

Supporting Information

Controlled Synthesis of Water-Dispersible and Superparamagnetic Fe_3O_4 Nanomaterials by Microwave-assisted Solvothermal Method: From Nanocrystals to Nanoclusters

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Table S1. Experimental parameters for the synthesis of Fe₃O₄ nanoparticles by Orthogonal experiment.

Samples	m _{FeCl₃·6H₂O/g}	m _{Na₃Cit/g}	m _{NaAc·3H₂O/g}	t/h	T/°C
1	0.8115	0.4	2	1	200
2	0.8115	0.4	4	2	220
3	0.8115	0.4	6	3	240
4	1.3525	0.4	2	2	240
5	1.3525	0.4	4	3	200
6	1.3525	0.4	6	1	220
7	1.8935	0.4	2	3	220
8	1.8935	0.4	4	1	240
9	1.8935	0.4	6	2	200

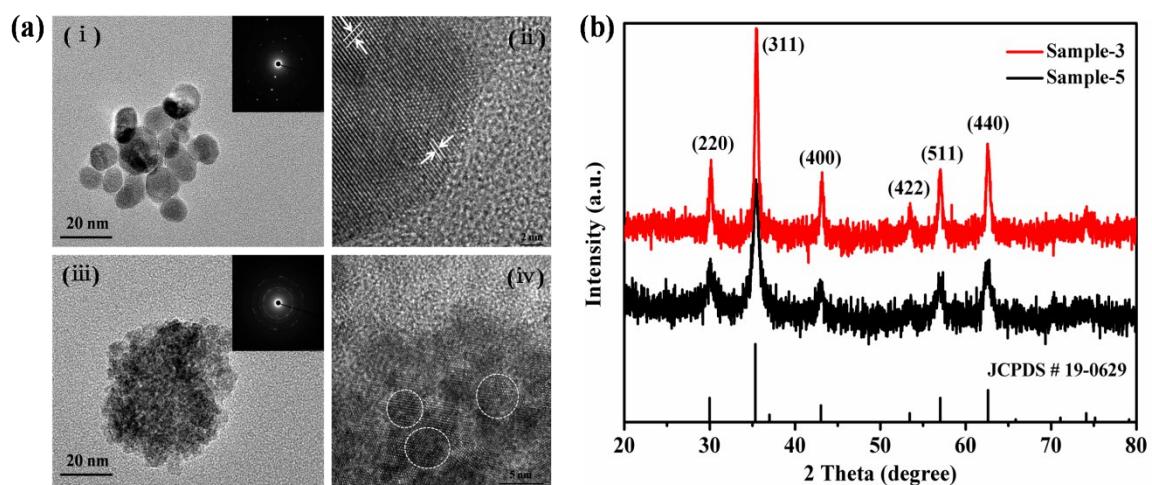


Figure S1. (a) Low magnification TEM images (i, iii) and high-resolution TEM images (ii, iv) and (b) XRD pattern s of SPION crystal (sample-3) and SPION cluster (sample -5).

Table S2. Experimental parameters for the synthesis of SPION crystals (Fe_3O_4 1-7) and SPION clusters (Fe_3O_4 8-14).

Samples	$m_{\text{FeCl}_3 \cdot 6\text{H}_2\text{O}}/\text{g}$	$m_{\text{Na}_3\text{Cit}}/\text{g}$	$m_{\text{NaAc} \cdot 3\text{H}_2\text{O}}/\text{g}$	t/h	T/ $^{\circ}\text{C}$
Fe_3O_4 -1	0.8115	0.4	4	3	200
Fe_3O_4 -2	0.8115	0.4	4	3	220
Fe_3O_4 -3	0.8115	0.4	4	3	240
Fe_3O_4 -4	0.8115	0.4	4	1	240
Fe_3O_4 -5	0.8115	0.4	4	2	240
Fe_3O_4 -6	0.8115	0.4	2	3	240
Fe_3O_4 -7	0.8115	0.4	6	3	240
Fe_3O_4 -8	1.3525	0.4	4	2	240
Fe_3O_4 -9	1.3525	0.4	4	2	200
Fe_3O_4 -10	1.3525	0.4	4	2	220
Fe_3O_4 -11	1.3525	0.4	4	1	200
Fe_3O_4 -12	1.3525	0.4	4	3	200
Fe_3O_4 -13	1.3525	0.4	2	2	200
Fe_3O_4 -14	1.3525	0.4	6	2	200

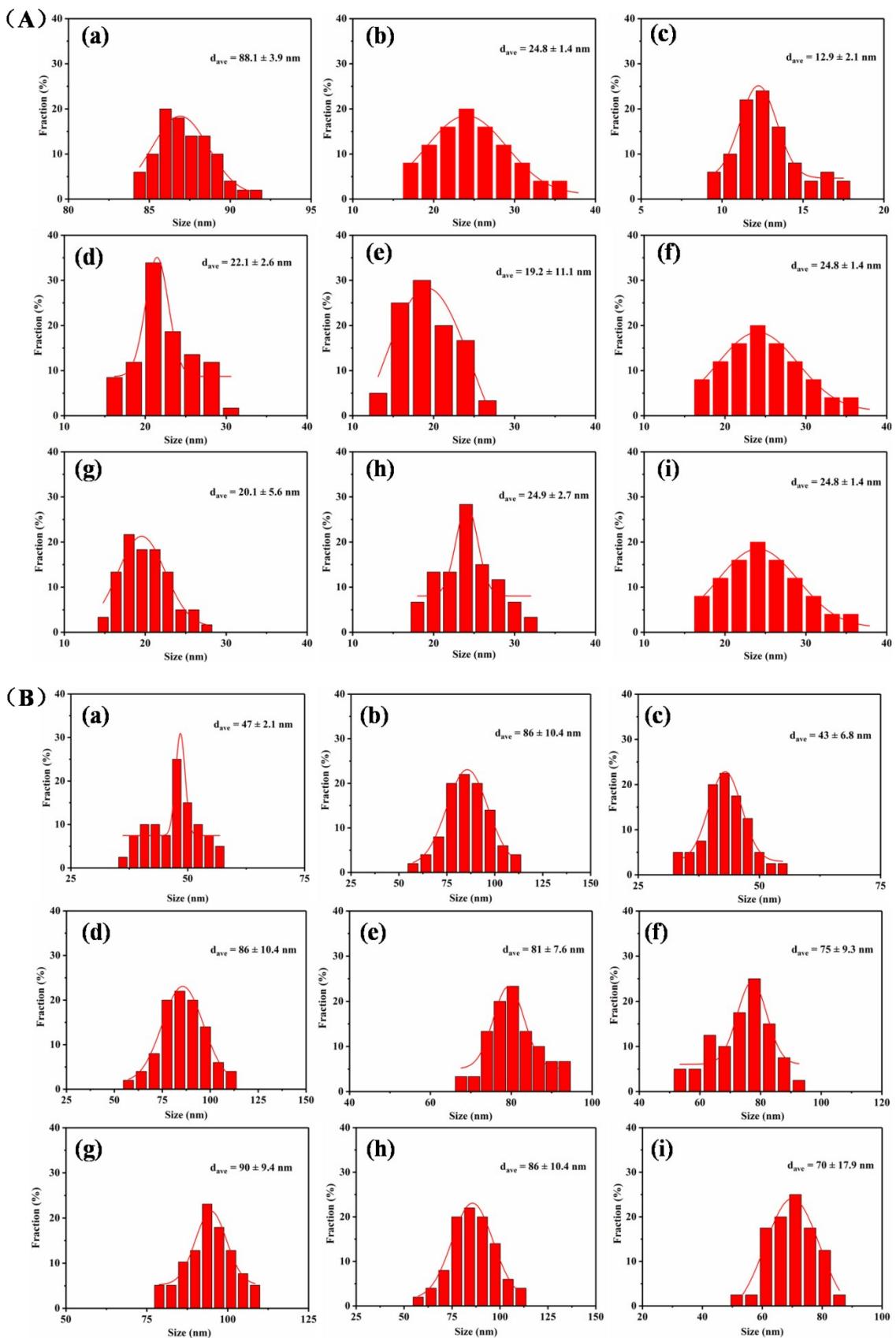


Figure S2. The size distribution of (A) SPION crystals (Fe_3O_4 1-7) and (B) SPION clusters (Fe_3O_4 8-14).

Table S3. Zeta potential for the synthesis of SPION crystals (Fe_3O_4 1-7) and SPION clusters (Fe_3O_4 8-14).

Samples	Fe_3O_4 -1	Fe_3O_4 -2	Fe_3O_4 -3	Fe_3O_4 -4	Fe_3O_4 -5	Fe_3O_4 -6	Fe_3O_4 -7
ζ potential ^a (mV)	-16.3	-12.5	-17.1	-15.6	-16.1	-34.3	-8.4
	± 0.6	± 0.4	± 0.3	± 0.3	± 0.2	± 0.9	± 0.1
Samples	Fe_3O_4 -8	Fe_3O_4 -9	Fe_3O_4 -10	Fe_3O_4 -11	Fe_3O_4 -12	Fe_3O_4 -13	Fe_3O_4 -14
ζ potential ^a (mV)	-22.9	-37.8	-33.5	-37.2	-23.4	-18.3	-14.2
	± 0.3	± 0.7	± 1.0	± 0.9	± 0.6	± 0.8	± 0.5

^a Values are averaged from three measurements in DI-water.



Figure S3. Photographs of the SPION crystal (Fe₃O₄-3, left) and SPION cluster (Fe₃O₄-9, right) suspended in DI-water.