)
Co1-O1	2.153(5)	Co2-O1 ^{vi}	1.998(4)	Al1-O4 ⁱ	1.924(4)
Co1-O1 ⁱ	2.153(5)	Co2-O1 ⁱ	1.998(4)	Al1-O5 ⁱⁱ	1.894(6)
Co1-O2	2.126(5)	Co2-O2 ^{vi}	2.370(6)	Al1-O7 ⁱⁱⁱ	1.842(6)
Co1-O2 ⁱ	2.126(5)	Co2-O2 ⁱ	2.370(6)	Al1-O8 ^{iv}	1.877(4)
Co1-O6	2.046(6)	Co2-O6	1.898(5)	Al1-O9 ^v	1.934(3)
Co1-O6 ⁱ	2.046(6)	Co2-O6 ^{vii}	1.898(5)	Al1-O10 ^{iv}	1.859(5)
P4-O3	1.590(4)	P2-O1	1.567(3)		
P4-O6	1.523(4)	P2-O4	1.534(6)		
P4-O7	1.488(5)	P2-O5	1.505(5)		
P4-O10	1.503(7)	P2-O8	1.523(3)		

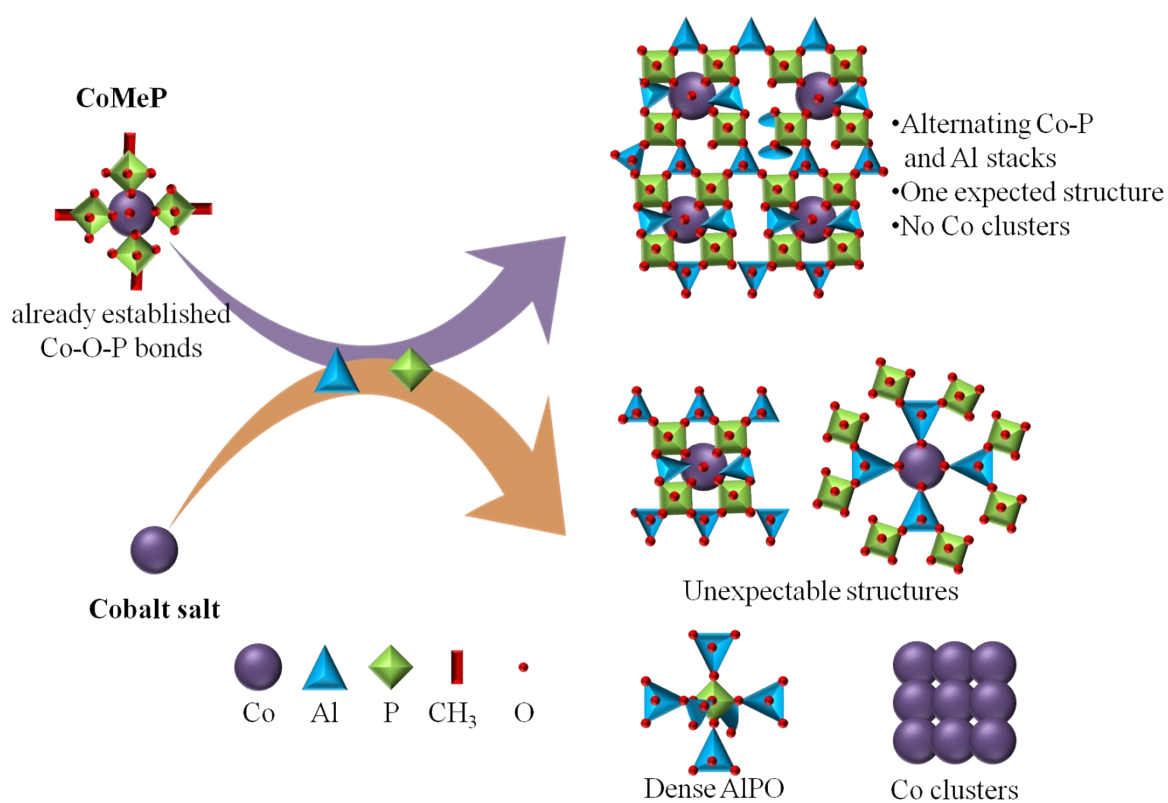


Figure S1. Simplified presentation of Co-AlPO formation starting from CoMeP versus cobalt salt.

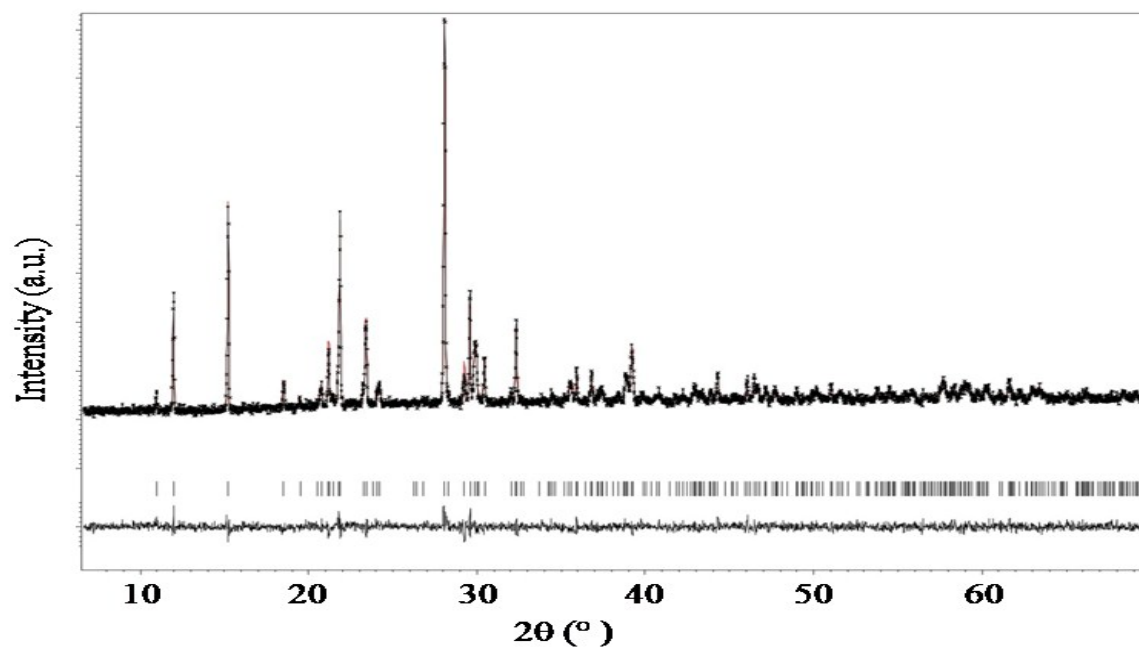


Figure S2. Rietveld refinement of the powder XRD pattern (Co-AlPO-4h) performed based on the structure obtained by single crystal XRD. The good matching confirms that the proposed structure of Co-AlPO is representative of the whole powder, which also appears free from secondary phase.